



*Supplement of*

## **Mechanistic insight into the kinetic fragmentation of norpinonic acid in the gas phase: an experimental and density functional theory (DFT) study**

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## S1. Synthesis of investigated compound

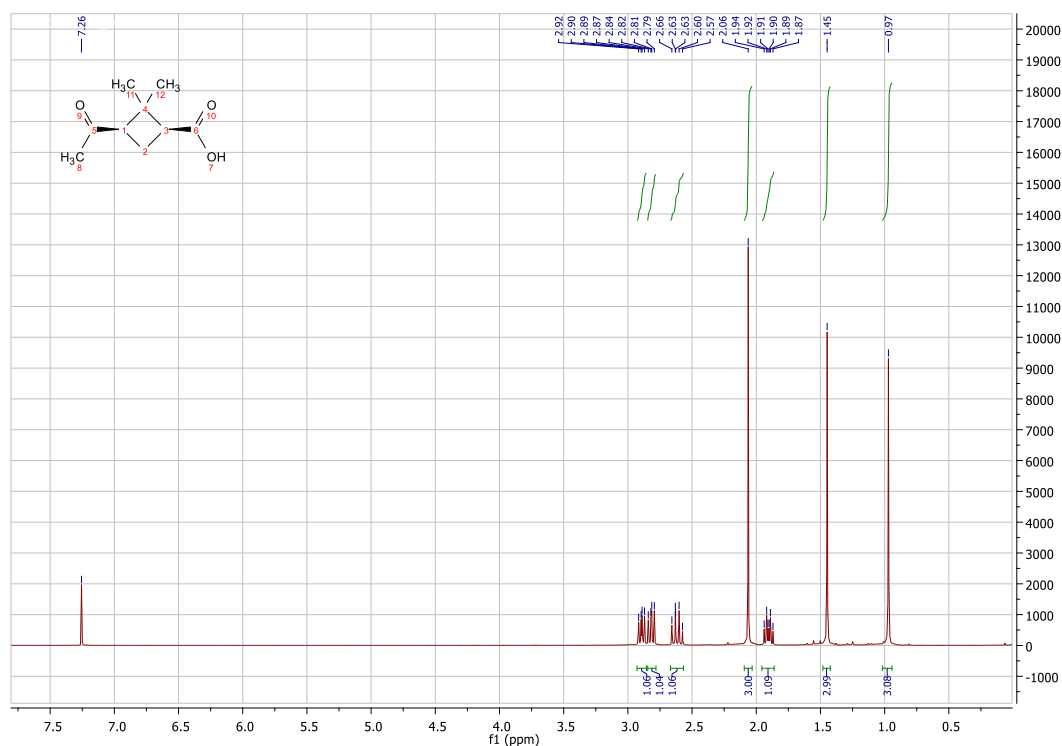
### S1.1 Synthesis protocol

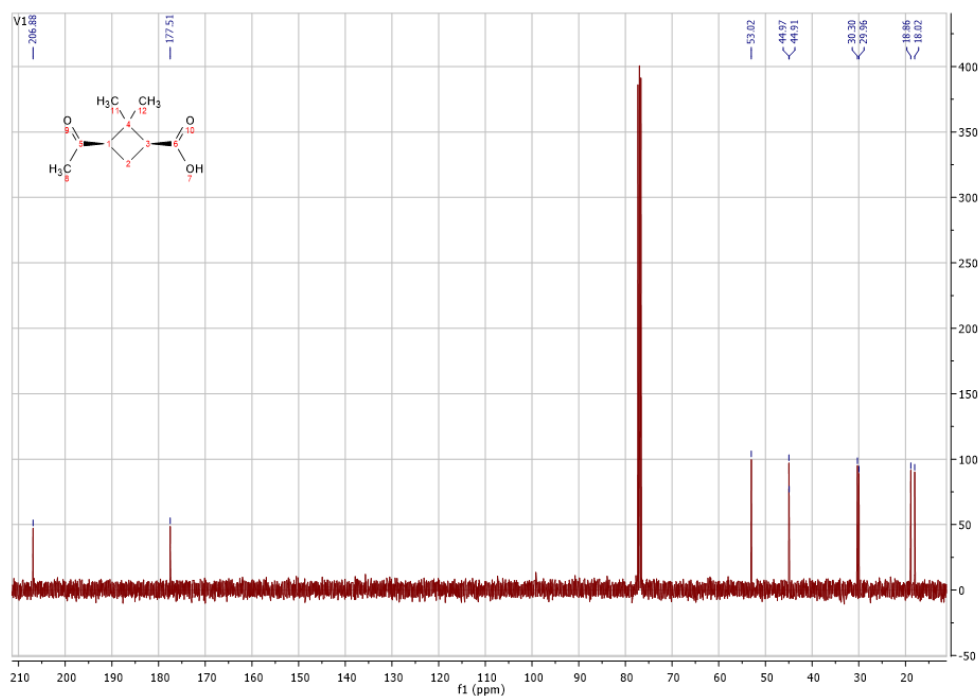
The commercially available (1S)-(-)-verbenone (CAS Registry No. 1196-01-6,  $\geq 94\%$ , Sigma-Aldrich) was used as starting material. The protocol for the synthesis of norpinonic acid covered one step - the oxidative cleavage of the double C=C bond using  $\text{RuCl}_3 \times \text{H}_2\text{O}$  (CAS Registry No. 14898-67-0, Sigma-Aldrich,  $\geq 99.98\%$ ). Other chemicals used in cis-norpinonic acid synthesis were as follows: sodium periodate (CAS Registry No. 7790-28-5, Sigma-Aldrich,  $\geq 99.8\%$ ), carbon tetrachloride (CAS Registry No. 56-23-5, Sigma-Aldrich,  $\geq 99.5\%$ ), acetonitrile (CAS Registry No. 75-05-8, POCH,  $\geq 99.9\%$ ), hexane (CAS Registry No. 110-54-3, POCH,  $\geq 99.9\%$ ), ethyl acetate (CAS Registry No. 60-29-7, POCH,  $\geq 99.9\%$ ), magnesium sulfate (CAS Registry No. 7487-88-9, POCH,  $\geq 98.5\%$ ).

To the stirred solution of (1S)-(-)-verbenone (2.5 mL, 16 mmol) in 2:2:3 carbon tetrachloride-acetonitrile-water (130 mL), catalytic  $\text{RuCl}_3$  hydrate (120 mg) and  $\text{NaIO}_4$  (13.4 g, 63 mmol) were added. The resulting mixture was stirred overnight at the room temperature, then diethyl ether (100 mL) was added and stirring for the next 5 min. Then mixture was extracted with diethyl ether ( $3 \times 100$  mL). The combined organic extracts were dried ( $\text{MgSO}_4$ ) and concentrated under reduced pressure. Norpinonic acid was isolated as white crystals after crystallization from diethyl ether (yield: 80%).

The structural identification of the final product based on IR, ESI-HR-MS,  $^1\text{H}$ ,  $^{13}\text{C}$  and 2D NMR analytical spectra proved the presence of cis-norpinonic acid stereoisomeric form.

### S1.2 Copies of $^1\text{H}$ and $^{13}\text{C}$ NMR spectra of isolated compound





## S2. Experimental procedures

### S2.1 Breakdown Curves

Collision spectra were recorded by varying the collision energy in incremental steps. Collision mass spectra were recorded with an energy resolution of 8–30 eV in the lab frame and 2–10 minutes of collection time at each step. Creation of the total breakdown curves was performed using absolute peak heights. Breakdown curves at five different argon pressures are presented in.

### S2.2 Extrapolation Procedure

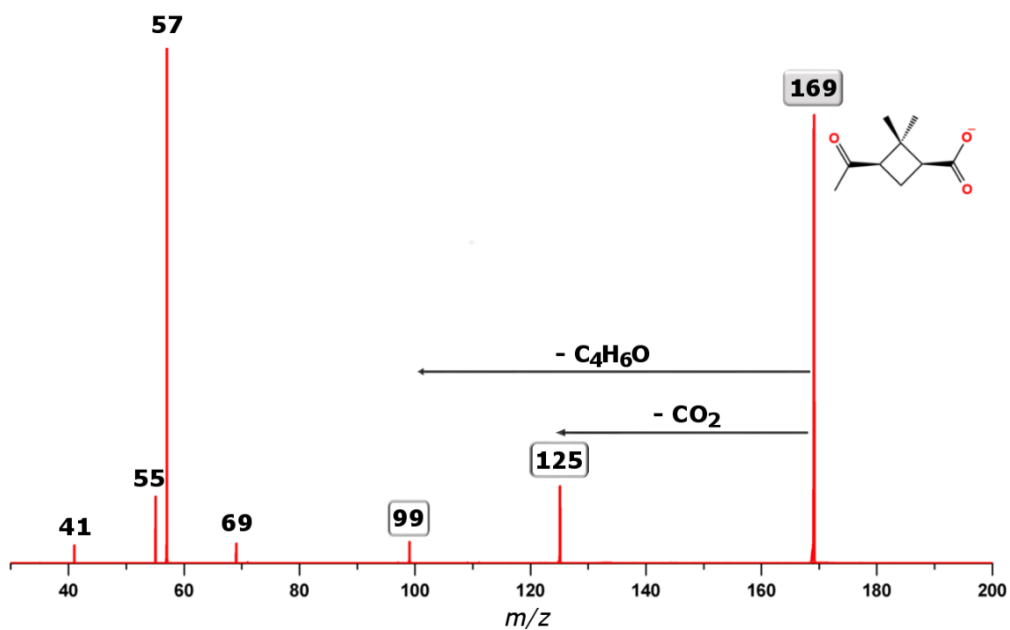
To determine the onset/threshold energies (at each gas pressure) to enable a comparison of the energetics of the observed processes, we used a simple extrapolation procedure. By performing a linear fit of the approximately linearly rising section of the breakdown curve, we define the onset energy at each gas pressure by calculation of the energy (X value) at zero intensity (Y = 0). To compare the experimental and theoretical energy values, we opted also for gas pressure extrapolation. We used the energies taken from extrapolation at five different gas pressures to define the onset energy by calculation of the energy (Y value) at zero gas pressure (X = 0). Summaries are presented in Tables.

### S2.3 Bimolecular reaction

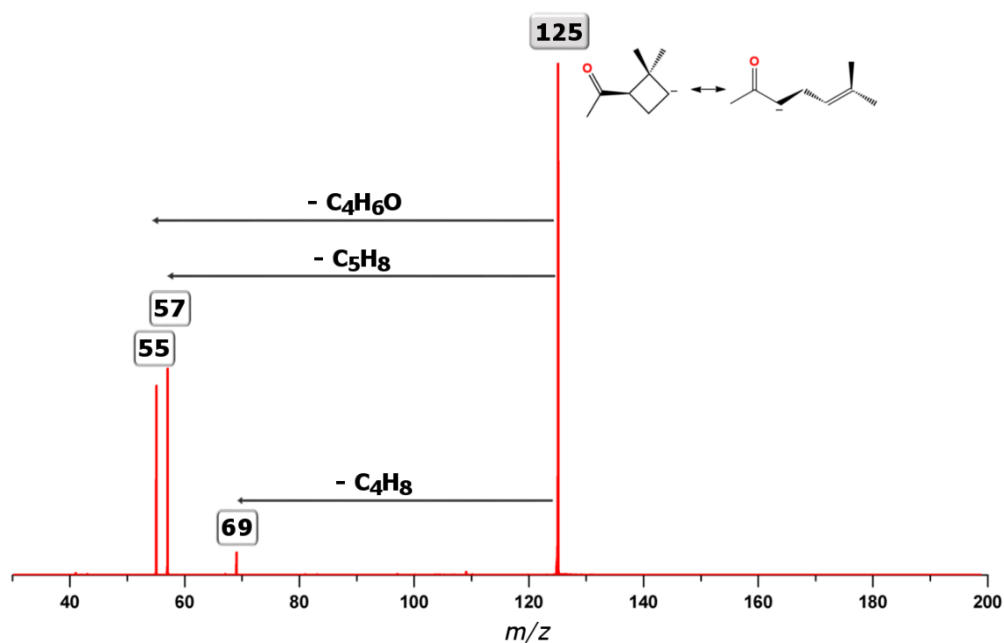
To perform the gas-phase reactions, we introduce neutral reagent vapors into the collision cell by attaching a flask filled with each reagent to the gas inlet system. The results of these experiments for reaction between methyl thiocyanate ( $\text{CH}_3\text{SCN}$ ) or dimethyl disulfide ( $\text{CH}_3\text{SSCH}_3$ ) and all ions are presented in Figures.

## S3. Experimental results

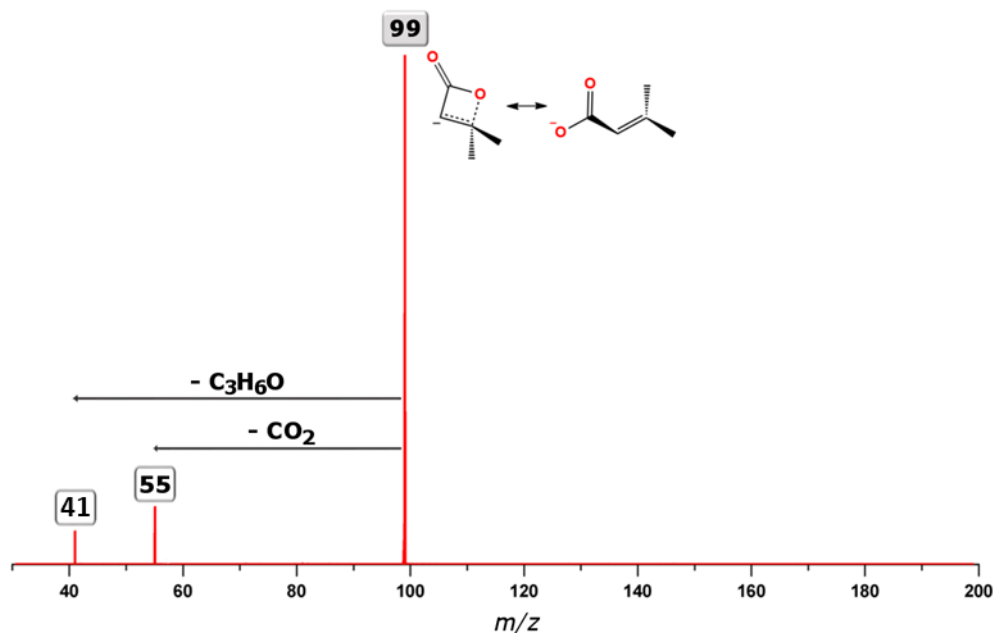
### S3.1 Fragment ion mass spectra



**Figure S1.** Fragment ion mass spectrum of  $C_9H_{13}O_3^-$  ( $m/z$  169) recorded with a ToF voltage of 3kV, taken at a collision energy of 3.8 eV (CM) with argon collision gas at nominal pressure of  $3.54 \times 10^{-4}$  mBar.

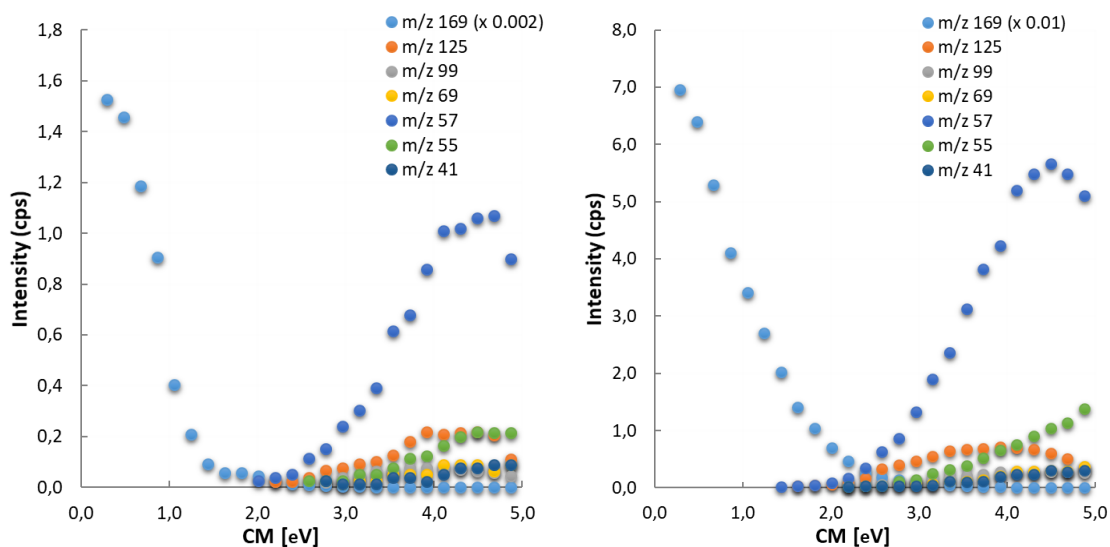


**Figure S2.** Fragment ion mass spectrum of  $C_8H_{13}O^-$  ( $m/z$  125) recorded with a ToF voltage of 3kV, taken at a collision energy of 4.1 eV (CM) with argon collision gas at nominal pressure of  $3.54 \times 10^{-4}$  mBar.

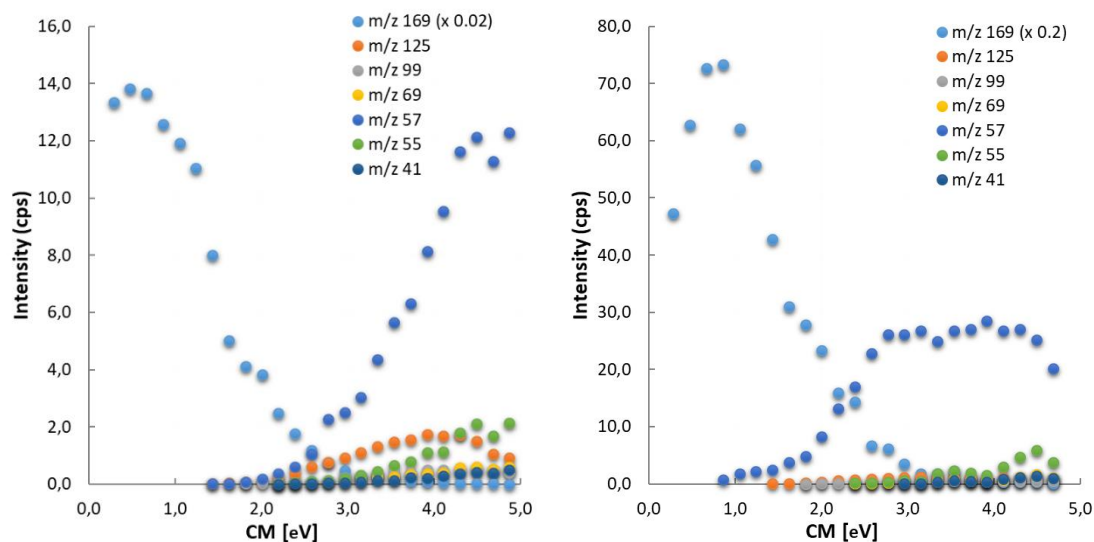


**Figure S3.** Fragment ion mass spectrum of  $C_5H_7O_2^-$  ( $m/z$  99) recorded with a ToF voltage of 3kV, taken at a collision energy of 4.3 eV (CM) with argon collision gas at nominal pressure of  $3.54 \times 10^{-4}$  mBar.

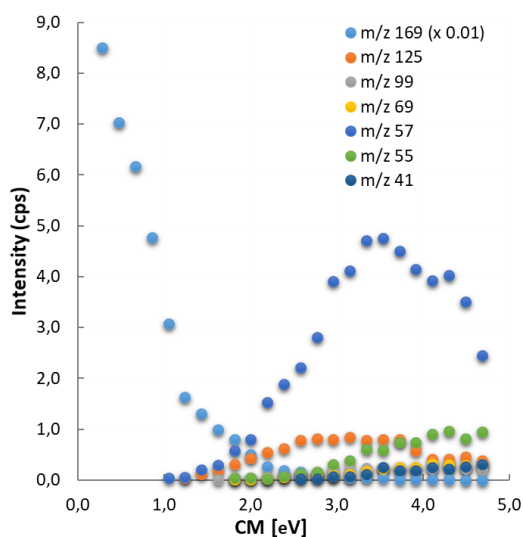
### S3.2 Breakdown curves and extrapolation results



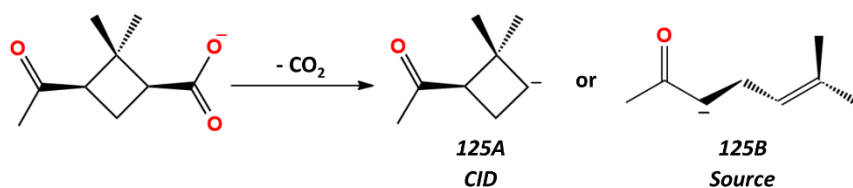
**Figure S4.** Breakdown curves for  $C_9H_{13}O_3^-$  ( $m/z$  169) recorded at an argon collision gas pressure of  $1.06 \times 10^{-4}$  mbar (left) and  $2.08 \times 10^{-4}$  mbar (right). The intensity of the  $m/z$  169 anion have been multiplied by a scaling factor for readability.



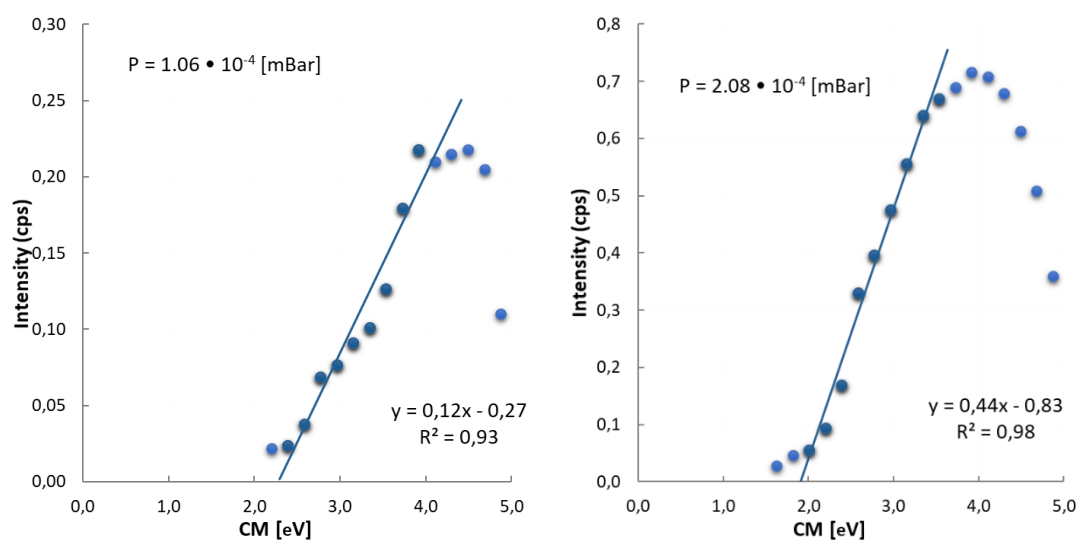
**Figure S5.** Breakdown curves for  $C_9H_{13}O_3^-$  ( $m/z$  169) recorded at an argon collision gas pressure of  $3.04 \times 10^{-4}$  mbar (left) and  $4.18 \times 10^{-4}$  mbar (right). The intensity of the  $m/z$  169 anion have been multiplied by a scaling factor for readability.



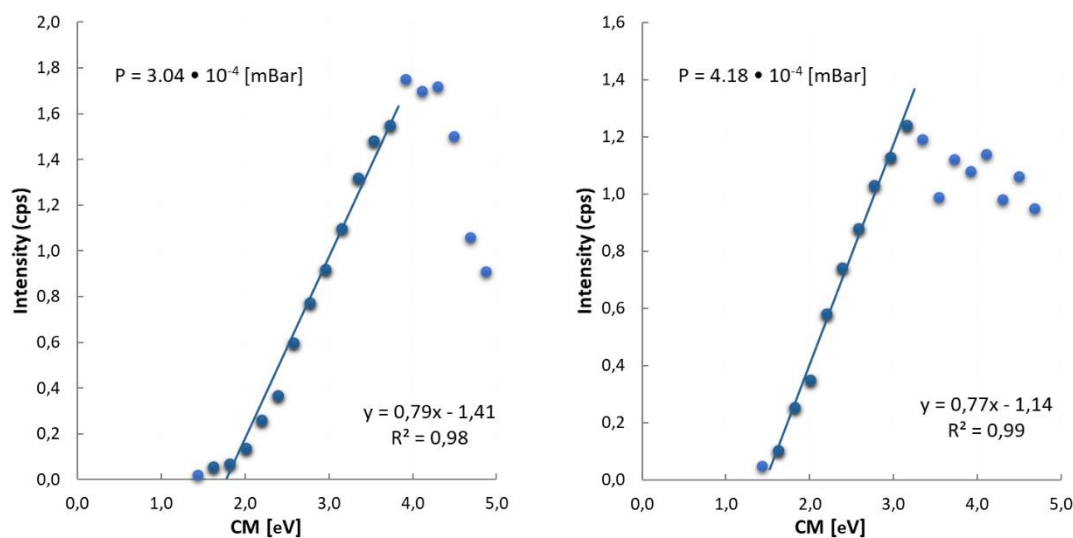
**Figure S6.** Breakdown curve for  $C_9H_{13}O_3^-$  ( $m/z$  169) recorded at an argon collision gas pressure of  $5.09 \times 10^{-4}$  mbar. The intensity of the  $m/z$  169 anion have been multiplied by a scaling factor for readability.



**Figure S7.** Decarboxylation of  $\text{C}_9\text{H}_{13}\text{O}_3^-$  ( $m/z$  169) ion leading to the formation of  $\text{C}_8\text{H}_{13}\text{O}^-$  ( $m/z$  125) ion. *Comment: From computational and experimental study is known that cyclic structure of  $m/z$  125 (125A) is formed in collision-induced dissociation (CID) experiment, while linear structure of  $m/z$  125 (125B) is formed in an ion source during ionization process.*

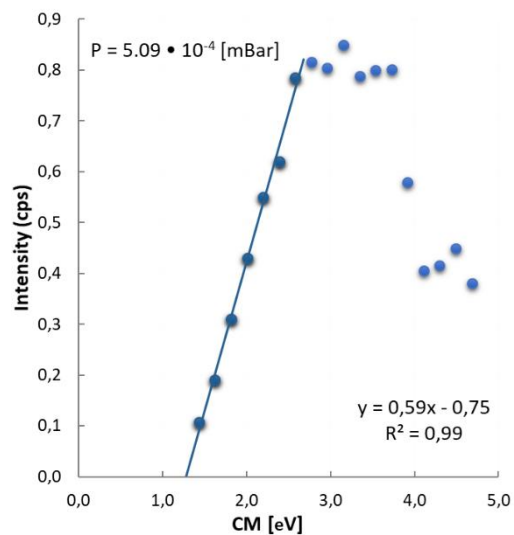


**Figure S8.** Extrapolation procedure for the decarboxylation of  $\text{C}_9\text{H}_{13}\text{O}_3^-$  ( $m/z$  169) recorded at an argon collision gas pressure of  $1.06 \times 10^{-4}$  mbar (left) and  $2.08 \times 10^{-4}$  mbar (right).

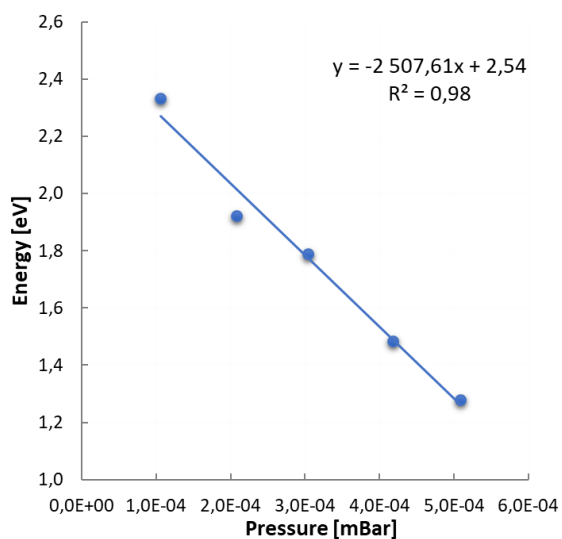


**Figure S9.** Extrapolation procedure for the decarboxylation of  $\text{C}_9\text{H}_{13}\text{O}_3^-$  ( $m/z$  169) recorded at an argon collision gas pressure of  $3.04 \times 10^{-4}$  mbar (left) and  $4.18 \times 10^{-4}$  mbar (right).





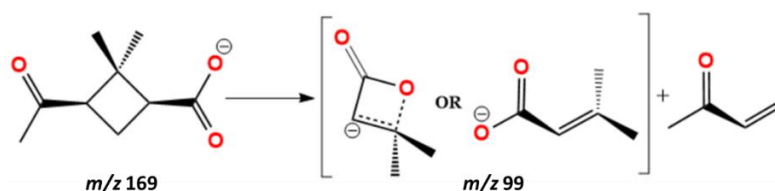
**Figure S10.** Extrapolation procedure for the decarboxylation of  $C_9H_{13}O_3^-$  ( $m/z$  169) recorded at an argon collision gas pressure of  $5.09 \times 10^{-4}$  mbar.



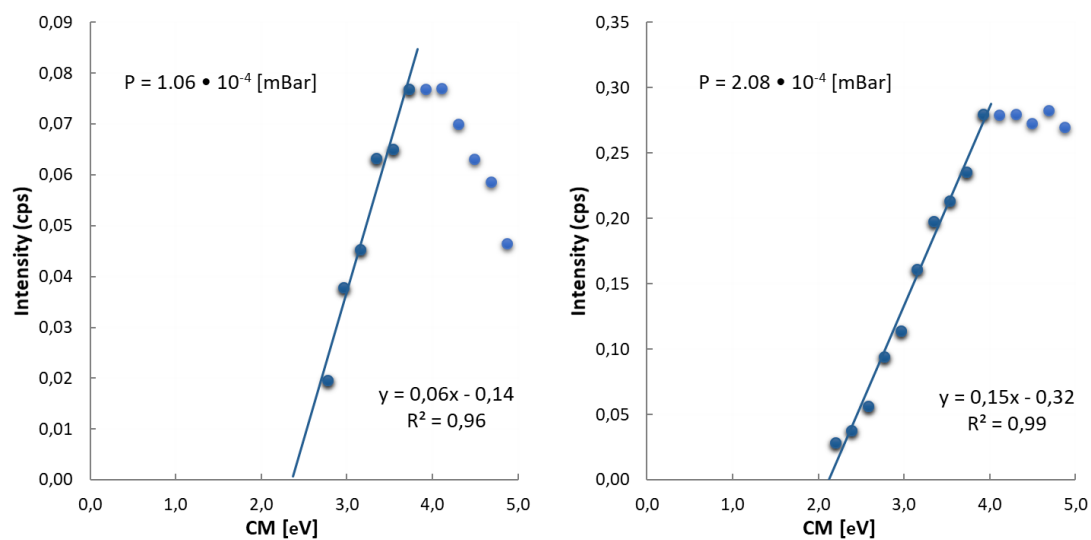
**Figure S11.** Extrapolation procedure for the decarboxylation of  $C_9H_{13}O_3^-$  ( $m/z$  169) at a gas pressure of 0 mbar.

**Table S1.** Summary of the values from the extrapolation procedure for decarboxylation of  $C_9H_{13}O_3^-$  ( $m/z$  169).

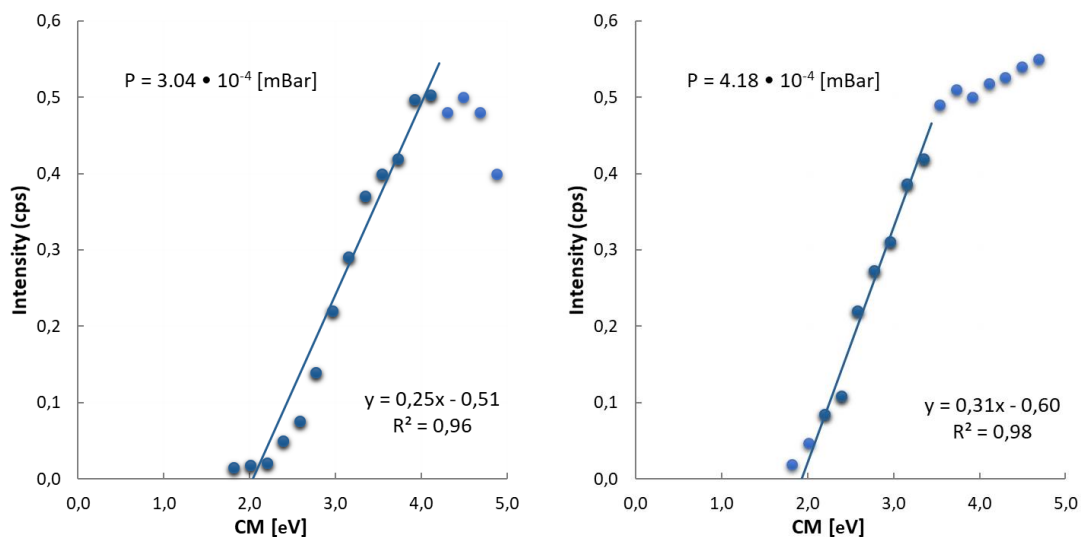
Pressure [mBar]:	Intercept	Slope	X at Y=0 [eV]	[kJ/mol]
$1.06 \cdot 10^{-4}$	-0.27	0.12	2,33	225
$2.08 \cdot 10^{-4}$	-0.83	0.44	1,92	186
$3.04 \cdot 10^{-4}$	-1.41	0.79	1,79	173
$4.18 \cdot 10^{-4}$	-1.12	0.77	1,49	143
$5.09 \cdot 10^{-4}$	-0.75	0.59	1,28	124
Extrapolated pressure [mBar]:	Intercept	Slope	Y at X=0 [eV]	[kJ/mol]
0	-2507,61	2,54	2,54	245



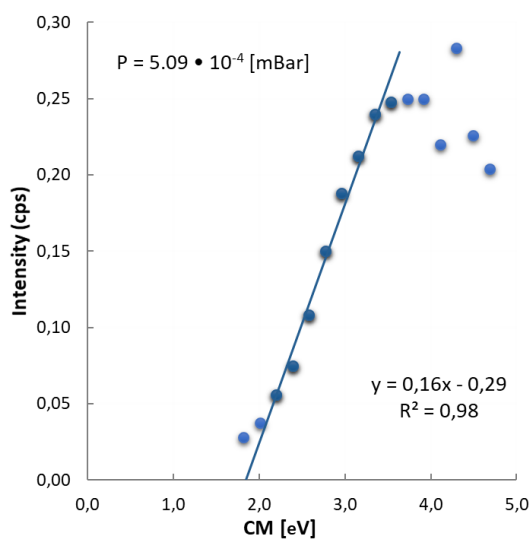
**Figure S12.** Elimination of  $C_4H_6O$  from  $C_9H_{13}O_3^-$  ( $m/z$  169) leading to the formation of  $C_5H_7O_2^-$  ( $m/z$  99) an ion, which can have two possible structure.



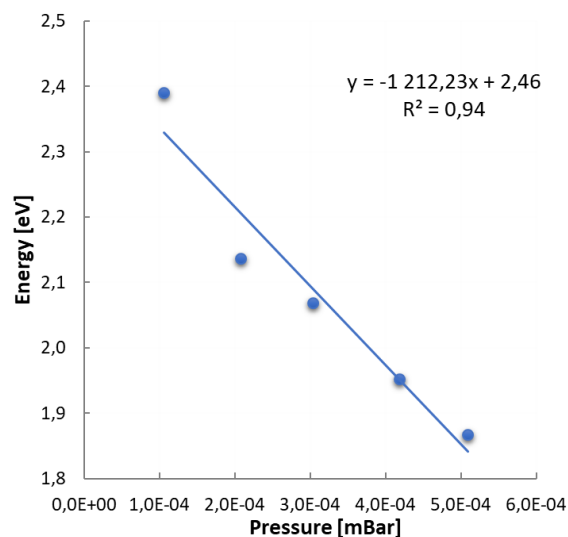
**Figure S13.** Extrapolation procedure for the elimination of  $C_4H_6O$  from  $C_9H_{13}O_3^-$  ( $m/z$  169) recorded at an argon collision gas pressure of  $1.06 \times 10^{-4}$  mbar (left) and  $2.08 \times 10^{-4}$  mbar (right).



**Figure S14.** Extrapolation procedure for the elimination of  $C_4H_6O$  from  $C_9H_{13}O_3^-$  ( $m/z$  169) recorded at an argon collision gas pressure of  $3.04 \times 10^{-4}$  mbar (left) and  $4.18 \times 10^{-4}$  mbar (right).



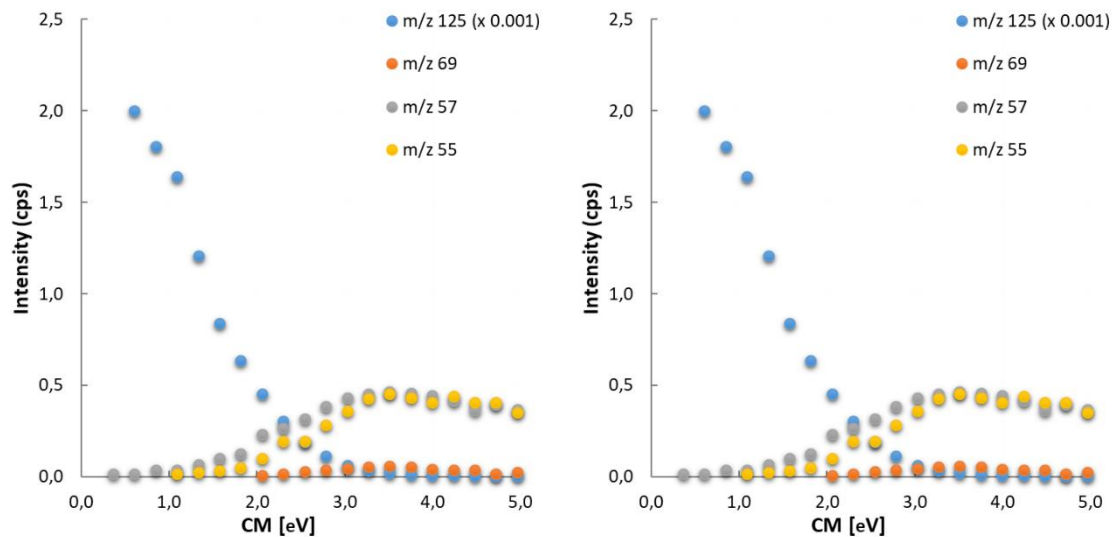
**Figure S15.** Extrapolation procedure for the elimination of  $C_4H_6O$  from  $C_9H_{13}O_3^-$  ( $m/z$  169) recorded at an argon collision gas pressure of  $5.09 \times 10^{-4}$  mbar.



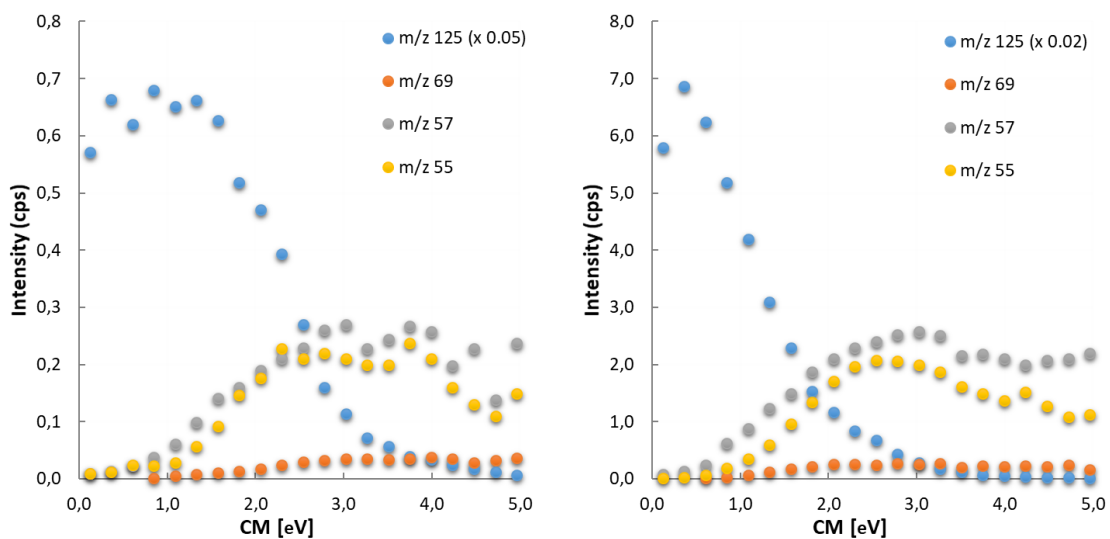
**Figure S16.** Extrapolation procedure for the elimination of  $C_4H_6O$  from  $C_9H_{13}O_3^-$  ( $m/z$  169) at a gas pressure of 0 mbar.

**Table S2.** Summary of the values from the extrapolation procedure for the elimination of  $C_4H_6O$  from  $C_9H_{13}O_3^-$  ( $m/z$  169).

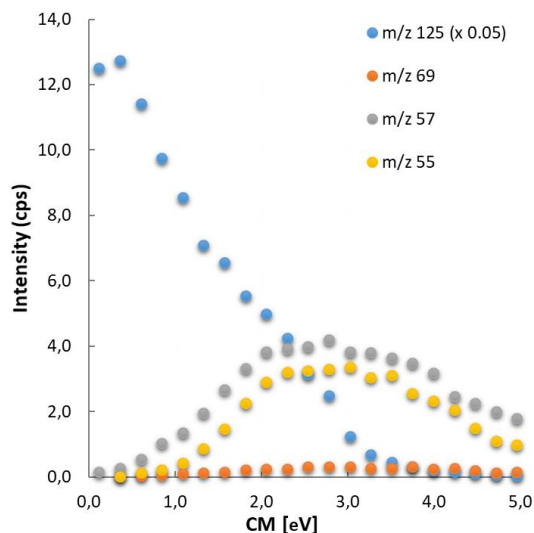
Pressure [mBar]:	Intercept	Slope	X at Y=0 [eV]	[kJ/mol]
$1.06 \cdot 10^{-4}$	-0.14	0.06	2,39	231
$2.08 \cdot 10^{-4}$	-0.32	0.15	2,14	206
$3.04 \cdot 10^{-4}$	-0.51	0.25	2,07	200
$4.18 \cdot 10^{-4}$	-0.60	0.31	1,95	188
$5.09 \cdot 10^{-4}$	-0.29	0.16	1,87	180
Extrapolated pressure [mBar]:	Intercept	Slope	Y at X=0 [eV]	[kJ/mol]
0	-1212,23	2,46	2,46	237



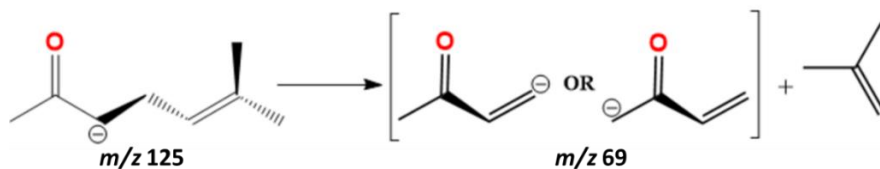
**Figure S17.** Breakdown curves for  $C_8H_{13}O^-$  ( $m/z$  125) recorded at an argon collision gas pressure of  $1.06 \times 10^{-4}$  mbar (left) and  $2.05 \times 10^{-4}$  mbar (right). The intensity of the  $m/z$  125 anion have been multiplied by a scaling factor for readability.



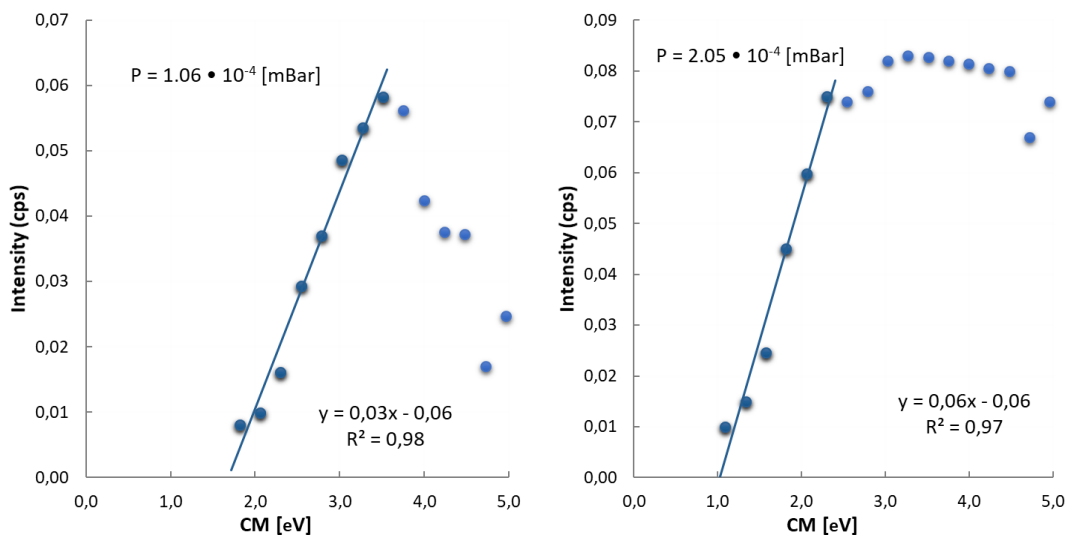
**Figure S18.** Breakdown curves for  $C_8H_{13}O^-$  ( $m/z$  125) recorded at an argon collision gas pressure of  $3.17 \times 10^{-4}$  mbar (left) and  $4.14 \times 10^{-4}$  mbar (right). The intensity of the  $m/z$  125 anion have been multiplied by a scaling factor for readability.



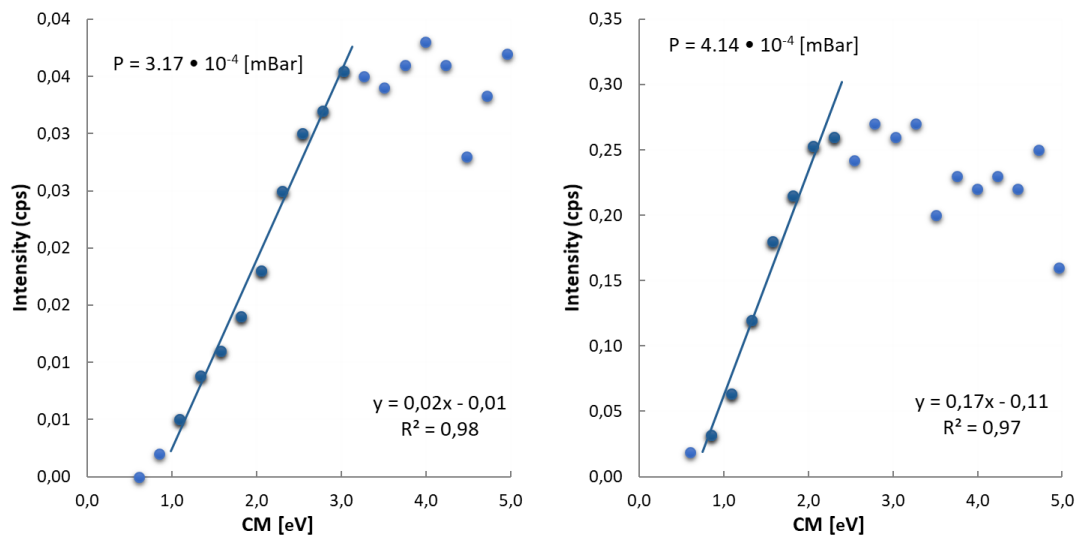
**Figure S19.** Breakdown curve for  $C_8H_{13}O^-$  ( $m/z$  125) recorded at an argon collision gas pressure of  $5.07 \times 10^{-4}$  mbar. The intensity of the  $m/z$  125 anion have been multiplied by a scaling factor for readability.



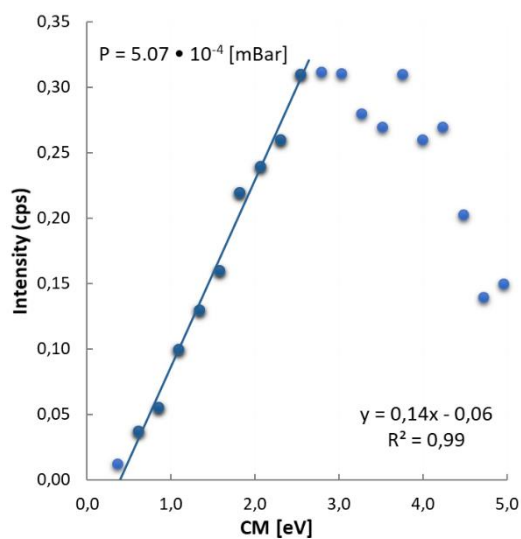
**Figure S20.** Elimination reaction of  $C_4H_8$  from  $C_8H_{13}O^-$  ( $m/z$  125) leading to the formation of  $C_4H_5O^-$  ( $m/z$  69) an ion, which can have two possible structure.



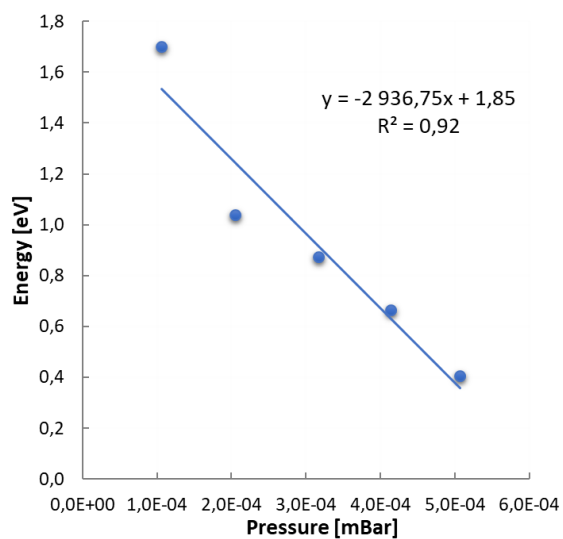
**Figure S21.** Extrapolation procedure for the elimination of  $C_4H_8$  from  $C_8H_{13}O^-$  ( $m/z$  125) recorded at an argon collision gas pressure gas pressure of  $1.06 \times 10^{-4}$  mbar (left) and  $2.05 \times 10^{-4}$  mbar (right).



**Figure S22.** Extrapolation procedure for the elimination of  $C_4H_8$  from  $C_8H_{13}O^-$  ( $m/z$  125) recorded at an argon collision gas pressure  $3.17 \times 10^{-4}$  mbar (left) and  $4.14 \times 10^{-4}$  mbar (right).



**Figure S23.** Extrapolation procedure the elimination of  $C_4H_8$  from  $C_8H_{13}O^-$  ( $m/z$  125) recorded at an argon collision gas pressure of  $5.07 \times 10^{-4}$  mbar.

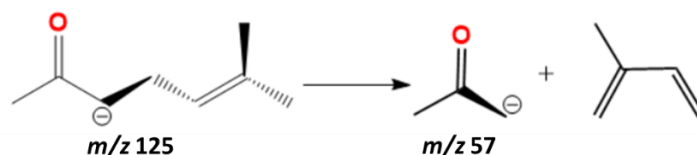


**Figure S24.** Extrapolation procedure for the elimination of  $C_4H_6O$  from  $C_8H_{13}O^-$  ( $m/z$  125) at a gas pressure of 0 mbar.

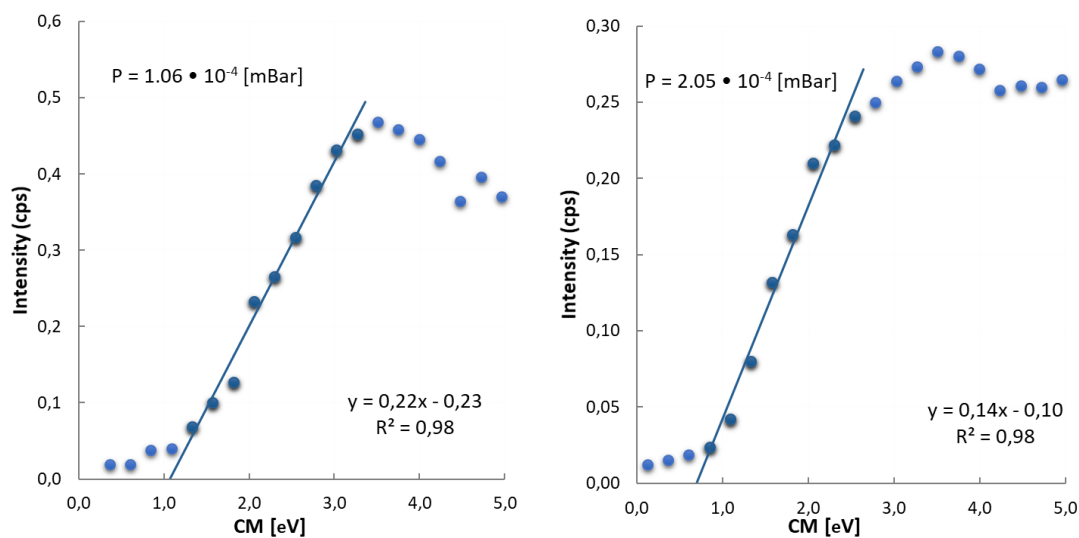


**Table S3.** Summary of the values from the extrapolation procedure for the elimination of  $C_4H_6O$  from  $C_8H_{13}O^-$  ( $m/z$  125).

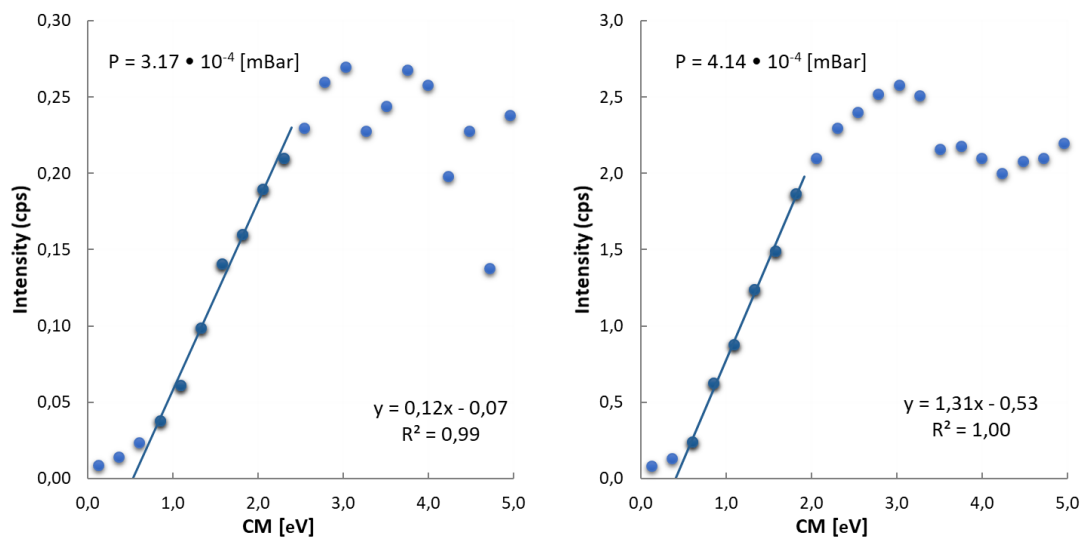
Pressure [mBar]:	Intercept	Slope	X at Y=0 [eV]	[kJ/mol]
$1.06 \cdot 10^{-4}$	-0.06	0.03	1,70	164
$2.05 \cdot 10^{-4}$	-0.06	0.06	1,04	100
$3.17 \cdot 10^{-4}$	-0.01	0.02	0,87	84
$4.14 \cdot 10^{-4}$	-0.11	0.17	0,66	64
$5.07 \cdot 10^{-4}$	-0.06	0.14	0,40	39
Extrapolated pressure [mBar]:	Intercept	Slope	Y at X=0 [eV]	[kJ/mol]
0	-2936,75	1,85	1,85	178



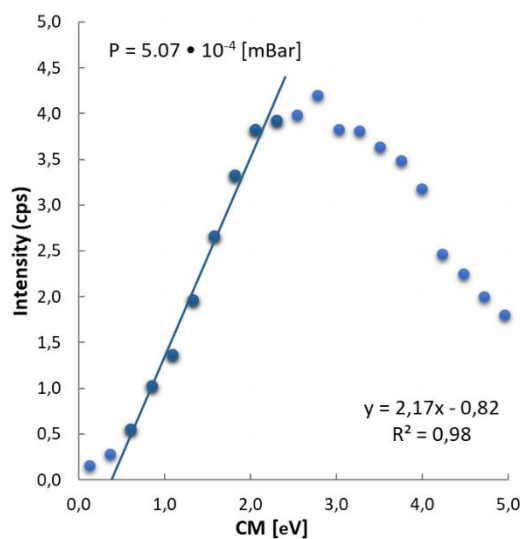
**Figure S25.** Elimination reaction of  $C_5H_8$  from  $C_8H_{13}O^-$  ( $m/z$  125) leading to the formation of  $C_3H_5O^-$  ( $m/z$  57) an ion.



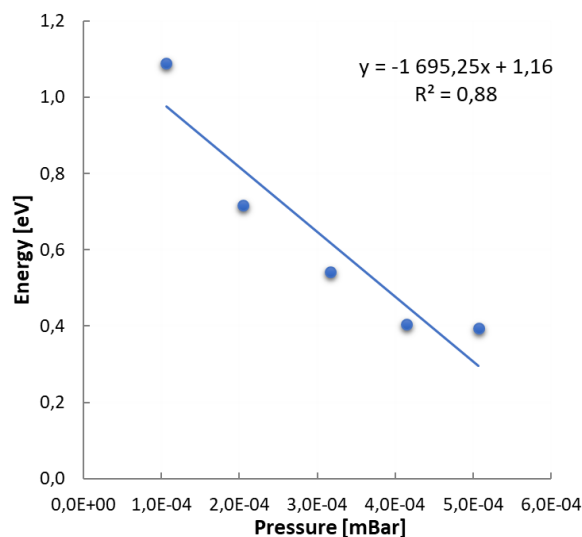
**Figure S26.** Extrapolation procedure for the elimination of  $C_5H_8$  from  $C_8H_{13}O^-$  ( $m/z$  125) recorded at an argon collision gas pressure gas pressure of  $1.06 \times 10^{-4}$  mbar (left) and  $2.05 \times 10^{-4}$  mbar (right).



**Figure S27.** Extrapolation procedure for the elimination of  $C_5H_8$  from  $C_8H_{13}O^-$  ( $m/z$  125) recorded at an argon collision gas pressure  $3.17 \times 10^{-4}$  mbar (left) and  $4.14 \times 10^{-4}$  mbar (right).



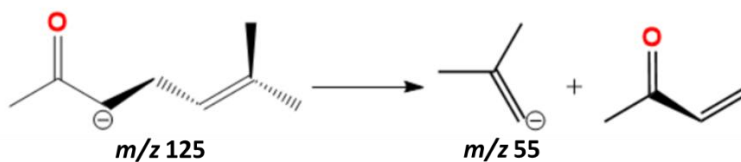
**Figure S28.** Extrapolation procedure the elimination of  $C_5H_8$  from  $C_8H_{13}O^-$  ( $m/z$  125) recorded at an argon collision gas pressure of  $5.07 \times 10^{-4}$  mbar.



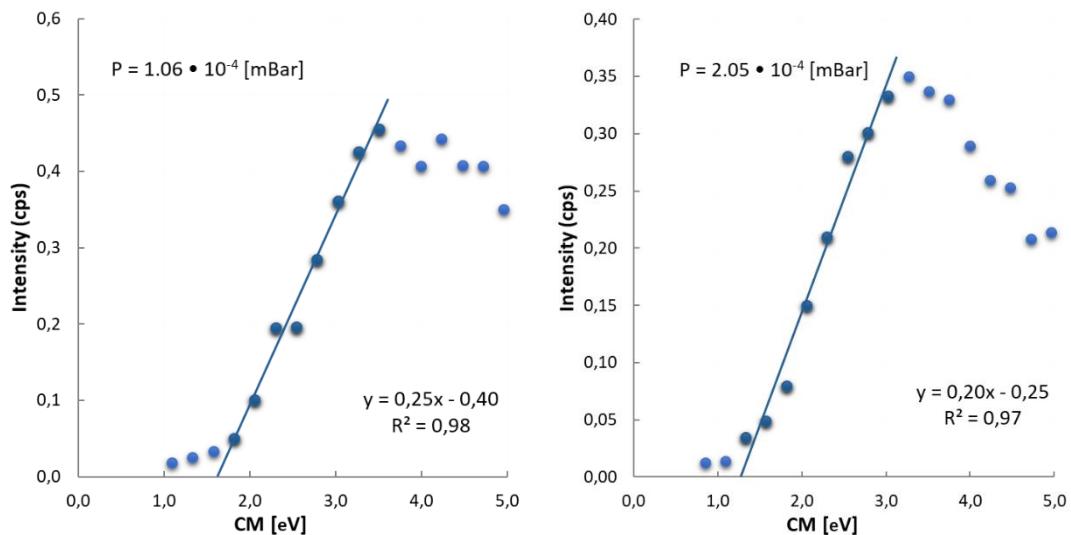
**Figure S29.** Extrapolation procedure for the elimination of  $C_5H_8$  from  $C_8H_{13}O^-$  ( $m/z$  125) at a gas pressure of 0 mbar.

**Table S4.** Summary of the values from the extrapolation procedure for the elimination of  $C_4H_6O$  from  $C_8H_{13}O^-$  ( $m/z$  125).

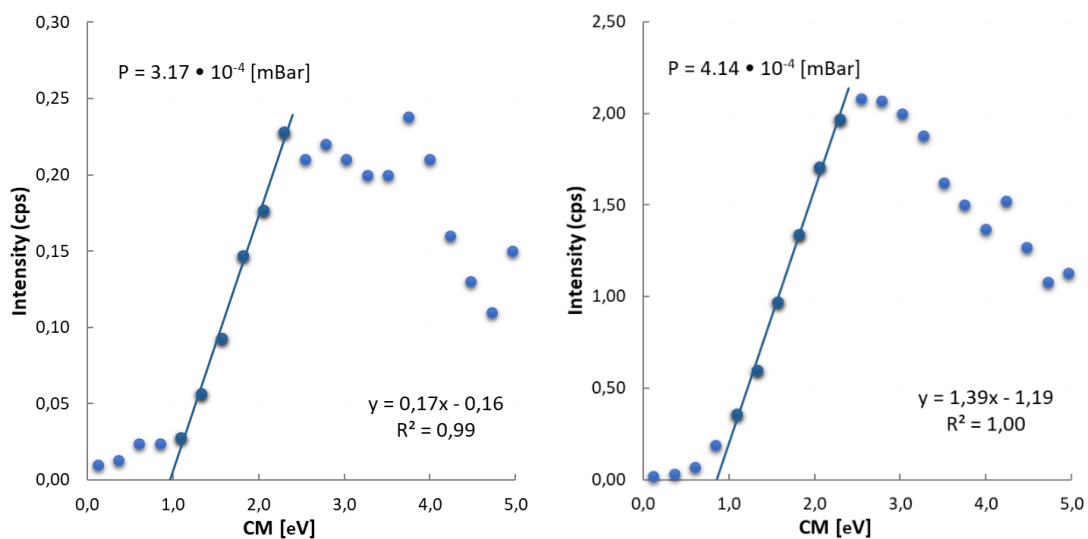
Pressure [mBar]:	Intercept	Slope	X at Y=0 [eV]	[kJ/mol]
$1.06 \cdot 10^{-4}$	-0.06	0.03	1,09	105
$2.05 \cdot 10^{-4}$	-0.06	0.06	0,72	69
$3.17 \cdot 10^{-4}$	-0.01	0.02	0,54	52
$4.14 \cdot 10^{-4}$	-0.11	0.17	0,41	39
$5.07 \cdot 10^{-4}$	-0.06	0.14	0,39	38
Extrapolated pressure [mBar]:	Intercept	Slope	Y at X=0 [eV]	[kJ/mol]
0	-1695,25	1,16	1,16	111



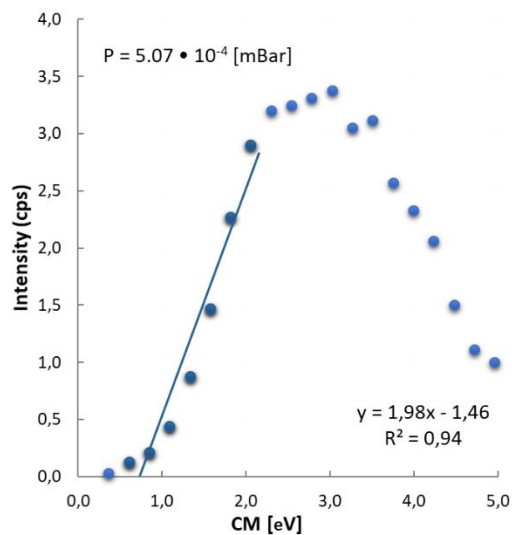
**Figure S30.** Elimination of  $C_4H_6O$  from  $C_8H_{13}O^-$  ( $m/z$  125) leading to the formation of  $C_4H_7^-$  ( $m/z$  55) a ion.



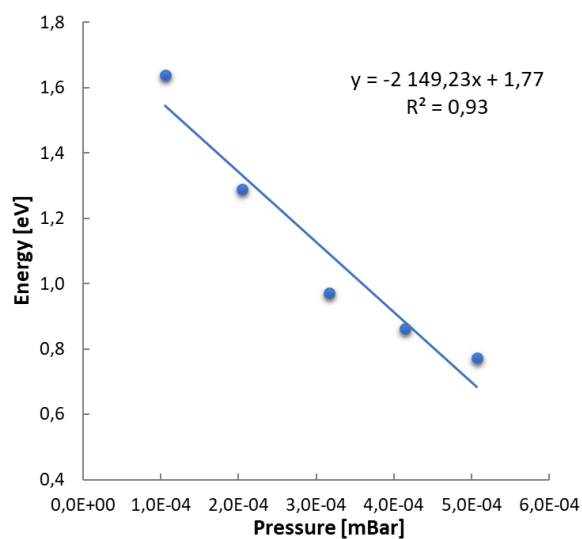
**Figure S31.** Extrapolation procedure for the elimination of  $C_4H_6O$  from  $C_3H_{13}O^-$  ( $m/z$  125) recorded at an argon collision gas pressure gas pressure of  $1.06 \times 10^{-4}$  mbar (left) and  $2.05 \times 10^{-4}$  mbar (right).



**Figure S32.** Extrapolation procedure for the elimination of  $C_4H_6O$  from  $C_3H_{13}O^-$  ( $m/z$  125) recorded at an argon collision gas pressure  $3.17 \times 10^{-4}$  mbar (left) and  $4.14 \times 10^{-4}$  mbar (right).



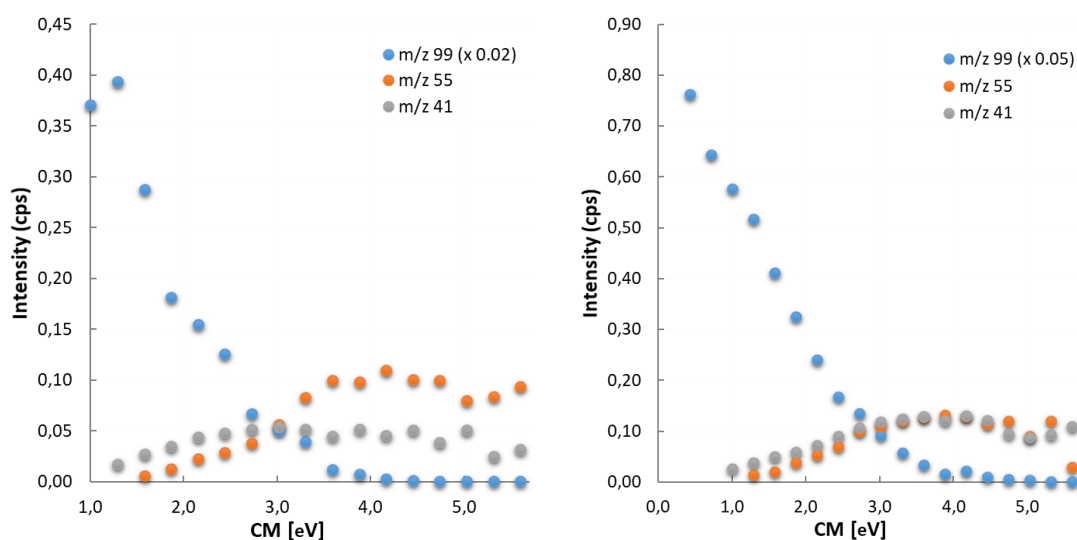
**Figure S33.** Extrapolation procedure the elimination of  $C_4H_6O$  from  $C_8H_{13}O^-$  ( $m/z$  125) recorded at an argon collision gas pressure of  $5.07 \times 10^{-4}$  mbar.



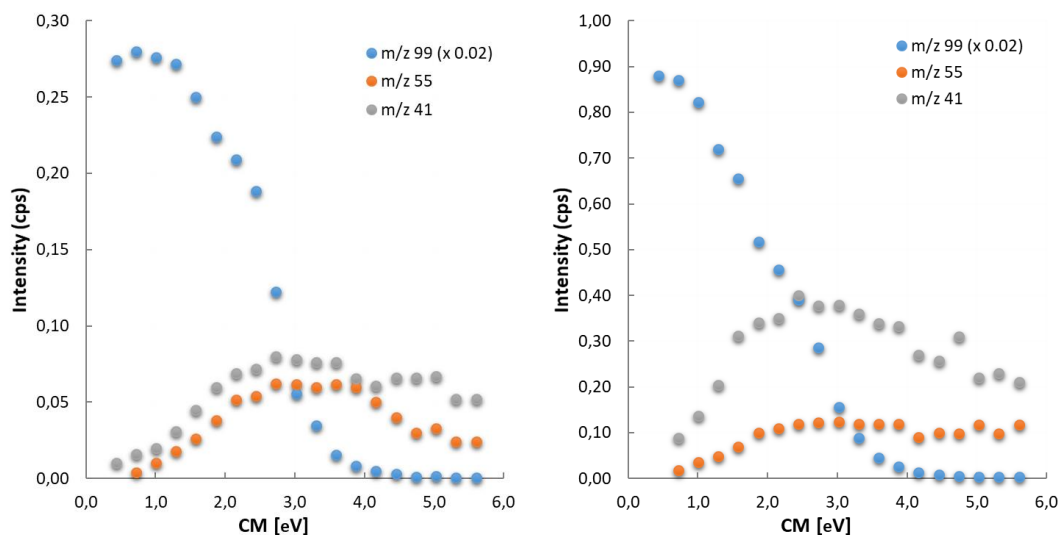
**Figure S34.** Extrapolation procedure for the elimination of  $C_5H_8$  from  $C_8H_{13}O^-$  ( $m/z$  125) at a gas pressure of 0 mbar.

**Table S5.** Summary of the values from the extrapolation procedure for the elimination of  $C_4H_6O$  from  $C_8H_{13}O^-$  ( $m/z$  125).

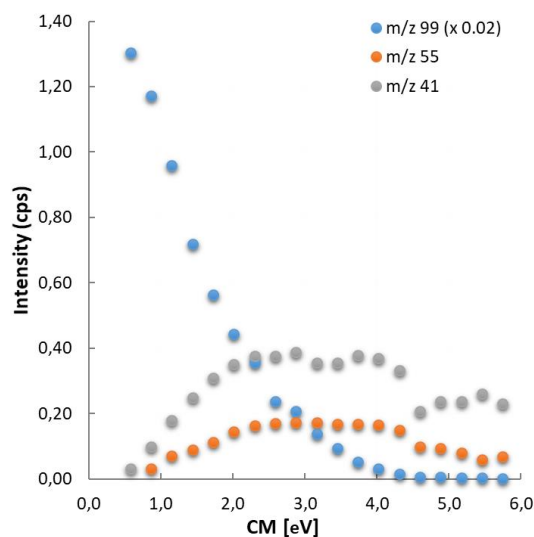
Pressure [mBar]:	Intercept	Slope	X at Y=0 [eV]	[kJ/mol]
$1.06 \cdot 10^{-4}$	-0.40	0.17	1,64	158,03
$2.05 \cdot 10^{-4}$	-0.25	0.20	1,29	124,44
$3.17 \cdot 10^{-4}$	-0.16	0.17	0,97	93,79
$4.14 \cdot 10^{-4}$	-1.19	1.39	0,86	83,23
$5.07 \cdot 10^{-4}$	-1.46	1.98	0,77	74,46
Extrapolated pressure [mBar]:	Intercept	Slope	Y at X=0 [eV]	[kJ/mol]
0	-2149,233	1,773	1,77	171



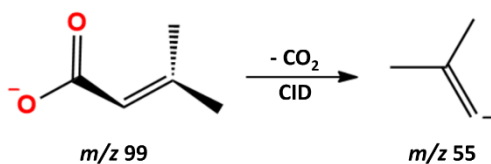
**Figure S35.** Breakdown curves for  $C_5H_7O_2^-$  ( $m/z$  99) recorded at an argon collision gas pressure of  $1.06 \times 10^{-4}$  mbar (left) and  $2.06 \times 10^{-4}$  mbar (right). The intensity of the  $m/z$  99 anion have been multiplied by a scaling factor for readability.



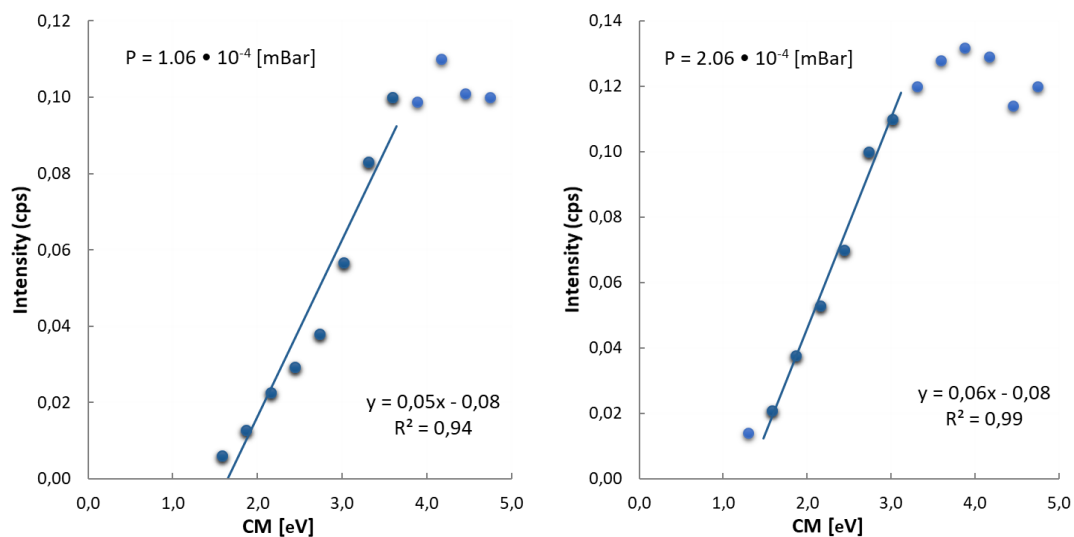
**Figure S36.** Breakdown curves for  $C_5H_7O_2^-$  ( $m/z$  99) recorded at an argon collision gas pressure of  $3.20 \times 10^{-4}$  mbar (left) and  $4.16 \times 10^{-4}$  mbar (right). The intensity of the  $m/z$  99 anion have been multiplied by a scaling factor for readability.



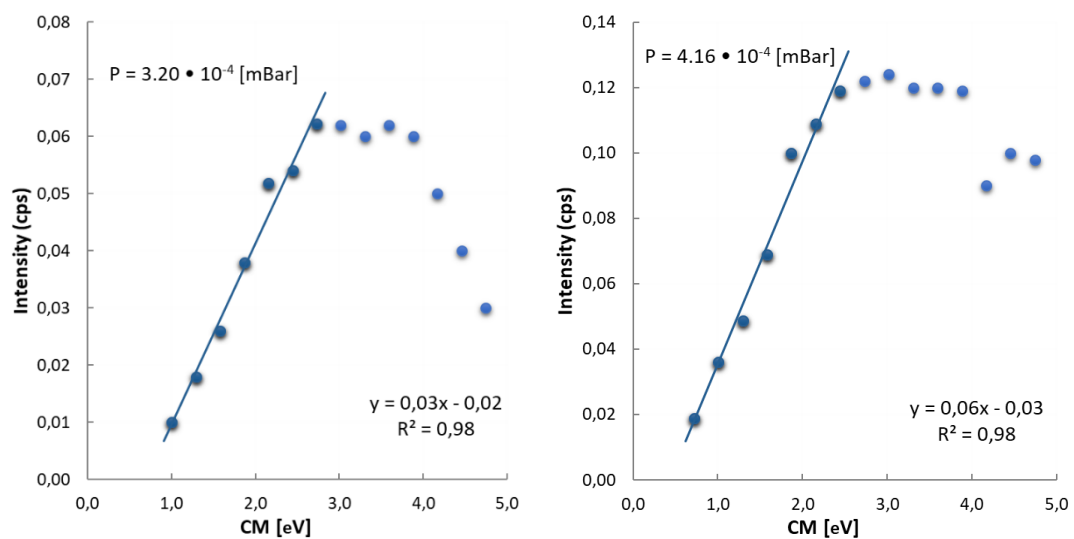
**Figure S37.** Breakdown curve for  $C_5H_7O_2^-$  ( $m/z$  99) recorded at an argon collision gas pressure of  $5.11 \times 10^{-4}$  mbar. The intensity of the  $m/z$  99 anion have been multiplied by a scaling factor for readability.



**Figure S38.** Decarboxylation reaction of  $C_5H_7O_2^-$  ( $m/z$  99) leading to the formation of  $C_4H_7^-$  ( $m/z$  55) an ion.

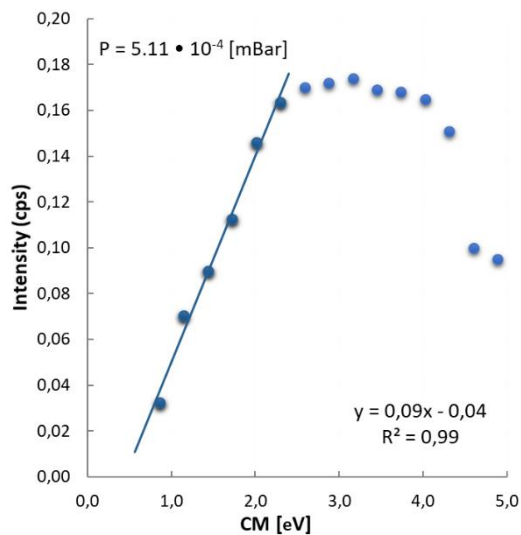


**Figure S39.** Extrapolation procedure for the decarboxylation of  $C_5H_7O_2^-$  ( $m/z$  99) recorded at an argon collision gas pressure of  $1.06 \times 10^{-4}$  mbar (left) and  $2.06 \times 10^{-4}$  mbar (right).

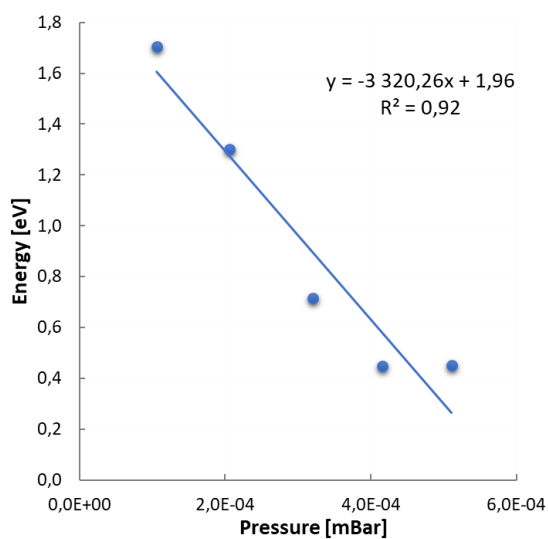


**Figure S40.** Extrapolation procedure for the decarboxylation of  $C_5H_7O_2^-$  ( $m/z$  99) recorded at an argon collision gas pressure of  $3.20 \times 10^{-4}$  mbar (left) and  $4.16 \times 10^{-4}$  mbar (right).





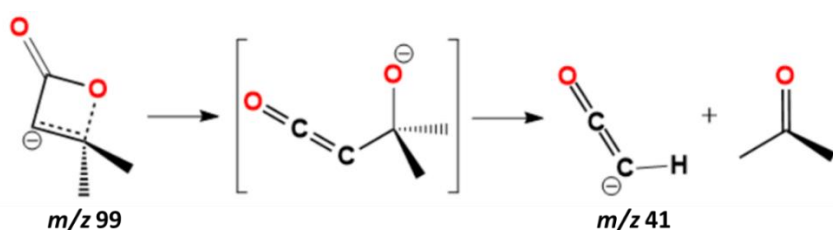
**Figure S41.** Extrapolation procedure for the decarboxylation of  $C_5H_7O_2^-$  ( $m/z$  99) recorded at an argon collision gas pressure of  $5.11 \times 10^{-4}$  mbar.



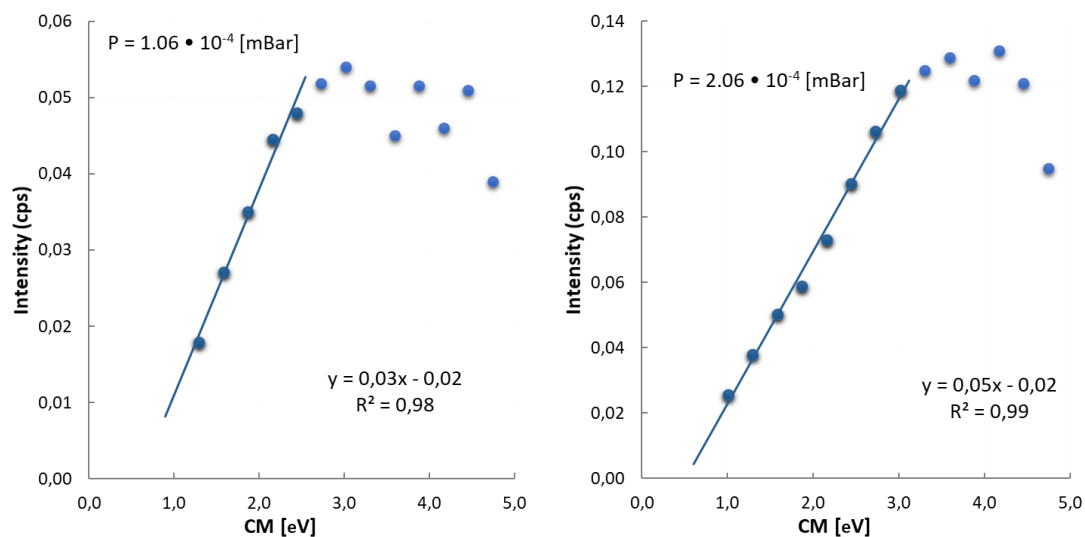
**Figure S42.** Extrapolation procedure for the decarboxylation of  $C_5H_7O_2^-$  ( $m/z$  99) at a gas pressure of 0 mbar.

**Table S6.** Summary of the values from the extrapolation procedure for the decarboxylation of  $C_5H_7O_2^-$  ( $m/z$  99).

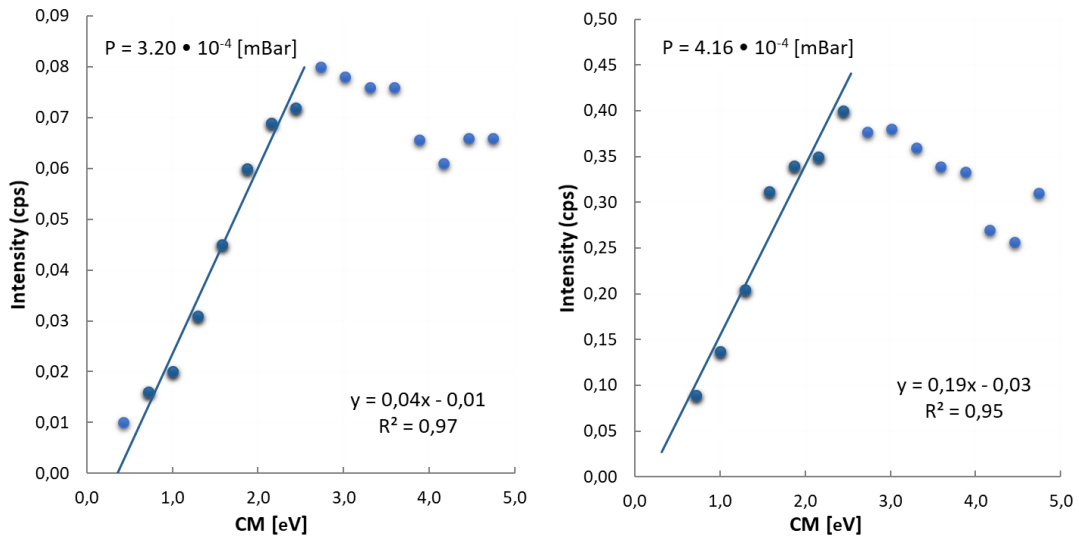
Pressure [mBar]:	Intercept	Slope	X at Y=0 [eV]	[kJ/mol]
$1.06 \cdot 10^{-4}$	-0.27	0.12	1,70	164
$2.06 \cdot 10^{-4}$	-0.83	0.44	1,30	126
$3.20 \cdot 10^{-4}$	-1.41	0.79	0,72	69
$4.16 \cdot 10^{-4}$	-1.12	0.77	0,45	43
$5.11 \cdot 10^{-4}$	-0.75	0.59	0,45	44
Extrapolated pressure [mBar]:	Intercept	Slope	Y at X=0 [eV]	[kJ/mol]
0	-3320,26	1,96	1,96	189



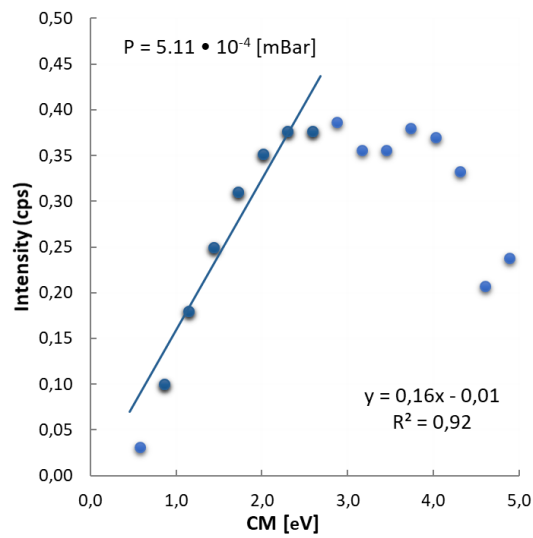
**Figure S43.** Elimination reaction of  $C_3H_6O$  from  $C_5H_7O_2^-$  ( $m/z$  99) leading to the formation of  $C_2HO^-$  ( $m/z$  41) an ion.



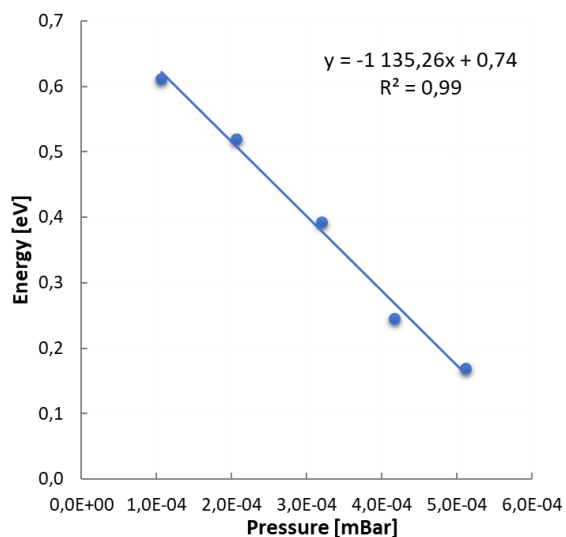
**Figure S44.** Extrapolation procedure for elimination of  $C_3H_6O$  from  $C_5H_7O_2^-$  ( $m/z$  99) recorded at an argon collision gas pressure gas pressure of  $1.06 \times 10^{-4}$  mbar (left) and  $2.06 \times 10^{-4}$  mbar (right).



**Figure S45.** Extrapolation procedure for elimination of  $C_3H_6O$  from  $C_5H_7O_2^-$  ( $m/z$  99) recorded at an argon collision gas pressure gas pressure of  $3.20 \times 10^{-4}$  mbar (left) and  $4.16 \times 10^{-4}$  mbar (right).



**Figure S46.** Extrapolation procedure for elimination of  $C_3H_6O$  from  $C_5H_7O_2^-$  ( $m/z$  99) recorded at an argon collision gas pressure gas pressure of  $5.11 \times 10^{-4}$  mbar.

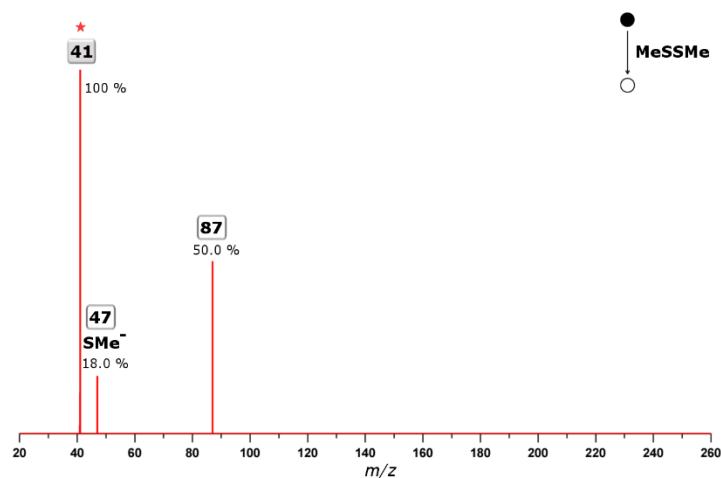


**Figure S47.** Extrapolation procedure for elimination of  $C_3H_6O$  from  $C_5H_7O_2^-$  ( $m/z$  99) at a gas pressure of 0 mbar.

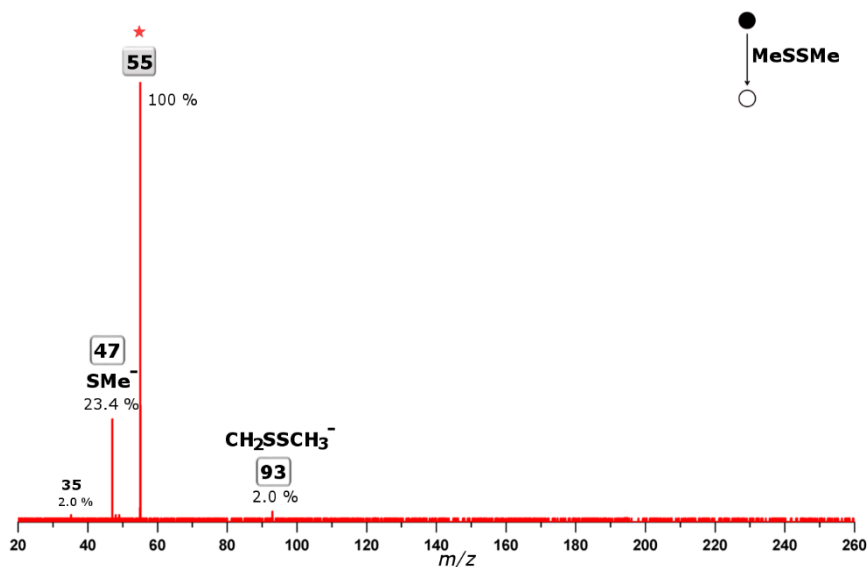
**Table S7.** Summary of the values from the extrapolation procedure for elimination of  $C_3H_6O$  from  $C_5H_7O_2^-$  ( $m/z$  99).

Pressure [mBar]:	Intercept	Slope	X at Y=0 [eV]	[kJ/mol]
$1,06 \cdot 10^{-4}$	-0,02	0,03	0,61	59
$2,06 \cdot 10^{-4}$	-0,02	0,05	0,52	50
$3,20 \cdot 10^{-4}$	-0,01	0,04	0,39	38
$4,16 \cdot 10^{-4}$	-0,03	0,19	0,25	24
$5,11 \cdot 10^{-4}$	-0,01	0,16	0,17	16
Extrapolated pressure [mBar]:	Intercept	Slope	Y at X=0 [eV]	[kJ/mol]
0	-1135,26	0,74	0,74	72

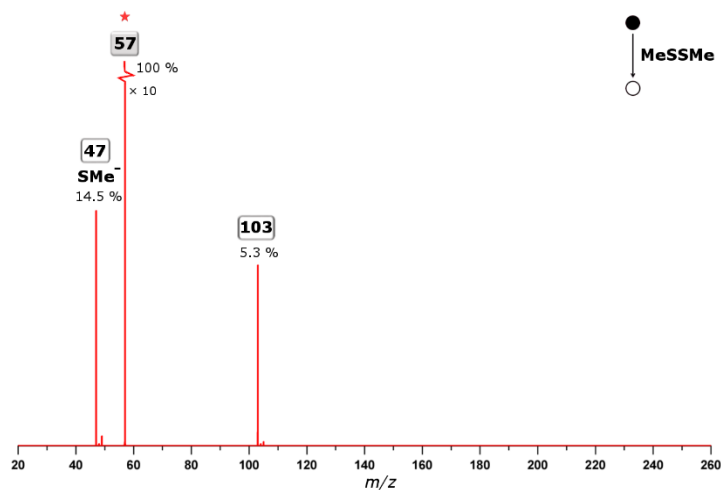
### S3.3 Reactions with dimethyl disulfide ( $\text{CH}_3\text{SSCH}_3$ )



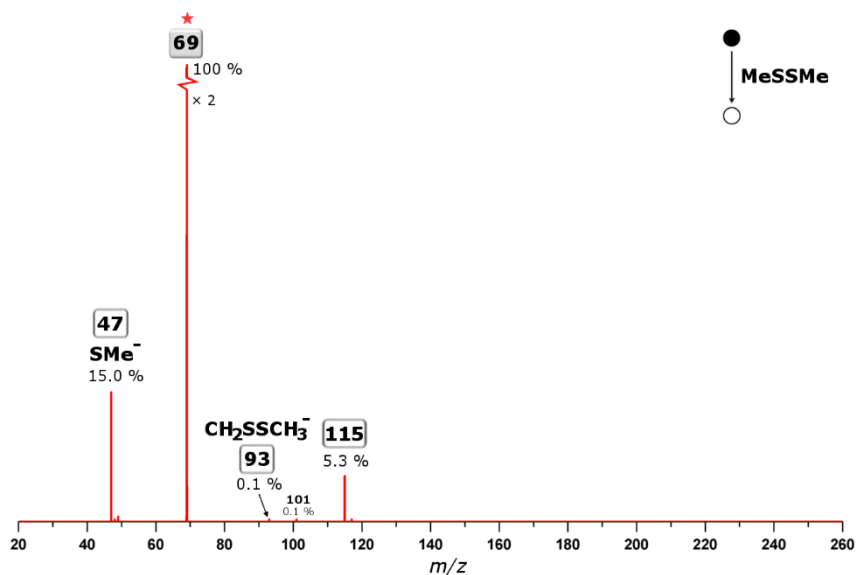
**Figure S48.** Mass spectrum of the reaction between dimethyl disulfide ( $\text{CH}_3\text{SSCH}_3$ ) and  $\text{C}_2\text{HO}^-$  ( $m/z$  41) recorded with a ToF voltage of 3kV, taken at a collision energy of 0.7 eV (CM) with reagent vapors at nominal pressures of  $3.43 \times 10^{-4}$  mBar.



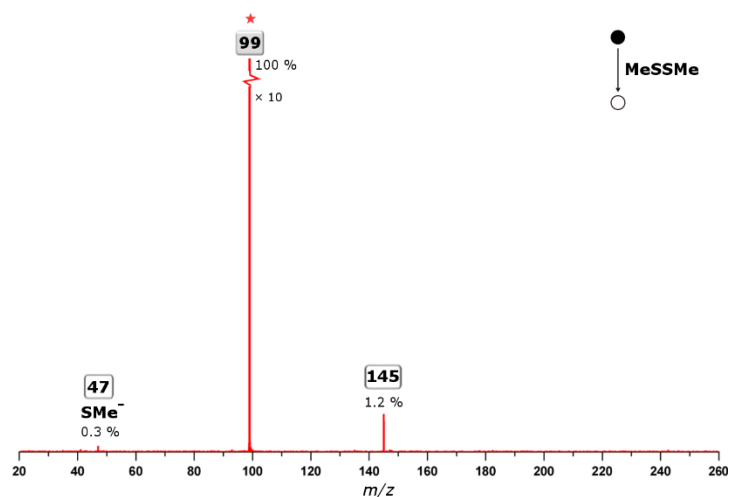
**Figure S49.** Mass spectrum of the reaction between dimethyl disulfide ( $\text{CH}_3\text{SSCH}_3$ ) and  $\text{C}_4\text{H}_7^-$  ( $m/z$  55) recorded with a ToF voltage of 3kV, taken at a collision energy of 0.63 eV (CM) with reagent vapors at nominal pressures of  $3.43 \times 10^{-4}$  mBar.



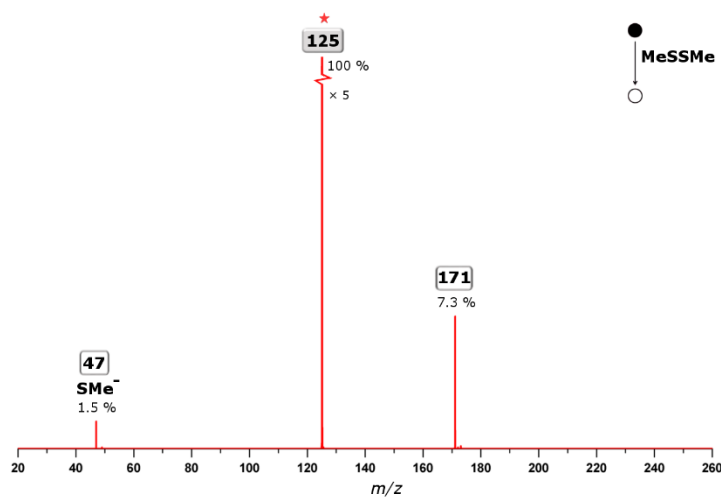
**Figure S50.** Mass spectrum of the reaction between dimethyl disulfide ( $\text{CH}_3\text{SSCH}_3$ ) and  $\text{C}_3\text{H}_5\text{O}^-$  ( $m/z$  57) recorded with a ToF voltage of 3kV, taken at a collision energy of 0.62 eV (CM) with reagent vapors at nominal pressures of  $3.43 \times 10^{-4}$  mBar.



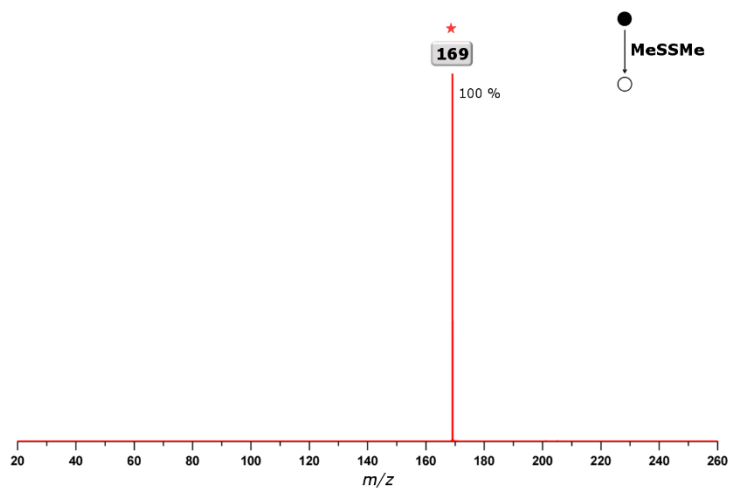
**Figure S51.** Mass spectrum of the reaction between dimethyl disulfide ( $\text{CH}_3\text{SSCH}_3$ ) and  $\text{C}_4\text{H}_5\text{O}^-$  ( $m/z$  69B) recorded with a ToF voltage of 3kV, taken at a collision energy of 0.62 eV (CM) with reagent vapors at nominal pressures of  $3.43 \times 10^{-4}$  mBar.



**Figure S52.** Mass spectrum of the reaction between dimethyl disulfide ( $\text{CH}_3\text{SSCH}_3$ ) and  $\text{C}_5\text{H}_6\text{O}_2^-$  ( $m/z$  99) recorded with a ToF voltage of 3kV, taken at a collision energy of 0.49 eV (CM) with reagent vapors at nominal pressures of  $3.43 \times 10^{-4}$  mBar.



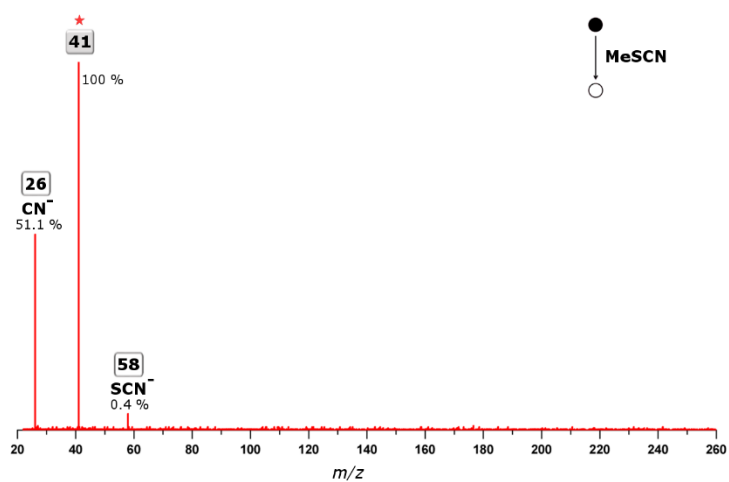
**Figure S53.** Mass spectrum of the reaction between dimethyl disulfide ( $\text{CH}_3\text{SSCH}_3$ ) and  $\text{C}_8\text{H}_{13}\text{O}^-$  ( $m/z$  125) recorded with a ToF voltage of 3kV, taken at a collision energy of 0.43 eV (CM) with reagent vapors at nominal pressures of  $3.43 \times 10^{-4}$  mBar.



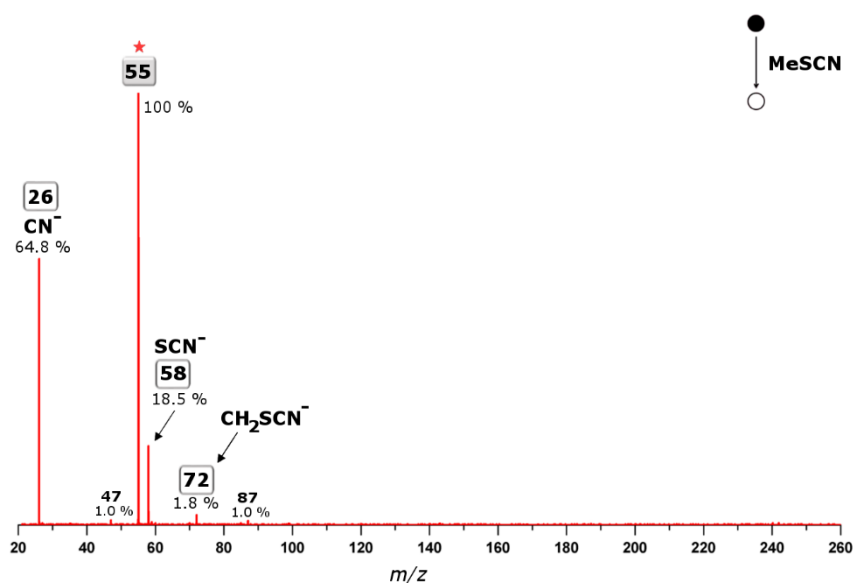
**Figure S54.** Mass spectrum of the reaction between dimethyl disulfide ( $\text{CH}_3\text{SSCH}_3$ ) and  $\text{C}_9\text{H}_{13}\text{O}_3^-$  ( $m/z$  169) recorded with a ToF voltage of 3kV, taken at a collision energy of 0.36 eV (CM) with reagent vapors at nominal pressures of  $3.43 \times 10^{-4}$  mBar. Please note the lack of the product in this reaction.



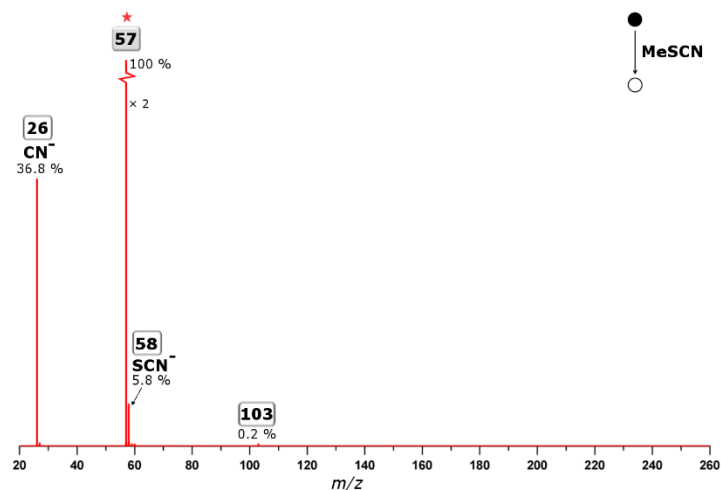
### S3.4 Reactions with methyl thiocyanate ( $\text{CH}_3\text{SCN}$ )



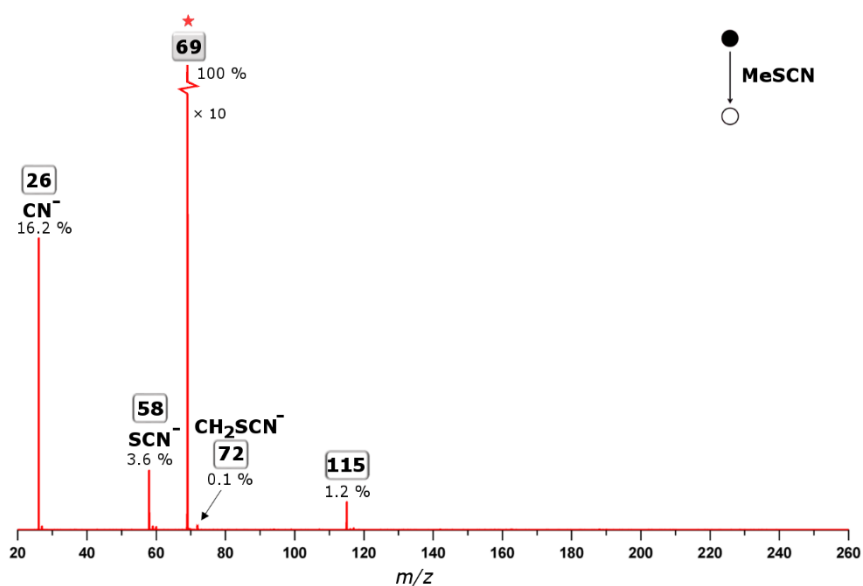
**Figure S55.** Mass spectrum of the reaction between methyl thiocyanate ( $\text{CH}_3\text{SCN}$ ) and  $\text{C}_2\text{HO}^-$  ( $m/z$  41) recorded with a ToF voltage of 3kV, taken at a collision energy of 0.64 eV (CM) with reagent vapors at nominal pressures of  $1.53 \times 10^{-4}$  mBar.



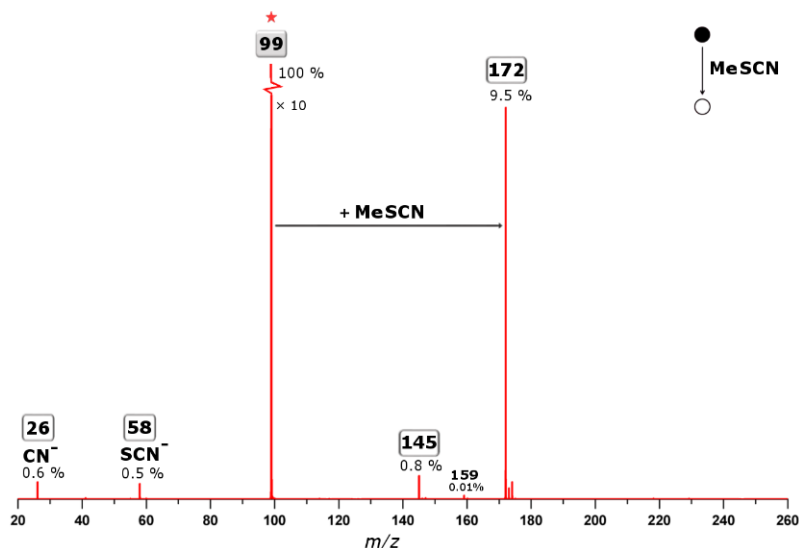
**Figure S56.** Mass spectrum of the reaction between methyl thiocyanate ( $\text{CH}_3\text{SCN}$ ) and  $\text{C}_4\text{H}_7^-$  ( $m/z$  55) recorded with a ToF voltage of 3kV, taken at a collision energy of 0.57 eV (CM) with reagent vapors at nominal pressures of  $1.53 \times 10^{-4}$  mBar.



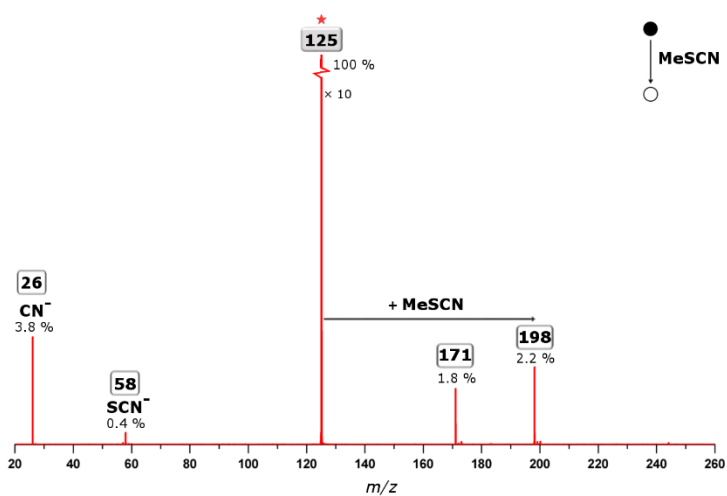
**Figure S57.** Mass spectrum of the reaction between methyl thiocyanate (CH<sub>3</sub>SCN) and C<sub>3</sub>H<sub>5</sub>O<sup>-</sup> (*m/z* 57) recorded with a ToF voltage of 3kV, taken at a collision energy of 0.56 eV (CM) with reagent vapors at nominal pressures of 1.53×10<sup>-4</sup> mBar.



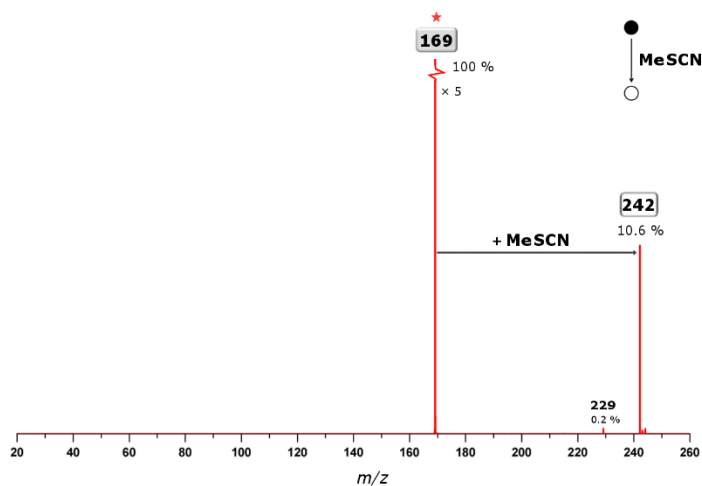
**Figure S58.** Mass spectrum of the reaction between methyl thiocyanate (CH<sub>3</sub>SCN) and C<sub>4</sub>H<sub>5</sub>O<sup>-</sup> (*m/z* 69B) recorded with a ToF voltage of 3kV, taken at a collision energy of 0.51 eV (CM) with reagent vapors at nominal pressures of 1.53×10<sup>-4</sup> mBar.



**Figure S59.** Mass spectrum of the reaction between methyl thiocyanate (CH<sub>3</sub>SCN) and C<sub>5</sub>H<sub>6</sub>O<sub>2</sub><sup>-</sup> (*m/z* 99) recorded with a ToF voltage of 3kV, taken at a collision energy of 0.42 eV (CM) with reagent vapors at nominal pressures of 1.53×10<sup>-4</sup> mBar.



**Figure S60.** Mass spectrum of the reaction between methyl thiocyanate (CH<sub>3</sub>SCN) and C<sub>8</sub>H<sub>13</sub>O<sup>-</sup> (*m/z* 125) recorded with a ToF voltage of 3kV, taken at a collision energy of 0.37 eV (CM) with reagent vapors at nominal pressures of 1.53×10<sup>-4</sup> mBar.



**Figure S61.** Mass spectrum of the reaction between methyl thiocyanate ( $\text{CH}_3\text{SCN}$ ) and  $\text{C}_9\text{H}_{13}\text{O}_3^-$  ( $m/z$  169) recorded with a ToF voltage of 3kV, taken at a collision energy of 0.30 eV (CM) with reagent vapors at nominal pressures of  $1.53 \times 10^{-4}$  mBar.

#### S4. Computed data for norpinonic acid

The mathematical models used for calculations in Gaussian 09 are called FEM (Final Element Method) and Symplex. These methods cut an N-dimensional space into small subsystems that can be described by N linear equations. The equations can be solved when the solution of at least one of them is known. Therefore, the programme guesses the solution of one and then recursively solves all the others. Once all solutions are obtained, the initial guess can be modified and the calculation repeated. This process is performed until the new solution gives the same result as the previous iteration, which is called convergence. Among other things, the chosen basis set influences the quality of the guesswork, while the theoretical model influences the type of calculations the matrixes will undergo (*Tomberg, A., 2013, Gaussian 09W Tutorial, an Introduction to Computational Chemistry Using G09W and Avogadro Software*).

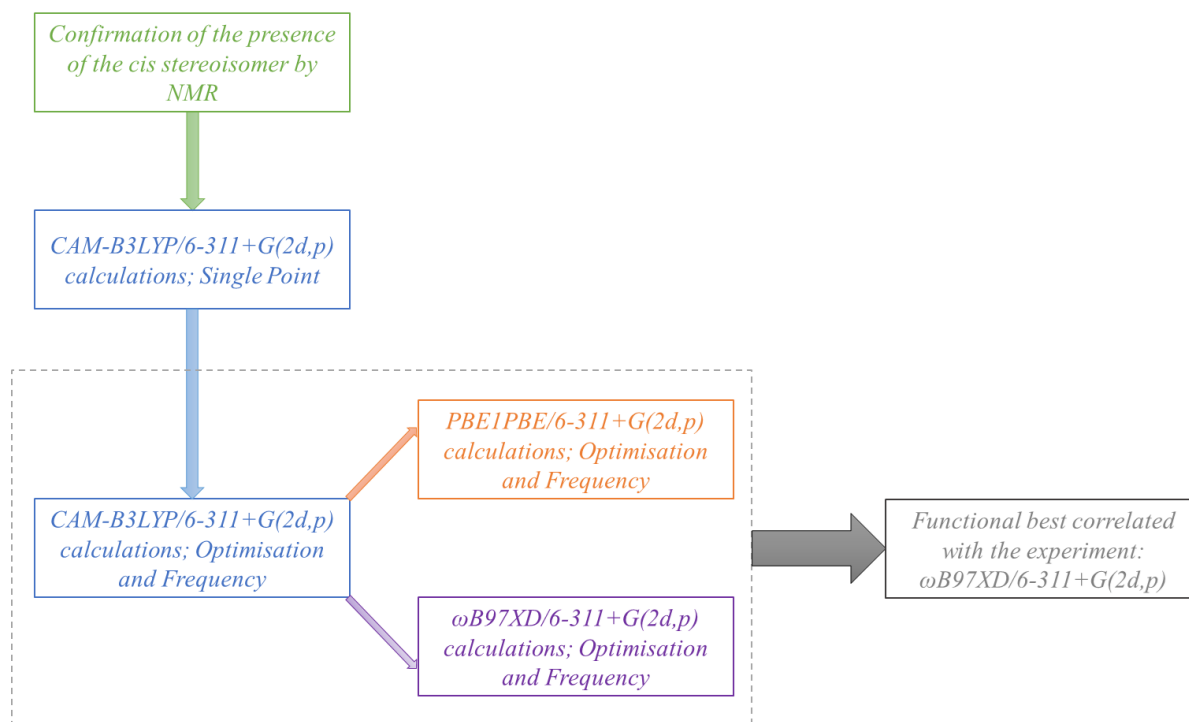
Because density functional methods are best regarded as approximations to the resolving for the electron density in a molecular system and the exact exchange correlation function is not known, comparative analysis of different DFT models in terms of performance is an important part of computational chemistry. Quite often, one particular set of functionals and functional bases works very well for a particular application, so for the study of structures not precisely described by quantum chemistry, calculations should be carried out with various functionals to get closer to experimental results.

CAM-B3LYP (Becke Three-Parameter Hybrid Functionals) long-range-corrected version of B3LYP, uses the non-local correlation provided by the LYP expression, which contains both local and non-local terms (*Axel D. Becke; Density-functional thermochemistry. III. The role of exact exchange. J. Chem. Phys. 1 April 1993; 98 (7): 5648–5652. <https://doi.org/10.1063/1.464913>; T. Yanai, D. Tew, and N. Handy, “A new hybrid exchange-correlation functional using the Coulomb-attenuating method (CAM-B3LYP),” *Chem. Phys. Lett.*, 393 (2004) 51-57. <https://doi.org/10.1016/j.cplett.2004.06.011>).*

$\omega$ B97XD is the latest functional from Chai and Head-Gordon includes long-range exact exchange, a small fraction of short-range exact exchange, a modified exchange density functional for short-range interaction, the correlation density functional and empirical dispersion corrections. (*J.-D. Chai and M. Head-Gordon, “Long-range corrected hybrid density functionals with damped atom-atom dispersion corrections,” Phys. Chem. Chem. Phys.*, 10 (2008) 6615-20. <https://doi.org/10.1039/B810189B>).

PBE1PBE Perdew, Burke and Ernzerhof’s 1996 basic function, transformed into a hybrid function by Adamo. This function uses 25% exact exchange and 75% DFT exchange (*J. P. Perdew, K. Burke, and*

M. Ernzerhof, "Generalized gradient approximation made simple," *Phys. Rev. Lett.*, 77 (1996) 3865-68. <https://doi.org/10.1103/PhysRevLett.77.3865>; C. Adamo and V. Barone, "Toward reliable density functional methods without adjustable parameters: The PBE0 model," *J. Chem. Phys.*, 110 (1999) 6158-69. DOI: 10.1063/1.478522).



**Figure S62.** A schematic representation of the strategy used to determine the functional.

#### S4.1 Computed fragmentation pathways

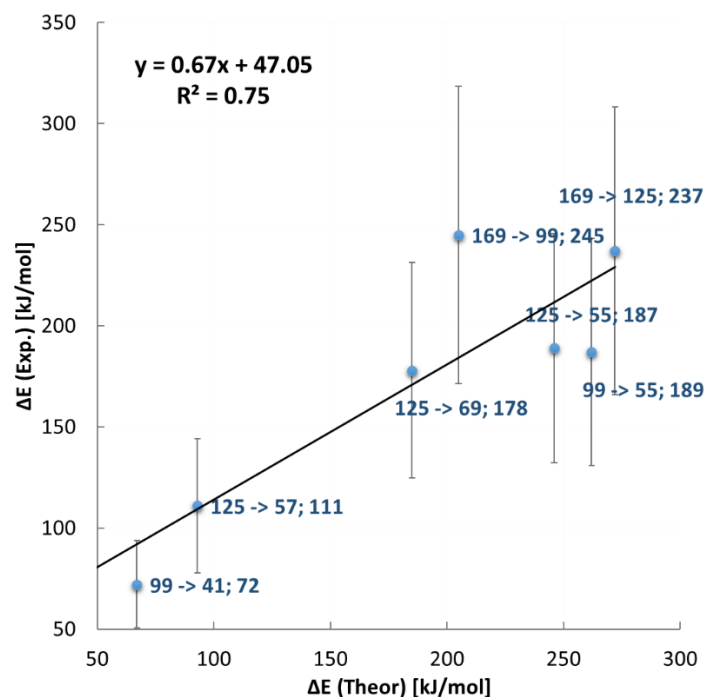
**Table S8.** Calculated electron energy values in kJ/mol for fragmentation reaction of *m/z* 169 anion.

Structure	Theoretical method		
	CAM-B3LYP/ 6-311+G(2d,p)	PBE1PBE/ 6-311+G(2d,p)	ωB97XD/ 6-311+G(2d,p)
TS_1	307	300	322
TS_2	266	287	301
TS_3	247	241	274
TS_4	176	172	197
TS_5	279	290	304
TS_6	273	294	304
TS_7	256	278	289
TS_8	170	181	201
TS_9	177	164	187
TS_10	178	171	198
TS_11	264	288	304
TS_12	302	318	341
TS_13	275	285	304
CX_99	158	169	187
CX_125B	254	280	288

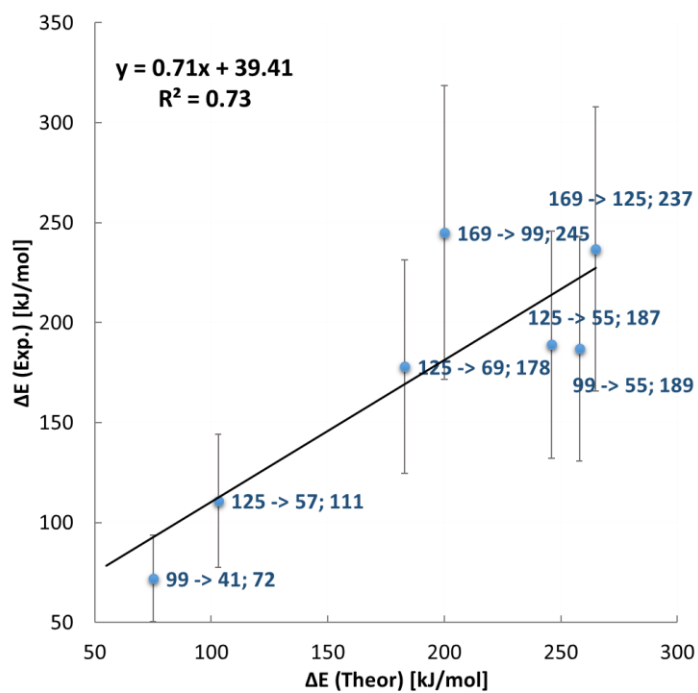
<b>IC_125B</b>	119	103	143
<b>IC_125C</b>	100	96	120
<b>IC_169</b>	82	79	90

**Table S9.** Calculated electron energy values and experimental values in kJ/mol for fragmentation reaction of *m/z* 169 anion.

	Theoretical method			Experimental energy
	CAM-B3LYP/ 6-311+G(2d,p)	PBE1PBE/ 6-311+G(2d,p)	ωB97XD/ 6-311+G(2d,p)	
Fragmentation reaction	169 → 99	200	205	245
	99 → 55	246	246	189
	99 → 41	75	67	72
	169 → 125	265	272	237
	125 → 69	183	185	178
	125 → 57	103	93	111
	125 → 55	258	262	187



**Figure S63.** A correlation between the theoretical and the experimental results obtained with PBE1PBE/6-311+g(2d,p) level of theory. Experimental threshold energies as a function of the theoretical threshold energies (kJ/mol).



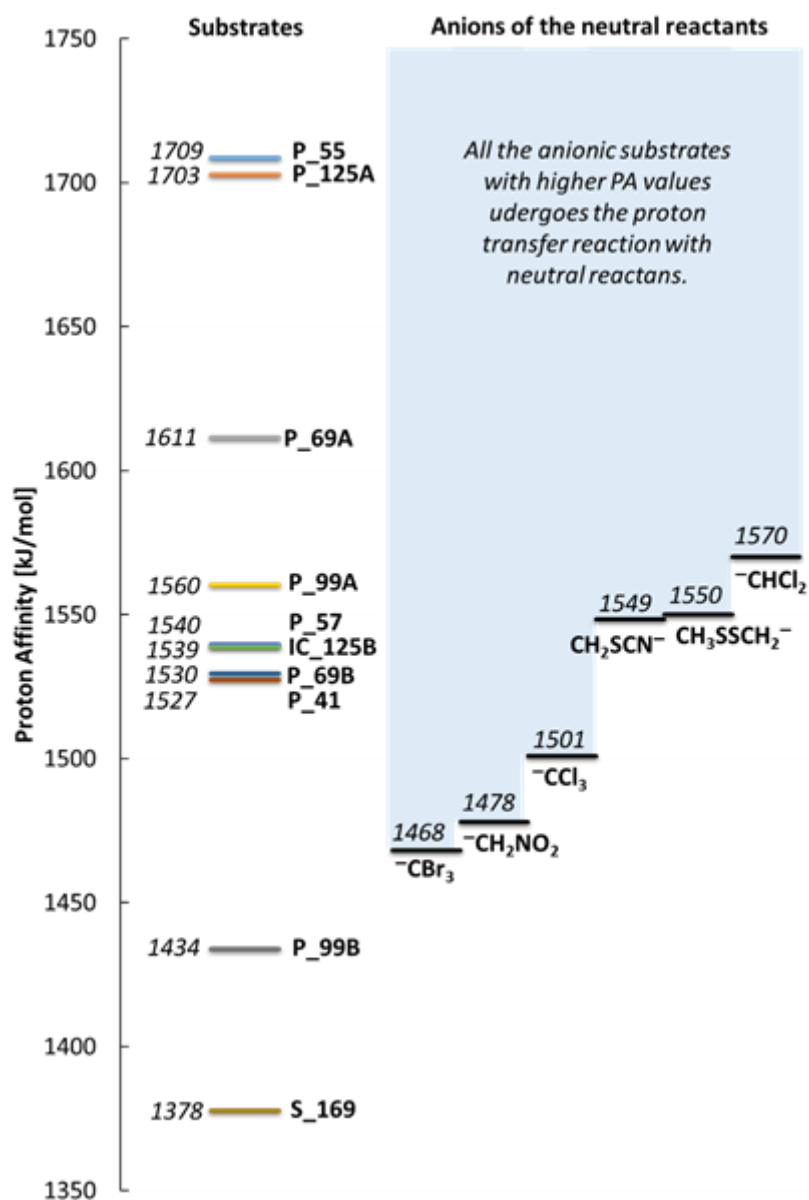
**Figure S64.** A correlation between the theoretical and the experimental results obtained with CAM-B3LYP/6-311+g(2d,p) level of theory. Experimental threshold energies as a function of the theoretical threshold energies (kJ/mol).

## S4.2 Proton affinity calculation

**Table S10.** Calculated proton affinity values in kJ/mol for anions and reagents.

Structure	Theoretical method		
	CAM-B3LYP/ 6-311+G(2d,p)	PBE1PBE/ 6-311+G(2d,p)	$\omega$ / 6-311+G(2d,p)
<b>Anions</b>			
P_41	1527	1535	1542
P_55	1709	1714	1719
P_57	1540	1545	1555
P_69A	1611	1618	1621
P_69B	1530	1532	1542
P_99A	1560	1545	1557
P_99B	1434	1445	1449
IC_125A	1539	1540	1558
P_125	1703	1701	1719
S_169	1378	1422	1430
<b>Neutral reagents</b>			
CH <sub>2</sub> Cl <sub>2</sub>	1569	1578	1581
CHCl <sub>3</sub>	1496	1501	1505
CHBr <sub>3</sub>	1467	1465	1479
CH <sub>3</sub> SSCH <sub>3</sub>	1525	1550	1565
CH <sub>3</sub> SCN	1548	1551	1561
CH <sub>3</sub> NO <sub>2</sub>	1478	1478	1497



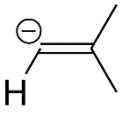

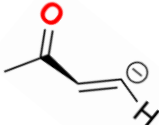
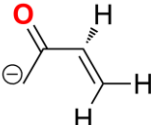


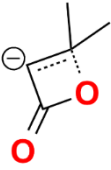
**Figure S65.** Proton affinity of the anionic fragment structures, compared with proton affinities of appropriate deprotonated reagents used in gas phase reactions obtained by theoretical calculation.

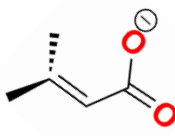
## S5. Geometries

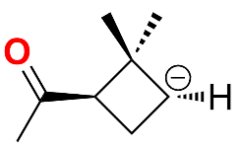
### S5.1 Anions/neutral structures/fragmentation structures

Anions					
	P_41	C	1.24478300	-0.09751100	-0.00000200
		H	2.19261000	0.38814900	0.00000800
		C	0.00642700	0.01585900	0.00000500
		O	-1.21248300	0.01272000	-0.00000300
		O	-1.21248300	0.01272000	-0.00000300
P_55	C	0.24376500	1.55782800	0.00003400	

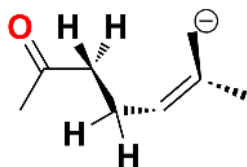
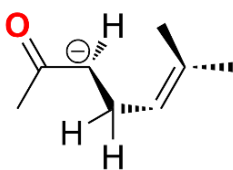
	C	0.02515300	0.22458800	-0.00003900
	H	1.34443700	1.73932000	-0.00006500
	C	1.08186300	-0.87834400	0.00000600
	H	0.99449700	-1.53854700	0.88110600
	H	0.99333700	-1.53975800	-0.88009400
	H	2.09050900	-0.45194100	-0.00096900
	C	-1.38018400	-0.35289400	0.00002400
	H	-1.57067800	-0.98994900	-0.88219300
	H	-1.57041200	-0.99079200	0.88165500
	H	-2.10527300	0.46460100	0.00041700
<b>P_57</b>	C	-1.19319300	-0.77858400	0.00004900
	H	-1.03374800	-1.85325400	0.00017500
	H	-2.21505300	-0.40827000	-0.00048800
	C	-0.13229300	0.10904100	0.00004700
	O	-0.18258700	1.38160900	0.00001000
	C	1.28538000	-0.48928900	-0.00003900
	H	1.82736500	-0.12622300	-0.88144300
	H	1.82726800	-0.12644800	0.88151700
	H	1.29549400	-1.58568100	-0.00019200
<b>P_69A</b>	C	1.96937200	0.35995400	0.00002100
	H	2.93105200	-0.19685000	0.00000500
	C	-0.45944500	-0.19496800	0.00000000
	O	-1.36657600	-1.04027100	0.00001300
	C	-0.83760000	1.27643800	-0.00000500
	H	0.05850300	1.89847100	0.00001300
	H	-1.45304200	1.48136600	-0.88279100
	H	-1.45308000	1.48137800	0.88275000
	C	0.95550200	-0.55665300	-0.00002400
	H	1.08220100	-1.65082600	-0.00003100
	<b>P_69B</b>	C	0.61300900	-0.04131300
	O	1.47823900	-0.94403900	-0.01960700
	C	0.81063800	1.30716300	0.01133600
	H	1.82155400	1.70395000	-0.00709300
	H	-0.02133500	2.01009600	0.03836000
	C	-0.79635200	-0.56439300	0.02442800
	H	-0.80960000	-1.65218700	0.07270600
	C	-1.94913300	0.09050500	-0.01790800
	H	-2.89297300	-0.43672200	0.00148000
	H	-1.99253100	1.17540400	-0.07486700
	<b>P_99A</b>	C	0.38682800	-0.00073600
C	-0.70638800	-0.00012400	0.07593600	
H	0.42126000	0.00352900	2.18965400	

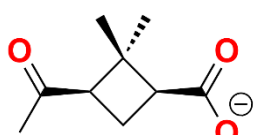
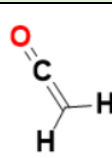
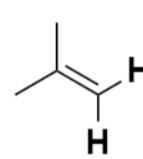
	C	-1.56291600	-1.25796600	-0.04602200
	H	-2.24665600	-1.34056600	0.80849500
	H	-2.16359700	-1.24935100	-0.96596100
	H	-0.91279200	-2.13585300	-0.05458700
	C	-1.56053000	1.25950500	-0.04690800
	H	-2.16049900	1.25158200	-0.96729700
	H	-2.24478700	1.34355900	0.80706400
	H	-0.90862500	2.13603400	-0.05530000
	C	1.33062400	-0.00051800	0.09372100
	O	0.32253900	-0.00123100	-0.96462700
	O	2.53871000	-0.00000700	-0.14567400


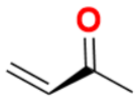
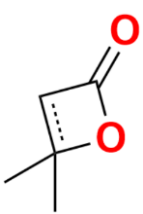
P_99B				
	C	-0.08253100	-0.68656500	-0.00050700
	C	1.12021300	-0.09474800	0.00014100
	H	-0.11111300	-1.77612600	-0.00128800
	C	2.38525100	-0.91919700	-0.00058300
	H	2.17073300	-1.99191300	-0.00192200
	H	3.00369000	-0.69149400	-0.88016300
	H	3.00317600	-0.69358700	0.87988600
	C	1.34089100	1.39419500	0.00128500
	H	1.92554300	1.69370300	0.88326300
	H	1.92584800	1.69509700	-0.88007100
	H	0.37736500	1.90672900	0.00178400
	C	-1.48365000	-0.06824100	-0.00022100
	O	-1.59094900	1.18592700	-0.00178600
	O	-2.40608600	-0.92281100	0.00151300

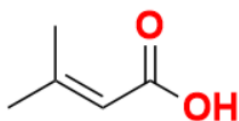
P_125				
	C	-1.72308000	1.27461800	-0.07010100
	C	-0.28699100	1.73925700	-0.48081400
	C	0.13927700	0.27251500	-0.68314600
	C	-1.14439000	-0.15552700	0.14258300
	H	-2.34487800	1.25154300	-0.98280900
	H	0.26681200	2.23598200	0.33085200
	H	-0.17867400	2.37991800	-1.37053700
	H	-0.03582200	-0.03900200	-1.71828500
	C	-1.91696400	-1.33237500	-0.44144100
	H	-1.39773700	-2.29057700	-0.29786300
	H	-2.89844200	-1.39806000	0.04665400
	H	-2.08782400	-1.19137400	-1.51426400
	C	-0.84646900	-0.44511300	1.61887000
	H	-1.79909100	-0.57786800	2.14351400
	H	-0.23090800	-1.34733400	1.76535000
	H	-0.35208900	0.40747000	2.09280400
	C	1.48021700	-0.28821500	-0.29767800
	O	1.86613800	-1.37887100	-0.70637600
	C	2.36516900	0.50894700	0.63901000

	H	3.25780000	-0.07026100	0.88373000
	H	2.64971800	1.45139800	0.15887100
	H	1.82141900	0.77449900	1.54929300
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<b>IC_125A</b>	C	-0.99836500	-0.89971200	-0.10591900
	C	0.15277700	-0.80105900	-1.07376800
	C	1.50823200	-0.95947500	-0.42720600
	C	-2.03495400	-0.08325700	0.07368200
	H	-0.92010300	-1.75238600	0.56634100
	H	0.08105900	0.12711400	-1.64643800
	H	0.02464100	-1.60177300	-1.81609100
	H	1.91559400	-1.96616100	-0.38747900
	C	-3.04633900	-0.33321100	1.15938600
	H	-3.07373800	0.49861900	1.87315700
	H	-4.05948600	-0.42659600	0.75024400
	H	-2.81869900	-1.24410600	1.71415100
	C	-2.29945200	1.15644000	-0.73360600
	H	-3.26612800	1.08911000	-1.24654700
	H	-2.35025500	2.03812100	-0.08487100
	H	-1.53398500	1.33686800	-1.48456700
	C	2.26804100	0.02559500	0.16475300
	O	3.39666700	-0.13683200	0.72132100
	C	1.75830000	1.47177700	0.17517900
	H	1.68396600	1.80247700	1.21429200
	H	2.50381600	2.10659000	-0.31146900
	H	0.79054400	1.62418900	-0.30630400
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<b>IC_125B</b>	C	-1.36203200	1.03394700	-0.13722400
	C	0.07145800	1.42244000	-0.45248000
	C	1.08119700	0.39391800	-0.85889000
	C	-1.96784800	-0.13954900	0.03304400
	H	-2.00961700	1.90826400	-0.03088100
	H	0.42526100	1.99160700	0.41546200
	H	-0.01946100	2.18988600	-1.24044900
	H	0.97359000	-0.01487000	-1.86138300
	C	-3.44260000	-0.19048500	0.34136000
	H	-3.98254000	-0.76471800	-0.42041800
	H	-3.62692700	-0.69503600	1.29670800
	H	-3.88381500	0.80651700	0.39220600
	C	-1.32018200	-1.49314500	-0.04282600
	H	-1.44163100	-2.02176500	0.91001400
	H	-1.81366600	-2.10811900	-0.80497900
	H	-0.26063100	-1.42007900	-0.27444400
	C	2.10492700	-0.14296700	-0.11257000
	O	2.90357700	-1.05183300	-0.50313100
	C	2.35046000	0.32952100	1.32469000



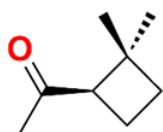
	H	2.20369600	-0.51984900	1.99793400
	H	3.39799000	0.62873100	1.41177900
	H	1.71685200	1.15201000	1.66288200
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<b>S_169</b>	C	1.12197900	0.02007100	-0.78957800
	C	0.00096400	-0.95644900	-1.17254900
	C	-1.04272000	0.10561000	-0.77881500
	C	0.08658300	0.86845900	0.01460100
	H	1.47635400	0.59725200	-1.64642100
	H	0.03819900	-1.81573900	-0.50454900
	H	-0.03346800	-1.31594400	-2.20253600
	H	-1.32732600	0.72502600	-1.63190400
	C	0.12700400	2.36845500	-0.20378800
	H	-0.71673800	2.86413500	0.28604400
	H	1.05741700	2.76996500	0.20355400
	H	0.09210100	2.61214100	-1.26833200
	C	0.11657100	0.55571800	1.50504100
	H	1.01913300	0.98785700	1.93879500
	H	-0.75844000	0.97157100	2.01408700
	H	0.16585800	-0.51666600	1.69526400
	C	-2.31733700	-0.23079100	-0.06047600
	C	2.34901800	-0.49417300	-0.00982100
	O	3.16352400	0.39902800	0.30975400
	O	-3.21895000	0.57955300	0.02216000
	O	2.38325700	-1.71580100	0.24526000
	C	-2.46727800	-1.59713600	0.56026600
	H	-3.42132900	-1.65730900	1.08016200
	H	-2.41668500	-2.36132300	-0.21904100
	H	-1.64642000	-1.80178200	1.24820200
<hr/>				
<b>Neutral fragments</b>				
<b>P_41N</b>	C	0.00642700	0.01585900	0.00000500
	O	-1.21236721	0.01398374	0.01699319
	C	1.20262468	-0.09365146	-0.00000176
	H	1.64977607	0.87843683	-0.00001276
	H	1.51182613	-0.62846311	0.87365204
<hr/>				
<b>P_55N</b>	C	0.02468900	0.22379100	-0.00001300
	C	1.07677400	-0.87793800	-0.00006700
	H	0.98789200	-1.53687900	0.87836400
	H	0.98771800	-1.53696000	-0.87841900
	H	2.08174300	-0.45147600	-0.00018600
	C	-1.38141400	-0.34424400	0.00015700
	H	-1.57346400	-0.97910700	-0.87960100
	H	-1.57329200	-0.97900600	0.88002400
	H	-2.10062500	0.47440300	0.00018100

	C	0.25549253	1.55919235	-0.00010088	
	H	-0.18247085	1.99689927	0.87253921	
	H	1.31030199	1.73883973	0.00184921	
<b>P_57N</b>	C	-0.00530100	0.17565700	-0.00006300	
	O	0.01888100	1.37695300	0.00001200	
	C	1.27887300	-0.62246700	-0.00036700	
	H	1.87467900	-0.33223100	-0.86655200	
	H	1.85406700	-0.36289300	0.88951900	
	H	1.11120700	-1.69818900	-0.01961800	
	C	-1.29960500	-0.59186100	0.00032000	
	H	-1.35224900	-1.22776100	-0.88711200	
	H	-1.33864300	-1.25556800	0.86767200	
	H	-2.14390900	0.09304200	0.01665800	
	<b>P_69N</b>	C	0.60330900	-0.05462800	0.03333200
	O	1.38490200	-1.03163500	-0.19398200	
	C	-0.82685400	-0.46160200	0.27541500	
	H	-0.91419000	-1.38893900	0.83889300	
	C	-1.92752100	0.11394900	-0.19480200	
	H	-2.91352300	-0.29634300	0.00370000	
	H	-1.87418300	1.01203700	-0.79992900	
	C	0.90405158	1.26541297	0.09343831	
	H	0.59370825	1.74351661	-0.81210237	
	H	0.39570955	1.70959274	0.92361462	
	H	1.96018860	1.38536926	0.21625968	
<b>P_99AN</b>	C	-0.70478600	-0.00001500	0.07386900	
	C	-1.55944900	-1.25461900	-0.04357300	
	H	-2.23953800	-1.33612400	0.81060800	
	H	-2.16193900	-1.24624100	-0.95923400	
	H	-0.91088900	-2.13007800	-0.05370500	
	C	-1.55756500	1.25575900	-0.04478300	
	H	-2.16041800	1.24722700	-0.96020300	
	H	-2.23723500	1.33934100	0.80953700	
	H	-0.90770200	2.13024400	-0.05624000	
	C	1.33087000	-0.00033100	0.09454500	
	O	0.32146700	-0.00133100	-0.96605500	
	O	2.53014900	-0.00003000	-0.14320000	
	C	0.38769400	-0.00028800	1.10368900	
	H	0.43288296	-0.87400880	1.71970030	
	H	0.43273601	0.87329384	1.71990810	
	<b>P_99BN</b>	C	-0.07963900	-0.68279700	-0.00052400
		C	1.11525000	-0.09423500	-0.00003900
H		-0.10713900	-1.76979100	-0.00100900	



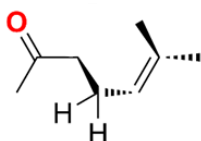
C	2.37750300	-0.91548300	-0.00024500
H	2.16300800	-1.98513100	-0.00093700
H	2.99419000	-0.68811600	-0.87779900
H	2.99377300	-0.68918300	0.87787700
C	1.33496300	1.39078800	0.00080000
H	1.91859900	1.68964200	0.88036700
H	1.91927400	1.69053400	-0.87801600
H	0.37467700	1.90290500	0.00065900
C	-1.47943500	-0.06801000	-0.00017300
O	-2.39454300	-0.92078300	0.00102900
O	-1.58954238	1.18556336	-0.00104038
H	-2.51904623	1.42560831	-0.00204427

### P\_125N



C	-0.28011600	1.74391200	-0.46002100
C	0.13927300	0.28078300	-0.68274900
C	-1.13841100	-0.15559700	0.14351000
H	0.28506300	2.23438700	0.34352900
H	-0.18950100	2.38815800	-1.34452800
H	-0.03948700	-0.01742100	-1.71781200
C	-1.92606700	-1.30532700	-0.46095600
H	-1.41984100	-2.27077800	-0.34198300
H	-2.90301300	-1.37042900	0.03042300
H	-2.10072300	-1.13800200	-1.52629900
C	-0.83219800	-0.48050000	1.60634900
H	-1.77886000	-0.62129300	2.13433800
H	-0.22241500	-1.38722200	1.72667400
H	-0.33006900	0.35585200	2.09435100
C	1.47225000	-0.29076400	-0.30191200
O	1.84214800	-1.38140000	-0.70106500
C	2.36764200	0.50345900	0.62223500
H	3.25339200	-0.08160600	0.86604600
H	2.65726200	1.43689400	0.13350400
H	1.83158500	0.78051500	1.53028100
C	-1.70813400	1.28085400	-0.03165600
H	-2.06131974	1.74169394	0.86711337
H	-2.45850732	1.35772012	-0.79055750

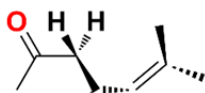
### IC\_125AN



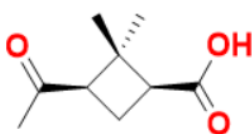
C	-0.99836500	-0.89971200	-0.10591900
C	0.15277700	-0.80105900	-1.07376800
C	-2.03495400	-0.08325700	0.07368200
H	-0.92010300	-1.75238600	0.56634100
H	0.08105900	0.12711400	-1.64643800
H	0.02464100	-1.60177300	-1.81609100
C	-3.04633900	-0.33321100	1.15938600
H	-3.07373800	0.49861900	1.87315700
H	-4.05948600	-0.42659600	0.75024400

H	-2.81869900	-1.24410600	1.71415100
C	-2.29945200	1.15644000	-0.73360600
H	-3.26612800	1.08911000	-1.24654700
H	-2.35025500	2.03812100	-0.08487100
H	-1.53398500	1.33686800	-1.48456700
C	2.26804100	0.02559500	0.16475300
O	3.39666700	-0.13683200	0.72132100
C	1.75830000	1.47177700	0.17517900
H	1.68396600	1.80247700	1.21429200
H	2.50381600	2.10659000	-0.31146900
H	0.79054400	1.62418900	-0.30630400
C	1.50823200	-0.95947500	-0.42720600
H	1.53034391	-1.60437696	0.42632345
H	2.05640284	-1.40936892	-1.22845764

IC_125BN				
C	-1.36203200	1.03394700	-0.13722400	
C	0.07145800	1.42244000	-0.45248000	
C	-1.96784800	-0.13954900	0.03304400	
H	-2.00961700	1.90826400	-0.03088100	
H	0.42526100	1.99160700	0.41546200	
H	-0.01946100	2.18988600	-1.24044900	
C	-3.44260000	-0.19048500	0.34136000	
H	-3.98254000	-0.76471800	-0.42041800	
H	-3.62692700	-0.69503600	1.29670800	
H	-3.88381500	0.80651700	0.39220600	
C	-1.32018200	-1.49314500	-0.04282600	
H	-1.44163100	-2.02176500	0.91001400	
H	-1.81366600	-2.10811900	-0.80497900	
H	-0.26063100	-1.42007900	-0.27444400	
C	2.10492700	-0.14296700	-0.11257000	
O	2.90357700	-1.05183300	-0.50313100	
C	2.35046000	0.32952100	1.32469000	
H	2.20369600	-0.51984900	1.99793400	
H	3.39799000	0.62873100	1.41177900	
H	1.71685200	1.15201000	1.66288200	
C	1.08119700	0.39391800	-0.85889000	
H	1.28846018	0.92973418	-1.76157655	
H	0.86014124	-0.62657551	-1.09261494	



S_169N				
C	1.06869600	-0.14685800	-0.76648300	
C	-0.07856200	-1.16918700	-0.84959300	
C	-1.06647900	0.00860600	-0.69117600	
C	0.08265300	0.88552500	-0.10202900	
H	1.35671800	0.22251700	-1.75419100	
H	-0.07932100	-1.83580800	0.01469700	
H	-0.14242500	-1.77073700	-1.75862600	

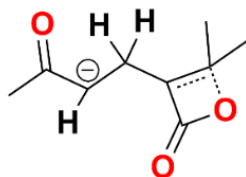




H	-1.33987000	0.39663700	-1.68210300
C	0.18363800	2.31354800	-0.61323000
H	-0.63894200	2.91669800	-0.21757300
H	1.12822500	2.76864700	-0.29897000
H	0.13894600	2.35065700	-1.70647600
C	0.14243300	0.85266500	1.42459900
H	1.08279900	1.29187400	1.77238600
H	-0.68442700	1.42802600	1.84475500
H	0.08026600	-0.16508500	1.82190900
C	-2.36905500	-0.14861300	0.05849400
C	2.34050000	-0.45019400	-0.03276900
O	3.31790700	0.26376900	-0.03901800
O	-2.84971000	0.78280400	0.67510600
O	2.30263900	-1.59651500	0.67950600
H	3.15971100	-1.67873500	1.12844200
C	-3.06312600	-1.48650300	-0.02258100
H	-4.07129700	-1.40417500	0.38358900
H	-3.09916600	-1.84421800	-1.05657600
H	-2.50210300	-2.23069900	0.55258100

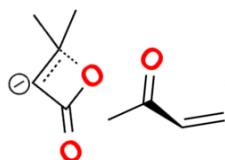
## 5.2 Products of the CID experiments

<b>IC_169</b>				
C	0.91118400	-0.45462200	-0.60549700	
C	-0.36076500	-1.24471000	-0.28481400	
C	-1.57722500	-0.58244700	-0.85806000	
C	1.20173900	0.90575700	0.06676000	
H	1.00282100	-0.37735500	-1.69161400	
H	-0.44274900	-1.37410500	0.79517600	
H	-0.19054200	-2.25656200	-0.68354900	
H	-1.55383300	-0.39795400	-1.93098500	
C	1.42564300	2.06401300	-0.87317700	
H	0.46774100	2.36920000	-1.29742400	
H	1.86402500	2.91422200	-0.34566800	

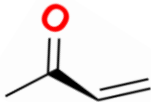
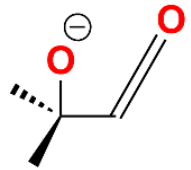
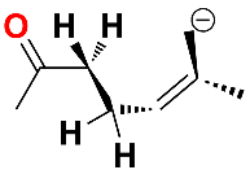


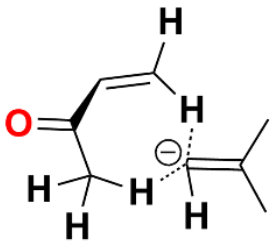
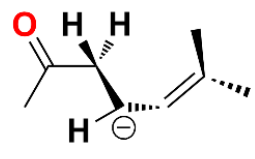
	H	2.09295500	1.77873400	-1.68787700
	C	0.36676300	1.28192400	1.26370300
	H	0.81394500	2.13825600	1.77393900
	H	-0.64248200	1.53400100	0.93895000
	H	0.29899200	0.45676100	1.97087400
	C	-2.72711800	-0.17482700	-0.21572500
	C	2.24075100	-0.86027600	-0.00896800
	O	2.51324400	0.36090400	0.52671100
	O	-3.70665900	0.41003200	-0.76879800
	O	2.92810100	-1.83360600	0.05992200
	C	-2.91302800	-0.41707900	1.28679000
	H	-3.82111600	-1.00894900	1.42228900
	H	-2.08969800	-0.92672100	1.79072400
	H	-3.08522000	0.54542100	1.77641400

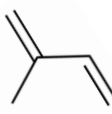
CX_99				
	C	1.70999700	0.11981700	0.89271600
	C	-1.70134500	0.71428000	-0.14612100
	C	-2.99880100	0.87284100	0.09613700
	C	2.83899600	-0.50813000	0.12164100
	H	1.35818700	0.03509200	1.90830400
	H	-1.27814900	-0.25816800	-0.36647800
	H	-0.97692400	1.52519800	-0.13400100
	H	-3.41507200	1.84775800	0.32737500
	C	4.23057400	-0.41699500	0.72846200
	H	4.31736900	-1.08962600	1.58757200
	H	5.00319700	-0.69164300	0.00212600
	H	4.41161600	0.60244400	1.06790400
	C	2.59608300	-1.87418800	-0.49976700
	H	3.37513500	-2.12742000	-1.22702400
	H	2.58584100	-2.64856200	0.27325300
	H	1.62921100	-1.87554500	-1.00207000
	C	-3.98836000	-0.21590200	0.08509600
	C	1.63963400	1.11460100	-0.06337500
	O	2.67179300	0.53393900	-0.89489200
	O	-5.16358500	0.00881700	0.31167600
	O	1.04742600	2.14645800	-0.36751500
	C	-3.52964400	-1.62577700	-0.21064300
	H	-4.38855300	-2.29098200	-0.15922600
	H	-3.07895700	-1.68233200	-1.20272700
	H	-2.77076400	-1.94321400	0.50597100



P°_70				
	C	1.92426400	0.17849900	-0.00018500
	C	0.87638900	-0.63128100	0.00029500
	H	1.81877600	1.25609900	-0.00070000
	H	2.93440800	-0.21067700	-0.00009800
	H	0.99772400	-1.70856400	0.00078100

	C	-0.53974600	-0.18798800	-0.00007100
	O	-1.41985300	-1.01938700	-0.00020100
	C	-0.85822300	1.28507300	0.00014400
	H	-1.93775600	1.40978100	0.00033700
	H	-0.43502400	1.77124100	0.88092400
	H	-0.43540000	1.77139700	-0.88073200
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<b>IC_99</b>				
	C	-0.53957400	-0.00012200	-0.81326000
	C	0.72445400	0.00000900	0.14288000
	H	-0.46860500	-0.00017600	-1.89520000
	C	1.54826200	-1.25702600	-0.25544500
	H	1.80491800	-1.29643600	-1.32263100
	H	2.47064400	-1.26057400	0.33202100
	H	0.97158300	-2.14566300	0.00557500
	C	1.54811400	1.25710300	-0.25557000
	H	2.47049500	1.26082400	0.33189900
	H	1.80476900	1.29643600	-1.32275900
	H	0.97132700	2.14569600	0.00535800
	C	-1.72973000	-0.00011400	-0.29502900
	O	0.38761800	0.00004800	1.42349400
	O	-2.80440400	0.00005200	0.16704200
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<b>IC_125B</b>				
	C	0.84785200	-0.35330100	0.40201800
	C	-0.13744500	0.50217600	0.90714600
	C	2.08600600	-0.07578200	-0.11704200
	H	0.59221300	-1.41606100	0.44630800
	H	0.01677700	1.57567400	0.91006600
	C	3.08960700	-1.15517300	-0.38270000
	H	3.56043400	-1.06429800	-1.37502400
	H	3.92831200	-1.16745100	0.33977000
	H	2.62266500	-2.14343100	-0.33531100
	C	2.57107200	1.33492300	-0.22463600
	H	2.98623900	1.73505400	0.72039500
	H	3.36410300	1.43266700	-0.97732100
	H	1.76030600	2.01563500	-0.50977900
	C	-2.24195000	-0.22582800	-0.08000400
	O	-2.53366800	-1.33545900	-0.49739900
	C	-2.53975500	0.98613200	-0.94323800
	H	-1.97906800	0.91539100	-1.87611300
	H	-3.60466900	0.97630500	-1.19226900
	H	-2.27475200	1.92087300	-0.45111200
	C	-1.52511800	0.00418000	1.22072900
H	-1.52898900	-0.94404600	1.77026700	
H	-2.07584500	0.74339800	1.81567200	
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<b>CX_125B</b>				
C	-1.52656200	0.22838400	-0.73896500	

	C	1.37698000	1.68552300	0.25799300
	C	2.56401800	1.09439500	0.17092700
	C	-2.68933600	-0.02070400	-0.11871200
	H	-1.72093400	0.44937900	-1.80991000
	H	0.41977800	1.15610900	0.12518700
	H	1.32677600	2.76142200	0.40104700
	H	3.49357400	1.65464400	0.22617000
	C	-4.06939200	-0.00975800	-0.75661000
	H	-4.57448200	-0.98282500	-0.66171200
	H	-4.73745600	0.72366700	-0.28036500
	H	-4.00298500	0.23389300	-1.81836400
	C	-2.75281300	-0.35741700	1.35780500
	H	-3.36443300	0.36439100	1.91981800
	H	-3.20597900	-1.34398600	1.53693100
	H	-1.74890300	-0.35996700	1.78267400
	C	2.75278800	-0.35340100	-0.07092100
	O	3.86143800	-0.76973200	-0.36874400
	C	1.57528500	-1.26837800	0.07873700
	H	1.81438100	-2.23475500	-0.36318400
	H	1.37995800	-1.40664500	1.14654900
H	0.64338600	-0.84932500	-0.33640300	
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<b>IC_125C</b>				
	C	-0.90564900	-0.07593300	-0.90452800
	C	0.16871700	-1.10180300	-0.81436900
	C	1.19765700	-0.96671800	0.36739800
	C	-1.96979400	-0.03204300	-0.01259800
	H	-0.89811400	0.60393200	-1.74946500
	H	0.74650500	-1.14582300	-1.74366000
	H	-0.26836900	-2.10318700	-0.67463400
	H	1.79006000	-1.88182400	0.43539900
	C	-3.04579500	0.99752100	-0.30487600
	H	-3.87271000	0.54459600	-0.86046600
	H	-3.45788300	1.41167400	0.61913000
	H	-2.65356100	1.82243000	-0.90484300
	C	-2.17610900	-0.83920300	1.09437800
	H	-3.06722300	-0.72505700	1.70034300
	H	-1.50893800	-1.65224800	1.35237800
	H	0.61804300	-0.83295100	1.28367700
	C	2.14145500	0.17702300	0.18346500
	O	3.27204800	0.02760800	-0.24833400
C	1.60061800	1.53784800	0.51127400	
H	1.39905700	1.60316000	1.58361500	
H	2.30233800	2.31555100	0.21165500	
H	0.62781600	1.63872400	0.01267700	
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<b>P<sup>o</sup>_69</b>				
C	0.85592800	-0.53275900	-0.27176500	

	C	2.01113500	-0.09946300	0.21080500	
	C	-0.45397700	0.11176700	-0.06523500	
	H	0.84189000	-1.45592800	-0.84665600	
	H	2.93630200	-0.62373600	0.00884700	
	H	2.06818700	0.78646500	0.83185600	
	C	-1.61039400	-0.82181700	0.15404100	
	H	-1.44948300	-1.43989200	1.04076000	
	H	-2.54554400	-0.27713400	0.27731200	
	H	-1.72036100	-1.50443600	-0.69291400	
	C	-0.60204200	1.43123400	-0.09352900	
	H	-1.56757200	1.89218400	0.07446000	
	H	0.23267800	2.08869600	-0.29957000	
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	<b>TS_1</b>	C	-1.72949000	1.20055800	-0.18635200
	C	-0.29789500	1.71574900	-0.42062100	
	C	0.33046800	0.36217900	-0.79552300	
	C	-1.35619100	-0.13100000	0.18879000	
	H	-2.29421500	1.21108700	-1.12403700	
	H	0.16962400	2.15342600	0.47146500	
	H	-0.22040400	2.48674500	-1.20071500	
	H	0.00565500	-0.00873900	-1.76595200	
	C	-2.06874000	-1.34302100	-0.36183300	
	H	-1.35445600	-2.16330400	-0.48081800	
	H	-2.85366400	-1.67932400	0.32730900	
	H	-2.52099400	-1.12892400	-1.33211100	
	C	-0.82476400	-0.39195500	1.58450500	
	H	-1.67902200	-0.59034400	2.24645400	
	H	-0.15491400	-1.25743200	1.61095100	
	H	-0.30182400	0.47674700	1.98324400	
	C	1.55041000	-0.23331700	-0.36360800	
	O	1.96629500	-1.32924400	-0.79371800	
	C	2.35225800	0.40264900	0.76961000	
	H	2.60597200	-0.36838900	1.50219100	
	H	3.29112300	0.78389500	0.35512800	
	H	1.84042800	1.22746400	1.26683500	
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<b>TS_2</b>	C	-1.27697700	-0.14758100	-0.03688900	
	C	1.06917100	1.52520900	-0.40738400	
	C	2.11824200	0.79798200	-0.81104800	
	C	-2.61449500	-0.16091200	-0.01620800	
	H	-0.87565100	-1.00028000	-0.62110700	
	H	0.63015500	1.44121900	0.57175800	
	H	0.59613600	2.23911300	-1.07005300	
	H	2.51733500	0.88448400	-1.81621800	
	C	-3.50519900	-1.19780100	-0.68111100	
	H	-4.14981800	-1.71158700	0.04716900	

	H	-4.18376100	-0.74250600	-1.41760000
	H	-2.90668500	-1.95199900	-1.19405500
	C	-3.41611400	0.90279400	0.70465500
	H	-4.08829100	1.44385300	0.02235400
	H	-4.05768200	0.47304500	1.48809000
	H	-2.74268200	1.62306900	1.16889100
	C	2.75027200	-0.22442400	0.01725200
	O	3.73585400	-0.84081400	-0.37019800
	C	2.12138800	-0.52908200	1.35281500
	H	2.57691500	-1.42824100	1.76376800
	H	2.28375100	0.30262200	2.04326400
	H	1.03571600	-0.64337800	1.24282600
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<b>TS_3</b>	C	-1.19777000	-0.67018400	0.08896100
	C	0.12848600	-0.07626800	0.05528600
	C	1.32004000	-0.75445300	-0.15473900
	C	-2.41055300	-0.07145200	-0.28938500
	H	-1.18203100	-1.75845300	0.14692800
	H	-0.72943400	-0.34809700	1.30719200
	H	0.13905600	1.00557500	0.11503200
	H	1.29466300	-1.81672400	-0.37521400
	C	-3.66031000	-0.72033300	0.22781600
	H	-3.89461300	-0.43819800	1.27948300
	H	-4.54341000	-0.44839300	-0.36270500
	H	-3.58314400	-1.81225800	0.21150500
	C	-2.45840600	1.42211100	-0.32517600
	H	-3.41152300	1.78341600	-0.72477800
	H	-2.35087200	1.88082700	0.68719800
	H	-1.66294000	1.85515700	-0.94051500
	C	2.60972700	-0.17616200	-0.04536900
	O	3.66388700	-0.80424700	-0.23225800
	C	2.72500600	1.30170900	0.32131600
	H	3.77963600	1.57129500	0.31533200
	H	2.18768100	1.93615300	-0.38700100
	H	2.30852800	1.49387200	1.31334400
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<b>TS_4</b>	C	1.15456500	0.06181000	1.08283000
	C	-0.07527700	-0.75827600	1.37763300
	C	-0.99985000	-0.91940500	0.15813400
	C	1.82732600	-0.06278400	-0.08028900
	H	1.43605400	0.83982000	1.78772000
	H	-0.61003800	-0.31769300	2.22712000
	H	0.20102900	-1.77018400	1.70533500
	H	-1.44285800	-1.91039000	0.05745800
	C	2.88859600	0.93924200	-0.45888300
	H	3.86759900	0.46119300	-0.56644700

	H	2.65277000	1.39331100	-1.42725800
	H	2.97314300	1.74144600	0.27930800
	C	1.44608200	-1.07605500	-1.04591900
	H	1.91281600	-0.98868400	-2.02880300
	H	1.47111000	-2.10642500	-0.68692900
	H	-0.02654000	-0.95280900	-0.77778400
	C	-1.94401700	0.09203100	-0.20036100
	O	-2.98708700	-0.13245300	-0.83681900
	C	-1.61923000	1.54625800	0.10797500
	H	-1.01317000	1.95414700	-0.70561900
	H	-2.55226500	2.10868300	0.15882300
	H	-1.04212800	1.67028600	1.02491900
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<b>TS_5</b>	C	-1.45228500	-0.04566300	-0.89297500
	C	0.93313900	1.44598700	-0.30483700
	C	2.22794200	1.10728800	-0.20171800
	C	-2.52872800	-0.04565400	-0.10331700
	H	-1.64168900	-0.49694800	-1.88164900
	H	-0.17699200	0.66792200	-0.50556900
	H	0.77067900	2.51477400	-0.50069800
	H	3.05762000	1.81161500	-0.30381200
	C	-3.89149300	-0.59688500	-0.46671100
	H	-4.20437300	-1.39250500	0.22344500
	H	-4.66958000	0.17718800	-0.41281100
	H	-3.89028100	-1.00663300	-1.47811500
	C	-2.47631800	0.53304000	1.29387700
	H	-3.21030900	1.33958800	1.42636400
	H	-2.71177500	-0.22390100	2.05420900
	H	-1.48540700	0.93346700	1.50687100
	C	2.68666100	-0.26482000	0.04255400
	O	3.87730600	-0.55356300	0.02859200
	C	1.65621300	-1.32545000	0.34765300
	H	2.09124800	-2.30780800	0.16518500
	H	1.38872700	-1.25250000	1.40531800
	H	0.73289200	-1.18281900	-0.21462500
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<b>TS_6</b>	C	0.15733900	-0.82413500	-0.76570200
	C	1.77164900	0.03139000	0.19415000
	H	0.70081900	-1.72313500	-1.07903600
	C	1.18692400	1.20295500	0.94337800
	H	1.98587500	1.73862100	1.46943100
	H	0.69435600	1.90265700	0.27015100
	H	0.45916400	0.86938400	1.68379400
	C	2.59933800	0.42563000	-1.00637700
	H	2.89459300	-0.45674300	-1.57659500
	H	2.03609900	1.08377200	-1.66946000

	H	3.51381900	0.94009100	-0.68396400
	C	-2.13871200	-0.12456700	-0.03208900
	O	-3.23391600	-0.44701000	0.43574900
	C	-1.92230500	1.34224200	-0.36053000
	H	-1.02980100	1.49404800	-0.96212400
	H	-2.80685700	1.69750200	-0.89242700
	H	-1.84128600	1.91673100	0.56427600
	C	-1.11373000	-1.09314400	-0.34480100
	H	-1.48828000	-2.11445300	-0.23461100
	C	2.16998000	-1.06612300	0.91940700
	H	2.77881000	-1.84056700	0.46699100
	H	1.71112100	-1.29732000	1.87295300
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<b>TS_7</b>	C	-1.24510300	-0.04273400	-0.75923700
	C	1.64083800	1.85553600	0.36759200
	C	2.62106800	0.99851800	0.11457000
	C	-2.44214500	-0.05397800	-0.16588400
	H	-1.30922500	0.23113400	-1.82675700
	H	0.61539600	1.51953100	0.48380900
	H	1.82845000	2.92358200	0.40581400
	H	3.63351600	1.34083800	-0.08164100
	C	-3.76458000	0.26635300	-0.83710800
	H	-4.46369400	-0.58030600	-0.78388200
	H	-4.27339900	1.11198300	-0.35292900
	H	-3.61666400	0.51737100	-1.88850200
	C	-2.60663200	-0.40664700	1.29748900
	H	-3.05681400	0.41837700	1.86728400
	H	-3.27065600	-1.27135800	1.43506700
	H	-1.64232200	-0.64436000	1.74564900
	C	2.42440200	-0.47249000	-0.04831400
	O	3.24683000	-1.08164000	-0.73667300
	C	1.22311200	-1.05237600	0.52557800
	H	1.22538200	-2.13946500	0.45716700
	H	1.03333600	-0.70926600	1.54339800
	H	0.21630000	-0.61803600	-0.07921100
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<b>TS_8</b>	C	-0.97973300	-0.01688800	-0.88331400
	C	-0.02131400	-1.00450600	-0.93381900
	C	-2.12954600	-0.03615200	-0.02083100
	H	-0.82250700	0.88678200	-1.46462500
	H	0.73417200	-0.96643500	-1.70756400
	H	-0.28824200	-2.00165700	-0.60406700
	C	-3.13600700	1.06547000	-0.25664000
	H	-3.52986200	1.02166700	-1.27610200
	H	-3.97250100	1.00154700	0.44102900
	H	-2.66660700	2.04644200	-0.13859600

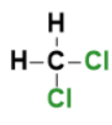
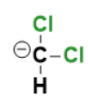
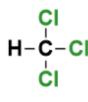
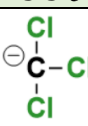
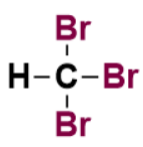

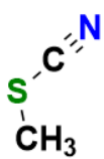


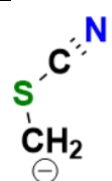
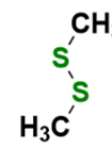
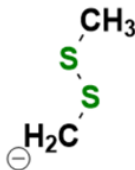
	C	-2.36662700	-0.93503900	0.95799900
	H	-3.28046600	-0.89853200	1.53912400
	H	-1.65720600	-1.71558900	1.20042500
	C	2.28965100	0.10435200	0.21297200
	O	3.21713100	0.12279900	-0.62215600
	C	1.77310800	1.42684000	0.75536100
	H	1.62634000	1.39590700	1.83721300
	H	2.46431000	2.22809300	0.49480400
	H	0.79342400	1.62641900	0.30281700
	C	1.56138200	-1.03849100	0.59205600
	H	1.97947800	-2.00512200	0.33371300
	H	0.93713000	-0.99542900	1.47637700
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<b>TS_9</b>	C	1.04978200	-0.39108900	-0.68162800
	C	-0.09854100	-1.38383000	-0.68246500
	C	-1.24306300	-0.42871100	-0.92580900
	C	0.53875100	0.83553000	0.04799700
	H	1.24707100	-0.10470300	-1.71912000
	H	-0.17697900	-1.88431000	0.28290800
	H	0.02619300	-2.16823300	-1.43605300
	H	-1.20012100	0.12319300	-1.86069900
	C	0.44732900	2.17674600	-0.59526000
	H	-0.60567900	2.45629000	-0.68721200
	H	0.94768900	2.92092700	0.02852700
	H	0.90430700	2.18244900	-1.58364400
	C	0.05925600	0.77691600	1.46533600
	H	0.73242500	1.37370700	2.08340500
	H	-0.94617600	1.19396600	1.54167600
	H	0.05275100	-0.23803700	1.85682100
	C	-2.41976500	-0.23881900	-0.19870200
	C	2.39903400	-0.47775000	0.00572700
	O	2.51981700	0.69205600	0.53908400
	O	-3.25713700	0.65852400	-0.45768100
	O	3.18189600	-1.41266500	0.02212700
	C	-2.73328700	-1.10738600	1.01475700
	H	-2.01501700	-1.90509800	1.20182800
	H	-2.80079000	-0.47302200	1.90203800
	H	-3.71925900	-1.55009300	0.86157000
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<b>TS_10</b>	C	0.94633200	0.39057300	0.60210500
	C	-0.96636600	1.19430300	-0.41413000
	C	-2.13274400	1.07805400	0.29761100
	C	1.72266600	-0.83219200	0.14311300
	H	0.56335400	0.56748500	1.59649600
	H	-0.79438600	0.58653800	-1.29099400
	H	-0.40984500	2.12038700	-0.39267400

	H	-2.35697400	1.79393100	1.08126200
	C	2.54054900	-1.54289300	1.20671600
	H	1.88204700	-2.09711000	1.88095400
	H	3.24446800	-2.25038200	0.75764000
	H	3.10198800	-0.81550300	1.79300700
	C	1.03850200	-1.81868900	-0.78130800
	H	1.76804500	-2.49620900	-1.23467300
	H	0.31360100	-2.41838800	-0.22601400
	H	0.51369000	-1.29233000	-1.57666100
	C	-3.08802900	0.02273700	0.15875000
	C	1.93033800	1.16982400	-0.06542200
	O	2.59274200	0.06105300	-0.64127200
	O	-4.11254500	-0.04874500	0.84477500
	O	2.22576200	2.32490900	-0.27584600
	C	-2.84467400	-1.06102000	-0.88140700
	H	-3.72258400	-1.70324000	-0.91978000
	H	-2.65919800	-0.63731500	-1.87009800
	H	-1.97132400	-1.65979200	-0.61589000
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<b>TS_11</b>	C	-0.27218600	-0.61350000	-0.88094500
	C	0.83914400	-0.08632100	-0.18463800
	H	-0.30826800	-1.65771200	-1.17379100
	C	1.86521200	-1.02086700	0.40381000
	H	1.90243900	-1.94542700	-0.17618800
	H	2.86979300	-0.59206200	0.47272200
	H	1.53383400	-1.27798100	1.41227200
	C	1.33943100	1.31934500	-0.39567300
	H	1.74992800	1.74739300	0.52322700
	H	2.14321600	1.28670500	-1.14466900
	H	0.54176500	1.95672000	-0.76813600
	C	-1.30816600	-0.06171100	-0.01449800
	O	-0.61786100	0.52231400	0.97237200
	O	-2.53380300	-0.11472900	-0.06159400
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<b>TS_12</b>	C	0.47683900	0.00000000	-0.89937100
	C	-0.69581200	0.00000000	0.08642900
	H	0.46325200	0.00000000	-1.98596000
	C	-1.56992200	1.25263000	-0.16466300
	H	-1.97927600	1.30025000	-1.18291000
	H	-2.39904000	1.25019900	0.54870400
	H	-0.96430900	2.14192200	0.01673400
	C	-1.56992200	-1.25263000	-0.16466300
	H	-2.39904000	-1.25019800	0.54870400
	H	-1.97927500	-1.30025000	-1.18291000
	H	-0.96430900	-2.14192200	0.01673500
	C	1.63117900	0.00000000	-0.27503300

	O	-0.15027700	0.00000000	1.32693400
	O	2.72375500	0.00000000	0.13865500

### 5.3 Proton transfer (PR) reaction

<b>CH<sub>2</sub>Cl<sub>2</sub></b>	C	0.00000000	0.00000000	0.76311300
	H	-0.89885900	0.00000000	1.37081900
	H	0.89885900	0.00000000	1.37081900
	Cl	0.00000000	1.47460800	-0.21530300
	Cl	0.00000000	-1.47460800	-0.21530300
<b>CHCl<sub>2</sub></b>	C	-0.02636900	0.95405500	0.00000000
	H	1.05476000	1.18343600	0.00000000
	Cl	-0.02636900	-0.20317000	1.50771100
	Cl	-0.02636900	-0.20317000	-1.50771100
<b>CHCl<sub>3</sub></b>	C	0.00000000	0.00000000	0.45285000
	H	0.00000000	0.00000000	1.53642700
	Cl	0.00000000	1.68170700	-0.08340200
	Cl	1.45640100	-0.84085300	-0.08340200
	Cl	-1.45640100	-0.84085300	-0.08340200
<b>CCl<sub>3</sub></b>	C	0.00000000	0.00000000	0.72425900
	Cl	0.00000000	1.70640500	-0.08520700
	Cl	1.47779000	-0.85320300	-0.08520700
	Cl	-1.47779000	-0.85320300	-0.08520700
<b>CHBr<sub>3</sub></b>	C	-0.00020600	-0.00002200	0.52270800
	H	-0.00021800	-0.00004800	1.60455900
	Br	0.43287100	1.79169500	-0.04514600
	Br	1.33543800	-1.27066100	-0.04514400
	Br	-1.76826800	-0.52102800	-0.04516200
<b>CBr<sub>3</sub></b>	C	0.00000000	0.00000000	0.81761900
	Br	0.00000000	1.87698300	-0.04672100
	Br	-1.62551500	-0.93849200	-0.04672100
	Br	1.62551500	-0.93849200	-0.04672100
<b>CH<sub>3</sub>SCN</b>	C	-1.48350600	0.80496100	0.00000400
	H	-2.51383700	0.44797700	-0.00003800
	H	-1.29544800	1.38932400	-0.89813500
	H	-1.29551300	1.38926300	0.89819600
	S	-0.46277500	-0.69990700	-0.00001800
	C	1.07262400	0.00209800	-0.00018700
	N	2.13921200	0.44708400	0.00019400
<b>CH<sub>2</sub>SCN</b>	C	-1.69524200	0.69300900	0.00004000
	H	-1.67518900	1.29592700	0.91025600
	H	-1.67581000	1.29530800	-0.91060500

	S	-0.55025900	-0.59283600	0.00004700
	C	1.13319900	-0.01565600	-0.00016800
	N	2.21820000	0.40428900	0.00005200
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<b>CH<sub>3</sub>SSCH<sub>3</sub></b>	S	0.90023900	0.49001000	-0.49082300
	S	-0.90024600	0.49002000	0.49081600
	C	-1.84888700	-0.78922200	-0.37651800
	H	-2.84597000	-0.77622000	0.07064200
	H	-1.92460200	-0.55383000	-1.43711900
	H	-1.40732800	-1.77493600	-0.23300000
	C	1.84889700	-0.78920500	0.37652900
	H	1.40751300	-1.77495800	0.23276400
	H	2.84606700	-0.77599500	-0.07043000
	H	1.92436800	-0.55397400	1.43718300
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<b>CH<sub>3</sub>SSCH<sub>2</sub></b>	S	0.97985200	-0.13747200	-0.00000100
	C	2.41222043	-0.79885926	-0.86913272
	H	2.47021033	-0.75823899	-1.98671794
	H	3.38215629	-0.35335769	-0.55106438
	H	2.62502582	-1.86894963	-0.64545271
	C	-0.78806600	0.74601200	0.00000500
	H	-0.59134407	1.68900827	0.50579855
	H	-1.39627571	0.07828002	0.60662141
	S	-1.58733886	1.04793285	-1.55770227