



*Supplement of*

## **Source apportionment study on particulate air pollution in two high-altitude Bolivian cities: La Paz and El Alto**

**Valeria Mardoñ**

*Correspondence to:* Valeria Mardoñez (valeria.mardonez@univ-grenoble-alpes.fr)

The copyright of individual parts of the supplement might differ from the article licence.

**Table S1. Bolivian air quality guidelines<sup>1</sup>**

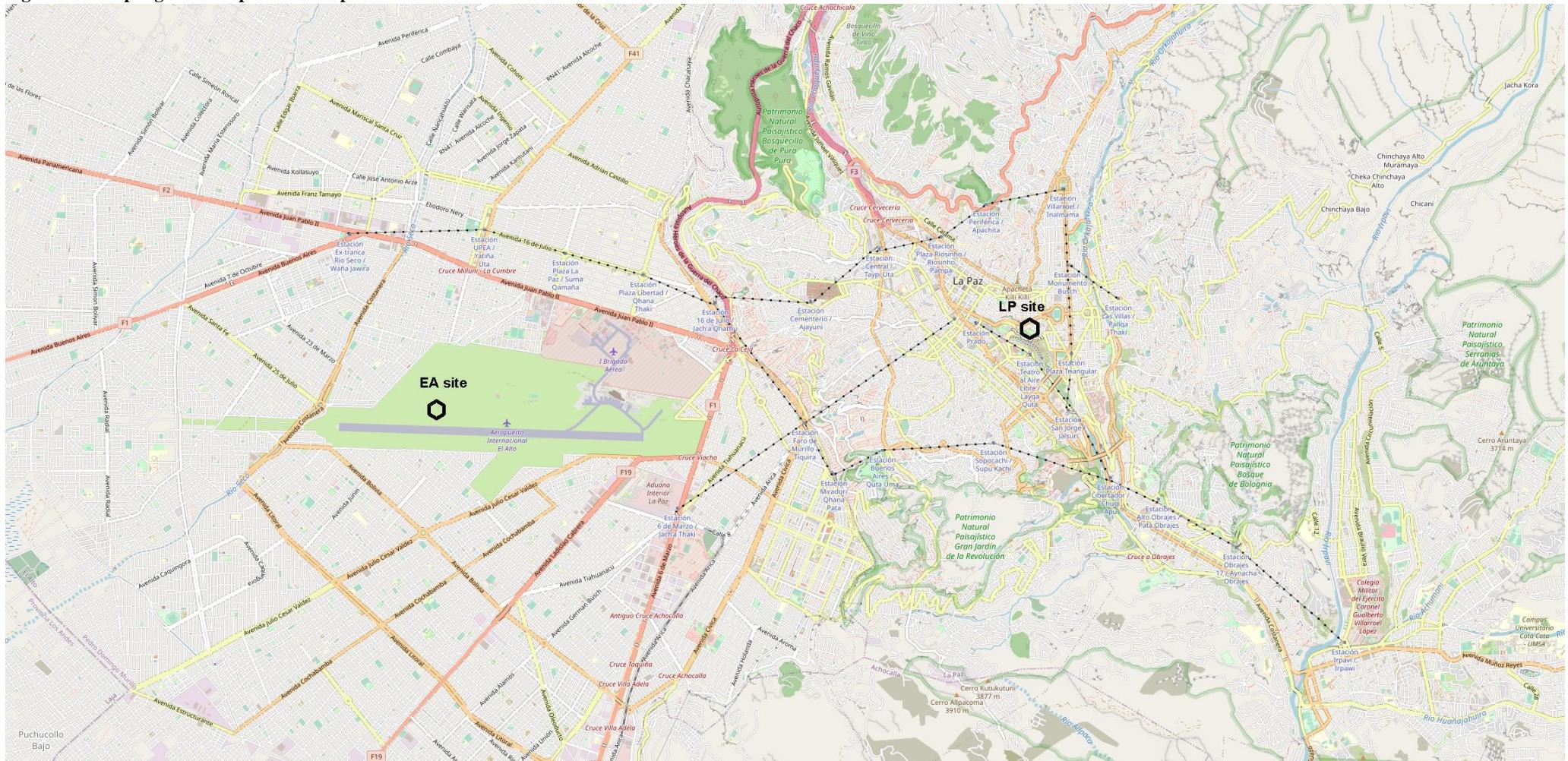
<i>Pollutant</i>	<i>Concentration</i>	<i>Period/ statistic characterization</i>
<i>Carbon Monoxide (CO)</i>	10 mg m <sup>-3</sup>	8h mean
	40 mg m <sup>-3</sup>	1h mean
<i>Sulfur dioxide (SO<sub>2</sub>)</i>	80 µg m <sup>-3</sup>	annual mean
	365 µg m <sup>-3</sup>	24h mean
<i>Nitrogen dioxide (NO<sub>2</sub>)</i>	150 µg m <sup>-3</sup>	24h mean
	400 µg m <sup>-3</sup>	1h mean
<i>Total suspended particles (TSP)</i>	260 µg m <sup>-3</sup>	24h mean
	75 µg m <sup>-3</sup>	annual mean
<i>Particles smaller than 10 µm (PM<sub>10</sub>)</i>	150 µg m <sup>-3</sup>	24h mean
	50 µg m <sup>-3</sup>	annual mean
<i>Ozone (O<sub>3</sub>)</i>	236 µg m <sup>-3</sup>	1h mean
<i>Lead (Pb)</i>	1.5 µg m <sup>-3</sup>	3-month mean

**Figure S1. Photographs taken at the sampling sites (Left: El Alto sampling site; right: La Paz sampling site)**



<sup>1</sup> The concentration values are referred to normal concentrations of pressure and temperature ( $\bar{T} = 298 K$ ,  $\bar{P} = 1013.5 hPa$ )

Figure S2. Sampling sites. ©OpenStreetMap contributors 2020. Distributed under a Creative Commons BY-SA License



**Table S2. Mean, median and standard deviation (sd) of the measured ambient concentrations (above the mean QL, after excluding outliers and samples collected during festivities or the day after, e.g. San Juan, Christmas and New Year). For STP concentrations, ambient concentrations in El Alto must be multiplied by a factor 1.52 and 1.46 for La Paz.**

Group	Specie	Analysis	Units	mean QL	El Alto (EA)				La Paz (LP)			
					mean	median	sd	n	mean	median	sd	n
carbonaceous aerosols	<b>particulate matter (PM)</b>	gravimetry thermal-optical analysis (Cavalli et al., 2010)	[ $\mu\text{g m}^{-3}$ ]	---	29.9	28.9	12.0	93	27.18	27.07	8.91	103
	<b>organic carbon (OC)</b>		[ $\mu\text{g m}^{-3}$ ]	0.21	3.51	3.48	1.57	93	3.85	3.68	1.78	103
	<b>elemental carbon (EC)</b>		[ $\mu\text{g m}^{-3}$ ]	0.02	1.46	1.41	0.60	93	1.59	1.57	0.78	103
polyhydric alcohols	<b>arabitol</b>	high performance liquid chromatography (HPLC) (Piot et al., 2012)	[ $\text{ng m}^{-3}$ ]	0.77	2.09	1.79	1.08	83	3.75	3.18	2.28	100
	<b>sorbitol</b>		[ $\text{ng m}^{-3}$ ]	0.79	9.13	5.34	10.77	75	13.88	7.75	17.38	87
	<b>mannitol</b>		[ $\text{ng m}^{-3}$ ]	0.77	3.82	3.37	2.67	88	6.73	5.30	4.53	97
monosaccharide anhydrides	<b>levoglucosan</b>		[ $\text{ng m}^{-3}$ ]	0.62	87.73	66.47	76.80	93	79.00	46.26	83.15	101
	<b>mannosan</b>		[ $\text{ng m}^{-3}$ ]	0.77	7.52	4.32	9.02	89	7.46	3.33	10.08	98
	<b>galactosan</b>		[ $\text{ng m}^{-3}$ ]	0.31	3.91	1.65	5.74	91	4.31	1.86	6.09	95
<b>saccharides</b>	<b>glucose</b>		[ $\text{ng m}^{-3}$ ]	0.77	14.35	13.81	5.84	93	22.07	22.33	8.81	102
ions	<b>methanesulfonic acid (MSA<sup>-</sup>)</b>	ionic chromatography (IC) (Jaffrezo et al., 2005)	[ $\text{ng m}^{-3}$ ]	0.06	3.93	3.57	1.87	93	4.38	4.05	2.48	88
	<b>chloride (Cl<sup>-</sup>)</b>		[ $\text{ng m}^{-3}$ ]	5.56	217.51	152.27	235.80	93	65.24	52.56	52.79	102
	<b>nitrate (NO<sub>3</sub><sup>-</sup>)</b>		[ $\text{ng m}^{-3}$ ]	11.02	609.72	551.21	359.84	92	555.16	482.23	321.65	102
	<b>sulfate (SO<sub>4</sub><sup>2-</sup>)</b>		[ $\text{ng m}^{-3}$ ]	5.29	1247.06	1068.66	723.87	93	1252.77	1098.03	769.54	102
	<b>oxalate (Ox<sup>-</sup>)</b>		[ $\text{ng m}^{-3}$ ]	1.64	50.07	45.75	43.71	73	44.95	24.11	58.18	91
	<b>sodium (Na<sup>+</sup>)</b>		[ $\text{ng m}^{-3}$ ]	10.47	52.05	46.35	32.64	86	40.94	39.59	21.10	95
	<b>ammonium (NH<sub>4</sub><sup>+</sup>)</b>		[ $\text{ng m}^{-3}$ ]	8.74	511.29	451.65	399.06	93	403.81	333.12	308.04	102
	<b>potassium (K<sup>+</sup>)</b>		[ $\text{ng m}^{-3}$ ]	2.76	77.41	68.44	52.61	93	73.34	56.82	55.77	102
	<b>magnesium (Mg<sup>+</sup>)</b>		[ $\text{ng m}^{-3}$ ]	0.62	25.14	22.72	12.72	93	24.33	22.45	12.77	102
	<b>calcium (Ca<sup>+</sup>)</b>		[ $\text{ng m}^{-3}$ ]	3.53	362.51	312.20	201.63	93	256.30	222.58	136.51	102
metals	<b>aluminum (Al)</b>	inductively coupled plasma atomic emission spectrometry (ICP-AES)	[ $\mu\text{g m}^{-3}$ ]	0.003	1.681	1.582	0.928	93	1.133	1.035	0.563	101
	<b>calcium (Ca)</b>		[ $\mu\text{g m}^{-3}$ ]	0.001	0.428	0.386	0.218	93	0.350	0.328	0.181	101
	<b>potassium (K)</b>		[ $\mu\text{g m}^{-3}$ ]	0.001	0.529	0.527	0.287	93	0.394	0.361	0.196	101
	<b>sodium (Na)</b>		[ $\mu\text{g m}^{-3}$ ]	0.001	0.137	0.135	0.078	93	0.110	0.106	0.060	101
	<b>magnesium (Mg)</b>		[ $\mu\text{g m}^{-3}$ ]	0.001	0.160	0.156	0.083	93	0.125	0.125	0.059	101
	<b>iron (Fe)</b>	inductively coupled plasma mass	[ $\mu\text{g m}^{-3}$ ]	0.001	0.932	0.900	0.501	93	0.669	0.590	0.357	101
	<b>lithium (Li)</b>		[ $\text{ng m}^{-3}$ ]	0.05	1.60	1.42	0.96	93	1.04	0.92	0.57	101
	<b>beryllium (Be)</b>		[ $\text{ng m}^{-3}$ ]	0.05	0.10	0.10	0.02	49	0.08	0.08	0.02	27
	<b>phosphor (P)</b>		[ $\text{ng m}^{-3}$ ]	0.05	37.08	34.43	17.91	93	25.48	22.66	11.94	101

<b>scandium (Sc)</b>	spectrometry (ICP-MS)  (Querol et al., 2001)	[ng m <sup>-3</sup> ]	0.05	0.24	0.21	0.13	35	0.20	0.20	0.09	25
<b>titanium (Ti)</b>		[ng m <sup>-3</sup> ]	0.05	80.15	74.41	44.71	93	55.83	50.55	27.76	101
<b>vanadium (V)</b>		[ng m <sup>-3</sup> ]	0.05	2.04	2.01	1.08	88	1.54	1.49	0.73	97
<b>chromium (Cr)</b>		[ng m <sup>-3</sup> ]	0.05	2.66	1.84	4.52	78	2.36	2.02	1.45	97
<b>manganese (Mn)</b>		[ng m <sup>-3</sup> ]	0.05	16.48	15.75	8.71	93	12.69	11.90	6.19	101
<b>cobalt (Co)</b>		[ng m <sup>-3</sup> ]	0.05	0.36	0.32	0.18	91	0.24	0.22	0.11	99
<b>nickel (Ni)</b>		[ng m <sup>-3</sup> ]	0.05	1.18	0.96	1.29	83	1.12	0.86	0.84	99
<b>copper (Cu)</b>		[ng m <sup>-3</sup> ]	0.05	2.89	2.51	1.89	90	4.25	3.91	2.52	101
<b>zinc (Zn)</b>		[ng m <sup>-3</sup> ]	0.05	12.72	12.63	5.51	93	11.68	11.24	5.47	100
<b>gallium (Ga)</b>		[ng m <sup>-3</sup> ]	0.05	0.40	0.39	0.23	90	0.29	0.26	0.15	101
<b>germanium (Ge)</b>		[ng m <sup>-3</sup> ]	0.05	0.11	0.11	0.04	21	0.13	0.11	0.07	8
<b>arsenic (As)</b>		[ng m <sup>-3</sup> ]	0.05	1.81	1.53	1.07	93	1.03	0.98	0.46	101
<b>selenium (Se)</b>		[ng m <sup>-3</sup> ]	0.05	0.09	0.10	0.02	16	0.10	0.09	0.03	17
<b>rubidium (Rb)</b>		[ng m <sup>-3</sup> ]	0.05	3.12	2.88	1.76	93	2.11	1.88	1.10	101
<b>strontium (Sr)</b>		[ng m <sup>-3</sup> ]	0.05	3.45	3.36	1.90	93	2.64	2.42	1.33	101
<b>yttrium (Y)</b>		[ng m <sup>-3</sup> ]	0.05	0.43	0.42	0.23	83	0.37	0.31	0.26	85
<b>zirconium (Zr)</b>		[ng m <sup>-3</sup> ]	0.40	7.49	7.63	1.88	93	6.25	5.99	1.86	100
<b>niobium (Nb)</b>		[ng m <sup>-3</sup> ]	0.05	0.30	0.29	0.13	93	0.23	0.23	0.09	101
<b>molibdene (Mo)</b>		[ng m <sup>-3</sup> ]	0.05	1.21	1.09	0.74	49	1.39	1.00	1.13	29
<b>cadmium (Cd)</b>		[ng m <sup>-3</sup> ]	0.05	0.14	0.13	0.06	90	0.09	0.08	0.02	57
<b>tin (Sn)</b>		[ng m <sup>-3</sup> ]	0.05	0.58	0.44	0.43	92	0.40	0.36	0.22	101
<b>antimony (Sb)</b>		[ng m <sup>-3</sup> ]	0.05	1.05	0.80	0.83	91	0.95	0.78	0.67	101
<b>cesium (Cs)</b>		[ng m <sup>-3</sup> ]	0.05	0.34	0.29	0.19	90	0.22	0.21	0.12	97
<b>barium (Ba)</b>		[ng m <sup>-3</sup> ]	0.05	14.60	14.48	7.24	90	15.64	15.58	7.46	101
<b>lanthanum (La)</b>		[ng m <sup>-3</sup> ]	0.05	0.82	0.81	0.42	92	0.58	0.54	0.27	101
<b>cerium (Ce)</b>		[ng m <sup>-3</sup> ]	0.05	1.65	1.62	0.90	93	1.22	1.16	0.67	101
<b>praseodymium (Pr)</b>		[ng m <sup>-3</sup> ]	0.05	0.20	0.20	0.10	86	0.17	0.17	0.06	94
<b>neodymium (Nd)</b>		[ng m <sup>-3</sup> ]	0.05	0.75	0.73	0.41	92	0.52	0.49	0.25	101
<b>samarium (Sm)</b>		[ng m <sup>-3</sup> ]	0.05	0.15	0.15	0.07	79	0.11	0.12	0.04	73
<b>europium (Eu)</b>		[ng m <sup>-3</sup> ]	0.05	0.07	0.07	0.01	26	0.07	0.07		1
<b>gadolinium (Gd)</b>		[ng m <sup>-3</sup> ]	0.05	0.14	0.14	0.06	73	0.11	0.10	0.05	73
<b>terbium (Tb)</b>		[ng m <sup>-3</sup> ]	0.05	0.07	0.07	0.01	46	0.07	0.07	0.01	18
<b>dysprosium (Dy)</b>		[ng m <sup>-3</sup> ]	0.05	0.12	0.11	0.04	61	0.10	0.08	0.05	59
<b>holmium (Ho)</b>		[ng m <sup>-3</sup> ]	0.05	0.07	0.07	0.01	47	0.07	0.07	0.00	6

	<b>erbium (Er)</b>		[ng m <sup>-3</sup> ]	0.05	0.07	0.07	0.01	35	0.07	0.07	0.01	34
	<b>thulium (Tm)</b>		[ng m <sup>-3</sup> ]	0.05				0				0
	<b>ytterbium (Yb)</b>		[ng m <sup>-3</sup> ]	0.05	0.07	0.07	0.01	34	0.07	0.07	0.02	57
	<b>lutetium (Lu)</b>		[ng m <sup>-3</sup> ]	0.05				0				0
	<b>hafnium (Hf)</b>		[ng m <sup>-3</sup> ]	0.40	0.40	0.40		1	0.42	0.41	0.01	9
	<b>tantalum (Ta)</b>		[ng m <sup>-3</sup> ]	0.05				0				0
	<b>tungsten (W)</b>		[ng m <sup>-3</sup> ]	0.05	0.24	0.16	0.23	84	0.20	0.13	0.15	81
	<b>thallium (Tl)</b>		[ng m <sup>-3</sup> ]	0.05	0.07	0.07	0.01	2				0
	<b>lead (Pb)</b>		[ng m <sup>-3</sup> ]	0.05	2.87	2.52	1.61	93	2.05	1.84	1.17	100
	<b>bismuth (Bi)</b>		[ng m <sup>-3</sup> ]	0.05	0.26	0.12	0.46	29	0.19	0.12	0.19	19
	<b>thorium (Th)</b>		[ng m <sup>-3</sup> ]	0.05	0.40	0.34	0.28	89	0.26	0.22	0.18	97
	<b>uranium (U)</b>		[ng m <sup>-3</sup> ]	0.05	0.09	0.09	0.03	48	0.10	0.08	0.06	36
<b>polyaromatic hydrocarbons (PAH)</b>	<b>phenanthrene (Phe)</b>	(Besombes et al., 2001)	[ng m <sup>-3</sup> ]	0.008	0.036	0.032	0.021	88	0.045	0.040	0.025	100
	<b>anthracene (An)</b>		[ng m <sup>-3</sup> ]	0.001	0.004	0.003	0.002	90	0.005	0.004	0.003	99
	<b>fluoranthene (Fla)</b>		[ng m <sup>-3</sup> ]	0.002	0.102	0.069	0.097	90	0.083	0.052	0.086	102
	<b>pyrene (Pyr)</b>		[ng m <sup>-3</sup> ]	0.003	0.123	0.082	0.119	91	0.105	0.070	0.107	102
	<b>triphenylene (Tri)</b>		[ng m <sup>-3</sup> ]	0.002	0.080	0.063	0.070	91	0.052	0.036	0.052	101
	<b>retene (Ret)</b>		[ng m <sup>-3</sup> ]	0.000	0.103	0.039	0.152	87	0.045	0.019	0.054	91
	<b>benzo(a)anthracene (BaA)</b>		[ng m <sup>-3</sup> ]	0.008	0.196	0.132	0.171	90	0.147	0.092	0.152	97
	<b>chrysene (Chr)</b>		[ng m <sup>-3</sup> ]	0.004	0.235	0.200	0.182	91	0.165	0.115	0.165	102
	<b>benzo(e)pyrene (BeP)</b>		[ng m <sup>-3</sup> ]	0.005	0.273	0.215	0.282	91	0.228	0.170	0.235	102
	<b>benzo(b)fluoranthene (BbF)</b>		[ng m <sup>-3</sup> ]	0.005	0.250	0.221	0.164	90	0.202	0.165	0.162	102
	<b>benzo(k)fluoranthene (BkF)</b>		[ng m <sup>-3</sup> ]	0.002	0.104	0.096	0.070	90	0.081	0.063	0.067	100
	<b>benzo(a)pyrene (BaP)</b>		[ng m <sup>-3</sup> ]	0.002	0.122	0.094	0.108	90	0.124	0.085	0.116	101
	<b>benzo(g,h,i)perylene (BghiP)</b>		[ng m <sup>-3</sup> ]	0.008	0.410	0.387	0.228	90	0.404	0.351	0.291	101
	<b>dibenzo(a,h)anthracene (DBahA)</b>		[ng m <sup>-3</sup> ]	0.000	0.010	0.006	0.010	88	0.008	0.005	0.009	101
	<b>indeno(1,2,3-cd)pyrene (IP)</b>		[ng m <sup>-3</sup> ]	0.005	0.218	0.214	0.136	91	0.196	0.155	0.144	102
	<b>coronene (Cor)</b>		[ng m <sup>-3</sup> ]	0.002	0.251	0.237	0.143	91	0.268	0.216	0.190	100
<b>alkanes</b>	<b>C11</b>	(GS-MS)	[ng m <sup>-3</sup> ]	0.070	0.698	0.712	0.358	16	0.559	0.455	0.431	20
	<b>C12</b>		[ng m <sup>-3</sup> ]	0.170	0.254	0.254	0.052	4	1.588	0.865	1.846	3
	<b>C13</b>		[ng m <sup>-3</sup> ]	0.115	0.217	0.184	0.095	19	0.284	0.231	0.244	14
	<b>C14</b>		[ng m <sup>-3</sup> ]	0.070	0.126	0.107	0.036	7	0.121	0.108	0.024	3
	<b>C15</b>		[ng m <sup>-3</sup> ]	0.070	0.381	0.103	0.686	6	0.253	0.124	0.385	12
	<b>C16</b>		[ng m <sup>-3</sup> ]	0.115	0.194	0.170	0.049	3	0.224	0.176	0.119	6

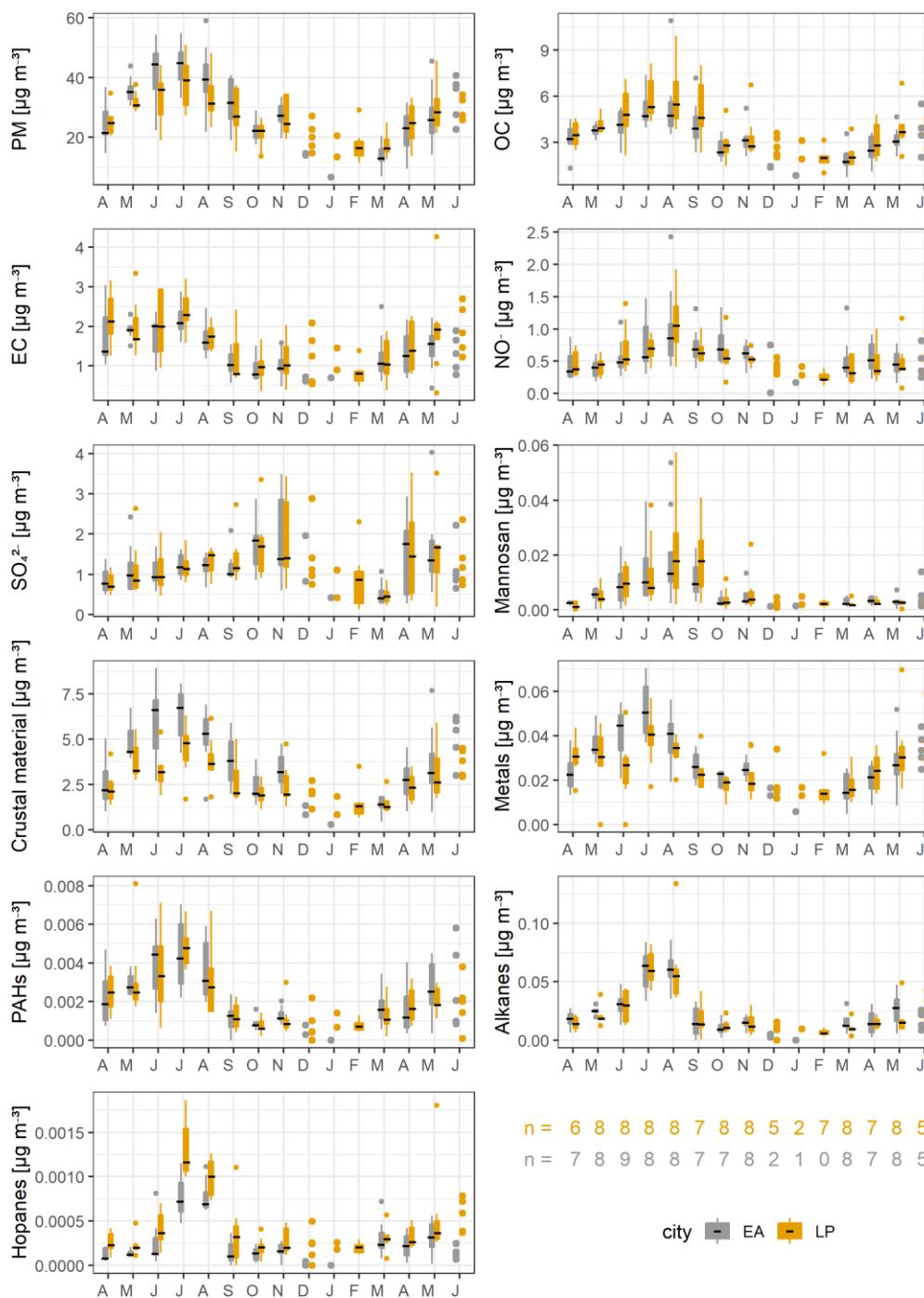
<b>C17</b>	(Golly, 2014)	[ng m <sup>-3</sup> ]	0.138	0.228	0.219	0.061	33	0.236	0.229	0.058	20
<b>C18</b>		[ng m <sup>-3</sup> ]	0.128	0.279	0.273	0.117	47	0.292	0.273	0.198	39
<b>C19</b>		[ng m <sup>-3</sup> ]	0.109	0.571	0.548	0.404	62	0.449	0.376	0.271	55
<b>C20</b>		[ng m <sup>-3</sup> ]	0.110	1.302	1.093	1.106	71	0.908	0.718	0.783	74
<b>C21</b>		[ng m <sup>-3</sup> ]	0.339	2.375	1.691	2.056	75	1.674	0.923	1.611	93
<b>C22</b>		[ng m <sup>-3</sup> ]	0.186	3.118	2.595	2.762	87	2.060	1.128	2.139	101
<b>C23</b>		[ng m <sup>-3</sup> ]	0.314	3.149	2.634	2.552	86	2.482	1.398	2.443	100
<b>C24</b>		[ng m <sup>-3</sup> ]	0.231	2.760	2.240	2.157	89	2.278	1.414	2.168	100
<b>C25</b>		[ng m <sup>-3</sup> ]	0.598	2.849	2.265	1.889	83	2.500	1.692	1.991	97
<b>C26</b>		[ng m <sup>-3</sup> ]	0.175	2.229	1.917	1.394	90	2.091	1.479	1.627	101
<b>C27</b>		[ng m <sup>-3</sup> ]	0.580	2.678	2.148	1.847	84	2.714	1.982	2.255	100
<b>C28</b>		[ng m <sup>-3</sup> ]	0.292	1.719	1.555	1.049	88	1.725	1.251	1.320	100
<b>C29</b>		[ng m <sup>-3</sup> ]	0.841	2.891	2.306	2.015	83	2.920	1.861	3.077	93
<b>C30</b>		[ng m <sup>-3</sup> ]	0.288	1.254	1.048	0.810	86	1.205	0.846	0.967	95
<b>C31</b>		[ng m <sup>-3</sup> ]	0.570	2.348	1.647	1.785	82	2.300	1.429	2.464	83
<b>C32</b>		[ng m <sup>-3</sup> ]	0.070	0.807	0.652	0.584	84	0.747	0.540	0.669	97
<b>C33</b>		[ng m <sup>-3</sup> ]	0.070	0.755	0.658	0.538	84	0.825	0.577	0.830	89
<b>C34</b>		[ng m <sup>-3</sup> ]	0.070	0.564	0.429	0.423	73	0.480	0.359	0.392	82
<b>C35</b>		[ng m <sup>-3</sup> ]	0.070	0.629	0.429	0.556	68	0.484	0.293	0.474	77
<b>C36</b>		[ng m <sup>-3</sup> ]	0.070	0.381	0.249	0.358	53	0.357	0.271	0.321	52
<b>C37</b>		[ng m <sup>-3</sup> ]	0.070	0.394	0.255	0.468	36	0.198	0.166	0.151	24
<b>C38</b>		[ng m <sup>-3</sup> ]	0.070	0.281	0.187	0.258	15	0.175	0.128	0.126	18
<b>C39</b>		[ng m <sup>-3</sup> ]	0.070	0.351	0.224	0.362	17	0.164	0.114	0.122	24
<b>C40</b>		[ng m <sup>-3</sup> ]	0.070	0.319	0.259	0.231	12	0.106	0.100	0.025	9
<b>Pristane</b>		[ng m <sup>-3</sup> ]	0.070	0.189	0.147	0.164	25	0.258	0.163	0.204	24
<b>Phytane</b>		[ng m <sup>-3</sup> ]	0.070	0.180	0.161	0.082	23	0.196	0.150	0.139	36
<b>Methyl PAH</b>	2-methyl-naphthalene	[ng m <sup>-3</sup> ]	0.019				0				0
	1-methyl-fluorene	[ng m <sup>-3</sup> ]	0.007				0	0.040	0.011	0.051	3
	3-methyl-phenanthrene	[ng m <sup>-3</sup> ]	0.025				0	0.125	0.125	0.141	2
	2-methyl-phenanthrene	[ng m <sup>-3</sup> ]	0.014				0	0.018	0.017	0.005	7
	2-methyl-anthracene	[ng m <sup>-3</sup> ]	0.014				0	0.041	0.041	0.035	2
	4/9-methyl-phenanthrene	[ng m <sup>-3</sup> ]	0.014				0	0.040	0.040		1
	1-methyl-phenanthrene	[ng m <sup>-3</sup> ]	0.007	0.010	0.010		1	0.022	0.013	0.029	9
	4-methyl-pyrene	[ng m <sup>-3</sup> ]	0.007	0.019	0.015	0.012	25	0.022	0.015	0.015	32

	<b>1-methyl-pyrene</b>	[ng m <sup>-3</sup> ]	0.007	0.024	0.017	0.016	27	0.025	0.017	0.018	37
	<b>1+3-Methyl-fluorene</b>	[ng m <sup>-3</sup> ]	0.007	0.016	0.015	0.005	13	0.017	0.014	0.006	14
	<b>Methyl-fluorene/pyrene</b>	[ng m <sup>-3</sup> ]	0.007	0.015	0.011	0.008	20	0.016	0.016	0.007	21
	<b>1-methylfluoranthene</b>	[ng m <sup>-3</sup> ]	0.007	0.017	0.015	0.009	21	0.020	0.014	0.013	26
	<b>3-methyl-chrysene</b>	[ng m <sup>-3</sup> ]	0.008	0.034	0.031	0.022	32	0.039	0.026	0.037	54
	<b>Methyl-chrysene/BenzoAnthracene</b>	[ng m <sup>-3</sup> ]	0.007	0.016	0.014	0.008	28	0.022	0.016	0.016	39
<b>thiophens</b>	<b>DBT (DiBenzoThiophen)</b>	[ng m <sup>-3</sup> ]	0.007	0.011	0.011	0.003	4	0.024	0.013	0.022	7
	<b>PheT(4,5) (Phenanthro(4,5 bcd)Thiophen)</b>	[ng m <sup>-3</sup> ]	0.007	0.012	0.011	0.004	4	0.016	0.011	0.016	16
	<b>BNT(2,1) (Benzo(b)Naphto(2,1 d)Thiophen)</b>	[ng m <sup>-3</sup> ]	0.027	0.027	0.027	0.006	7	0.040	0.039	0.020	14
	<b>BNT(1,2) (Benzo(b)Naphto(1,2 d)Thiophen)</b>	[ng m <sup>-3</sup> ]	0.007	0.009	0.010	0.001	3	0.016	0.010	0.009	9
	<b>BNT(2,3) (Benzo(b)Naphto(2,3 d)Thiophen)</b>	[ng m <sup>-3</sup> ]	0.007	0.010	0.010	0.002	12	0.017	0.015	0.009	12
	<b>DNT(2,1) (Dinaphto (2,1) Thiophen)</b>	[ng m <sup>-3</sup> ]	0.007	0.014	0.011	0.008	3	0.014	0.012	0.006	3
	<b>BPT(2,1) (Benzo(b)Phenanto(2,1d)thiophen)</b>	[ng m <sup>-3</sup> ]	0.007				0				0
<b>hopanes</b>	<b>HP1 (Trisnorneohopane)</b>	[ng m <sup>-3</sup> ]	0.012	0.060	0.041	0.048	46	0.074	0.049	0.065	64
	<b>HP2 (17<math>\alpha</math>(H)-Trisnorhopane)</b>	[ng m <sup>-3</sup> ]	0.013	0.047	0.038	0.031	54	0.056	0.039	0.044	94
	<b>HP3 (17<math>\alpha</math>(H)-21<math>\beta</math>(H)-Norhopane)</b>	[ng m <sup>-3</sup> ]	0.007	0.143	0.097	0.122	87	0.215	0.147	0.183	99
	<b>HP4 (17<math>\alpha</math>(H)-21<math>\beta</math>(H)-Hopane)</b>	[ng m <sup>-3</sup> ]	0.011	0.179	0.119	0.150	83	0.255	0.168	0.212	99
	<b>HP5 (17<math>\alpha</math>(H)-21<math>\beta</math>(H)-22S-Homohopane)</b>	[ng m <sup>-3</sup> ]	0.007	0.077	0.050	0.064	77	0.109	0.069	0.093	98
	<b>HP6 (17<math>\alpha</math>(H)-21<math>\beta</math>(H)-22R-Homohopane)</b>	[ng m <sup>-3</sup> ]	0.007	0.059	0.035	0.053	72	0.081	0.050	0.074	99
	<b>HP7 (17<math>\alpha</math>(H)-21<math>\beta</math>(H)-22S-Bishomohopane)</b>	[ng m <sup>-3</sup> ]	0.007	0.057	0.036	0.053	73	0.077	0.043	0.072	100
	<b>HP8 (17<math>\alpha</math>(H)-21<math>\beta</math>(H)-22R-Bishomohopane)</b>	[ng m <sup>-3</sup> ]	0.007	0.042	0.026	0.034	66	0.058	0.037	0.051	94
	<b>HP9 (17<math>\alpha</math>(H)-21<math>\beta</math>(H)-22S-Trishomohopane)</b>	[ng m <sup>-3</sup> ]	0.007	0.044	0.023	0.043	68	0.057	0.031	0.055	89
	<b>HP10 (17<math>\alpha</math>(H)-21<math>\beta</math>(H)-22R-Trishomohopane)</b>	[ng m <sup>-3</sup> ]	0.007	0.038	0.024	0.028	57	0.043	0.024	0.039	77
<b>methoxyphenols</b>	<b>DMPT (6,10,14-trimethyl-2-pentadecanone)</b>	[ng m <sup>-3</sup> ]	0.055	0.749	0.524	0.637	89	0.984	0.631	1.345	102
	<b>Vanillin</b>	[ng m <sup>-3</sup> ]	0.017	0.199	0.162	0.119	22	0.172	0.144	0.101	22
	<b>Acetovanillone</b>	[ng m <sup>-3</sup> ]	0.055	0.280	0.236	0.169	27	0.272	0.095	0.520	16
	<b>Guaiacyl acetone</b>	[ng m <sup>-3</sup> ]	0.057	0.239	0.226	0.117	15	0.161	0.143	0.086	14
	<b>Coniferylaldehyde</b>	[ng m <sup>-3</sup> ]	0.056	0.260	0.224	0.137	4	0.352	0.281	0.199	4

	<b>Vanillic acid</b>	[ng m <sup>-3</sup> ]	0.018	0.577	0.423	0.449	46	0.713	0.533	0.723	40
	<b>Homovanillic acid</b>	[ng m <sup>-3</sup> ]	0.014	0.218	0.175	0.079	3	0.087	0.087	0.034	2
	<b>Syringol</b>	[ng m <sup>-3</sup> ]	0.006				0	0.066	0.066		1
	<b>4-methylsyringol</b>	[ng m <sup>-3</sup> ]	0.006				0	0.032	0.032		1
	<b>4-propenylsyringol</b>	[ng m <sup>-3</sup> ]	0.028				0	0.093	0.093		1
	<b>Acetosyringone</b>	[ng m <sup>-3</sup> ]	0.028	0.398	0.353	0.227	33	0.420	0.409	0.220	31
	<b>Syringyl acetone</b>	[ng m <sup>-3</sup> ]	0.017	0.181	0.181		1	0.089	0.089		1
	<b>Sinapyl aldehyde</b>	[ng m <sup>-3</sup> ]	0.056	0.160	0.160		1	0.348	0.348	0.204	2
	<b>Syringic acid</b>	[ng m <sup>-3</sup> ]	0.018	0.322	0.283	0.182	34	0.442	0.331	0.328	33
<b>sterols</b>	<b>Cholesterol</b>	[ng m <sup>-3</sup> ]	0.056	2.300	1.926	1.412	11	1.534	1.413	0.858	22
<b>methyl-nitricatechols</b>	<b>3methylcatechol</b>	[ng m <sup>-3</sup> ]	0.070				0	0.133	0.133		1
	<b>4-methylcatechol</b>	[ng m <sup>-3</sup> ]	0.084				0				0
	<b>4nitroguaiacol</b>	[ng m <sup>-3</sup> ]	0.300	0.414	0.414		1				0
	<b>4nitrocatechol</b>	[ng m <sup>-3</sup> ]	0.140	1.001	0.806	0.601	14	1.602	1.458	1.130	22
	<b>3-methyl-6-nitrocatechol</b>	[ng m <sup>-3</sup> ]	0.124				0				0
	<b>4-methyl-5-nitrocatechol</b>	[ng m <sup>-3</sup> ]	0.140	0.236	0.248	0.022	3	0.381	0.334	0.165	12
	<b>3-methyl-5-nitrocatechol</b>	[ng m <sup>-3</sup> ]	0.070	0.355	0.292	0.122	3	0.523	0.581	0.245	13
	<b>3methyl4nitrocatechol</b>	[ng m <sup>-3</sup> ]	0.124	0.195	0.195		1	0.925	0.281	1.096	7

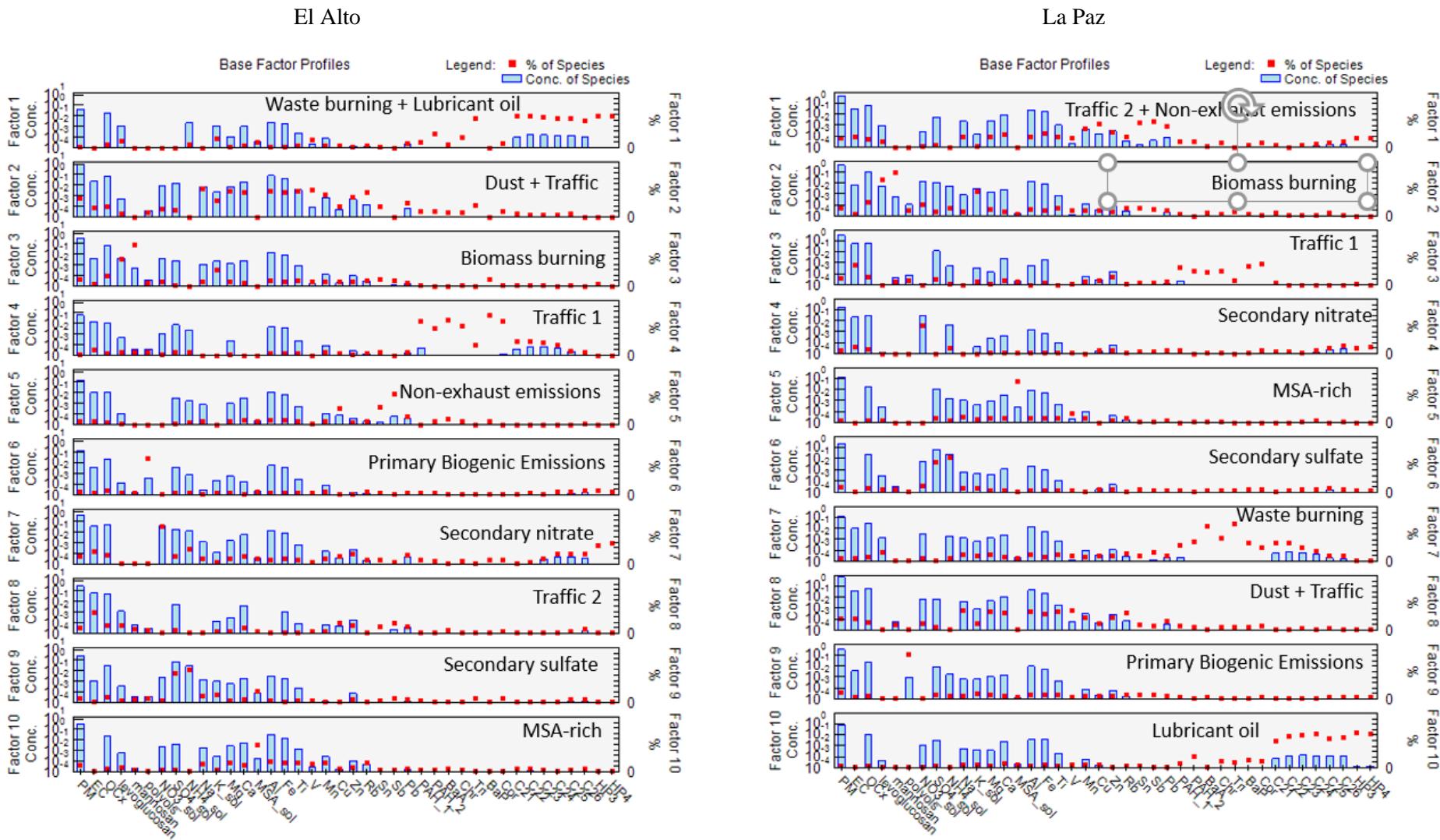
Table S3. Species analyzed from fuel samples collected at La Paz and El Alto

	<b>Sample#</b>	<b>Al</b>	<b>Cr</b>	<b>Mn</b>	<b>Fe</b>	<b>Co</b>	<b>Ni</b>	<b>Cu</b>	<b>Zn</b>	<b>As</b>	<b>Ag</b>	<b>Cd</b>	<b>Pb</b>
		mg/l											
<i>gasoline</i>	1	13.7	3.96	63.0	2.81	0.05	0.23	0	0.85	4.00	0.12	0.002	0.10
<i>gasoline</i>	2	8.23	3.84	67.3	0.00	0.03	0.10	0	1.68	3.47	0.09	0.003	0.10
<i>gasoline</i>	3	8.54	3.51	61.7	0.00	0.03	0.07	0	1.71	3.07	0.01	0.004	0.10
<i>diesel</i>	4	14.9	9.82	0.66	0.00	0.02	0.13	0	7.85	4.00	0.01	0.005	0.25
<i>diesel</i>	5	15.8	8.81	0.41	0.00	0.02	0.13	0	6.55	3.74	0.00	0.007	0.27
<i>diesel</i>	6	35.0	10.2	0.91	11.6	0.02	0.11	0	7.49	4.36	0.01	0.011	0.28

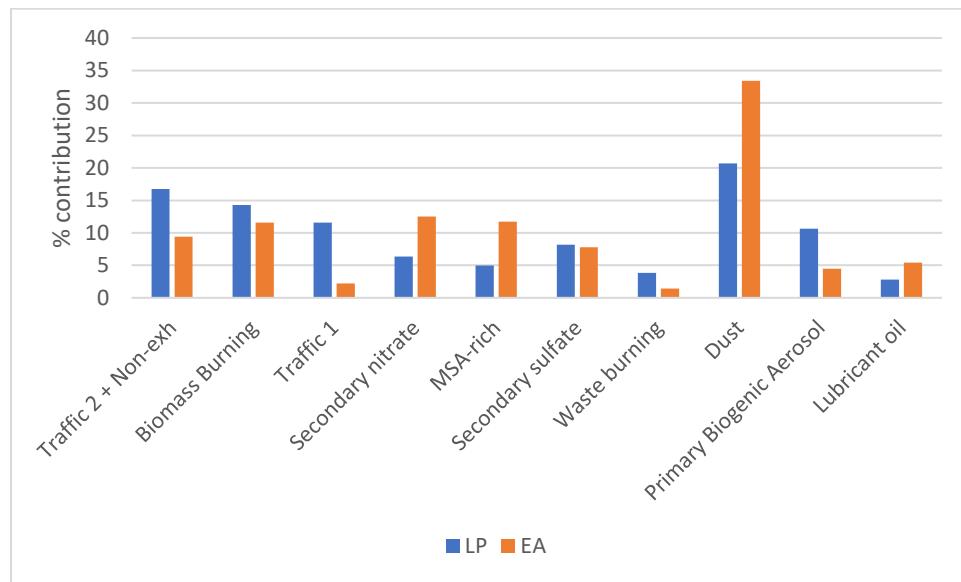


**Figure S3.** Monthly variation of major contributor species to PM<sub>10</sub>, crustal material comprises: Al, Fe, Ti, Ca, Mg, K, Mn, P; metals comprises: Co, Ni, Cu, Zn, As, Rb, Sr, Cd, Sn, Sb, Pb; PAHs is the sum of: Phe, An, Fla, Pyr, Tri, Ret, BaA, Chr, BeP, BbF, BkF, BaP, BghiP, DBahA, IP, Cor; alkanes are the sum of C<sub>20</sub>-C<sub>31</sub>, hopanes the sum of HP3 and HP4; and n is the number of filters collected in the corresponding month at each site.

**Figure S4. Chemical profile of single-site sources of PM<sub>10</sub>**



**Figure S5. Percentage contribution of sources to total ambient PM (single site approach)**



**Table S4. Bootstrap mapping of single site solution El Alto**

**Table S5. Bootstrap mapping of single site solution La Paz**

	Traffic 2 + Non-exhaust	Biomass Burning	Traffic 1	Secondary nitrate	MSA-rich	Secondary sulfate	Waste burning	Dust	Primary Biogenic Aerosols	Lubricant oil	Unmapped
<b>Boot Factor 1</b>	84	1	3	1	1	0	4	3	0	3	0
<b>Boot Factor 2</b>	0	100	0	0	0	0	0	0	0	0	0
<b>Boot Factor 3</b>	11	0	80	0	0	1	5	1	0	2	0
<b>Boot Factor 4</b>	0	0	0	99	0	0	0	0	0	1	0
<b>Boot Factor 5</b>	0	0	0	0	100	0	0	0	0	0	0
<b>Boot Factor 6</b>	0	0	0	0	0	100	0	0	0	0	0
<b>Boot Factor 7</b>	2	0	2	0	0	0	94	1	0	1	0
<b>Boot Factor 8</b>	0	0	0	0	0	0	1	98	0	1	0
<b>Boot Factor 9</b>	0	0	0	0	0	0	0	0	100	0	0
<b>Boot Factor 10</b>	3	0	1	0	0	0	3	1	0	92	0

**Table S6. Spearman correlations between chloride and each of the resolved sources of PM. Strongest correlations are found between Cl- and Waste burning, secondly with TR1 and Non-exhaust.**

		Waste burning	Sec. sulfate	TR1	MSA-rich	Lubricant	BB	Dust	Sec. Nitrate	Non-Exhaust	PBA	TR2
Spearman	Cl- EA	0.75	-0.22	0.57	0.34	0.28	0.47	0.59	-0.24	0.67	-0.19	0.25
	Cl- LP	0.67	0.01	0.61	0.39	0.49	0.53	0.45	0.19	0.57	-0.25	0.44

#### Test of Similarity of Chemical Profiles

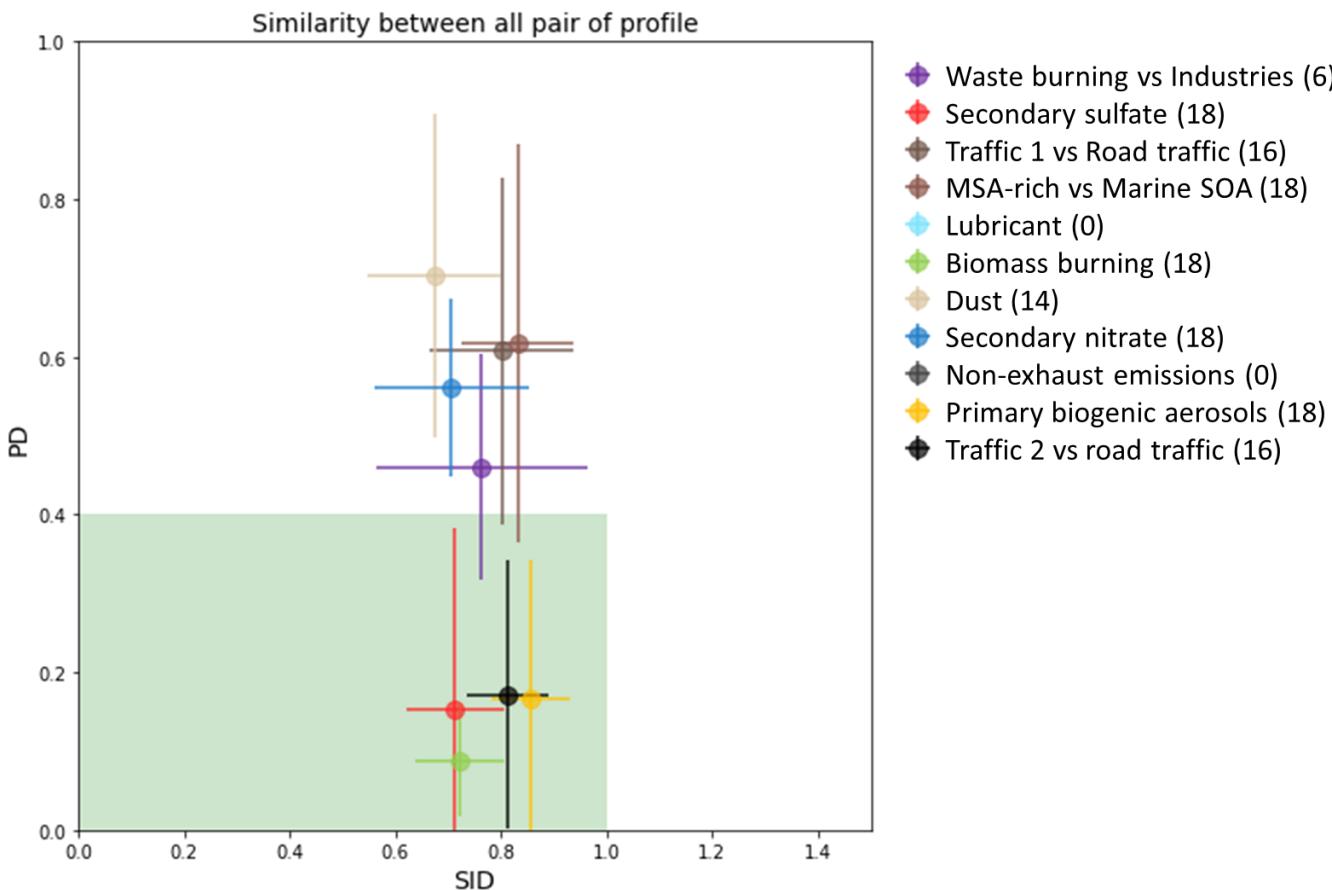
In source apportionment studies, factors are labeled according to their chemical profile and its similarity to what was previously reported in the literature. Although the source identification is based on the specific tracers of the different sources, the exact chemical profile of sources with the same name could vary from site to site. Thus, a metric to quantitatively evaluate the similarity between two factors was proposed by Belis et al. 2015 and Pernigotti & Belis, 2018. Two parameters are considered to establish given similarity: the Pearson distance (PD) and the similarity Identity distance (SID), defined as follows:

$$PD = 1 - r^2 \quad [S1]$$

$$SID = \frac{\sqrt{2}}{m} \sum_{j=1}^m \frac{|x_j - y_j|}{x_j + y_j} \quad [S2]$$

Where m is the number of common species existing between the two compared profiles and, x and y are the relative mass of the species j in each factor profile. According to Pernigotti & Belis, 2018, PD < 0.4 and SID < 1 are considered as acceptable criteria for profile similarity.

This method was used to evaluate the similarity of the resolved factors in La Paz-El Alto with what was found in France by Borlaza et al., 2021 and Weber et al., 2019 (Figure S6). From the 11 resolved sources, only 8 were comparable with what was obtained from the French-sites source apportionment, out of which 4 resulted to be significantly similar to what was observed in France: primary biogenic aerosols, biomass burning, secondary sulfate and traffic 2. Although waste burning was not among the identified sources in France, this factor was compared to the industrial emissions observed in France. The disparities between the remaining factors (dust, secondary nitrate, MSA-rich and traffic 1) was largest in the PD parameter, which is highly sensitive to variation in the major mass fractions of the PM, and will be discussed below.



**Figure S6.** Similarity plot of the sources of  $\text{PM}_{10}$  in La Paz-El Alto and the sources identified by Borlaza et al. 2021 and Weber et al. 2019, by pairs, of profiles belonging to the same factor/source category. The mean  $\pm$  standard deviation (sd) for each source category are plotted (circles correspond to the mean and lines to the sd on both axis). The green box highlights the acceptable area for profile similarity according to Pernigotti & Belis, 2018.

It was observed that the largest difference between the dust profiles, was the relative abundance of EC, OC, sulfate and nitrate assigned to the factors. The abundance of such species in the dust profiles in France was considerably higher compared to La Paz. In contrast, the relative concentrations of Al and Fe were generally higher for LP-EA. This shows that not only the composition of dust but the aging of the air masses carrying it are different. Similarly, for secondary nitrate, the OC and EC relative concentrations found in the secondary nitrate factor in LP-EA was generally higher than in France. This is likely because the main source of secondary-nitrate precursors in LP-EA are vehicular gaseous emissions and since the process through which secondary nitrate is formed is a relatively fast chemical process, the PMF was not able to fully separate both factors. On the other hand, nitrate relative concentrations were 2 to 4 times higher than what was observed in LP\_EA. The largest difference observed in the MSA-rich chemical profiles, that led to high PD values, were the relative abundances of OC, sulfate, Al and Fe. Specifically, OC and sulfate where repeatedly higher in France, which gives an idea of the aging processes that secondary marine organic aerosols undergo. In contrast, in LP-EA, Al and Fe relative masses were significantly higher than in France. This is likely due to the mixing that takes place during the transport of this secondary aerosols across the Altiplano until reaching the

metropolis. Traffic 1 had noticeably higher concentrations of Al and Fe in LP-EA but lower OC, sulfates and nitrates. In contrast, high OC relative concentrations in traffic 2, primary biogenic emissions, secondary sulfate and biomass burning, in both LP-EA and France pull the PD towards lower values.

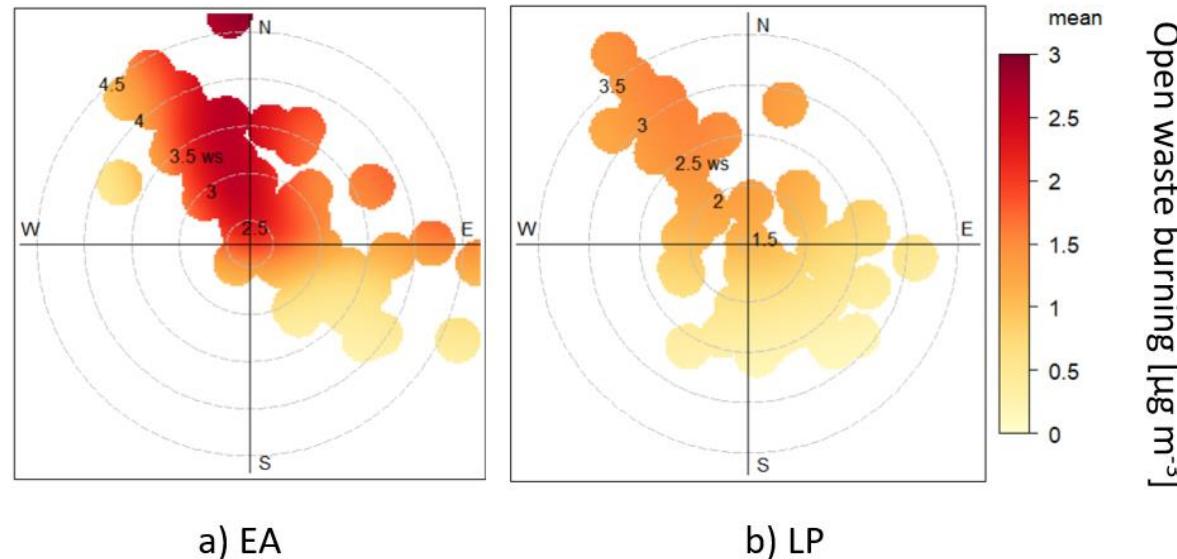


Figure S7. Polar plot showing the mean concentrations attributed to open waste burning and the associated wind speed ( $\text{m s}^{-1}$ ) and wind direction.

## References

- Belis, C. A., Pernigotti, D., Karagulian, F., Pirovano, G., Larsen, B. R., Gerboles, M., and Hopke, P. K.: A new methodology to assess the performance and uncertainty of source apportionment models in intercomparison exercises, *Atmos. Environ.*, 119, 35–44, <https://doi.org/10.1016/j.atmosenv.2015.08.002>, 2015.
- Besombes, J. L., Maître, A., Patissier, O., Marchand, N., Chevron, N., Stoklov, M., and Masclet, P.: Particulate PAHs observed in the surrounding of a municipal incinerator, *Atmos. Environ.*, 35, 6093–6104, [https://doi.org/10.1016/S1352-2310\(01\)00399-5](https://doi.org/10.1016/S1352-2310(01)00399-5), 2001.
- Borlaza, L. J. S., Weber, S., Uzu, G., Jacob, V., Cañete, T., Micallef, S., Trébuchon, C., Slama, R., Favez, O., and Jaffrezo, J. L.: Disparities in particulate matter ( $\text{PM}_{10}$ ) origins and oxidative potential at a city scale (Grenoble, France) - Part 1: Source apportionment at three neighbouring sites, *Atmos. Chem. Phys.*, 21, 5415–5437, <https://doi.org/10.5194/acp-21-5415-2021>, 2021.
- Cavalli, F., Viana, M., Yttri, K. E., Genberg, J., and Putaud, J. P.: Toward a standardized thermal-optical protocol for measuring atmospheric organic and elemental carbon:

The eusaar protocol, *Atmos. Meas. Tech.*, 3, 79–89, 2010.

Golly, B.: Etude des sources et de la dynamique atmosphérique de polluants organiques particulaires en vallées alpines : apport de nouveaux traceurs organiques aux modèles récepteurs, 292, 2014.

Jaffrezo, J. L., Aymoz, G., and Cozic, J.: Size distribution of EC and OC in the aerosol of Alpine valleys during summer and winter, *Atmos. Chem. Phys.*, 5, 2915–2925, <https://doi.org/10.5194/acp-5-2915-2005>, 2005.

Pernigotti, D. and Belis, C. A.: DeltaSA tool for source apportionment benchmarking, description and sensitivity analysis, *Atmos. Environ.*, 180, 138–148, <https://doi.org/10.1016/j.atmosenv.2018.02.046>, 2018.

Piot, C., Jaffrezo, J. L., Cozic, J., Pissot, N., El Haddad, I., Marchand, N., and Besombes, J. L.: Quantification of levoglucosan and its isomers by High Performance Liquid Chromatography-Electrospray Ionization tandem Mass Spectrometry and its applications to atmospheric and soil samples, *Atmos. Meas. Tech.*, 5, 141–148, <https://doi.org/10.5194/amt-5-141-2012>, 2012.

Querol, X., Alastuey, A., Rodriguez, S., Plana, F., Mantilla, E., and Ruiz, C. R.: Monitoring of PM10 and PM2.5 around primary particulate anthropogenic emission sources, *Atmos. Environ.*, 35, 845–858, [https://doi.org/10.1016/S1352-2310\(00\)00387-3](https://doi.org/10.1016/S1352-2310(00)00387-3), 2001.

Weber, S., Salameh, D., Albinet, A., Alleman, L. Y., Waked, A., Besombes, J. L., Jacob, V., Guillaud, G., Meshbah, B., Rocq, B., Hulin, A., Dominik-Sègue, M., Chrétien, E., Jaffrezo, J. L., and Favez, O.: Comparison of PM10 sources profiles at 15 french sites using a harmonized constrained positive matrix factorization approach, *Atmosphere (Basel.)*, 10, 1–22, <https://doi.org/10.3390/atmos10060310>, 2019.