



Supplement of

Chemical composition of secondary organic aerosol particles formed from mixtures of anthropogenic and biogenic precursors

Yunqi Shao et al.

Correspondence to: Yunqi Shao (yunqi.shao@manchester.ac.uk)

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Section 1: Supplementary figures

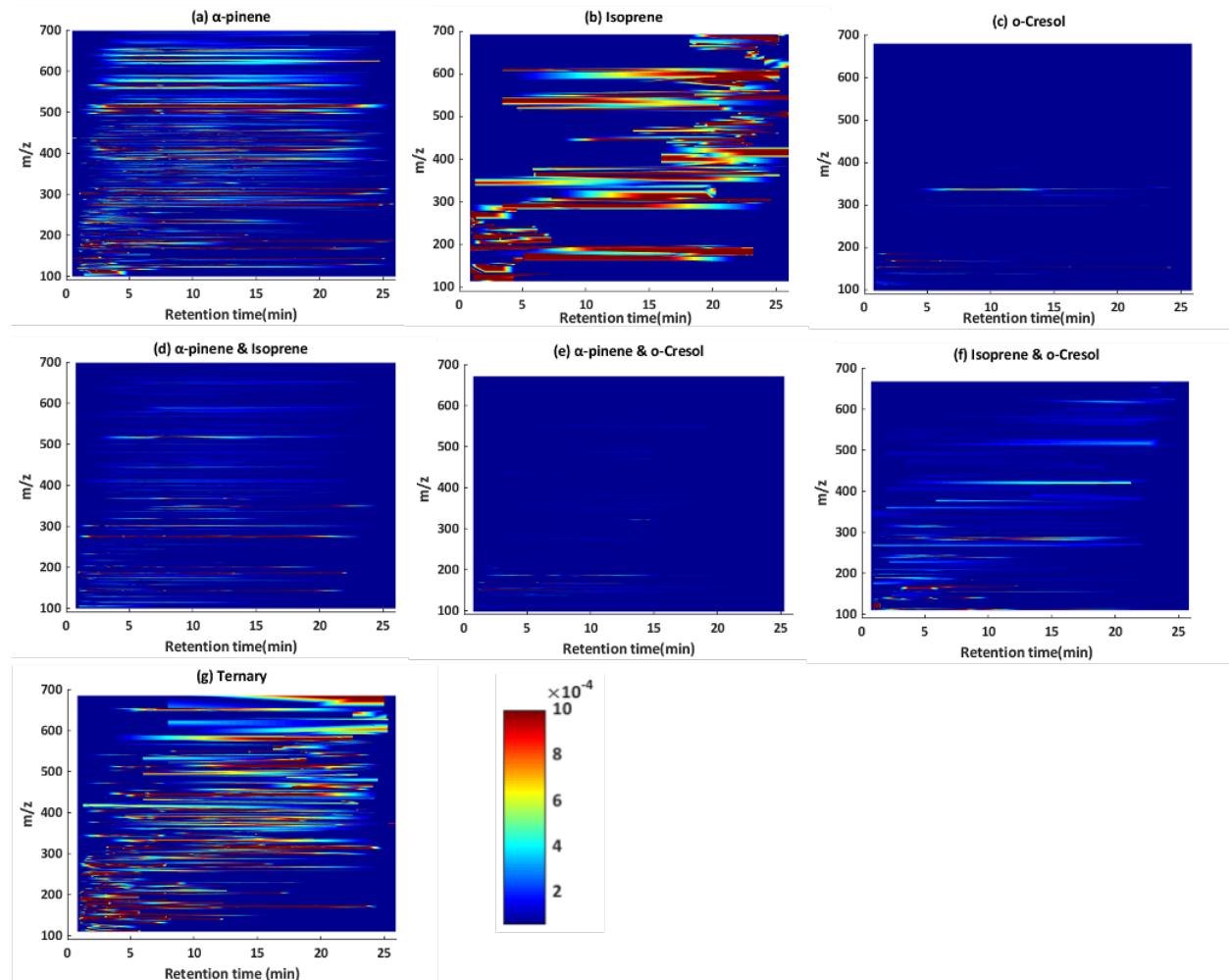


Figure S1: General Mass spectra m/z vs retention time with normalized signal intensity as colour) for compounds detected in Negative ionization mode for single and mixture experiment.

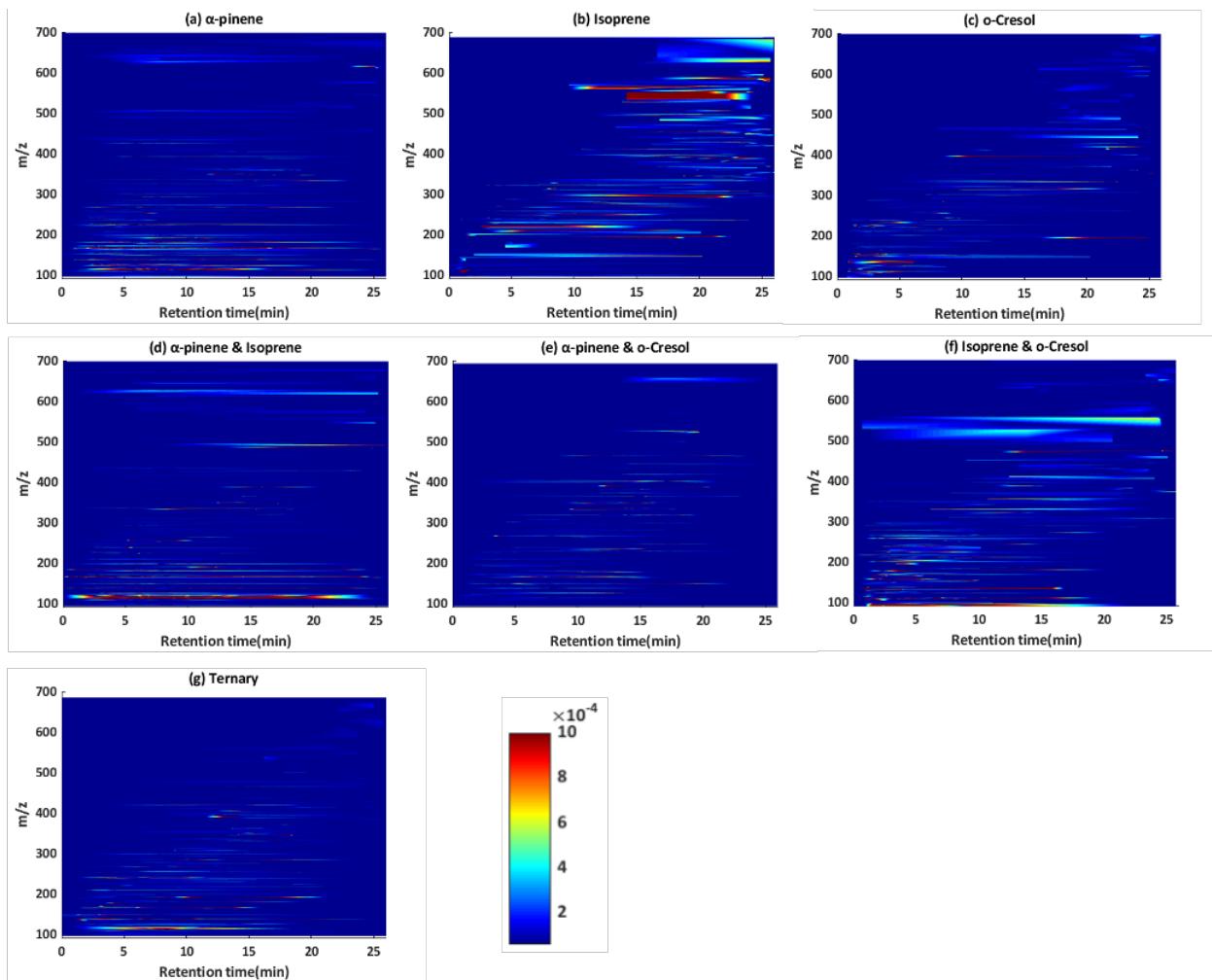


Figure S2: General Mass spectra m/z vs retention time with normalized signal intensity as colour) for compounds detected in Positive ionization mode for single and mixture experiment

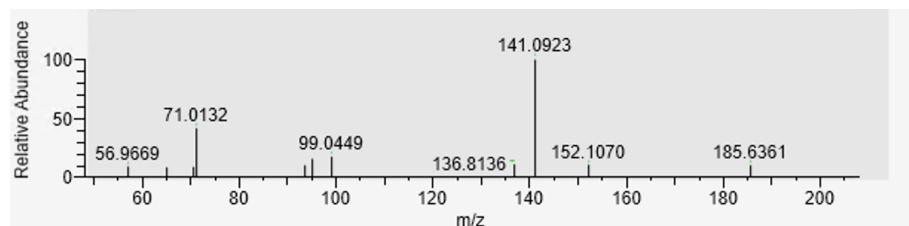


Figure S3: The fragmentation spectra for $C_9H_{14}O_4$ in binary α -pinene/isoprene system.

Section 2: Supplementary Tables

The product ions of deprotonated species of C₇H₇NO₄ and C₇H₇NO₃ in single o-cresol system, binary α -pinene/o-cresol and Isoprene/o-cresol system will be discussed here in order to illustrate the methodology to assign SOA compound structure. For all of the C₇H₇NO₄ isomer, the fragment ion at m/z 138 due to neutral loss of NO group by odd electron cleavage that often happen in ring structures with nitrogen containing group (Hayen et al., 2002; Fu et al., 2006; Pereira et al., 2015). C₇H₇NO₄(i) and (III) also has the combine loss of NO and hydrogen radical (m/z=137), while the 109 m/z fragment corresponding to the combined loss of NO and CO from the deprotonated C₇H₇NO₄. For the C₇H₇NO₃ compound, the fragment ion m/z=122 has formed as a result of loss of NO from m/z 152, the loss of combined NO and hydrogen radicals resulting forming the 121m/z fragment and the combined loss of CH and NO leading to formation of 109 m/z fragment. These fragmentation ions of deprotonated molecules suggested that the C₇H₇NO₄ are the methyl-nitrocatechol and the C₇H₇NO₃ is methyl-nitrophenol(Kitanovski et al., 2012).

Table S1: Deprotonated molecular species for C₇H₇NO₄ isomers and C₇H₇NO₃, obtained from the use of the Orbitrap_LCMS in binary α-pinene/o-cresol and Isoprene/o-cresol system respectively.

Precursors System	Formula	Retention Time (min)	[M-H] ⁻	Fragment ion [m/z]	Loss [Da]	Suspected Fragment ion MF
o-cresol	C ₇ H ₇ NO ₄ (i)	4.46	168	166	2	C ₇ H ₅ NO ₄
				138	30	C ₇ H ₆ O ₃
				122	46	C ₇ H ₆ O ₂
				108	60	C ₆ H ₄ O ₂
	C ₇ H ₇ NO ₄ (ii)	7.40	168	138	30	C ₇ H ₆ O ₃
				137	31	C ₇ H ₅ O ₃
				122	46	C ₇ H ₆ O ₂
				109	59	C ₆ H ₅ O ₂
	C ₇ H ₇ NO ₄ (iii)	8.93	168	138	30	C ₇ H ₆ O ₃
				137	31	C ₇ H ₅ O ₃
				109	59	C ₆ H ₅ O ₂
α-pinene/o-cresol	C ₇ H ₇ NO ₄ (i)	4.52	168	166	2	C ₇ H ₅ NO ₄
				138	30	C ₇ H ₆ O ₃
				108	60	C ₆ H ₄ O ₂
	C ₇ H ₇ NO ₄ (ii)	7.53	168	138	30	C ₇ H ₆ O ₃
				137	31	C ₇ H ₅ O ₃
				109	59	C ₆ H ₅ O ₂
	C ₇ H ₇ NO ₄ (iii)	9.08	168	138	30	C ₇ H ₆ O ₃
				137	31	C ₇ H ₅ O ₃
				109	59	C ₆ H ₅ O ₂
Isoprene/o-cresol	C ₇ H ₇ NO ₃	10.19	152	122	30	C ₇ H ₆ O ₂
				121	31	C ₇ H ₅ O ₂
				109	43	C ₆ H ₅ O ₂
				166	2	C ₇ H ₅ NO ₄
	C ₇ H ₇ NO ₄ (i)	4.52	168	138	30	C ₇ H ₆ O ₃
				137	31	C ₇ H ₅ O ₃
				108	60	C ₆ H ₄ O ₂
				138	30	C ₇ H ₆ O ₃
Isoprene/o-cresol	C ₇ H ₇ NO ₄ (ii)	7.53	168	137	31	C ₇ H ₅ O ₃
				109	59	C ₆ H ₅ O ₂
				138	30	C ₇ H ₆ O ₃
				137	31	C ₇ H ₅ O ₃
	C ₇ H ₇ NO ₄ (iii)	9.14	168	109	59	C ₆ H ₅ O ₂
				122	30	C ₇ H ₆ O ₂
				121	31	C ₇ H ₅ O ₂
				109	43	C ₆ H ₅ O ₂

Table S2: The total normalized peak area and normalized mass concentration attributed to nC>21 molecules that were found in all repeat experiments in selected system. The normalized peak area of nC>21 molecules obtained from the both ionization mode in orbitrap-LCMS, and the particulate mass concentration of each system obtained from HR-TOF-AMS measurement. The normalized mass concentration equals the particulate mass concentration multiply the normalized peak area of nC>21 molecules.

Particulate mass concentration at the end of experiment(ug/m ³)	Negative ionization mode		Positive ionization mode	
	Normalized Peak Area of nC >21 molecules	Normalized mass concentration of nC >21 molecules(ug/m ³)	Normalized Peak Area of nC >21 molecules	Normalized mass concentration of nC >21 molecules(ug/m ³)
a) α-pinene	361	0.008	3.01	0.003
b) Isoprene	0.4	0.007	0.002	0.040
d) α-pinene/isoprene	102	0.011	1.15	0.002
e) α-pinene/o-cresol	150	0.004	0.73	0.019
f) Isoprene/o-cresol	22	0.0006	0.013	0.006
g) Ternary	85	0.002	0.16	0.012

Reference:

Fu, X., Zhang, Y., Shi, S., Gao, F., Wen, D., Li, W., Liao, Y., and Liu, H.: Fragmentation study of hexanitrostilbene by ion trap multiple mass spectrometry and analysis by liquid chromatography/mass spectrometry, Rapid Communications in Mass Spectrometry, 20, 2906-2914, <https://doi.org/10.1002/rcm.2683>, 2006.

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Kitanovski, Z., Grgić, I., Yasmeen, F., Claeys, M., and Čusak, A.: Development of a liquid chromatographic method based on ultraviolet-visible and electrospray ionization mass spectrometric detection for the identification of nitrocatechols and related tracers in biomass burning atmospheric organic aerosol, Rapid Communications in Mass Spectrometry, 26, 793-804, <https://doi.org/10.1002/rcm.6170>, 2012.

Pereira, K. L., Hamilton, J. F., Rickard, A. R., Bloss, W. J., Alam, M. S., Camredon, M., Ward, M. W., Wyche, K. P., Muñoz, A., Vera, T., Vázquez, M., Borrás, E., and Ródenas, M.: Insights into the Formation and Evolution of Individual Compounds in the Particulate Phase during Aromatic Photo-Oxidation, Environmental Science & Technology, 49, 13168-13178, 10.1021/acs.est.5b03377, 2015.