



Supplement of

Measurement report: Intensive biomass burning emissions and rapid nitrate formation drive severe haze formation in the Sichuan Basin, China – insights from aerosol mass spectrometry

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Fig. S1 Molar ratios of nitrate to sulphate vs. ammonium to sulphate during daytime





Fig. S2 Variation of nitrate and sulphate as Ox/ALWC increases during (a) daytime (left column) and (b) nighttime (right column). The data of nitrate and sulphate concentrations were grouped into different bins according to 20 μ g/m³ increment of Ox, and 100 μ g/m³ increment of ALWC. The mean (square), 50th (horizontal line inside the box), 25th and 75th percentiles (lower and upper box), and 10th and 90th percentiles (lower and upper whiskers) of the box chart are marked in (a).



Fig. S3 Diurnal variations of (a) chemical composition in NR-PM_{2.5}, (b) O₃, NO₂, and SO₂, (c) RH
and temperature, (d) planet boundary layer height (PBLH) and solar radiation (SR). The PBLH was
derived from the European Centre for Medium-Range Weather Forecasts (ECMWF) dataset of
ERA5 hourly data (https://cds.climate.copernicus.eu/cdsapp#!/home).



Fig. S4 Mass spectrum profile of OA factors resolved by PMF for (a) 3-, (b) 4-, and (c) 5-factor

solutions



Fig. S5 OOA concentration as a function of (a) Ox and (c) ALWC during daytime. (b) and (d) are the same as (a) and (c) but during nighttime. The data of OOA concentration are grouped into different bins according to 20 μ g/m³ increment of Ox and 100 μ g/m³ increment of ALWC during both daytime and nighttime. The colour scale represents O₃/Ox ratios in (a) and (b). The mean (square), 50th (horizontal line inside the box), 25th and 75th percentiles (lower and upper box), and 10th and 90th percentiles (lower and upper whiskers) of the box chart are marked in (b).

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Fig. S7 Average nitrate formation rate as a function of $PM_{2.5}$ concentration during H1, H2 and H3



Fig. S8 Fire maps of areas around Deyang during (a) non-haze, (b) H2 and H3 periods. The Fire
Maps were acquired from Fire Information for Resource Management System (FIRMS) developed
by the National Aeronautics and Space Administration (NASA). The data of VIIRS (375m) was
used (https://firms.modaps.eosdis.nasa.gov/active_fire/).



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Table S1 Summary of mass concentrations of PM2.5 compositions measured during winter in

different cities.									
Species	This study ^a	Chengdu ^b	Chongqing ^b	Xi'an ^c	Changzhou	Beijing ^d			
OA	39.2 ± 3.9	N.A.	N.A.	64.2 ± 40.6	31.2 ± 11.9	103 ± 33			
NO ₃ -	$29\ \pm 14$	$33.4~{\pm}29.5$	$15.8~{\pm}9.5$	$27.7~\pm20.4$	$24.1~{\pm}11.8$	43 ± 11			
$\mathrm{NH}_{4^{+}}$	$15.1~{\pm}6.4$	$12~{\pm}7.9$	$11.3~\pm5.2$	$12.5~\pm9.1$	$13.1~{\pm}3.7$	$14.9~{\pm}5.1$			
SO ₄ ²⁻	10 ± 4.2	16.6 ± 13	$17.5~\pm7.4$	$17.6~{\pm}14.2$	$18.7~\pm7.6$	$47\ \pm 15$			
Cl	5.2 ± 4.1	N.A.	$1.6~{\pm}1.2$	5.1 ± 4.0	N.A.	35.4 ± 7.9			
Reference		(Huang et	(Wang et	(Duan et	(Ye et al.,	(Elser et			
		al., 2021)	al., 2018)	al., 2021)	2017)	al., 2016)			

^a The data of rainy hours (0:00-9:00 25 December, 2021 & 0:00-8:00 1 January, 2022) were removed.

^b The concentrations of water-soluble inorganic ions were measured by an ion chromatography,

vhile the compositions in other cities listed in the table were measured by AMS.

^c The concentrations of different compositions were measured before COVID-19 lockdown.

^d The concentrations of different compositions were measured during extreme haze episodes.

78

Table S2 Description of the PMF solutions

Factor numbers	fpeak	Q/Qexp	Comment			
2	0	3.13	Too few factors and large residuals.			
	0	2.64	Optimum PMF solution. Q/Qexp decreases by			
			15.7 %. Temporal profile and diurnal variation of			
2			different factors are consistent with external tracers.			
3	0		The factors (HOA, BBOA, and OOA) resolved also			
			represent major OA sources around the observation			
			site.			
		2.25	Q/Qexp decreases by 14.8 %. A new factor (factor2 in			
			Fig. S4 (b)) with high m/z 55/ m/z 57, which makes itself			
			look like cooking organic aerosol (COA), is separated.			
4	0		However, the diurnal profile of this factor does not show			
			apparent peaks at noon and in the evening. The			
			observation site is not affected by intense cooking			
			emissions either.			
		2.03	Q/Qexp decreases by 10 %. OOA is split into two factors			
~	0		with similar time series. One of the factors (factor4 in			
5	0		Fig. S4 (c)) has too low m/z 44 signal intensity, which is			
			not reasonable for OOA.			

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Table S3 Summary of meteorological parameters, mass concentrations of PM2.5 species, OA

87	factors and elemental ratios during different episodes.								
	H1	H2	H3	F1	F2	F3	Overall		
Meteorological Parameters									
T (°C)	$8.2\ \pm 2.7$	$6.6\ \pm 2.8$	$8.0\ \pm 2.4$	$6.2\ \pm 2.3$	$1.5\ \pm 3.2$	$4.7\ \pm 1.3$	$7.3\ \pm 2.8$		
RH (%)	80.7 ± 11.3	81.5 ± 11.7	79.7 ± 12.4	$99.9\ \pm 0.3$	$99.0\ \pm 1.4$	$99.5\ \pm 0.7$	81.0 ± 12.4		
WS (m/s)	$0.7\ \pm 0.5$	$0.7\ \pm 0.4$	$0.7\ \pm 0.4$	$0.9\ \pm 0.4$	$0.7\ \pm 0.2$	$0.6\ \pm 0.2$	$0.7\ \pm 0.5$		
$SR (W/m^2)^a$	$297\ \pm 156$	318 ± 157	$276~{\pm}162$	470	500	75	$276~{\pm}164$		
$PM_{2.5}$ species (µg/m ³)									
Org	$45.6 \ \pm 18.4$	42.1 ± 11.5	42.2 ± 11.8	$53.3\ \pm 12.8$	$45.0\ \pm 12.7$	$45.4\ \pm 8.4$	39.2 ± 15.7		
NO ₃ -	$34.3\ \pm 17.4$	33.5 ± 11.1	$30.2\ \pm11.0$	41.1 ± 17.8	$22.3\ \pm 12.7$	$23.0~{\pm}4.9$	29.0 ± 13.9		
SO_4^{2-}	$10.5\ \pm 4.7$	$11.5~\pm3.4$	$10.3\ \pm 3.8$	$14.6~{\pm}6.7$	$10.3\ \pm 2.2$	$13.0~\pm6.1$	$10.1\ \pm 4.2$		
$\mathrm{NH_{4}^{+}}$	$16.9~\pm7.6$	$17.1~\pm5.0$	$15.8~{\pm}5.5$	$21.3\ \pm 8.6$	$15.3\ \pm 3.8$	$13.6~\pm2.5$	$15.1~\pm6.4$		
Chl	$5.8~{\pm}3.3$	$5.1~{\pm}2.8$	5.9 ± 4.4	$8.3\ \pm 3.6$	$16.2\ \pm 10.9$	$8.1\ \pm 2.7$	5.2 ± 4.1		
BC	$6.7\ \pm 2.6$	6.2 ± 2.3	$7.1~\pm2.2$	$8.3\ \pm 1.9$	$9.3\ \pm 2.8$	$9.3\ \pm 1.0$	$6.2\ \pm 2.8$		
$OA(\mu g/m^3)$									
HOA	$11.7~\pm7.6$	$10.5\ \pm 6.9$	$8.6\ \pm 5.1$	$13.9~{\pm}4.6$	$16.3\ \pm7.7$	$11.0\ \pm 2.9$	8.9 ± 6.5		
BBOA	$7.7\ \pm 5.5$	$10.0~\pm4.7$	$10.6~{\pm}4.6$	$11.2~\pm3.8$	$14.7\ \pm 5.2$	$13.0\ \pm 3.0$	$8.9\ \pm 5.4$		
OOA	$18.3~\pm7.4$	$20.5\ \pm 6.9$	$15.5~{\pm}4.4$	$22.3\ \pm10.1$	$12.8~{\pm}5.1$	$12.1\ \pm 2.7$	$16.3\ \pm 6.8$		
Elemental ratios ^b									
O/C	$0.70\ \pm 0.14$	0.71 ± 0.14	$0.68\ \pm 0.12$	0.66 ± 0.13	$0.54\ \pm 0.13$	$0.55\ \pm 0.06$	0.71 ± 0.14		
H/C	$1.54\ \pm 0.03$	$1.56\ \pm 0.02$	$1.58\ \pm 0.02$	1.56 ± 0.01	$1.59\ \pm 0.01$	$1.59\ \pm 0.01$	$1.56\ \pm 0.09$		
$\overline{\mathrm{OS}_{\mathrm{c}}}$	-0.12 ± 0.39	-0.14 ± 0.28	-0.23 ± 0.25	-0.23 ± 0.26	-0.51 ± 0.22	-0.48 ± 0.12	-0.14 ± 0.31		

^a The daily maximum was used for calculating the average values during different episodes

^b The O/C and H/C were determined by the parameterization proposed by Canagaratna et al. (2015).

90 The $\overline{OS_c}$ was calculated as 2O/C - H/C recommended by (Heald et al., 2010).

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94 **References**

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96 Canagaratna, M. R., Jimenez, J. L., Kroll, J. H., Chen, Q., Kessler, S. H., Massoli, P., Hildebrandt Ruiz,
97 L., Fortner, E., Williams, L. R., Wilson, K. R., Surratt, J. D., Donahue, N. M., Jayne, J. T., and
98 Worsnop, D. R.: Elemental ratio measurements of organic compounds using aerosol mass
99 spectrometry: characterization, improved calibration, and implications, Atmos. Chem. Phys., 15,
100 253-272, https://doi.org/10.5194/acp-15-253-2015, 2015.

- Duan, J., Huang, R., Chang, Y., Zhong, H., Gu, Y., Lin, C., Hoffmann, T., and O'Dowd, C.: Measurement
 report of the change of PM2.5 composition during the COVID-19 lockdown in urban Xi'an: Enhanced
 secondary formation and oxidation, Sci. Total Environ., 791, 148126, https://doi.org/
 104 10.1016/j.scitotenv.2021.148126, 2021.
- Elser, M., Huang, R., Wolf, R., Slowik, J. G., Wang, Q., Canonaco, F., Li, G., Bozzetti, C., Daellenbach,
 K. R., Huang, Y., Zhang, R., Li, Z., Cao, J., Baltensperger, U., El-Haddad, I., and Pr év αt, A. S. H.:
 New insights into PM2.5 chemical composition and sources in two major cities in China during
 extreme haze events using aerosol mass spectrometry, Atmos. Chem. Phys., 16, 3207-3225,
 https://doi.org/10.5194/acp-16-3207-2016, 2016.
- Heald, C. L., Kroll, J. H., Jimenez, J. L., Docherty, K. S., DeCarlo, P. F., Aiken, A. C., Chen, Q., Martin,
 S. T., Farmer, D. K., and Artaxo, P.: A simplified description of the evolution of organic aerosol
 composition in the atmosphere, Geophys. Res. Lett., 37, https://doi.org/ 10.1029/2010GL042737,
 2010.
- Huang, X., Zhang, J., Zhang, W., Tang, G., and Wang, Y.: Atmospheric ammonia and its effect on PM2.5
 pollution in urban Chengdu, Sichuan Basin, China, Environ. Pollut., 291, 118195, https://doi.org/
 10.1016/j.envpol.2021.118195, 2021.
- Wang, H., Tian, M., Chen, Y., Shi, G., Liu, Y., Yang, F., Zhang, L., Deng, L., Yu, J., Peng, C., and Cao,
 X.: Seasonal characteristics, formation mechanisms and source origins of PM2.5 in two megacities in
 Sichuan Basin, China, Atmos. Chem. Phys., 18, 865-881, https://doi.org/ 10.5194/acp-18-865-2018,
- **120** 2018.
- Ye, Z., Liu, J., Gu, A., Feng, F., Liu, Y., Bi, C., Xu, J., Li, L., Chen, H., Chen, Y., Dai, L., Zhou, Q., and
 Ge, X.: Chemical characterization of fine particulate matter in Changzhou, China, and source
 apportionment with offline aerosol mass spectrometry, Atmos. Chem. Phys., 17, 2573-2592,
 https://doi.org/ 10.5194/acp-17-2573-2017, 2017.
- 125