



*Supplement of*

**The water-insoluble organic carbon in PM<sub>2.5</sub> of typical Chinese urban areas: light-absorbing properties, potential sources, radiative forcing effects, and a possible light-absorbing continuum**

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1 **Text S1. Water soluble inorganic ions measurements**

2 3 anions (Cl<sup>-</sup>, NO<sub>3</sub><sup>-</sup> and SO<sub>4</sub><sup>2-</sup>) and 4 cations (Na<sup>+</sup>, K<sup>+</sup>, Ca<sup>2+</sup> and NH<sub>4</sub><sup>+</sup>) were analyzed with ion-  
3 chromatography (761 Compact IC, Metrohm, Switzerland). Anions were separated on a Metrohm Metrosep  
4 A sup5-250 column with 3.2 mM Na<sub>2</sub>CO<sub>3</sub> and 1.0 mM NaHCO<sub>3</sub> as the eluent and 35 mM H<sub>2</sub>SO<sub>4</sub> for a  
5 suppressor. Cations were measured using a Metrohm Metrosep C4-150 column with 2 mM sulfuric acid as  
6 the eluent. The injection loop volume for anion and cation was 100 μL. The water-soluble ions analyses  
7 were duplicated for several filter samples, and the overall relative standard deviations were generally less  
8 than 4%.

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10 **Text S2. Calculation of MAC for particulate light-absorbing OC**

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12 MAC can be compared with MAE only after considering the particulate effect ( $\xi$ ) and in the small-particle  
13 limit (diameter  $\ll \lambda$ ) (Sun et al., 2007):

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$$\text{MAC}(\lambda) = \text{MAE}(\lambda) \times \xi(\lambda) \quad (\text{S1})$$

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$$\xi(\lambda) = \frac{9n}{(n^2 - k(\lambda)^2 + 2) + 4n^2k(\lambda)^2} \quad (\text{S2})$$

17 where  $n$  and  $k$  represent the real and imaginary parts of the complex refractive index ( $m = n + ik$ ),  
18 respectively. In this study, we assume a constant  $n$  value of 1.55 (Lu et al., 2015), and the wavelength-  
19 dependent  $k$  is calculated as:

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$$k(\lambda) = \frac{\rho \times \lambda \times \text{MAE}(\lambda)}{4\pi \times \left(\frac{\text{OA}}{\text{OC}}\right)} \quad (\text{S3})$$

21 where  $\rho$  is the density of particle and was fixed to 1.2 g/cm<sup>3</sup>. The OA/OC ratios are 1.51, 1.91, 2.30 for  
22 WIOC, HULIS and non-HULIS, respectively (Kiss et al., 2002).

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**Table S1.** Summary of concentration and light absorption of extractible organic carbon components in PM<sub>2.5</sub> from ten Chinese cities

Component	Unit	Warm seasons rang (avg <sup>a</sup> ± std <sup>b</sup> )	Cold seasons rang (avg ± std)	Annual avg ± std
WIOC		1.45–5.25 (2.29 ± 0.95)	1.93–12.9 (4.87 ± 2.89)	3.65 ± 2.53
HULIS-C	μgC/m <sup>3</sup>	1.37–4.31 (2.46 ± 0.77)	2.10–7.64 (4.63 ± 1.49)	3.60 ± 1.62
Non-HULIS-C <sup>c</sup>		1.28–5.61 (2.36 ± 1.16)	1.55–7.96 (4.09 ± 1.51)	3.27 ± 1.60
EX-OC <sup>d</sup>		4.19–12.8 (7.11 ± 2.38)	6.25–25.2 (13.6 ± 5.22)	10.5 ± 5.23
Abs <sub>365, WIOC</sub>	Mm <sup>-1</sup>	1.27–7.69 (2.76 ± 1.77)	2.78–38.5 (10.4 ± 8.69)	6.80 ± 7.44
Abs <sub>365, HULIS</sub>		1.04–6.05 (2.96 ± 1.4)	2.27–17.2 (8.72 ± 3.75)	5.99 ± 4.08
Abs <sub>365, non-HULIS</sub>		0.53–2.72 (1.25 ± 0.68)	1.11–8.50 (3.76 ± 2.27)	2.57 ± 2.11
Abs <sub>365, EX-OC</sub>		3.24–16.1 (6.98 ± 3.54)	6.42–55.4 (22.9 ± 13.0)	15.4 ± 12.6
WIOC/EX-OC		15.4–48.7 (32.5 ± 6.97)	19.8–57.5 (34.2 ± 8.12)	33.4 ± 7.55
HULIS-C/EX-OC		25.2–45.4 (35.1 ± 5.29)	19.2–47.6 (35.3 ± 6.32)	35.2 ± 5.77
Non-HULIS-C/EX-OC	%	21.1–43.8 (32.4 ± 5.92)	23.4–36.6 (30.5 ± 4.34)	31.4 ± 5.17
Abs <sub>365, WIOC</sub> /Abs <sub>365, EX-OC</sub>		20.3–50.6 (38.4 ± 9.06)	29.4–69.6 (42.5 ± 10.1)	40.5 ± 9.73
Abs <sub>365, HULIS-C</sub> /Abs <sub>365, EX-OC</sub>		32.2–55.1 (42.8 ± 6.54)	22.5–57.9 (40.5 ± 7.90)	41.6 ± 7.28
Abs <sub>365, non-HULIS-C</sub> / Abs <sub>365, EX-OC</sub>		9.51–26.8 (18 ± 4.43)	6.80–30.4 (17.0 ± 5.58)	17.5 ± 5.02

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<sup>a</sup> avg: average

<sup>b</sup> std: standard deviation

<sup>c</sup> The concentration of non-HULIS-C is calculated by the difference between WSOC and HULIS-C

<sup>d</sup> The concentration of extractible organic carbon (EX-OC) is the sum of WSOC and WIOC.

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**Table S2.** The person correlation coefficients ( $r$ ) of concentrations and Abs<sub>365</sub> of WIOC with water soluble ions in cold and warm seasons.

	Concentrations of WIOC		Abs <sub>365</sub> of WIOC	
	Warm seasons	Cold seasons	Warm seasons	Cold seasons
K <sup>+</sup>	0.04	0.61**	0.25	0.48*
Cl <sup>-</sup>	0.46	0.92**	0.63**	0.90**
NO <sub>3</sub> <sup>-</sup>	0.38	0.29	0.53*	0.18
SO <sub>4</sub> <sup>2-</sup>	0.44	0.69**	0.51*	0.63**
NH <sub>4</sub> <sup>+</sup>	0.51*	0.51*	0.59*	0.49*

\* Significance at  $p < 0.05$  level

\*\* Significance at  $p < 0.01$  level

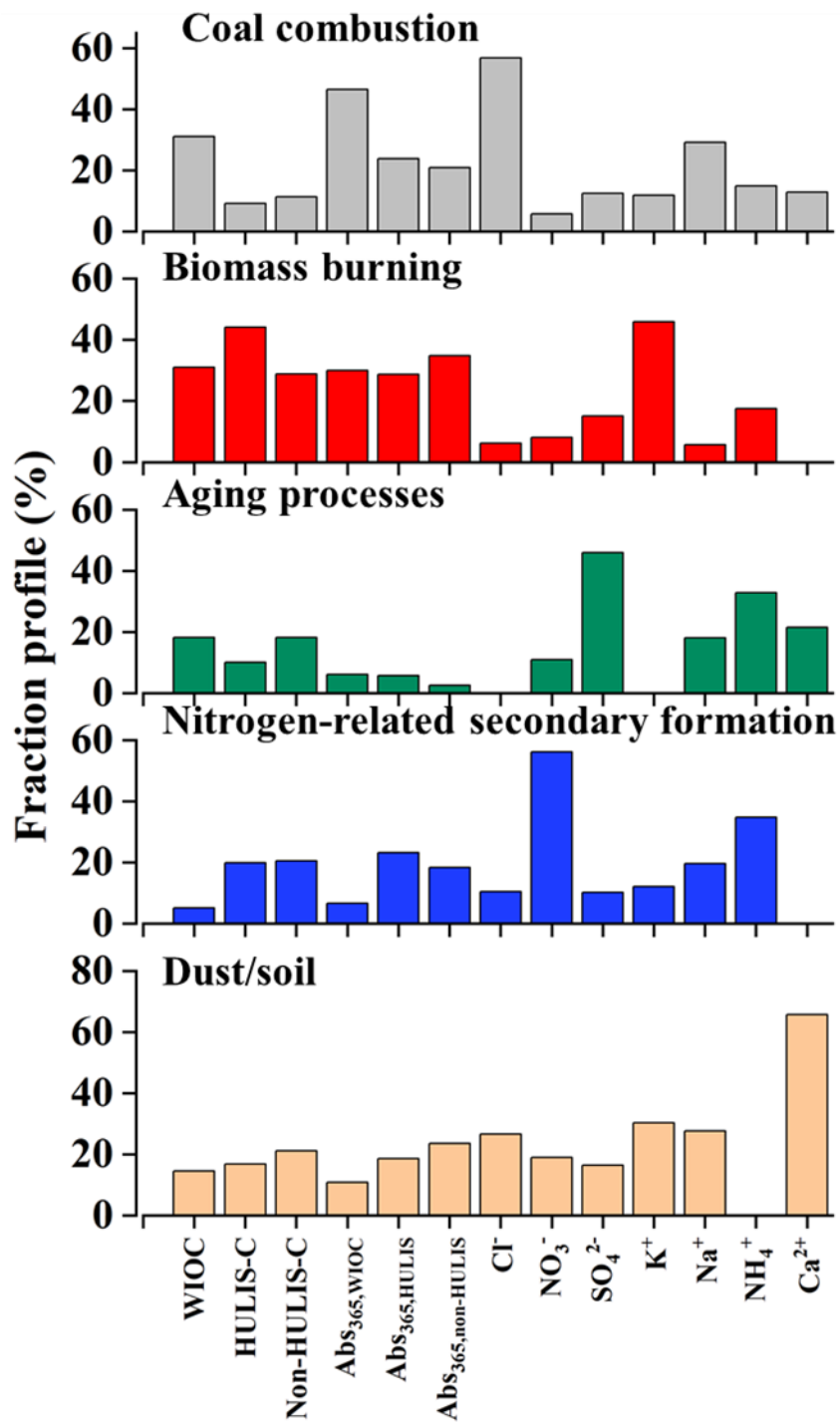
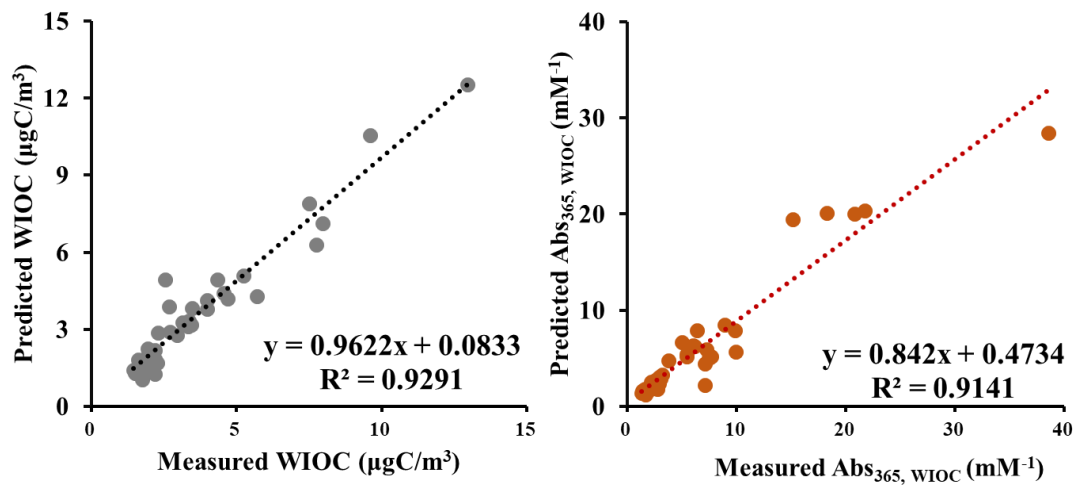


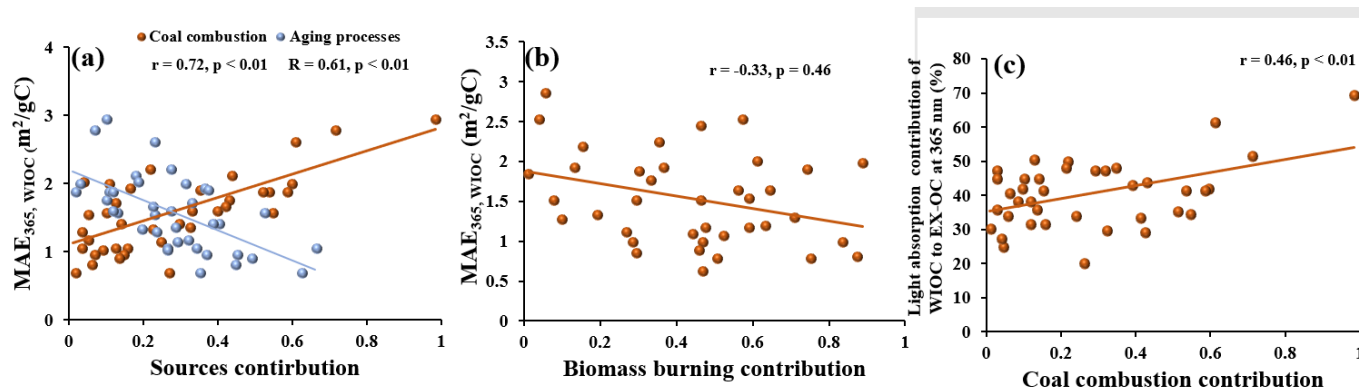
Figure S1. Source profiles for five sources resolved by the positive matrix factorization (PMF) model.

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**Figure S2.** PMF-predicted versus measured values of (a) concentrations and (b) Abs<sub>365</sub> of WIOC.

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**Figure S3.** (a) The correlation of mass absorption efficient (MAE<sub>365</sub>) of WIOC to relative contribution of coal combustion (brown dots) and aging processes (blue dots). (b) The relationship between the MAE<sub>365</sub> of WIOC and relative contribution of biomass burning. (c) The relationship between relative contribution of coal combustion and light absorption contribution of WIOC to EX-OC at 365 nm.

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## References:

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Kiss, G., Varga, B., Galambos, I., Ganszky, I. (2002), Characterization of water-soluble organic matter isolated from atmospheric fine aerosol, *Journal of Geophysical Research-Atmospheres*, 107(D21).<https://doi.org/10.1029/2001jd000603>

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Lu, Z.F., Streets, D.G., Winijkul, E., Yan, F., Chen, Y.J., Bond, T.C., Feng, Y., Dubey, M.K., Liu, S., Pinto, J.P., Carmichael, G.R. (2015), Light Absorption Properties and Radiative Effects of Primary Organic Aerosol Emissions, *Environmental Science & Technology*, 49(8), 4868-4877.<https://doi.org/10.1021/acs.est.5b00211>

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Sun, H.L., Biedermann, L., Bond, T.C. (2007), Color of brown carbon: A model for ultraviolet and visible light absorption by organic carbon aerosol, *Geophysical Research Letters*, 34(17).<https://doi.org/10.1029/2007gl029797>

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