



Supplement of

Inverse modeling of SO_2 and NO_x emissions over China using multisensor satellite data – Part 1: Formulation and sensitivity analysis

Yi Wang et al.

Correspondence to: Jun Wang (jun-wang-1@uiowa.edu) and Yi Wang (yi-wang-4@uiowa.edu)

The copyright of individual parts of the supplement might differ from the CC BY 4.0 License.

20 Averaging kernel impacts on data assimilation



Figure S1. OMPS SO₂ Vertical Column Density (VCD) retrievals in Arpil 2018. (a) and (b) are operational VCDs and the VCDs that are modified through averaging kernel according to formula S1, respectively. (c) is the differences between the modified and operational VCDs. (d) is scatter plot of modified VCDs versus operational VCDs. Linear correlation coefficient (R), linear regression equation, root mean squared error (RMSE), normalized mean bias (NMB), mean bias (MB), and number of observations (N) are shown over the scatter plot.

30

$$\Omega_{v}^{new} = \Omega_{v}^{opl} (1 + (\mathbf{A} - \mathbf{1}) \cdot (\mathbf{S}^{opl} - \mathbf{S}^{new}))$$
(S1)

Equation S1 is used to calculate modified SO₂ (NO₂) vertical column density Ω_v^{new} . In equation S1, Ω_v^{opl} is 35 OMPS operational SO₂ (NO₂) VCD, *A* is averaging kernel vector, *I* is a vector with all elements of 1, S^{opl} is SO₂ (NO₂) is the shape factor that is used in OMPS trace gas retrieving process, and S^{opl} is SO₂ (NO₂) is shape factor from GEOS-Chem simulation of this paper.

40



Figure S2. OMPS NO₂ Vertical Column Density (VCD) retrievals in October 2013. (a) and (b) are operational VCDs and the VCDs that are modified through averaging kernel according to formula S1, respectively. (c) is the differences between the modified and operational VCDs. (d) is scatter plot of modified VCDs versus operational VCDs. Linear correlation coefficient (R), linear regression equation, root mean squared error (RMSE), normalized mean bias (NMB), mean bias (MB), and number of observations (N) are shown over the scatter plot.

The iteration halt criterion



55



Figure S3. Normalized cost function (the ratio of the cost function at a iteration to that at the 1^{st} iteration) for E-SO₂ (a) and E-NO₂ (b). A iteration is accepted (solid circle) if its cost function value is smaller than that of any previous iterations, otherwise not accepted (empty circle). The 1^{st} iteration (prior) is defined as not accepted. The iterations that are selected based on the halt criterion are marked with red cross.

We choose less than 3% reduction in the cost function between two successful iterations as a criterion to halt the minimization. This selection is to expediate the computation while still maintain the similar accuracy for the optimization; further tests show that the more iterations
60 (after <3% reduction of cost function) doesn't yield discernable difference in the cost function values (Fig. S3) and optimization results (Table S1 and S2).

Iteration	1 (prior)	5 (selected)	6	7	8	9	10	11
SO ₂ [Gg S]	1166	748	744	Not accepted ^a	746	Not accepted	749	Not accepted

Table S1. Anthropogenic SO₂ emissions for October 2013 from E-SO₂ at each iteration.

65 ^aPosterior emission total amount at the iteration that is not accepted (cost function value is not smaller than that of any previous iterations) is not shown.

Iteration	1 (prior)	6 (selected)	7	8	9	10	11	12
NO _x [Gg N]	714	672	667	Not acceptedª	666	666	666	Not accepted

Table S2. Anthropogenic NO_x emissions for October 2013 from E-NO₂ at each iteration.

^aPosterior emission total amount at the iteration that is not accepted (cost function value is not smaller than that

70 of any previous iterations) is not shown.

Averaging kernel (scattering weight) impacts on evaluation



80

75

Figure S4. Taylor diagrams for comparing of VCDs of SO_2 (a) and NO_2 (b) from the GEOS-Chem simulations (squares for prior, triangles for posterior E-SO₂ (a) or E-NO₂ (b), and diamonds for E-joint) with that from the OMI (label 1 for operational level 3 SO₂ (a) or level 3 NO₂ (b) and label 2 for the level 3 SO₂ (a) that are modified by considering the vertical profiles from the GEOS-Chem simulation with which is to be compared or the level 2 NO₂ (b) that are modified by considering the vertical profiles from the GEOS-Chem simulation with which is to be compared) in October 2013 over China.

$$\Omega_s^{opl} = \Omega_v^{opl} \cdot M^{opl} \tag{S2}$$

$$M^{new} = \boldsymbol{W} \cdot \boldsymbol{S}^{new} \tag{S3}$$

$$\Omega_{v}^{opl} = \frac{\Omega_{s}^{opl}}{M^{new}} \tag{S4}$$

85

Equation S2 is used to convert OMI SO₂ (NO₂) vertical column density Ω_{v}^{opl} to SO₂ (NO₂) slant column density Ω_{s}^{opl} by multiplying SO₂ (NO₂) air mass factor M^{opl} from OMI product. Equation S3 is used to calculate new SO₂

(NO₂) air mass factor M^{new} , where W is SO₂ (NO₂) scattering weight, and S^{new} is SO₂ (NO₂) shape factor that from the GEOS-Chem simulation with which is to be compared. Equation S4 is used to calculate new OMI SO₂ (NO₂) vertical column density Ω_v^{opl} .

NO₂ lifetime

95

105



Figure S5. Relative change of GEOS-Chem NO₂ VCDs when NH₃ emissions reduce to 50% (a) and 20% (b), respectively at OMPS overpassing time, and a dataset of climatological aerosol surface area is used, regardless of scenarios of NH₃ emissions. (c) and (d) are similar to (a) and (b), respectively, but a dataset of climatological AOD rather than climatological aerosol surface area is used.

Figure S5 a and b show the relative change GEOS-Chem NO₂ VCDs with NH₃ emissions reducing to 50% and 20%, respectively, in the situation that a dataset of climatological aerosol surface area is used, regardless of scenarios of NH₃ emissions, hence reflecting the impact of photolysis of O₃ and NO₂ on NO₂ lifetime when NH₃ emissions are reduced. The situation that a dataset of climatological AOD is used is shown in Fig. S4 c and d ; thus they are the impact of N₂O₅ chemistry on NO₂ lifetime. Apparently, the impact of photolysis of O₃ and NO₂

thus they are the impact of N₂O₅ chemistry on NO₂ lifetime. Apparently, the impact of photolysis of O₃ and N on NO₂ lifetime caused by the reduction of NH₃ emissions is negligible compared to that of N₂O₅ chemistry.