Characterization of refractory aerosol particles collected in the tropical upper troposphere–lower stratosphere (UTLS) within the Asian tropopause aerosol layer (ATAL)

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Abstract. Aerosol particles with diameters larger than 40 nm were collected during the flight campaign StratoClim 2017 within the Asian tropopause aerosol layer (ATAL) of the 2017 monsoon anticyclone above the Indian subcontinent. A multi-impactor system was installed on board the aircraft M-55 Geophysica, which was operated from Kathmandu, Nepal. The size and chemical composition of more than 5000 refractory particles/inclusions of 17 selected particle samples from seven different flights were analyzed by use of scanning electron microscopy (SEM) and transmission electron microscopy (TEM) combined with energy dispersive X-ray (EDX) microanalysis. Based on chemical composition and morphology, the refractory particles were assigned to the following particle groups: extraterrestrial, silicates, Fe-rich, Al-rich, Hg-rich, other metals, C-rich, soot, Cl-rich, and Ca-rich.

Most abundant particle groups within the refractory particles are silicates and C-rich (non-volatile organics). In samples taken above the tropopause, extraterrestrial particles are becoming increasingly important with rising altitude. The most frequent particle sources for the small (maximum in size distribution $D_{P,max} = 120$ nm) refractory particles carried into the ATAL are combustion processes at the ground (burning of fossil fuels/biomass burning) and the agitation of soil material. The refractory particles in the ATAL represent only a very small fraction (< 2 % by number for particles > 40 nm) of the total aerosol particles, which are dominated by species like ammonium, sulfate, nitrate, and volatile organics. During one flight, a large number of very small ($D_{P,max} = 25$ nm) cinnabar particles (HgS) were detected, which are supposed to originate from a ground source such as coal combustion or underground coal fires.
1 Introduction

Each year during the summer monsoon from June to September, the Asian tropopause aerosol layer (ATAL) develops inside the Asian monsoon anticyclone (AMA) (Vernier et al., 2018; Zhang et al., 2019). The ATAL forms in the upper-troposphere–lower-stratosphere (UTLS) region at altitudes of about 14 to 18 km, corresponding to potential temperature levels of 360 to 420 K (Hanumanthu et al., 2020). The lateral dimensions of the AMA can extend from the eastern Mediterranean up to East Asia (Vernier et al., 2011).

During the summer monsoon, large-scale convection in the Himalayan region provides a strong upward transport of gases and particles from the ground to the ATAL. Because of “eddy shedding” (Dethof et al., 1999; Popovic and Plumb, 2001; Pan et al., 2016), a smaller fraction of the particles is even transported into the lower stratosphere and can be subject to long-range transport there (Vogel et al., 2016, 2019; Fujiwara et al., 2021).

Model analysis by Fairlie et al. (2020) showed the dominance of regional anthropogenic emissions of particle precursors like sulfate, nitrate, ammonia, and organic aerosol particles from China and the Indian subcontinent in affecting observed aerosol concentrations in the ATAL. The first in situ mass spectrometric analysis of aerosol particles within the ATAL (Appel et al., 2022) determined that the particles in the ATAL consist mainly of ammonium nitrate and organics. It was further found that up to 70% of these are formed from the conversion of inorganic and organic gas-phase precursors rather than from a direct uplift of primary particles from the boundary layer. Höpfner et al. (2019) gained important insights into the formation of ammonium nitrate from agricultural emissions of ammonia uplifted within the Asian monsoon in the UTLS region.

But still, all details of processes involved in nucleation of inorganic compounds (including sulfuric acid, ammonium salts, and nitric acid) as well as secondary organic aerosol (SOA) formation in the AMA are not completely understood. It is also not clear whether aerosol particles from the AMA (either transported from below or newly formed within the AMA) are a relevant source of the stratospheric Junge layer (Appel et al., 2022).

An increased particle concentration has a variety of severe atmospheric implications, including the high relevance for the climate system. First, aerosol particles within the ATAL can directly influence/affect the radiative budget at the top of the atmosphere (Vernier et al., 2015). Second, aerosol particles inside the AMA are involved in ice-cloud formation below the tropopause and in the tropical transition layer (Wagner et al., 2020; Ueyama et al., 2018). They can act as heterogeneous ice nuclei influencing cirrus cloud properties (see e.g., Liu et al., 2009; Fadnavis et al., 2013).

It was shown in modeling studies that some of UTLS particles in the AMA should include refractory components (Fadnavis et al., 2013; Lau et al., 2018; Ma et al., 2019), even though the main components are sulfuric acid/sulfates, nitric acid/nitrates, water, ammonium, and organic compounds (Appel et al., 2022; Höpfner et al., 2019).

The physicochemical properties of refractory particles (concentration, chemical composition, size, mixing state) are important for model calculations of all the processes mentioned above. In lower-stratosphere aerosol particle studies conducted outside the AMA, most refractory particles are assumed to originate from meteoric ablation, space debris, or rocket exhaust (e.g., Borrmann et al., 2010; Murphy et al., 2014; Ebert et al., 2016; Schneider et al., 2021). Apart from volcanic eruptions and wildfires, the transport of refractory particles from the Earth’s surface is considered to be of minor importance (Kremser et al., 2016).

For the ATAL, vertical transport of gases and particles from the boundary layer plays a much larger role. Based on the investigation of the transport pathways and the dynamics in the tropics, several studies conclude that the most important refractory particles in the ATAL are mineral dust, black carbon, metal (oxides), and to a smaller extent meteoric material (e.g., Lelieveld et al., 2018; Lau et al., 2018; Ma et al., 2019; Bossolasco et al., 2021). A few studies even state that the main constituents within the ATAL are mineral dust and black carbon aerosol (Bossolasco et al., 2021; Ma et al., 2019). On the Indian subcontinent, numerous anthropogenic sources exist for small refractory particles which have a low enough inertia to enable transport to such high altitudes (e.g., Lawrence and Lelieveld, 2010).

Some of the particles originating on the ground will experience scavenging and chemical processing during transport through convective mixed-phase clouds. Consequently, the chemical composition of aerosol particles in the ATAL may differ from the ground conditions (Froyd et al., 2009; Jost et al., 2017). However, experimental data on physicochemical properties of refractory particles within the ATAL and their sources are sparse (Vernier et al., 2022). This contribution aims to improve the experimental data available in order to gain a better understanding of particle transport into the ATAL.

Please be aware that different definitions of the term “refractory” and “non-volatile” are used in the literature. For example, during StratoClim 2017, Mahnke et al. (2021) used a multi-channel condensation particle counter and an optical particle spectrometer (ultra-high-sensitivity aerosol spectrometer, UHSAS-A) for detecting total aerosol densities of submicrometer-sized particles in the ATAL. In these measurements, particles which have passed through a heated tube section of about 1 m length at 270 °C within COPAS (the COndensation PArticle counting System) are classified as non-volatile.

In this way, these measurements of some of the secondary sulfate, nitrate, and organic particles could be classified as non-volatile (i.e., thermostable at up to 270 °C).
In this work, the term refractory refers to all particles that are stable during electron bombardment under high-vacuum conditions in the scanning electron microscope.

2 Methods

2.1 Flight campaign StratoClim 2017

Within the EU Seventh Framework Programme project StratoClim (Stratospheric and upper tropospheric processes for better climate predictions), a stratospheric aircraft campaign (StratoClim 2017) was conducted in July–August 2017 in Kathmandu, Nepal. One main goal was the characterization of the Asian tropopause aerosol layer (ATAL) within the 2017 monsoon anticyclone. In total, eight flights were performed by the Russian high-altitude aircraft M-55 Geophysica. The aircraft was equipped with a variety of in situ and remote sensing equipment for the measurement of particle and gas composition. The flights took place every second day during the period 27 July–10 August. Sampling details can be found in Table 1.

The flights were conducted from Kathmandu (Nepal) Tribhuvan International Airport (TIA) with a total flight time of about 31 h. Three flights (nos. 2, 4, 5) took place exclusively above Nepal. These flights were carried out along an axis parallel to the Himalaya over almost the entire east–west extension of this country. Three further flights (nos. 3, 7, 8) were performed over northeastern India. These flight patterns allowed the study of the horizontal structure of the AMA over large parts of its north–south extension, although the flight tracks did not reach beyond the anticyclone core (von Hobe et al., 2021). For more details, see Khaykin et al. (2022).

According to Bucci et al. (2020) and Brunamonti et al. (2018), the first half of the StratoClim 2017 campaign (nos. 1–4) was less affected by regional convective activity than the second half (nos. 5–8). The minimum and maximum flight height during the particle collection periods of the particle samples studied in detail are given in Table 1.

The boxplot illustration in Fig. 1b illustrates the potential temperature (\(\Theta\)) difference from the 1 Hz calculated \(\Theta\) level of the cold point tropopause (CPT) throughout the sampling period. Positive (negative) \(\Delta\Theta\) values indicate a sampling above (underneath) the CPT. The CPT potential temperature is extracted from ERA-Interim data (Weigel et al., 2021a). The potential temperature (\(\Theta\)) was calculated based on UCSE data of ambient temperature and pressure as defined by the World Meteorological Organization (WMO, 1966). For the vertical temperature and pressure distribution during the impactor collection phases in StratoClim 2017, the WMO-compliant \(\Theta\) values deviate by no more than \(\sim 1\, \text{K}\) from the results according to the refined \(\Theta\) calculation (Baumgartner et al., 2020).

2.2 Sampling technique

Particle samples were taken by the inlet line of COPAS (Curtius et al., 2005; Weigel et al., 2009; Borrmann et al., 2010) with a Y-shape manifold. According to Weigel et al. (2009), the inlet efficiency is comparable with the inlet system char-
Table 1. Parameters for particle samples (coarse: equivalent projected area diameter > 0.4 µm; fine: equivalent projected area diameter 0.04–0.4 µm).

<table>
<thead>
<tr>
<th>Flight no.</th>
<th>Date [mm/dd/yy]</th>
<th>Sample no.</th>
<th>Particle size fraction</th>
<th>Sampling start [UTC]</th>
<th>Sampling duration [min]</th>
<th>Flight height [km]</th>
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<td>2.1</td>
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<td>03:22</td>
<td>13</td>
<td>12.5–15.0</td>
</tr>
<tr>
<td>3</td>
<td>07/31/17</td>
<td>3.6</td>
<td>coarse</td>
<td>05:09</td>
<td>13</td>
<td>19.8</td>
</tr>
<tr>
<td>4</td>
<td>08/02/17</td>
<td>4.5</td>
<td>coarse</td>
<td>10:15</td>
<td>15</td>
<td>17.5–18.0</td>
</tr>
<tr>
<td>5</td>
<td>08/04/17</td>
<td>5.2</td>
<td>coarse</td>
<td>04:35</td>
<td>18</td>
<td>16.3–17.0</td>
</tr>
<tr>
<td></td>
<td></td>
<td>5.2f</td>
<td>fine</td>
<td>04:35</td>
<td>18</td>
<td>16.3–17.0</td>
</tr>
<tr>
<td>6</td>
<td>08/06/17</td>
<td>6.5</td>
<td>coarse</td>
<td>10:05</td>
<td>15</td>
<td>16.2</td>
</tr>
<tr>
<td>7</td>
<td>08/08/17</td>
<td>7.1</td>
<td>coarse</td>
<td>04:29</td>
<td>17</td>
<td>12.6–14.3</td>
</tr>
<tr>
<td></td>
<td></td>
<td>7.1f</td>
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<td>17</td>
<td>12.6–14.3</td>
</tr>
<tr>
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<td></td>
<td>7.4</td>
<td>coarse</td>
<td>05:29</td>
<td>17</td>
<td>13.0–18.0</td>
</tr>
<tr>
<td></td>
<td></td>
<td>7.4f</td>
<td>fine</td>
<td>05:29</td>
<td>17</td>
<td>13.0–18.0</td>
</tr>
<tr>
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<td>08/10/17</td>
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<td>coarse</td>
<td>09:19</td>
<td>17</td>
<td>12.0–16.0</td>
</tr>
<tr>
<td></td>
<td></td>
<td>8.1f</td>
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<td>12.0–16.0</td>
</tr>
<tr>
<td></td>
<td></td>
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<td>coarse</td>
<td>09:40</td>
<td>17</td>
<td>16.0–17.0</td>
</tr>
<tr>
<td></td>
<td></td>
<td>8.2f</td>
<td>fine</td>
<td>09:40</td>
<td>17</td>
<td>16.0–17.0</td>
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<tr>
<td></td>
<td></td>
<td>8.4</td>
<td>coarse</td>
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<td>17</td>
<td>17.0</td>
</tr>
<tr>
<td></td>
<td></td>
<td>8.5f</td>
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</tr>
<tr>
<td></td>
<td></td>
<td>8.6</td>
<td>coarse</td>
<td>11:01</td>
<td>17</td>
<td>17.1–17.8</td>
</tr>
</tbody>
</table>

* Local time is UTC + 5.45.

The orifices of the individual dual-stage impactors are 0.75 and 0.25 mm in diameter. Air velocity in the second nozzle is at the speed of sound and is thus controlling the impactor flow, which was calculated to be around 7.7 cm$^3$ s$^{-1}$. During UTLS sampling, temperature in the COPAS system varied between 272 and 290 K and pressure between 50 and 67 hPa. In these conditions in the UTLS, 50% efficiency cut-offs (calculated according to Raabe et al., 1988) are a ∼400 nm aerodynamic diameter for the first impactor stage and ∼40 nm for the second. Please note that under UTLS conditions the strictness of the impactor size discrimination is inferior compared to tropospheric conditions.

In this study, particle samples of the first impactor stage are referred to as the coarse fraction (> 400 nm) and those of the second stage as the fine fraction (∼40–400 nm).

A purge flow system is added to the Multi-MINI, which floods the tubing and the interior of the manifold prior to each sampling with ambient air to avoid any carryover of particles from previous measurements. The purge flow extends to the front of the first impaction nozzle, so the potential volume affected by carryover is minimal. The purge time (7 min) was chosen so that the tube and manifold volume could be filled at least 10 times with the current aerosol. Such a purge flow system has proven to be crucial in UTLS aerosol parti-
cle sampling in order to minimize sampling artifacts, above all a carryover from the boundary layer (Ebert et al., 2016).

During StratoClim 2017, the separate impactors were operated in the UTLS for 13–18 min each (Table 1). This sampling time was a compromise between receiving a sufficient number of refractory particles and simultaneously avoiding an overloading of the sampling substrate by the dominating semi-volatile sulfate, organic, and/or nitrate particles, which would hinder accurate electron microscopic analysis of the much smaller number of refractory particles. Furthermore, the chosen sampling time allowed us to collect six UTLS particle samples during each flight.

Particles were collected on Ni transmission electron microscopy (TEM) grids (S162-N9, Plano GmbH, Wetzlar, Germany). A total of 42 dual-stage impactor samples were collected during seven mission flights. The individual sample labels follow this sequence. For example, sample 8.5 corresponds to the fifth particle sampling in flight 8. The first flight of the campaign (27 July) was exclusively used for testing the sampling setup, including blind sampling for detection and elimination of possible particulate artifacts.

The main goal of the present study is the physicochemical characterization and source apportionment of refractory particles within the ATAL. These non-volatile particles are expected to be very small (< 500 nm) and to occur in very small numbers. Furthermore, they will often be embedded in or agglomerated with the dominating secondary sulfate, nitrate, and/or organic particles. Please note that the analyzed refractory inclusions can be much smaller than the lower cutoff diameter of the sampling device.

It should be emphasized here again that the term “refractory” is used in the present paper for all particles which are stable during electron bombardment under the high-vacuum conditions of the electron microscope in contrast to the sulfate-, organic-, and nitrate-containing particles, which evaporate quickly.

### 2.3 Sampling and analysis strategy

A major challenge of particle sampling in the UTLS region is avoiding sampling artifacts which can be caused by abrasion within the aircraft sampling line (manifold, inlet, collector) or by carryover of particles from the boundary layer.

As the particle concentrations in the UTLS are very low and refractory particles/inclusions represent the smallest share, even a small contribution of refractory artifacts will severely distort the results. In order to minimize the risk of artifacts, impactor sampling has the advantage that all collected particles are deposited within a very small area (impaction spot) on the sampling substrate. Since TEM substrates are almost particle-free before sampling and in impactor collection, the impaction spot on the sampling substrate is very small (< 1 mm²), the number of artifact particles is negligible if the number of collected particles within the impaction spot is high, and no artifact particles originate from the sampling line itself. In several procedural blank tests (e.g., complete flight 1), no particulate artifacts were detected in samples after the rinsing unit was used.

Nevertheless, for any individual ambient collection there is still the risk of artifact introduction into the samples due to individual events during installation, in flight, or during removal and transport of the particulate samples.

To minimize the risk of interpreting artifact particles as real refractory components, all samples for which errors were recorded during the sample change or afterwards during handling that could have led to possible contamination were sorted out (in total five samples).

Furthermore, samples were excluded when too few particles were found on the substrate as in this case it cannot be guaranteed that the number of potential refractory artifact particles is negligible. Based on this criterion, many of the received particle samples from flight nos. 2, 3, 4, 5, and 6 had to be excluded.

Finally, samples which were selected for analysis but with fewer than 25 refractory particles found were also excluded for final data analysis for statistical reasons.

In this way, only 17 out of 84 received particle samples were investigated in detail. Of these samples, 6 were fine stage and 11 coarse stage. In five cases it was possible to analyze fine- and coarse-stage pairs of the same sample (nos. 5.2, 7.1, 7.4, 8.1, and 8.2).

### 2.4 Characterization of refractory particles/inclusions by electron microscopy

Individual particle analysis was performed in a FEI Quanta 200 (Eindhoven, the Netherlands) FEG environmental scanning electron microscope (ESEM) equipped with an energy-dispersive X-ray (EDX) detector (EDAX, Tilburg, the Netherlands). As the instrument was operated under high-vacuum conditions only, we will refer to the method as scanning electron microscopy (SEM) throughout the paper.

A detailed analysis of the dominating sulfates and nitrate particles was not intended. Instead, we focus on the detection of refractory particles/inclusions. In a first step, many refractory particles were detected using backscattered electron (BSE) imaging by their higher average atomic number (leading to higher brightness) compared to the dominating sulfate, nitrate, and organic species.

This procedure was successful in detecting externally mixed high-Z refractory particles, but low-Z refractory particles (dominantly carbonaceous particles) and completely embedded refractory inclusions stayed undetected.

For the detection of embedded refractory inclusions, several thousands to tens of thousands volatile particles in each sample were evaporated. This could be achieved by focusing the electron beam on each individual particle for some seconds with SEM. In this way, refractory residuals were detected in 1%–2% of the (vaporized) volatile particles.

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This time-intensive analytical step (vaporization of 270 000 individual particles) was also necessary to provide a statistically significant number of refractory particles for individual particle samples. Using this approach, it was possible to analyze a significant number of refractory particles/inclusions (28–741 particles per sample, 5033 in total) in 17 flight samples (Table 2).

Additional measurements were performed in a Jeol 2100 transmission electron microscope, which was equipped with an Oxford INCA EDX system to determine the mineralogical phase of nanometer-sized Hg-rich particles.

3 Results

Even though refractory particles play an important role in many atmospheric processes in the UTLS (see Introduction) they only account for a small fraction of the total aerosol population in the ATAL. In this SEM–EDX study, heterogeneous inclusions of refractory components were observed in around 2 % of the analyzed ATAL particles.

During StratoClim 2017, simultaneous aerosol mass spectrometric measurements with the ERICA-LAMS (ERC Instrument for Chemical composition of Aerosols laser ablation mass spectrometer) instrument were conducted. The ERICA-LAMS can measure refractory and non-refractory aerosol components by laser ablation and the ionization technique followed by time-of-flight aerosol mass spectrometry. The instrument has been described before in detail (Hünig et al., 2022; Dragoneas et al., 2022) and is only briefly reviewed here. Particles enter the system through a pressure-controlled inlet (Molleker et al., 2020). The following aerodynamic lens focuses particles into a narrow beam. The particles are optically detected by scattering light when passing through two laser beams. This setup provides the particle’s time of flight and its velocity. By using a calibration with particles of known size, density, and shape, we can derive the aerodynamic diameter. The \( d_{50} \) cutoff of the ERICA-LAMS is thus limited by the optical particle detection efficiency to 180 nm. The detected and sized particles are ablated and ionized by a single triggered laser shot (wavelength = 266 nm), and the ions are guided into a time-of-flight mass spectrometer. As a result, the ERICA-LAMS provides bipolar mass spectra and size of individual particles.

In these measurements, it was determined that between 20 % and 50 % of all measured particles (by number) include refractory material, depending on altitude (Appel et al., 2022). The values between the two techniques differ, obviously, for different reasons. The refractory number abundance of 2 % determined by SEM–EDX measurements refers to refractory heterogeneous inclusions within ATAL particles only (see Sect. 2.4), while the value of 20 %–50 % derived from mass spectrometry (MS) reflects the proportion of particles that provide any refractory signal, including dissolved refractory elements, which may also play an important role (e.g., for extraterrestrial material, see Schneider et al., 2021, and discussion in Sect. 4.3).

Contrary to earlier modeling studies (Fadnavis et al., 2013; Lau et al., 2018; Ma et al., 2019; Bossolasco et al., 2021), we found that refractory particles (including desert dust) make up the minority aerosol components in the ATAL.

In all samples, the secondary components (sulfate, nitrate, and organic aerosol particles) are the dominant particle types. These mainly volatile species, however, are not in the scope of this work and are not regarded further. Detailed data on the concentration and distribution of the volatile main species of UTLS particles during StratoClim 2017 can be found in Höpfner et al. (2019), Yu et al. (2022), and Appel et al. (2022).

In total, 5033 refractory particles/inclusions were detected within the 17 selected UTLS particle samples. Based on EDX spectra and morphological criteria, these refractive particles/inclusions were classified into 10 particle groups: silicate, extraterrestrial, Ca-rich, Cl-rich, Fe-rich, Al-rich, other metals, soot, C-rich, and Hg-rich. All refractory particles, which fit in none of these groups were summarized in an eleventh “other” group. The classification criteria, secondary electron images, and EDX spectra of all groups are shown in Figs. S3 and S4 in the Supplement. Ni and C are often present in the EDX particle spectra because of the Ni grid sample substrate used. To take this into account, the classification was carried out in two steps. In the first step, all elements (except Ni) with \( Z \geq 4 \) are used for the classification of carbon-dominated particle groups (soot and C-rich). In the second classification step, all elements with \( Z \geq 8 \) are used for the classification of all other particles (Fig. S3).

It must be noted that because of the small size of most detected refractory particles (\( D_p < 100 \) nm for 50 % of all detected refractory particles), only the major elements of such small particles can be detected by EDX and minor elements are often not clearly distinguishable from the spectrum background.

The absolute number of detected particles for each group and flight sample is given in Table 2, and their relative abundances are in Fig. 2.

The smoothed relative size distributions of five refractive particle groups are plotted in Fig. 3 (all diameters are projected area diameters). Because of the limited number of particles, this distribution can only be shown for the most abundant refractory particle groups.

Approximately 30 % (1499 out of 5033) of refractory particles were Hg-rich particles with very small diameters (maximum of size distribution \( D_{P,\text{max}} = 25 \) nm). These particles occur almost exclusively in all samples of flight 8. Based on the absolute particle numbers, it is assumed that the Hg-rich particles are an additional load. Since relative abundances are compositional data (i.e., they have a constant sum), the comparison of samples from different days can be misleading. Therefore, the Hg-rich particles were excluded in the further
Table 2. Number of analyzed refractory particles/inclusions after evaporation of the volatile matrix.

<table>
<thead>
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<th>Sample number</th>
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<th>4.5</th>
<th>5.2</th>
<th>5.2f*</th>
<th>6.5</th>
<th>7.1</th>
<th>7.1f*</th>
<th>7.4</th>
<th>7.4f*</th>
<th>8.1</th>
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<th>8.5f*</th>
<th>8.6</th>
<th>Total</th>
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<tbody>
<tr>
<td>Extraterrestrial</td>
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<td>16</td>
<td>35</td>
<td>25</td>
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<td>33</td>
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* f denotes fine-stage sample 0.04–0.4 µm equivalent projected diameter.

Figure 2. Relative number abundance of refractory aerosol particles/inclusions within the 17 UTLS flight samples from seven different flights during StratoClim 2017 (f denotes fine stage).

Comparisons of the relative particle abundance in this section and are discussed separately (Sect. 4.4).

3.1 Extraterrestrial material

Mg-rich silicates as well as Mg- and Fe-rich particles were classified as extraterrestrial (chondritic composition as proposed by Rietmeijer, 1998). Following the Rietmeijer classification, the extraterrestrial group has an average relative abundance of 11% within the refractory particles (2%–40% in the individual samples). The average particle diameter ($D_p$) was 290 nm.

3.2 Silicates

All particles with Si and O as major elements but without Mg were classified as silicates. As minor elements Na, Al, K, Ca, and Fe were often found. In total, 41% of all detected refrac-
tory particles were classified as silicates. They were found in all 17 samples with an abundance between 21 % and 58 %, which means these particles belong to the most abundant refractory particle group. \(D_{\text{P, max}}\) for the silicates was found to be 120 nm (\(D_{\text{P}} = 170\) nm).

3.3 Ca-rich
All particles with Ca and O as major elements were classified as Ca-rich. Additionally, carbon was found in most of these particles as the main element.

The abundance of the Ca-rich particle group is mostly low. In only 5 of the 17 samples does this group contribute to more than 2 % of the detected refractory particles (maximum 7 %). \(D_{\text{P}}\) of the Ca-rich particles is 210 nm.

3.4 Cl-rich
Particles with Cl as the main peak were classified as Cl-rich. Besides Cl, only carbon and oxygen were detected as the main peaks. Single Cl-rich particles were detected in 15 of the 17 samples but always with very low abundances (0 %—4 %). With 510 nm, these particles have the largest \(D_{\text{P}}\) of all refractory particle groups.

3.5 Fe-rich
Particles with Fe and O as the main peak (without detectable Mg) were classified as Fe-rich. Most of these particles show no other metal peaks; only in single particles were very small peaks of Al, Cr, or Mn detected. Fe-rich particles were observed in all 17 particle samples with an average abundance of 9 % (2 %—30 %). \(D_{\text{P}}\) of Fe-rich particles was 250 nm.

3.6 Al-rich
Aluminum- and oxygen-rich particles were classified as Al-rich. These particles play only a minor role in the UTLS, and only some individual particles were detected (average 1 %, range 0 %—4 %). \(D_{\text{P, max}}\) was found at 125 nm (\(D_{\text{P}} = 250\) nm).

3.7 Other metals
Besides Al- and Fe-rich particles, about 300 refractory particles of different metals (or metal oxides) were detected and summarized in the group named “other metals”. Most of these particles were Mn-rich (78), Cr-rich (71), Zn-rich (63), Sn-rich (26), Pb-rich (20), W-rich (17), or Ti-rich (14). Additionally, a few individual particles of Ba, La, Mg, Sb, Cu, and Ce were found. \(D_{\text{P}}\) of this group was 290 nm.

3.8 C-rich and soot
All refractory particles, which show only carbon and oxygen peaks, were classified as soot or C-rich. Soot can be recognized in SEM by its typical morphology of agglomerates of spherical primary particles. As the lateral resolution of the SEM instrument is limited and the characteristic morphology described is often no longer observable for very small soot particles (\(D_{\text{P}} \leq 100\) nm), this morphological criterion cannot be applied for these particles. Thus, the abundance of the soot group represents a minimum share and the received average diameter of this group (\(D_{\text{P}} = 250\) nm) will be too high. Soot particles were detected in 13 of the 17 samples. All refractory carbon-rich particles which could not clearly be identified as soot were categorized as C-rich. C-rich particles (or non-volatile organic compounds, NVOCs) were detected in all 17 samples with an average abundance of 17 % (3 %—41 %). \(D_{\text{P}}\) of the C-rich group was 180 nm. As the TEM mesh is also excited in the case of very small particles, the C background in the spectrum is significantly increased in many of these spectra, which generally makes the clear identification of small carbonaceous (C-rich and soot) particles more difficult. Therefore, the proportion of refractory carbonaceous particles shown in this work is only a minimum proportion. Other studies have clearly demonstrated the special importance of organic particles in the ATAL (e.g., Appel et al., 2022).
4 Discussion

4.1 Sources of the refractory particles in the AMA

4.1.1 Extraterrestrial

For the classification of stratospheric particles, a broader classification than was used here is often applied. For example, in the NASA cosmic dust catalogue (Warren et al., 2011) not only chondritic compositions but also compositions strongly modified by ablative heating or melting during passage through the atmosphere are classified as “cosmic”. Following the NASA definition (which is not completely applicable to this work as it is defined for optical microscopic data of large super-µm particles), all particles from our extraterrestrial-, silicate-, and Fe-rich groups would be classified as cosmic. The NASA definition is appropriate for stratospheric particles, collected well above the tropopause where little terrestrial admixture is present. For particles collected below the tropopause and especially inside the ATAL, where terrestrial silicate material increasingly enters, an unambiguous attribution of Mg-, Si-, and/or Fe-containing particles to a terrestrial or extraterrestrial source is difficult. Thus, the classification of extraterrestrial particles in the upper troposphere used here is associated with a substantial uncertainty.

Furthermore, it has to be considered that after ablation processes, a part of the incoming extraterrestrial material (iron) may be “dissolved” in sulfuric acid droplets (discussion in Sect. 4.3). If the total extraterrestrial input should be estimated, both the dissolved fraction and the refractory particles/inclusions must be considered.

4.1.2 Silicates

The particles of the silicate group (Si- and O-rich and Mg-free) show the typical element signatures of terrestrial silicates with minor elements, for example Na, Al, K, Ca, or Fe.

Soil, coal burning and modified extraterrestrial material are the three most important sources for the silicates. Only the smallest soil particles managed transport into the UTLS, while the larger ones sedimented prior to this due to their inertia.

4.1.3 Ca-rich

As most of the Ca-rich particles also contain C and S as major elements (sometimes Mg and K as minor elements), they are interpreted as calcium carbonates (calcite/dolomite) or calcium sulfates (gypsum/anhydrite). The main sources of these Ca-rich particles are soil and industrial combustion processes such as coal and fossil fuel burning.

4.1.4 Cl-rich

Particles containing only Cl, C, and O as the main elements are interpreted as an organochlorine compound from industrial or secondary processes. An internal mixture of sea salt and organic particles is also conceivable, but it is unlikely, as Na and Mg were not detected in any of these particles.

4.1.5 Fe-rich

While the highest observed abundance of the Fe-rich group for samples from flights 2–7 was 11% (range 2%–11%), in samples of flight 8, significantly higher abundances of up to 30% (range 7%–30%) were encountered. Since flight 8 is characterized by a strong updraft (Sect. 4.3 and 4.4) and there is an increased input of terrestrial refractory particles in these samples, it is assumed that a large fraction of these Fe-rich particles stem from terrestrial sources, most likely due to industrial high-temperature processes (e.g., in forges or smelters) and burning of fossil fuels. However, the terrestrial/extraterrestrial origin of Fe-rich particles in the UTLS is a subject of current debate and an extraterrestrial origin of single Fe-rich particles cannot be excluded (Ebert et al., 2016).

4.1.6 Al-rich

Al-rich particles are supposed to be mainly aluminum oxide and originate either from Al-rich minerals or from solid rocket fuel exhausts (Mackinnon et al., 1982; Cziczo et al., 2002). Al-rich particles in the stratosphere and their impact on stratospheric ozone have been studied since the early 1970s (Hoshizaki et al., 1975; Denison et al., 1994; Jackman et al., 1998; Danilin et al., 2001).

Cofer et al. (1991) measured a bimodal size distribution of aluminum oxide particles in the Space Shuttle plume with peaks at <0.3 and 2 µm, while in this study no supermicrometer-sized Al particles were detected.

4.1.7 Other metals

Anthropogenic high-temperature processes and burning of fossil fuels are assumed to be the main sources of the diverse metal/metal oxide particles (Mn, Cr, Zn, Sn, Pb, W, Ti) detected in this study.

4.1.8 Soot

During data analysis, the assumption that the abundance of the small refractory particles/inclusions would be the same in the coarse and fine stages was disproved for the soot group. Soot is observed at higher abundances for the fine-stage particles (3%–22%) in contrast to the coarse-stage ones (0%–5%). As soot agglomerates have a very small aerodynamic diameter (small $D_p$, low density, and high pore volume) and – in contrast to all other refractory particles – they were often
not embedded within larger particles, a much higher abundance of these particles was found for the fine stage. Fossil fuel burning, industrial processes, and traffic are the most likely terrestrial sources. However, no strong enrichment of soot was observed for flights 7 and 8, which were characterized by significant updraft.

4.1.9 C-rich

The sources for the small refractory carbon-rich particles and NVOCs are nucleation processes. Sources of such nucleation particles can be either natural or anthropogenic primary emissions at the ground (e.g., fossil fuel burning) as well as secondary atmospheric processes. During the StratoClim campaign in 2017, it was detected that organics in general along with ammonium, sulfate, and nitrate are the main constituents of the ATAL (Yu et al., 2022).

4.2 Absolute concentration of refractory particles in the AMA

All abundances of refractory particles discussed so far are relative proportions. Direct estimation of absolute concentrations is associated with large uncertainties. The two main uncertainties are the poorly known collection efficiency in the airborne particle collection system under extreme and variable ambient conditions and the inhomogeneous deposition of the secondary particles (splattering) on the TEM grids. Therefore, the main discussion is focused on the relative proportions to avoid misinterpretations.

Nevertheless, an estimation can be made about the quantitative proportion of refractory particles by the known total aerosol concentration (Appel et al., 2022; Yu et al., 2022; Höpfner et al., 2019) and the determined ratio of the refractive particles/volatile particles in this work. During StratoClim 2017, a total of 5033 refractory particles/residuals within 270,000 analyzed volatile sulfate, nitrate, and organic particles was found (≈1.9 % by number). The total particulate concentration in the ATAL during StratoClim 2017 was on the order of 1–2 μg m\(^{-3}\) (Appel et al., 2022; Yu et al., 2022; Höpfner et al., 2019). Taking into account the small size of most refractory particles/inclusions, we estimate that the total concentration of refractory particles in the ATAL above Nepal during StratoClim 2017 will be <10 ng m\(^{-3}\).

This estimate is based on our definition of refractory particles and represents a lower estimate, as we do not capture all refractory residues (especially refractory organic particles) in the SEM analysis described above.

4.3 Variability in the relative abundance of refractory particles in the AMA

For the whole StratoClim 2017 campaign, the silicate group was the main refractory particle group (41 %), followed by carbonaceous particles (17 %) and extraterrestrial particles (11 %). Fe-rich particles and those classified as other metals also occur at higher relative abundance (8 % each), while only minor portions (1 %–4 % each) were determined of the Ca-rich, chloride, Al-rich, and soot groups. All percentages given above are calculated without Hg particles (see Sect. 3.1), which are discussed separately (Sect. 4.4).

Only a small fraction of refractory particles emitted at the ground reach the lower stratosphere, and only a small fraction of extraterrestrial particles reach the AMA. This can be seen from the average ratio of ground-emitted/extraterrestrial refractory particles, which drops from 16.6 in all samples collected below the tropopause (2743 ground-emitted particles/165 extraterrestrial particles) to 4.2 in all samples collected above the tropopause (549 ground-emitted particles/131 extraterrestrial particles). During StratoClim 2017, terrestrial particles dominate the composition of refractory particles clearly in the ATAL and slightly above the tropopause.

In our measurements the relative number abundance of the extraterrestrial particles decreases from 19.3 % above the tropopause to 5.7 % below the tropopause (within the refractory particles). In the same campaign, a substantial decrease from 13.3 % to 0.3 % was also detected by Schneider et al. (2021) by single-particle mass MS with the ERICA instrument. A variety of stratospheric MS measurements have detected signatures of elements from meteoric material within a significant fraction of the dominant sulfuric acid/sulfate particles (Murphy et al., 1998, 2014; Cziczo et al., 2001; Froyd et al., 2009; Schütze et al., 2017).

However, the results of Schneider et al. (2021) and our measurements cannot be compared directly due to the different size range analyzed and the different sensitivity of the two analytical techniques. Furthermore, the complete extraterrestrial material detected by ERICA will not be present in the form of heterogeneous refractory particles. Instead, after ablation, some fraction of the extraterrestrial material may be dissolved within sulfuric acid droplets (Kremser et al., 2016). For example, Murphy et al. (2014) assumed that Fe, Ni, and Mg within UTLS particles are dissolved, while Si and Al may be present in the form of refractory solid inclusions. Fe and Mg occurring in small amounts within sulfate particles collected at 2–8 km height in the vicinity of a tropopause fold over the western Pacific in 2013 were also interpreted as dissolved components originating from meteoric ablation (Adachi et al., 2022).

The presence of Fe and Mg in dissolved form would explain the fact that in our SEM study only a very small number of solid Mg-rich Fe particles were found and that the particles classified as extraterrestrial mainly consist of Mg-rich silicates. Furthermore, this partial dissolution of some meteoric elements will also strongly modify the composition of the resulting silicates (in contrast to the composition of original meteoric material). Thus, it is possible that some of the particles classified as silicates in this study will have an extraterrestrial origin. At least in the samples collected above.
the tropopause, a relevant part of the silicate particles may be residuals of ablation processes, even when it is not possible to differentiate them from terrestrial silicates by their main elemental composition.

The relative abundance of all terrestrial refractory particle groups shows quite low variability within the 17 different flight samples regardless of the specific flight altitude. This is remarkable as some samples were collected at the lowermost levels of the ATAL (e.g., sample 2.1 in 12.5–15 km), some samples well within the ATAL (just below the tropopause), and some samples significantly above the tropopause (e.g., sample 3.6 in the free stratosphere at 19.8 km). This implies that a small fraction of the refractory particles emitted at the surface of larger sulfate, nitrate, and/or organic particles. The small size is a clear indicator that these particles are formed by nucleation. The most probable terrestrial source for this of Hg-particle precursors will be the burning of fossil fuels (e.g., coal burning). Almost identical Hg-rich particles were found by Weinbruch et al. (2022) in samples directly taken at the stacks of diesel- and coal-fired power plants on Svalbard. During fossil fuel burning, mercury mainly passes into the gas phase as Hg\(^0\). However, in the original plume directly, parts of Hg\(^0\) may adsorb or be imaged (Fig. 4 right inset) and diffusion patterns could be obtained (Fig. 4 left inset). The indexing confirmed HgS (cinnabar) as the phase for all Hg-rich particles studied. Nucleation of HgS particles can only take place under reducing conditions. Such conditions can exist in soils or during fossil fuel burning at the ground but are unlikely in the UTLS. Therefore, we conclude that the small HgS particles were already formed at the ground and then entered the UTLS as primary particles with an appropriate updraft. Most probably, the small HgS particles will agglomerate quite fast on the surface of other existing particles (organics, sulfate, nitrates, or soot). This assumption is also supported by the fact that cinnabar particles were only found in samples from one flight (flight 8). If these particles had been formed in a secondary process in the lower stratosphere or near the

4.4 Mercury-rich particles

Besides the refractory particle groups discussed so far, Hg-rich particles were detected in all samples of flight 8 (10 August 2017). They were the most abundant particle group in flight 8 (1491 particles in total, on average 45 % of all refractory particles), while Hg-rich particles were almost absent in the 11 samples of flights 2–7 (in total only eight Hg-rich particles).

Mercury is of high interest due to its toxicity and its ability to undergo long-range transport in the atmosphere. Natural sources include volcanic emissions, geothermal sources, and biomass burning. South Asia is known to show enhanced anthropogenic Hg emissions (Kumari et al., 2015) from metal refining, incineration of waste, smelters, and manufacturing units, as well as coal and oil combustion (Pirrone et al., 2010).

All Hg-rich particles during StratoClim 2017 are very small with a \(D_{p,max}\) of 25 nm (Fig. 3). They agglomerate on the surface of larger sulfate, nitrate, and/or organic particles. The relative abundance of all terrestrial refractory particle groups and the flight height was observed, a dependence on meteorology of the specific flight day was seen.

Bucci et al. (2020) showed that there was an enhanced convective influence in the second part of the StratoClim 2017 campaign (flights 5–8). This also becomes visible in the increasing abundance of specific groups of refractive particles for flights 7 and 8.

For flight 7, the abundance of other metals is enhanced (on average from 3.3 % for flights 2–6 to 13.3 %). On this day various small metallic/alloy particles (dominantly Cr-, Mn-, and Zn-rich) from a specific ground region were transported into the ATAL. The ground regions from which the strongest input (or updraft) was observed indicate industrial high-temperature emissions in the Indo-Gangetic Plain–northern India as the source for these particles.

Flight 8 represents a special situation. In these samples the relative abundance of the Fe-rich and other metal group is increased in comparison to the results from all samples of flights 2–6 (Table 2). Additionally, a large share of Hg-rich particles were detected in all samples of flight 8. These Hg-rich particles were almost absent in all other flight samples.

All three refractory particle groups are supposed to originate from fossil fuel burning or industrial high-temperature processes. The specific source (or sources) is (are) located in northeastern India or southern China. A detailed discussion of sources and pathways of the Hg-rich particles is given in the following subsection.
tropopause, they would instead have been found as a general component in all flight samples of StratoClim 2017 collected near the tropopause and especially in the samples of flight 3, taken in the free stratosphere, which was not the case.

To identify a possible cinnabar source region at the ground, back trajectories were calculated based on the Chemical Lagrangian Model of the Stratosphere (ClaMS) using high-resolution ERA5 reanalysis (for details see Vogel et al., 2023a, b). The trajectories were calculated starting at the specific UTLS particle sampling time and location back to the start of the monsoon season (1 June 2017). Endpoints are shown for all trajectories that had reached the model boundary layer by then. Furthermore, in order to identify the position of the strongest uplift of air along the back trajectories, the mean location of the strongest change in potential temperature along the back trajectories (running mean over 6 h) was calculated. The results for flight sample 8.5 ( exemplary for all very similar graphs of the flight-8 samples) are shown in Fig. 5 (the results for all other samples are given in Fig. S2 in the Supplement). The frequency distribution (fd) of air mass origins shows that the possible source regions for an entry of terrestrial particles are located in the Indo-Gangetic Plain and northeastern Indian subcontinent and in southern China. Anthropogenic emissions in the Indo-Gangetic Plain are higher compared to other regions in India, caused by the dense concentration of industries as well as by the very high population density in this area. Air masses transported from the Indo-Gangetic Plain (or passing it) take up the anthropogenic emissions and are mainly uplifted along the southern edge of the Himalaya or by strong convection to UTLS altitudes.

These specific source regions as well as meteorological conditions for particle transport from the ground to the UTLS for this flight are discussed in detail by Bucci et al. (2020), who studied the impact of deep convective transport on the ATAL during the StratoClim 2017 campaign. They describe that the CO values during flight 8 decrease from around 80 ppm to below 60 ppm (see also Lee et al., 2021), and \( \text{O}_3 \) concentrations increase from around 120 ppbv to around 150 ppbv. This corresponds to an increasing mixing of stratospheric air and a decrease in the convective influence from 100 % to around 50 %. For the HgS particles of flight 8, this could indicate a stratospheric source. However, Bucci et al. (2020) also show in a detailed analysis that flight 8 captured some very intense overshoots and convective outflows from exceptionally fast (less than an hour) and localized plumes. These events are only visible as sharp peaks in the CO concentrations and happened on such a small temporal and spatial scale that they will not be visible in, for example, most satellite data (with coarser resolution). These outflows may
be responsible for the fast convective transport of HgS particles from ground sources during flight 8.

India and China are among the largest emitters of atmospheric mercury in the world (Jetashree et al., 2022; Wu et al., 2006). For example, the Indian Jharia region, which is located directly within the identified source region, with the Jharia coal fields (23.75°N, 86.42°E), produces most of India’s coal. Jharia coal mines are India’s most important storehouse of prime coke coal and consist of 23 large underground and nine large open-cast mines. Furthermore, there have been persistent smoldering coal field fires in this region for more than a century. The Hg concentration of the coal is higher than the world average, and the coal field fires are known to be a source of Hg pollution in the mining area (Raj et al., 2017).

According to Nádudvari et al. (2022), HgS particles are found above underground coal deposit fires and thermally affected waste dumps from hard coal mining due to the reducing environment in the bituminous surface layer. Therefore, one likely possibility is that the detected cinnabar particles in flight 8 originate from underground coal fires in northern India or southern China and/or from industrial coal burning. Cinnabar may have formed either under or at the ground under reducing conditions and have been instantly adsorbed on the surface of other aerosol particles transported into the UTLS by a strong updraft, maybe during the intense overshoot events observed during flight 8.

If there were a case of strong convective input of coal combustion (at or under the ground) during flight 8, an increased level of soot, CO, and other Hg particles such as HgCl₂ (Srivastava et al., 2006; Peng et al., 2021) would also be expected, which is not the case. While the observations of Bucci et al. (2020) can explain the CO values, the non-discovery of other Hg species and the lack of clearly increased soot abundance contradict a ground source being the source of the determined HgS particles.

In all the previous literature, a stratospheric origin of particulate Hg³¹ components in the UTLS was inferred. Murphy et al. (1998) discussed Hg-rich particles from UTLS and stratospheric aircraft experiments. Hg was detected in a large number of MS particle spectra during one flight leg near the tropopause south of Houston to 10°N. In contrast, no mercury was found during flights in a remote continental surface site (Idaho Hill, Colorado) and a remote marine surface site (Kennaook/Cape Grim, Tasmania). During different aircraft campaigns in the tropics and middle latitudes, Murphy et al. (2006) detected Hg-containing particles close to the tropopause, while no Hg-containing particles were detected below 5 km height. They concluded that particulate Hg³¹ most likely originates from oxidation of gaseous Hg⁰ in the lower stratosphere and not from a primary terrestrial Hg-particle source. This conclusion is also supported by Lyman and Jaffe (2012). During aircraft measurements at 6–7 km altitude, Hg³¹ was positively correlated with stratospheric tracers (ozone and potential vorticity), indicating that Hg³¹ increased with increasing stratospheric influence.

The Hg chemistry in the atmosphere is quite complex and the subject of current scientific research. Excellent overviews of possible mercury oxidation pathways in the atmosphere are, for example, Schroeder and Munthe (1998), Holmes et al. (2010), Shah et al. (2016), Obrist et al. (2018), and Lyman et al. (2020). They summarize the possible role of OH, O₃, bromine, photochemistry, and aqueous-phase reduction. Gratz et al. (2015) reported results from the 2013 Nitrogen, Oxidants, Mercury and Aerosol Distributions, Sources and Sinks campaign, which supported the role of bromine as the dominant oxidant of mercury in the upper troposphere and the importance of subtropical anticyclones for the formation of Hg³¹.

This is worth highlighting because during the StratoClim flight campaign in 2017, Adcock et al. (2021) detected enhanced bromine values in the UTLS.

Up to now it has been concluded that the source of Hg³¹ particles in the UTLS must be completely attributed to stratospheric oxidation of gaseous Hg⁰. The observation of HgS particles in the UTLS during flight 8 shows the possibility that quick convective outflows of very fast and localized plumes could cause direct transport of HgS particles in the UTLS region.

This transport from the ground could also be responsible for the Hg particles described in Murphy et al. (2006) for a lower-stratospheric sample. Even though no mineralogical-phase information is given by Murphy et al. (2006), their observed Hg-rich particles seem to be identical to the HgS (cinnabar) particles observed by us. The size (10–20 nm in diameter), mixing state (attached to sulfate particles), Hg : S ratio, and beam sensitivity in STEM (scanning transmission electron microscope; volatilization under STEM conditions in a few seconds) were identical to those of our cinnabar particles found during StratoClim. Thus, the presence of HgS particles in the UTLS seems not to be limited to the specific conditions within the ATAL during the 2017 monsoon anticyclone.

5 Conclusion

It was shown that within the 2017 monsoon anticyclone there is a predominantly terrestrial input of refractory particles into the ATAL.

In contrast to prior modeling studies, we found that refractory particles (including desert dust) play only a minor role in the total composition of aerosol particles within the ATAL. In SEM measurements, about 2% by number of the typical ATAL particles (main components: ammonium, sulfate, nitrate, and organics) show visible inclusions/agglomerates of refractory particles. The main components within the refractory particles were silicates and NVOCs. In addition, Fe-rich particles, other metal-rich particles (Mn, Cr, Zn), and
extraterrestrial particles were found, as well as some small quantities of soot and Ca-, Cl-, and Al-rich particles.

In general, most refractory particles found are very small. The maximum of the dN/d\(\log D_p\) distribution is \(\sim 125\) nm. This also means that, besides some very small terrestrial soil particles, nucleation processes are the predominant source. For most refractory particles these are mainly anthropogenic combustion processes (coal burning, biomass burning, industrial processes). For the NVOCs secondary atmospheric processes are additionally important.

The variability in the relative number abundance of individual ground-emitted refractory particle groups was quite low during StratoClim 2017 for most sampling days and at different flight altitudes between \(\sim 12–19\) km. This suggests that there was generally a roughly uniform background composition for the terrestrial refractory particles in the ATAL and also just beyond the tropopause (lower stratosphere). Extraterrestrial refractory particles play a larger role above the tropopause; within the ATAL their relative abundance is low.

During flights 7 and 8, additional refractory particles were detected. These particles originated from an additional input from special anthropogenic ground sources and were rapidly transported into the UTLS under enhanced convective influence. The ground regions from which the strongest input (or updraft) was observed indicate industrial emissions in the Indo-Gangetic Plain for flight 7, which additionally introduced various metals and metal oxides into the ATAL here.

The exact source of the Hg-rich particles found in flight 8 remains unclear. As previously described in the literature, they could have originated in the lower stratosphere. However, the TEM characterization of cinnabar (HgS) suggests a ground source such as coal combustion or underground coal fires.

Data availability. The complete SEM original data set is available for the community and can be accessed by request to Martin Ebert (mebert@geo.tu-darmstadt.de) of the Technical University of Darmstadt.

Supplement. The supplement related to this article is available online at: https://doi.org/10.5194/acp-24-4771-2024-supplement.

Author contributions. RW: potential temperature, CPT data analysis and classification of particle origin, data discussion and interpretation; SW: manuscript data interpretation and discussion; LS: SEM data evaluation, interpretation and discussion; KK: data evaluation and classification, interpretation and discussion; SL: TEM measurements, TEM data interpretation and discussion; FK: classification, comparison, and interpretation of refractory (SEM and MS) particle data; FP: field experiment data, data discussion and interpretation (meteorology, ATAL, UTLS); GG: field experiment data, data discussion and interpretation (meteorology, ATAL, UTLS); BV: back-trajectory analysis, air mass origin discussion; SB: manuscript data interpretation and discussion.

Competing interests. The contact author has declared that none of the authors has any competing interests.

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