



## Measurement report: High contributions of halocarbon and aromatic compounds to atmospheric volatile organic compounds in an industrial area

Ahsan Mozaffar<sup>1,2,3</sup>, Yan-Lin Zhang<sup>1,2,3</sup>, Yu-Chi Lin<sup>1,2,3</sup>, Feng Xie<sup>1,2,3</sup>, Mei-Yi Fan<sup>1,2,3</sup>, and Fang Cao<sup>1,2,3</sup>

<sup>1</sup>Yale-NUIST Center on Atmospheric Environment, International Joint Laboratory on Climate and Environment Change, Nanjing University of Information Science and Technology, Nanjing 210044, China

<sup>2</sup>Key Laboratory Meteorological Disaster, Ministry of Education and Collaborative Innovation Center on Forecast and Evaluation of Meteorological Disaster, Nanjing University of Information Science and Technology Nanjing 210044, China

<sup>3</sup>Jiangsu Provincial Key Laboratory of Agricultural Meteorology, College of Applied Meteorology, Nanjing University of Information Science and Technology, Nanjing 210044, China

**Correspondence:** Yan-Lin Zhang (dryanlinzhang@outlook.com)

Received: 2 April 2021 – Discussion started: 5 May 2021

Revised: 18 October 2021 – Accepted: 27 October 2021 – Published: 13 December 2021

**Abstract.** Volatile organic compounds (VOCs) are key components of tropospheric chemistry. We investigated ambient VOCs in an industrial area in Nanjing, China, between July 2018 and May 2020. The sum of the suite of measured total VOC (TVOC) concentrations was  $59.8 \pm 28.6$  ppbv (part per billion by volume) during the investigation period. About twice the TVOC concentrations were observed in the autumn ( $83 \pm 20$  ppbv) and winter ( $77.5 \pm 16.8$  ppbv) seasons compared to those in spring ( $39.6 \pm 13.1$  ppbv) and summer ( $38.8 \pm 10.2$  ppbv). In previous studies in Nanjing, oxygenated VOCs (OVOCs) and halocarbons were not measured, and the current TVOC concentration without halocarbons and OVOCs was similar to the previous investigation in the same study area. However, it was twofold higher than the one reported in the nonindustrial suburban area of Nanjing. Due to the industrial influence, the halocarbons VOC group ( $14.3 \pm 7.3$  ppbv, 24 %) was the second-largest contributor to the TVOCs after alkanes ( $21 \pm 7$  ppbv, 35 %), which is in contrast with the previous studies in Nanjing and also in almost all other regions in China. Relatively high proportions of halocarbons and aromatics were observed in autumn (25.7 % and 19.3 %, respectively) and winter (25.8 % and 17.6 %, respectively) compared to those in summer (20.4 % and 11.8 %, respectively) and spring (20.3 % and 13.6 %, respectively). According to the potential source contribution function (PSCF), short-distance transport from the surrounding industrial areas and cities was the main reason for the high VOC concentrations in the study area. According to positive matrix factorization (PMF) model results, vehicle-related emissions (33 %–48 %) contributed to the major portion of the ambient VOC concentrations. Aromatics, followed by alkenes, were the top contributors to the loss rate of OH radicals ( $L_{OH}$ ; 37 % and 32 %, respectively). According to the empirical kinetic modelling approach (EKMA) and relative incremental reactivity (RIR) analysis, the study area was in the VOC-sensitive regime for ozone ( $O_3$ ) formation during all measurement seasons. Therefore, alkenes and aromatics emissions from automobiles need to be decreased to reduce secondary air pollution formation in the study area.

## 1 Introduction

Air pollution characterized by severe ozone ( $O_3$ ) and haze pollution is a big problem in urban and industrial areas in China (He et al., 2019; Hui et al., 2018, 2019; Tan et al., 2018b; Jia et al., 2016; Feng et al., 2016). In recent years,  $O_3$  concentrations above the national standard and severe haze events have been frequently reported (He et al., 2019; Hui et al., 2019; Sheng et al., 2018; Feng et al., 2016; Tan et al., 2018b; Jia et al., 2016). As a precursor of  $O_3$  and secondary organic aerosol (SOA), volatile organic compounds (VOCs) are largely responsible for the severe air pollution in China (Song et al., 2018; Hui et al., 2019, 2018; He et al., 2019). Unfortunately, anthropogenic VOC emissions have been increasing over the last 2 decades in China, and they are expected to do so in the future (Mozaffar and Zhang, 2020, and references therein).

Atmospheric VOC has plenty of sources; it can be emitted from various anthropogenic and biogenic sources. Besides, it can also be formed in the atmosphere. Anthropogenic VOC sources mainly include industrial emissions, vehicle exhaust, solvent usage, biomass burning, and fuel evaporation. On the other hand, vegetation is the main biogenic source of VOC. In developed areas in China, vehicle exhaust and industrial emissions are the two major VOC sources (He et al., 2019; Hui et al., 2018, 2019; Mo et al., 2017; Song et al., 2018; An et al., 2014; Mozaffar and Zhang, 2020). Vehicle-related sources are more dominant in the North China Plain (NCP), central China (CC), and the Pearl River Delta (PRD) regions. But industry-related sources are more influential in the Yangtze River Delta (YRD) area (Zhang et al., 2017, 2018; Meng et al., 2015; Sun et al., 2019; He et al., 2019; An et al., 2017; Mozaffar and Zhang, 2020; Shao et al., 2016). Alkanes, alkenes, aromatics, oxygenated VOCs (OVOCs), and halocarbons are the most common VOC groups in the atmosphere (Hui et al., 2019, 2018; Hung-Lung et al., 2007; Song et al., 2018; Tiwari et al., 2010; He et al., 2019; Na et al., 2001). The seasonal change affects VOC concentration and composition. For example, the contribution from biogenic and solvent utilization increases in summer, and the contribution from combustion sources increases in winter (Mo et al., 2017; Song et al., 2018; An et al., 2014). The chemical reactivity of VOCs depends on their chemical composition. For instance, alkenes and aromatics are generally more reactive than alkanes (Carter, 2010). An analysis of the OH radical loss rate ( $L_{OH}$ ) is commonly used to understand the chemical reactivity of VOCs (Hui et al., 2018).

Industries are an important source of VOCs, and different reactive and hazardous VOC emissions from industries have already been reported in different areas of the Earth (Z. Zhang et al., 2018; Na et al., 2001; Hung-Lung et al., 2007; Yan et al., 2016; Tiwari et al., 2010; Shi et al., 2015). For instance, Z. Zhang et al. (2018) reported a high concentration of alkanes ( $21.3 \pm 17.8 \text{ mg m}^{-3}$  out of the total  $23.1 \pm 24.5 \text{ mg m}^{-3}$ ) and a lifetime cancer risk from

different aromatics and halocarbons in a petroleum refinery in Guangzhou, China. A high concentration of OVOCs ( $829.7 \pm 1076.7 \text{ ppb C}$ , where ppb is parts per billion, out of a total of  $1317.3 \pm 1184.5 \text{ ppb C}$ ) was observed in an industrial area in Ulsan, South Korea (Na et al., 2001). Hung-Lung et al. (2007) mentioned a high concentration of aromatics ( $\sim 90 \text{ ppb}$  out of a total of  $\sim 160 \text{ ppb}$ ) in an industrial area in Taiwan. A high concentration of halocarbons ( $9590.2 \text{ mg m}^{-3}$  out of a total of  $19\,652 \text{ mg m}^{-3}$ ) was observed in an iron smelting plant in Liaoning, China (Shi et al., 2015). Y. Zhang et al. (2018) mentioned a high concentration of alkanes ( $39.4 \text{ ppbv}$ , parts per billion by volume, out of a total of  $94.15 \text{ ppbv}$ ) and aromatics ( $18.9 \text{ ppbv}$  out of a total of  $94.15 \text{ ppbv}$ ) in a petrochemical industrial area in Shanghai, China. Therefore, VOC composition varies among the industries/industrial areas in different regions. Mostly, short-term investigations are performed to characterize the VOCs in industry-affected areas. In the current study, we carried out a comprehensive investigation on VOC in an industrial area in Nanjing between July 2018 and May 2020. Several VOC investigations have already been performed in the Nanjing industrial area, but OVOCs and halocarbons were not measured in those studies (An et al., 2017, 2014). However, OVOCs and halocarbons are already mentioned as being one of the highest concentrated VOC groups in other industrial regions (Na et al., 2001; Shi et al., 2015). In the current study area, a high concentration of alkanes ( $19.6 \text{ ppbv}$  out of the total  $43.5 \text{ ppbv}$ ) and alkenes ( $11.1 \text{ ppbv}$  out of the total  $43.5 \text{ ppbv}$ ) was observed in a previous investigation (An et al., 2014). Besides the incomplete VOC measurements,  $O_3$  formation sensitivity to its precursors was not investigated properly by using a photochemical box model in Nanjing. Moreover, the source apportionment of VOCs was not conducted for different seasons of a year.

In the current study, we report the variations in concentrations and compositions of VOC during the observation period. We present the possible source areas and potential sources of VOC based on a potential source contribution function (PSCF) and positive matrix factorization (PMF) model analysis. We also report the chemical reactivity of the VOC using  $L_{OH}$ . We also present the sensitivity analysis of  $O_3$  formation using the empirical kinetic modelling approach (EKMA) and relative incremental reactivity (RIR) analysis. Therefore, this study provides valuable information to the scientific community and policymakers.

## 2 Material and methods

### 2.1 Sampling site description, gases analysis, and meteorology data

Field measurements were carried out at the Nanjing University of Information Science and Technology ( $32.1^\circ \text{N}$ ,  $118.4^\circ \text{E}$ ) for about 1 month in winter, spring, and summer and 3 months in autumn between July 2018 and May 2020.

The sampling site was on the rooftop of a building ( $\sim 20$  m). The sampling site is surrounded by different chemical and petrochemical industries, steel plants, gas stations, high traffic roads, and residential areas. A detailed description of the sampling site can be found elsewhere (Mozaffar et al., 2020).

We analysed ambient air VOCs using an online gas chromatography flame ionization detector mass spectrometer (GC-FID/MS) instrument (AC-GCMS 1000; Guangzhou HeXin Instrument Co., LTD., China). The FID detector analysed C2–C5 VOCs, and the MS analysed C6–C12 VOCs. The instrument analysed one sample every hour. During the investigation period, we inspected and calibrated the instrument regularly to ensure the accuracy of the data (Mozaffar et al., 2020). We monitored the  $O_3$  concentrations using a 49i  $O_3$  analyser (Thermo Fisher Scientific Inc., USA).  $NO$ ,  $NO_2$ , and  $NO_x$  concentrations were measured using a 42i  $NO$ – $NO_2$ – $NO_x$  analyser (Thermo Fisher Scientific Inc., USA).  $SO_2$  concentrations were followed using a 43i  $SO_2$  analyser (Thermo Fisher Scientific Inc., USA), and  $CO$  concentrations were measured using a 48i  $CO$  analyser (Thermo Fisher Scientific Inc., USA). We also measured temperature and relative humidity, wind speed, wind direction, and solar radiation with HMP155 (Vaisala, Finland), 010C (Met One Instruments, Inc., USA), 020C (Met One Instruments, Inc., USA), and CNR4 (Kipp & Zonen, The Netherlands) analysers, respectively. A detailed description of the instrumentation, sampling procedure, analysis, quality control, and calibration procedure can be found elsewhere (Mozaffar et al., 2020).

## 2.2 Positive matrix factorization (PMF) model and potential source contribution function (PSCF)

We used the positive matrix factorization (PMF) model (U.S. Environmental Protection Agency, U.S. EPA, version 5.0) for the source apportionment of VOCs. A detailed description of the model can be found elsewhere (Hui et al., 2019; Song et al., 2018). We used 20 potential VOC tracers (Figs. S1–S4) in the PMF model. The error fraction was set to 20 % for the sample data uncertainty estimation. We explored the PMF factor number from 4 to 8 to determine the optimal number of sources. Finally, we decided to choose a 7- to 8-factor solution for different seasons, as  $Q_{\text{true}}/Q_{\text{robust}}$  was  $\sim 1.0$ ,  $Q_{\text{true}}/Q_{\text{expected}}$  ranged from 0.99 to 1.45 (Hui et al., 2019), and strong correlations (0.7–0.8) were observed between the concentrations extracted from the model and the observed concentrations of each compound (He et al., 2019).

We used the potential source contribution function (PSCF) to locate possible source areas of VOCs for different seasons during the investigation period. We used ZeFir analysis software to do the PSCF analysis and the Hysplit4 model to cluster the backward trajectories (Petit et al., 2017). Backward trajectories at the sampling site were estimated using the National Centres for Environmental Prediction data (<ftp://arlftp.arlhq.noaa.gov/pub/archives/gdas1>, last access: 27 September

2020). We estimated 72 h of backward trajectories 24 times a day, arriving at 500 m above the ground surface, using the HYSPLIT4 model. For the PSCF analysis, we divided the geographic region covered by the back trajectories into an array of  $0.1^\circ \times 0.1^\circ$  grid cells. The mean TVOC concentration was used as the VOC reference value. More details about the PSCF analysis can be found in previous studies (Chen et al., 2018).

## 2.3 OH radical loss rate ( $L_{OH}$ )

To evaluate the daytime photochemistry of VOCs, we estimated their OH radical loss rate ( $L_{OH}$ ). The following equation was used to estimate the  $L_{OH}$  ( $s^{-1}$ ) (Zhang et al., 2020).

$$L_{OH} = [VOC]_i \times k_{OH,i}, \quad (1)$$

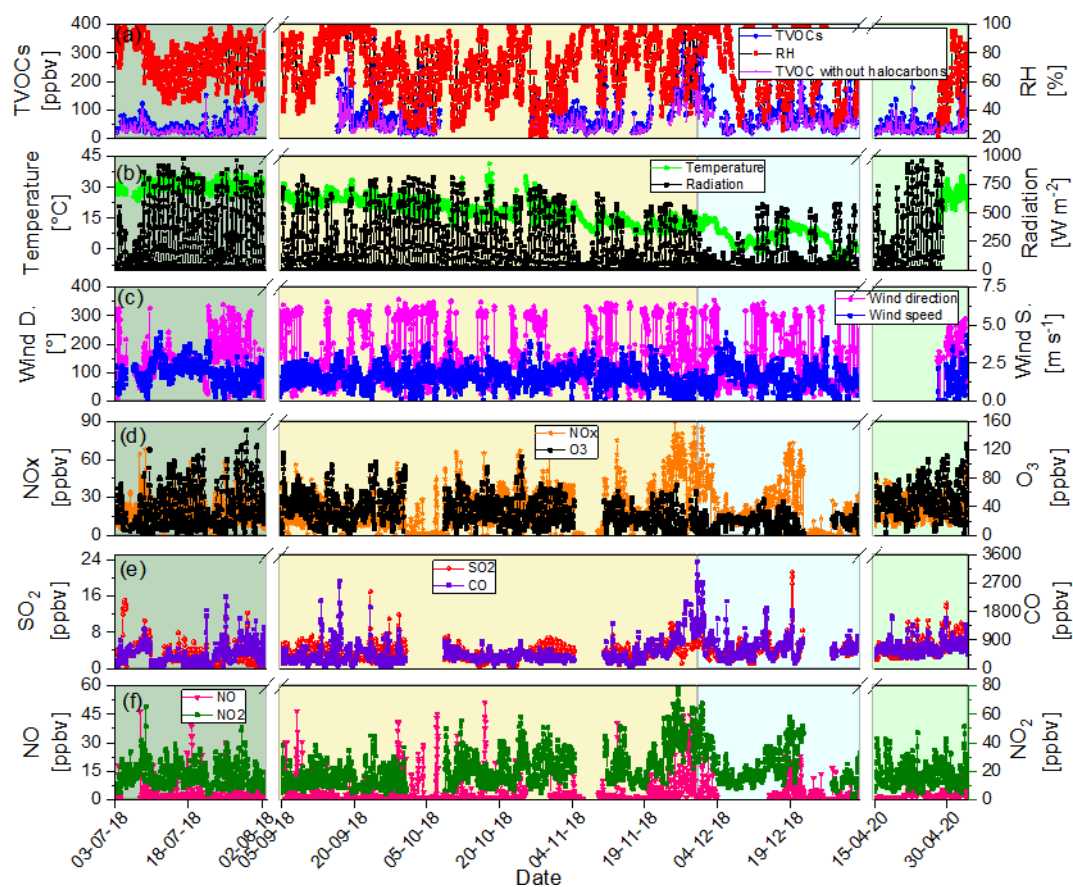
where  $[VOC]_i$  is the concentration of VOC species  $i$  (molecules per cubic centimetre), and  $k_{OH,i}$  (cubic centimetres per molecule per second) is the reaction rate constant of  $i$  VOC with the OH radical. The  $k_{OH}$  values for the VOCs are collected from Carter (2010; Table S1).

## 2.4 Empirical kinetic modelling approach (EKMA) and relative incremental reactivity (RIR)

The empirical kinetic modelling approach (EKMA) is a well-known procedure to develop the  $O_3$  formation reduction strategy by testing the relationship between ambient  $O_3$  and its precursors (He et al., 2019; Hui et al., 2018; Vermeuel et al., 2019; Tan et al., 2018b). To obtain data for the EKMA isopleth, we used the Framework for 0-D Atmospheric Model (F0AM; v3.2; Wolfe et al., 2016), a photochemical box model run by Master Chemical Mechanism (MCM) v3.2 chemistry (Jenkin et al., 1997, 2003, 2015; Saunders et al., 2003). The F0AM-MCM box model can simulate 16 940 reactions of 5733 chemical species. The box model was run using the VOCs and gas concentrations and the meteorological data as input. In total, 61 VOCs were constrained in the model as the rest of the observed VOC species reactions are not included yet in MCM. These constrained VOCs are listed in Table S1. To generate the  $O_3$  isopleth from the model-simulated data, a total of 121 reduction scenarios ( $11 NO_x \times 11$  VOC concentrations) were simulated, and the maximum  $O_3$  produced at each model scenario was saved.

The relative incremental reactivity (RIR; Cardelino and Chameides, 1995) is also used to test the  $O_3$  formation sensitivity of its precursors. We also utilized the F0AM-MCM box model data to estimate the RIR. The RIR is simply defined as the percentage change in  $O_3$  formation per percentage change in the precursor's concentration. In this study, we reduced the precursor concentration by 10 % for the RIR estimation. The RIR was estimated using the following equation.

$$RIR(X) = \frac{[P_{O_3}(X) - P_{O_3}(X - \Delta X)] / P_{O_3}(X)}{[\Delta X] / [X]}, \quad (2)$$



**Figure 1.** Time series of hourly meteorological parameters, inorganic air pollutants, TVOCs, and TVOCs without halocarbon concentrations during the observation period at Nanjing. The green, yellow, cyan, and light green shaded areas indicate summer, autumn, winter, and spring seasons, respectively. The discontinuity of the measured data is due to instrument failure.

where  $[X]$  is the observed concentration of a precursor  $X$ , and  $[\Delta X]$  is the changes in the concentration of  $X$ .  $P_{O_3}(X)$  and  $P_{O_3}(X-\Delta X)$  are the simulated net  $O_3$  production, using the observed and reduced concentration of the precursor  $X$ , respectively.

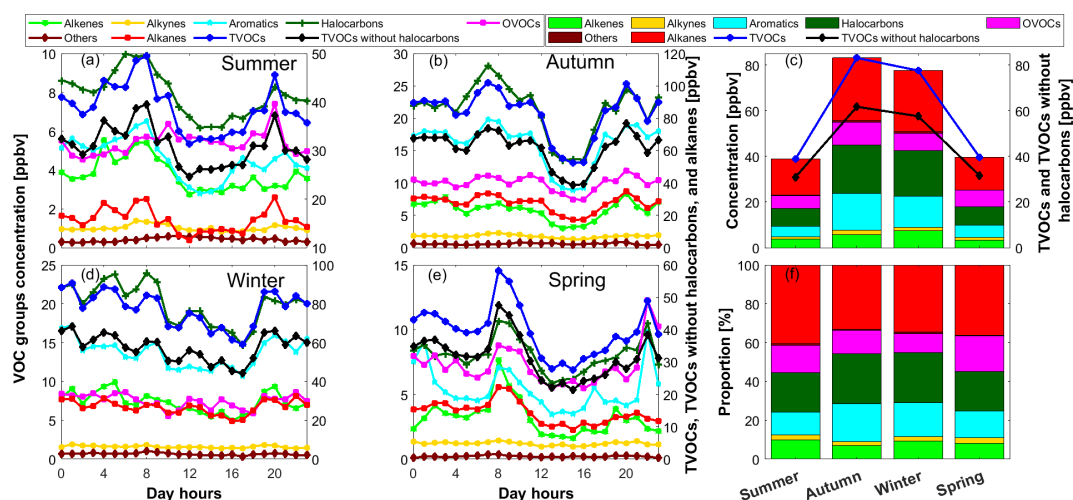
### 3 Results and discussion

#### 3.1 Overview of the metrological conditions and air pollutants concentrations

The time series of the hourly data is shown in Fig. 1. The discontinuity of the time series data is due to the failure of the instruments and the COVID-19-related lockdowns. The data measured between July and August 2018 are termed summertime data. Similarly, data collected between September and November 2018 are autumn time data, data collected between December 2018 and January 2019 are wintertime data, and data collected between April and May 2020 are springtime data. Overall, the observed temperature and solar radiation gradually decreased from summer to winter and increased back to the summertime level in spring. The tem-

perature ranged between  $-5.7$  and  $41.4^\circ\text{C}$  during the measurement period. The relative humidity values varied from 18 % to 100 %; high values were generally observed in winter and autumn. During the observation period, wind speeds ranged between  $0.1$  and  $7.5\text{ ms}^{-1}$ . Wind prevailed at the sampling site from many directions during the measurement periods; more details about the wind direction will be discussed in Sect. 3.3.2. The  $O_3$  and  $NO_x$  concentrations varied from 2 to 160 ppbv and 0.4 to 90 ppbv, respectively. Whereas high  $O_3$  concentrations ( $> 80$  ppbv) were observed in summer and spring, high  $NO_x$  concentrations were measured in winter and at the end of autumn. The CO and  $SO_2$  concentrations ranged from 83 to 3398 ppbv and 0.5 to 21 ppbv, respectively. In general, high CO and  $SO_2$  concentrations were observed in the winter and spring. The measured NO and  $NO_2$  concentrations varied from 0.4 to 51 ppbv and 1 to 79 ppbv, respectively. In general, high NO and  $NO_2$  concentrations were observed in autumn and winter. The TVOC concentrations estimated with all the measured VOCs varied between 9 and 393 ppbv during the observation period, and the high values were measured in autumn and winter. More details about the





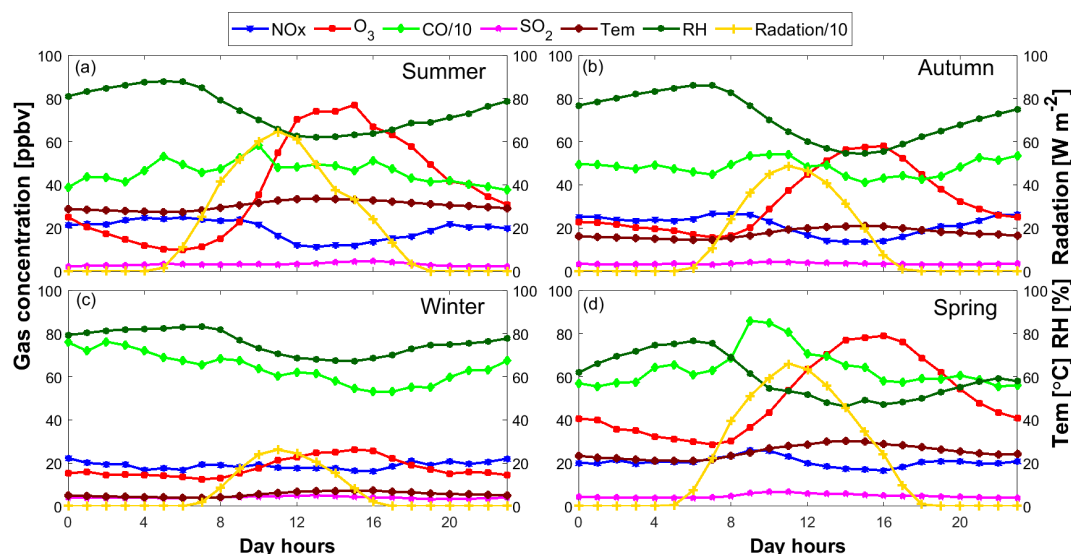
**Figure 2.** Diurnal variations in different VOC groups, TVOCs, and TVOCs without halocarbons concentrations in different seasons (a, b, d, e). Seasonal variations in average concentrations and the proportion of different VOC groups, TVOCs, and TVOCs without halocarbons (c, f).

above-mentioned parameters will be discussed in the following section.

### 3.2 Concentration and composition of VOCs

In total, 100 VOCs were observed in the Nanjing industrial area, including 27 alkanes, 11 alkenes, 1 alkyne, 17 aromatics, 31 halocarbons, 12 OVOCs, and 1 other (carbon disulfide; Table S2). Ethane ( $5.8 \pm 2.5$  ppbv), propane ( $4.2 \pm 1.5$  ppbv), and ethylene ( $3 \pm 1.6$  ppbv) were the most abundant VOCs in the study area during the observation period. However, we observed season-wise variations in the order of abundant VOC species (Table S2). For instance, acetone was the third-highest concentrated VOC in spring. The above-mentioned four VOC species are also frequently mentioned as being the most abundant VOCs in different regions in China (Deng et al., 2019; He et al., 2019; Li et al., 2018; Ma et al., 2019). We compared the individual VOC concentrations with the available data presented in recent investigations. The individual VOC concentrations in the current observation were similar to those reported in the previous investigation in the same study area (An et al., 2017). However, they were almost twice those found in a nonindustrial suburban area in Nanjing (Wu et al., 2020; Table S2). Some of the differences may be due to the differences in the observation period. The reported yearly concentrations (Wu et al., 2020) were probably estimated over continuous measurement data for a year. However, in the current observation, the measurements were not continuously performed during all the days of a year. Individual VOC concentrations in autumn were about 1.4 times lower than those measured in Beijing in October and November (Li et al., 2015). The wintertime individual VOC concentrations were also about 1.4 times lower than those measured in

Shanghai during November–January (Y. Zhang et al., 2018). But, the yearly individual VOC concentrations in the current observation were similar to those measured in Guangzhou from June to May (Zou et al., 2015). During the observation period, the concentrations of different VOC groups were of the order of alkanes ( $21 \pm 7$  ppbv; 35 %) > halocarbons ( $14.3 \pm 7.3$  ppbv; 24 %) > aromatics ( $9.9 \pm 5.8$  ppbv; 17 %) > OVOCs ( $7.5 \pm 1.9$  ppbv; 13 %) > alkenes ( $5 \pm 1.9$  ppbv; 8 %) > alkynes ( $1.4 \pm 0.3$  ppbv; 2 %) > others ( $0.5 \pm 0.2$  ppbv; 1 %). Relatively high proportions of halocarbons and aromatics were observed in autumn (25.7 % and 19.3 %, respectively) and winter (25.8 % and 17.6 %, respectively) compared to those measured in summer (20.4 % and 11.8 %, respectively) and spring (20.3 % and 13.6 %, respectively; Fig. 2f). It could be related to the burning of biomass and fossil fuel for additional heating. Similar to the observation in the current study, the alkane family is generally the most abundant VOC group in China (Mozaffar and Zhang, 2020). The relatively high contribution of halocarbons to the TVOCs could be related to the industrial emissions in the study area. However, halocarbons and OVOCs were neither measured in previous investigations in the same study area (An et al., 2014, 2017; Shao et al., 2016) nor in another suburban area in Nanjing (Wu et al., 2020). Either the aromatics or alkenes group was mentioned as being the second most abundant VOC group in those studies in Nanjing, which is the third and fifth most abundant VOC group in the current investigation. The TVOC concentration estimated with all the measured VOCs was  $59.8 \pm 28.6$  ppbv over the whole observation period, and relatively higher TVOC concentrations were measured in autumn ( $83 \pm 20$  ppbv) and winter ( $77.5 \pm 16.8$  ppbv), compared to those observed in spring ( $39.6 \pm 13.1$  ppbv) and summer ( $38.8 \pm 10.2$  ppbv). The TVOC concentra-



**Figure 3.** Diurnal variations in weather conditions and  $\text{NO}_x$ ,  $\text{O}_3$ , CO, and  $\text{SO}_2$  concentrations in different seasons. Note that the plotted CO concentrations and solar radiation values are reduced 10-fold for a better visualization.

tions without halocarbons were  $45.4 \pm 20.4$ ,  $61.7 \pm 14.6$ ,  $57.4 \pm 11.8$ ,  $31.6 \pm 10.9$ , and  $30.9 \pm 8.2$  ppbv during the whole observation period, autumn, winter, spring, and summer, respectively. As mentioned before, halocarbons and OVOCs were not reported in the previous investigations in Nanjing (An et al., 2017; Wu et al., 2020). The current TVOC concentration without halocarbons and OVOCs was similar to the previous investigation in the same study area; however, it was twofold higher than the one reported in the nonindustrial suburban area in Nanjing. The diurnal variation in the TVOC, alkenes, aromatics, halocarbons, OVOC, and alkanes concentrations showed a double-hump structure (Fig. 2a, b, d, e). This double-hump pattern indicates the contribution of traffic emissions during the rush hours in the morning and evening. The lowest concentration of the TVOCs and different VOC groups was reached between 12:00 and 16:00 LT (local time). Conversely, the highest concentration of  $\text{O}_3$  was reached in that period (Fig. 3). The lowest  $\text{O}_3$  concentrations were observed in winter, which was consistent with the solar radiation.

### 3.3 Sources of VOCs

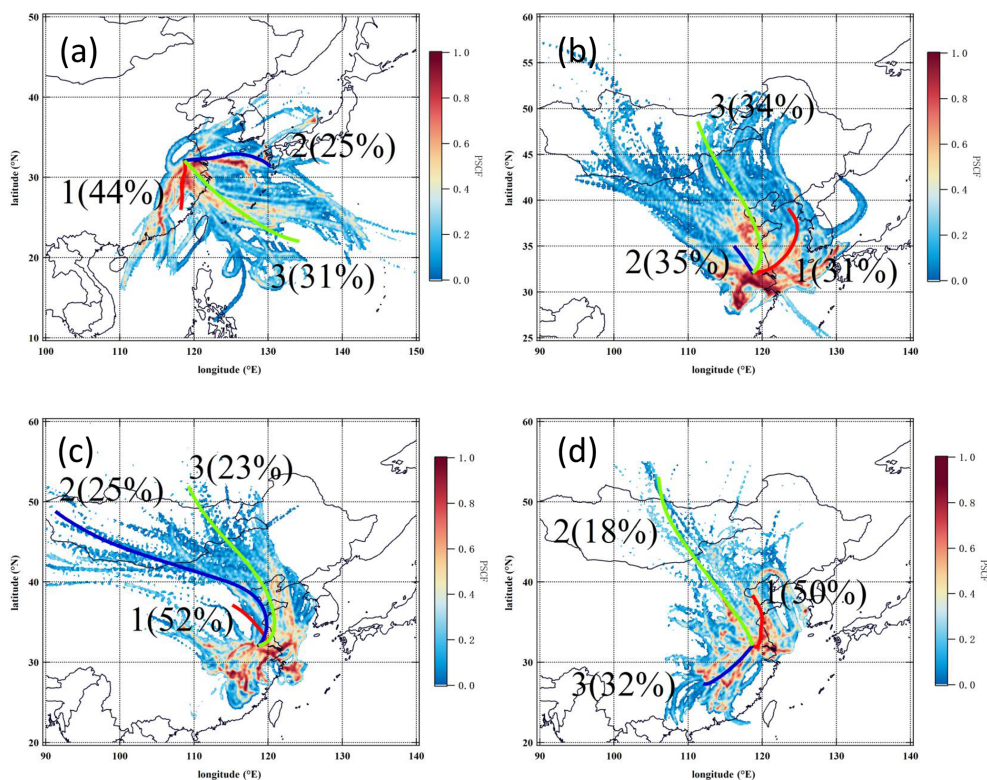
#### 3.3.1 Potential source contribution function (PSCF)

Besides the local sources, both the long-distance and short-distance transport of air mass could bring VOCs to the study area. Figure 4 shows the wind cluster and PSCF analysis results for different seasons. During summer, the major air masses were short-distance transport from the southern (44 %) direction and two long-distance types of transport from the southeastern (31 % and 25 %) direction. High PSCF values were in the nearby southern and southeastern directions; therefore, VOC pollution in the study area was mainly

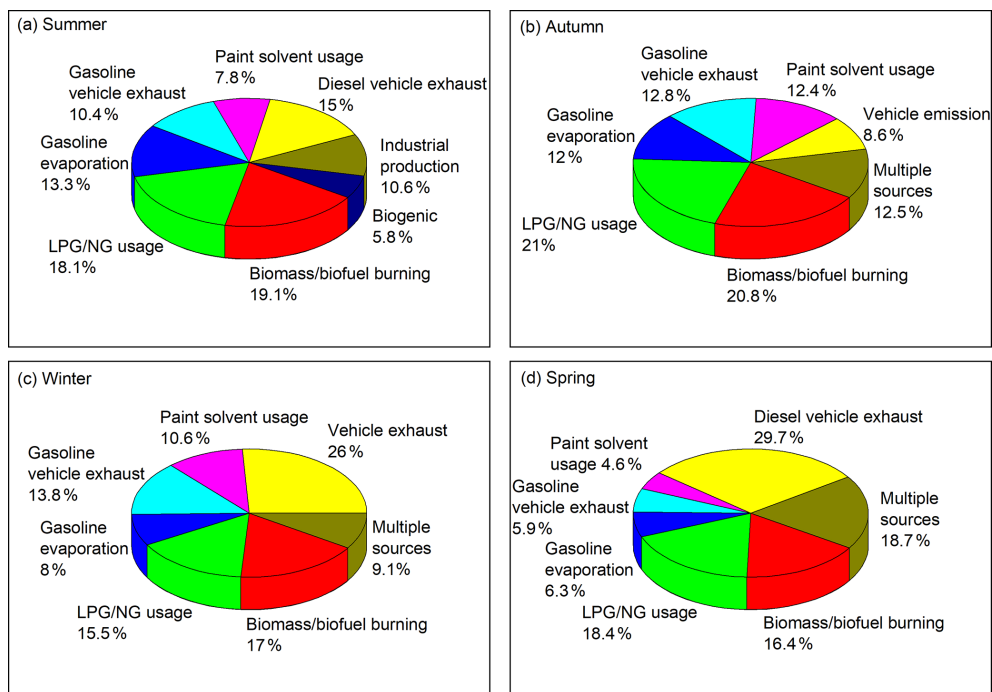
affected by short-distance transport from the southern and eastern directions. During autumn, the dominant air masses were short-distance transport from the northwestern direction (35 %) and long-distance transport from the northern (34 %) direction. However, according to the PSCF analysis, VOC pollution was mainly influenced by the short-distance transport from the southern direction. During winter, short-distance transport from the northwestern (52 %) direction was the major incoming air masses to the study area. According to the PSCF values, the short-distance air masses from the south and north mainly transported VOC to the receptor site. During the spring, air masses were primarily transported from the north (50 %) and the southwest (32 %). A minor long-distance air mass was transported from the northwestern (18 %) direction. Atmospheric VOCs were mainly transported by these air masses to the study site, mostly from the nearby areas. Overall, the high PSCF values were concentrated around the measurement site. Therefore, short-distance transport from the surrounding areas and cities was the main reason for the high VOC concentration. The above conclusion makes perfect sense as the sampling site is surrounded by different chemical and petrochemical industries, steel plants, gas stations, high traffic roads, and residential areas.

#### 3.3.2 PMF model analysis

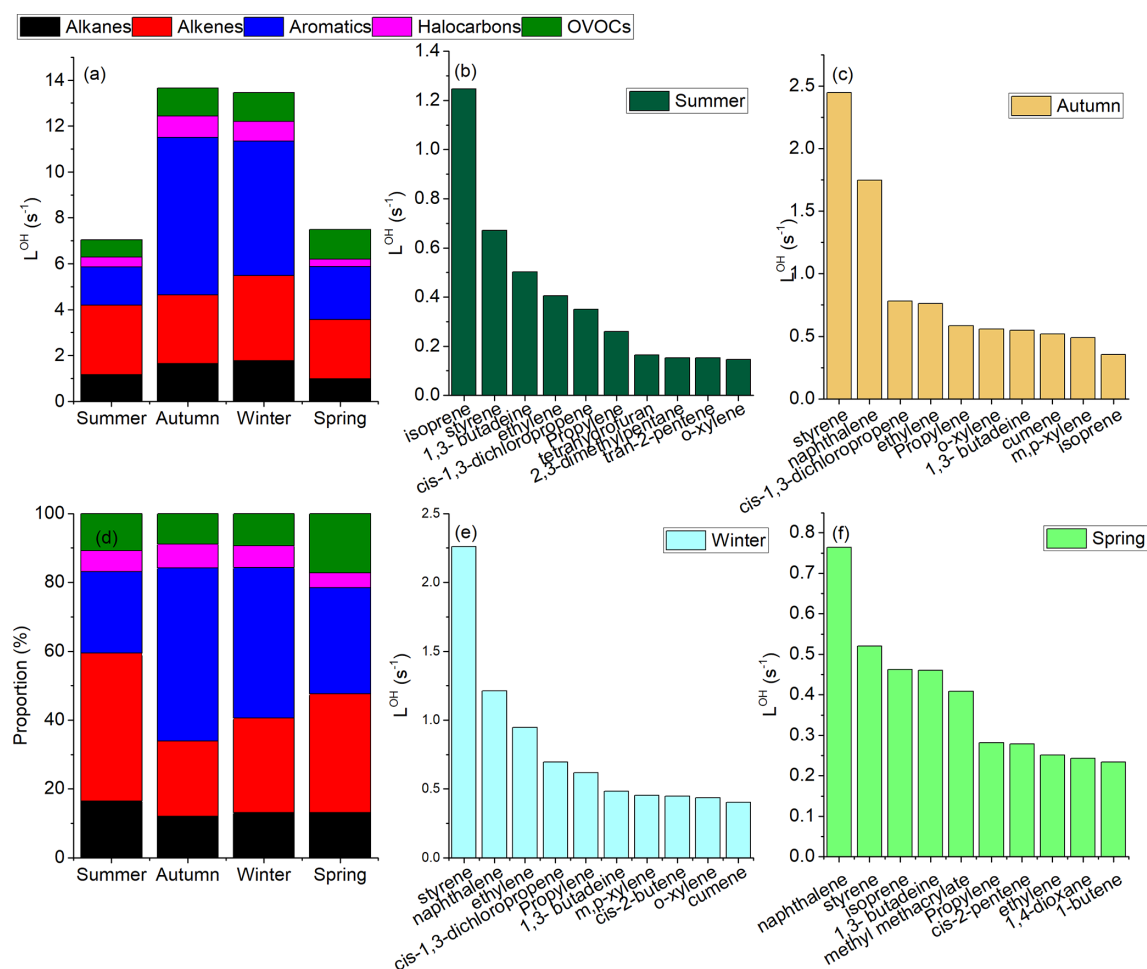
According to the PMF model analysis, five VOC sources were common during all the measurement seasons. They were biomass/biofuel burning, liquefied petroleum gas/natural gas (LPG/NG) usage, gasoline evaporation, gasoline vehicle exhaust, and paint solvent usage (Sect. S1). The biogenic source was distinguished only in the sum-



**Figure 4.** Wind cluster and PSCF analysis during (a) summer (b) autumn, (c) winter, and (d) spring, based on the 72 h backward air mass trajectories from the study area.



**Figure 5.** Relative contributions of different sources to ambient VOCs in the Nanjing industrial area during different seasons.



**Figure 6.** Contribution to OH loss rates of different VOC groups and the top 10 VOC species in different seasons.

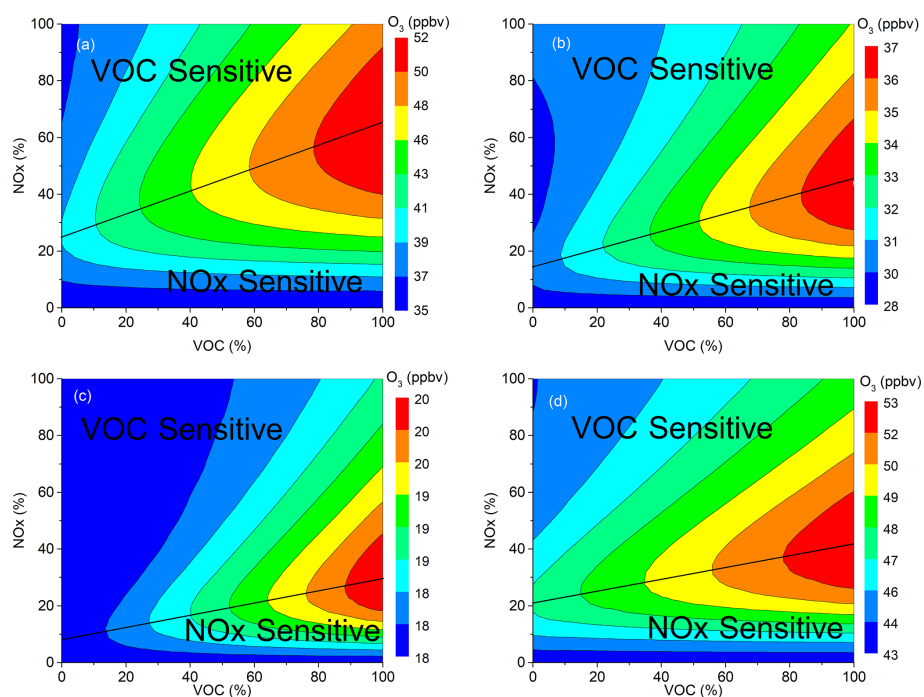
mer. Figure 5 shows the relative contributions of different sources to ambient VOCs during different seasons. Overall, vehicle-related sources contributed the most to the ambient VOC concentrations. The total contributions of vehicle-related emissions were about 39 %, 33 %, 48 %, and 42 % in summer, autumn, winter, and spring, respectively. The contributions of biomass/biofuel burning sources were about 19 %, 21 %, 17 %, and 16.4 % in summer, autumn, winter, and spring, respectively. Besides these two sources, LPG/NG usage (18 %, 21 %, 16 %, and 18 %, respectively) and paint solvent usage (8 %, 12 %, 11 %, 5 %, respectively) were two other important VOC sources during those four seasons.

### 3.4 Chemical reactivity ( $L_{OH}$ )

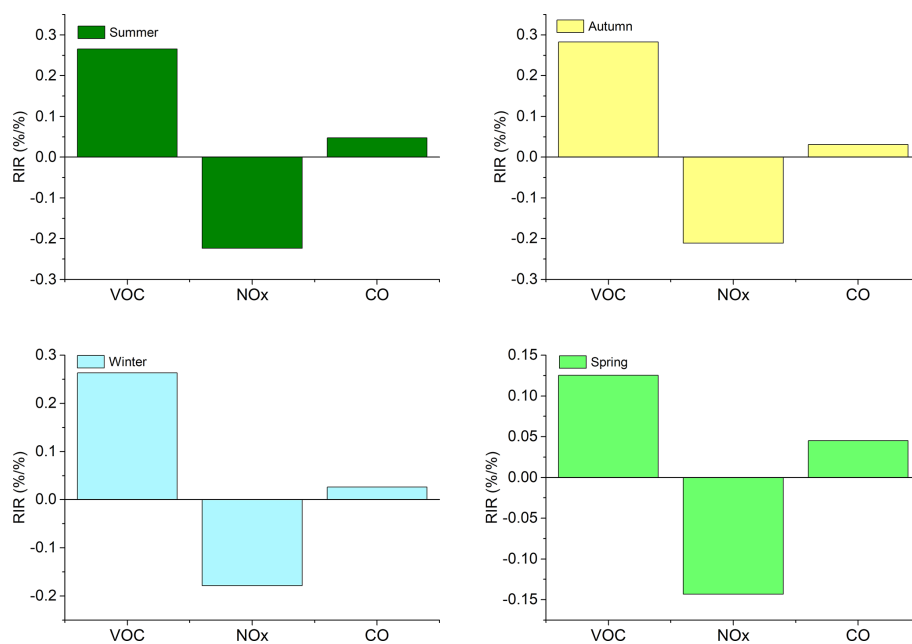
The estimated loss rates of OH radical ( $L_{OH}$ ) with VOCs were about twofold higher in autumn ( $13.7 \text{ s}^{-1}$ ) and winter ( $13.5 \text{ s}^{-1}$ ) compared to those in summer ( $7 \text{ s}^{-1}$ ) and spring ( $7.5 \text{ s}^{-1}$ ) (Fig. 6a). The relatively high  $L_{OH}$  values in autumn and winter were due to the relatively high VOC concentrations in these seasons (Fig. 2). The average  $L_{OH}$  value

was  $10.4 \pm 3.6 \text{ s}^{-1}$  over the four seasons. It was in a similar range to the values determined in Guangzhou ( $10.9 \text{ s}^{-1}$ ), Chongqing ( $10 \text{ s}^{-1}$ ), Xian ( $1.6\text{--}16.2 \text{ s}^{-1}$ ), and Tokyo ( $7.7\text{--}13.4 \text{ s}^{-1}$ ) but higher than the values estimated in Shanghai ( $2.9\text{--}5$ ,  $6.2 \text{ s}^{-1}$ ) and Beijing ( $7 \text{ s}^{-1}$ ; Tan et al., 2019; Zhu et al., 2019; Yoshino et al., 2012; Song et al., 2020). While alkene was the highest contributor to the  $L_{OH}$  in summer ( $3 \text{ s}^{-1}$ ; 43 %) and spring ( $2.6 \text{ s}^{-1}$ ; 35 %), aromatics were the maximum contributor in autumn ( $6.9 \text{ s}^{-1}$ ; 50 %) and winter ( $5.9 \text{ s}^{-1}$ ; 44 %; Fig. 6d). An increase in the OH loss rate by OVOCs was observed in spring (17 %) compared to the other seasons (10 %, 8 %, and 9 % in summer, autumn, and winter, respectively). Over the four seasons, the contribution of VOC groups to  $L_{OH}$  exhibited the following trend: aromatics > alkenes > alkanes > OVOCs > halocarbons. Similar to the current study, the aromatic group is also mentioned as being the maximum contributor to  $L_{OH}$  in different regions in China. However, the alkene group is generally reported as being the top contributor to  $L_{OH}$  (Zhang et al., 2020; Zhao et al., 2020; Hui et al., 2018; Song et al., 2020). Figure 6 also shows the top 10 VOCs contributing to  $L_{OH}$  for the differ-





**Figure 7.** An O<sub>3</sub> isopleth diagram for (a) summer (b) autumn, (c) winter, and (d) spring, based on percentage changes in VOCs and NO<sub>x</sub> concentrations in Nanjing and the corresponding modelled O<sub>3</sub> production.



**Figure 8.** The RIR values of the VOC, NO<sub>x</sub>, and CO for the different seasons in Nanjing.

ent seasons. Whereas isoprene was the highest contributor to  $L_{OH}$  in summer, it was styrene in autumn and winter. On the other hand, naphthalene was the main contributor to  $L_{OH}$  in the spring. Overall, styrene, naphthalene, ethylene, and isoprene were the main contributors to  $L_{OH}$ . In previous studies in China, these compounds are also mentioned as being one of the highest contributors to  $L_{OH}$  (Zhao et al., 2020; Hui et al., 2018; Song et al., 2020).

### 3.5 Sensitivity analysis of $O_3$ formation

Figure 7 shows the EKMA isopleth diagrams of  $O_3$  for different seasons. In all the diagrams, VOC and  $NO_x = 100\%$  is the base case. The ridgeline divided the diagrams into two regimes, with the VOC-sensitive regime above the line and  $NO_x$ -sensitive regime below the line. For all the seasons, the study area fell above the ridgeline. Moreover,  $O_3$  production decreased with the decrease in VOC concentration. Therefore, the study area was in the VOC-sensitive regime for  $O_3$  formation during all the seasons. As a case study,  $O_3$  formation sensitivity to its precursors was tested on a high  $O_3$  concentration day (29 July 2018; maximum 126 ppbv). During the high  $O_3$  episode, the study area was also in the VOC-sensitive regime for  $O_3$  formation (Fig. S5). We also employed the RIR analysis to evaluate the  $O_3$  production sensitivity to VOC,  $NO_x$ , and CO concentrations (Fig. 8). The RIR value of VOC was the highest during all the seasons. It indicates that the  $O_3$  production was more sensitive to the reduction in the VOC concentration. This finding is consistent with the EKMA isopleth diagrams (Fig. 7). Except for the spring, the RIR values of CO were very small relative to those for the VOC. It indicates that the CO concentrations were relatively less important for the  $O_3$  formation during those seasons. The RIR values for  $NO_x$  were negative during all the seasons, implying that the  $O_3$  formation was in the  $NO_x$ -titration regime in the study area. From the above analysis, it is evident that a reduction in the VOC concentration in the study area will be the most efficient way to reduce the  $O_3$  formation. The previous two studies performed in Nanjing also concluded with the same finding based on VOC/ $NO_x$  ratios and RIR analysis (An et al., 2015; Xu et al., 2017). Our findings are also consistent with the previous studies performed in other regions of China (Tan et al., 2018b; He et al., 2019; Feng et al., 2019; Ma et al., 2019). However, the  $NO_x$ -sensitive regions for  $O_3$  formation are also found in China (Tan et al., 2018a; Jia et al., 2016).

## 4 Conclusions

Nanjing is one of the biggest industrial cities in China. We performed a long-term investigation of ambient VOCs in Nanjing. Compared to the previous investigation in the current study area, similar TVOC concentrations were observed. However, about twofold higher TVOC concentrations were observed compared to the one previously reported in a non-

industrial suburban area in Nanjing. TVOC concentrations were about twice as high in autumn and winter compared to those observed in summer and spring. Generally, haze pollution occurs most frequently in autumn and winter. Therefore, VOC concentration reduction in these seasons is an important step to reduce haze pollution in the study area. The halocarbon group was the second-largest contributor to the TVOCs following alkanes, indicating the impact of industrial emissions on the local air. As halocarbons are carcinogenic, their emissions need to be reduced. The short-distance air mass from the surrounding areas and cities was the main reason for the high VOC concentration. Hence, local emissions need to be reduced. Vehicle-related emissions were the major VOC source in the study area; thus, emission reduction from this source should have a higher priority. Aromatics and alkenes were the major contributors to the  $L_{OH}$ . Therefore, these two VOC groups should have a higher priority in emission reduction policies and strategies. During all the seasons, the study area was in the VOC-sensitive regime for  $O_3$  formation. Therefore, VOC emission reduction is the most effective way to decrease the local  $O_3$  formation.

**Data availability.** All the data presented in this article can be accessed through <https://osf.io/bm6cs/> (last access: 17 December 2020; Mozaffar, 2021).

**Supplement.** The supplement related to this article is available online at: <https://doi.org/10.5194/acp-21-18087-2021-supplement>.

**Author contributions.** YLZ designed and supervised the project. MYF, FX, YCL, FC, and AM conducted the measurements. AM analysed the data and prepared the paper. All authors contributed to the discussions in order to improve the article.

**Competing interests.** The authors declare that they have no conflict of interest.

**Acknowledgements.** The authors are thankful for the funding support from the National Nature Science Foundation of China (grant no. 41977305), the Provincial Natural Science Foundation of Jiangsu (grant no. BK20180040), and the Jiangsu Innovation and Entrepreneurship Team. We are also grateful to Zijin Zhang and Meng-Yao Cao, for their help with the sampling.

**Disclaimer.** Publisher's note: Copernicus Publications remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

**Financial support.** This research has been supported by the National Nature Science Foundation of China (grant no. 41977305), the Provincial Natural Science Foundation of Jiangsu (grant no. BK20180040), and the Jiangsu Innovation and Entrepreneurship Team.

**Review statement.** This paper was edited by Nga Lee Ng and reviewed by four anonymous referees.

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