

## Technical note: Measurement of chemically resolved volume equivalent diameter and effective density of particles by AAC-SPAMS

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Abstract. Size and effective density ( $\rho_e$ ) are important properties of aerosol particles and are related to their influences on human health and the global climate. The volume equivalent diameter  $(D_{ve})$  is an intrinsic property that is used to evaluate particle size. Three definitions of  $\rho_e$  are generally used to characterize the physical property of a particle as an alternative to particle density, in which only the  $\rho_{\rm e}^{\rm II}$ , defined as the ratio of particle density  $(\rho_p)$  to a dynamic shape factor  $(\chi)$ , has the characteristic of being independent of particle size. However, it is still challenging to simultaneously characterize the  $D_{ve}$  and  $\rho_e^{II}$  of aspherical particles. Here, we present a novel system that classifies particles with their aerodynamic diameter  $(D_a)$  by aerodynamic aerosol classifier (AAC) and determines their vacuum aerodynamic diameter  $(D_{va})$  by single-particle aerosol mass spectrometry (SPAMS) to achieve a measurement of  $D_{ve}$  and  $\rho_e^{II}$ . The reliability of the AAC-SPAMS system for accurately obtaining  $D_{ve}$  and  $\rho_{\rm e}^{\rm II}$  is verified based on the result that the deviation between the measured and theoretical values is less than 6 % for the size-resolved spherical polystyrene latex (PSL). The AAC-SPAMS system was applied to characterize the  $D_{ve}$  and  $\rho_{e}^{II}$ of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> and NaNO<sub>3</sub> particles, suggesting that these particles are aspherical and their  $\rho_{e}^{II}$  is independent of particle size. Finally, the AAC-SPAMS system was deployed in a field measurement, showing that it is a powerful technique to characterize the chemically resolved  $D_{ve}$  and  $\rho_{e}^{II}$  of particles in real time.

## 1 Introduction

Size and particle density  $(\rho_p)$  are critical parameters of aerosol particles in quantifying the impact of aerosols on air quality, human health and global climate change (Buseck and Posfai, 1999; Pöschl, 2005; Pitz et al., 2003). Effective density ( $\rho_e$ ) has been adopted to characterize the physical property of a particle as an alternative to  $\rho_p$ , since  $\rho_p$  for aspherical aerosol particles is hardly measured (Sumlin et al., 2018; Katrib et al., 2005). Size and  $\rho_e$  govern the transport properties of a particle both in the atmosphere and in the human respiratory system (Seinfeld and Pandis, 1998; Liu and Daum, 2008) and directly and/or indirectly influence the potential of the particle to absorb or reflect solar radiation (Tang, 1997; Zhao et al., 2019; Liu and Daum, 2008).  $\rho_e$  can also provide information concerning particle morphology (Yon et al., 2015) and serve as a tracer for atmospheric processing (Guo et al., 2014; Yin et al., 2015; Liu et al., 2015). However, the quantitative relationship between aerosol properties, namely size and  $\rho_{e}$ , and their effects on air quality, human health, and global climate change are not yet well understood, which is partly because important aerosol properties cannot be measured by current techniques.

#### 1.1 Size

Size is a fundamental property of particles and can be parameterized by the physical quantity of volume equivalent diameter ( $D_{ve}$ ). Defined as the diameter of a spherical particle with the same volume as the particle (DeCarlo et al., 2004),  $D_{ve}$  is an intrinsic physical quantity that can be used to evaluate the actual size of the particle. However, to date, atmospheric science usually describes particle size by other diameter definitions, such as the electric mobility diameter ( $D_m$ ), aerodynamic equivalent diameter ( $D_a$ ), and vacuum aerodynamic equivalent diameter ( $D_{va}$ ), whose relationships with  $D_{ve}$  are shown in Eqs. (1)–(3), respectively:

$$\frac{D_{\rm m}}{C_{\rm c}(D_{\rm m})} = \frac{D_{\rm ve}}{C_{\rm c}(D_{\rm ve})}\chi_{\rm t},\tag{1}$$

$$D_{\rm a} = D_{\rm ve} \sqrt{\frac{\rho_{\rm p} C_{\rm c} \left( D_{\rm ve} \right)}{\chi_{\rm t} \cdot \rho_0 \cdot C_{\rm c} \left( D_{\rm a} \right)}},\tag{2}$$

$$D_{\rm va} = \frac{\rho_{\rm p}}{\rho_0} \frac{D_{\rm ve}}{\chi_{\rm v}},\tag{3}$$

where  $C_c(D)$  is the Cunningham slip correction factor,  $\chi_t$ and  $\chi_v$  represent the aerosol dynamic shape factor ( $\chi$ ) in the transition regime and in the free-molecule regime, respectively, and  $\rho_0$  represents the unit density of 1.0 g/cm<sup>3</sup>. From these definitions, it can be seen that  $D_m$ ,  $D_a$ , and  $D_{va}$  are originally derived from  $D_{ve}$ , but in actuality they do not reflect the actual size of the aspherical particle. Meanwhile,  $D_{ve}$  of aspherical particles cannot be easily obtained, which limits its application in the scientific community.

### **1.2 Effective density**

At present, three definitions of  $\rho_e$  are introduced in atmospheric science (DeCarlo et al., 2004): the first definition ( $\rho_e^{I}$ ) is the ratio of the measured particle mass ( $m_p$ ) to the particle volume (V) calculated assuming a spherical particle with a diameter equal to the measured  $D_m$ , the second definition ( $\rho_e^{II}$ ) is the ratio of  $\rho_p$  to  $\chi$  (Hand and Kreidenweis, 2002), and the third definition ( $\rho_e^{II}$ ) is the ratio of  $D_m$  and  $D_{va}$ , expressed in Eqs. (4)–(6), respectively.

$$\rho_{\rm e}^{\rm I} = \frac{6m_{\rm p}}{\pi D_{\rm m}^3},\tag{4}$$

$$\rho_{\rm e}^{\rm II} = \frac{\rho_{\rm p}}{\chi},\tag{5}$$

$$\rho_{\rm e}^{\rm III} = \frac{D_{\rm va}}{D_{\rm m}} \,\rho_0. \tag{6}$$

The definitions of  $\rho_e^{I}$  and  $\rho_e^{III}$  can be derived into their final forms, as shown in the Eqs. (7) and (8), respectively.

$$\rho_{\rm e}^{\rm I} = \frac{\rho}{\chi_{\rm f}^3} \cdot \left(\frac{C_{\rm c}\left(D_{\rm ve}\right)}{C_{\rm c}\left(D_{\rm m}\right)}\right)^3 \tag{7}$$

$$\rho_{\rm e}^{\rm III} = \rho \cdot \frac{C_{\rm c} \left( D_{\rm ve} \right)}{\chi^2 \cdot C_{\rm c} \left( D_{\rm m} \right)} \tag{8}$$

Equation (7) is derived from combining Eq. (1) with Eq. (4), in which  $m_p$  is equal to  $1/6\rho \cdot D_{ve}^3$ . The detailed derivation of Eq. (8) is presented in Schneider et al. (2006). A variety of methods are developed to characterize  $\rho_{\rm e}^{\rm I}$ and  $\rho_{\rm e}^{\rm III}$ , among which the more advanced methods are to achieve the measurement of the chemically resolved effective density. Combining a single-particle soot photometer (SP2) with a (volatility) tandem differential mobility analyzer ((VT)DMA) can measure the  $\rho_e^{I}$  of particles mixed with soot (Y. Zhang et al., 2016; Wu et al., 2019; Han et al., 2019). The measurement of chemically resolved  $\rho_{\rm e}^{\rm III}$  can be achieved by coupling a DMA with an online aerosol mass spectrometer such as the Single-Particle Laser Ablation Time-of-Flight Mass Spectrometer (SPLAT) (Zelenyuk et al., 2005, 2006; Alexander et al., 2016), aerosol mass spectrometer (AMS) (Dinar et al., 2006; Schneider et al., 2006; Kiselev et al., 2010), aerosol time-of-flight mass spectrometer (ATOFMS) (Spencer and Prather, 2006; Spencer et al., 2007), and single-particle aerosol mass spectrometry (SPAMS) (G. Zhang et al., 2016; Zhai et al., 2017). However, the  $\rho_{e}^{I}$  and  $\rho_{e}^{III}$  are demonstrated to have the inherent characteristic of decreasing with increasing particle size, which will be presented in a separate publication. Therefore, it will introduce systemic error when assessing the particle's impact on visibility, human health, and climate change from the physical quantities in  $\rho_e^{I}$  and  $\rho_e^{III}$ . In contrast,  $\rho_e^{II}$  is independent of particle size. For example, for soot particles with  $\chi$  of 2.5 and  $\rho_p$  of 1.80 g/cm<sup>3</sup>, the calculated  $\rho_e^{I}$ ,  $\rho_e^{II}$ , and  $\rho_e^{III}$  are 0.43, 0.72, and 0.45 g/cm<sup>3</sup> at  $D_{\rm m}$  of 40 nm and 0.22, 0.72, and 0.36 g/cm<sup>3</sup> at  $D_{\rm m}$  of 550 nm, respectively. The big gap between the three definitions of effective density suggests that they should be carefully treated when characterizing the particles. However, the  $\rho_{e}^{II}$  has not been widely applied in atmospheric sciences because of the lack of measurement techniques. Previous literature has tried to retrieve the  $\rho_e^{\text{II}}$  and the real part in the refractive index (n) through a fitting procedure that compares the measured light-scattering intensity of particles  $(R_{\text{meas}})$  to the theoretical values  $(R_{\text{theory,test}})$  calculated by a series of *n* and  $\rho_{e}^{II}$  values (Moffet and Prather, 2005; Moffet et al., 2008; G. Zhang et al., 2016). Moffet and Prather (2005) successfully obtained  $\rho_e^{II}$  for spherical particles via single-particle mass spectrometry. However, subject to the accuracy of Mie theory for the aspherical particles, dry NaCl and calcium-rich dust particles failed to fit the  $R_{\text{theory,test}}$  well to  $R_{\text{meas}}$  (Moffet et al., 2008). Similarly, G. Zhang et al. (2016) failed to simultaneously retrieve  $\rho_e^{II}$ and n for  $(NH_4)_2SO_4$  and  $NaNO_3$  particles. To the best of our knowledge, there is no appropriate technique to achieve the measurement of  $\rho_{\rm e}^{\rm II}$  for aspherical particles.

The aim of the present work is to develop a method to simultaneously obtain  $D_{ve}$  and  $\rho_e^{II}$  for aspherical particles. For simplicity, the symbol  $\rho_e$  in the following text refers to the definition of  $\rho_e^{II}$ . The established system of aerodynamic aerosol classifier single-particle aerosol mass spectrometry (AAC-SPAMS) is capable of characterizing the  $D_a$  and  $D_{va}$ 



**Figure 1.** Schematic diagram of the AAC-SPAMS system (0.3 L/min). The diffusion drying tube is filled with orange silica gel, which reduces the RH to 5%-15%.

of particles, which can be applied to theoretically derive  $D_{ve}$ and  $\rho_e$ . To verify the reliability of the AAC-SPAMS system, we apply it to measure the  $D_{ve}$  and  $\rho_e$  of the spherical particles of polystyrene latex (PSL). The results are in good agreement with the theoretical values. Finally, the AAC-SPAMS system is applied to measure the  $D_{ve}$  and  $\rho_e$  for (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> and NaNO<sub>3</sub> particles and for the chemically resolved atmospheric particles.

### 2 Experimental section

### 2.1 Measurement system

Figure 1 shows a schematic diagram of the AAC-SPAMS system. The particles are first dried by a diffusion drying tube (TSI 9302, USA), classified by AAC (Cambustion Ltd., UK) based on the  $D_{\rm a}$ , and then transported into SPAMS, where the  $D_{\rm va}$  and the mass spectra of individual particles are obtained. The working principle of the AAC is described in detail elsewhere (Tavakoli and Olfert, 2013). AAC consists of two coaxial cylinders that rotate at the same rotational speed. Polydisperse particles enter into the space between the cylinders (i.e., the classification column) and experience a centrifugal force that causes them to move toward the outer cylinder. The particles to be classified can leave the classification column with the particle-free sheath flow and finally exit the AAC with the sample flow. Thus, the  $D_a$  values of classified particles can be derived from their relationship with their relaxation time  $(\tau)$ , as shown in Eq. (9):

$$\tau = \frac{C_{\rm c}\left(D_{\rm a}\right) \cdot \rho_0 \cdot D_{\rm a}^2}{18\mu},\tag{9}$$

where  $\mu$  is the gas dynamic viscosity. Particles with large relaxation times impact and adhere to the outer cylinder, while particles with small relaxation times exit the classifier with the exhaust flow. In this study, the exhaust flow from the AAC was about 0.3 L/min, and the size resolution parameter of the AAC was set as 40.

Detailed information about the operation of SPAMS (Hexin Analytical Instrument Co., Ltd., China) is given elsewhere (Li et al., 2011). Briefly, the particles are introduced into the vacuum system through a 0.1 mm critical orifice and are gradually collimated into a beam in the aerodynamic lens. Two continuous diode Nd:YAG laser beams (532 nm) are used to aerodynamically size the particles, which are subsequently desorbed and ionized by a pulsed laser (266 nm) that is triggered based on the velocity of a specific particle. The generated positive and negative ions are recorded with the corresponding particle size. The  $D_{va}$  of the particle is related to the transit time between the two laser beams (532 nm) in SPAMS, which can be obtained by using a calibration curve generated from the measured transit times of a PSL series with predefined sizes (nominal diameters).

#### 2.2 Laboratory experiments

Dried spherical PSL (Nanosphere Size Standards, Duke Scientific Corp., Palo Alto) ( $\rho_p = 1.055 \text{ g/cm}^3$  and  $\chi = 1.0$ ) with  $D_{ve}$  values of  $203 \pm 5$ ,  $310 \pm 6$ ,  $510 \pm 5$ , and  $740 \pm 6$  nm were used in the AAC-SPAMS system, and the  $D_{ve}$  was verified by Scanning Mobility Particle Sizer (Model 3938, TSI Inc., USA). The PSL particles were first classified by AAC, and then their  $D_{va}$  values were obtained by SPAMS. ACC-SPAMS was also applied to the particles of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>  $(\rho_{\rm p} = 1.77 \text{ g/cm}^3)$  and NaNO<sub>3</sub>  $(\rho_{\rm p} = 2.26 \text{ g/cm}^3)$  with  $D_{\rm a}$ values of 250.0, 350.0, 450.0, and 550.0 nm. Besides, to obtain the measurement uncertainty of the AAC, the  $D_a$  values of these PSL particles were measured to be  $212.8 \pm 0.2$ ,  $324.7 \pm 0.4$ ,  $529.9 \pm 0.4$ , and  $767.5 \pm 0.4$  by the system of AAC condensation particle counter (CPC), respectively. It shows that the AAC has the deviations of 1.1 %, 1.3 %, 0.8 %, and 0.7 % for determining the  $D_a$  values of the particles.

### 2.3 Ambient sampling

For field observations, the AAC-SPAMS system was deployed in Science and Technology Enterprise Accelerator A2 Block, Guangzhou, China, to characterize the  $D_{ve}$ ,  $\rho_e$ , and chemical compositions of aerosol particles. The sampling inlet was hung 2.5 m from the third floor ( $\sim$  12 m above ground level). Ambient aerosol particles were introduced into the AAC through a 5 m long conductive silicone tube with an inner diameter of 6 mm and a PM2.5 cyclone inlet. The sampling flow from the PM<sub>2.5</sub> cyclone inlet was 3 L/min, and the residence time in the conductive silicone tube was approximately 5 s. Particles with the  $D_a$  of 250.0, 350.0, 450.0, and 550.0 nm were classified by the AAC. The sampling time for the particles of each  $D_a$  was approximately 10 min. From 6– 8 July 2019, approximately 129 869 ionized particles were obtained from nine rounds of measurement. The sampling details are shown in Table S1 in the Supplement. The number of ionized particles with the  $D_a$  of 250.0, 350.0, 450.0, and 550.0 nm is 35 609, 38 374, 31 910, and 23 976, respectively. The sampled  $\sim 100\,000$  particles are first classified by using an adaptive resonance theory neural network (ART-2a)

(Song et al., 1999) with a vigilance factor of 0.75, a learning rate of 0.05, and 20 iterations.

# 2.4 Theoretical derivation of $D_{ve}$ and $\rho_e$ from $D_a$ and $D_{va}$

In this study, the calculations of  $D_{ve}$  and  $\rho_e$  for unknown particles are theoretically derived from  $D_a$  and  $D_{va}$ . Combining Eqs. (2) and (3), we obtain Eq. (10):

$$C_{\rm c}(D_{\rm a})\frac{D_{\rm a}^2}{D_{\rm va}} = D_{\rm ve}C_{\rm c}(D_{\rm ve})\frac{\chi_{\rm v}}{\chi_{\rm t}}.$$
(10)

Based on the approximation between  $\chi_v$  and  $\chi_t$  ( $\chi_v \approx \chi_t = \chi_a$ ) (DeCarlo et al., 2004), Eq. (10) becomes Eq. (11):

$$C_{\rm c}(D_{\rm a})\frac{D_{\rm a}^2}{D_{\rm va}} = D_{\rm ve}C_{\rm c}(D_{\rm ve}).$$
 (11)

The Cunningham slip correction factor is calculated by Eq. (12) (Peng and Bi, 2020):

$$C_{\rm c}(D) = 1 + \frac{\lambda}{D} \left( A + B \cdot \exp\left(\frac{C \cdot D}{\lambda}\right) \right),\tag{12}$$

where  $\lambda$  is the mean free path of the gas molecules and *A*, *B*, and *C* are empirically determined constants specific to the analysis system. The values of *A*, *B*, and *C* are 2.33, 0.966, and -0.498 provided by the manual of the AAC, respectively. Substituting Eq. (12) into Eq. (11) obtains Eq. (13).

$$\frac{D_{a}^{2}}{D_{va}} + \frac{D_{a} \cdot \lambda}{D_{va}} \left( A + B \cdot \exp\left(\frac{C \cdot D_{a}}{\lambda}\right) \right)$$
$$= D_{ve} + \lambda \left( A + B \cdot \exp\left(\frac{C \cdot D_{ve}}{\lambda}\right) \right)$$
(13)

If the  $D_a$  and  $D_{va}$  of an unknown particle can be measured, its  $D_{ve}$  could be calculated according to Eq. (13). Finally, the  $\rho_e$  value of the particles is calculated by the  $D_{va}$  and  $D_{ve}$ values according to Eq. (14), which is obtained by combining Eqs. (3) and (5):

$$\rho_{\rm e} = \frac{\rho_{\rm p}}{\chi_{\rm a}} = \frac{D_{\rm va}}{\rho_0 \cdot D_{\rm ve}}.$$
(14)

Thus, we can obtain both the  $D_{ve}$  and  $\rho_e$  values of unknown particles based on the  $D_a$  and  $D_{va}$  values. Because the AAC and SPAMS instruments have the ability to determine  $D_a$  and  $D_{va}$ , the AAC-SPAMS system developed in this study can be used to obtain the  $D_{ve}$  and  $\rho_e$  values for unknown particles.

### 3 Results and discussion

# 3.1 Verification of the AAC-SPAMS system to obtain $D_{ve}$ and $\rho_e$

The  $D_{va}$  distribution of PSL particles with predefined  $D_{ve}$  values after being screened by the AAC is shown in Fig. S1

in the Supplement. Gaussian fitting is used to obtain the peak  $D_{\rm va}$  for each size PSL with  $R^2$  fitting coefficient over 0.98. Each fitting has a full width at half maximum (FWHM) of 6.6%, 4.4%, 2.3%, and 2.2%, and the corresponding peaks are 215.8, 319.0, 532.1, and 803.5 nm, respectively. Substituting the  $D_a$  and  $D_{va}$  values of PSL into Eq. (11), the measured  $D_{ve}(D_{ve,me})$  of PSL from the AAC-SPAMS system is 203.6, 309.7, 511.6, and 737.2 nm, respectively (Fig. 2a). Thus, the deviations between the theoretical  $D_{ve}$  ( $D_{ve,th}$ ) and  $D_{\rm ve,me}$  values are 0.3%, -0.1%, 0.3%, and -0.4%, respectively. On the other hand, the measured  $\rho_e(\rho_{e,me})$  values of the particles, calculated from the  $D_{va}$  and  $D_{ve.me}$  values with Eq. (14), are 1.1, 1.0, 1.0, and 1.1 g/cm<sup>3</sup>, respectively (Fig. 2b). Comparing to the theoretical  $\rho_e(\rho_{e,th})$  (i.e., 1.055 g/cm<sup>3</sup> of PSL particles), the deviations of  $\rho_{e,me}$  are determined to be 4.3 %, -5.2 %, -5.2 %, and 4.3 %, respectively. That is, the deviations of  $D_{ve,me}$  and  $\rho_{e,me}$  obtained by the AAC-SPAMS system are within 1 % and 6 %, respectively.

## 3.2 Application of the AAC-SPAMS system for obtaining D<sub>ve</sub> and ρ<sub>e</sub> of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> and NaNO<sub>3</sub>

Figure S2 shows the  $D_{va}$  distributions of  $(NH_4)_2SO_4$  and NaNO<sub>3</sub> particles with D<sub>a</sub> values of 250.0, 350.0, 450.0, and 550.0 nm screened by the AAC. The  $D_{ya}$  peaks are obtained by Gaussian fitting, with  $R^2$  values over 0.93 and FWHM values ranging from 7.6% to 10.6%. The (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> particles have Dva values of 300.0, 418.0, 551.1, and 695.1 nm (Fig. S2), which correspond to particles possessing  $D_{\text{ve.me}}$ values of 177.3, 254.4, 331.8, and 409.3 nm, respectively, according to Eq. (11). Substituting the values of  $D_{va}$  and  $D_{ve,me}$ into Eq. (12), the  $\rho_{e,me}$  values are 1.7, 1.6, 1.6, and 1.7 g/cm<sup>3</sup> (Fig. 3a), respectively. Similarly, the selected NaNO<sub>3</sub> particles are determined to have  $D_{va}$  values of 321.0, 454.9, 599.8, and 755.3 nm (Fig. S2), corresponding to D<sub>ve.me</sub> values of 150.1, 218.2, 287.0, and 355.9 nm, respectively. The  $\rho_{e,me}$  values of the NaNO<sub>3</sub> particles are 2.2, 2.0, 2.0, and 2.1 g/cm<sup>3</sup> (Fig. 3b), respectively. Figure 3 also shows that the  $\rho_{e,me}$  values of the NaNO<sub>3</sub> and (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> particles with different  $D_a$  deviate from their average values with the maximum of 5.9% and 4.8%, respectively, which are identical with the deviation for the  $\rho_{e,me}$  of PSL particles. These deviations may be derived from the calibration of particle  $D_{va}$ from the SPAMS. While the  $R^2$  of size calibration curve is 0.999, the curve of exponential function is found to slightly deviate from the data points measured by SPAMS. For example, the size calibration function produces the deviation of -4.4% and 3.1% from the data points of 310 and 740 nm, respectively.

Taking the systematic error into account, the slight difference of the  $\rho_{e,me}$  values for the four sizes suggests that the  $\rho_e$  of  $(NH_4)_2SO_4$  and NaNO<sub>3</sub> particles is independent of particle size from 250.0 to 550.0 nm. It is determined by the definition of effective density used in this study, which



Figure 2. (a) Comparison between the measured  $D_{ve}(D_{ve,me})$  and the theoretical  $D_{ve}(D_{ve,th})$  of the PSL particles. (b) Comparison between the measured  $\rho_e$  ( $\rho_{e,me}$ ) and the theoretical  $\rho_e$  ( $\rho_{e,th}$ ) of the PSL particles.



**Figure 3.** (a) Comparison between the measured  $\rho_e$  ( $\rho_{e,me}$ ) and average  $\rho_e$  ( $\rho_{e,a}$ ) values of the (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> particles. (b) Comparison between the measured  $\rho_e$  ( $\rho_{e,me}$ ) and average  $\rho_e$  ( $\rho_{e,a}$ ) values of the NaNO<sub>3</sub> particles.

keeps constant as long as the  $\chi_a$  of the particles does not change with particle size for pure compound. The average  $\rho_{e,me}$  values of  $(NH_4)_2SO_4$  and NaNO<sub>3</sub> particles are calculated to be  $1.7 \pm 0.1$  and  $2.1 \pm 0.1$  g/cm<sup>3</sup>, which are lower than the  $\rho_p$  of  $(NH_4)_2SO_4$  (1.77 g/cm<sup>3</sup>) and NaNO<sub>3</sub> (2.27 g/cm<sup>3</sup>). This is partly caused by the  $\chi_a$ , which can be used to parameterize the morphology. According to Eq. (14), the  $\chi_a$  with different  $D_a$  are calculated to be 1.04, 1.11, 1.11, and 1.04 for  $(NH_4)_2SO_4$  particles and 1.03, 1.14, 1.14, and 1.08 for NaNO<sub>3</sub> particles. Thus, the average  $\chi_a$  values of the  $(NH_4)_2SO_4$  and NaNO<sub>3</sub> particles are determined to be  $1.07\pm0.04$  and  $1.10\pm0.05$ , respectively, indicating that these particles are aspherical.

The asphericity of  $(NH_4)_2SO_4$  determined by AAC-SPAMS system is consistent with the previous studies reporting that the  $\chi_a$  of  $(NH_4)_2SO_4$  were larger than the value of 1.03 (Zelenyuk et al., 2006; Beranek et al., 2012; G. Zhang et al., 2016). However, previous studies found that the NaNO<sub>3</sub> particles had different morphology. G. Zhang et al. (2016) observed that NaNO<sub>3</sub> had the  $\chi_a$  of 1.09–1.13, while Hoffman et al. (2004) found that NaNO<sub>3</sub> particle had a round droplet-like shape even at 15% RH, supported by the consistency between the measured value of "anhydrous" so-

lution droplet (Zelenyuk et al., 2005). Eclectically, Tang and Munkelwitz (1994) studied that most of the NaNO<sub>3</sub> particles crystallized between 20 % and 30 % RH but some persisted down to 10 % RH to keep solution droplets. Notably, the spherical NaNO<sub>3</sub> particles at low RH observed by Hoffman et al. (2004) were dried in the sticky carbon tape, which might affect the phase transition of droplet-like NaNO<sub>3</sub> particles. In this study, most NaNO<sub>3</sub> particles might crystallize because the RH of the aerosol flow carrying the NaNO<sub>3</sub> particles was reduced to below 20 % through the diffusion drying tube. The asphericity of the crystallized NaNO<sub>3</sub> particles is supported by their FWHM values of the  $D_{va}$  distributions, which are consistent with that of aspherical (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> (Figs. S1 and S2).

## **3.3** Application of the AAC-SPAMS system for measuring the chemically resolved *D*<sub>ve</sub> and *ρ*<sub>e</sub>

SPAMS can obtain information on the chemical composition of individual particles, implying that the AAC-SPAMS system has the ability to simultaneously characterize  $D_{ve}$ ,  $\rho_e$ , and the chemical compositions of particles in real time. It is worth noting that the freshly emitted soot particles exhibit the largest  $\chi$  (~ 2.5) in the actual atmosphere (Peng et al., 2016).

D <sub>a</sub> (nm)	K-rich	EC-S	K-Na	Amine
250.0	$193.1\pm8.2$	$192.2\pm8.1$	$193.8\pm21.9$	$190.6\pm4.6$
350.0	$284.0\pm28.4$	$280.8\pm9.3$	$271.9 \pm 18.0$	$284.8 \pm 18.2$
450.0	$364.7\pm21.1$	$357.8\pm6.9$	$342.5\pm7.3$	$367.9\pm9.7$
550.0	$416.6\pm28.3$	$439.5\pm5.4$	$397.3\pm29.7$	$442.5\pm7.4$
D <sub>a</sub> (nm)	EC-N-S	OC-N-S	OC-EC-N-S	Metal-rich
250.0	$188.5\pm5.9$	$200.8 \pm 17.9$	$195.4\pm8.9$	$189.0\pm6.7$
350.0	$281.3\pm9.3$	$295.7\pm29.8$	$294.0\pm32.3$	$277.0\pm9.1$
450.0	$358.0\pm5.8$	$398.3 \pm 44.3$	$428.9 \pm 24.0$	$342.9 \pm 10.0$
550.0	$453.2\pm16.4$	$547.4 \pm 14.7$	570.9	$407.4 \pm 14.5$

**Table 1.** The measured mean  $D_{ve}$  and its standard deviation for the eight particle types at  $D_a$  values of 250.0, 350.0, 450.0, and 550.0 nm from nine rounds of measurements.



Figure 4. Variation in  $\rho_e$  of the eight particle types with  $D_{ve}$ . The solid lines represent the range of the  $\rho_e$  and  $D_{ve}$  measured from nine rounds, and the data points stand for the average values.

It meets the upper limit for the approximation between the  $\chi_t$  and  $\chi_v$  (DeCarlo et al., 2004).

As an example, the AAC-SPAMS system was deployed in the field to obtain the chemically resolved  $D_{ve}$  and  $\rho_e$  values for unknown aerosol particles. The sampled ~ 100 000 particles are classified into eight major particle types with distinct chemical compositions: K-rich, EC-S, K-Na, amine, EC-N-S, OC-N-S and OC-EC-N-S, and Metal-rich, representing 97 % of the detected particle population. Details of the chemical composition and number fraction of the eight types of particles are presented in the Figs. S3 and S4, respectively, which are discussed in the Supplement.

We used Gaussian fitting to obtain the  $D_{va}$  peaks for each particle type with D<sub>a</sub> values of 250.0, 350.0, 450.0, and 550.0 nm. Following this, we calculated the  $D_{ve}$  values of the atmospheric particles with Eq. (11). Table 1 presents the average  $D_{ve}$  values of the eight particle types, for which the standard deviation is calculated based on nine samples. The average  $D_{ve}$  at  $D_a$  values of 250.0, 350.0, 450.0, and 550.0 nm shows wide ranges: from 188.5 to 200.8, 271.9 to 295.7, 342.5 to 428.9, and 397.3 to 570.9 nm, respectively, which are caused by the different chemical composition. The result indicates that particles with significantly different  $D_{ve}$ might possess the same  $D_a$ . Furthermore, the large standard deviation of Dve, such as 21.9 nm for K-Na at 250.0 nm, 32.3 nm for OC-EC-N-S at 350.0 nm, and 44.3 nm for OC-N-S at 450.0 nm, indicates that the  $D_{ve}$  of particles is remarkably different even for particles with the same type and same  $D_{a}$ .

According to  $D_{ve}$  and  $D_{va}$ , we calculated the  $\rho_e$  of each particle type by Eq. (12). Figure 4 shows the variations of the  $\rho_e$  with  $D_{ve}$  for nine particle samples. For pure compounds, such as  $(NH_4)_2SO_4$  and NaNO<sub>3</sub> particle,  $\rho_e$  theoretically does not change with particle size. However, the sampled particles have experienced complex atmospheric processes. Therefore,  $\rho_e$  has a very wide distribution for each type of particles with a similar  $D_{\rm ve}$ . Specifically, the  $\rho_{\rm e}$  of K-Na increases with  $D_{ve}$ , while the  $\rho_e$  of OC-N-S and OC-EC-N-S decreases with  $D_{ve}$ , which may be influenced by the particle shape or the material density. Additionally, the average  $\rho_e$  of each type of particle is in the order from small to large:  $1.2 \pm 0.2$  g/cm<sup>3</sup> for OC-EC-N-S,  $1.3 \pm 0.2$  g/cm<sup>3</sup> for OC-N-S,  $1.4 \pm 0.1$  g/cm<sup>3</sup> for K-rich,  $1.4 \pm 0.1$  g/cm<sup>3</sup> for amine,  $1.5 \pm 0.1$  g/cm<sup>3</sup> for EC-N-S,  $1.5 \pm 0.1$  g/cm<sup>3</sup> for EC-S,  $1.6 \pm 0.1$  g/cm<sup>3</sup> for K-Na, and  $1.6 \pm 0.1$  g/cm<sup>3</sup> for Metalrich. It is reasonable to find that the average  $\rho_e$  of internally mixed particles distributes in the range of their material densities ( $\rho_m$ ). For instance, the OC-EC-N-S, OC-N-S, K-rich, and amine particles, mainly comprised of internally mixed sulfate and organics, have the average  $\rho_e$  between that of sulfate with  $\rho_{\rm m}$  of 1.77 g/cm<sup>3</sup> and organic aerosols with  $\rho_{\rm m}$  of 1.2 g/cm<sup>3</sup> (Cross et al., 2007).

#### 4 Conclusion

We develop an AAC-SPAMS system to first achieve the measurement of the  $D_{ve}$  and  $\rho_e$  (defined as the ratio of  $\rho_p$  to  $\chi$ ) of the aspherical particles through characterizing their  $D_a$ and  $D_{va}$ . The reliability of the AAC-SPAMS system is verified by accurately measuring the  $D_{ve}$  and  $\rho_e$  of PSL. Applying the AAC-SPAMS system to determine the  $D_{ve}$  and  $\rho_e$  of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> and NaNO<sub>3</sub> particles shows that these particles are aspherical and their  $\rho_e$  are independent of particle size. Coupled with the ability of SPAMS to characterize the chemical composition of individual particles, the AAC-SPAMS system is demonstrated to be capable of characterizing the  $D_{\rm ve}, \, \rho_{\rm e}(\rho_{\rm p}/\chi)$  and chemical compositions of atmospheric particles simultaneously, showing the potential application of this system in field observations. The approach achieves the measurement of chemically resolved  $D_{\rm ve}$  and  $\rho_{\rm e}(\rho_{\rm p}/\chi)$ and provides the possibility to determine their quantitative relationship with other particle properties, which would be a benefit for further reduction of the uncertainty associated with the effects of particles on air quality, human health, and radiative forcing.

*Data availability.* Data in this study are available at https://doi.org/10.5281/zenodo.4655206 (Peng, 2021).

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*Author contributions.* The idea for the study was conceived by LP and GHZ. All experiments were performed by LP with the assistance of LL. LP wrote the paper, which was reviewed by GHZ and XHB. All co-authors discussed the results and commented on the manuscript.

*Competing interests.* The authors declare that they have no conflict of interest.

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