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Measurement report: Ice nucleating abilities of biomass burning, African dust, and sea spray aerosol particles over the Yucatán Peninsula

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Abstract. Most precipitation from deep clouds over the continents and in the intertropical convergence zone is strongly influenced by the presence of ice crystals whose formation requires the presence of ice nucleating particles (INPs). Although there are a large number of INP sources, the ice nucleating abilities of aerosol particles originating from oceans, deserts, and wildfires are poorly described at tropical latitudes. To fill this gap in knowledge, the National Autonomous University of Mexico micro-orifice uniform deposit impactor droplet freezing technique (UNAM-MOUDI-DFT) was constructed to measure the ice nucleating activity of aerosol samples that were collected in Sisal and Mérida, Yucatán (Mexico) under the influence of cold fronts, biomass burning (BB), and African dust (AD) intrusions during five short-term field campaigns between January 2017 and July 2018.

The three different aerosol types were distinguished by their physicochemical properties. Marine aerosol (MA), BB, and AD air masses were found to contain INPs; the highest concentrations were in AD (from 0.071 to $36.07 \, \mathrm{L^{-1}}$ at temperatures between -18 and $-27\,^{\circ}\mathrm{C}$), followed by MA (from 0.068 to $18.90 \, \mathrm{L^{-1}}$ at temperatures between -15 and $-28\,^{\circ}\mathrm{C}$) and BB (from 0.063 to $10.21 \, \mathrm{L^{-1}}$ at temperatures between -20 and $-27\,^{\circ}\mathrm{C}$). However, MA had the highest surface active site densities ($n_{\rm s}$) between -15 and $-30\,^{\circ}\mathrm{C}$. Additionally, supermicron particles contributed more than $72\,\%$ of the total INP concentration independent of aerosol type.

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1 Introduction

Approximately 60% of the Earth's surface is covered by clouds at any time; however, a lack of detailed knowledge about the aerosol-cloud interactions introduces large uncertainties in projections of climate change (Lohmann and Feichter, 2005; Boucher et al., 2013). Globally, the precipitation in the intertropical convergence zone (ITCZ) and over the continents is dominated by cold rain; therefore, it is very important to fully understand the formation of ice in clouds (Mülmenstädt et al., 2015). Primary ice particle formation takes place in the atmosphere via homogeneous and heterogeneous ice nucleation. Homogeneous ice nucleation typically occurs at temperatures below -38 °C and relative humidities with respect to ice (RH_i) above 140 % (Knopf et al., 2011; Kanji et al., 2017). In contrast, heterogeneous nucleation, in which ice nucleating particles (INPs) promote phase transition, occurs at temperatures warmer than -38 °C and lower RH; than those required for homogeneous nucleation (Kanji et al., 2017). Although there are several modes through which ice particles can form heterogeneously (i.e., deposition nucleation, pore-condensation freezing, immersion freezing, condensation freezing, and contact freezing), immersion freezing has been reported as the most important ice nucleation mode for mixed-phase clouds (Knopf et al., 2011; Murray et al., 2012).

The presence of INPs in clouds affects their microphysical properties, lifetime, precipitation formation, and the planetary radiative balance (DeMott et al., 2003, 2010; Boucher et al., 2013; Kanji et al., 2017). At temperatures higher than -38 °C, 1 in 10^6 (or fewer) atmospheric aerosol particles can act as INPs (DeMott et al., 2010; Lohmann et al., 2016). Although it remain partially unclear which physicochemical properties make an aerosol particle a good INP, its size (diameter $> 0.5 \,\mu\text{m}$), crystal structure (similar to the ice lattice structure), chemical composition (hygroscopicity and ability to form hydrogen bonds), insolubility (solid surface for the ice embryo formation), and the presence of active sites have been reported as key factors (Pruppacher and Klett, 1997; Murray et al., 2012; Kanji et al., 2017). As summarized by Kanji et al. (2017), aerosol particles, such as mineral and desert dust, crystalline salts, volcanic ash, organic and glassy particles, metallic particles, biomass and fossil combustion particles, marine aerosol (MA), and biological particles, have all been shown to act as INPs under varying thermodynamic conditions as a function of the ice nucleation mode. A short summary of the main results of the ice nucleating abilities found for mineral dust, biomass burning (BB), and MA particles in the immersion freezing mode is provided below. Note that a direct comparison of the INP concentrations reported in different studies needs to be considered carefully as there are factors associated with each technique or method, such as sampling volume, droplet size, aerosol loading, aerosol particle size range, and temperature range, among others, that could be influencing specific results.

Mineral dust is mainly composed of clays (e.g., kaolinite, montmorillonite, illite), feldspars, and quartz, among others (e.g., Zimmermann et al., 2008; Ardon-Dryer and Levin, 2014). The ice nucleating abilities of pure minerals, clays, and surrogates of natural mineral dust particles have been previously evaluated in the immersion freezing mode (e.g., Marcolli et al., 2007; Augustin-Bauditz et al., 2014; Lüönd et al., 2010; Murray et al., 2011; Wheeler et al., 2015; Harrison et al., 2019). These studies found that mineral dust particles will catalyze ice formation at temperatures as high as −15.2 °C. Boose et al. (2016) analyzed 15 ambient mineral dust samples, including some collected after they were transported from distant sources. The authors reported that the efficient ice nucleating abilities of those samples were related to the presence of K-feldspar, which is in agreement with earlier results reported by Atkinson et al. (2013). Boose et al. (2016) also show that airborne dust samples transported from distant sources had lower ice nucleation activity likely due to particle aging. Ardon-Dryer and Levin (2014) evaluated the ice nucleating abilities of mineral dust from the eastern Mediterranean region via immersion freezing and found that INP concentrations varied from 0.16 to 234 L⁻¹ at temperatures between -11.8 and -28.9 °C. Reicher et al. (2019) found similar INP concentrations (from 0.1 to $1000 L^{-1}$) in the same region, during dust storms, and at temperatures ranging from -18.2 to -38.2 °C. Price et al. (2018) reported INP concentrations from 0.1 to $100 L^{-1}$ at -12 and −23 °C under dusty conditions around Cape Verde, Africa, while Gong et al. (2020a) found INP concentrations between 0.0003 and $0.2 L^{-1}$ at temperatures between -5 and $-24 \,^{\circ}$ C under low dust emissions on one of the Cape Verde islands.

A total of 9 of the 21 different biomass fuels tested by Petters et al. (2009) produced INPs, with swamp sawgrass smoke identified as the most efficient fuel reporting an INP fraction of $\approx 1:100$ particles. Umo et al. (2015) evaluated the ice nucleating activity of coal fly ash (CFA), coal bottom ash, domestic bottom ash, and wood bottom ash. CFA was found to be the most efficient with a T_{50} (the temperature at which 50% of the droplets freeze) of around -16 °C. The authors identified particle morphology and chemical composition as the key physicochemical properties controlling the ice nucleating abilities of the evaluated BB particles. In controlled laboratory experiments, Levin et al. (2016) analyzed 22 different biomass fuels and found that the INP concentrations ranged between 10^2 and 10^4 per liter at -30 °C. They also observed increased INP concentrations for intense flaming combustion (i.e., > 0.95). Prenni et al. (2012) found that INP concentrations from wildfires and prescribed fires in Colorado and Wyoming (United States) ranged between 3.4 and $90 L^{-1}$ at $-30 \,^{\circ}$ C. Likewise, McCluskey et al. (2014) found that mean INP concentrations during prescribed burns varied from 6.36 to 16.7 L⁻¹ at −30 °C in Colorado and from 0.64 to $34.03 L^{-1}$ between -22 and $-30 \,^{\circ}$ C in Georgia. However, McCluskey et al. (2014) showed that INP concentrations were higher during wildfires in Colorado (i.e., Hewlett wildfire and High Park wildfire) with mean values of 15.90 to 79.43 L⁻¹ (between -22 and -30 °C) and 7.18 to 75.98 L⁻¹ (between -22 and -31 °C). It is important to note that aerosol particles emitted by BB contain smoke and ash (Kanji et al., 2017). In general, BB particles are enriched in organic compounds and to a lesser extent trace metals (Saarikoski et al., 2007). Jahn et al. (2020) showed that BB particles may contain calcite, sylvite, dolomite, potassium sulfate, and quartz. However, the mineral composition will differ depending on the combustion conditions, biomass source, and the chemical composition, among others.

Several field campaigns have been conducted to evaluate the ice nucleating abilities of aerosol particles in marine environments. It has been shown that the composition of the MA strongly depends on the particle size, with the highest organic content found in submicron particles (O'Dowd et al., 2004; Prather et al., 2013; Quinn et al., 2015). Bigg (1973) collected aerosol samples in Australia, Schnell and Vali (1975) in Huntington Beach (California, US), in the Caribbean off Nassau (Bahamas), and in Vancouver and Nova Scotia (Canada), Rosinski et al. (1987) over the Pacific Ocean, and Rosinski et al. (1988) in the Gulf of Mexico (GoM). More recently, the ice nucleating abilities of particles present in the Arctic sea surface microlayer (Wilson et al., 2015; Irish et al., 2017, 2019) and the Arctic ambient aerosol were evaluated (DeMott et al., 2016; Creamean et al., 2018; Wex et al., 2019). Similar studies were performed over the eastern Mediterranean by Gong et al. (2019), over the Southern Ocean by McCluskey et al. (2018) and Welti et al. (2020), over the North Atlantic by Wilbourn et al. (2020), and over the tropical Atlantic by DeMott et al. (2016), Welti et al. (2018), Ladino et al. (2019), Ladino et al. (2020), and Gong et al. (2020a). Welti et al. (2020) presented an analysis of the INP concentrations on ship-based measurements for different zones in the Arctic, Atlantic, Pacific, and Southern oceans. The different climatic zones covered in the Welti et al. (2020) study were divided depending on the latitude. For the tropical zone (23.5° N-23.5° S), the authors report that the concentrations of INPs vary between 0.001 and $0.1 L^{-1}$ at temperatures from -5 to -25 °C and between 0.1 and 10^5 L⁻¹ at lower temperatures, i.e., from -24 to -38 °C. It is important to note that in some of the aforementioned studies, onset freezing temperatures as high as -3 °C were reported which were associated with marine organic material likely of biological origin.

Aerosol particles can be transported over long distances far from their emission sources (e.g., Griffin et al., 2001; Taylor, 2002; Wu et al., 2004; Prenni et al., 2009). For example, DeMott et al. (2003), Prospero and Lamb (2003), Prospero and Mayol-Bracero (2013), and Ramírez-Romero et al. (2021) showed how mineral dust particles from the Saharan desert could reach the Caribbean region, the United States, and Mexico. The highest probability for mineral dust to reach the Caribbean and Mexico occurs during the mid-summer drought (MSD), a period of a relative minimum in precip-

itation within the rainy season typically between mid-July and mid-August. Similarly, Peppler et al. (2000) and Saide et al. (2015) showed how BB particles from Central America and southern Mexico could impact the Yucatán Peninsula, Mexico. As shown by Ríos and Raga (2018), there is a clear BB seasonality in Mexico and Central America from December to June with the largest burned areas between April and May. Finally, the arrival of air masses to southern Mexico behind cold fronts has also been documented (DiMego et al., 1976; Cavazos, 1997), which are characterized by low temperatures, dry conditions, and strong winds flowing over the GoM. These bring marine aerosol particles into the Yucatán Peninsula, as shown by Ladino et al. (2019). During longrange travel, the physicochemical properties (e.g., size, composition, morphology, hygroscopicity, etc.) of the transported aerosol particles can be modified due to atmospheric aging that likely affects their ice nucleating abilities (e.g., Kanji et al., 2017).

The majority of the field studies related to INPs have been carried out at middle and high latitudes with a limited number of studies in the tropics. As the tropical regions have significantly different characteristics than higher latitudes (e.g., flora, fauna, marine biological activity, sea surface temperature, air temperature, relative humidity, RH, cloud cover, atmospheric circulation), there is a compelling need to improve the current understanding of the role of aerosol particles emitted at tropical latitudes as they impact cloud formation (Yakobi-Hancock et al., 2014). Given that Mexico is annually impacted by aerosol particles transported from Africa, North America, and Central America, the present study evaluates the ice nucleating abilities of these three types of aerosol particles.

2 Methodology

2.1 Sampling location

Five short-term field campaigns were carried out between January 2017 and July 2018, three in Mérida (20.98° N, 89.64° W) and two in Sisal (21.16° N, 90.04° W), as part of the African Dust and Biomass Burning Over Yucatán (AD-ABBOY) project. Both sites are located in the Yucatán Peninsula (Fig. 1) in southeastern Mexico. The Yucatán Peninsula is surrounded by the GoM in the north and west, the Caribbean Sea in the east, and Central America in the south. Most of the peninsula (i.e., 86%) experiences a warm, tropical wet climate, while the remaining 14% has a dry and semi-dry climate. The annual average temperature and RH are 26°C and 79%, respectively (INEGI, 2019). Its soil is karstic, in which the principal components are limestone, dolomite, and gypsum (Herrera-Silveira et al., 2004b).

Mérida, the capital of Yucatán State located about 35 km inland from the coast of the GoM, has a population of 892 363 (INEGI, 2015). The main industries are tourism,

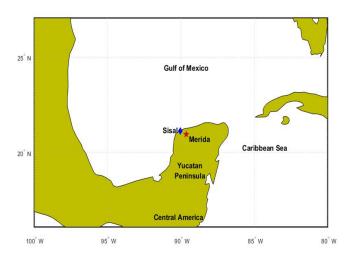


Figure 1. Map showing the Yucatán Peninsula and the sampling locations, i.e., Sisal (blue diamond) and Mérida (red star).

manufacturing, textiles, and commerce (García de Fuentes and Morales, 2000). The city has an average temperature of 25 °C and an annual accumulated precipitation of about 900 mm (INEGI, 2017). The sampling site in Mérida was located on the rooftop of the school of chemistry of the Universidad Autónoma de Yucatán (SC-UADY), located in the central-western part of the city, 3 km from downtown.

Sisal is located 49 km from Mérida and has 1837 inhabitants (INEGI, 2010). It is a small coastal village, most people work in fishing-related activities, and there are no nearby industrial emissions (Santoyo, 2017). The mean temperature and RH are 27 °C and 80.8 %, respectively. Sisal is affected yearly by cold fronts between November and January and by convective storms and precipitation from the end of April to October (Santoyo, 2017). The aerosol sampling took place on the rooftop of the engineering institute of the Universidad Nacional Autónoma de México (EI-UNAM), which is 50 m from the shoreline. The dates of the sampling periods for Mérida and Sisal are shown in Table 1. A subset of the samples (those with the full size range available, i.e., from 0.32 to 10 µm) collected during the five campaigns was chosen for this study, as summarized in Table S1.

The presence of BB particles and the intrusions of AD into the Yucatán Peninsula have been previously documented (e.g., Yokelson et al., 2009; Kishcha et al., 2014; Ríos and Raga, 2018; Raga et al., 2021; Trujano-Jiménez et al., 2021; Ramírez-Romero et al., 2021). Ríos and Raga (2018) reported that within the BB season, the maximum fire density is observed in April. Likewise, Kishcha et al. (2014), Raga et al. (2021), and Ramírez-Romero et al. (2021) indicate that July is the period with the highest likelihood for the AD influx to the western Caribbean within the MSD. Therefore, April and July were chosen as the sampling periods to capture the presence of BB and AD particles in Mérida, respectively.

Table 1. Summary of the sampling periods, the instrumentation used, and the type of collected samples (for chemical composition and INP analysis) from Mérida and Sisal during 2017 and 2018. MA, BB, and AD refer to marine aerosol, biomass burning, and African dust, respectively. The ^a, ^b, and ^c denote the samples collected for chemical analysis, samples collected for INP analysis, and samples not collected during this period, respectively.

Aerosol type	Place	2017	2018
MA	Sisal	21 Jan–2 Feb LasAir II 310A MOUDI 100NR ^a MOUDI 100R ^b	С
ВВ	Mérida	13 Apr–31 May LasAir II 310A FH62C14 Partisol 2025i ^a MOUDI 100R ^b	26 Mar–8 Apr LasAir II 310A FH62C14 Partisol 2025i ^a MOUDI 100R ^b
AD	Sisal	С	3 Jul–16 Jul LasAir II 310A MiniVol ^a MOUDI 100R ^b
	Mérida	С	3 Jul–16 Jul LasAir III 310C MiniVol ^a MOUDI 100R ^b

Given that the Yucatán Peninsula is encircled by the GoM, MA is ubiquitous throughout the Peninsula. Therefore, the MA composition was assessed in the remote coastal village of Sisal between January and February, a time of the year when the presence of particles such as BB and AD are least likely.

2.2 Instrumentation

2.2.1 Ice nucleating particles

The evaluation of the ice nucleating abilities of the ambient aerosol particles consists of three steps: (i) collection of the particles, (ii) activation with the droplet freezing technique (DFT), and (iii) calculation of the INP concentration as a function of temperature and particle size.

Collection of aerosol particles

Aerosol particles were collected by inertial impaction on hydrophobic glass coverslips of $22 \text{ mm} \times 22 \text{ mm}$ (HR3-215, Hampton Research) using a micro-orifice uniform deposit impactor (MOUDI 100R; MSP), as shown in Fig. 2a. To secure the glass coverslips on the micro-orifice uniform deposit impactor (MOUDI) stages, substrate holders were used as reported by Mason et al. (2015a). The MOUDI has eight stages to separate particles as a function of their aerodynamic diameter (cut sizes of 0.18, 0.32, 0.56, 1.0, 1.8, 3.2,

5.6, and 10.0 µm). The flow rate was set to 30.0 L min⁻¹ for the five campaigns, except for Sisal 2018, with a flow rate of 25.8 L min⁻¹. The sampling period was typically 6 h from 08:00 to 14:00 (local time; all times are in central standard time); however, on some specific days, more than one sample per day was collected. After each sampling, the glass coverslips were stored in petri dishes for between 3 and 22 months at 4 °C prior to their analysis with the DFT in Mexico City (Fig. 2a). As recently highlighted by Beall et al. (2020), the temperature and the length of the storage can impact the ice nucleation abilities of MA samples. Although, this was not evaluated in the present study, future studies will evaluate how the storage procedure impacts results.

The UNAM-DFT

The DFT has been employed in various studies to evaluate the ability of aerosol particles to act as INPs via immersion freezing (e.g., Koop et al., 2000; Iannone et al., 2011; Mason et al., 2015a, b; Wheeler et al., 2015; Si et al., 2018; Irish et al., 2019). A DFT was built in the Micro- and Mesoscale Interactions Laboratory of the Atmospheric Science Center at UNAM based on the design described in Mason et al. (2015a). The results delivered by the Mason et al. (2015a) MOUDI-DFT were compared against those reported by the Colorado State University Continuous Flow Diffusion Chamber (CSU-CFDC). As shown in DeMott et al. (2017), the median INP concentrations from both devices were in good agreement.

The newly built DFT consists of four components: (i) cold stage, (ii) humid/dry air system, (iii) optical microscope with video recording system, and (iv) data acquisition system (Fig. 3).

The cold stage consists of a sample holder, a cooling block, and a heating block all aligned and stacked on top of each other with the sample holder at the top and the cooling block at the bottom (Fig. S1a). The cooling block is made of stainless steel $(4 \text{ cm} \times 4 \text{ cm} \times 2 \text{ cm})$, in which a cooling fluid continuously circulates to maintain a constant temperature. A copper heating block $(4 \text{ cm} \times 4 \text{ cm} \times 1 \text{ cm})$ is located above the cooling block. Two resistance heaters (100 W and 120 V) are placed in the middle of the heating block to increase its temperature to control the temperature of the sample holder. The temperature of the heating block is regulated by a temperature controller via a thermocouple. A sample holder $(4 \text{ cm} \times 2 \text{ cm} \times 1 \text{ cm})$ made of stainless steel is placed on top of the heating block. To ensure good thermal contact between the sample holder, the heating block, and the cold block, the three are attached by screws on each corner. The sample holder is divided in two parts (i.e., top and bottom; Fig. S1c) with a Teflon spacer between them. Aerosol samples collected on the glass coverslips (Fig. 2a) are placed between the two sections of the sample holder. Afterwards, these sections are fixed in place with four screws. As shown in Fig. S1b, a circular glass window is placed on the top of the sample holder to avoid any interference of the ambient air. An optical microscope (Axiolab Zeiss, Germany) with a $5 \times /0.12$ magnification objective is coupled with the sample holder (Fig. 3).

To maintain the temperature of the cooling block at ca. -80 °C, polydimethylsiloxane circulates through it with the help of a pump inside the cooling bath (PRO-RP1090, LAUDA). A humid/dry air system is required to form liquid droplets on the aerosol particles deposited on the glass coverslips. While humid air is obtained when the nitrogen (grade 4.8, INFRA) passes through a bubbler filled with pure water (LAL reagent water, Associates of Cape Cod, Inc.), dry conditions are experienced when nitrogen is injected and flows towards the sample. The air is directed by stainless steel tubes (Swagelok), and the flows are controlled by four Swagelok valves (Fig. 3). Each experiment is recorded by a video camera (MC500-W, JVLAB) attached to the microscope, and the video images are used for further analysis. The temperature at the center of the sample holder was obtained with a resistance temperature detector (RTD) with a precision of 0.1 °C and an uncertainty of 0.1 to 0.56 °C between 0 and -40 °C, respectively. The RTD is connected to a FieldLogger device (RS485, NOVUS) to acquire the data (Fig. 3).

The aim of the UNAM-DFT is to mimic ice particle formation via immersion freezing; hence, a full experiment is performed as follows. At room temperature, the coverslip containing the aerosol particles is placed on the cold stage. The sample holder is isolated from the ambient atmosphere by tightening the screws and positioning the circular glass window on top of the heating block when the temperature is at 0 °C (Fig. 2b). Humid air is directed towards the sample holder to induce liquid droplet formation by water vapor condensation, as shown in Fig. 2c. Once the droplets have reached a diameter of 170 µm (on average), dry air is directed toward the sample holder to shrink the droplets and minimize the contact between them (Fig. 2d). Approximately 30-40 droplets are formed on each glass coverslip. Afterwards, the humid/dry air system is closed, and the sample holder is completely isolated by closing the valves. Finally, the temperature of the sample holder is decreased from 0 to -40 °C at a cooling rate of 10 °C min⁻¹. During the temperature ramp, the droplets freeze, promoted by the aerosol particles immersed within the droplets (Fig. 2e); otherwise, they are expected to freeze homogeneously close to -38 °C.

Frozen fraction

The temperatures from the sample holder and the images in the videos are analyzed to determine the freezing temperature of each droplet. The first metric employed was the frozen fraction (F_{ice}):

$$F_{\rm ice} = \frac{N_{\rm ice}}{N_{\rm ice} + N_{\rm droplets}},\tag{1}$$

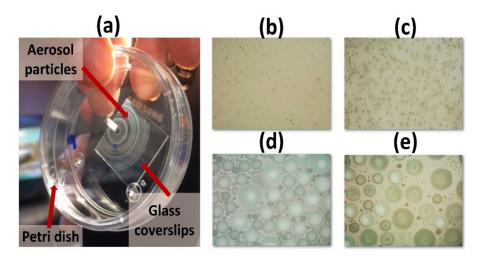


Figure 2. (a) Aerosol particles collected on a hydrophobic glass coverslip, (b) aerosol particles seen from the microscope ($d = 5.6-10 \,\mu\text{m}$), (c) water vapor condensation on the aerosol particles, (d) liquid droplets, and (e) frozen (opaque colors) and unfrozen droplets as seen by the optical microscope.

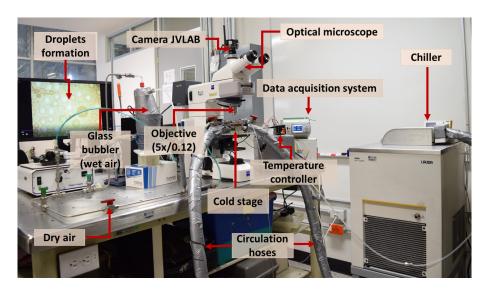


Figure 3. UNAM-DFT experimental setup with its main components. Figure S1 shows a zoom of the cold stage.

where N_{ice} is the number of frozen drops (dimensionless) and N_{droplets} is the number of unfrozen droplets (dimensionless) (Kanji et al., 2017).

INP concentration

The INP number concentration (L^{-1}) was calculated using the following expression (Mason et al., 2015a):

$$[INP(T)] = -\ln\left(\frac{N_u(T)}{N_o}\right) \cdot \left(\frac{A_{\text{deposit}}}{A_{\text{DFT}}V}\right) \cdot N_o \cdot f_{\text{ne}}$$

$$\cdot f_{\text{nu},0.25-0.10 \,\text{mm}} \cdot f_{\text{nu},1 \,\text{mm}}, \qquad (2)$$

where $N_{\rm u}(T)$ is the number of unfrozen droplets at a temperature $T(^{\circ}{\rm C})$, N_0 is the total number of droplets (dimen-

sionless), $A_{\rm deposit}$ is the total area of the aerosol particles deposited on the MOUDI hydrophobic glass cover slip (cm²), $A_{\rm DFT}$ is the area of the sample analyzed by the DFT (cm²), V is the volume of air through the MOUDI (L), $f_{\rm ne}$ is a correction factor to account for the uncertainty associated with the number of nucleation events in each experiment (dimensionless), and $f_{\rm nu}$ is a correction factor to account for changes in particle concentration across each MOUDI sample (dimensionless).

Equation (2) considers the possibility that multiple particles can be present within a droplet (Vali, 1971). As part of the data analysis, two corrections are included. The first is related to the total area covered by the particles deposited on the coverslips, which varies between 425 mm² and 605 mm² as a function of the MOUDI stage. The second is related to

inhomogeneities in particle deposit. The correction factor associated with the nucleation events (f_{ne}) was calculated with a 95% confidence level (Mason et al., 2015a). More details on these corrections can be found in Mason et al. (2015a).

2.2.2 Surface active site density (n_s)

The n_s values were derived at -15, -20, -25, and -30 °C using Eq. (3):

$$n_{\rm s}(T) = \frac{[{\rm INP}(T)]}{S_{\rm tot}}, \tag{3}$$

where [INP(T)] is the INP concentration at temperature T and S_{tot} is the total surface area of all aerosol particles. See the Supplement for more details on how S_{tot} was calculated.

2.2.3 Aerosol number and mass concentration

The particle size distributions for equivalent optical diameters (EODs) between 0.3 and 25 μ m were obtained with optical particle counters, a PMS LasAir II 310A (0.3 to 25 μ m) and PMS LasAir III 310C (0.3 to 10 μ m), as listed in Table 1, operated at a flow rate of 28.3 L min⁻¹ and at a sampling rate of 11 Hz. The particle mass concentrations, i.e., PM₁₀ and PM_{2.5}, were obtained with a FH62C14 Thermo Scientific Inc., operated at a flow rate of 16.7 L min⁻¹.

2.2.4 Chemical composition

For the chemical analysis, aerosol particles were collected on 47 mm Teflon filters (Pall Science) for 48 h in Mérida in 2017 and 2018 and in Sisal in 2017. A 24 h sample was also made in Sisal in 2018. Note that there were two different sampling periods in Sisal to evaluate the interannual variability in aerosol loading. The aerosol particles were collected using a MiniVol (TAS, Airmetrics), a Partisol (2025i, Thermo Fisher Scientific), and a cascade impactor (MOUDI 100NR, MSP) (see Table 1).

The Teflon filters were analyzed by X-ray fluorescence (XRF) to determine the elemental composition with an X-ray spectrometer at the Laboratorio de Aerosoles, Instituto de Física, UNAM (Espinosa et al., 2012). The spectrometer uses an Oxford Instruments (Scotts Valley, CA, USA) X-ray tube with an Rh anode and an Amptek X-123SDD spectrometer (Bedford, MA, USA) operated at 50 kV and a current of 500 μA, irradiating during 900 s per spectrum. A set of thin film standards (MicroMatter Co., Vancouver, Canada) was used for the calibration of the instrument. The spectra were analyzed using the AXIL (QXAS) software. The chemical composition was quantified using the methodology reported by Espinosa et al. (2010). Once the concentration of each analyzed element was obtained, the fraction (in percentage) with respect to the total mass concentration was derived.

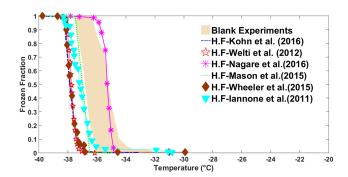


Figure 4. Activation curve for the blank experiments (rose shaded area) obtained with the UNAM-DFT using pure water. Colored lines correspond to homogeneous freezing (H.F.) curve literature data. Dashed blue line: Kohn et al. (2016); red stars: Welti et al. (2012); magenta asterisks: Nagare et al. (2016); dotted green line: Mason et al. (2015a); brown diamonds: Wheeler et al. (2015); and cyan triangles: Iannone et al. (2011).

2.2.5 Meteorological variables

Wind speed, wind direction, air temperature, and RH were obtained using meteorological stations. During Sisal 2017, a Davis (Vantage Pro2) meteorological station located 20 m away from the other instruments was used. Meteorological data during the four other sampling periods were obtained from meteorological sensors from the University Network of Atmospheric Observatories (RUOA), which were installed at SC-UADY and EI-UNAM. Additionally, back trajectories of the air masses arriving in Mérida and Sisal were estimated with the Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) model from the National Oceanic and Atmospheric Administration (NOAA) over 24 h and 13 d. The 24 h back trajectories were run for the BB events as they were likely locally emitted. On the other hand, the back trajectories for the cold fronts (MA) and African dust intrusions (AD) were run for 13 d as the particles carried by those events take several days to reach Mexico.

3 Results and discussion

3.1 UNAM-DFT performance

The sensitivity response of the UNAM-DFT was evaluated using blank experiments with pure water (LAL Reagent Water, Associates of Cape Cod, Inc.). In this case, new glass coverslips were placed on the sample holder, and droplets ($d=126\,\mu\mathrm{m}$ on average) were formed on them by the condensation of water vapor on the substrate surface. The freezing temperature of each droplet was later determined.

Figure 4 shows that most of the droplets (i.e., 95%) froze homogeneously, i.e., with no particles in them and at temperatures below $-36\,^{\circ}\text{C}$ (rose shaded area). The freezing temperatures of the blank experiments are in agreement with

previous homogeneous freezing values reported for liquid droplets using different methods (Welti et al., 2012; Kohn et al., 2016; Nagare et al., 2016). This indicates that the substrate does not impact the heterogeneous freezing results discussed below. The small differences between our results and published homogeneous freezing curves can be attributed to differences in the droplet's size, the purity of the water, the exposure time of the droplets to low temperatures, and the freezing technique. For example, Kohn et al. (2016) used the Portable Immersion Mode Cooling chAmber (PIMCA) coupled with the Portable Ice Nucleation Chamber (PINC) with droplet sizes ranging from 10 to 14 µm in diameter (dashed blue line). Welti et al. (2012) (red stars) and Nagare et al. (2016) (magenta asterisks) used the Immersion Mode Cooling chAmber – Zurich Ice Nucleation Chamber (IMCA-ZINC) with $\approx 20 \,\mu m$ droplets in diameter. Note that droplet sizes obtained with the UNAM-DFT are much larger than those used in the aforementioned studies. Additionally, the homogeneous freezing experiments in the UNAM-DFT take around 4 min in contrast with the residence time of droplets in the PIMCA-PINC and the IMCA-ZINC which are less than 10 s. Iannone et al. (2011) and Wheeler et al. (2015) also performed homogeneous freezing experiments using a DFT. Both studies found that 90 % of the droplets froze at -37 °C (brown diamonds and cyan triangles, respectively). Although the total times of the experiments between the present study and those performed by Iannone et al. (2011) and Wheeler et al. (2015) are comparable, the size of the droplets may be slightly different.

Additionally, as shown by Lacher et al. (2021), INP concentrations reported by the UNAM-DFT during the Puy de Dôme Ice Nucleating Intercomparison Campaign (PICNIC) are in good agreement with the values reported by other online and offline techniques, such as the Portable Ice Nucleation Experiment (PINE) instrument, the Ice Nucleation Spectrometer of the Karlsruhe Institute of Technology (IN-SEKT), the FRankfurt Ice Deposition freezinG Experiment (FRIDGE), the Colorado State University Ice Spectrometer (IS), the Ice Nucleation Droplet Array (INDA), the Leipzig Ice Nucleation Array (LINA), and the LED-based Ice Nucleation Detection Apparatus (LINDA). The results in Lacher et al. (2021) indicate that the UNAM-MOUDI-DFT is very robust with a high level of confidence.

3.2 Chemical composition vs. air mass

To corroborate if the aerosol particles collected in January, April, and July in the Yucatán Peninsula were influenced by MA, BB, and AD air masses, respectively, the elemental chemical composition obtained by XRF was evaluated, as shown in Fig. 5. The pie charts show the composition of particles under background conditions for Mérida (a) and Sisal (b). For Mérida, the background corresponds to those particles collected on days with PM₁₀ levels below the 10th percentile in the absence of BB and AD. For Sisal, the back-

ground corresponds to those particles collected in January in the absence of BB and AD. Although Mérida and Sisal are rather close (ca. 49 km), the elemental composition in both sites differs with the former likely influenced by urban emissions and the later by MA. Na and Cl are ubiquitous in Mérida and Sisal due to their proximity to the GoM. Since karstic sediments form the soil in Yucatán, the presence of Ca is expected as well.

Given that Sisal is a remote coastal Village with almost no industrial activities, the background aerosol composition measured in January can be considered as representative of the MA composition for the Yucatán Peninsula as the presence of BB and AD particles is very low. Figure 5b indicates that chlorine, sodium, sulfur, calcium, potassium, and magnesium were the most abundant elements influenced by MA. The presence of these elements agrees with results reported in studies focused on MA (Parungo et al., 1986; Xia and Gao, 2010; Prather et al., 2013). Prather et al. (2013) classified MA particles in four categories: inorganic sea salts (SSs) mainly composed of NaCl, inorganic sea salt plus organic carbon (SS-OC), biological particles, and organic particles (OCs) in which organic species can be coupled with ions such as sulfur, magnesium, and calcium. O'Dowd et al. (2004), Ovadnevaite et al. (2014), and Lee et al. (2015) reported that the fraction of SSs in the marine aerosol was 74 %, 14 %–85 %, and 72 %, respectively. Furthermore, the high percentage of sulfur is associated with dimethyl sulfide (DMS) emissions related to the marine biological activity linked to phytoplankton blooms (Yoch, 2002; Barnes et al., 2006). Herrera-Silveira et al. (2004a) characterized the phytoplankton in the GoM between 1999 and 2002, finding that diatoms, dinoflagellates, and chlorophytes are the most abundant types of phytoplankton on the coast of Sisal. Therefore, these microorganisms could contribute to the sulfur emissions in this zone.

Besides Na and Cl, sulfur is the most abundant element observed in the background aerosol particles in Mérida (Fig. 5a). The presence of sulfur can be attributed to anthropogenic sources, such as light- and heavy-duty vehicles with diesel engines that are prevalent in Mérida. Given the proximity of Mérida to the GoM, part of the sulfur measured in Mérida could originate from DMS. To assess the influence of the likely BB and AD particles collected in April and July in Mérida, respectively, the enrichment factor (against the background composition) was calculated for each of the 16 analyzed elements (Fig. 5c). Sulfur, potassium, aluminum, nickel, and iron were found to have an enrichment factor above 1.0 under the influence of BB (red bars). Higher concentrations of S and K were found in Mérida during the BB season, in agreement with Reid et al. (2005) who reported that particles emitted during BB could contain trace inorganic species (e.g., sulfates, nitrate, chlorine, calcium, and potassium). Potassium, a key element emitted during BB, is an important nutrient for the plants and is absorbed in the woody material through their roots (Jones et al., 2007).

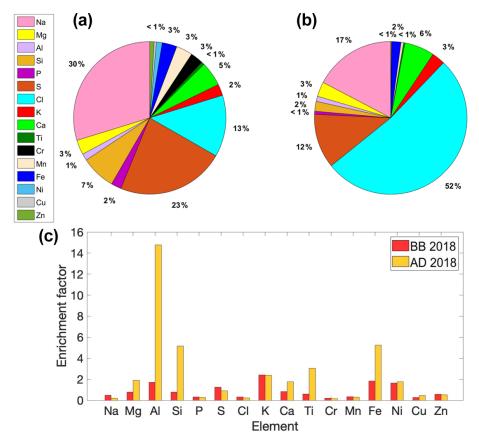


Figure 5. Percentage of elemental mass concentrations for background samples in Mérida (a) and Sisal (b). Enrichment factors are calculated for samples obtained under the influence of BB (red bars) and AD (yellow bars) plumes from background conditions in Mérida (c).

Moreover, the mass concentration of $PM_{2.5}$ positively correlated with K (> 0.6) (Fig. S2a), corroborating that the sampled air masses during this season contained a comparatively high mass fraction of particles emitted from BB when $PM_{2.5}$ was high. Note that Li et al. (2003) found that soot aggregates emitted by BB may contain potassium salts (K_2SO_4 , KNO₃, and KCl) and, in minor quantity, elements such as iron, calcium, and magnesium.

Figure 5c shows the significant enrichment of elements such as silicon, aluminum, and iron, which are characteristic of mineral dust particles, during the AD period (yellow bars). Additionally, positive correlation coefficients (≥ 0.8) between the mass concentrations of PM_{2.5} and Mg, Al, Si, K, Ti, and Fe were found (Fig. S2b). Rosinski et al. (1988) reported high percentages of these elements in aerosol particles collected during July and August in 1986 over the GoM. The authors attributed this composition to air masses originating in the east (probably from Africa). The elements Al, Si, Ca, and Fe are typically found in mineral dust particles (Linke et al., 2006; Querol et al., 2019).

Figure S2c shows the total organic carbon concentration (TOC) for the BB and AD periods, providing further evidence that the two air masses were completely different, with

higher TOC concentrations (by $2.0\,\mu g\,m^{-3}$) observed during the BB period.

Overall, important differences in elemental composition are evident in the aerosol particles collected in January, April, and July in the Yucatán Peninsula, corroborating the presence of MA, BB, and AD particles as suggested by previous studies (e.g., Yokelson et al., 2009; Kishcha et al., 2014; Ríos and Raga, 2018; Ladino et al., 2019; Raga et al., 2021; Trujano-Jiménez et al., 2021; Ramírez-Romero et al., 2021).

3.2.1 Aerosol size distribution vs. air mass

Figure 6 shows the average aerosol size distributions sampled under the influence of the different air masses: MA (cyan), BB (red), and AD (yellow). For all three air masses, the highest particle concentration (reported by the LasAir) was observed for particles with EODs between 0.3 and 0.5 μm . For particles larger than 0.5 μm , the concentration decreases with size. Out of the three air masses, AD reported the highest particle concentration for particles ranging from 0.3 to 5.0 μm . Likewise, the highest particle concentration for particles between 5.0 and 25 μm was found in the MA air masses. Except for the smallest size bin (i.e., 0.3–0.5 μm), aerosol particles

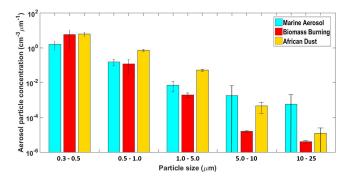


Figure 6. Average aerosol size distribution derived from the optical particle counters for the three air masses: MA (cyan bars), AD (yellow bars), and BB (red bars). The error bars represent the standard deviation.

measured during the BB season showed the lowest concentration in comparison to the other two air masses.

The size distribution for MA measured in Sisal is comparable to those reported by Si et al. (2018) for three different marine sites with lower concentrations (North Pacific, North Atlantic, and Arctic oceans). Generally, the coast of Sisal is very calm with moderate wave activity; however, during the MA sampling period, two cold fronts impacted the Yucatán Peninsula, increasing the wind speed up to 28 km h⁻¹. High wind speeds generate breaking waves that produce sea spray particles by bubble bursting. Figure S3a shows examples of back trajectories obtained with HYSPLIT during the presence of one of the cold fronts, indicating that air masses crossed the GoM prior to their arrival in Sisal.

The AD particle number concentration measured in Sisal was found to be lower compared with data published from other studies. For example, Gong et al. (2020b) reported that the aerosol concentrations measured on São Vicente island, Cape Verde, under the influence of AD for particles larger than 1.0 µm varied between 3 and 71 cm⁻³. Kaaden et al. (2009) found that under the influence of Saharan dust particles in Morocco, the mean aerosol concentration for particles larger than 1.0 µm was 70 cm⁻³. The lower particle number concentration measured in the Yucatán Peninsula is attributed to the long distance the AD particles traveled before arriving at Mexico. While crossing the Atlantic, many particles can be removed from the atmosphere by dry deposition and, to a lesser extent, by wet deposition as cloud formation is low during this season.

Although coarse particles were measured during the BB season at a much lower concentration, those particles may likely be associated with background particles such as resuspended dust. Field studies of BB emissions have mostly focused on submicron particles (e.g., Pósfai et al., 2003; Zhang et al., 2011). Guyon et al. (2005) found that during BB events in the Amazonia, the maximum submicron particle concentration was around $100 \, \text{cm}^{-3}$ in fresh and detrained smoke.

3.3 INP concentration vs. air mass

The INP concentration as a function of temperature and air mass type for particles between 0.32 and $10\,\mu m$ is shown in Fig. 7a. Samples obtained under the influence of MA produced particles that froze at the warmest temperatures of the three different air masses with onset freezing temperatures as high as -5 °C. The AD particles were able to activate ice particle formation at temperatures below -9 °C. BB particles were the least efficient with onset freezing temperatures colder than -19 °C.

The INP concentrations for MA (cyan circles) were found to range from 0.068 to $18.90 L^{-1}$ at temperatures between -5 and -31 °C. The present results were found to be higher than those reported by Mason et al. (2016) for the west coast of Canada when the air masses were coming from the northwest (i.e., 0.10 to $6.1 L^{-1}$ at temperatures from -15 to -25 °C), by Si et al. (2018) and Irish et al. (2019) in the Arctic (0.004 to $0.67 L^{-1}$ at temperature ranging from -15 to -25 °C), by Mason et al. (2016) in Alert (0.05 to 0.99 L⁻¹ at temperatures from -15 to -25 °C), and by DeMott et al. (2016) in Puerto Rico (0.0002 to $0.02 L^{-1}$ at temperatures from -6 to -24 °C). However, the present results are similar to those reported by Mason et al. (2016) on the east coast of Canada for southeastern air masses $(0.38 \text{ to } 2.8 \text{ L}^{-1} \text{ at tem-}$ peratures from -15 to -25 °C). Recall that the MA samples in Sisal were collected during the winter and in the presence of cold fronts. Therefore, the high wind speeds associated with the cold fronts, and hence wave breaking, could have played an important role in the high INP concentrations measured in Sisal, as discussed in Ladino et al. (2019). Additionally, given that air temperatures, RH, and sea surface temperatures are considerably different between the tropics and higher latitudes, these differences could be partially responsible for the latitudinal differences observed in INP concentrations.

The aerosol particles collected during the AD period produced the highest INP concentrations with values ranging between 0.071 and $36.07L^{-1}$ at temperatures from -9 to −29 °C (yellow squares). As shown in Fig. 5, the AD particles were enriched in Al, Fe, and Si, which are present in different minerals and clays (Querol et al., 2019). Although 2018 was a more active AD season than 2017, as characterized by higher PM_{2.5} and PM₁₀ concentrations (Ramírez-Romero et al., 2021), the INP concentrations measured in 2018 in the Yucatán Peninsula were lower than values reported elsewhere. For example, Ardon-Dryer and Levin (2014) found that the INP concentrations in the eastern Mediterranean ranged between 0.16 and 234 L⁻¹ at temperatures between -11.8 and -28.9 °C, and larger values (up to $10^3 \, \mathrm{L}^{-1}$) at temperatures from -18 to $-29 \, ^{\circ}\mathrm{C}$ were reported by Reicher et al. (2019). Price et al. (2018) found that in Cape Verde, the INP concentrations were an order of magnitude higher (0.1 and $10^2\,L^{-1}$ at temperatures from about -10to -25 °C) than those measured in Sisal and Mérida. Note

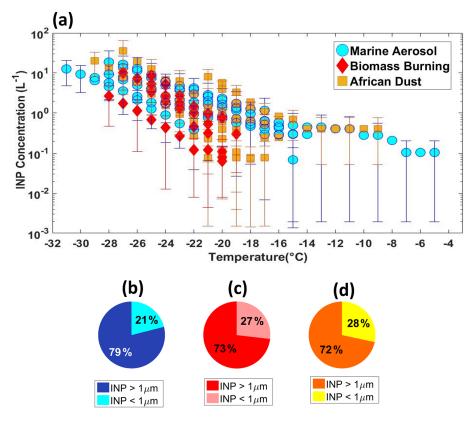


Figure 7. (a) INP concentrations as a function of temperature for MA (cyan circles), AD (yellow squares), and BB (red diamonds) for aerosol particles with sizes ranging between 0.32 and 10 μm. The pie charts illustrate the contribution of supermicron and submicron particles to the total INP concentrations for (b) MA, (c) BB, and (d) AD.

that the measurements reported by Ardon-Dryer and Levin (2014), Price et al. (2018), and Reicher et al. (2019) were performed relatively close to the mineral dust source, i.e., the Sahara and Arabian deserts. The lower values measured in the Yucatán Peninsula during the AD period can be associated with the long distance (> 8000 km) that the mineral dust particles traveled before reaching Mexico. During the long-range transport, the physicochemical properties of the AD particles may experience chemical aging, likely impacting their ice nucleating abilities, which is consistent with the results of Boose et al. (2016). Also, those particles may have been exposed to different dilution and removal processes during long-range transport, likely reducing the concentration of INPs.

The measured INP concentration during the BB period was found to range between 0.063 and $10.21\,\mathrm{L^{-1}}$ at temperatures from -19 to $-28\,^{\circ}\mathrm{C}$ (red diamonds), the lowest INP concentrations, on average, of the three aerosol types. BB particles have been reported as inefficient INPs via immersion freezing with a higher potential to catalyze ice particles at temperatures below $-40\,^{\circ}\mathrm{C}$ via deposition nucleation (Kanji et al., 2017). The concentrations of INPs measured in Mérida during the BB periods are lower than those reported by Prenni et al. (2012), McCluskey et al. (2014), and

Levin et al. (2016). The lower concentrations reported in the present study can be attributed to the long distance between the burning areas (likely southern Mexico and Central America) and the sampling site. Additionally, it has been shown that the type of fuel or biomass can determine the ice nucleating abilities of the BB emitted particles. McCluskey et al. (2014) found clear differences in the ice nucleating abilities of different prescribed burns and wildfires.

Given that the UNAM-MOUDI-DFT can differentiate between the INP concentrations as a function of the aerosol aerodynamic size, the contributions of the submicron and supermicron particles to the total INP concentrations were assessed for the three aerosol types. Figure 7b to d shows that supermicron particles are the major contributor (> 72 %) to the INP concentrations for all aerosol types. Note that this contribution is largest for the MA. These results are in agreement with previous studies (e.g., Mason et al., 2015b, 2016; Ladino et al., 2019; Gong et al., 2020a). The current results and the aforementioned studies highlight the importance of supermicron particles in ice cloud formation.

3.4 Surface active site density

Figure 8 summarizes the n_s values calculated for the three periods, i.e., MA, BB, and AD. As shown in Fig. S4, the n_s

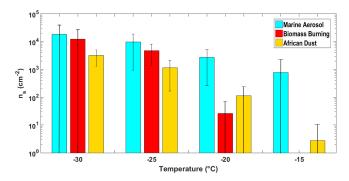


Figure 8. Surface active site density (n_s) as a function of temperature for MA, BB, and AD particles. The error bars represent the standard deviation.

values reported in the literature for MA, BB, and AD cover several orders of magnitude. The $n_{\rm s}$ obtained for the MA and BB particles overlap with the published range of $n_{\rm s}$ values. The MA and BB $n_{\rm s}$ values were found to agree with those reported by DeMott et al. (2016) and Umo et al. (2015), respectively. At -15 and $-20\,^{\circ}$ C, the $n_{\rm s}$ values for the AD period were found to agree with the values reported by Gong et al. (2020a). However, in comparison to the literature, lower $n_{\rm s}$ values were found for the AD period at temperatures below $-20\,^{\circ}$ C. Note that some of the $n_{\rm s}$ data in the comparison refer to pure components or aerosol types, while our analysis includes the entire atmospheric aerosol variability, i.e., also the surface of particles not acting as ice nucleating particles.

Although the AD particles had the highest INP concentrations between -9 and -29 °C (Fig. 7), Figure 8 shows that the MA had the highest $n_{\rm s}$ values in the Yucatán Peninsula (at least in winter), followed by AD at temperatures above -25 °C. On the other hand, BB particles had lower $n_{\rm s}$ values at -20 and -15 °C but comparable values to the other air masses at -25 and -30 °C.

Given that AD particles traveled > $8000 \,\mathrm{km}$ before reaching the Yucatán Peninsula, the low n_{S} values calculated for the AD particles are not unexpected. It is likely that the most efficient INPs could have been washed out while traveling over the Atlantic. Also, it is well known that aerosol aging can strongly affect the ice nucleating abilities of mineral dust particles (Kanji et al., 2017). Therefore, the composition of the AD particles that arrived at the Yucatán Peninsula may significantly differ from the AD particles found closer to their source with different ice nucleation efficiencies.

Large n_s values can be likely associated with the presence of biological particles, as was the case in Si et al. (2018) which linked the high n_s values with the presence of terrestrial biological particles. However, McCluskey et al. (2018) and DeMott et al. (2016) showed that these kind of particles can also be of marine origin as there is large biological activity in marine environments. Rodriguez-Gomez et al. (2020) showed different terrestrial and marine microorganisms that

were identified in Sisal which can be linked with the high n_s values found for the MA samples.

4 Conclusions

The UNAM-MOUDI-DFT apparatus was successfully built at the Universidad Nacional Autónoma de México. This new device can be used to study ice formation via immersion freezing and can discriminate the INP concentrations as a function of the aerosol aerodynamic size from 0.32 µm to 10 µm. The performance of the UNAM-MOUDI-DFT was assessed by conducting homogeneous freezing experiments with pure water and evaluated against other online and offline cloud chambers during the Puy de Dôme Ice Nucleating Intercomparison Campaign (PICNIC), which is discussed in a separate publication. The overall performance of the new device was found to be very good, providing high-quality results. The UNAM-MOUDI-DFT is expected to help the scientific community understand the main sources of INPs at tropical latitudes and to provide information urgently needed for the development of new ice nucleation parameterizations that include aerosol particles emitted in the tropics.

Aerosol particles originating from three different sources (marine aerosol, biomass burning, and African dust) were collected between January 2017 and July 2018 in Sisal and Mérida, Yucatán (Mexico). The three different aerosol types were characterized by their physicochemical properties (i.e., elemental composition, aerosol concentration, and particle size). Furthermore, the samples were analyzed with the newly developed UNAM-MOUDI-DFT device, allowing for the evaluation of the potential impact of different aerosol particles on mixed-phase cloud formation in the Yucatán Peninsula. This is the first such comprehensive study ever conducted in Mexico and among the first ones at tropical latitudes.

The three different air masses were found to contain INPs with different ice nucleating efficiencies. The cold fronts, responsible for advecting marine aerosol particles onto the Yucatán Peninsula in winter, were found to be an important source of efficient INPs. Out of the three aerosol types analyzed in the present study, marine aerosol particles were identified as having the highest n_s values and the warmest onset freezing temperatures. Therefore, this is the aerosol type with the highest potential to impact the local hydrological cycle. Although marine aerosol particles are ubiquitous in the Yucatán Peninsula since it is surrounded by the GoM and the Caribbean Sea, synoptic phenomena such as cold fronts and tropical cyclones may enhance their concentration and likely their organic content. Based on the present results, it is desirable to conduct similar measurements during the phytoplankton bloom season, under the influence of a tropical cyclone, and in October–November when there is a much lower probability of the occurrence of BB, AD, and cold fronts.

In the absence of cold fronts, AD may be an important driver in mixed-phase cloud formation and, hence, in precipitation development. Given that these particles are not locally emitted, it is very important to improve the current understanding of the long-range transport of AD particles to Mexico, including their interannual variability.

Under the influence of all three air masses, supermicron particles were found to be the major contributor to the total INP concentrations. The present results highlight the importance of including the sampling of supermicron particles in future field measurements.

Data availability. The data from the present work can be found in the Supplement (Tables S3–S14). Also, they are available upon request to the corresponding author.

Supplement. The supplement related to this article is available online at: https://doi.org/10.5194/acp-21-4453-2021-supplement.

Author contributions. FC, GBR, and LAL designed the field campaigns and the experiments. FC, CRR, DC, WG, MG, AKB, and LAL built the UNAM-MOUDI-DFT. FC, CRR, HAO, DR, TA, BF, and LAL carried out the aerosol measurements. JM and HAO performed the chemical analyses. GBR, DB, DR, BF, JSK, JYH, and LAL installed the equipment and provided the infrastructure for the ADABBOY project. FC, CRR, and LAL wrote the paper with contributions from all coauthors.

Competing interests. The authors declare that they have no conflict of interest

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