



Supplement of

Isotopic evidence for dominant secondary production of HONO in near-ground wildfire plumes

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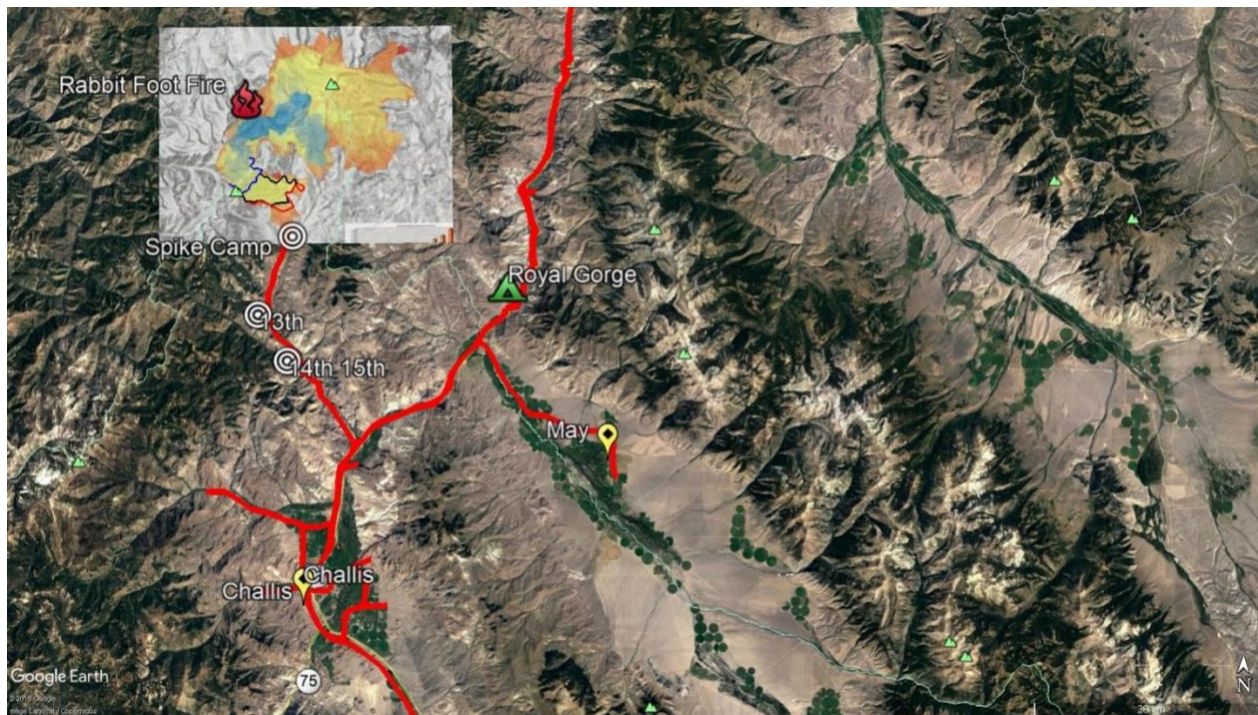


Figure S1. WE-CAN 2018 sampling map for Rabbit Foot fire (RF) in the Challis area of Salmon-Challis National Forest in central Idaho, from August 9th to August 18th 2018. Of note are important dates, which give the locations of the different approaches to the RFF during mobile measurements up MCR (H. Munro, “An Investigation of Nitrous and Nitric Acid Diurnal Cycles in Biomass Burning Plumes”, Thesis, University of New Hampshire, Durham, NH, 2019).

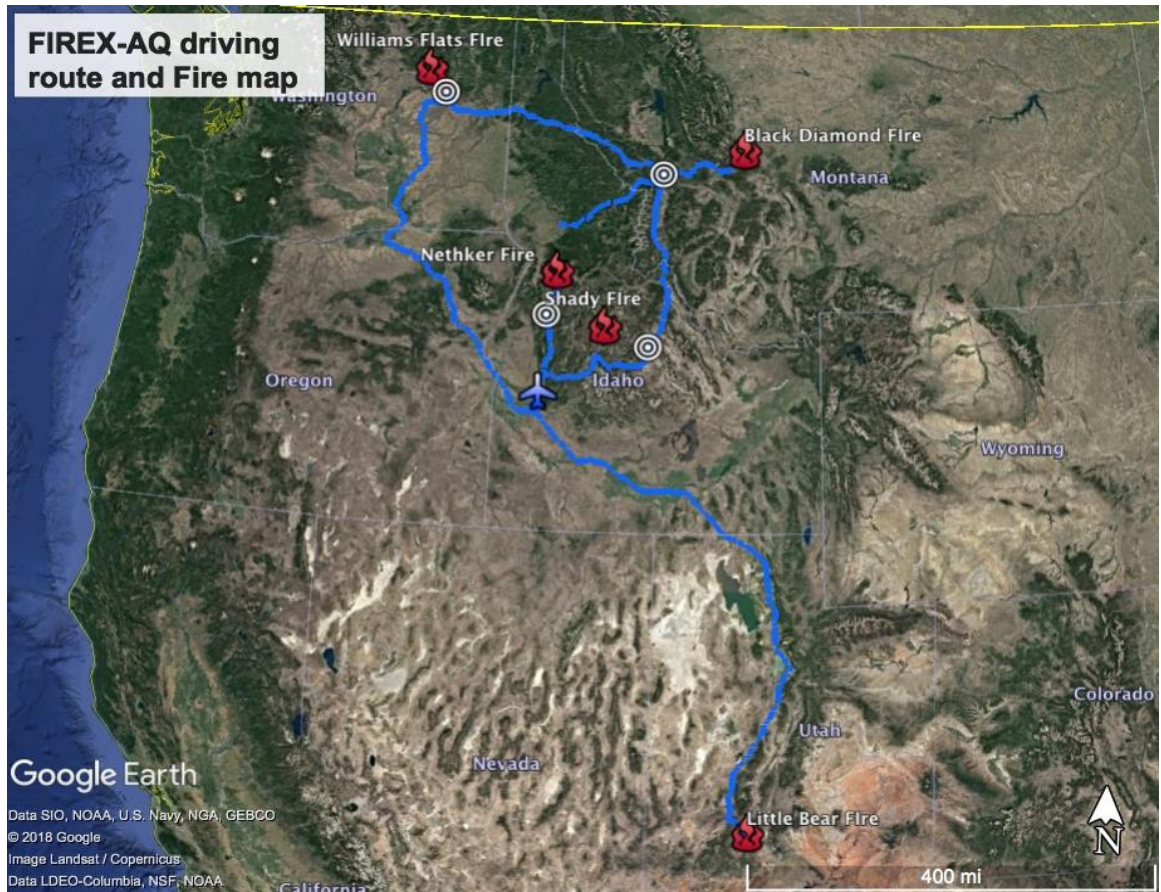


Figure S2. Driving map during FIREX-AQ. From Jul 24 to Aug 22 of 2019, we sampled smoke from five different fires including Shady fire (Idaho), Black Diamond fire (Montana), Williams Flats fire (Washington), Nethker fire (Idaho), and Little Bear fire (Utah).

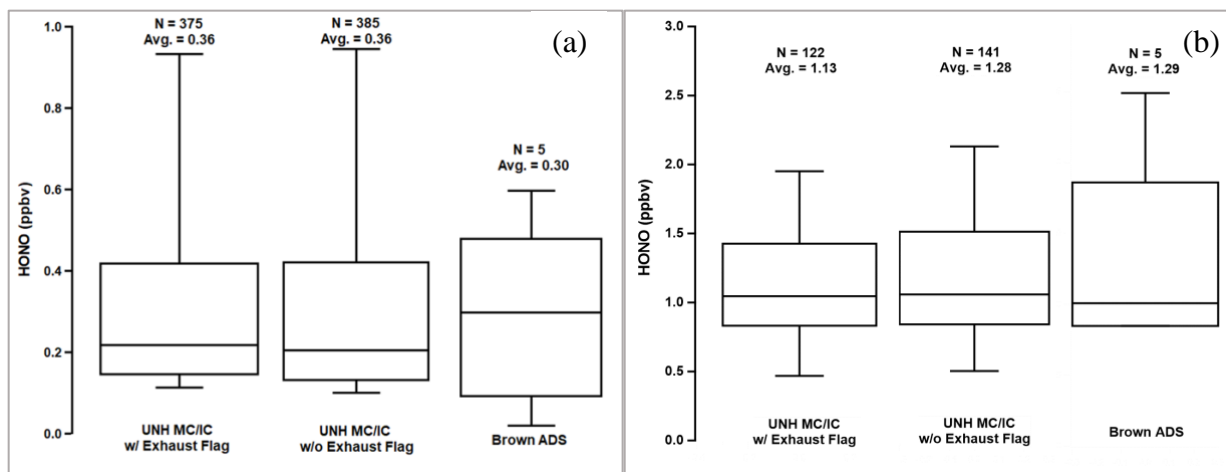


Figure S3. Comparison of HONO concentrations measured with MC/IC and that recovered from ADS collection for isotopic analysis for Williams Flats fire (a) and Nethker fire (b). The whiskers from bottom to top represent 10%, 25%, 50%, 75% and 90% quartiles.

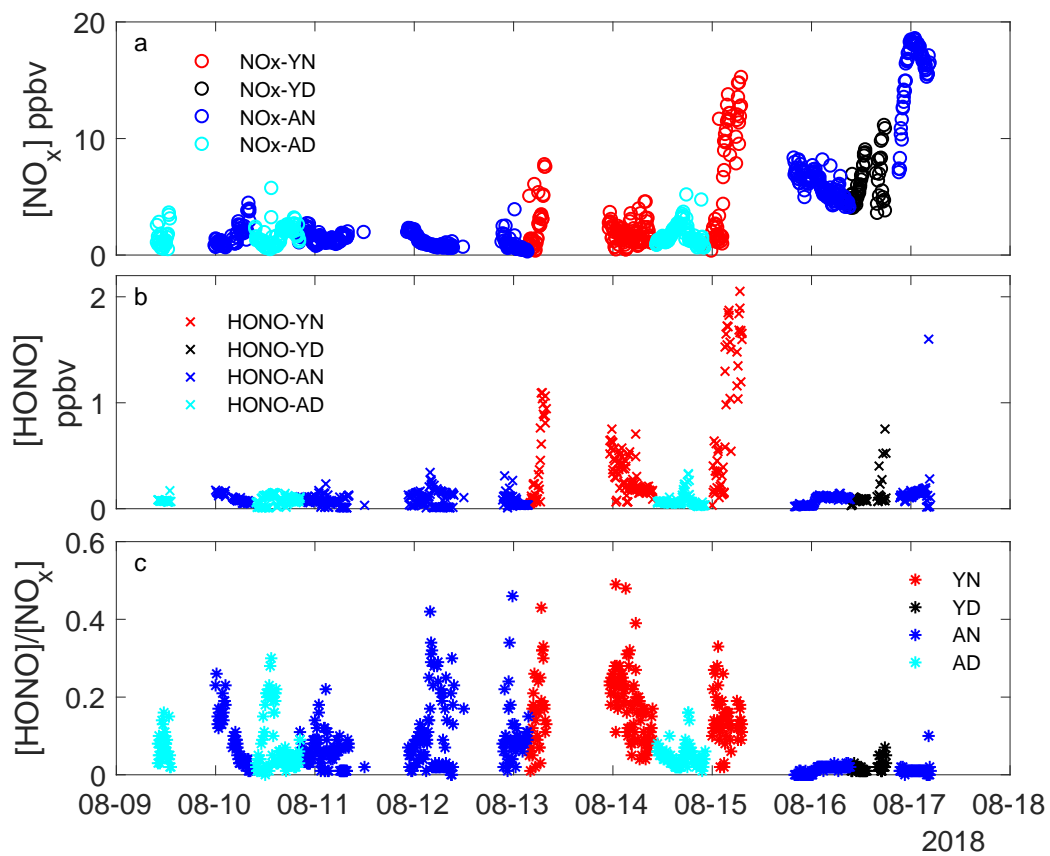


Figure S4. Time history of HONO/NO_x during WE-CAN 2018 sampling.

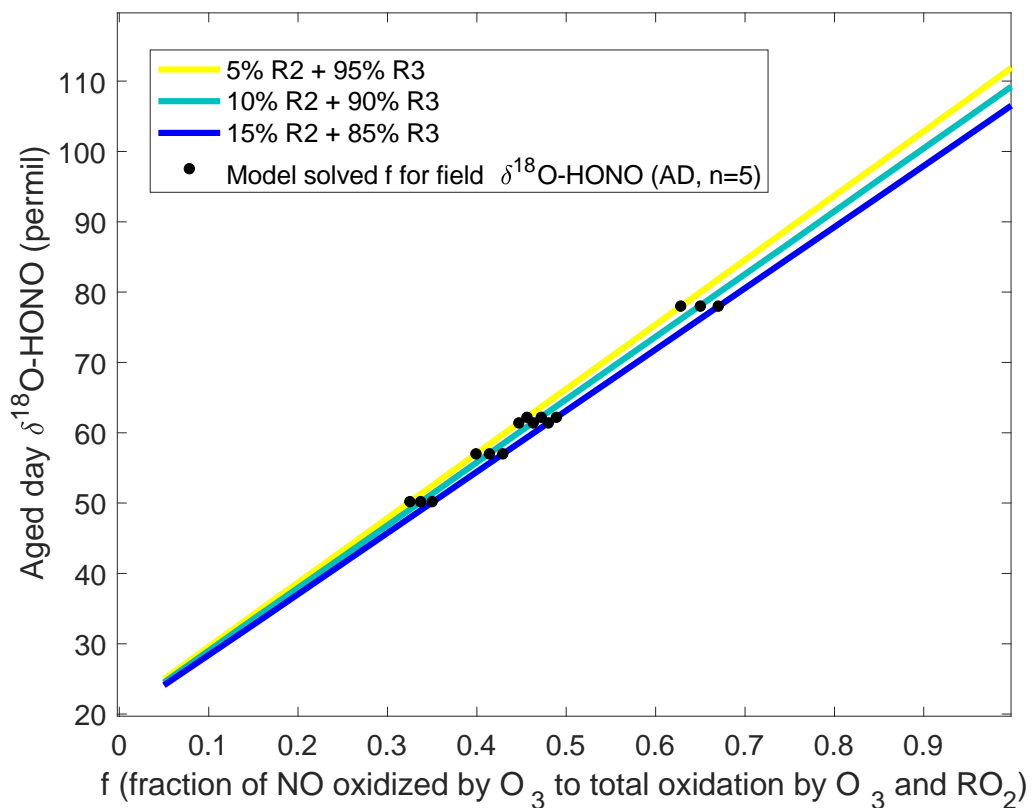


Figure S5. Modeling results for δ¹⁸O-HONO in aged daytime smoke as a function of fraction of NO oxidized to NO₂ via O₃ to that via O₃ and RO₂ together ($f_{O_3/(O_3+RO_2)}^{NO}$). The contribution of R2 to HONO production is varied from 5% to 15% following M1 in Fig. 4, and R3 accounts for the remaining secondary HONO contribution.

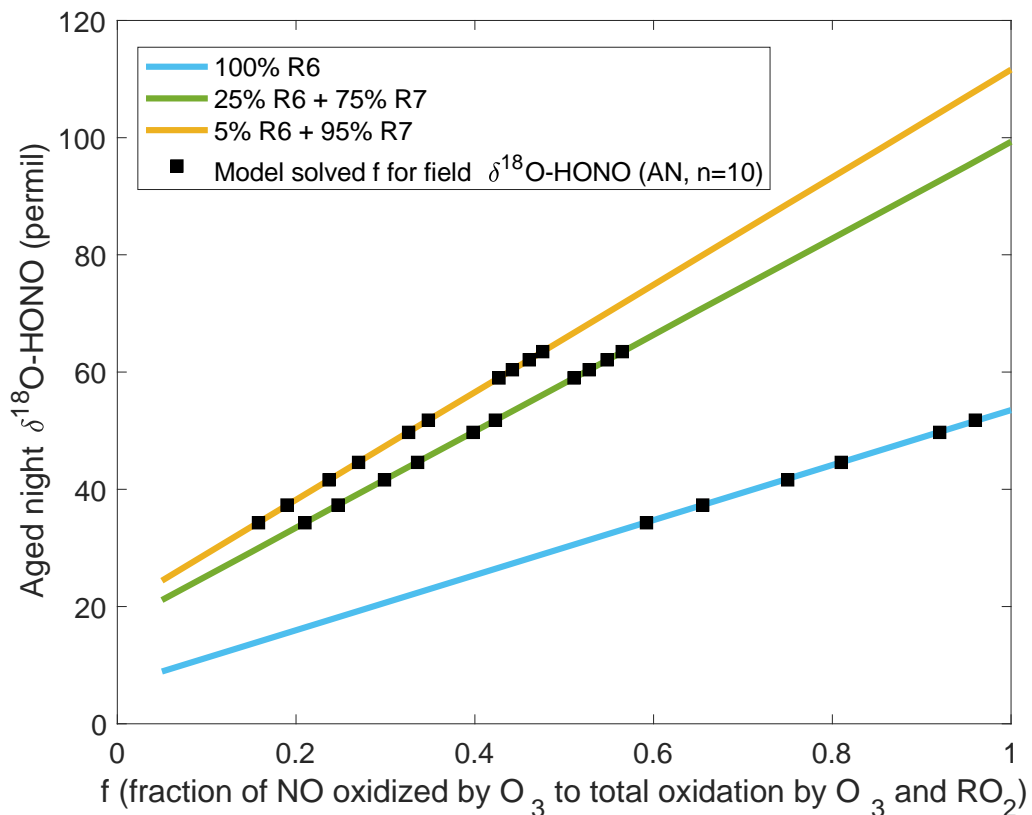


Figure S6. Modeling results for $\delta^{18}\text{O-HONO}$ in aged nighttime smoke as a function of fraction of NO oxidized to NO_2 via O_3 to that via O_3 and RO_2 together ($f_{\text{O}_3/(\text{O}_3+\text{RO}_2)}^{\text{NO}}$). Three scenarios with various contributions of R6 and R7 are modeled. Black squares are model solved $f_{\text{O}_3/(\text{O}_3+\text{RO}_2)}^{\text{NO}}$ to each field-observed $\delta^{18}\text{O-HONO}$ under this condition.

Table S1. Secondary formation and sink reactions and their isotopic enrichment factors for ^{15}N and ^{18}O ($^{15}\epsilon$ and $^{18}\epsilon$) estimated in this work.

Condition	Reaction	$^{15}\epsilon$ ‰	$^{18}\epsilon$ ‰
day	$\text{NO} + \text{OH} \longrightarrow \text{HONO}$	+10	+15
day	$\text{HONO} + \text{h}\nu \longrightarrow \text{NO} + \text{OH} (\lambda < 400 \text{ nm})$	-2.5	-3.5
day	$\text{HONO} + \text{OH} \longrightarrow \text{NO}_2 + \text{H}_2\text{O}$	N/A	N/A
day	$\text{HNO}_3/\text{NO}_3^- + \text{h}\nu \longrightarrow \text{HONO} + \text{NO}_x$	≤ -47.9	6 to 13
day	organics + $\text{H}_2\text{O} + \text{NO}_2 + \text{h}\nu$	-2.9 to -4.5	-5.7 to -8.9
night	$\text{HONO} (\text{g}) \longrightarrow \text{HONO} (\text{ads})$	-2	-4
night	$2\text{NO}_2 + \text{H}_2\text{O} \longrightarrow \text{HNO}_3 + \text{HONO}$	-2.9 to -4.5	-5.7 to -8.9
night	$\text{NO}_2 + \text{H}_2\text{O} + \text{Fe}_x\text{O}_y$ (and/or quinone)	-2.9 to -4.5	-5.7 to -8.9

Table S2. Rate coefficients used in this work (Burkholder et al. 2019^a)

Reaction	Rate coefficient (298 K) cm ³ s ⁻¹
HO ₂ + NO → OH + NO ₂	8.0 × 10 ⁻¹²
NO + O ₃ → NO ₂ + O ₂	1.9 × 10 ⁻¹⁴
C ₂ H ₅ O ₂ + NO → product	8.7 × 10 ⁻¹²
CH ₃ O ₂ + NO → CH ₃ O + NO ₂	7.7 × 10 ⁻¹²
RO ₂ + NO → product	8.0 × 10 ⁻¹²
RO ₂ + NO → RONO ₂	3.2 × 10 ⁻¹³

^a Burkholder, J. B., Sander, S. P., Abbatt, J., Barker, J. R., Cappa, C., Crounse, J. D., Dibble, T. S., Huie, R. E., Kolb, C. E., Kurylo, M. J., Orkin, V. L., Percival, C. J., Wilmouth, D. M. and Wine P. H.: "Chemical Kinetics and Photochemical Data for Use in Atmospheric Studies, Evaluation No. 19," JPL Publication 19-5, Jet Propulsion Laboratory, Pasadena, 2019.

Table S3. Modeling results for $\Delta\delta^{15}\text{N}_{\text{HONO-NO}_x}$ (‰).

Remaining HONO	$\Delta\delta^{15}\text{N}_{\text{HONO-NO}_x}$ (5% R2+95% R3)	$\Delta\delta^{15}\text{N}_{\text{HONO-NO}_x}$ (10% R2+90% R3)	$\Delta\delta^{15}\text{N}_{\text{HONO-NO}_x}$ (15% R2+85% R3)	$\Delta\delta^{15}\text{N}_{\text{HONO-NO}_x}$ (10% R2+85% R3 +5% R4)	$\Delta\delta^{15}\text{N}_{\text{HONO-NO}_x}$ (10% R2+80% R3 +10% R4)	$\Delta\delta^{15}\text{N}_{\text{HONO-NO}_x}$ (10% R2+75% R3 +15% R4)
0.9	-1.8	-1.2	-0.5	-3.4	-5.6	-7.8
0.8	-1.6	-0.9	-0.3	-3.1	-5.3	-7.5
0.7	-1.3	-0.6	0.0	-2.8	-5.0	-7.2
0.6	-0.9	-0.3	0.4	-2.5	-4.7	-6.9
0.5	-0.5	0.1	0.8	-2.1	-4.3	-6.5
0.4	0.0	0.7	1.3	-1.5	-3.7	-5.9
0.36	0.2	0.9	1.5	-1.3	-3.5	-5.7
0.32	0.5	1.2	1.8	-1.0	-3.2	-5.4
0.28	0.8	1.5	2.1	-0.7	-2.9	-5.1
0.24	1.2	1.8	2.5	-0.4	-2.6	-4.8
0.2	1.6	2.2	2.9	0.0	-2.2	-4.4
0.16	2.1	2.8	3.4	0.6	-1.6	-3.8
0.12	2.8	3.4	4.1	1.2	-1.0	-3.2
0.08	3.7	4.3	5.0	2.1	-0.1	-2.3
0.04	5.3	5.9	6.6	3.7	1.5	-0.7
0.02	6.8	7.5	8.1	5.3	3.1	0.9

Table S4. Modeling results for $\delta^{18}\text{O}$ aged daytime and aged nighttime (%).

	Day			Night		
	$\delta^{18}\text{O}$ - HONO (5%R2+ 95%R3)	$\delta^{18}\text{O}$ - HONO (10%R2+ 90%R3)	$\delta^{18}\text{O}$ - HONO (15%R2+ 95%R3)	$\delta^{18}\text{O}$ - HONO (100%R6)	$\delta^{18}\text{O}$ - HONO (25%R6+ 75%R7)	$\delta^{18}\text{O}$ - HONO (5%R6+ 95%R7)
0.05	25.0	24.5	24.1	8.9	21.1	24.4
0.15	34.2	33.5	32.7	13.6	29.3	33.6
0.25	43.3	42.4	41.4	18.3	37.6	42.8
0.35	52.5	51.3	50.1	23.0	45.8	52.0
0.429	59.8	58.4	57.0	26.7	52.3	59.2
0.592	74.7	72.9	71.2	34.3	65.7	74.2
0.65	80.0	78.1	76.2	37.1	70.5	79.5
0.75	89.2	87.0	84.9	41.8	78.7	88.7
0.85	98.3	96.0	93.6	46.5	86.9	97.8
0.95	107.5	104.9	102.3	51.2	95.1	107.0
1	112.1	109.4	106.7	53.5	99.3	111.6