Atmos. Chem. Phys., 23, 4559–4576, 2023 https://doi.org/10.5194/acp-23-4559-2023 © Author(s) 2023. This work is distributed under the Creative Commons Attribution 4.0 License.





Measurement report: Molecular-level investigation of atmospheric cluster ions at the tropical high-altitude research station Chacaltaya (5240 m a.s.l.) in the Bolivian Andes

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Received: 28 October 2022 – Discussion started: 14 December 2022 Revised: 6 March 2023 – Accepted: 21 March 2023 – Published: 14 April 2023

Abstract. Air ions are the key components for a series of atmospheric physicochemical interactions, such as ion-catalyzed reactions, ion-molecule reactions, and ion-induced new particle formation (NPF). They also control atmospheric electrical properties with effects on global climate. We performed molecular-level measurements of cluster ions at the high-altitude research station Chacaltaya (CHC; 5240 m a.s.l.), located in the Bolivian Andes, from January to May 2018 using an atmospheric-pressure-interface time-of-flight mass spectrometer. The negative ions mainly consisted of $(H_2SO_4)_{0-3} \cdot HSO_4^-$, $(HNO_3)_{0-2} \cdot NO_3^-$, SO_5^- , $(NH_3)_{1-6} \cdot (H_2SO_4)_{3-7} \cdot HSO_4^-$, malonic-acid-derived, and CHO / CHON $\cdot (HSO_4^- / NO_3^-)$ cluster ions. Their temporal variability exhibited distinct diurnal and seasonal patterns due to the changes in the corresponding neutral species' molecular properties (such as electron affinity and proton affinity) and concentrations resulting from the air masses arriving at CHC from different source regions. The positive ions were mainly composed of protonated amines and organic cluster ions but exhibited no clear diurnal variation. H_2SO_4 –NH₃ cluster ions likely contributed to the NPF process, particularly during the wet-to-dry transition period and the dry season, when CHC was more impacted by air masses originating from source regions with elevated SO₂ emissions. Our study provides new insights into the chemical composition of atmospheric cluster ions and their role in new particle formation in the high-altitude mountain environment of the Bolivian Andes.

1 Introduction

Air ions regulate the electrical properties of the atmosphere by serving as carriers of electrical charges (Williams, 2009). They also play an important role in atmospheric chemistry by participating/catalyzing ion-molecule reactions and ion-induced new particle formation (NPF; Hirsikko et al., 2011). The formation of tropospheric ions is initiated through simple-structured ions, such as O^+ , N_2^+ , O^- , and O_2^- , mainly from radioactive decay in the soil (e.g., radon and gamma radiation), thunderstorm activity (lightning), and galactic cosmic rays (GCRs). These ions can transfer their charges to other compounds, leading to the subsequent production of an assortment of ions, such as the bisulfate ion (HSO_4^-) , nitrate ion (NO_3^-) , hydronium ion (H_3O^+) , and ammonium ion (NH₄⁺; Smith and Spanel, 1995; Hirsikko et al., 2011). Depending on their sizes, air ions are usually classified into cluster ions (diameter ≤ 1.6 nm) that are charged molecules or molecular clusters and charged particles (diameter > 1.6 nm; Hirsikko et al., 2005, 2011; Komppula et al., 2007).

Cluster ions exist almost always in the troposphere and can undergo frequent ion-molecule reactions during their lifetime (~ 100 s; Manninen et al., 2010; Hirsikko et al., 2011). Their chemical composition, in addition to the initial ionization, also depends on the concentrations of the parent neutral species (Eisele, 1986). Bianchi et al. (2017) showed that the diurnal cycle of negative organic ions followed the variations of their neutral molecules' concentrations in a boreal forest, since the higher concentrations of neutral molecules would result in a larger probability of them being charged. Moreover, molecular properties of the neutral species, such as electron affinity (EA) and proton affinity (PA), are also important for determining cluster ion composition. Cluster ions derived from molecules with higher EA (e.g., HSO₄ and NO₃) or PA (e.g., trimethylamine $-C_3H_9N$ – and pyridine $-C_5H_5N$) tend to obtain the ambient negative or positive charge, respectively (Ferguson and Arnold, 1981; Hirsikko et al., 2011). Because of the strong EA or PA, it is almost unlikely that the ions derived from those molecules will further transfer their charges to other neutral compounds via ionmolecule reactions. Thus, these negative $(HSO_4^- \text{ and } NO_3^-)$ and positive $(C_3H_{10}N^+ \text{ and } C_5H_6N^+)$ ions are usually more abundant than other ions in the atmosphere (Eisele, 1986; Ehn et al., 2010; Bianchi et al., 2017; Frege et al., 2017). In contrast, charge transfer occurs more easily for ions derived from neutral species of lower EA or PA.

Atmospheric cluster ions can contribute to NPF via ioninduced nucleation (Yu, 2010). Since the discovery of this mechanism in the first cloud chamber study in the early 1900s (Wilson, 1911), ion-induced nucleation has been known to be an important source of atmospheric aerosol particles. Recently, a series of chamber studies conducted at the CLOUD (Cosmics Leaving Outdoor Droplets) facility at CERN (the European Centre for Nuclear Research) showed that aerosol-nucleation rates are substantially enhanced in the presence of some specific cluster ions, such as sulfuric acidammonia (H₂SO₄–NH₃) cluster ions (Kirkby et al., 2011; Schobesberger et al., 2015), pure H₂SO₄ cluster ions (Kirkby et al., 2011), and organic cluster ions (Kirkby et al., 2016). Field measurements have also suggested an important role of atmospheric ions in ion-induced nucleation (Manninen et al., 2010; Hirsikko et al., 2011; Rose et al., 2018; Jokinen et al., 2018; Yan et al., 2018; Beck et al., 2021). Among them, the onsets of high-altitude NPF events, compared to those occurring in the lower troposphere, are often associated with more abundant cluster ions (Lee et al., 2003; Venzac et al., 2008; Boulon et al., 2010; Sellegri et al., 2019). Such increases are due to the higher GCR intensity and lower condensation sink (CS) in the high-altitude regions. As a result, potentially larger contributions of cluster ions to aerosol formation would be expected (Smith and Spanel, 1995; Hirsikko et al., 2011).

However, the molecular-level understanding of ambient cluster ions and their influence on NPF in high-altitude environments (in the troposphere) is still very limited. Two mountaintop studies in the Alps show that, depending on the air mass origins, NPF could be triggered by sulfuric acid-ammonia clusters or nitrate (or sulfuric acid) clustering with highly oxygenated organic molecules (Bianchi et al., 2016; Frege et al., 2017). Another study in the Himalayas found that NPF was mainly driven by organic vapors of biogenic origin (Bianchi et al., 2021). Recently, frequent and intensive NPF events were observed at the high-altitude research station Chacaltava (CHC; 16.3505° S, 68.1314° W; 5240 m a.s.l.) located in the Bolivian Andes (Rose et al., 2015a), but the exact mechanism and the role of cluster ions in the aerosol-nucleation process remain unclear. Therefore, a detailed investigation of cluster ions at CHC, including their molecular composition, temporal variation (diurnal and seasonal), and source regions, is needed in order to understand their role in atmospheric processes such as NPF in the study regions.

Here we present measurements of atmospheric ions from January to May 2018 at CHC. The dataset is part of the Southern hemisphere high ALTitude Experiment on particle Nucleation And growth (SALTENA) field experiment campaign (Bianchi et al., 2022). During the study period, the sampled air masses originated from various source regions, such as the Amazon basin to the east and the Altiplano and the Pacific Ocean to the west (Fig. 1a; Aliaga et al., 2021). Temporal evolutions (diurnal and/or seasonal variations) of both negative- and positive-ion composition are investigated, and their potential connections with source regions and NPF are discussed. Our study thus adds important observational information on a better understanding of atmospheric ions



Figure 1. (a) True-color satellite image showing the location of CHC (blue circle) and its surrounding area. The yellow line presents the boundary of the Amazon basin. Red triangles denote the volcanoes in this area. (b) A zoomed-in true-color satellite image showing the distance between CHC and the La Paz–El Alto metropolitan area (orange circle). Image sources: Esri, DigitalGlobe, GeoEye, i-cubed, USDA FSA, USGS, AEX, Getmapping, Aerogrid, IGN, IGP, swisstopo, and the GIS User Community.

and provides new insights into their role in high-altitude NPF in the troposphere of the Bolivian Andes.

2 Methods

2.1 Measurement site description

The high-altitude research station CHC is \sim 140 m below the summit of Mount Chacaltaya (5380 m a.s.l.) with an open view to the south and west (Andrade et al., 2015). The La Paz–El Alto metropolitan area (with 1.7 million inhabitants) is \sim 1–1.6 km lower (in altitude) and \sim 15 km south of CHC (Fig. 1b). The seasonal meteorological conditions at CHC depend on the cycle between the wet (November to March; wetto-dry transition period in April) and dry (May to September; dry-to-wet transition period in October) seasons driven by large-scale tropical circulation (Rose et al., 2015a; Bianchi et al., 2022). This pattern also affects the source regions of air masses arriving at CHC (Aliaga et al., 2021). Additionally, due to the strong diurnal cycle of the planetary boundary layer (PBL) height and the thermally induced winds in the mountainous terrain, CHC is often affected by a polluted PBL transported from the La Paz-El Alto metropolitan area during daytime (Wiedensohler et al., 2018), whereas at night CHC is located in the residual layer or tropical free troposphere (Rose et al., 2015b; Chauvigné et al., 2019).

2.2 Instrumentation

All the instruments involved in this study were installed in a temperature-controlled measurement room (at ~ 25 °C). All data are reported in local time (UTC – 4).

2.2.1 Measurements of atmospheric cluster ions

The composition of cluster ions was measured by an atmospheric-pressure-interface time-of-flight mass spectrometer (APi-TOF, Aerodyne Research Inc. & Tofwerk AG). The APi-TOF mass spectrometer consists of an atmospheric-pressure-interface (APi) module and a time-offlight (TOF) mass spectrometer. The APi module allows the instrument to sample ions from ambient air directly. The positive or negative ions within the sampled airflow are focused and guided by two quadrupoles and an ion lens, with a gradually decreasing pressure (from atmospheric pressure to $\sim 10^{-4}$ mbar), before entering the TOF mass spectrometer ($\sim 10^{-6}$ mbar). A more detailed description of this instrument is given in Junninen et al. (2010). In this study, ambient air was sampled through a ~ 1.5 m stainless-steel tube with a total sample flow of 14 standard liters per minute (SLPM) to ensure laminar flow during sampling, and 0.8 SLPM of the total flow entered APi-TOF.

During the wet season, APi-TOF was first operated in negative mode to measure negative cluster ions (January) and then switched to positive mode to measure positive cluster ions (February to March). During the wet-to-dry transition period (April) and the dry season (May), the instrument was changed back to negative mode to investigate the potential seasonality of negative ion composition. It is important to note that, similarly to Frege et al. (2017), for better characterization of the connection between cluster ions and NPF, we only included the ion data observed under the cloud-free condition in this study (to avoid influence from, e.g., light-ning activity). CHC was considered to be affected by clouds when relative humidity (RH) exceeded 95 %, as suggested by a previous study at the same location (Rose et al., 2015a). From January to May 2018, the proportions of cloud-free hours in the total measurement time were 72 %, 78 %, 79 %, 98 %, and 98 %, respectively (Fig. S1 in the Supplement).

2.2.2 Measurements of H₂SO₄ and oxidized organic molecules

Concentrations of H₂SO₄ and oxidized organic molecules (OOMs) were measured using a nitrate ion (NO_3^-) -based chemical-ionization atmospheric-pressure-interface time-offlight mass spectrometer (CI-APi-TOF, Aerodyne Research Inc. & Tofwerk AG; Jokinen et al., 2012). The instrument is a combination of APi-TOF and a CI unit, which has been widely used to measure H₂SO₄ in the atmosphere (Jokinen et al., 2012; Bianchi et al., 2016; Zha et al., 2018). In this study, a soft X-ray source (L9490, Hamamatsu) was used to charge nitric acid (HNO₃) in a sheath flow of 20 SLPM to produce the reagent ion, NO_3^- . H₂SO₄ and OOMs in the sample flow (10 SLPM) were then charged by either proton transfer or the formation of an adduct with the reagent ion during the \sim 200 ms residence time in the CI unit. A calibration factor of $1.5 \times 10^{10} \text{ cm}^{-3}$ for H₂SO₄ was determined (with sampling loss corrected) according to the approach by Kürten et al. (2012). The same calibration coefficient was adopted for determining OOM concentrations in this study, which could result in an underestimation of their concentrations due to a lower charging efficiency of OOMs than H_2SO_4 by NO_3^- (Hyttinen et al., 2015).

2.2.3 Auxiliary measurements

The number concentration and size distribution of atmospheric ions and neutral particles were measured with a neutral cluster and air ion spectrometer (NAIS, Airel Ltd.; Mirme and Mirme, 2013). The instrument can detect naturally charged air ions and total particles with mobility diameters from 1.4–50 and 3–50 nm, respectively. It is important to note that the size range of the NAIS was corrected based on a side-by-side comparison with an updated version of NAIS (designed for measurements under low-pressure environments; Mirme et al., 2010) at CHC. Thus, the size range of detection was slightly different from the traditionally reported ranges (i.e., 0.8–42 and 2–42 nm, respectively; Manninen et al., 2010). The details of the instrument used can be found in Rose et al. (2017). Particle number size distributions between 10 and 500 nm were measured by a mobility particle size spectrometer (MPSS; Wiedensohler et al., 2012), and the data were used for calculating the CS, which represents the loss rate of condensing vapors and cluster ions on preexisting particles (Kulmala et al., 2001).

Meteorological parameters, such as temperature, RH, and global radiation, were also measured simultaneously at CHC. Detailed descriptions can be found in Bianchi et al. (2022).

2.3 Simulation of air mass origin and history

To understand the source regions and transport pathways of the air masses arriving at CHC, we used the results of air mass history analysis obtained from FLEXible PARTicle dispersion-Weather Research and Forecasting model (FLEXPART-WRF) simulations described in Aliaga et al. (2021). In brief, a Lagrangian transport and dispersion model (FLEXPART-WRF; version 3.3.2; Brioude et al., 2013) was used to calculate the air mass history during the campaign period. The backward simulation was driven by the high-spatial-resolution and high-temporal-resolution meteorological output from the WRF model (version 4.0.3; Skamarock et al., 2019). In the simulation, 20000 particles were continuously released every hour from a 10 m-deep layer (0-10 m a.g.l.) at CHC and traced back in the atmosphere for 96 h. The output of the FLEXPART-WRF is the sourcereceptor relationship (SRR, in seconds), which is calculated for each geographical grid cell included in the simulation. The SRR value depends on the particle's residence time and the number of particles in the output grid cells. Clustering analysis was conducted by applying a series of pretreatments (e.g., log-polar grid transformation and grid cell preprocessing) and a k-means clustering algorithm (Lloyd, 1982) to the calculated SSR dataset (see Aliaga et al., 2021, for more details).

2.3.1 Major air mass pathways

Six air mass pathways (PWs) representing air masses arriving at CHC were determined from the clustering analysis. They are named based on their clock positions from CHC (e.g., 03_PW indicates the pathway with its centroid located in the three-o'clock direction (east, 90°) of CHC, Fig. 2a). The characteristics of these air mass pathways, such as source region, transport distance, and transport time, were distinct from each other (Table 1). A detailed description of the air mass pathways and their characteristics can be found in Aliaga et al. (2021).

The influence of each air mass pathway on CHC varied with time and was estimated by its SRR percentage (SRR[%]_{pathway}) as in Eq. (1):

$$SRR[\%]_{pathway} = \frac{SRR_{pathway}}{SRR_{total}} \times 100,$$
 (1)



Figure 2. Influence of the six air pathways on CHC from January to May 2018. (a) Horizontal profile of the air mass pathways, adapted from Aliaga et al. (2021). (b) Frequency of the representative periods for each pathway (the highest 10% of their corresponding SRR[%]_{pathway}) in different months.

Table 1. Overview of the six air mass pathways extracted from Aliaga et al. (2021).

Pathway	Direction to CHC	Representative source region	Transport distance (km)* Median (25 %–75 %)	Transport time (h) Median (25 %–75 %)
03_PW	East	Amazon basin and eastern/southeastern lowlands	518 (413–608)	51 (45–57)
05_PW	South and southeast	Southeastern lowlands and southern Altiplano	428 (303–567)	45 (36–52)
07_PW	Southwest	Pacific Ocean, coastal area, western Altiplano, and La Paz–El Alto	721 (577–896)	54 (45–61)
08_PW	West	Western Altiplano and Lake Titicaca, coastal area	238 (198–279)	36 (29–43)
11_PW	North and northwest	Amazon basin, western Altiplano, coastal area	465 (326–563)	53 (46–59)
12_PW	North	Amazon basin	76 (49–95)	27 (21–33)

* Distance between CHC and each pathway's center point (see Fig. 2a).

where $SRR_{pathway}$ and SRR_{total} are the residence times of a specific air mass pathway and the total (96 h = 345 600 s) in the simulation, respectively.

2.3.2 Identification of representative periods for each air mass pathway

Air mass history analysis shows that the air sampled at CHC was typically a mixture of multiple pathways. Thus, the cluster ion composition observed during the study period was often influenced by multiple source regions concurrently. To characterize the influence of every single pathway on cluster ion composition, periods when an air mass pathway exerted its largest impact on CHC (the highest 10% of its SRR[%]_{pathway} values; Fig. 2b) during the whole study period are identified as the representative periods of 03_PW

(covering, e.g., the Amazon basin) are more frequently seen during the wet season (highest in January), whereas 08_PW (covering, e.g., the Altiplano region) has most of its representative periods during the dry season (highest in May). Note that the SRR[%]_{pathway} of any individual pathway rarely reached 40% due to the relatively short study period (see Fig. S2), and thus the representative periods cannot be directly identified via SRR[%]_{pathway} values. In contrast, such representative periods were determined by using a certain threshold (e.g., >70%) in a previous study at CHC, which was based on a more than 6-year dataset (Koenig et al., 2021).



Figure 3. Correlation between the TIC measured by APi-TOF and condensation sink, colored by the fraction of small ions (defined as concentrations of ions with diameter < 2 nm to the total ion concentrations) determined from NAIS data. Data are shown at a time resolution of 1 h.

3 Results and discussion

3.1 Variation in total ion count

During the study period, the total ion count (TIC) observed by APi-TOF at CHC varied between < 10 and 100 counts per second (cps; Figs. 3 and S3). Similar variations in the TIC were also observed in the long-term cluster ion measurements at the high-altitude station Jungfraujoch in Switzerland (JFJ; 3454 m a.s.l.; Frege et al., 2017) and were attributed to the seasonal changes in ion precursor and sink. Like the winter-summer seasonality of the JFJ, CHC and its adjacent mountain areas are frequently covered by snow in the wet season and are mostly free of snow in the dry season (Bianchi et al., 2022; Koenig et al., 2021). Thus, in contrast with the generally stable GCR flux (primarily controlled by the decadal-scale solar cycle; Shuman et al., 2015), a reduced radioactive decay from the soil and a lower ion production rate could be expected at CHC in the wet season than in the dry season.

However, the TIC measured by APi-TOF was significantly higher in the wet season $(41 \pm 23 \text{ cps}, \text{mean} \pm \text{standard} \text{ de$ $viation})$ and the wet-to-dry transition period $(56 \pm 32 \text{ cps})$ than in the dry season $(14 \pm 11 \text{ cps}; \text{ Fig. S3})$. Considering the slight negative correlation (Pearson's correlation coefficient, R: -0.41; Fig. 3) between the TIC and CS (representing the loss rate of condensing vapors and cluster ions on preexisting particles; Kulmala et al., 2001), the observed TIC fluctuation may be related (at least partially) to the varying CS ($\sim 1 \times 10^{-4} \text{ s}^{-1}$ in the wet season to $\sim 5 \times 10^{-2} \text{ s}^{-1}$ in the dry season). Moreover, the cluster ions measured by APi-TOF usually account for only a small fraction of the total atmospheric ions (Rose et al., 2018). Changes in the fraction of small ions in total atmospheric ions can potentially lead to a fluctuation in TIC (Frege et al., 2017). This is illustrated in Fig. 3, where a smaller TIC determined from APi-TOF is associated with a lower fraction of smaller ions (< 2 nm) observed by NAIS (mostly cluster ions). However, for better characterization of the influences of different ion compositions on CHC and their diurnal and seasonal relative changes, we normalized the observed ion signal to the TIC for APi-TOF measurements.

3.2 Negative ions

3.2.1 Main negative ions and their diurnal variation

A number of negative ions were consistently observed at CHC throughout the study period (i.e., in the wet, wet-to-dry transition, and dry seasons). Based on the chemical composition of the observed negative ions, we classified them into eight groups as follows: sulfuric acid $((H_2SO_4)_{0-3} \cdot HSO_4)$, nitric acid $((HNO_3)_{0-2} \cdot NO_3)$, SO₅, sulfuric acid-ammonia $((NH_3)_{1-6} \cdot (H_2SO_4)_{3-7} \cdot HSO_4^-)$, malonic-acid-derived (MA-derived, including C₃H₃O₄⁻, $C_3H_4O_4 \cdot NO_3^-$, and $C_3H_4O_4 \cdot HSO_4^-$), oxidized organic molecules $(CHO / CHON \cdot (HSO_4^- / NO_3^-))$, others (other identified negative ions, such as IO_3^-), and unidentified ions. The campaign-average diurnal variations of these eight negative ion groups are shown in Fig. 4. Different diurnal patterns of each negative ion group were observed, mainly due to changes in concentrations of the parent neutral species and their unique physicochemical properties (e.g., the EA and PA of molecules; Ferguson and Arnold, 1981; Bianchi et al., 2017; Hirsikko et al., 2011).

 $(HNO_3)_{0-1} \cdot NO_3^-$ and $(H_2SO_4)_{0-3} \cdot HSO_4^-$ were among the highest in the signal of all negative ion groups in all seasons, making up 37 % (whole day) and 20 % (daytime: 07:00– 19:00 and hereafter) of negative ions at CHC during the study period, respectively. The EA of their neutral molecules (HSO₄ and NO₃) is higher than that of most of the neutral species in the atmosphere and thus hinders the direct electron transfer from HSO₄⁻ and NO₃⁻ to other molecules through ion-molecule reactions (Ferguson and Arnold, 1981). As a result, these ion groups were found to dominate negative cluster ions at CHC and other locations, such as a number of remote sites in the United States (Eisele, 1986), a boreal forest site in Finland (Ehn et al., 2010; Bianchi et al., 2017), and the JFJ in Switzerland (Frege et al., 2017).

Distinct diurnal patterns were observed for the $(HNO_3)_{0-1} \cdot NO_3^-$ and $(H_2SO_4)_{0-3} \cdot HSO_4^-$ ion groups. $(HNO_3)_{0-1} \cdot NO_3^-$ exhibited a relatively flat diurnal pattern (see Fig. 4), with similar fractions at daytime (35%) and nighttime (19:00–07:00; 39%). Such a diurnal pattern could result from the high EA of the NO₃ molecule (4.01 eV and an additional ~ 1 eV per HNO₃; Ferguson and Arnold, 1981) and its relatively abundant (usually several parts per billion by volume) parent neutral species (e.g., HNO₃ and N₂O₅) with multiple sources in the atmosphere (e.g., anthropogenic emission and lightning; Martin et al., 2007). In contrast,



Figure 4. Diurnal variation of the negative ions (fraction of TIC), CS, and neutral H₂SO₄ concentrations at CHC, averaged over the periods when negative ions were measured (i.e., January, April, and May 2018).

 $(H_2SO_4)_{0-3}$ ·HSO₄⁻ exhibited a strong diurnal variation. While the fraction of $(H_2SO_4)_{0-3}$ HSO₄ remained low (2%) during nighttime, it started to increase after sunrise (shortly after 07:00) and reached a maximum (30%) at around 10:00. Despite an EA comparable to that of the NO₃ molecule (4.75 eV for HSO₄; Wang et al., 2000), the strong diurnal variation of $(H_2SO_4)_{0-3} \cdot HSO_4^-$ is a result of the photochemical production of neutral H₂SO₄. The influence of neutral H_2SO_4 on $(H_2SO_4)_{0-3}$ $\cdot HSO_4^-$ is indicated by their similar diurnal patterns (R: 0.52; see Fig. S4a). Similarly, a higher level of $(NH_3)_{1-6} \cdot (H_2SO_4)_{3-7} \cdot HSO_4^-$ was only observed with the presence of abundant $(H_2SO_4)_{0-3}$ •HSO⁻₄ during daytime. It is also important to note that the decreases in $(H_2SO_4)_{0-3} \cdot HSO_4^-$ and $(NH_3)_{1-6} \cdot (H_2SO_4)_{3-7} \cdot HSO_4^$ at around noontime (12:00; see Fig. 4) coincided with an enhanced CS, indicating the influence of a higher ion sink in addition to the decrease in neutral H₂SO₄ concentrations (Boulon et al., 2010; Frege et al., 2017).

The MA-derived ion group is distinguished from the CHO / CHON•(HSO₄⁻ / NO₃⁻) ion group by its abundance in total organic ions. This group of ions, mainly composed of $C_3H_3O_4^-$, is formed from the deprotonation of malonic acid, with a higher EA (~4.60 eV) than that of the NO₃ molecule (4.01 eV; Ravi Kumar et al., 2005). Malonic acid, similarly to HNO₃, has multiple origins in the atmosphere, such as primary and secondary anthropogenic sources, biogenic sources, and the degradation of larger organic com-

pounds (Braban et al., 2003). It is also one of the main dicarboxylic acids which make up a substantial fraction of total carbon in aerosol particles (Kawamura and Bikkina, 2016). However, the concentration of gas-phase malonic acid is less well documented and is estimated to be in the range of 10^7 to 10^9 cm⁻³, up to 3 orders of magnitude higher than the typically reported ambient H₂SO₄ concentration (Fang et al., 2020). Thus, the fraction of the MA-derived ion group was high during nighttime (12 %; see Fig. 4). In contrast, its fraction decreased significantly (to 5 %) during daytime due to the increase in (H₂SO₄)₀₋₃·HSO⁻₄, which has an even higher EA (4.75 eV for HSO₄; Wang et al., 2000).

The $CHO/CHON \cdot (HSO_4^-/NO_3^-)$ ion overall molecular formula of group, with an $C_{2-15}H_{2-26}O_{2-13}N_{0-2}$ ·NO₃ / HSO₄, constituted a significant fraction of negative ions (31%) at CHC. These organic ions are formed through the adduction between primary charge carriers, such as HSO_4^- and NO_3^- , and neutral OOMs. These OOMs are likely the oxidation products of the volatile organic compounds (VOCs) from the Amazon, the Altiplano, and the adjacent La Paz-El Alto metropolitan area. Their chemical composition is potentially affected by the changing air pathways covering different VOC source regions (Aliaga et al., 2021) and by the different conditions during daytime/nighttime due to the evolution of different atmospheric layers (Beck et al., 2022). While the diurnal variation was relatively small (34% for the

nighttime and 27 % for the daytime), the ion composition of CHO / CHON•(HSO₄⁻ / NO₃⁻) could also be significantly different between daytime and nighttime due to the availability of the charging ions (see more discussions in Sect. 3.2.2). A previous study from a boreal forest shows that organic ions are mainly composed of CHO / CHON•NO₃⁻ during nighttime and that the fraction of CHO / CHON•HSO₄⁻ increases with the HSO₄⁻ signal during daytime (Bianchi et al., 2017). This is also shown by the slightly positive correlation between the CHO / CHON•HSO₄⁻ signal fraction and the total neutral OOM concentration during daytime (*R*: 0.25; see Fig. S4b), whereas no clear dependence was found between CHO / CHON•(HSO₄⁻ / NO₃⁻) and the total neutral OOM concentration.

The SO_5^- ion group, consisting of SO_5^- ions and/or $O_2 \cdot SO_3^-$ cluster ions (Bork et al., 2013; Frege et al., 2017), exhibited a lower fraction (< 5%) than the aforementioned ion groups during the study period. Similar to that of the $(HNO_3)_{0-1} \cdot NO_3^-$ ion group, no diurnal pattern was evident for the SO_5^- ion group. This may be the result of its different major formation pathways during daytime and nighttime (Bork et al., 2013; Frege et al., 2017). Daytime production of SO₅⁻ ions is likely associated with photooxidation of SO₂ (similar to the formation pathway of H_2SO_4 ; Ehn et al., 2010; Schobesberger et al., 2015). This is shown in the positive correlation (R is 0.46 for daytime data in Fig. S4c) between the neutral H₂SO₄ concentration and the signal fraction of the SO_5^- ion group. During nighttime, however, the SO_5^- ion group is mainly composed of $O_2 \cdot SO_3^-$ cluster ions, which are possibly formed via the oxidation of SO_2 with O_3^- (producing SO_3^-) and subsequent addition of O_2 (Bork et al., 2013).

3.2.2 Seasonalities of negative ions

For better seasonality comparison at the high-altitude CHC, we calculated the average mass spectra of the negative ion groups for each season (Fig. 5 for daytime and Fig. S5 for nighttime). Distinct seasonalities (wet season, wet-todry transition period, and dry season) were found for the majority of the negative ion groups at CHC, including $(H_2SO_4)_{0-3} \cdot HSO_4^-, (NH_3)_{1-6} \cdot (H_2SO_4)_{3-7} \cdot HSO_4^-, SO_5^-, and$ organic cluster ions, as shown in the averaged daytime mass spectra (Fig. 5; a more detailed reason will be discussed below). However, the signals of some other negative ion groups, e.g., MA-derived ions and $(HNO_3)_{0-2} \cdot NO_3^-$, were generally stable (with differences ≤ 20 %) across the seasons. Such unclear seasonalities can be attributed to the high EA (Ferguson and Arnold, 1981; Ravi Kumar et al., 2005) and/or the stability of the parent neutral species (Martin et al., 2007; Kerminen et al., 2000; Bikkina et al., 2021). Similar patterns can also be found in the average nighttime mass spectra among the seasons (Fig. S5).

The $(H_2SO_4)_{0-3} \cdot HSO_4^-$ group exhibited a much higher contribution in the dry season (May) than in the wet season (January) and the wet-to-dry transition period (April). The daytime fraction of $(H_2SO_4)_{0-3} \cdot HSO_4^-$ increased continuously from 16% in the wet season to 20% in the wet-to-dry transition period and to 30% in the dry season. The maximum number of H_2SO_4 molecules increased concurrently from two to four in the cluster ions (i.e., from $(H_2SO_4)_2 \cdot HSO_4^-$ to $(H_2SO_4)_4 \cdot HSO_4^-$). Similar trends were also found for other H_2SO_4 -related ions, such as $(NH_3)_{1-6} \cdot (H_2SO_4)_{3-7} \cdot HSO_4^-$ and SO_5^- during daytime (Fig. 5).

The seasonal variations of the aforementioned H₂SO₄related ion groups are likely due to the changes in neutral H₂SO₄ (Fig. S6) linked to the changing synoptic-scale wind patterns carrying different air masses with varying SO2 (Bianchi et al., 2022). The air mass pathways 07_PW and 08 PW, covering the western and northern Altiplano (see Table 1 and Fig. 2a), where active volcanic degassing of SO₂ has been reported (Moussallam et al., 2017; Carn et al., 2017), had their largest influence on CHC in the dry season (i.e., May; see Fig. 2). The corresponding daytime fractions of $(H_2SO_4)_{0-3}$ ·HSO₄⁻ from these two pathways (Fig. 6a) were also the highest (27 % and 32 %, respectively). In contrast, air mass pathways 03_PW and 12_PW, originating in the Amazon basin and the eastern/southeastern lowlands, exerted their most significant impact on CHC in the wet season (i.e., January), with lower daytime fractions of $(H_2SO_4)_{0-3}$ ·HSO₄ (13% and 14%, respectively). The low fractions of H₂SO₄-related cluster ions in the wet season are also consistent with the lower SO_2 level in the Amazon basin compared to the Altiplano (Andreae et al., 1990). As for the wet-to-dry transition period (i.e., April), 05_PW covering both the southeastern lowlands and the southern Altiplano (where volcanic degassing is also significant; Carn et al., 2017) had an evident influence on CHC, resulting in a substantial level of H₂SO₄-related cluster ions (21% for daytime). It is also noted that, because of the much lower nocturnal neutral H₂SO₄ concentrations, the nighttime fractions of H₂SO₄-related cluster ions in all air mass pathways (Fig. 6b) were generally low (< 3%), and no clear seasonality was found.

The organic cluster ion group exhibited a distinct seasonal variation from $(H_2SO_4)_{0-3} \cdot HSO_4^-$. The signal fraction of organic cluster ions was higher in the wet season (31 % for day-time and 32 % for nighttime) than in the dry season (23 % for daytime and 27 % for nighttime; Figs. 5 and S5), but it was highest for the wet-to-dry transition period (46 % for daytime and 52 % for nighttime; see Fig. 6).

The seasonal changes in organic cluster ions could be due to the combined effect of different meteorological conditions and VOCs from different air mass origins (see Fig. 2). The air masses that originated from the Amazon basin and the lowlands (03_PW and 12_PW) showed their largest impact on CHC in the wet season (i.e., January). They contained higher



Figure 5. Mass spectra of negative ions at CHC averaged between 07:00 and 19:00 in (**a**) the wet season (January), (**b**) the wet-to-dry transition period (April), and (**c**) the dry season (May). The normalized signal intensities from 400 to 800 Th are multiplied by a factor of 5 for better visualization.



Figure 6. The fractions of the negative ion groups observed at CHC determined during the representative periods of each air mass pathway (described in Sect. 2.3.2) for (a) daytime (07:00–19:00) and (b) nighttime (19:00–07:00). A sketch of the horizontal profile of the air mass pathways (Fig. 2a) is shown in the upper-right corner for clarity.

fractions of organic cluster ions, which were 35 % and 34 % for daytime and 50 % and 45 % for nighttime, respectively (Fig. 6). In the dry season (i.e., May), however, the changes in air mass origin towards the Altiplano and the Pacific Ocean led to a lower content of organic cluster ions. The organic cluster ion fractions for 07_PW and 08_PW (largest influence on CHC in the dry season) in May were 23 % and 19 % for daytime and 29 % and 27 % for nighttime, respectively. As for the wet-to-dry transition period (i.e., April), due to the combined influences of biogenic and anthropogenic VOC sources from 05_PW (evident impact on CHC in April), covering the southeastern lowlands and the southern Altiplano, the corresponding organic cluster ion fractions from this air mass pathway were also the highest (41 % for daytime and 53 % for nighttime; see Fig. 6).

A further investigation of the organic ion group shows that the seasonal trends of the individual organic ions also varied (Fig. 7 for daytime and Fig. S7 for nighttime). Although the majority of the organic cluster ions at CHC were more abundant during the wet season (Figs. 7a and S7a), fractions of CHO / CHON·HSO $_4^-$ increased during the dry season. The observed increases in CHO / CHON·HSO₄⁻ cluster ions could be associated with the increased HSO_4^- / NO_3^- ratios in the dry season (Figs. 5 and S5). Similar increases in CHO / CHON·HSO₄⁻ cluster ions were also found to relate to the ratio of HSO_4^- / NO_3^- in a boreal forest environment (Bianchi et al., 2017). In addition, changes in OOM composition between the wet and dry seasons may also play a role (Figs. 7b and S7b), as NO₃⁻ tends to cluster with OOMs containing hydroxyl and hydroperoxyl functional groups (Hyttinen et al., 2015), while some other observed OOMs may be more efficiently charged by HSO_4^- .

The seasonal variations of the individual organic cluster ions are likely caused by different air masses (Figs. 7b and S7b). The air masses influenced by tropical rainforest vegetation from the Amazon basin are dominated by isoprene (C_5H_8) emissions and isoprene oxidation products (Bianchi et al., 2022). This region corresponds to 03_PW and 12_PW (largest impact on CHC in the wet season in January) consisting of relatively higher fractions of organic cluster ions with OOMs containing four to five carbon atoms (50 % and 46 %for nighttime and 29 % and 32 % for daytime, respectively). In contrast, when the air masses were more influenced by the Altiplano (i.e., 05_PW, 07_PW, and 08_PW, with more anthropogenic emissions and less vegetation) in the wet-todry transition period and the dry season, organic cluster ions with six to eight carbon atoms, potentially originating from anthropogenic sources (e.g., toluene (C7H8); Huang et al., 2019; Cai et al., 2022), had higher contributions. The signal fractions of these organic ions were thus the highest in these air mass pathways, accounting for 36 %-39 % for nighttime and 37 %–39 % for daytime. For all the air mass pathways, fractions of organic ions with more than nine carbon atoms were relatively low (< 10%). This might be due to their lower volatilities compared to OOMs with smaller carbon



Figure 7. (a) Mass defect plot of organic cluster ions during nighttime (19:00–07:00). The color code indicates ratios (in log scale) between the median signals of each ion detected in May (I_{May}) of the dry season and January (I_{Jan}) of the wet season. The marker size is proportional to the log-transformed median signals of ions in May. (b) Fraction of organic cluster ions from different air mass pathways as a function of carbon atom numbers during nighttime (19:00–07:00). A similar figure based on daytime data (07:00– 19:00) is in the Supplement (Fig. S7). Note that MA-derived ions were not included in this figure.

numbers (Donahue et al., 2012), resulting in a larger probability of them being removed during their transport to CHC (e.g., condensing on preexisting particles).

3.3 Positive ions

Several positive cluster ion groups were consistently observed in February and March (i.e., the wet season) during the study period. Based on their chemical composition, the positive cluster ions measured at CHC are classified into four groups (Fig. 8): (1) a series of protonated amines, including trimethylamine ($C_3H_9N\cdot H^+$), pyridine ($C_5H_7N\cdot H^+$), aniline ($C_6H_7N\cdot H^+$), and benzylamine ($C_7H_9 N\cdot H^+$); (2) organic cluster ions consisting of OOMs (identified as $C_{3-24}H_{6-39}O_{2-12}N_{0-2}$) clustered with positive charge carriers such as protons (H^+), ammonium (NH_4^+), and aminium (NH^+) ions; (3) contamination ions; (4) unidentified ions (likely organic ions in higher masses; Bianchi et al., 2021). Contamination in the positive cluster ions includes



Mass/charge (Th)

Figure 8. Averaged mass spectra of positive ions at CHC in February and March 2018, when APi-TOF was operating in positive-ion mode (see Sect. 2.2.1), during (a) daytime (07:00–19:00) and (b) nighttime (19:00–07:00). The normalized signal intensities from 400 to 800 Th are multiplied by a factor of 5 for better visualization.

ethylhexylglycerin (e.g., $C_{11}H_{24}O_3 \cdot NH^+$), which is widely used in cosmetics (Aerts et al., 2016), and polydimethylsiloxane (e.g., $(C_2H_6OSi)_7 \cdot NH_4^+$), possibly from instrument tubing (Bianchi et al., 2014). In contrast with the negative cluster ions, the four positive cluster ion groups were generally stable with smaller diurnal variability over the study period (Fig. 9). This is similar to the diurnal patterns determined in previous studies in a boreal forest environment (Ehn et al., 2010) and at the JFJ (Frege et al., 2017). However, due to the unavailable measurements of the corresponding neutral species (e.g., amines), the exact reason for such weak diurnal variations observed in different locations remains unclear.

The protonated amines were the most abundant positiveion group (46%), with no significant diurnal variations. Nighttime contributions of this ion group (47%) were similar to its daytime contributions (45%; Fig. 9). They also dominated the positive-ion spectra observed in different environments, such as a boreal forest (Ehn et al., 2010), the JFJ (Frege et al., 2017), and the free troposphere (Schulte and Arnold, 1990). Their sources have not been fully identified (Kosyakov et al., 2020), but they are widely used as solvents and dyes (Sims et al., 1989), which may be potential sources of these ions observed at CHC.

Positive organic cluster ions were also relatively abundant (19%) at CHC during the wet season. Similar to the negative organic ions in the wet season (Fig. 6), this reflects the in-



Figure 9. Diurnal variation of positive-ion groups at CHC, averaged over measurements in February and March 2018 (when APi-TOF was operating in positive-ion mode; see Sect. 2.2.1).

fluence of air masses originating from the Amazon basin and the eastern/southeastern lowlands (e.g., 03_PW and 11_PW). Differences in the positive organic ion signals between night-



Figure 10. Time series of the (**a**) size distribution of aerosol particles (measured with NAIS and MPSS), (**b**) signal fractions of the $(H_2SO_4)_{0-3} \cdot HSO_4^-$ ion group and neutral H_2SO_4 concentration, and (**c**) signal fractions of the $(NH_3)_{1-6} \cdot (H_2SO_4)_{3-7} \cdot HSO_4^-$ and negative organic cluster ion groups, observed at CHC from 7 to 14 May 2018, when NPF occurred frequently.

time (21%) and daytime (18%) were small, which is similar to the negative organic cluster ions (see Fig. 3). A further investigation of the relationship between these positive ions and their neutral species is, unfortunately, not possible due to the unavailability of CI-APi-TOF data in February and March caused by instrumental issues.

3.4 Potential connections between atmospheric ions and NPF events

During the SALTENA campaign from January to May 2018, NPF events were frequently observed at CHC (Fig. S8). While most of them occurred from April (the wet-to-dry transition period, 21 events) to May (dry season, 26 events), NPF events seldom occurred during the wet season from January to March (8 events in total). This is consistent with a previous study performed at CHC (Rose et al., 2015a), which also found that NPF events mainly occurred during the dry season.

Previous field studies at high-altitude mountain sites have shown that NPF events can be triggered by different compounds, such as low-volatility neutral OOMs (Bianchi et al., 2021), neutral H₂SO₄ and OOMs (Bianchi et al., 2016), and H₂SO₄–NH₃ cluster ions (Frege et al., 2017). While the signal fractions of the negative organic cluster ions did not seem to have a strong correlation with the onset of the NPF events, the fractions of the (NH₃)_{1–6}•(H₂SO₄)_{3–7}•HSO₄⁻ (associated with $(H_2SO_4)_{0-3} \cdot HSO_4^-$ always increased concurrently with the number concentration of small particles (see example NPF events on, e.g., 7, 10, 13, and 14 May 2018; Fig. 10). The number of NPF days increased when more H_2SO_4 molecules were present in the $(NH_3)_{1-6} \cdot (H_2SO_4)_{3-7} \cdot HSO_4^$ cluster ion (Fig. 11a). In particular, more than half of the NPF events (28 out of the total 55 of events) were observed in the presence of $(NH_3)_{4-6} \cdot (H_2SO_4)_7 \cdot HSO_4^-$. The majority (35 events) of all the NPF events exhibited clear nucleation and growth processes (i.e., Class 1 events; the classification is defined following the approach by Yli-Juuti et al., 2009; Fig. 11b). In contrast, only Class 2 (similar to Class 1 but with less clarity) and bump events (early growth of the newly formed particles is interrupted) were observed when only $(H_2SO_4)_2 \cdot HSO_4^-$ was observed.

Moreover, higher levels (up to an order of magnitude) of $(NH_3)_{1-6} \cdot (H_2SO_4)_{3-7} \cdot HSO_4^-$ and $(H_2SO_4)_{0-3} \cdot HSO_4^-$ ions as well as negative organic cluster ions charged by HSO_4^- were also observed during the NPF days (Fig. 12). In contrast, other negative ion groups (e.g., the majority of the negative organic cluster ions charged by NO_3^-) were more abundant during the non-NPF days. Our observations indicate a potentially important role of $(NH_3)_{1-6} \cdot (H_2SO_4)_{3-7} \cdot HSO_4^-$ cluster ions in NPF events at CHC from January to May 2018, particularly in the wet-to-dry transition period and the dry season.



Figure 11. Connection between the maximum number of H_2SO_4 (in addition to HSO_4^-) observed in the $(H_2SO_4)_2 \cdot HSO_4^-$ and $(NH_3)_{1-6} \cdot (H_2SO_4)_{3-7} \cdot HSO_4^-$ cluster ions during NPF events and (a) the number of NPF days; (b) proportions of different NPF classes.

The majority of the observed NPF events occurred when CHC was more impacted by air masses originating from source regions with elevated SO₂ emissions (05_PW, 07_PW, and 08_PW). This is similar to the observations from the high-altitude station JFJ (Frege et al., 2017). Moreover, consistent with the previous findings from Rose et al. (2015a), CHC was affected by the frequent arrival of air masses from the Pacific Ocean during the dry season (see also Fig. 2). The NPF during the dry season at CHC also seems to be triggered once the air masses arrive on the continent (Rose et al., 2015a). However, these air masses of marine origin may not directly contain the nucleating species (also reflected in the poor connection between NPF events and the levels of methanesulfonic acid ions $(CH_3SO_3^-)$ or IO_3^- ; data not shown). In addition, inconsistent time series of CH₃SO₃⁻ and IO₃⁻ prohibit us from drawing further conclusions about the role of marine-derived compounds in NPF at Chacaltaya during the dry season. Besides, the fraction of large positive organic cluster ions (mass range from 500 to 800 Th; Fig. S9) was found to increase during the NPF events in the wet season. These organic cluster ions usually contained at least 10 carbon atoms (Fig. S10). Such large positive organic ions have been found to contribute to NPF in the Himalayas (Bianchi et al., 2021), and thus their contribution to NPF cannot be completely ruled out at CHC.



Figure 12. Mass defect plot of differences in negative cluster ion composition between NPF and non-NPF days. The negative ion composition of NPF events was averaged over all NPF days from 08:00 to 12:00 in January, April, and May 2018 (when APi-TOF was operating in negative-ion mode; see Sect. 2.2.1) at CHC. The ion composition of non-NPF days was averaged over non-NPF days from 08:00 to 12:00 for the same period. The *x* axis is the exact mass of cluster ions, and the *y* axis is the mass defect. The color code indicates ratios (in log scale) between median signals of each ion determined in NPF events (I_{NPF}) and non-NPF periods ($I_{non-NPF}$). The marker size is proportional to the log-transformed median signals of ions observed in the NPF events that occurred in January, April, and May. Note that (HNO₃)₀₋₂•NO₃⁻ cluster ions were not included here.

4 Conclusions

In this study, both negative and positive atmospheric ions were measured at a high-altitude research station (CHC) in the Bolivian Andes for 5 months, from January to May 2018, using an APi-TOF mass spectrometer. Negative ions were mainly composed of $(H_2SO_4)_{0-3}$ ·HSO₄, $(HNO_3)_{0-2} \cdot NO_3^-$, SO_5^- , $(NH_3)_{1-6} \cdot (H_2SO_4)_{3-7} \cdot HSO_4^-$, MAderived, and CHO/CHON•(HSO $_4^-$ /NO $_3^-$) ion groups. Positive ions mainly consisted of a series of protonated amines $(C_{3-7}H_{7-9}N\cdot H^+)$ and organic cluster ions CHO / CHON \cdot (H⁺ / NH⁺ / NH⁺). Distinct diurnal variation was observed for the negative ions and attributed mainly to the changes in the corresponding neutral species' concentrations and/or their EA / PA. An example is H₂SO₄related cluster ions, the diel temporal variation of which was mainly due to the photochemical production of neutral H₂SO₄ during daytime. Strong seasonality of negative ions was also found, such as for H₂SO₄-related cluster ions owing to changes in SO₂ and the resulting neutral H₂SO₄ concentrations. The seasonal variation was mainly because of the differences in source regions of air masses arriving at CHC from the wet to dry seasons. In contrast, no significant diurnal variation was observed for the positive ions. The comparison between NPF and non-NPF days infers that H₂SO₄-NH₃ cluster ions contribute to the aerosol-nucleation process at CHC, particularly in the wet-to-dry transition period and the dry season, when CHC was more impacted by air masses

originating from source regions with elevated SO_2 emissions. The results further indicate that atmospheric ion composition at CHC is directly affected by air masses from different source regions.

Measurements of atmospheric ions in the field will improve understanding of atmospheric physical and chemical processes in the study regions, as the ions play important roles in atmospheric chemistry through participation in or catalysis of ion-molecule reactions and ion-induced new particle formation. Our study thus provides new insights into the chemical composition of atmospheric ions and their potential role in high-altitude NPF in the Bolivian Andes, where both natural (e.g., biogenic and volcanic) and anthropogenic emissions are important.

Data availability. The data that are involved in the figures can be found at https://doi.org/10.5281/zenodo.7271286 (Zha, 2022).

Supplement. The supplement related to this article is available online at: https://doi.org/10.5194/acp-23-4559-2023-supplement.

Author contributions. QZ, WH, and FB analyzed the data. DA conducted the air mass history analysis. QZ, WH, DA, OP, LH, AMK, CW, JE, YG, MA, CM, and FB collected the data and operated the instruments during the measurement campaign. QZ and WH wrote the manuscript with contributions from JC, VS, SC, DW, RK, MA, CM, and FB. All the authors commented on the manuscript.

Competing interests. The contact author has declared that none of the authors has any competing interests.

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Acknowledgements. We thank the Bolivian staff of the IIF-UMSA (Physics Research Institute, UMSA) working at CHC for the long-term observations performed within the framework of the Global Atmosphere Watch (GAW) and the Aerosol, Clouds and Trace Gases Research Infrastructure (ACTRIS). We thank the IRD (Institut de Recherche pour le Développement) for the logistic and financial support during the campaign, including shipping and customs concerns. We thank the CSC-IT Center for Science, Finland, for the generous computational resources that allowed the WRF and FLEXPART-WRF simulations to be conducted.

Financial support. This research has been supported by the European Commission, Horizon 2020 Framework Programme (grant nos. 850614 and 742206), the H2020 Marie Skłodowska-Curie

Actions (grant no. 764991), and the Knut and Alice Wallenberg Foundation (grant no. 2017.0165).

Open-access funding was provided by the Helsinki University Library.

Review statement. This paper was edited by Zhibin Wang and reviewed by two anonymous referees.

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