



Enhanced soot particle ice nucleation ability induced by aggregate compaction and densification

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Abstract. Soot particles, acting as ice nucleating particles (INPs), can contribute to cirrus cloud formation, which has an important influence on climate. Aviation activities emitting soot particles into the upper troposphere can potentially impact ice nucleation (IN) in cirrus clouds. Pore condensation and freezing (PCF) is an important ice formation pathway for soot particles in the cirrus regime, which requires the soot INP to have specific morphological properties, i.e., mesopore structures. In this study, the morphology and pore size distribution of two kinds of soot samples were modified by a physical agitation method without any chemical modification by which more compacted soot sample aggregates could be produced compared to the unmodified sample. The IN activities of both fresh and compacted soot particles with different sizes, 60, 100, 200 and 400 nm, were systematically tested by the Horizontal Ice Nucleation Chamber (HINC) under mixed-phase and cirrus-cloud-relevant temperatures (T). Our results show that soot particles are unable to form ice crystals at $T > 235$ K (homogeneous nucleation temperature, HNT), but IN is observed for compacted and larger-sized soot aggregates (> 200 nm) well below the homogeneous freezing relative humidity (RH_{hom}) for $T < \text{HNT}$, demonstrating PCF as the dominating mechanism for soot IN. We also observed that mechanically compacted soot particles can reach a higher particle activation fraction (AF) value for the same T and RH condition compared to the same aggregate size fresh soot particles. The results also reveal a clear size dependence for the IN activity of soot particles with the same degree of compaction, showing that compacted soot particles with large sizes (200 and 400 nm) are more active INPs and can convey the single importance of soot aggregate morphology for the IN ability. In order to understand the role of soot aggregate morphology for its IN activity, both fresh and compacted soot samples were characterized systematically using particle mass and size measurements, comparisons from TEM (transmission electron microscopy) images, soot porosity characteristics from argon (Ar) and nitrogen (N_2) physisorption measurements, as well as soot–water interaction results from DVS (dynamic vapor sorption) measurements. Considering the soot particle physical properties along with its IN activities, the enhanced IN abilities of compacted soot particles are attributed to decreasing mesopore width and increasing mesopore occurrence probability due to the compaction process.

1 Introduction

Black carbon (BC) particles are estimated as the second-most important forcing for climate warming only after CO₂ (Ramanathan and Carmichael, 2008; Bond et al., 2013). BC particles can influence the radiation balance in the atmosphere directly by scattering or absorbing shortwave radiation and indirectly by acting as cloud condensation nuclei (CCN) or ice nucleating particles (INPs) in the atmosphere to form water droplets or ice crystals (Bond et al., 2013; Jacobson, 2004), thereby changing cloud properties. For example, McGraw et al. (2020) suggested that BC particles influence cirrus cloud formation by acting as INPs and competing with the homogeneous freezing of aerosol solution droplets and exhibit large uncertainty in their global net radiative forcing on climate. Recently, Schneider et al. (2021) reported even higher RH_i (relative humidity with respect to ice) for homogeneous freezing of sulfate droplets compared to Koop et al. (2000), which implies that soot particles are even more likely to compete with homogeneous freezing based on the new freezing parameterization suggested in Schneider et al. (2021). As aviation activities emit BC with a great amount of water vapor in contrail plumes under high-altitude and cold-temperature (T) conditions, aviation soot particles can be activated as ice crystals, which potentially regulates cirrus cloud coverage in aviation corridors.

In the cirrus regime, ice crystals can be formed via homogeneous freezing of solution droplets or heterogeneous freezing on the surface of an INP (Cziczo et al., 2013; Lohmann et al., 2020). Homogeneous freezing, requiring low T and high RH, i.e., RH_{hom} (homogeneous freezing relative humidity) conditions, can face competition from heterogeneous freezing at RH < RH_{hom} because INPs lower the energy barrier for the ice embryo formation and facilitate ice crystal activation (Vali et al., 2015), which can deplete water vapor that would otherwise allow humidity levels to reach as high as RH_{hom}. Due to a hydrophobic surface and associated low water interaction ability, soot particles have been assumed to be poor INPs as requiring low T and high RH for ice nucleation (IN). DeMott (1990) investigated the IN activity of acetylene (C₂H₂) soot in an expansion cloud chamber at $T < 253$ K and suggested that soot particles are able to form ice crystals via immersion-mode freezing when T is lower than HNT (homogeneous nucleation temperature) of supercooled liquid water droplets. Möhler et al. (2005) reported that propane (C₃H₈) flame soot particles with low organic carbon (OC) content form ice crystals via deposition nucleation at $T < \text{HNT}$. Kanji and Abbatt (2006) studied the ice nucleation activity of *n*-hexane soot with a cold stage facility under RH conditions below water saturation at cirrus cloud T and observed ice crystals only at the instrument threshold, suggesting that *n*-hexane soot can be activated via deposition nucleation but is a poor INP. Mahrt et al. (2018) studied the IN activities of six kinds of soot particles in a continuous-flow diffusion chamber under mixed-phase and cirrus cloud

conditions. The authors found that porous soot particles are able to form ice crystals below RH_{hom} conditions, which the authors attributed to pore condensation and freezing (PCF) (Marcolli, 2014) instead of deposition nucleation. By contrast, nonporous and hydrophobic soot particles only freeze homogeneously at $T < \text{HNT}$ (Möhler et al., 2005; Koehler et al., 2009; Mahrt et al., 2018). PCF was also reported by Nichman et al. (2019) and Zhang et al. (2020) in their studies of soot IN activity at $T < \text{HNT}$. Recently, Kanji et al. (2020) and Falk et al. (2021) demonstrated that soot particles do not form ice at $T > \text{HNT}$. This finding supports the relevance of soot ice crystal activation via a PCF mechanism under cirrus conditions and rules out the role of immersion-mode freezing.

According to Marcolli (2014) and Marcolli et al. (2021), the PCF process occurs following three steps. Firstly, supercooled water condenses into mesopores (2–50 nm in width) due to the inverse Kelvin effect below water saturation conditions. Next, the supercooled porewater freezes homogeneously at $T < \text{HNT}$ or freezes heterogeneously if active sites are available within the pore. Finally, the pore ice grows out of the pore and forms a macroscopic ice crystal. The PCF mechanism emphasizes the role of pore size distribution (PSD) and soot–water interaction ability in soot particle IN (Marcolli, 2014; David et al., 2020; Marcolli et al., 2021). Given the large heterogeneity of soot properties including chemical characteristics, like chemical composition, surface polarity and soot–water contact angle, as well as physical properties, like aggregate mobility size, particle fractal dimension (D_f), soot porosity and PSD, the dominating predictor of PCF for soot particles is still to be revealed. Previous studies suggested that the water interaction history of soot particles, e.g., water droplet or ice crystal formation processes, can lead to enhanced IN for compacted soot aggregates (Colbeck et al., 1990; Ma et al., 2013; China et al., 2015; Bhandari et al., 2019; Mahrt et al., 2020b), suggesting soot aggregate size and the porosity characteristics are crucial for its IN activity.

The IN ability of an INP is known to be dependent on its size (Pruppacher and Klett, 1997; Archuleta et al., 2005; Connolly et al., 2009; Mahrt et al., 2018; Nichman et al., 2019) since active sites on INPs promoting IN scale with the particle surface area. In the case of soot particle and PCF, pore structures generated among primary particle networks in the aggregate are important for soot IN and should also scale with the soot aggregate size. Zhang et al. (2020) suggested that there is a PCF size threshold (~ 200 nm) for soot particles addressed in their study at 227 K, but it depends on the soot type. However, aviation emissions tend to be comprised of soot particles with a number size distribution mode around 100 nm (Bond et al., 2013). For example, the APEX (Aircraft Particle Emissions eXperiment) campaign shows that the majority of soot particles emitted by an aero-engine range from ~ 3 to 100 nm with a geometric number mean diameter (GMD) of ~ 10 –35 nm (Wey et al., 2007; Kinsey et

al., 2010). Nonetheless, larger-sized soot particles from contrail ice crystal residuals were detected in field studies. For instance, aircraft-contrail-released soot particles with a typical size larger than 400 nm were detected in field sampling studies (Twohy and Gandrud, 1998; Petzold et al., 1998). In order to make the IN experimental results both comparable to the literature and relevant to the real atmosphere, soot particles with a mobility size down to 60 nm and up to 400 nm were investigated in this study.

There is some evidence in the literature showing that soot particle IN ability is controlled by its morphology. Nichman et al. (2019) reported that soot samples with lower branching are more IN-active than fractal soot particles. Mahrt et al. (2020b) also reported significant enhancement in the IN ability of C₃H₈ flame soot due to a compacted morphology induced by cloud processing. In a separate study, Mahrt et al. (2020a) observed significantly enhanced soot IN after aging soot particles in water or diluted sulfuric acid (H₂SO₄) solution, suggesting that the particle structure collapse caused by droplet evaporation could contribute to this IN promotion. The above studies either used different soot samples to represent different morphologies or potentially introduced chemical modifications upon changing the particle morphology, and thus a change in chemical composition cannot be excluded and the exclusive role of morphology in soot ice nucleation cannot be differentiated. In this study, soot particle morphology was changed only by mechanical stirring, preserving the chemical composition to avoid confounding effects of chemical changes.

For the purposes of this study, a unique soot sample preparation method was used to generate fresh and mechanically compacted soot particles in the size range spanning from 60 to 400 nm. The particle IN activity was measured in a continuous-flow diffusion chamber under both mixed-phase and cirrus cloud conditions. Systematic measurements to characterize soot particle morphological properties, including soot aggregate mass and size, microscopic images for single soot aggregate morphology, as well as soot sample PSD were performed to interpret the corresponding IN experimental results in addition to soot–water interaction abilities analyzed from soot–water vapor isotherms.

2 Experimental methods

The schematic of the experimental setup is shown in Fig. 1. Firstly, fresh soot samples were directly used or prepared as compacted powders depending on the time of physical agitation. Subsequently, soot powders were either aerosolized by a dry dispersion setup (Fig. 2) to generate aerosol samples or used for bulk sample offline characterization measurements. The soot sample PSD was measured by both argon (Ar) and nitrogen (N₂) sorption techniques, and the soot–water interaction ability was measured by dynamic vapor sorption (DVS) measurements. Aerosolized soot particles were size-

selected by a differential mobility analyser (DMA, classifier 3080, with a 3081 column and a polonium radiation source, TSI Inc.) to generate monodisperse aerosol sample flow, with a mobility size of 60, 100, 200 or 400 nm. The monodisperse aerosol sample flow was diluted by a factor of ~ 6 and split for downstream IN experiments and soot particle online characterization measurements, as depicted in Fig. 1. Soot particle IN abilities were measured by the Horizontal Ice Nucleation Chamber (HINC) (Lacher et al., 2017; Mahrt et al., 2018) based on the continuous-flow diffusion chamber technique developed by Kanji and Abbatt (2009). In parallel, the mass and size distribution of soot particles were measured by a centrifugal particle mass analyser (CPMA, Cambustion Ltd., Cambridge, UK) and a scanning mobility particle sizer (SMPS, Classifier 3082, Column 3081, TSI Inc.) system. Additionally, soot aggregates for transmission electron microscopy (TEM) image analysis were collected by the Zurich Electron Microscope Impactor (ZEMI) (Mahrt et al., 2020b). The excess aerosol flow was pumped to the exhaust.

2.1 Soot sample preparation and aerosolization

In this study, two different types of commercial carbon black samples were used as fresh soot samples, including FW200 and Printex90 (PR90, Orion Engineered Carbons GmbH, OEC, Frankfurt am Main, Germany). FW200 and PR90 samples are used because they are commercially available carbon black products which allow for (1) large sample sizes, as was required in this study, and (2) the comparison and validation of experiment reproducibility. Furthermore, the primary particle sizes of FW200 and PR90 are about 13 and 14 nm, respectively (see Table 1); both are close to aviation soot particles, which have a mean primary particle diameter of 15 nm (Delhaye et al., 2017). Moreover, FW200 and PR90 contain different volatile content of 20 % and 1 %, respectively (see Table 1). This difference also makes our samples representative of atmospheric soot particles, with varying volatile content caused by aging processes during their transportation (Li et al., 2018; Ditas et al., 2018). As shown in Fig. 2, the Teflon-coated magnetic bar was used to stir and transfer a part of kinetic energy to soot powders, causing the displacement of single primary particles, which directly changes the soot sample porosity and makes it more compacted and densified compared to the fresh sample. In the first several hours of agitation, the masses of 400 nm FW200 (Fig. 3a) and PR90 (Fig. 3b) particles are ~ 10.8 and ~ 13.2 fg initially, and the mass value increases to plateau values of ~ 14.3 and ~ 16.2 fg after a 3 h agitation. The particle mass of a fixed mobility size continues to increase, which means the particle effective density is also increasing and signifies compaction and densification. As shown in Fig. 3, size-selected 400 nm soot particle mass is reasonably constant after 2-week agitation as the mass variability falls within the uncertainty range of ± 0.6 % and ± 1.5 % for FW200 and PR90 soot 400 nm particles, respectively. This indicates the maximum degree

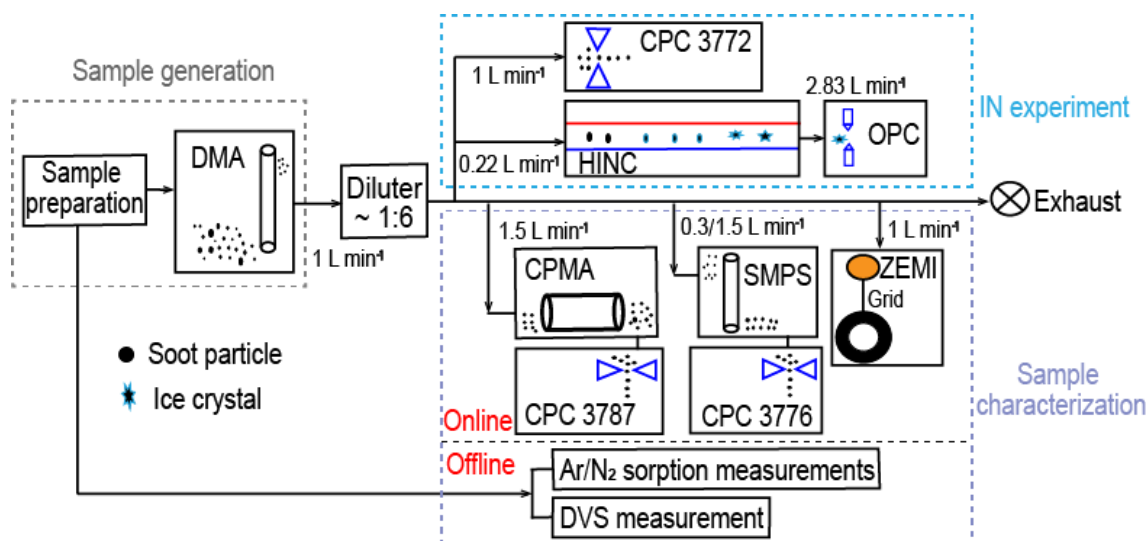


Figure 1. The experimental schematic. The arrows show aerosol flow direction. DMA: Differential Mobility Analyser; CPC: Condensation Particle Counter; HINC: Horizontal Ice Nucleation Chamber; OPC: Optical Particle Counter; CPMA: Centrifugal Particle Mass Analyser; ZEMI: Zurich Electron Microscope Impactor; DVS: dynamic vapor sorption.

of compaction is achieved within 3 h of mechanical agitation. When the same sample has been stirred/compacted for an additional 2 weeks, no further mass increase was detected. Compared to the fresh soot sample, the number size distribution mode of the polydisperse agitated particles shifts to a smaller value (Fig. A1), showing that the peak size mode shifts from ~ 257 nm for fresh FW200 soot to ~ 219 nm for the densified sample, and that of fresh PR90 soot changes from ~ 300 to ~ 264 nm. The change in aerosol particle size distribution (Fig. A1) covers the size range from ~ 50 to 400 nm and from ~ 60 to 800 nm for FW200 and PR90 soot, respectively, which encompasses the monodisperse size ranges (60 to 400 nm) investigated in this study. It also suggests that soot particle morphology change induced by agitation occurs for all the size-selected soot samples in this study.

In the following experiments, fresh soot powder was dispersed into soot aerosol and referred to fresh samples, termed FW200fresh and PR90fresh. Compacted soot particles, called FW200comp and PR90comp hereafter, mean soot samples which were agitated for at least 2 weeks and that the mass of a size-selected soot particle already reaches the stable value similar to the case shown in Fig. 3. In total, four kinds of soot samples (FW200, FW200comp, PR90 and PR90comp) were used in this study at four different sizes. Simultaneously with agitation, the dry soot powder while being stirred (fresh or compacted for more than 2 weeks) was aerosolized with a N_2 flow of velocity of ~ 35 m s $^{-1}$ through a venturi nozzle, creating large enough shear forces to break down a dispersed soot agglomerate into small aggregates. A distinction of this soot particle “aging” method is that no chemical aging effects are involved, thus allowing us exclu-

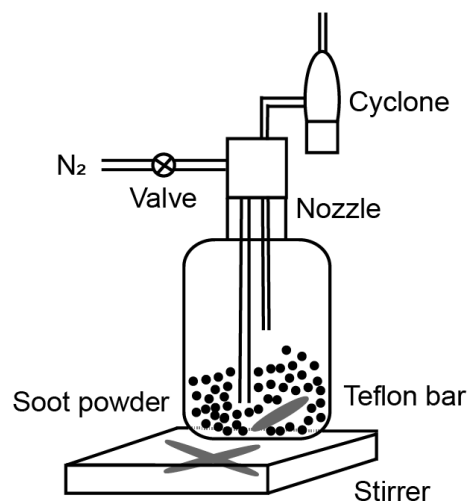


Figure 2. The schematic of the soot aerosol sample generation setup.

sively to investigate the influence of morphology changes on soot IN ability.

2.2 Ice nucleation experiments

The IN experiments were performed by running RH scans in HINC from ice saturation (S_i) values of unity to water saturation (S_w) conditions of 1.1 at T ranging from 243 to 218 K with a step of 5 K. The HINC RH scan was conducted at a S_i rate of 0.02 min $^{-1}$ at each T . The principle of HINC has been explained in detail by Lacher et al. (2017) and Mahrt et al. (2018). In brief, HINC is a continuous-flow dif-

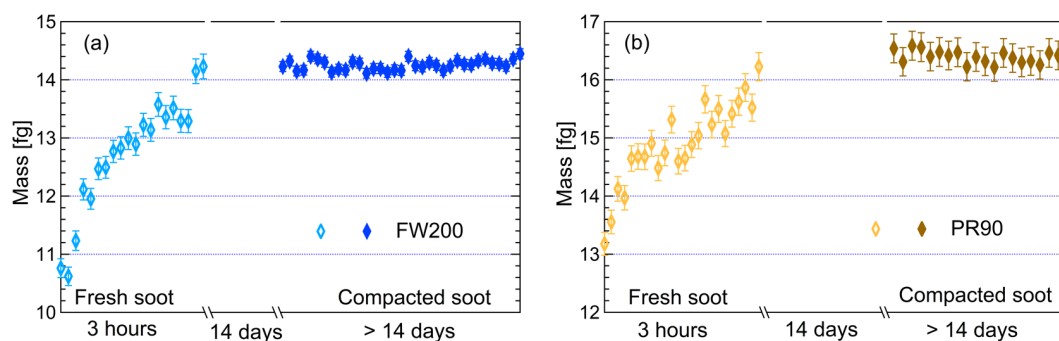


Figure 3. The mass of 400 nm FW200 (a) and PR90 (b) soot particles as a function of agitation time. The measurement uncertainty mainly originated from CPC counting ($\pm 10\%$) (Olfert et al., 2006). The error bars for fresh samples indicate the upper limit of the uncertainty and the error bars for compacted samples represent 1 standard deviation of corresponding measurements.

fusion chamber and creates supersaturation conditions with respect to ice and/or water by using the difference between the diffusion rate of water vapor and air. Two copper plates coated with an ice layer at different T ($T_{\text{top}} > T_{\text{bottom}}$) form the nucleation and growth parts of the chamber. The aerosol sample flow is jacketed between two N_2 sheath flows with a ratio of 1 : 12 and is confined in the center of the chamber, where the T and RH conditions for the aerosol particles can be controlled by adjusting the T difference between T_{top} and T_{bottom} . The RH uncertainty range for each RH and T condition during the RH scan in the chamber is influenced by the T uncertainty of four thermocouples mounted on each chamber wall, which is ± 0.1 K. According to Mahrt et al. (2018), the RH uncertainty at $T = 218$ K is reported to be $\pm 5\%$ RH_w when conditioning the chamber at 105 % RH_w with a sample-to-sheath-flow ratio of 1 : 12, and the uncertainty is smaller under warmer T and lower RH conditions. By feeding in soot sample aerosol with a movable injector into the chamber at a fixed position, the IN activity of the soot particles with a transition time ~ 14 s under a fixed T and RH condition was measured. In order to calculate the activation fraction (AF; see Eq. 1), a Condensation Particle Counter (CPC; Model 3772, TSI Inc.) was used to monitor the number concentration of particles entering the chamber, and the number of water droplets or ice crystals exiting the chamber is recorded by an Optical Particle Counter (OPC; MetOne, GT-526S) in six size bins, including 0.3, 1.0, 2.0, 3.0, 4.0 and 5.0 μm . The AF ratio is defined as

$$\text{AF} = \frac{n_{>1\mu\text{m}}}{n_{\text{total}}}, \quad (1)$$

where $n_{>1\mu\text{m}}$ is the number of particles (activated as water droplets or ice crystals) larger than 1 μm recorded by the OPC and n_{total} represents the total number of soot particles entering the chamber recorded by the CPC.

2.3 Particle characterization methods

To characterize the morphology and porosity of soot particles in this study, both online soot aerosol aggregate and offline bulk soot sample powder measurements were performed. These measurements aim to investigate the changes in the compacted soot samples compared to the fresh samples for soot aggregate mass and size, D_f , microscopic TEM images and soot–water interaction ability, as well as soot sample PSD, N_2 -BET (Brunauer–Emmett–Teller) specific surface area and porosity. The relevant properties are summarized in Table 1.

2.3.1 Online soot aggregate characterization

Online soot aggregate property measurements mean that the soot particles for characterization were taken from the same aerosol flow as for IN experiments. Thus, the characterized soot particle population is identical to ice- or water-activated soot INPs or CCN. Considering that a HINC RH scan at a fixed T takes ~ 40 min, during which the fresh soot might become compacted to some extent (see Fig. 3), soot samples were always replaced with fresh soot for every sample RH ramp during the IN experiments. Soot samples agitated for 2 weeks were used to produce compacted soot aerosols. Both fresh and compacted 60, 100, 200 and 400 nm soot particles were sampled. In order to monitor the changes in the particle morphology and its influence on soot IN ability, the particle size and mass distribution of size-selected soot were measured synchronously with IN experiments. The single particle mass of a size-selected soot aerosol sample was derived from the CPMA mass scan measurements running with a CPC (Model 3787, TSI Inc., 1.5 L min^{-1}). Simultaneously, the size distribution of monodisperse soot particles was scanned by the SMPS setup operating with a CPC (Model 3776, TSI Inc.) in high-flow mode (1.5 L min^{-1}) for 60 and 100 nm or in low-flow mode (0.3 L min^{-1}) for 200 and 400 nm soot particles. Both the CPMA and SMPS raw data scans were fitted by a bimodal lognormal distribution

model. The uncertainty in particle mass and mobility diameter was derived from the standard deviation of the mode mass from the CPMA scans and mode mobility diameter from SMPS scans from at least 20 to 100 runs per sample. The mass-mobility exponent D_f was obtained by plotting the single particle mass as a function of the particle mobility diameter and by fitting the data with a power function. Equation (A1) in Appendix A is used to fit mass-mobility data. In addition, the size distributions of polydisperse fresh and compacted soot particles were also measured and presented in Appendix A. The proportion of double-charged particles is less than $\sim 25\%$ and $\sim 29\%$ for the 60 and 100 nm size-selected particles, respectively, and approximately smaller than 20% for 200 and 400 nm (see Figs. A2 to A5). In order to take TEM images for soot aggregate morphology analysis, soot aggregates were deposited on 400 mesh Cu grids with a formvar/carbon support film (TED, PELLA, INC.), utilizing the ZEMI setup (Mahrt et al., 2018). The sample grids were visualized at a magnification value of 75k.

2.3.2 Offline soot particle characterization

Bulk soot sample property measurements, termed offline measurements in Fig. 1 including Ar, N₂ and water vapor gas physisorption isotherm measurements, were not conducted simultaneously with IN experiments. Soot powder samples for these offline measurements were prepared in the same way as for online measurements. Therefore, different standard methods and analysis approaches are used to characterize soot cavity characteristics. In the following, we describe the different gas physisorption measurements used and associated analysis models applied.

Gas physisorption measurements: N₂, Ar and water vapor

Gas physisorption measurements can be used to evaluate particle structural and surface properties by measuring the interaction characteristics of bulk particulate samples with respect to probe gas pressure levels. Different probe gases have their advantages and disadvantages of addressing specific particle properties. N₂ is widely used for physisorption measurements to characterize porous material PSD in the literature (Jelinek and Kovats, 1994; Kruk et al., 1997; Galarneau et al., 1999; Hayati-Ashtiani, 2011; Kupgan et al., 2017). Some studies reported that the results from N₂ physisorption measurements might be biased due to its quadrupolar nature and the interaction with the substance surface functional groups, which affects the orientation of the molecule adsorbed. For instance, pore structure diameters for porous material derived from N₂ isotherms can be substantially underestimated (Jelinek and Kovats, 1994; Kruk et al., 1997; Lowell et al., 2004). More recently, Ar has become a more favored noble gas in some laboratory studies (Gardner et al., 2001; Thommes et al., 2012; Sing, 2014b) as a result of its

single atomic shape and its inertness with the substrate surface polarities. Thus, we performed both N₂ and Ar isotherm measurements to compare the differences of soot porosity results derived from both measurements. The DVS measurement, which has the advantage of reflecting the interaction ability of soot with water vapor, was also conducted to understand soot IN activities at low T with supersaturated water vapor. However, quantitative water vapor isotherm data analysis requires the soot–water contact angle, which is not well constrained in the literature. Thus, contact angle assumptions are made for approximate calculations in this study. By comparing the results of these three measurements, morphology differences between fresh and compacted soot particles and the influence on soot particle IN activities are discussed.

The Ar and N₂ adsorption and desorption isotherms for soot samples were measured at 77 and 87 K, respectively, at varying adsorbate gas pressure. Before the measurement, all samples were outgassed for 2 h at 573 K under vacuum. The raw data are presented as the volume of the adsorptive gas adsorbed per gram of sample at standard temperature and pressure (STP) corresponding to each relative pressure (p/p_0) condition, which is indicated by the ratio of adsorptive vapor pressure (p) to the adsorptive vapor saturation pressure (p_0).

Before the DVS measurement, soot samples were dried for 4 h at 573 K and then cooled down to ambient T under vacuum. The measurement was conducted at 298 K by a gravimetric dynamic method measuring the mass of water vapor adsorbed or desorbed by the soot sample under varying p/p_0 (or RH) conditions, assuming a quasi-equilibrium state of the soot sample under this vapor pressure. The initial sample mass is used as a reference for following mass measurements. For each sample, an adsorption isotherm was obtained from $p/p_0 = 5\%$ to 90% with a step of 10%, and then a desorption isotherm down to $p/p_0 = 5\%$ with the same p/p_0 resolution was also measured. Finally, the sample mass change (Δm) in percentage as a function of p/p_0 can be obtained, representing the amount of water adsorbed or released per gram of soot sample under various p/p_0 conditions.

PSD analysis

Sorption isotherms can be used to calculate the PSD, which is of interest for describing the structure of carbon black aggregates. In the following we apply the term “pore” to describe the cavity structures in the soot aggregate. The corresponding term “pore radius” is hereby defined as a radius equivalent to the cylindrical pore half-width that leads to the same vapor pressure reduction due to the inverse Kelvin effect. The PSD results reflecting soot cavity characteristics as a function of pore size can be derived from the desorption isotherms by applying the Kelvin equation with specific assumptions. The Kelvin equation is given as the following:

$$r_k = -\frac{2\gamma_{sl}v_s \cos(\theta)}{RT \ln(p/p_0)}, \quad (2)$$

Table 1. Soot sample characterization results.

Soot type	FW200fresh	FW200comp	PR90fresh	PR90comp
Volatile matter at 950 °C (wt wt % ⁻¹) ^a	20		1	
Primary particle size nm ^a	13		14	
Fractal dimension (D_f) ^b	2.64	2.62	2.66	2.66
$N_2 S_{BET}$ m ² g ⁻¹ ^c	552.4 ± 5.1	567.3 ± 7.3	336.5 ± 2.7	332.9 ± 3.0
Total pore volume at $p/p_0 = 0.99$ (N ₂) mm ³ g ⁻¹	1543	1431	1135	755
Total pore volume at $p/p_0 = 0.99$ (Ar) mm ³ g ⁻¹	1788	865	967	662
Number of voids per aggregate ^d	1.88 (136)	0.27 (147)	1.71 (154)	0.28 (143)

^a Information from the manufacturer. ^b Calculated from the SMPS-CPMA particle size and mass measurement results by using Eq. (A1).

^c $N_2 S_{BET}$ is calculated by the formulation presented in Appendix B. ^d Defined as a two-dimensional void encompassed by primary particle clusters per aggregate and derived from TEM image analysis. The total number of soot aggregates analyzed is shown in the parentheses, respectively.

where r_k is the pore radius required for capillary condensation induced by the inverse Kelvin effect, γ_{sl} denotes the interfacial tension between the solid and liquid phases, v_s is the adsorptive molar volume, θ is the contact angle between the soot surface and the adsorptive gas which can be taken as 0° for N₂ and Ar, and R is the ideal gas constant with the value of 8.314 J mol⁻¹ K⁻¹. The application of the Kelvin equation for PSD analysis relies on the following assumptions. (1) The equation is valid over the complete pore size range addressed. (2) The pores are rigid and of a cylindrical structure. (3) The capillary filling or desorption of each pore does not depend on its location within the pore network inside the aggregate. (4) The adsorption or desorption on the pore walls proceeds exactly in the same way as on the corresponding open surface.

Liquid can condense in mesopores (2–50 nm) when N₂ or Ar saturation ratios are above the BET range ($p/p_0 > 0.15$ or 0.1), due to capillary effects according to the Kelvin equation. This condensation step is reversible with decreasing p/p_0 , which refers to the desorption branch of the isotherms and is in equilibrium state. By applying the Brunauer–Joyner–Holanda (BJH) approach (Barrett et al., 1951; Lowell et al., 2004), the PSD can be calculated from the desorption branch of the respective sorption isotherm. A more detailed description of the PSD results determination by the BJH approach can be found in the International Union of Pure and Applied Chemistry (IUPAC) recommendation for calculation of the distribution of mesopores (Thommes et al., 2015). The BJH approach calculates the pore size from the Kelvin equation at each p/p_0 value, and the cumulative adsorbed gas volume corresponding to a pore size value can also be retrieved. However, the desorption step includes the desorption from the adsorbent surface in addition to a release of N₂ or Ar from pores, and these surface desorbates are irrelevant to pore structures during the desorption process. To distinguish between the liquid released from pores and that which is desorbed from the material surface, the De Boer thickness equation was applied (Thommes and Cychoosz, 2014; Thommes et al., 2015). With the BJH method, pore radius from 1 to

200 nm can be calculated, but values close to the limits have to be interpreted carefully. At the lower limit, inter-molecular forces lead to an enhanced condensation of liquid. Therefore, the pore radius below 10 nm calculated by the Kelvin equation is an underestimation of the real pore radius by approximately 30 % (Lowell et al., 2004). The upper limit of 400 nm is defined by the highest $p/p_0 \approx 0.99$ that can be reached during the measurement. Due to the irregular geometry of carbon black aggregates, the intra-aggregate cavity volume among primary particles cannot be clearly distinguished from the outer-aggregate volume between primary particle clusters. Therefore, the pore volume at the highest pore radius is uncertain. Regardless of the limitation of the BJH method, the calculated PSD results can be used to compare different types of soot and to investigate the effect of particle compaction on the spatial pore density of soot aggregates in this study.

Analogously to the BJH method applied for N₂ and Ar isotherms, a PSD analysis approach can be formulated for DVS data on the basis of the Kelvin equation as well. The DVS isotherm PSD analysis method is based on the theory proposed by Wheeler (1951) and more recently formulated by Shkol'nikov and Sidorova (2007) and is also similar to the approach applied by Mahrt et al. (2020b). A six- or seven-order polynomial function is used for the extrapolation of the discrete isotherm data points into a polynomial expression instead of a three-order polynomial stepwise fitting for every three data points used by Mahrt et al. (2020b). The desorption branch in water vapor sorption measurement was used to derive the PSD as this branch is more associated with the equilibrium gas–liquid phase transition. Because soot–water contact angle was not measured, three contact angle values including 0, 45 and 75°, representing surface wettability from hydrophilic to hydrophobic, were used to cover the possible values for soot samples in this study. With a contact angle value assumption, pore radius can be calculated as a function of p/p_0 , according to Eq. (2). Under each water vapor p/p_0 condition, the mass of water vapor is measured in the DVS measurement. Subtracting the amount of water

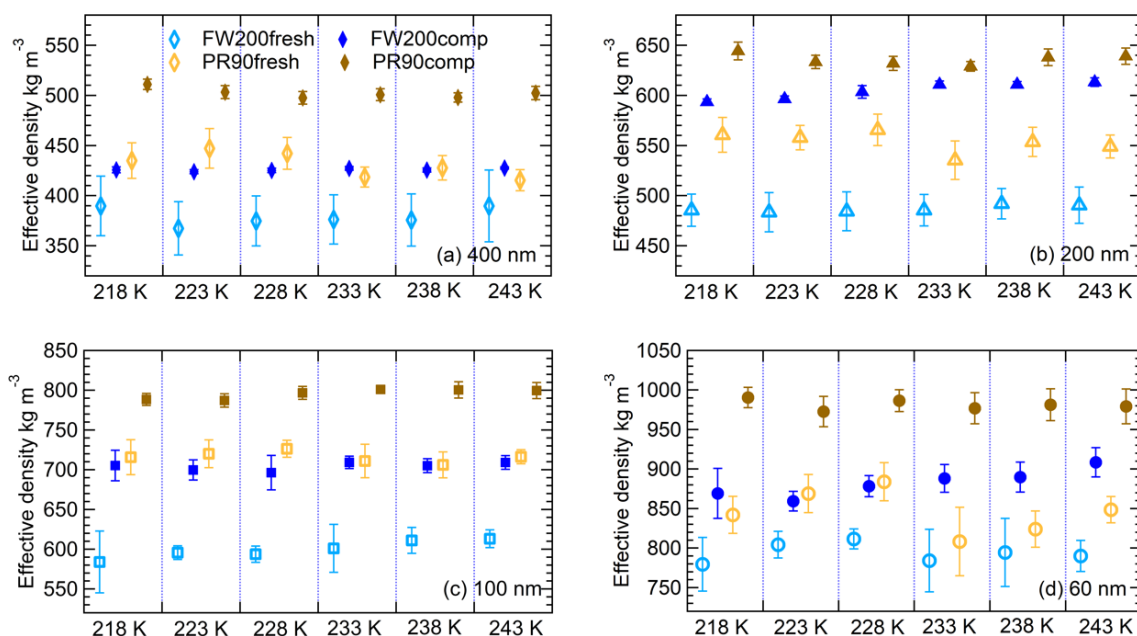


Figure 4. The effective density (ρ_{eff}) of 400 nm (a), 200 nm (b), 100 nm (c) and 60 nm (d) size-selected fresh and compacted FW200 and PR90 soot particles at each IN experiment T (x axis). The error bar is propagated from the particle mass and mobility diameter measurement uncertainties.

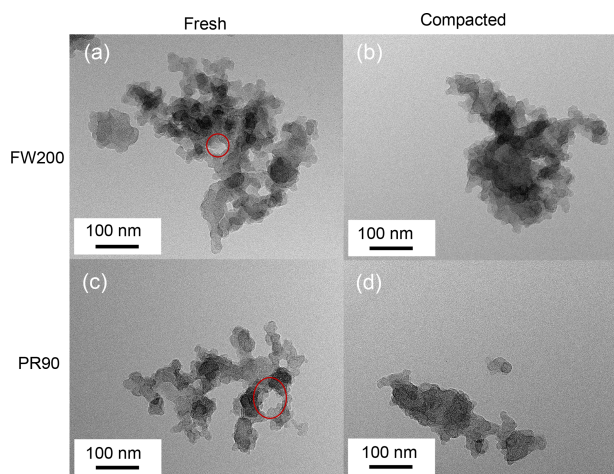


Figure 5. TEM images for 400 nm fresh and compacted FW200 and PR90 soot aggregates at a magnification value 75k. The red circle is used to indicate intra-aggregate voids with sizes of tens of nanometers.

sticking on the material surfaces and wall area, water volume can be calculated which is equivalent to the pore volume. Hence, the pore volume distribution can be determined as a function of pore radius. A detailed formulation of this approach is provided in Appendix B.

3 Results and discussion

3.1 Evidence of soot aggregate densification (compaction)

Particle effective density. Soot particles are fractal aggregates consisting of numerous spherical primary particles. Primary particles form branch-like structures, which are not rigid and susceptible to rearrangement by mechanical forces upon physical agitation. The particle effective density (ρ_{eff}) can be an indicator of the spatial density of intra-aggregate primary particles within a given mobility size soot aggregate. As shown in Fig. 3, the mass change in 400 nm size-selected soot particles after physical agitation shows a ρ_{eff} increment. The ρ_{eff} is calculated by the following equation:

$$\rho_{\text{eff}} = \frac{6m}{\pi \cdot D_m^3}, \quad (3)$$

where m is an average value for particle mass measured by the CPMA, and D_m is average particle mobility size derived from the SMPS measurements. The ρ_{eff} results at all sizes for both fresh and compacted particles are presented in Fig. 4, showing the ρ_{eff} of size-selected soot particles for the IN experiments at different T . The agitation process increases FW200 and PR90 soot particles ρ_{eff} by more than 10% and 15%, respectively, showing that all compacted FW200 and PR90 particles are denser than the fresh ones with the same mobility size. This evidence of soot aggregate compaction suggests that its cavity structure and intra-aggregate void vol-

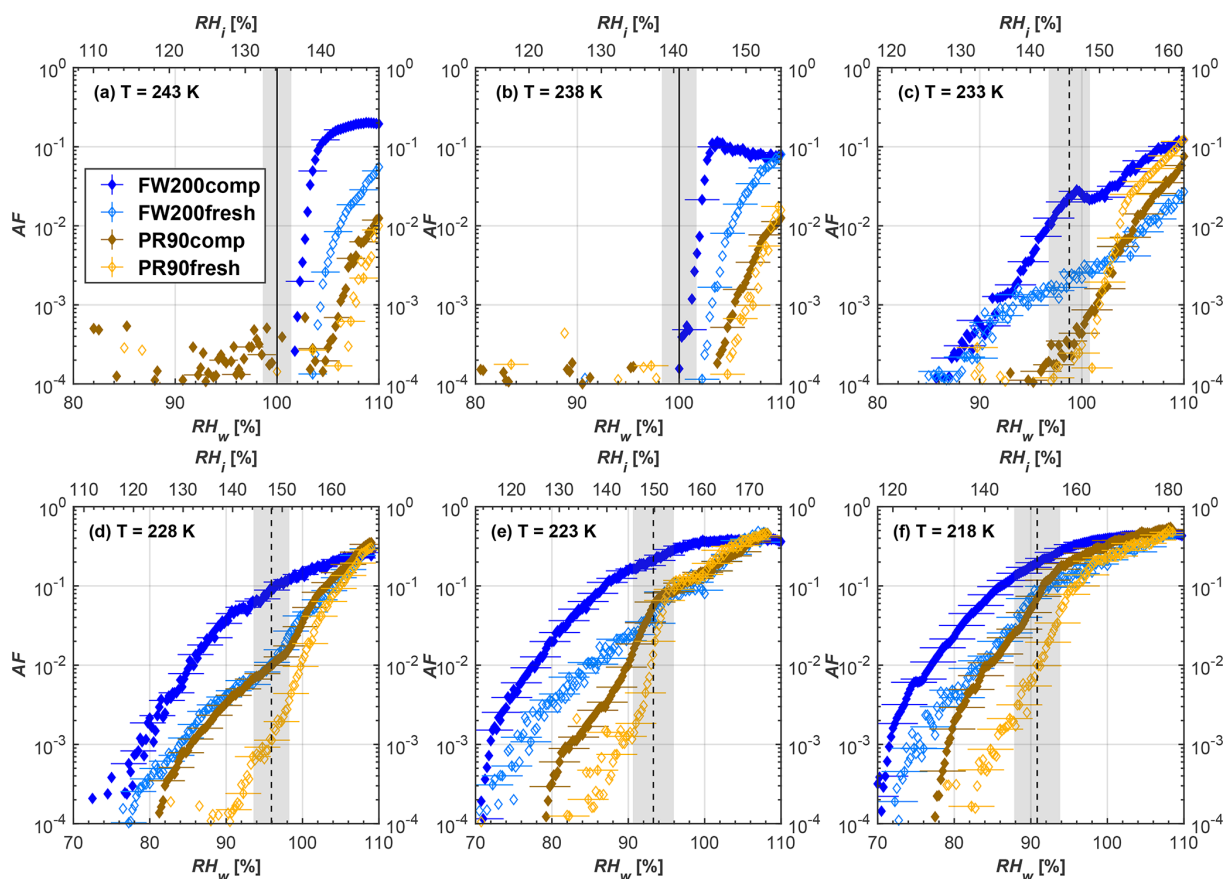


Figure 6. Average of three AF curves as a function of RH_w and RH_i from the $1\ \mu\text{m}$ OPC channel for 400 nm fresh and compacted FW200 and PR90 soot particles at different T . For 243 and 238 K the AF curves represent water droplet formation (see text Sect. 3.2.1 for details). Black solid lines represent water saturation conditions according to Murphy and Koop (2005). Black dashed lines denote the expected RH values for solution droplet homogeneous freezing at each T (Koop et al., 2000). The grey shading shows the possible variation range in RH that aerosol in HINC can encounter for the calculated homogeneous freezing RH values at each T .

ume (pore size) could be modified as a result of the mechanical stirring, potentially resulting in mesopore enrichment.

Soot aggregate compaction. As shown in Fig. 5, visual microscopic evidence of soot structure compaction can also be confirmed. Intra-aggregate voids, indicated by red circles with sizes of tens of nanometers, can be seen from fresh FW200 and PR90 soot aggregate images (Fig. 5a and c), showing a lacy or open-branched structure. Herein, the intra-aggregate void is defined as the two-dimensional cavity space encompassed by soot primary particle clusters in a soot aggregate. However, well-agitated soot aggregate TEM images (Fig. 5b and d) show a compact and dense structure with a rare presence of cavity structures. This finding coincides with higher ρ_{eff} values for compacted soot shown in Fig. 4. For each type of soot sample, the number of voids per soot aggregate was calculated and provided in Table 1. In general, this kind of void can be observed for fresh soot aggregate, whereas the void occurrence probability for compacted soot aggregates is much less than unity, demonstrating that the void structure universally exists in fresh soot aggregates and

is reduced upon aggregate compaction. It can be inferred that the whole primary particle network or spatial pore structures inside a soot aggregate should have been modified during the agitation process, generating compacted soot particles with the PSD shifting to smaller pore sizes. These intra-aggregate voids may also explain why fresh soot samples have a larger total pore volume reported from gas sorption measurements (see Table 1). This finding agrees with the intra-aggregate void volume measurement results for bulk carbon black materials performed by Joyce et al. (2009), who suggested that the void volume decreases with increasing sample compression level.

3.2 Soot particle ice nucleation results

3.2.1 Fresh and compacted soot particle activation fraction curves

The AF curves for fresh and compacted FW200 and PR90 soot with different particle sizes are presented in Figs. 6 to 9, respectively. Our soot particles only show water droplet

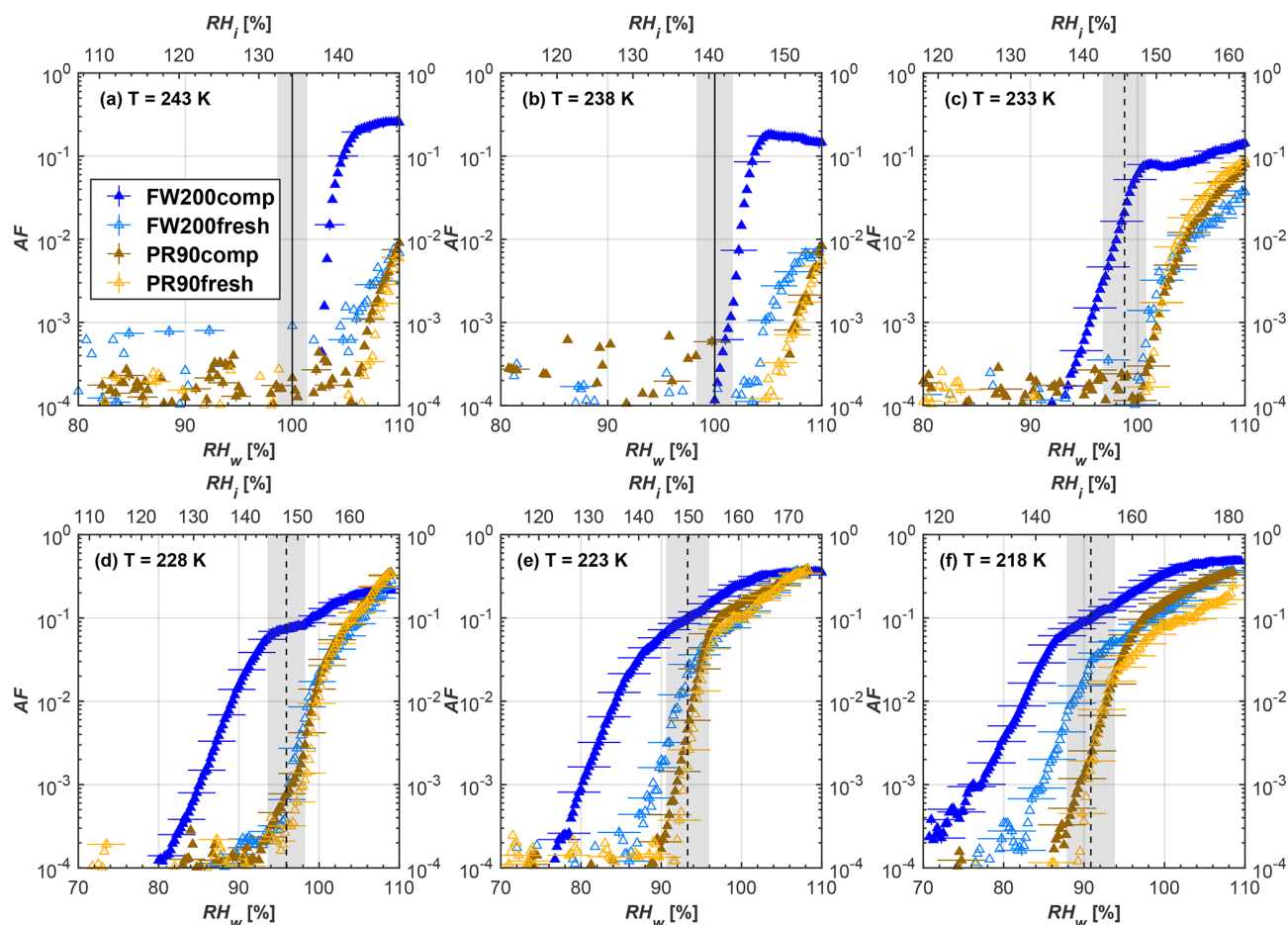


Figure 7. Average of three AF curves as a function of RH_w and RH_i from the $1\ \mu\text{m}$ OPC channel for 200 nm fresh and compacted FW200 and PR90 soot particles at different T . For 243 and 238 K the AF curves represent water droplet formation (see text Sect. 3.2.1 for details). Black solid lines represent water saturation conditions according to Murphy and Koop (2005). Black dashed lines denote the expected RH values for solution droplet homogeneous freezing at each T (Koop et al., 2000). The grey shading shows the possible variation range in RH that aerosol in HINC can encounter for the calculated homogeneous freezing RH values at each T .

formation at $T > \text{HNT}$ and nucleate ice at $T < \text{HNT}$. Evidence of soot–water droplet formation at $T > \text{HNT}$ is presented in Figs. C1 to C4, showing that OPC signals from the $5\ \mu\text{m}$ channel are absent for $T = 243$ and 238 K. Since droplet growth rates are much lower than those of ice crystals due to the difference between RH_i and RH_w , any ice crystals nucleating onto the soot particles at $T < 233$ K would be able to grow up to $5\ \mu\text{m}$ in HINC and be detected by the OPC. This is also consistent with the results reported by Lacher et al. (2017) and Mahrt et al. (2018). For $T < \text{HNT}$, 200 and 400 nm compacted FW200 and 400 nm compacted PR90 soot particles generally can reach a higher AF value under the same T and RH condition compared to the fresh particles, suggesting compaction enhanced IN activities at $T < \text{HNT}$ and soot–water interactions for $T > \text{HNT}$. Aggregate compaction also promotes 100 nm FW200 soot activation at $T \geq 233$ K as shown in Fig. 8. However, the rest of the AF curves for small-sized (60 and 100 nm) particles do not

show significant differences between compacted and fresh FW200 or PR90 soot. Overall, compacted soot particles with larger sizes are more effective INPs in the cirrus regime compared to the uncompacteds. We attribute the soot particle IN activity below RH_{hom} at $T < \text{HNT}$ to the PCF process. In the following, we compare the differences in soot particle IN between fresh and compacted soot particles produced from both BC materials.

FW200. For 400, 200 and 100 nm compacted FW200 soot particles, our results are in good agreement with Mahrt et al. (2018), who studied the IN activity of the same BC material with the same sizes but utilized a fluidized bed aerosol generator (FBG) to produce soot aerosol samples. It is very likely that the beads and rotating motion of the transition belt in the FBG setup can also result in soot aggregate compaction, which has a similar compaction effect on the fresh soot to that induced by Teflon bar mechanical agitation in this study. However, fresh soot particles without physi-

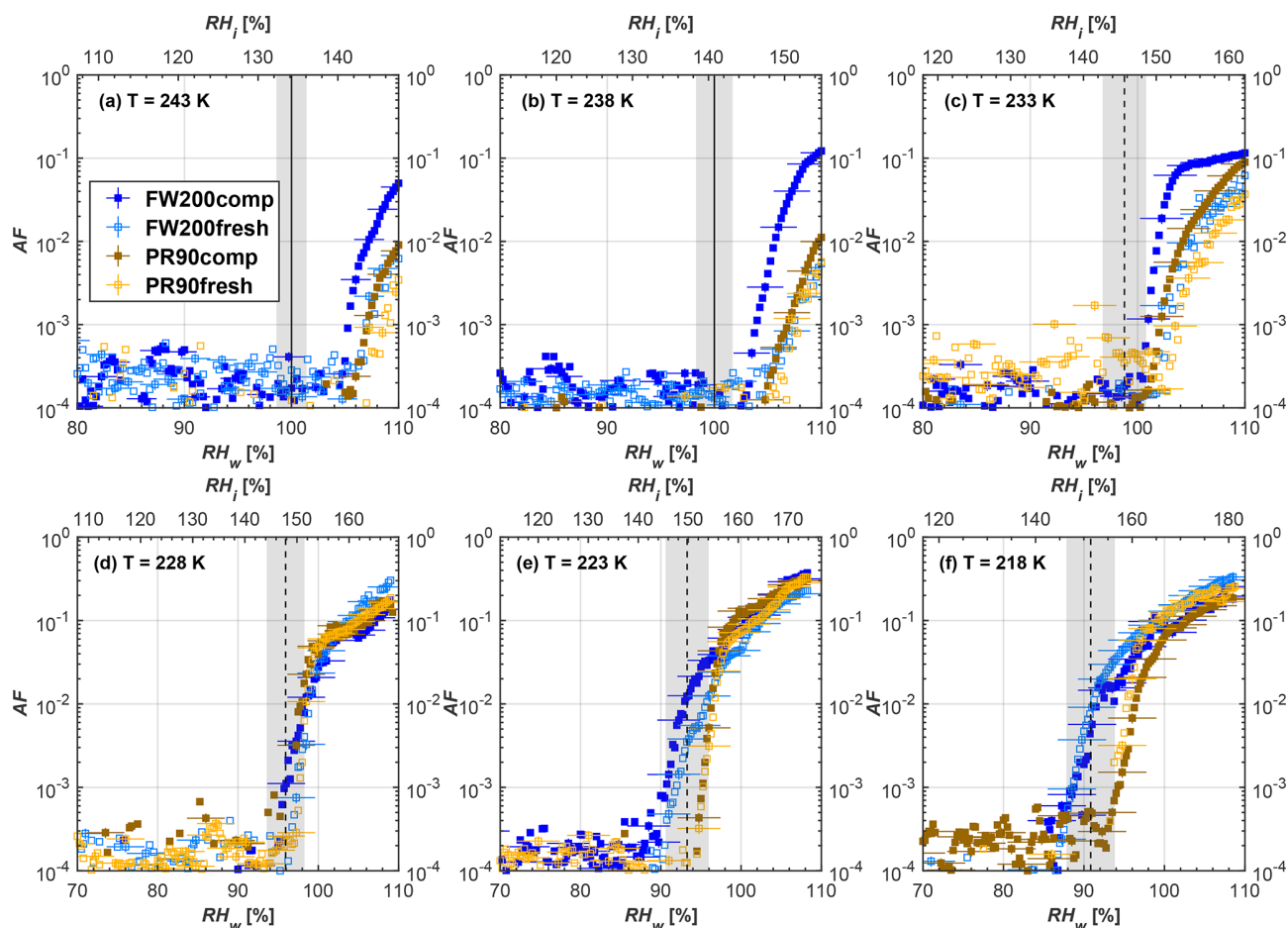


Figure 8. Average of three AF curves as a function of RH_w and RH_i from the 1 μm OPC channel for 100 nm fresh and compacted FW200 and PR90 soot particles at different T . For 243 and 238 K the AF curves represent water droplet formation (see text Sect. 3.2.1 for details). Black solid lines represent water saturation conditions according to Murphy and Koop (2005). Black dashed lines denote the expected RH values for solution droplet homogeneous freezing at each T (Koop et al., 2000). The grey shading shows the possible variation range in RH that aerosol in HINC can encounter for the calculated homogeneous freezing RH values at each T .

cal aging may have a loose structure. As shown in Fig. 6, 400 nm compacted FW200 soot particles are more effective INPs than the fresh sample, as the compacted particles require a lower RH to reach the same AF value at $T < 233$ K. We ascribe this to an enrichment in small mesopores for compacted FW200 soot particles. After agitation, macropores (> 50 nm in width) and larger mesopores (> 20 nm in width) among fresh soot aggregates might be compacted into smaller mesopores which induce an inverse Kelvin effect and lead to capillary condensation under RH conditions well below water saturation conditions, thereby promoting PCF activation according to the soot–PCF framework (Marcolli et al., 2021). Given that physical agitation is the only modification to soot samples, the enhancement of IN ability must result exclusively from the changes in morphology (resulting in mesopore enrichment) induced by soot aggregate compaction. At $T = 233$ K, 400 nm fresh and compacted FW200 soot particles show comparable AF values

at $RH_w < 94\%$, but FW200fresh AFs are lower than that of FW200comp at $RH_w < 94\%$, as shown in Fig. 6c. This is also because FW200comp soot contains more mesopores relevant to PCF activation compared to the fresh sample, showing that 400 nm FW200comp reaches higher AFs than the fresh particles, because the 400 nm FW200fresh sample may reach its limit for PCF-relevant mesopores, whereas a larger fraction of FW200comp sample has more mesopores, satisfying the PCF process when reaching at the same AF level at $RH_w = 94\%$ for 233 K. At $T > \text{HNT}$, the enriched mesopores by aggregate compaction can also promote water droplet formation for 400 nm soot particles, as shown in Fig. 6a and b, in which compacted soot can form more water droplets than the fresh one under the same RH and T condition. This promoted soot–water interaction ability also results from the pore structure enrichment induced by aggregate compaction, considering that mesopores can make contributions to soot–water uptake by facilitating water capillary

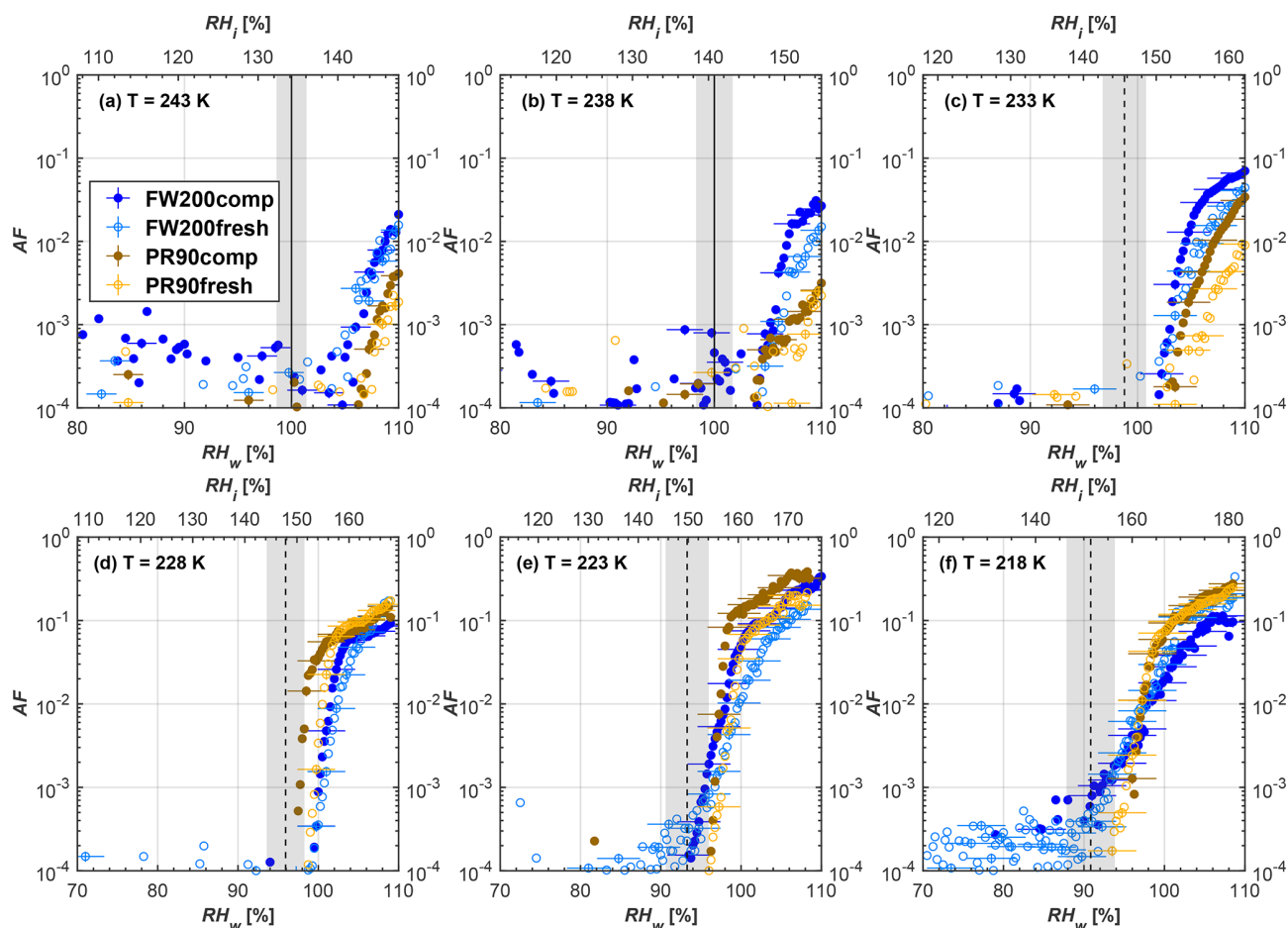


Figure 9. Average of three AF curves as a function of RH_w and RH_i from the 1 μm OPC channel for 60 nm fresh and compacted FW200 and PR90 soot particles at different T . For 243 and 238 K the AF curves represent water droplet formation (see text Sect. 3.2.1 for details). Black solid lines represent water saturation conditions according to Murphy and Koop (2005). Black dashed lines denote the expected RH values for solution droplet homogeneous freezing at each T (Koop et al., 2000). The grey shading shows the possible variation range in RH that aerosol in HINC can encounter for the calculated homogeneous freezing RH values at each T .

condensation (Persiantseva et al., 2004; Popovicheva et al., 2008a, b). The IN activities of 200 nm FW200 soot particles are similar to those of 400 nm FW200, but smaller-sized soot samples (60 and 100 nm) only freeze homogeneously (see Figs. 7 to 9) at $T < \text{HNT}$. At $T > \text{HNT}$, these small-sized soot particles (< 400 nm) form water droplets only above water saturation conditions. We believe that small-sized aggregates (100 and 60 nm) do not possess enough mesopores of the right size or structure due to the limited number of primary particles making up these small aggregates, resulting in a limited intra-aggregate void volume.

PR90. Size-selected 400 nm PR90 soot particles show similar but less pronounced IN enhancement after compaction, in comparison to FW200 soot. At $T \leq 228$ K below HNT, PR90comp soot particles reach a higher AF value under the same T and RH condition compared to PR90fresh, which means PR90comp are more active INPs. This is also because compacted PR90 soot contains mesopores of relevant proper-

ties (size and structure) to promote PCF activation. However, there is no difference between the PR90comp and PR90fresh soot at $T = 233$ K (see Fig. 6c). This is because of the homogeneous freezing dependence on T and the limitation of ice embryo growth with such a small mesopore volume in PR90 soot. In addition, the homogeneous freezing nucleation rate decreases exponentially with increasing T (Ickes et al., 2015). Additionally, the small PSD (see Sect. 3.4) in PR90 soot aggregates limits the volume of supercooled water engaging in porewater homogeneous freezing. Both inhibit the PCF process by leading to a small homogeneous freezing rate (Ickes et al., 2015; Vali et al., 2015; Koop and Murray, 2016). On the other hand, the pore volume in PR90 soot aggregate may be too small for the volume of water to freeze on the timescale of our experiments (~ 10 s) or to allow the ice embryo to grow outside of the pore. David et al. (2020) reported a similar case where silica particles with smaller mesopores (2.8 or 3.3 nm) are less active INPs via

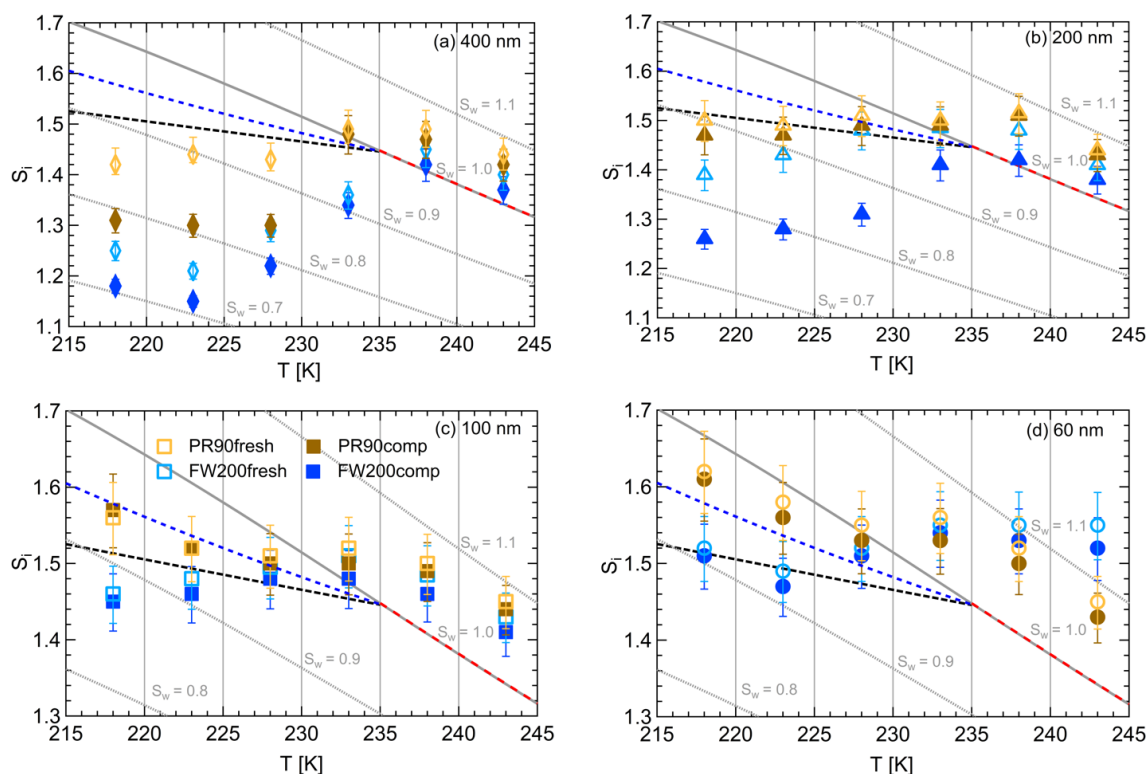


Figure 10. Onset saturation values (at $AF = 0.1\%$) with respect to ice (S_i) as a function of T and soot particle size (indicated in each panel) for fresh and compacted FW200 and PR90 soot samples. Black dashed lines denote the homogeneous freezing conditions for solution droplets at $T < HNT$, according to Koop et al. (2000). Blue dashed lines denote the homogeneous freezing conditions for solution droplets at $T < HNT$, according to Schneider et al. (2021). Grey dotted lines represent constant water saturation conditions calculated based on Murphy and Koop (2005). Red dashed lines indicate the $1\ \mu\text{m}$ size water droplet survival conditions for $T > HNT$, where the phase of the particle $> 1\ \mu\text{m}$ can be either ice or liquid water. The S_i uncertainty caused by the temperature uncertainty ($\pm 0.1\ \text{K}$) of the HINC chamber is indicated as error bars. Each data point was derived from the AF curves in Figs. C1 to C4 (see Appendix C), in which each AF curve is an average of at least three individual RH ramps.

PCF than those particles with larger mesopores (9.1 nm) at 233 K. At $T > HNT$, PR90comp 400 nm soot particles do not tend to be more effective CCN than PR90fresh, suggesting the enrichment of small mesopores caused by agitation has a limited influence on promoting PR90 soot–water interaction abilities. Smaller-sized ($< 400\ \text{nm}$) PR90 soot particles do not show a significant difference for IN ability between aggregates with and without compaction but require homogeneous conditions to form ice crystals (Figs. 7 to 9) at $T < HNT$. At $T > HNT$, small-sized ($< 400\ \text{nm}$) PR90 soot particles form water droplets at $RH_w > 105\%$, which implies their hydrophobic surface and low water interaction ability.

There are also differences between FW200 and PR90 IN activities. Firstly, PR90 soot particles are less active INPs compared to FW200, as shown by the fact that the same-sized fresh or compacted PR90 soot particles require a higher RH than that of FW200 soot with the same compaction level to reach the same AF value at the same T . According to the PCF mechanism (Marcolli, 2014; Marcolli et al., 2021), this probably results from the differences in soot sample properties,

including sample PSD and soot–water contact angle, both of which are determinators for the PCF mechanism. Thus, the PR90 soot PCF process occurs under higher saturation conditions and requires low T if PR90 soot is originally less porous than FW200 and/or of a lower surface wettability. Secondly, aggregate compaction-induced IN promotion for PR90 soot is not as significant as for FW200 soot. As can be seen from Fig. 7, soot aggregate compaction can exert IN enhancement for 200 nm FW200 soot particles but not for PR90 soot particles. This is also determined by the intrinsic nature of the soot sample and may suggest that aggregate compaction-induced mesopore enrichment for PR90 soot is not as strong as for FW200 soot or may simply be due to a higher contact angle of PR90 at $T > HNT$, where PR90 soot particles are poorer CCN and require higher RH conditions to form droplets than FW200 soot with the same sizes, also suggesting a paucity of pore structures and a low water interaction ability of PR90 soot.

3.2.2 Fresh and compacted soot particle onset ice nucleation results

In this section, the IN ability of fresh and compacted FW200 and PR90 soot particles will be compared in terms of their onset ice saturation (S_i) results, and the IN dependence on particle size will be discussed. The onset S_i value is a measure to evaluate the IN ability of INPs, using the threshold S_i value required for the INP to reach a prescribed AF value (0.1 %) at a fixed T . The onset S_i values for each soot sample particle at different T are presented in Fig. 10, derived from the AF curves of the 1 μm OPC channel in Figs. 6 to 9 in which each curve is an average of three RH ramp experiments. In Figs. C5 to C8, we show the size dependence for the sample type compared to Figs. 6 to 9, where we show different samples for a given size.

FW200. From Fig. 10a, soot aggregate compaction decreases the onset S_i value for 400 nm FW200comp soot particles by at least 0.08 compared to the fresh sample when $T < 233$ K in the cirrus regime. As shown in Fig. 10b, the onset S_i for 200 nm FW200 can be decreased by as much as 0.15 when $T < 233$ K. In particular, there is also a reduction in the onset S_i value at $T = 233$ K for 200 nm FW200comp soot particles showing a smaller S_i value compared to 200 nm fresh FW200 soot particles, which means that soot aggregate compaction can even promote, albeit by a small S_i reduction, 200 nm FW200comp to nucleate ice. Both compacted and fresh 100 nm FW200 soot particles are still more competitive INPs than the solution droplets freezing homogeneously at $T \leq 223$ K (see Fig. 10c). However, the onset condition values for fresh and compacted particles are overlapped, which means limited IN enhancement for 100 nm compacted particles, as discussed before. Finally, 60 nm FW200 soot particles cannot facilitate ice crystal formation below homogeneous conditions proposed by Koop et al. (2000) at $T < \text{HNT}$, showing that the error bars are across the homogeneous freezing line even at the lowest $T = 218$ K. If we consider the homogeneous freezing conditions of sulfate particles proposed by Schneider et al. (2021), 60 and 100 nm soot particles are active INPs via PCF at $T \leq 223$ K, as their S_i values are below the corresponding RH_{hom} condition (blue dotted line, Fig. 10). At $T > \text{HNT}$ all data points lie within error bars and above the 1 μm water droplet limit line (the limit above which we cannot distinguish between ice and water solely based on the 1 μm OPC channel) for both compacted and fresh FW200 soot of all sizes, suggesting compaction does not have pronounced effects on FW200 soot droplet activation activities. Kanji et al. (2020) suggested that soot particles are poor INPs in immersion-mode freezing at $T > \text{HNT}$ and can only form water droplets under these conditions. These results are also consistent with the findings reported by Friedman et al. (2011), Mahrt et al. (2018) and Falk et al. (2021), in which the authors noted that soot particles do not nucleate ice for $T > \text{HNT}$.

PR90. In general, PR90 soot requires a higher onset S_i value than FW200 soot with the same compaction level to reach the same $\text{AF} = 0.1$ % at $T < \text{HNT}$, indicative of less active IN abilities. Nonetheless, soot compaction significantly enhances 400 nm PR90 soot IN at $T \leq 228$ K, similarly to that of FW200 soot discussed above. In Fig. 10a, 400 nm compacted PR90 soot onset S_i values can be reduced by more than 0.1, in contrast to the fresh particles. Fresh and compacted 200 nm PR90 soot still can compete with homogeneous freezing at $T = 218$ and 223 K as the S_i values are smaller than the RH_{hom} condition, but the error bars intercept with the homogeneous freezing threshold calculated according to Koop et al. (2000) as depicted in Fig. 10b. PR90 soot particles with or without compaction of sizes 60 and 100 nm only freeze homogeneously at $T < \text{HNT}$ even considering the higher limit of homogeneous freezing proposed by Schneider et al. (2021), also suggesting that the agitation process does not promote PR90 soot IN, as shown by onset value error bars being in touch at all T (see Fig. 10c and d). At $T > \text{HNT}$, all sizes of fresh and compacted PR90 soot particles form water droplets under similar RH conditions at $S_w > 1.05$, suggesting soot particle structure compaction does not influence PR90 soot–water droplet formation, as discussed previously.

In brief, aggregate compaction can enhance soot particle IN ability significantly by lowering the onset S_i values required for PCF activation at $T < \text{HNT}$ in the cirrus regime and showing a clear size dependence. Firstly, soot aggregate compaction-induced IN enhancement shows a size dependence, suggesting that this compaction effect on soot IN ability is more pronounced for large soot particles (> 200 nm) and that physical agitation might not compact the small soot aggregate morphology as effectively as for large soot aggregates to induce IN enhancement. The size dependence effect is observed for the IN of soot particles with the same compaction level, showing that large particles require a lower onset S_i value than smaller ones at the same T , which coincides with the results reported by Mahrt et al. (2018) and Nishman et al. (2019) and also what is generally known about mineral dust ice nucleation (Archuleta et al., 2005; Welti et al., 2009; Kanji and Abbatt, 2010). This is because large soot aggregates have a higher probability of possessing the mesopores of the right size for the PCF mechanism. Large soot aggregates contain a larger number of primary particles and are more susceptible to a primary particle network change caused by external forces, whereas small soot aggregates contain a limited number of primary particles and are more analogous to the fresh soot basic unit which is much closer to primary particle clusters or a single primary particle. Thus, large soot aggregates better present the aggregate compaction effect on ice nucleation. The PCF activation has a positive relation to the number of mesopores, implying that the PCF activation requires sufficient mesopores with the right characteristic for capillary condensation and ice nucleation within the mesopores. Furthermore, an individual

mesopore even with the appropriate pore size for PCF probably plays a limited or implicit role in the event. For instance, dynamic molecular simulation results suggest that a single 3 nm pore in a silica slab does not facilitate ice formation, whereas ice crystal can grow quickly in an array of pores with the same width under the same simulation environment (David et al., 2019), suggesting that a bridging effect from the pore network promotes the growth of pore ice to a macroscopic ice crystal. In this study, experiments for 60 nm FW200 and PR90 soot are performed to extend a lower size limit for the investigation of soot particle IN activities in the cirrus regime since sub-100 nm particles are often observed in the upper troposphere (Petzold and Schröder, 1998; Twohy and Gandrud, 1998; Cziczko and Froyd, 2014). According to Zhang et al. (2020), there might be a size threshold between 100 and 400 nm for soot particles to be active INPs under cirrus cloud conditions. However, it is difficult to define such a threshold size for soot particles to trigger the PCF mechanism under cirrus conditions. As can be seen from the aforementioned discussion, the IN ability dependence on size is different for FW200 and PR90 soot in this study, showing that PR90 is more dependent on its size than FW200 soot as 200 nm PR90 soot onset results are close to the homogeneous freezing line, whereas those of 200 nm FW200 soot are well below the RH_{hom} limit. Last but not least, our IN results for 60 and 100 nm FW200 soot particles suggest that the freshly emitted aviation soot particle majority with a small mobility size (≤ 100 nm) could act as INP candidates in the cirrus cloud regime if they have properties (pores and surface wettability) similar to FW200 soot, and the homogeneous freezing limit considered is analogous to that proposed by the homogeneous freezing parameterization of Schneider et al. (2021).

3.3 Soot–water interaction ability

Water vapor sorption measurement is an approach to characterize the cavity properties and water interaction ability of porous material. In previous studies, Mahrt et al. (2020b) used DVS measurement to evaluate the PSD of C_3H_8 flame soot, and David et al. (2020) utilized this measurement to estimate the water–solid contact angle of porous silica. According to the IUPAC (Sing et al., 1985; Thommes et al., 2015), sorption includes two aspects of interaction between adsorptive gas and the solid, i.e., adsorption and desorption. As a function of water vapor p/p_0 levels, the Δm values of both adsorption and desorption branches for each sample are plotted in Fig. 11. Based on the IUPAC classification (Sing et al., 1985; Thommes et al., 2015), the measured isotherms for compacted and fresh FW200 soot can be classified as a type-IV isotherm, and PR90 soot follows type V. Both types of isotherms indicate the presence of mesopores in the sample. A larger isotherm hysteresis area for fresh and compacted FW200 soot implies they contain more mesopores for capillary water condensation than PR90 soot. Additionally, a larger isotherm slope for FW200 than PR90 soot with the

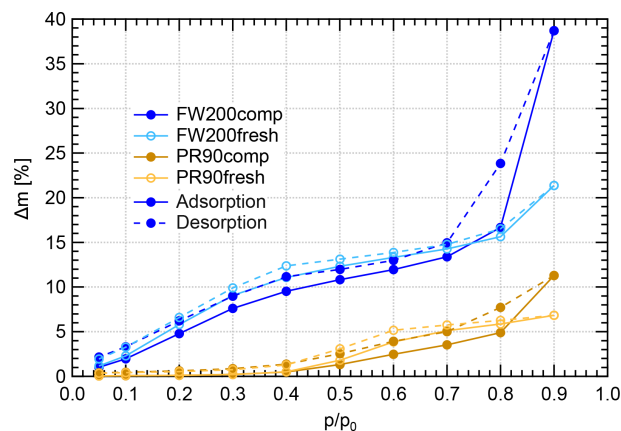


Figure 11. Water dynamic vapor sorption (DVS) isotherms at 298 K for compacted and fresh FW200 and PR90 soot. The y axis indicates the mass change fraction (Δm) with respect to the original sample mass, and the x axis stands for the relative water vapor pressure (p/p_0) levels. Solid lines indicate adsorption branches, and the dotted lines indicate desorption branches.

same compaction level at $p/p_0 < 0.3$ indicates a more wettable surface and a lower contact angle for FW200 soot. Both lower contact angle and abundance in mesopore structures facilitate the PCF process, generally supporting the discussion in Sect. 3.2. Hence, FW200 soot particles are more effective INPs due to an increased water uptake ability compared to PR90 soot of the same compaction level.

FW200 soot. In general, soot–water interaction abilities are influenced by the density of local surface active sites which are hydrophilic and can accommodate water molecules more easily and by soot porosity (Popovitcheva et al., 2000, 2008a, b). At a low p/p_0 range (< 0.3), soot–water uptake is attributed to its active site density. Persiantseva et al. (2004) suggested that the water interaction of soot samples under low RH conditions is representative of its water contact angle, and a concave isotherm curve usually stands for a low contact angle. As shown in Fig. 11, isotherms for fresh and compacted FW200 soot both show the same kind of sensitivity to increasing p/p_0 , suggesting that both soot samples contain a considerable number density of active sites and have a similar water contact angle. Figure 11 also shows that there is an inflection point for FW200 soot around $p/p_0 = 0.4$, which means the completion of monolayer water coverage and the start of capillary condensation induced by fine pore structures. Furthermore, the higher water uptake ability for fresh FW200 around $p/p_0 = 0.4$ suggests it contains more micropores (< 2 nm) than compacted FW200 because small-sized pores can induce capillary condensation at lower RH than larger pores. In addition, vapor isotherm hysteresis implies the existence of mesopores. For $p/p_0 > \sim 0.4$, water may form multiple layers over the soot surfaces and mesopore capillary condensation might be induced by an inverse Kelvin effect which is indicated by the hysteresis loop

when considering the desorption isotherm. The hysteresis loop for fresh and compacted FW200 soot can be attributed to Type H4 (Thommes et al., 2015), which is typical of materials that are both microporous and mesoporous. A considerably larger hysteresis loop for FW200comp means the sample contains plenty of mesopores, whereas a smaller loop for FW200fresh implies that there are a relatively lower number of mesopores. Assuming the soot–water contact angle to be the same for both fresh and compacted FW200, a smaller relative pressure of the hysteresis loop knee in the desorption branch (near $p/p_0 = 0.7$) for FW200comp than that of FW200fresh (around $p/p_0 = 0.8$) indicates that the mesopore size distribution of FW200comp shifts to a smaller pore size range given that small mesopores fulfil the Kelvin equation at low p/p_0 levels (Lowell et al., 2004). Note that the hysteresis in the desorption branch of FW200 soot, even at low p/p_0 levels, suggests the presence of ink-bottle pore structures which cannot release absorbed liquid with decreasing environmental RH (Lowell et al., 2004). In brief, water sorption results demonstrate that agitation will enrich the mesopore structures in FW200 soot samples but should not change soot particle surface wettability. This supports the IN enhancement for compacted FW200 soot particles in PCF activation at $T < \text{HNT}$.

PR90 soot. The Type V isotherm with a Type H3 hysteresis loop (Thommes et al., 2015) for PR90 soot shows the existence of smaller mesopores but a higher soot–water contact angle compared to FW200 soot. It is evident that PR90 soot contains far fewer active sites for water molecule adsorption at low p/p_0 levels than those of FW200 soot as PR90 soot presents a lower isotherm slope and a smaller Δm at low pressures as shown in Fig. 11, indicating that its contact angle is higher than FW200 (Kireeva et al., 2009). The flat isotherm at low pressures, suggesting a low ability for multiple-layer water adsorption, also manifests the weak water interaction ability over the PR90 soot sample surface, i.e., a hydrophobic surface (Lowell et al., 2004). For $p/p_0 < 0.4$, there is no significant water uptake activity difference between PR90comp and PR90fresh soot, suggesting physical agitation cannot modify their surface wettability. In addition, fresh PR90 soot adsorption and desorption isotherms level over those of compacted PR90 when $p/p_0 > 0.4$ and reach the largest difference at $p/p_0 = 0.6$, and then they become smaller than those of PR90comp soot when $p/p_0 > 0.8$. This can be explained similarly to the case for FW200 soot, and it suggests that compaction increases large-sized mesopores but decreases small-sized mesopores or micropores in PR90 soot, which can already lead to capillary condensation at low p/p_0 levels. A larger hysteresis loop for PR90comp soot than PR90fresh for $p/p_0 > 0.4$ suggests that the agitation process increases the availability of mesopores responsible for capillary condensation. Therefore, compacted PR90 soot contains more larger mesopores than the fresh sample, but both PR90 soot samples are more hydrophobic, suggesting a higher water contact angle than FW200 soot. Thus, the water sorption

isotherms for PR90 soot qualitatively explain the enhanced IN for PR90comp soot particles and the relatively poor IN ability compared to FW200 soot particles of the same size and the same compaction level.

In conclusion, DVS measurements provide soot–water interaction results consistent with soot particle IN results. Firstly, they reveal that a lower water contact angle contributes to more active IN for FW200 soot than PR90 soot (see Sect. 3.2). Meanwhile, DVS results demonstrate that physical agitation does not change soot surface active site density for water uptake and the contact angle. In addition, DVS isotherms demonstrate the porosity change in soot samples with and without compaction. Agitation-induced aggregate compaction causes the enrichment in large-sized mesopores and enhances water uptake ability, supporting the IN ability enhancement. From 400 nm soot particle IN results in Sect. 3.2, aggregate compaction can make PR90 soot particles as active as fresh FW200, suggesting that change in soot particle morphology can make originally hydrophobic soot IN-active just due to changes in mesopore structure, highlighting the very direct role of particle morphology in ice nucleation.

3.4 Soot particle pore size distribution

Soot aggregate pore structures need to be optimal to play a crucial role in the PCF process. On the one hand, the pore needs to be small enough to induce the inverse Kelvin effect and water capillary condensation, while on the other, the pore size is required to be large enough to be super critical with respect to the ice embryo size in order to support its growth out of the pore (Marcolli, 2020; Marcolli et al., 2021). Given a defined pore structure, with an approximate contact angle value and at a known T , the RH condition required for the pore capillary condensation and pore ice growth can be calculated according to the soot–PCF framework developed by Marcolli et al. (2021). In general, there exist different pore types, including cracks, cavities or capillaries, which can be like wedges, cones, and cylindrical and other geometric structures. For the sake of a PCF process, mesopores, which are defined as cylindrical with diameters between 2 and 50 nm (Sing et al., 1985), are the main interest in this study. The simplifications and assumptions of soot pore structure are introduced in Sect. 2.3.2 and will be held in this section as well.

3.4.1 Pore size distribution retrieved from Ar and N₂ isotherms

According to the BJH approach, the PSD results for soot samples can be analyzed from Ar and N₂ isotherms, which are independent of the contact soot–water angle. Based on Ar and N₂ physisorption measurement isotherms provided in Figs. D1 and D2, the calculated PSD results as a function of pore radius are presented in Fig. 12. The PSD results

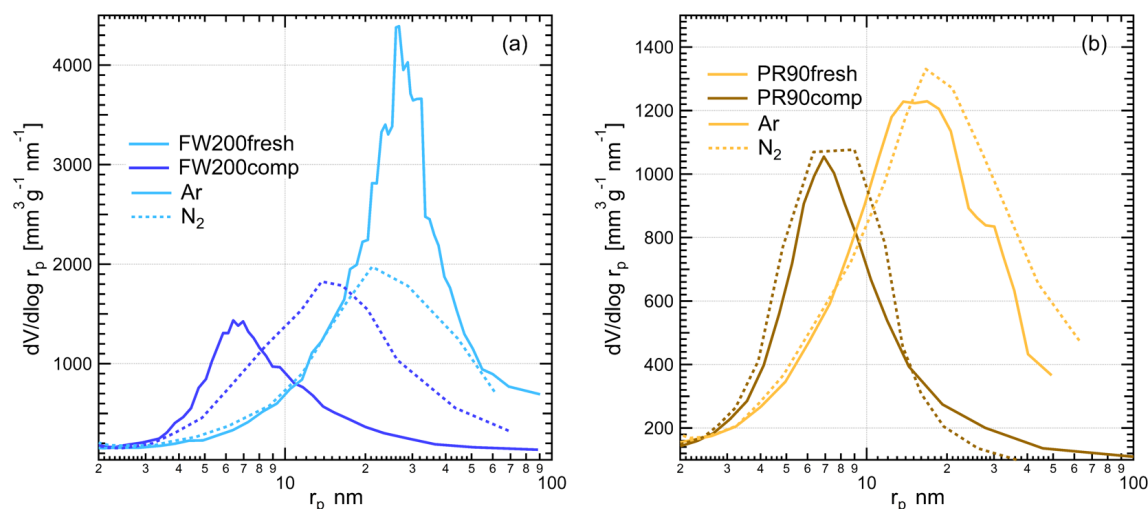


Figure 12. The PSD as a function of pore size (r_p) for fresh and compacted (a) FW200 and (b) PR90 soot based on Ar and N_2 adsorption and desorption isotherms (Figs. C1 and C2). Solid lines and dashed lines are for results derived from Ar isotherms and N_2 isotherms, respectively.

generally show that physical agitation shifts soot mesopore size distribution to smaller size ranges in comparison to the fresh samples. Compacted FW200 soot PSD derived from Ar BJH changes to a markedly smaller distribution mode from ~ 27 (21 nm, N_2 BJH) to ~ 7 nm (13 nm, N_2 BJH) compared to the fresh soot (see Fig. 12a). Analogously to FW200 soot, PR90 soot Ar PSD peak decreases from ~ 14 down to ~ 6 nm after compaction (see Fig. 12b). N_2 BJH analysis for compacted PR90 soot shows a PSD reduction from ~ 16 to ~ 7 nm compared to fresh PR90 soot. In general, the shift in PSD to small pore sizes for compacted soot samples suggests that small mesopores with radii approximately < 10 nm can be appreciably enriched after agitation, which would favor the PCF process, thereby enhancing soot IN via PCF. The mesopore enrichment contributes to compacted soot particle IN ability enhancement because the smaller the mesopore size is, the lower the RH is that will be required to trigger the PCF mechanism at the same T . For example, the Ar PSD result for FW200 soot, presenting a larger mesopore size distribution for pore radius from ~ 3 to ~ 12 nm, directly supports onset results in Fig. 10, as shown where 200 and 400 nm FW200comp soot shows ice nucleation at much smaller RH reaching the onset AF value (0.1 %) at the same T as the FW200fresh soot. According to the PCF process (Marcolli, 2014), 400 nm FW200 soot particles with a 60° water contact angle assumption (Marcolli et al., 2021) need ~ 4 nm mesopores to present onset $S_i = 1.24$ at 228 K (Fig. 10a) in order to induce capillary water condensation. If the contact angle is assumed to be 45° , the mesopore should be around 6 nm. From Fig. 12a, FW200comp soot contains more mesopores around 4–6 nm than the fresh sample, explaining that 400 nm FW200comp soot can show such a low onset S_i value, whereas 400 nm fresh FW200 soot containing fewer 4–6 nm mesopores requires higher onset S_i at the same

T . In general, the increased mesopore availability around 10 nm contributes to the enhanced IN ability of compacted soot particles because of the pore morphology changes upon aggregate structure compaction.

However, both fresh soot samples contain more larger-sized pore structures than compacted soot by showing a cross in the PSD curves of fresh and compacted soot sample derived from the same gas sorption technique. Fresh FW200 soot has more mesopores $> \sim 12$ nm (Ar PSD) or $> \sim 18$ nm (N_2 PSD) than the compacted soot sample, and the similar case for PR90 soot is about $> \sim 9$ nm (Ar PSD) or $> \sim 11$ nm (N_2 PSD). These pore structures are too large to trigger inverse Kelvin effects under the same RH condition as required by smaller mesopores (< 10 nm), making them PCF-irrelevant. For example, a 12 nm mesopore in a 400 nm FW200 soot aggregate with a water contact assumption 60° requires $S_i = 1.38$ at 233 K to induce capillary water condensation according to the Kelvin equation, which is already higher than the onset S_i value for 400 nm fresh (1.36) and compacted (1.34) FW200 soot particles shown in Fig. 10a. Hence, these large mesopores in fresh soot particles do not contribute to the IN activity. Besides, fresh soot samples also contain more pore structures larger than 50 nm in comparison to the compacted soot (see Fig. 12), which is consistent with their larger cumulative pore volumes (Table 1). These larger pore structures are probably intra-aggregate voids (macropores > 50 nm) among primary clusters or outer-aggregate voids as shown in TEM images (Fig. 5), which can be filled by Ar and N_2 near the saturation condition during sorption measurements.

Gas sorption measurements also manifest the structural difference between soot samples and the derived PSD dependence on a gas agent (Ar and N_2). Comparing the PSD results between FW200 and PR90 soot with the same ph-

ysisorption measurement, there is a difference to be noted even though they reveal similar mesopore size change effects exerted by agitation. The FW200 soot PSD curve can reach a much higher peak and contains a larger quantity of mesopores than PR90 soot with the same compaction level, supporting FW200 soot being more active in PCF activation than PR90 soot (Sect. 3.2). A higher porosity level of FW200 soot than PR90 soot is also proven by its larger cumulative pore volume from BJH analysis (see Table 1). In addition, the agitation can lead to a more significant Ar PSD change to fresh FW200 soot than to fresh PR90 soot, showing compacted FW200 soot Ar PSD peak-mode reduction from ~ 27 to ~ 7 nm, whereas the case for PR90 soot is from ~ 14 to ~ 6 nm and with a smaller cumulative pore volume change (see Table 1). This results from the difference in the soot sample original primary particle network. Likely, PR90 soot, comprising a small fraction of volatile content ($\sim 1\%$), may have a tight primary particle connectivity. Compared to PR90 soot, FW200 soot may have looser and less intimate primary particle connections, providing a higher probability of primary particle rearrangement when the physical force is imposed. For example, Marcolli et al. (2021) used the concept of overlap coefficient for primary particles to depict the tightness of primary particle connection, which has an important influence on the mesopore sizes of soot aggregates. Furthermore, the pore structure in FW200 soot aggregates is not as stable as that in PR90 soot aggregates, also resulting from its lower carbonization (Table 1). Therefore, agitation-induced aggregate compaction generates more smaller mesopores for FW200 soot than for PR90 soot. Furthermore, compacted PR90 soot has a comparable amount of mesopore structures in the size range ~ 6 – 7 nm to that of fresh FW200 soot, both showing a volume-based PSD value of $\sim 1000 \text{ mm}^3 \text{ g}^{-1} \text{ nm}^{-1}$ in this pore size range. This explains the comparable IN ability of 400 nm PR90comp and FW200fresh at $T \leq 228$ K. Again, it implies that originally hydrophobic soot can be active INPs if it contains sufficient mesopores relevant to PCF activation. Finally, Fig. 12 shows a divergence between Ar and N_2 PSD results for the same soot sample, suggesting the PSD result is dependent on the probe gas nature. As introduced in Sect. 2.3.2, the quadrupole moment of N_2 affects the orientation of the molecule during the sorption process (Thommes et al., 2012), thereby influencing isotherms measured and the PSD results. Nonetheless, these measurements allow one to conclude the same fact that the physical agitation shifts the PSD mode of fresh soot to small sizes and leads to significant medium mesopore enrichment and IN enhancement for compacted soot.

3.4.2 Pore size distribution retrieved from water vapor isotherms

In this section, we present the calculated soot sample PSD results by applying the Kelvin equation-based formulation in Appendix B to corresponding DVS isotherms, assuming dif-

ferent soot–water contact angle values for soot samples. The contact angle ranging from 0 to 180° , and from completely wettable to non-wettable, can be understood as a measure of surface wettability (Pruppacher and Klett, 1997; Lohmann et al., 2016). Reasonable assumptions for FW200 and PR90 soot need to be made by comparing similar water isotherms to the other soot samples with reported soot–water contact angle values in the literature. For instance, Persiantseva et al. (2004) used the sessile drop technique to measure the contact angle of water droplets on different soot pellets covering a broad range of wettability in conjunction with water vapor isotherms. The authors reported that the water contact angle on the aircraft combustor produced soot surface spans from 60 to 80° . By comparing the isotherm characteristics, their study can serve as a good reference for soot–water contact angle estimates in this study because we collected similar types of isotherms. Also note that the contact angle of different soot samples can vary in a broad range. For example, Shin et al. (2010) found that the hydrophobic graphene contact angle can be up to approximately $91 \pm 1^\circ$ tested by using a charge coupled device (CCD) camera. However, Kireeva et al. (2009) reported that the soot–water contact angle can be as low as 28° for hydrophilic black carbon. Moreover, the water contact angle for INPs even with the same material might show a distribution as a result of surface heterogeneity. For example, both Welti et al. (2012) and Marcolli et al. (2007) found that a distribution of contact angles describes the mineral INPs' IN ability better than a single contact angle value arbitrarily used for IN parameterization. Considering the heterogeneity and complex surface properties of soot particles, to ease the discussion, different contact angle values are assumed to describe averaged soot particle wettability and to simplify data analysis. In this study, three contact angle values (0 , 45° and 75°) for both FW200 and PR90 soot samples were used for the PSD formulation for DVS analysis. According to DVS isotherm analysis in Sect. 3.3, a low contact angle (45°) may fit the FW200 real surface better, whereas a higher contact value (75°) for PR90 might be suitable for its more hydrophobic surfaces. Marcolli et al. (2021) used a contact angle mean value 60° for FW200 soot in the PCF–soot framework study, which is also within the contact angle assumption range in this study. In brief, these three contact angle values can cover the possible contact angle range of these soot samples and present the role of contact angle in soot–water interaction activities. The corresponding soot PSD results with different soot–water contact angle assumptions are presented in Fig. 13.

The PSD analysis in Fig. 13 demonstrates that small mesopores generated after compaction are responsible for stronger water interaction abilities of compacted soot compared to fresh soot. PSD curves with the 0° contact angle assumption are used as a perfect case for soot samples and to serve as a PSD benchmark. Assuming a 45° contact angle for FW200 soot, the PSD curve for FW200comp stays above FW200fresh in the radius range from approximately

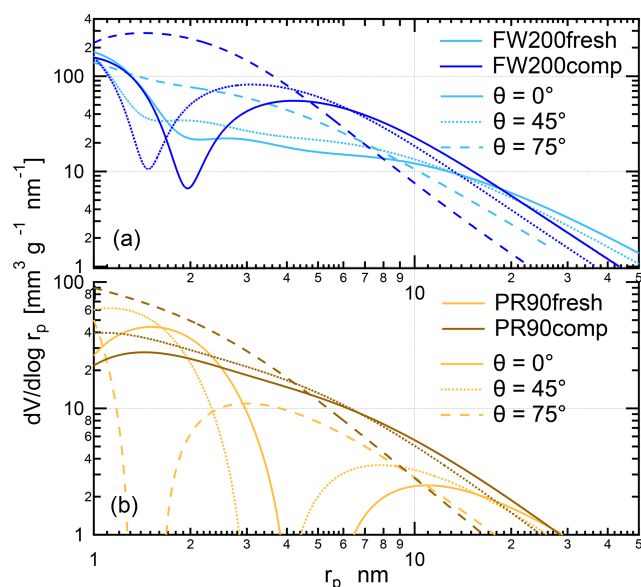


Figure 13. The PSD as a function of pore size (r_p) for fresh and compacted (a) FW200 and (b) PR90 soot with contact angle assumptions $\theta = 0^\circ$, $\theta = 45^\circ$ and $\theta = 75^\circ$ indicated by solid, dotted and dashed lines corresponding to DVS isotherms shown in Fig. 11.

2 to 14 nm in Fig. 13a (dotted lines), suggesting compacted FW200 soot contains more PCF-relevant mesopores which are required to be smaller than ~ 10 nm at $T < \text{HNT}$ (Marcolli, 2014, 2020). PR90comp soot mainly contains pore structures smaller than ~ 17 nm, as shown by the PSD curve (dashed line) in Fig. 13b calculated with a soot–water contact angle 75° . However, PR90fresh soot shows less mesopore abundance in the pore size range from 2 to 9 nm as its PSD curve stays below that of PR90comp soot. It is just the paucity in this pore size range that leads to the poor IN ability of PR90 fresh soot particles. Additionally, the PSD curves retrieved from DVS isotherms showing a monotonic decrease with increasing pore size are different from PSD curves derived from Ar/N₂ measurements in Fig. 12 showing a unimodal distribution as a function of pore radius. This originates from contact angle dependence of soot–water interaction and is also because of the different nature of the probe gas.

Overall, PSD results for FW200 soot derived from DVS measurements coincide with the N₂ or Ar PSD results to imply that physical agitation-induced soot compaction results in the increment of small mesopores required for the PCF activation and thereby promotes compacted soot being more active INPs. We believe soot PSD results derived from DVS measurements are more relevant to soot particle IN activities as they both are associated with soot–water interaction activities. In particular, mesopore abundance is a more important predictor of soot particle IN via the PCF mechanism. As discussed in Sect. 3.2, both soot particle IN and compaction-induced IN enhancement show size dependence. We suggest

that the size dependence is in fact mesopore abundance dependence as a function of aggregate size. In this context, each mesopore can be viewed as a local PCF-active site. Only when the mesopore number density reaches a critical amount can the aggregate nucleate ice under a PCF-favored T and RH condition. For example, the PSD results in Fig. 13a can explain why 400 nm FW200comp soot AF increases with a larger slope than that of the fresh soot particles at 233 K as presented in Fig. 6. According to the soot PCF mechanism (Marcolli, 2014), cylindrical mesopores with a radius smaller than ~ 9 nm are relevant to PCF at $T = 233$ K and $\text{RH}_w < 94\%$, assuming a 45° contact angle for FW200 soot. The PSD results in Figs. 12a and 13a for FW200 soot both suggest that FW200comp soot has more mesopores smaller than 9 nm in comparison to FW200fresh soot.

Similarly, the mesopore increment for compacted PR90 soot (see Fig. 13b) also explains its enhanced water interaction ability and promoted IN activity. Particularly, the reason for that compaction can promote PR90 soot IN significantly at 228 K but not at 233 K can also be explained by its PSD results. With a soot–water contact angle assumption 75° , PR90 soot shows the existence of small mesopores between ~ 2 and ~ 17 nm (Fig. 13b). Though these mesopores are small enough to induce capillary condensation but are too small to provide enough space for the ice embryo to grow out of the pore at 233 K or to nucleate ice in a small volume of water on the short timescale of our experiments (~ 10 s), i.e., the PCF activation is limited by ice crystal growth at $T = 233$ K or freezing rate. For example, Marcolli et al. (2021) reported that a three-membered-ring pore in a tetrahedral primary particle packing is smaller in pore volume than a four-membered-ring pore in a cubic packing and is more likely to have the ice growth limitation scenario. This can be the possible case for PR90comp soot. The second possible scenario is the limitation of a low homogeneous freezing rate for supercooled porewater because the supercooled porewater homogeneous freezing rate is both T and the porewater volume dependent. Based on the parameterization for homogeneous freezing in confined structures (Ickes et al., 2015; Marcolli, 2020), calculation can be made to show some evidence. For example, 400 nm PR90comp soot particles with a 75° water contact angle show onset $S_1 = 1.31$ at $T = 228$ K, suggesting its 3 nm mesopore is responsible for porewater capillary condensation. However, assuming a cylindrical pore with a radius of 3 nm and a height 400 nm, supercooled porewater homogeneous freezing at 228 K takes ~ 0.1 s, whereas it takes ~ 25 s at 233 K which is longer than the particle residence time in the HINC chamber in this study. Therefore, PR90comp soot shows a strong T dependence and a PCF suppression from $T = 228$ to 233 K. Again, linking the PSD results analyzed from DVS measurements for FW200 and PR90 soot to the IN results, both low soot–water contact angle and mesopore abundance exert advantages for FW200 soot in PCF activation and contribute to its higher IN effectiveness compared to PR90 soot.

4 Conclusion and summary

The effect of soot particle morphology changes on its IN ability has been systematically investigated in this study. The soot aggregate morphology was modified only by applying physical agitation. Under mixed-phase and cirrus cloud T conditions, size-selected soot particle IN activities were measured under varying RH conditions at a fixed T . Soot aggregate morphological properties were characterized both online and offline for IN result interpretation, including effective density of size-selected soot particles, soot aggregate microscopic TEM images, bulk sample soot–water interaction ability tested by DVS measurements, as well as soot sample PSD analysis based on Ar, N₂ and water sorption measurements. Soot aggregate densification (compaction) caused by physical agitation is shown by a mass increase in size-selected soot particles after a certain period of agitation. Further evidence is observed from more compacted soot aggregate microscopic TEM images for these aggregates dispersed by agitated soot samples, demonstrating that the physical agitation can modify soot morphological properties and squeeze soot aggregate by shrinking intra-aggregate structures. The experimental results show that there is a significant IN enhancement for compacted 400 and 200 nm FW200 and compacted 400 nm PR90 soot particles. The size dependence shows that a larger aggregate size for the same soot sample is more IN-active because the larger soot aggregates are more sensitive to the induced aggregate densification, causing mesopore enrichment. With decreasing particle size, the physical-agitation-caused mesopore increment plays a limited role in promoting soot IN abilities. The limited IN activity of small soot aggregates (≤ 200 nm) arises from the low probability of PCF-relevant mesopores within a small soot aggregate since small aggregates comprised of a limited number of primary particles cannot form sufficient mesopore structures. PCF activation may require a cluster of mesopores for IN rather than a few solitary mesopores. Importantly, qualitative analysis of DVS isotherms and PSD quantitative calculation based on vapor isotherms (water, Ar and N₂) reveal the role of contact angle and morphology in the soot particle PCF IN process. It is demonstrated that compacted soot nucleating ice crystals under lower RH conditions than the fresh soot at the same T results from the enrichment in the mesopore population. DVS measurements also show that soot–water contact angle does not change after soot compaction, demonstrating the single importance of soot particle PSD in the IN enhancement of compacted soot via the PCF mechanism. Compacted FW200 soot active IN ability is attributed to its abundant mesopores and low water contact angle. Notably, 200 and 400 nm compacted PR90 soot with a high contact angle can be competitive INPs to fresh FW200 soot particles at $T < \text{HNT}$, suggesting that the existence of PCF-relevant mesopores, as a prerequisite for soot PCF activation, is more important than the contact angle. In conclusion, soot PCF activation relies

more on the mesopore availability, and a lower soot–water contact angle also facilitates the PCF process.

In summary, findings in this study have revealed the implications of soot particles in cirrus cloud formation through a PCF pathway strongly depending on its morphology. Our study suggests that compacted soot particles with abundant mesopore structures and a low contact angle favor ice activation via PCF under low T and RH conditions and potentially make considerable contributions to cirrus cloud formation competing with liquid droplet homogeneous freezing. For instance, soot aggregates were detected in contrail residuals and demonstrated to be engaged in contrail evolution and cirrus cloud formation (Twohy and Gandrud, 1998; Petzold et al., 1998). Some aero-engine soot particles emitted directly in the upper troposphere can act as INPs in a first ice formation cycle and release residuals upon sublimation of this ice. The released residuals can have a compacted aggregate structure (Bhandari et al., 2019; Mahrt et al., 2020b). After such a cloud processing scenario, originally fresh and hydrophobic large-sized soot particles will become more active INPs, thereby impacting cirrus cloud formation more significantly. Therefore, this laboratory study implies the importance of cloud processing effects on the IN activity of soot particles. Soot particles, however, may undergo different atmospheric aging processes simultaneously, such as cloud processing and external material coating, resulting in more complex property changes (Zhang et al., 2008; George et al., 2015; Bhandari et al., 2019) than the single compactness change in this study. Consequently, other property changes induced by atmospheric aging, such as surface wettability change, should also be considered when evaluating atmospheric soot IN. Furthermore, our 60 and 100 nm soot particle IN activity results suggest that near source aviation soot particles with a particle size distribution smaller than 100 nm (Lobo et al., 2015; Moore et al., 2017; Liati et al., 2019) are potentially PCF-relevant compared to the homogeneous freezing parameterization reported by Schneider et al. (2021), which also extends the IN data for size-selected soot particles to a lower size limit given that the smallest size-selected soot particle IN is 100 nm reported in the literature (Mahrt et al., 2018; Nichman et al., 2019). Our study demonstrates that even in the absence of chemical modifications, morphology alone can directly regulate the IN ability of soot particles. In the future, comprehensive soot property characterization with an emphasis on soot PSD is necessitated to evaluate soot IN ability as well as better estimates of the soot–water contact angle. Finally, synthetic aviation fuels, which contain a negligible amount of aromatics and are gaining an increasing share in aviation fuel usage, can lead to a reduction in aero-engine particulate matter mass and number emissions (Duong et al., 2018; Xue et al., 2019) and significant decrease in the GMD of aero-engine-emitted particle population by up to 40 % (Corporan et al., 2007, 2011). Considering the IN dependence on soot particle size, it implies that the soot emission from aviation activities fueled by

synthetic fuels should have a decreasing impact on cirrus formation. Bräuer et al. (2021) reported that low-aromatic bio-fuel blends considerably reduce aircraft emitted ice by 40 %. However, Kärcher et al. (2021) noted that a small number of active soot particles in aviation contrails may still modulate cirrus cloud microphysics, by decreasing the ice crystal number concentration but increasing the ice crystal mean size. Hence, further constraining the understanding of the IN activity of small-sized soot particles is necessary to mitigate aviation soot impacts on climate.

Appendix A: SMPS and CPMA measurements for soot samples

A1 Raw size distribution of soot samples

SMPS measurement results show that the agitated particle size distribution curve shifts to smaller size ranges and the peak mode value decreases compared to those of the fresh particles. As a function of mobility size, the particle number size distribution curves for fresh and compacted FW200 and PR90 soot samples are presented in Fig. A1. This demonstrates that agitation-induced particle physical property change occurs over the entire size range of the fresh particles.

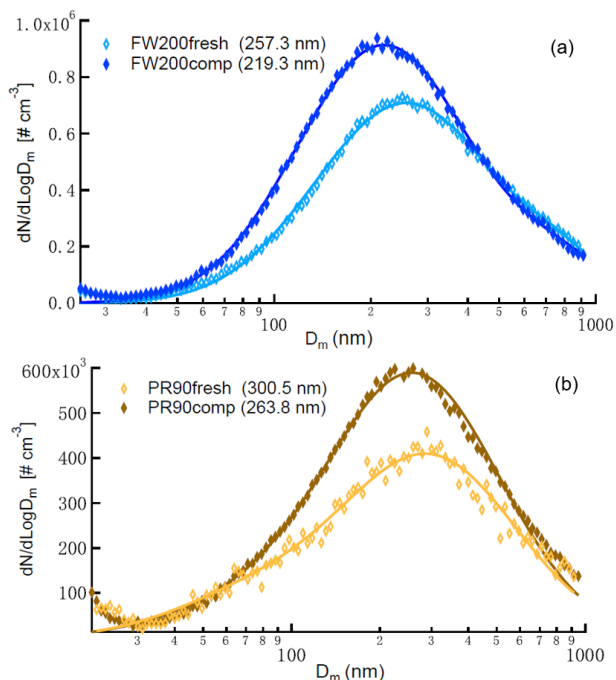


Figure A1. The size distribution of polydisperse fresh and compacted FW200 (a) and PR90 (b) soot particles as a function of mobility size (D_m). Number values in the legend indicate the size distribution mode of the aerosol. The size distribution is shown without multiple charge correction. Solid lines are lognormal fitting for the soot aerosol indicated by data points with the same color.

Table A1. The sheath–aerosol flow configuration used for DMA size selection and SMPS size distribution measurements. D_m stands for the size selection value used by DMA. The aerosol-to-sheath flow ratio in the DMA and SMPS is indicated by F_{DMA} and F_{SMPS} , respectively. The flow rate is in volumetric liters per minute.

D_m (nm)	F_{DMA}	F_{SMPS}	Size scanning range
60	1 : 18	1.5 : 9	8–317 nm
100	1 : 15	1.5 : 7.5	9–359 nm
200	1 : 12	0.3 : 3.0	14–594 nm
400	1 : 7.3	0.3 : 1.8	19–851 nm

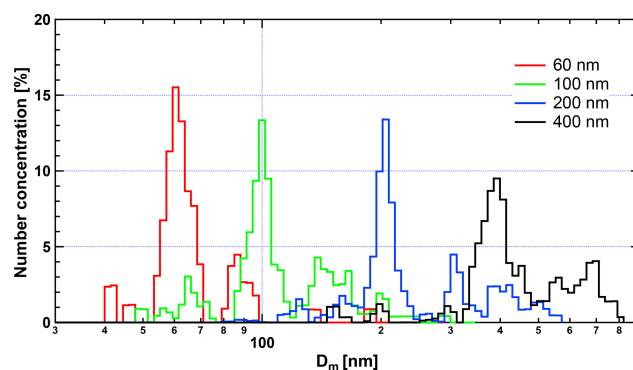


Figure A2. Number size distribution as a function of mobility size (D_m) for 60, 100, 200 and 400 nm size-selected FW200 fresh soot particles.

A2 The quality of soot particle size selection

The DMA selects aerosol particles based on their electrical mobility size, equivalent to the diameter of a perfect spherical particle. Fractal soot particles are not spherical, and thus size-selected soot aerosol particles deviate from the size selection value and show a biased particle size distribution. We, therefore, measured the particle size distribution of our size-selected soot samples to check the DMA size selection quality (see Figs. A2 to A5). The DMA has a classifier 3080 running with a 3081 column and a polonium radiation source. The SMPS system for particle size distribution measurements has a classifier 3082 and a 3081 column with an X-ray neutralizer, coupling with a CPC 3776 (a low-flow mode 0.3 L min^{-1} and a high-flow mode 1.5 L min^{-1}). The DMA and the SMPS system were operated with the same flow configurations (Table A1) as for all IN experiments.

A3 Particle fractal dimension calculation

The fractal dimension D_f (presented in Table 1), describing soot particle morphology, is derived from the power-law relation between particle mobility size and mass (Schmidt-Ott et al., 1990; Olfert and Rogak, 2019).

$$m = C D_m^{D_f}, \quad (\text{A1})$$

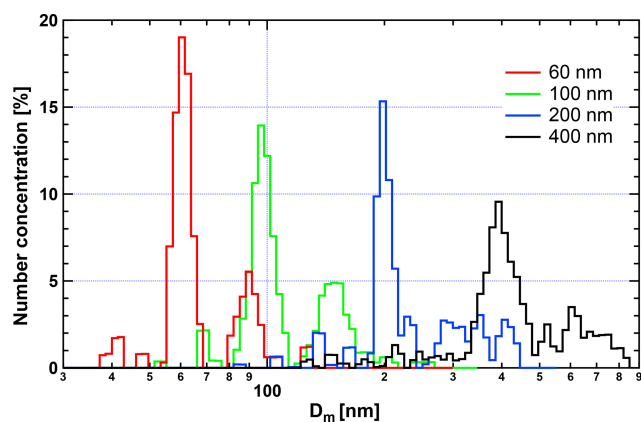


Figure A3. Number size distribution as a function of mobility size (D_m) for 60, 100, 200 and 400 nm size-selected FW200 compacted soot particles.

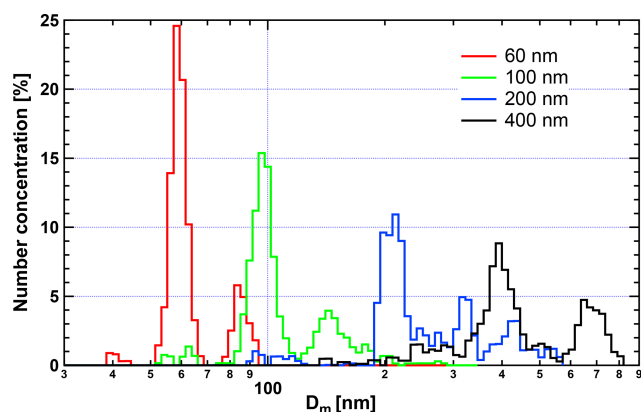


Figure A4. Number size distribution as a function of mobility size (D_m) for 60, 100, 200 and 400 nm size-selected PR90 fresh soot particles.

where C is the mass-mobility constant.

Appendix B: Formulation for the BET and PSD analysis method based on DVS isotherms

B1 BET method formulation

The BET model, which was developed by Brunauer et al. (1938), is a common approach to estimate specific surface area values. Taking gas molecule interactions in adjacent layers into account, Langmuir theory (Langmuir, 1918) of monolayer adsorption was extended to calculate the surface area of the adsorbent due to adsorption by multiple layers of adsorbates. The BET formula is given by

$$\frac{1}{V \left(\frac{p_0}{p} - 1 \right)} = \frac{1}{V_m C} + \frac{C - 1}{V_m C} \frac{p}{p_0}. \quad (\text{B1})$$

Here, V is the volume of the adsorbate adsorbed by per-gram sample under each relative pressure (p/p_0) condition, C is

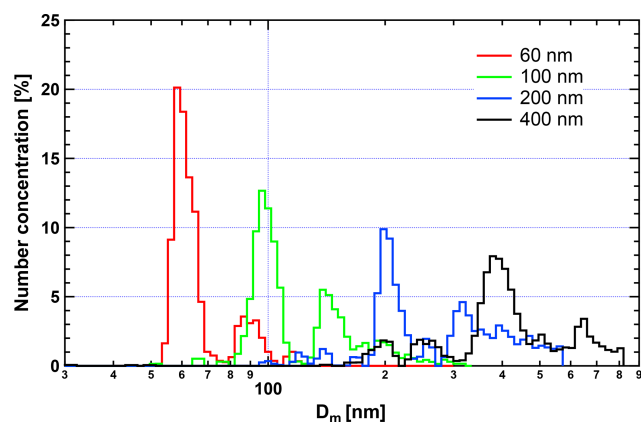


Figure A5. Number size distribution as a function of mobility size (D_m) for 60, 100, 200 and 400 nm size-selected PR90 compacted soot particles.

a constant and V_m stands for the volume of the adsorbate gas equivalent to a monolayer coverage over the adsorbent. Plotting the raw data as a function of p/p_0 results in a linear section of the curve, which is the BET range. The V_m and C values can be easily determined by firstly working out the slope and intercept of the linear function. Finally, the BET-specific surface area (S_{BET}) can be calculated by the equation below:

$$S_{\text{BET}} = \frac{V_m \rho \bar{N} A}{\bar{M}}, \quad (\text{B2})$$

where \bar{N} is the Avogadro number 6.02×10^{23} , ρ is the adsorbate density in the gas phase, $\rho_{\text{N}_2} = 0.809 \times 10^{-3} \text{ g cm}^{-3}$ (Sing, 2014a), A is the cross-sectional area of a probe gas molecule, $A_{\text{N}_2} = 0.162 \text{ nm}^2$ (Sing, 2014a), and \bar{M} is the molar mass of the adsorbate, $\bar{M}_{\text{N}_2} = 28 \text{ g mol}^{-1}$. In this study, the BET range for N_2 is from $p/p_0 = 0.05$ to 0.15. The S_{BET} values for N_2 physisorption measurements are shown in Table 1.

B2 PSD formulation for DVS data based on the Kelvin equation

With a focus on understanding the mesopore PSD, an analysis method based on the Kelvin equation was applied. The Kelvin equation can estimate the critical pore size required for mesopore capillary condensation as a function of soot-water contact angle (θ) and p/p_0 . The equation is given as Eq. (2) in Sect. 2.3.2. The interfacial tension γ_{sl} between the solid- and liquid-phase water is taken as 72 mJ m^{-2} at 298 K (Floriano and Angell, 1990; Hruby et al., 2014). The water molar volume v_s is taken as $18.02 \text{ cm}^3 \text{ mol}^{-1}$. According to the Kelvin equation, larger pores require larger p/p_0 values than smaller pores to trigger capillary condensation assuming the sample with a uniform θ distribution on its surface. With decreasing p/p_0 level in the desorption process, large

pores release adsorbed water first. In this case, the Δm at a p/p_0 value corresponds to the amount of water desorbed by the pore structures (porewater) whose r_k (see Eq. 2) are larger than the value calculated by the Kelvin equation at this p/p_0 level. The equation to calculate the desorbed water volume or porewater volume is given by

$$V_w = \frac{\Delta m}{\rho_w}. \quad (\text{B3})$$

Here, ρ_w is the density of bulk water at 298 K, taking $0.99623 \text{ g cm}^{-3}$ according to Marcolli (2017). Wheeler (1951) suggested that the amount of water desorbed from the pore is comprised of two parts, including the volume of water freely released from the pore by inverse capillary condensation and the volume of a quasi-liquid layer with a thickness t over an open surface. Thus, the radius of a pore (r_p) capable of hosting water calculated by the Kelvin equation equals the sum of r_k and t , given as below:

$$r_p = r_k + t, \quad (\text{B4})$$

with decreasing p/p_0 , V_w increases as a function of decreasing r_p . Thus, the relation between V_w and r_p can be derived. Note that the change in V_w at each p/p_0 desorption step refers to the sum of the liquid desorption by the Kelvin effect and the desorption from the surface of the adsorption film within pores emptied at previous steps. Considering this and assuming a cylindrical pore shape, Shkol'nikov and Sidorova (2007) proposed a PSD function as below:

$$\frac{dV_p}{dr_p} = \left(\frac{r_p}{r_k}\right)^2 \cdot \frac{dV_w}{dr_p}. \quad (\text{B5})$$

Then, the expression for PSD as a function of r_p can be established by assuming a fixed value for θ .

In order to derive the function of PSD with pore size r_p as the variable, the quasi-liquid layer thickness (t) is still needed. Hence, the following data analysis will be conducted. The DVS raw data are logged as the Δm of water adsorbed by the sample to the raw sample weight at varying and discrete p/p_0 values. A high-order polynomial function was first applied to fit the expression for Δm as a function of p/p_0 in MATLAB with the coefficient $R^2 > 0.999$. Thus, the approximate function $\Delta m = f(p/p_0)$ was obtained. The mole number of water molecules (n_w) adsorbed by per-unit mass of the adsorbent can be given as

$$n_w = \frac{\Delta m}{M_w}, \quad (\text{B6})$$

where M_w , equalling $18.105 \text{ g mol}^{-1}$, is the molar mass of water. Then n_w and V_w can be expressed by p/p_0 , according to Eqs. (B6) and (B3), respectively. The excess surface work (Adolphs and Setzer, 1996a, b; Churaev et al., 1998), describing the energetic characterization of the adsorption and desorption process, was applied in this study. The Gibbs

free energy change (ΔG) during the adsorption and desorption process, which describes the thermal dynamic state of the process and is given as Eq. (B7), can be formulated as a function of n_w and p/p_0 . Essentially, ΔG is a function of p/p_0 and given as

$$\Delta G = n_w RT \ln \left(\frac{p}{p_0} \right). \quad (\text{B7})$$

According to Adolphs (2007), there is a critical p/p_0 value and a critical value n_c for ΔG to reach a minimum denoting the water monolayer formation over the sample particle surfaces. The value of n_c can be determined by calculating the p/p_0 value for the minimum extremum of ΔG . The thickness (t) of the quasi-liquid water layer is also needed to determine r_p and is given as

$$t = \frac{n_w}{n_c} \tau. \quad (\text{B8})$$

$\tau = 0.24 \text{ nm}$ is the single water molecule thickness, according to Georgi et al. (2017). Finally, given a p/p_0 , the pore radius (r_p) can be calculated by substituting Eqs. (2) (see Sect. 2.3.2) and (B8) into Eq. (B4). By solving simultaneous Eqs. (B3), (B4) and (B5), the PSD results can be achieved. The corresponding results are presented in Fig. 13 (see Sect. 3.4).

Appendix C: Activation fraction curves of soot particles with different sizes at T from 243 to 218 K

In Figs. C1 to C4, the AF values derived from the $5 \mu\text{m}$ OPC channel for different soot samples with the same sizes (400, 200, 100 and 60 nm) are plotted versus RH values at different T . The absence of a $5 \mu\text{m}$ OPC signal at $T = 243$ and 238 K demonstrates that there is no ice nucleation onto soot particles since ice crystals if formed would be able to grow to $5 \mu\text{m}$ at this temperature. Therefore, the signals in the $1 \mu\text{m}$ OPC channel for $T = 243$ and 238 K are water droplets but not ice crystals. To compare soot particle IN ability dependence on size, the AF values for the same fresh and compacted FW200 and PR90 soot samples with different sizes as a function of RH at a fixed T are plotted in Figs. C5 to C8, corresponding to a $1 \mu\text{m}$ OPC channel. Each curve is an average of at least three individual RH scan experiments. Detailed discussions on IN results are presented in the main text in Sect. 3.2.

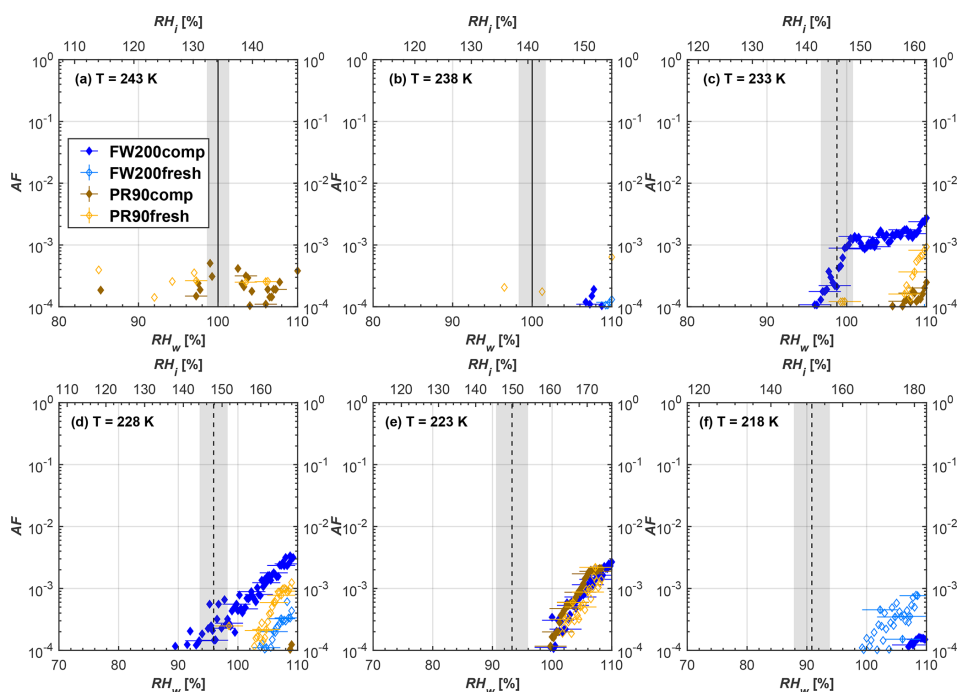


Figure C1. Averaged AF curves as a function of RH_w and RH_i from the $5\ \mu\text{m}$ OPC channel for 400 nm fresh and compacted FW200 and PR90 soot particles at different T . Black solid lines represent water saturation conditions according to Murphy and Koop (2005). Black dashed lines denote the expected RH values for solution droplet homogeneous freezing at each T (Koop et al., 2000). The grey shading shows the possible variation range in RH that aerosol in HINC can encounter for the calculated homogeneous freezing RH values at each T .

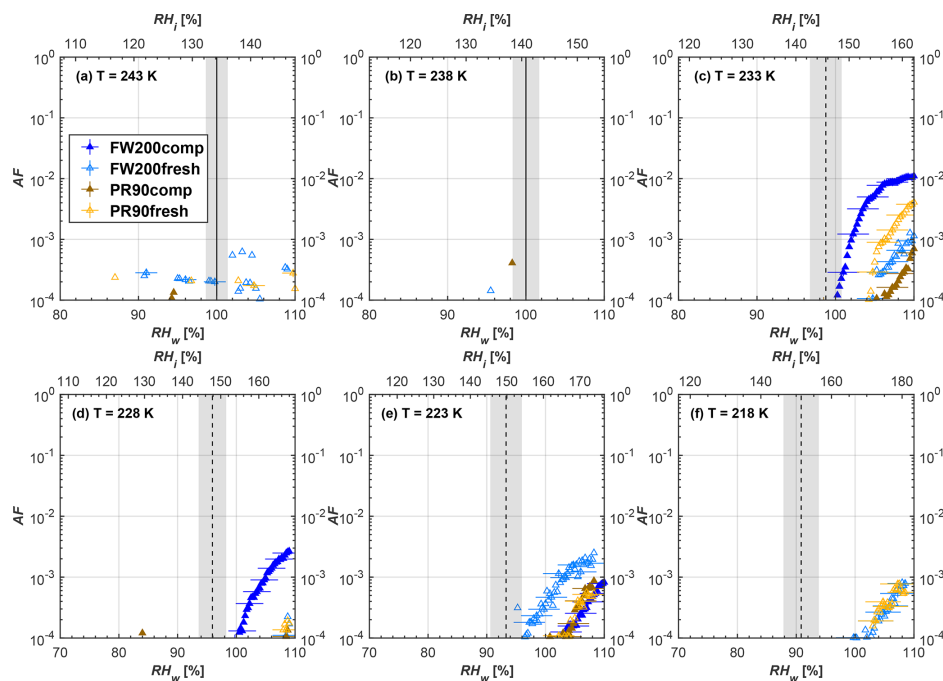


Figure C2. Averaged AF curves as a function of RH_w and RH_i from the $5\ \mu\text{m}$ OPC channel for 200 nm fresh and compacted FW200 and PR90 soot particles at different T . Black solid lines represent water saturation conditions according to Murphy and Koop (2005). Black dashed lines denote the expected RH values for solution droplet homogeneous freezing at each T (Koop et al., 2000). The grey shading shows the possible variation range in RH that aerosol in HINC can encounter for the calculated homogeneous freezing RH values at each T .

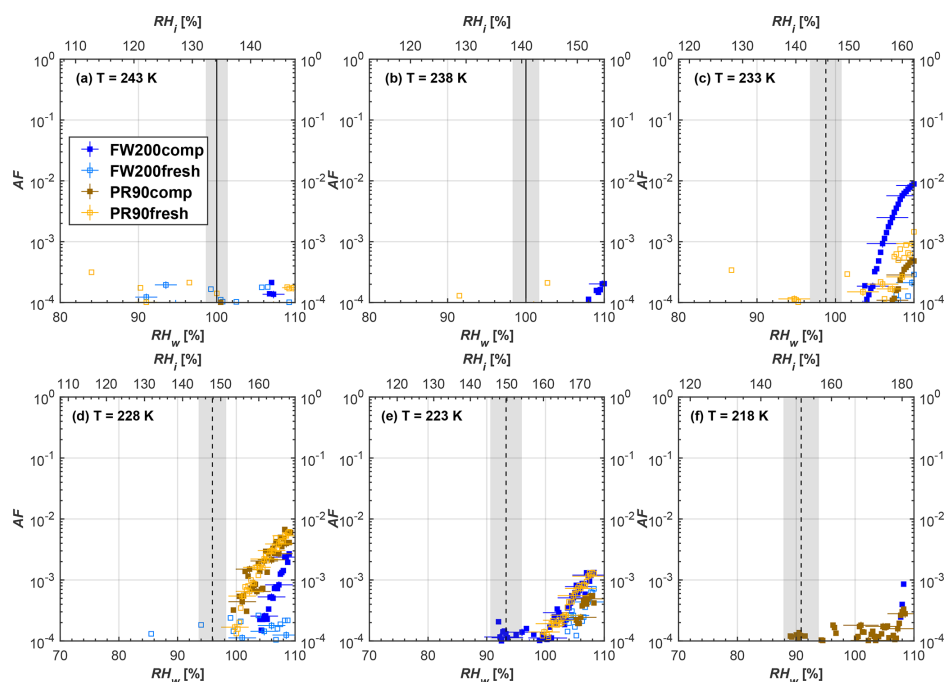


Figure C3. Averaged AF curves as a function of RH_w and RH_i from the 5 μm OPC channel for 100 nm fresh and compacted FW200 and PR90 soot particles at different T . Black solid lines represent water saturation conditions according to Murphy and Koop (2005). Black dashed lines denote the expected RH values for solution droplet homogeneous freezing at each T (Koop et al., 2000). The grey shading shows the possible variation range in RH that aerosol in HINC can encounter for the calculated homogeneous freezing RH values at each T .

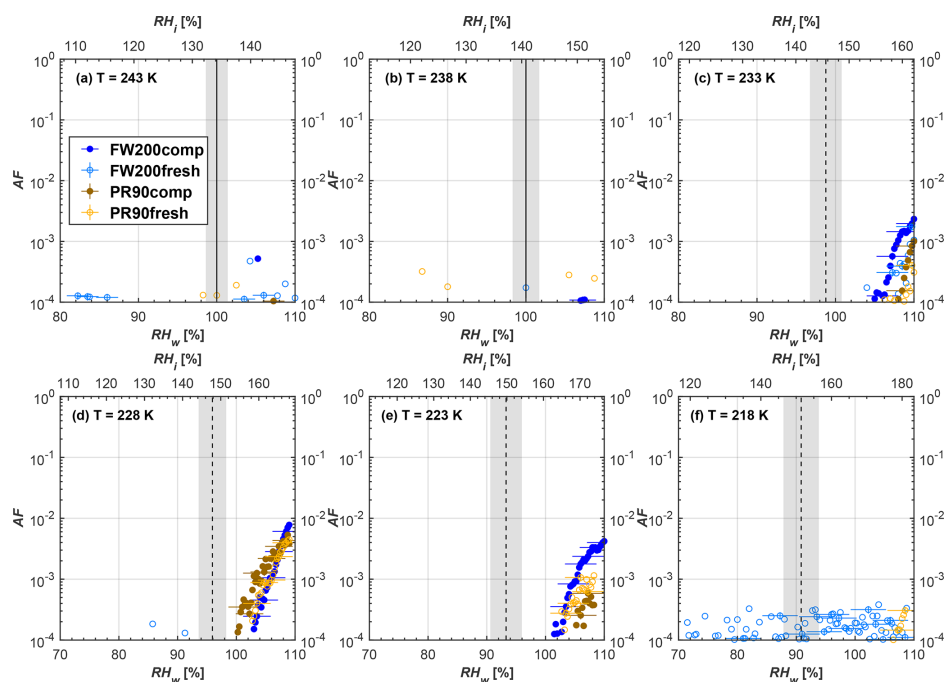


Figure C4. Averaged AF curves as a function of RH_w and RH_i from the 5 μm OPC channel for 60 nm fresh and compacted FW200 and PR90 soot particles at different T . Black solid lines represent water saturation conditions according to Murphy and Koop (2005). Black dashed lines denote the expected RH values for solution droplet homogeneous freezing at each T (Koop et al., 2000). The grey shading shows the possible variation range in RH that aerosol in HINC can encounter for the calculated homogeneous freezing RH values at each T .

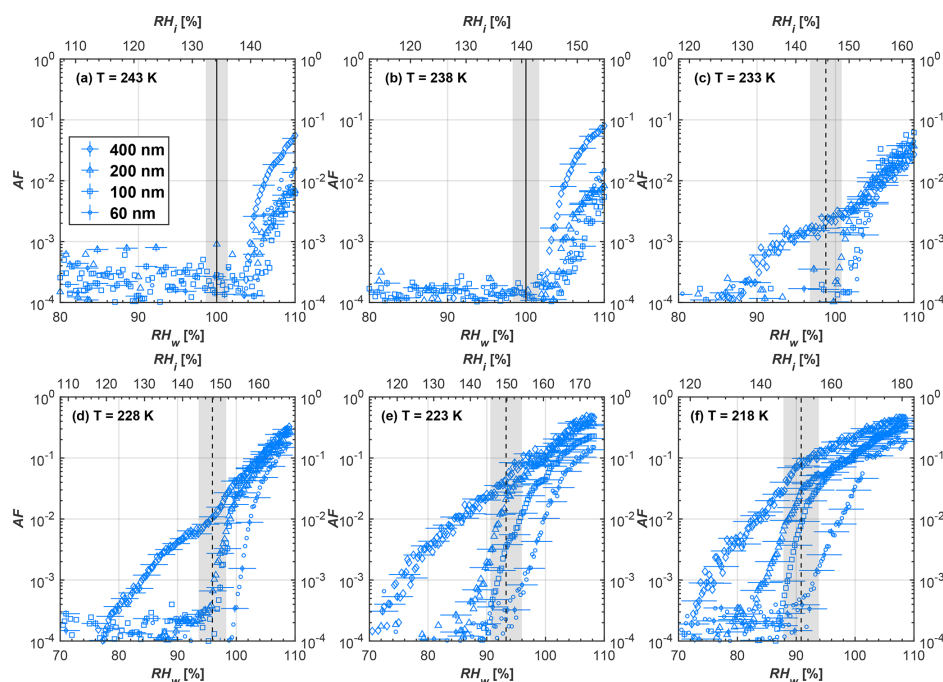


Figure C5. Averaged AF curves as a function of RH_w and RH_i from the 1 μm OPC channel for size-selected fresh FW200 soot particles at different T . Black solid lines represent water saturation conditions according to Murphy and Koop (2005). Black dashed lines denote the expected RH values for solution droplet homogeneous freezing at each T (Koop et al., 2000). The grey shading shows the possible variation range in RH that aerosol in HINC can encounter for the calculated homogeneous freezing RH values at each T .

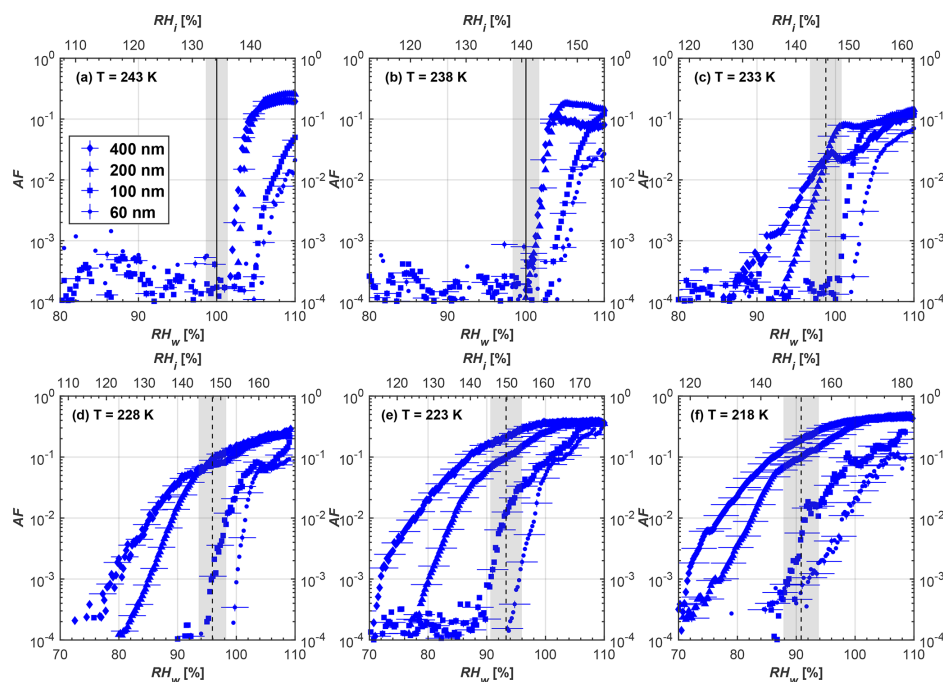


Figure C6. Averaged AF curves as a function of RH_w and RH_i from the 1 μm OPC channel for size-selected compacted FW200 soot particles at different T . Black solid lines represent water saturation conditions according to Murphy and Koop (2005). Black dashed lines denote the expected RH values for solution droplet homogeneous freezing at each T (Koop et al., 2000). The grey shading shows the possible variation range in RH that aerosol in HINC can encounter for the calculated homogeneous freezing RH values at each T .

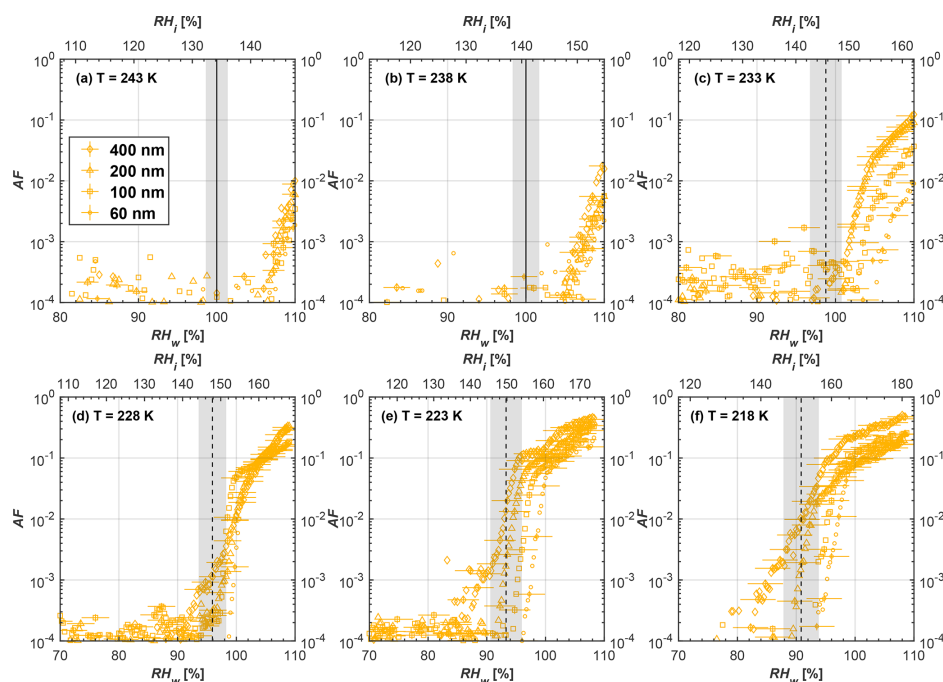


Figure C7. Averaged AF curves as a function of RH_w and RH_i from the 1 μm OPC channel for size-selected fresh PR90 soot particles at different T . Black solid lines represent water saturation conditions according to Murphy and Koop (2005). Black dashed lines denote the expected RH values for solution droplet homogeneous freezing at each T (Koop et al., 2000). The grey shading shows the possible variation range in RH that aerosol in HINC can encounter for the calculated homogeneous freezing RH values at each T .

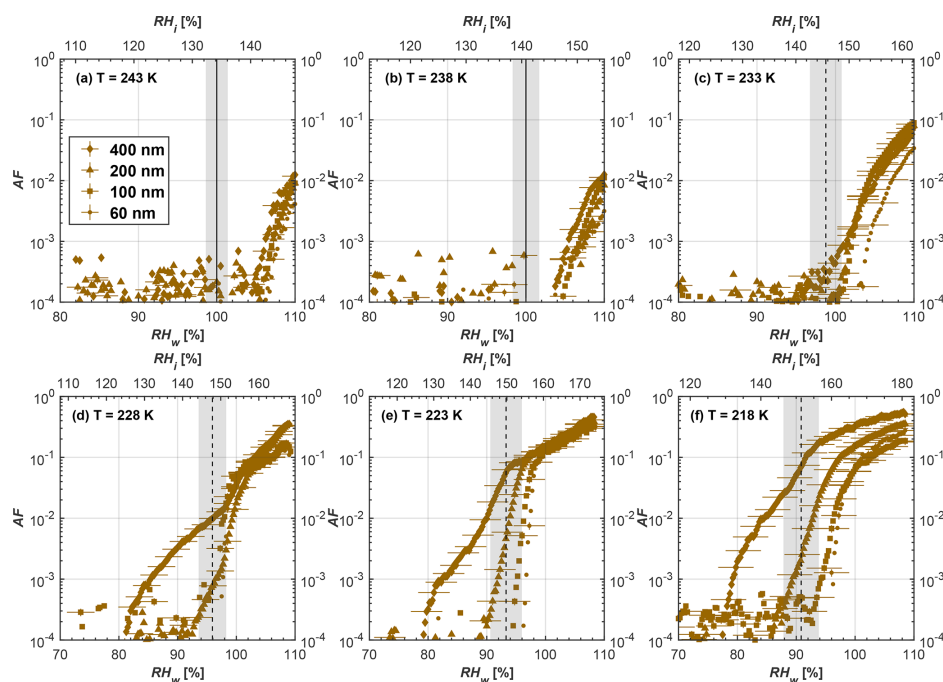


Figure C8. Averaged AF curves as a function of RH_w and RH_i from the 1 μm OPC channel for size-selected compacted PR90 soot particles at different T . Black solid lines represent water saturation conditions according to Murphy and Koop (2005). Black dashed lines denote the expected RH values for solution droplet homogeneous freezing at each T (Koop et al., 2000). The grey shading shows the possible variation range in RH that aerosol in HINC can encounter for the calculated homogeneous freezing RH values at each T .

Appendix D: Gas sorption isotherms

Ar and N₂ adsorption and desorption isotherms are plotted as adsorbed gas volume under the STP (standard temperature and pressure) condition versus the relative pressure (p/p_0) in Figs. D1 and D2, respectively. Both measurements are based on the manometric method, which measures the amount of gas removed from the gas phase to evaluate the gas adsorption activity of the soot sample. Ar physisorption measurements are conducted at 87 K, and the N₂ measurements are performed at 77 K. The p/p_0 range is from near zero to approximately unity. Both adsorption and desorption processes are performed. Each curve stands for a single run.

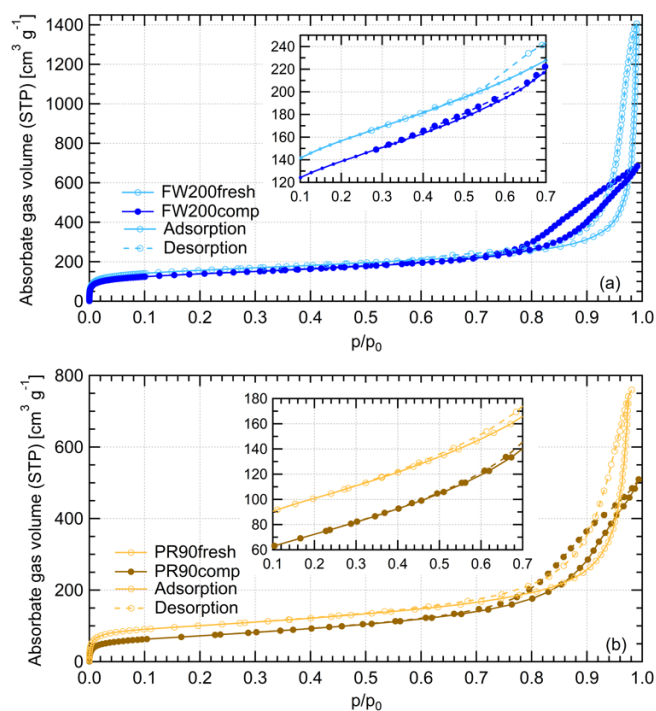


Figure D1. Ar adsorption and desorption isotherms plotted as the volume of adsorbed gas for per-gram soot sample at standard temperature and pressure (STP) as a function of the relative pressure conditions for fresh and compacted (a) FW200 and (b) PR90 soot. Solid lines indicate adsorption branches, and the dashed lines indicate desorption branches.

Data availability. The data presented in this publication are available at <https://doi.org/10.3929/ethz-b-000511281> (Gao et al., 2022).

Author contributions. KG and ZAK designed the experiments and interpreted the data. KG conducted the measurements and collected the raw data. KG wrote the manuscript and prepared all the figures with the help from ZAK. FF helped with the analysis of Ar and N₂ sorption isotherms using BET and BJH approaches. All

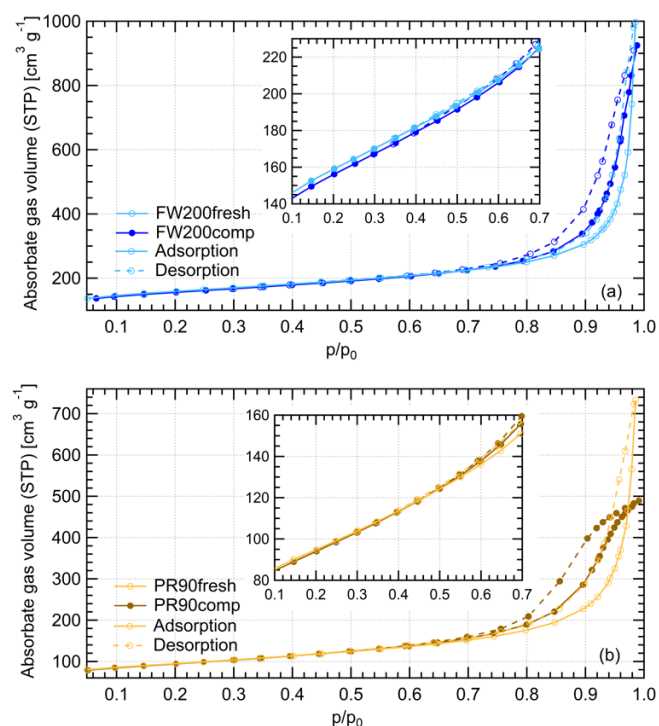


Figure D2. N₂ adsorption and desorption isotherms plotted as the volume of adsorbed gas for per-gram soot sample at STP as a function of the relative pressure conditions for fresh and compacted (a) FW200 and (b) PR90 soot. Solid lines indicate adsorption branches, and the dashed lines indicate desorption branches.

authors discussed, reviewed and edited the manuscript. ZAK supervised the project.

Competing interests. The contact author has declared that neither they nor their co-authors have any competing interests.

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References

- Adolphs, J.: Excess surface work—A modelless way of getting surface energies and specific surface areas directly from sorption isotherms, *Appl. Surf. Sci.*, 253, 5645–5649, <https://doi.org/10.1016/j.apsusc.2006.12.089>, 2007.
- Adolphs, J. and Setzer, M. J.: A Model to Describe Adsorption Isotherms, *J. Colloid Interf. Sci.*, 180, 70–76, 1996a.
- Adolphs, J. and Setzer, M. J.: Energetic Classification of Adsorption Isotherms, *J. Colloid Interf. Sci.*, 184, 443–448, <https://doi.org/10.1006/jcis.1996.0639>, 1996b.
- Archuleta, C. M., DeMott, P. J., and Kreidenweis, S. M.: Ice nucleation by surrogates for atmospheric mineral dust and mineral dust/sulfate particles at cirrus temperatures, *Atmos. Chem. Phys.*, 5, 2617–2634, <https://doi.org/10.5194/acp-5-2617-2005>, 2005.
- Barrett, E. P., Joyner, L. G., and Halenda, P. P.: The Determination of Pore Volume and Area Distributions in Porous Substances. I. Computations from Nitrogen Isotherms, *J. Am. Chem. Soc.*, 73, 373–380, <https://doi.org/10.1021/ja01145a126>, 1951.
- Bhandari, J., China, S., Chandrakar, K. K., Kinney, G., Cantrell, W., Shaw, R. A., Mazzoleni, L. R., Giroto, G., Sharma, N., Gorkowski, K., Gilardoni, S., Decesari, S., Facchini, M. C., Zanca, N., Pavese, G., Esposito, F., Dubey, M. K., Aiken, A. C., Chakrabarty, R. K., Moosmuller, H., Onasch, T. B., Zaveri, R. A., Scarnato, B. V., Fialho, P., and Mazzoleni, C.: Extensive Soot Compaction by Cloud Processing from Laboratory and Field Observations, *Sci. Rep.-UK*, 9, 11824, <https://doi.org/10.1038/s41598-019-48143-y>, 2019.
- Bond, T. C., Doherty, S. J., Fahey, D. W., Forster, P. M., Berntsen, T., DeAngelo, B. J., Flanner, M. G., Ghan, S., Kärcher, B., Koch, D., Kinne, S., Kondo, Y., Quinn, P. K., Sarofim, M. C., Schultz, M. G., Schulz, M., Venkataraman, C., Zhang, H., Zhang, S., Bellouin, N., Guttikunda, S. K., Hopke, P. K., Jacobson, M. Z., Kaiser, J. W., Klimont, Z., Lohmann, U., Schwarz, J. P., Shindell, D., Storelvmo, T., Warren, S. G., and Zender, C. S.: Bounding the role of black carbon in the climate system: A scientific assessment, *J. Phys. Chem. A*, 118, 5380–5552, <https://doi.org/10.1002/jgrd.50171>, 2013.
- Bräuer, T., Voigt, C., Sauer, D., Kaufmann, S., Hahn, V., Scheibe, M., Schlager, H., Huber, F., Le Clercq, P., Moore, R. H., and Anderson, B. E.: Reduced ice number concentrations in contrails from low-aromatic biofuel blends, *Atmos. Chem. Phys.*, 21, 16817–16826, <https://doi.org/10.5194/acp-21-16817-2021>, 2021.
- Brunauer, S., Emmett, P. J., and Teller, E.: Absorption of gasses in multimolecular layers, *J. Am. Chem. Soc.*, 60, 309–319, 1938.
- China, S., Kulkarni, G., Scarnato, B. V., Sharma, N., Pekour, M., Shilling, J. E., Wilson, J., Zelenyuk, A., Chand, D., Liu, S., Aiken, A. C., Dubey, M., Laskin, A., Zaveri, R. A., and Mazzoleni, C.: Morphology of diesel soot residuals from supercooled water droplets and ice crystals: implications for optical properties, *Environ. Res. Lett.*, 10, 114010, <https://doi.org/10.1088/1748-9326/10/11/114010>, 2015.
- Churaev, N. V., Setzer, M. J., and Adolphs, J.: Influence of Surface Wettability on Adsorption Isotherms of Water Vapor, *J. Colloid Interf. Sci.*, 197, 327–333, <https://doi.org/10.1006/jcis.1997.5292>, 1998.
- Colbeck, I., Appleby, L., Hardman, E. J., and Harrison, R. M.: The optical properties and morphology of cloud-processed carbonaceous smoke, *J. Aerosol Sci.*, 21, 527–538, [https://doi.org/10.1016/0021-8502\(90\)90129-L](https://doi.org/10.1016/0021-8502(90)90129-L), 1990.
- Connolly, P. J., Möhler, O., Field, P. R., Saathoff, H., Burgess, R., Choulaton, T., and Gallagher, M.: Studies of heterogeneous freezing by three different desert dust samples, *Atmos. Chem. Phys.*, 9, 2805–2824, <https://doi.org/10.5194/acp-9-2805-2009>, 2009.
- Corporan, E., DeWitt, M. J., Belovich, V., Pawlik, R., Lynch, A. C., Gord, J. R., and Meyer, T. R.: Emissions Characteristics of a Turbine Engine and Research Combustor Burning a Fischer-Tropsch Jet Fuel, *Energ. Fuels*, 21, 2615–2626, <https://doi.org/10.1021/ef070015j>, 2007.
- Corporan, E., Edwards, T., Shafer, L., DeWitt, M. J., Klingshirn, C., Zabarnick, S., West, Z., Striebich, R., Graham, J., and Klein, J.: Chemical, Thermal Stability, Seal Swell, and Emissions Studies of Alternative Jet Fuels, *Energ. Fuels*, 25, 955–966, <https://doi.org/10.1021/ef101520v>, 2011.
- Cziczo, D. J. and Froyd, K. D.: Sampling the composition of cirrus ice residuals, *Atmos. Res.*, 142, 15–31, <https://doi.org/10.1016/j.atmosres.2013.06.012>, 2014.
- Cziczo, D. J., Froyd, K. D., Hoose, C., Jensen, E. J., Diao, M., Zondlo, M. A., Smith, J. B., Twohy, C. H., and Murphy, D. M.: Clarifying the Dominant Sources and Mechanisms of Cirrus Cloud Formation, *Science*, 340, 1320–1324, 2013.
- David, R. O., Marcolli, C., Fahrni, J., Qiu, Y., Perez Sirkin, Y. A., Molinero, V., Mahrt, F., Bruhwiler, D., Lohmann, U., and Kanji, Z. A.: Pore condensation and freezing is responsible for ice formation below water saturation for porous particles, *P. Natl. Acad. Sci. USA*, 116, 8184–8189, <https://doi.org/10.1073/pnas.1813647116>, 2019.
- David, R. O., Fahrni, J., Marcolli, C., Mahrt, F., Brühwiler, D., and Kanji, Z. A.: The role of contact angle and pore width on pore condensation and freezing, *Atmos. Chem. Phys.*, 20, 9419–9440, <https://doi.org/10.5194/acp-20-9419-2020>, 2020.
- Delhaye, D., Ouf, F.-X., Ferry, D., Ortega, I. K., Penanhoat, O., Peillon, S., Salm, F., Vancassel, X., Focsa, C., Irimiea, C., Harivel, N., Perez, B., Quinton, E., Yon, J., and Gaffie, D.: The MERMOSE project: Characterization of particulate matter emissions of a commercial aircraft engine, *J. Aerosol Sci.*, 105, 48–63, <https://doi.org/10.1016/j.jaerosci.2016.11.018>, 2017.
- DeMott, P. J.: An Exploratory Study of Ice Nucleation by Soot Aerosols, *J. Appl. Meteorol.*, 29, 1072–1079, [https://doi.org/10.1175/1520-0450\(1990\)029<1072:AESOIN>2.0.CO;2](https://doi.org/10.1175/1520-0450(1990)029<1072:AESOIN>2.0.CO;2), 1990.
- Ditas, J., Ma, N., Zhang, Y., Assmann, D., Neumaier, M., Riede, H., Karu, E., Williams, J., Scharffe, D., Wang, Q., Saturno, J., Schwarz, J. P., Katich, J. M., McMeeking, G. R., Zahn, A., Hermann, M., Brenninkmeijer, C. A. M., Andreae, M. O., Poschl, U., Su, H., and Cheng, Y.: Strong impact of wildfires on the abundance and aging of black carbon in the lowermost stratosphere, *P. Natl. Acad. Sci. USA*, 115, E11595–E11603, <https://doi.org/10.1073/pnas.1806868115>, 2018.
- Duong, L. H., Reksowardojo, I. K., Soerawidjaja, T. H., Pham, D. N., and Fujita, O.: The sooting tendency of aviation biofuels and jet range paraffins: effects of adding aromat-

- ics, carbon chain length of normal paraffins, and fraction of branched paraffins, *Combust. Sci. Technol.*, 190, 1710–1721, <https://doi.org/10.1080/00102202.2018.1468323>, 2018.
- Falk, J., Korhonen, K., Malmberg, V. B., Gren, L., Eriksson, A. C., Karjalainen, P., Markkula, L., Bengtsson, P.-E., Virtanen, A., Svenningsson, B., Pagels, J., and Kristensen, T. B.: Immersion Freezing Ability of Freshly Emitted Soot with Various Physico-Chemical Characteristics, *Atmosphere*, 12, 1173, <https://doi.org/10.3390/atmos12091173>, 2021.
- Floriano, M. A. and Angell, C. A.: Surface tension and molar surface free energy and entropy of water to -27.2 degree.C, *J. Phys. Chem.*, 94, 4199–4202, <https://doi.org/10.1021/j100373a059>, 1990.
- Friedman, B., Kulkarni, G., Beránek, J., Zelenyuk, A., Thornton, J. A., and Cziczo, D. J.: Ice nucleation and droplet formation by bare and coated soot particles, *J. Geophys. Res.*, 116, D17203, <https://doi.org/10.1029/2011jd015999>, 2011.
- Galarneau, A., Desplandier, D., Dutartre, R., and Renzo, F. D.: Micelle-templated silicates as a test bed for methods of mesopore size evaluation, *Micropor. Mesopor. Mat.*, 27, 297–308, <https://hal.archives-ouvertes.fr/hal-00172394>, 1999.
- Gao, K., Friebel, F., Zhou, C.-W., and Kanji, Z. A.: Enhanced soot particle ice nucleation ability induced by aggregate compaction and densification, ETH Zurich [data set], <https://doi.org/10.3929/ethz-b-000511281>, 2022.
- Gardner, L., Kruk, M., and Jaroniec, M.: Reference Data for Argon Adsorption on Graphitized and Nongraphitized Carbon Blacks, *J. Phys. Chem. B*, 105, 12516–12523, <https://doi.org/10.1021/jp011745+>, 2001.
- George, C., Ammann, M., D’Anna, B., Donaldson, D. J., and Nizkorodov, S. A.: Heterogeneous photochemistry in the atmosphere, *Chem. Rev.*, 115, 4218–4258, <https://doi.org/10.1021/cr500648z>, 2015.
- Georgi, N., Kolesnikov, A., Uhlig, H., Möllmer, J., Rückriem, M., Schreiber, A., Adolphs, J., Enke, D., and Gläser, R.: Characterization of Porous Silica Materials with Water at Ambient Conditions. Calculating the Pore Size Distribution from the Excess Surface Work Disjoining Pressure Model, *Chem.-Ing.-Tech.*, 89, 1679–1685, <https://doi.org/10.1002/cite.201700042>, 2017.
- Hayati-Ashtiani, M.: Characterization of Nano-Porous Bentonite (Montmorillonite) Particles using FTIR and BET-BJH Analyses, *Part. Part. Syst. Char.*, 28, 71–76, <https://doi.org/10.1002/ppsc.201100030>, 2011.
- Hruby, J., Vins, V., Mares, R., Hykl, J., and Kalova, J.: Surface Tension of Supercooled Water: No Inflection Point down to -25 degrees C, *J. Phys. Chem. Lett.*, 5, 425–428, <https://doi.org/10.1021/jz402571a>, 2014.
- Ickes, L., Welti, A., Hoose, C., and Lohmann, U.: Classical nucleation theory of homogeneous freezing of water: thermodynamic and kinetic parameters, *Phys. Chem. Chem. Phys.*, 17, 5514–5537, <https://doi.org/10.1039/c4cp04184d>, 2015.
- Jacobson, M. Z.: Climate response of fossil fuel and biofuel soot, accounting for soot’s feedback to snow and sea ice albedo and emissivity, *J. Geophys. Res.-Atmos.*, 109, D21201, <https://doi.org/10.1029/2004jd004945>, 2004.
- Jelinek, L. and Kovats, E. S.: True surface areas from nitrogen adsorption experiments, *Langmuir*, 10, 4225–4231, <https://doi.org/10.1021/la00023a051>, 1994.
- Joyce, G. A., Henry, W. M., and Magee, R. W.: Advances in structure measurements of carbon black, *Rubber World*, 240, 27–35, 2009.
- Kanji, Z. A. and Abbatt, J. P. D.: Laboratory studies of ice formation via deposition mode nucleation onto mineral dust and n-hexane soot samples, *J. Geophys. Res.*, 111, D16204, <https://doi.org/10.1029/2005jd006766>, 2006.
- Kanji, Z. A. and Abbatt, J. P. D.: The University of Toronto Continuous Flow Diffusion Chamber (UT-CFDC): A Simple Design for Ice Nucleation Studies, *Aerosol Sci. Tech.*, 43, 730–738, <https://doi.org/10.1080/02786820902889861>, 2009.
- Kanji, Z. A. and Abbatt, J. P. D.: Ice nucleation onto arizona test dust at cirrus temperatures: effect of temperature and aerosol size on onset relative humidity, *J. Phys. Chem. A*, 114, 935–941, <https://doi.org/10.1021/jp908661m>, 2010.
- Kanji, Z. A., Welti, A., Corbin, J. C., and Mensah, A. A.: Black Carbon Particles Do Not Matter for Immersion Mode Ice Nucleation, *Geophys. Res. Lett.*, 47, e2019GL086764, <https://doi.org/10.1029/2019gl086764>, 2020.
- Kärcher, B., Mahrt, F., and Marcolli, C.: Process-oriented analysis of aircraft soot-cirrus interactions constrains the climate impact of aviation, *Commun. Earth Environ.*, 2, 113, <https://10.1038/s43247-021-00175-x>, 2021.
- Kinsey, J. S., Dong, Y., Williams, D. C., and Logan, R.: Physical characterization of the fine particle emissions from commercial aircraft engines during the Aircraft Particle Emissions eXperiment (APEX) 1–3, *Atmos. Environ.*, 44, 2147–2156, <https://doi.org/10.1016/j.atmosenv.2010.02.010>, 2010.
- Kireeva, E. D., Popovicheva, O. B., Persiantseva, N. M., Khokhlova, T. D., and Shonija, N. K.: Effect of black carbon particles on the efficiency of water droplet freezing, *Colloid J.*, 71, 353–359, <https://doi.org/10.1134/s1061933x09030090>, 2009.
- Koehler, K. A., DeMott, P. J., Kreidenweis, S. M., Popovicheva, O. B., Petters, M. D., Carrico, C. M., Kireeva, E. D., Khokhlova, T. D., and Shonija, N. K.: Cloud condensation nuclei and ice nucleation activity of hydrophobic and hydrophilic soot particles, *Phys. Chem. Chem. Phys.*, 11, 7759, <https://doi.org/10.1039/b916865f>, 2009.
- Koop, T. and Murray, B. J.: A physically constrained classical description of the homogeneous nucleation of ice in water, *J. Chem. Phys.*, 145, 211915, <https://10.1063/1.4962355>, 2016.
- Koop, T., Luo, B., Tsias, A., and Peter, T.: Water activity as the determinant for homogeneous ice nucleation in aqueous solutions, *Nature*, 406, 611–614, <https://doi.org/10.1038/35020537>, 2000.
- Kruk, M., Jaroniec, M., and Sayari, A.: Application of Large Pore MCM-41 Molecular Sieves To Improve Pore Size Analysis Using Nitrogen Adsorption Measurements, *Langmuir*, 13, 6267–6273, <https://doi.org/10.1021/la970776m>, 1997.
- Kupgan, G., Liyana-Arachchi, T. P., and Colina, C. M.: NLDFT Pore Size Distribution in Amorphous Microporous Materials, *Langmuir*, 33, 11138–11145, <https://doi.org/10.1021/acs.langmuir.7b01961>, 2017.
- Lacher, L., Lohmann, U., Boose, Y., Zipori, A., Herrmann, E., Bukowiecki, N., Steinbacher, M., and Kanji, Z. A.: The Horizontal Ice Nucleation Chamber (HINC): INP measurements at conditions relevant for mixed-phase clouds at the High Altitude Research Station Jungfraujoch, *Atmos. Chem. Phys.*, 17, 15199–15224, <https://doi.org/10.5194/acp-17-15199-2017>, 2017.

- Langmuir, I.: The Adsorption of Gases on Plane Surfaces of Glass, Mica and Platinum, *J. Am. Chem. Soc.*, 40, 1361–1403, 1918.
- Li, M., Bao, F., Zhang, Y., Song, W., Chen, C., and Zhao, J.: Role of elemental carbon in the photochemical aging of soot, *P. Natl. Acad. Sci. USA*, 115, 7717–7722, <https://doi.org/10.1073/pnas.1804481115>, 2018.
- Liat, A., Schreiber, D., Alpert, P. A., Liao, Y., Brem, B. T., Corral Arroyo, P., Hu, J., Jonsdottir, H. R., Ammann, M., and Dimopoulos Eggenschwiler, P.: Aircraft soot from conventional fuels and biofuels during ground idle and climb-out conditions: Electron microscopy and X-ray micro-spectroscopy, *Environ. Pollut.*, 247, 658–667, <https://doi.org/10.1016/j.envpol.2019.01.078>, 2019.
- Lobo, P., Durdina, L., Smallwood, G. J., Rindlisbacher, T., Siegerist, F., Black, E. A., Yu, Z., Mensah, A. A., Hagen, D. E., Miake-Lye, R. C., Thomson, K. A., Brem, B. T., Corbin, J. C., Abegglen, M., Sierau, B., Whitefield, P. D., and Wang, J.: Measurement of Aircraft Engine Non-Volatile PM Emissions: Results of the Aviation-Particle Regulatory Instrumentation Demonstration Experiment (A-PRIDE) 4 Campaign, *Aerosol Sci. Tech.*, 49, 472–484, <https://doi.org/10.1080/02786826.2015.1047012>, 2015.
- Lohmann, U., Lüönd, F., and Mahrt, F.: *An Introduction to Clouds: From the Microscale to Climate*, Edn. 1, Cambridge University Press, Cambridge, UK, 2016.
- Lohmann, U., Friebel, F., Kanji, Z. A., Mahrt, F., Mensah, A. A., and Neubauer, D.: Future warming exacerbated by aged-soot effect on cloud formation, *Nat. Geosci.*, 13, 674–680, <https://doi.org/10.1038/s41561-020-0631-0>, 2020.
- Lowell, S., Shields, J. E., Thomas, M. A., and Thommes, M.: *Characterization of Porous Solids and Powders: Surface Area, Pore Size and Density*, Particle Technology Series, Kluwer Academic Publishers, Dordrecht, Netherlands, <https://doi.org/10.1007/978-1-4020-2303-3>, 2004.
- Ma, X., Zangmeister, C. D., Gigault, J., Mulholland, G. W., and Zachariah, M. R.: Soot aggregate restructuring during water processing, *J. Aerosol Sci.*, 66, 209–219, <https://doi.org/10.1016/j.jaerosci.2013.08.001>, 2013.
- Mahrt, F., Marcolli, C., David, R. O., Grönquist, P., Barthazy Meier, E. J., Lohmann, U., and Kanji, Z. A.: Ice nucleation abilities of soot particles determined with the Horizontal Ice Nucleation Chamber, *Atmos. Chem. Phys.*, 18, 13363–13392, <https://doi.org/10.5194/acp-18-13363-2018>, 2018.
- Mahrt, F., Alpert, P. A., Dou, J., Gronquist, P., Arroyo, P. C., Ammann, M., Lohmann, U., and Kanji, Z. A.: Aging induced changes in ice nucleation activity of combustion aerosol as determined by near edge X-ray absorption fine structure (NEXAFS) spectroscopy, *Environ. Sci.*, 22, 895–907, <https://doi.org/10.1039/c9em00525k>, 2020a.
- Mahrt, F., Kilchhofer, K., Marcolli, C., Grönquist, P., David, R. O., Rösch, M., Lohmann, U., and Kanji, Z. A.: The Impact of Cloud Processing on the Ice Nucleation Abilities of Soot Particles at Cirrus Temperatures, *J. Geophys. Res.-Atmos.*, 125, 1–23, <https://doi.org/10.1029/2019jd030922>, 2020b.
- Marcolli, C.: Deposition nucleation viewed as homogeneous or immersion freezing in pores and cavities, *Atmos. Chem. Phys.*, 14, 2071–2104, <https://doi.org/10.5194/acp-14-2071-2014>, 2014.
- Marcolli, C.: Pre-activation of aerosol particles by ice preserved in pores, *Atmos. Chem. Phys.*, 17, 1595–1622, <https://doi.org/10.5194/acp-17-1595-2017>, 2017.
- Marcolli, C.: Technical note: Fundamental aspects of ice nucleation via pore condensation and freezing including Laplace pressure and growth into macroscopic ice, *Atmos. Chem. Phys.*, 20, 3209–3230, <https://doi.org/10.5194/acp-20-3209-2020>, 2020.
- Marcolli, C., Gedamke, S., Peter, T., and Zobrist, B.: Efficiency of immersion mode ice nucleation on surrogates of mineral dust, *Atmos. Chem. Phys.*, 7, 5081–5091, <https://doi.org/10.5194/acp-7-5081-2007>, 2007.
- Marcolli, C., Mahrt, F., and Kärcher, B.: Soot PCF: pore condensation and freezing framework for soot aggregates, *Atmos. Chem. Phys.*, 21, 7791–7843, <https://doi.org/10.5194/acp-21-7791-2021>, 2021.
- McGraw, Z., Storelvmo, T., Samset, B. H., and Stjern, C. W.: Global Radiative Impacts of Black Carbon Acting as Ice Nucleating Particles, *Geophys. Res. Lett.*, 47, e2020GL089056, <https://doi.org/10.1029/2020gl089056>, 2020.
- Möhler, O., Linke, C., Saathoff, H., Schnaiter, M., Wagner, R., Mangold, A., Krämer, M., and Schurath, U.: Ice nucleation on flame soot aerosol of different organic carbon content, *Meteorol. Z.*, 14, 477–484, <https://doi.org/10.1127/0941-2948/2005/0055>, 2005.
- Moore, R. H., Thornhill, K. L., Weinzierl, B., Sauer, D., D’Ascoli, E., Kim, J., Lichtenstern, M., Scheibe, M., Beaton, B., Beyersdorf, A. J., Barrick, J., Bulzan, D., Corr, C. A., Crosbie, E., Jurkat, T., Martin, R., Riddick, D., Shook, M., Slover, G., Voigt, C., White, R., Winstead, E., Yasky, R., Ziemba, L. D., Brown, A., Schlager, H., and Anderson, B. E.: Biofuel blending reduces particle emissions from aircraft engines at cruise conditions, *Nature*, 543, 411–415, <https://doi.org/10.1038/nature21420>, 2017.
- Murphy, D. M. and Koop, T.: Review of the vapour pressures of ice and supercooled water for atmospheric applications, *Q. J. Roy. Meteor. Soc.*, 131, 1539–1565, <https://doi.org/10.1256/qj.04.94>, 2005.
- Nichman, L., Wolf, M., Davidovits, P., Onasch, T. B., Zhang, Y., Worsnop, D. R., Bhandari, J., Mazzoleni, C., and Cziczo, D. J.: Laboratory study of the heterogeneous ice nucleation on black-carbon-containing aerosol, *Atmos. Chem. Phys.*, 19, 12175–12194, <https://doi.org/10.5194/acp-19-12175-2019>, 2019.
- Olfert, J. and Rogak, S.: Universal relations between soot effective density and primary particle size for common combustion sources, *Aerosol Sci. Tech.*, 53, 485–492, <https://doi.org/10.1080/02786826.2019.1577949>, 2019.
- Olfert, J. S., Reavell, K. S., Rushton, M. G., and Collings, N.: The experimental transfer function of the Couette centrifugal particle mass analyzer, *J. Aerosol Sci.*, 37, 1840–1852, <https://doi.org/10.1016/j.jaerosci.2006.07.007>, 2006.
- Persiantseva, N. M., Popovicheva, O. B., and Shonija, N. K.: Wetting and hydration of insoluble soot particles in the upper troposphere, *J. Environ. Monitor.*, 6, 939–945, <https://doi.org/10.1039/b407770a>, 2004.
- Petzold, A. and Schröder, F. P.: Jet engine exhaust aerosol characterization, *Aerosol Sci. Tech.*, 28, 62–76, <https://doi.org/10.1080/02786829808965512>, 1998.
- Petzold, A., Strom, J., Ohlsson, S., and Schröder, F. P.: Elemental composition and morphology of ice-crystal residual particles in cirrus clouds and contrails, *Atmos. Res.*, 49, 21–34, [https://doi.org/10.1016/S0169-8095\(97\)00083-5](https://doi.org/10.1016/S0169-8095(97)00083-5), 1998.
- Popovicheva, O., Persiantseva, N. M., Shonija, N. K., DeMott, P., Koehler, K., Petters, M., Kreidenweis, S., Tishkova, V., Demirdjian, B., and Suzanne, J.: Water interaction with hydrophobic and

- hydrophilic soot particles, *Phys. Chem. Chem. Phys.*, 10, 2332–2344, <https://doi.org/10.1039/b718944n>, 2008a.
- Popovicheva, O. B., Persiantseva, N. M., Tishkova, V., Shonija, N. K., and Zubareva, N. A.: Quantification of water uptake by soot particles, *Environ. Res. Lett.*, 3, 025009, <https://doi.org/10.1088/1748-9326/3/2/025009>, 2008b.
- Popovicheva, O. B., Persiantseva, N. M., Trukhin, M. E., Rulev, G. B., Shonija, N. K., Ya Buriko, Y., Starik, A. M., Demirdjian, B., Ferry, D., and Suzanne, J.: Experimental characterization of aircraft combustor soot: Microstructure, surface area, porosity and water adsorption, *Phys. Chem. Chem. Phys.*, 2, 4421–4426, <https://doi.org/10.1039/b0043451>, 2000.
- Pruppacher, H. R. and Klett, J. D.: *Microphysics of Clouds and Precipitation*, Edn. 2, Kluwer Academic Publishers, Dordrecht, the Netherlands, 1997.
- Ramanathan, V. and Carmichael, G.: Global and regional climate changes due to black carbon, *Nat. Geosci.*, 1, 221–227, <https://doi.org/10.1038/ngeo156>, 2008.
- Schmidt-Ott, A., Baltensperger, U., Gaggeler, H. W., and Jost, D. T.: Scaling behaviour of physical parameters describing agglomerates, *J. Aerosol Sci.*, 21, 711–717, [https://doi.org/10.1016/0021-8502\(90\)90037-X](https://doi.org/10.1016/0021-8502(90)90037-X), 1990.
- Schneider, J., Höhler, K., Wagner, R., Saathoff, H., Schnaiter, M., Schorr, T., Steinke, I., Benz, S., Baumgartner, M., Rolf, C., Krämer, M., Leisner, T., and Möhler, O.: High homogeneous freezing onsets of sulfuric acid aerosol at cirrus temperatures, *Atmos. Chem. Phys.*, 21, 14403–14425, <https://doi.org/10.5194/acp-21-14403-2021>, 2021.
- Shin, Y. J., Wang, Y., Huang, H., Kalon, G., Wee, A. T., Shen, Z., Bhatia, C. S., and Yang, H.: Surface-energy engineering of graphene, *Langmuir*, 26, 3798–3802, <https://doi.org/10.1021/la100231u>, 2010.
- Shkol'nikov, E. I. and Sidorova, E. V.: Analytical equation for calculating the pore size distribution from adsorption data, *Dokl. Phys. Chem.*, 412, 4–7, <https://doi.org/10.1134/s0012501607010022>, 2007.
- Sing, K. S. W.: Assessment of Surface Area by Gas Adsorption, in: *Adsorption by Powders and Porous Solids (Second Edition)*, edited by: Rouquerol, F., Rouquerol, J., Sing, K. S. W., Llewellyn, P., and Maurin, G., Academic Press, 237–268, <https://doi.org/10.1016/b978-0-08-097035-6.00007-3>, 2014a.
- Sing, K. S. W.: Adsorption by Active Carbons, in: *Adsorption by Powders and Porous Solids*, edited by: Rouquerol, F., Rouquerol, J., Sing, K. S. W., Llewellyn, P., and Maurin, G., Academic Press, 321–391, <https://doi.org/10.1016/b978-0-08-097035-6.00010-3>, 2014b.
- Sing, K. S. W., Everett, D. H., Haul, R. A. W., Moscou, L., Pierotti, R. A., Rouquerol, J., and Siemieniowska, T.: Reporting Physisorption Data for Gas/Solid Systems with Special Reference to the Determination of Surface Area and Porosity, *Pure Appl. Chem.*, 57, 603–619, <https://doi.org/10.1351/pac198557040603>, 1985.
- Thommes, M. and Cychosz, K. A.: Physical adsorption characterization of nanoporous materials: progress and challenges, *Adsorption*, 20, 233–250, <https://doi.org/10.1007/s10450-014-9606-z>, 2014.
- Thommes, M., Cychosz, K. A., and Neimark, A. V.: Advanced Physical Adsorption Characterization of Nanoporous Carbons, in: *Novel Carbon Adsorbents*, edited by: Tascón, J. M. D., Elsevier, 107–145, <https://doi.org/10.1016/b978-0-08-097744-7.00004-1>, 2012.
- Thommes, M., Kaneko, K., Neimark, A. V., Olivier, J. P., Rodriguez-Reinoso, F., Rouquerol, J., and Sing, K. S. W.: Physisorption of gases, with special reference to the evaluation of surface area and pore size distribution (IUPAC Technical Report), *Pure Appl. Chem.*, 87, 1051–1069, <https://doi.org/10.1515/pac-2014-1117>, 2015.
- Twohy, C. H. and Gandrud, B. W.: Electron microscope analysis of residual particles from aircraft contrails, *Geophys. Res. Lett.*, 25, 1359–1362, <https://doi.org/10.1029/97gl03162>, 1998.
- Vali, G., DeMott, P. J., Möhler, O., and Whale, T. F.: Technical Note: A proposal for ice nucleation terminology, *Atmos. Chem. Phys.*, 15, 10263–10270, <https://doi.org/10.5194/acp-15-10263-2015>, 2015.
- Welti, A., Lüönd, F., Stetzer, O., and Lohmann, U.: Influence of particle size on the ice nucleating ability of mineral dusts, *Atmos. Chem. Phys.*, 9, 6705–6715, <https://doi.org/10.5194/acp-9-6705-2009>, 2009.
- Welti, A., Lüönd, F., Kanji, Z. A., Stetzer, O., and Lohmann, U.: Time dependence of immersion freezing: an experimental study on size selected kaolinite particles, *Atmos. Chem. Phys.*, 12, 9893–9907, <https://doi.org/10.5194/acp-12-9893-2012>, 2012.
- Wey, C. C., Anderson, B. E., Wey, C., Miake-Lye, R. C., Whitefield, P., and Howard, R.: Overview on the Aircraft Particle Emissions Experiment (APEX), *J. Propul. Power*, 23, 898–905, <https://doi.org/10.2514/1.26406>, 2007.
- Wheeler, A.: Reaction rates and selectivity in catalyst pores, in: *Advances in Catalysis*, edited by: Frankenburg, W. G., Komarewsky, V. I., Rideal, E. K., Emmett, P. H., and Taylor, H. S., Academic Press, 249–327, [https://doi.org/10.1016/S0360-0564\(08\)60109-1](https://doi.org/10.1016/S0360-0564(08)60109-1), 1951.
- Xue, X., Hui, X., Vannorsdall, P., Singh, P., and Sung, C.-J.: The blending effect on the sooting tendencies of alternative/conventional jet fuel blends in non-premixed flames, *Fuel*, 237, 648–657, <https://doi.org/10.1016/j.fuel.2018.09.157>, 2019.
- Zhang, C., Zhang, Y., Wolf, M. J., Nichman, L., Shen, C., Onasch, T. B., Chen, L., and Cziczo, D. J.: The effects of morphology, mobility size, and secondary organic aerosol (SOA) material coating on the ice nucleation activity of black carbon in the cirrus regime, *Atmos. Chem. Phys.*, 20, 13957–13984, <https://doi.org/10.5194/acp-20-13957-2020>, 2020.
- Zhang, R., Khalizov, A. F., Pagels, J., Zhang, D., Xue, H., and McMurry, P. H.: Variability in morphology, hygroscopicity, and optical properties of soot aerosols during atmospheric processing, *P. Natl. Acad. Sci.*, 105, 10291–10296, 2008.