



Supplement of

The promotion effect of nitrous acid on aerosol formation in wintertime in Beijing: the possible contribution of traffic-related emissions

Yongchun Liu et al.

Correspondence to: Yongchun Liu (liuyc@buct.edu.cn), Weigang Wang (wangwg@iccas.ac.cn), and Markku Kulmala (markku.kulmala@helsinki.fi)

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1 Supplement information

Non-refractory PM_{2.5} (NR-PM_{2.5}) measurement. Concentration of NR-PM_{2.5} was 2 measured with a ToF-ACSM (Aerodyne Co. Ltd., USA). The operation protocol and 3 the configuration of ToF-ACSM has been described well in previous work (Fröhlich et 4 al., 2013). Namely, PM_{2.5} particles from the inlet were focused by a PM_{2.5} aerodynamic 5 lens (Williams et al., 2013), and then vaporized by a standard vaporizer heated at 600 6 °C followed by electronic ionization (EI, 70 eV). The non-refractory components 7 including chloride, nitrate, sulfate, ammonia and organics were measured using a time-8 9 of-flight mass spectrometer with unit mass resolution (UMR). The concentrations of the above species were calculated based on the measured fragments signals, the signal 10 ions (SI), the fragment table, the measured ionization efficiency (IE) of nitrate and the 11 12 corresponding relative ionization efficiency (RIE) for sulfate, chloride, ammonia and organics. IE calibration of nitrate was performed using 300 nm dry NH4NO3 every 13 month during this observation study. 14

15 VOCs measurement. VOCs were measured using a Single Photo Ionization Time-offlight Mass spectrometer (SPI-ToF-MS 3000R, Hexin Mass Spectrometry). 0.8 L min⁻ 16 ¹ of filtered air was sucked from the whole sampling tube and heated to 80 °C in the 17 inlet. VOCs were selectively enriched continuously through a polydimethylsiloxane 18 (PDMS) membrane, and then ionized by VUV light (10.5 eV) with a deuterium lamp. 19 The concentration of VOCs was determined with the time-of-flight mass spectrometer 20 (ToF-MS) based upon external standard curves of PAMS and TO-15 standard gases 21 (Linde Electronics & Specialty Gases, USA). VOCs with m/z from 40 to 300 were 22

recorded with 3 min of time resolution, while hourly averaged concentration werereported in this work. Calibration was performed every week.

25 HONO measurement. HONO in ambient air directly sampled from the window of the laboratory was absorbed by a solution containing 0.06 mol L⁻¹ sulfnilamide in 1 mol L⁻ 26 ¹ HCl, and then transformed into an azo dye by N-(1-naphthyl) ethylene-27 diaminedihydrochloride (0.8 mml L⁻¹). The azo dye was pumped into Teflon absorption 28 cells (Liquid Core Waveguide, LCW) and detected by a mini-spectrometer with a diode 29 array detector (Ocean Optics, SD2000). The HONO concentrations was obtained by 30 31 subtracting the calibrated signal of the second coil from the first coil using external nitrile standard solutions. Zero point calibration was performed every day using 32 scrubbed zero air (Tong et al., 2016). 33

34 **Photolysis rate constants of HONO and O₃**. Photolysis rate constants of $NO_2(J_{NO2})$, HONO(J_{HONO}) and O₃(J_{O3}) under clear sky conditions were calculated according to the 35 solar zenith angle and the location using a box model (FACSIMILE 4). NO₂ photolysis 36 37 sensor (J_{NO2}, Metcon) was unavailable, while UVB is always available during our observation study. However, it was available from Aug 17 to Sep 16, 2018. A calibration 38 function between the measured UVB light intensity and J_{NO2} was established to correct 39 the influence the climatological O₃ column, aerosol optical depth and cloud cover on 40 surface UV light intensity from Aug 17 to Sep 16, 2008. As shown in Figure S10, the 41 model well predicted the J_{NO2} . Then the J_{NO2} during this campaign study was predicted 42 using the model. We further confirmed the calculated J_{NO2} by comprising the OH 43 concentration estimated by the J_{O1D} according to the equation ($c_{OH}=J_{O1D}\times 2\times 10^{11}$ 44

molecules cm⁻³) (Tan et al., 2019) and the measured OH concentration at Huairou,
which is 60 km northeast from BUCT, form Jan 11 to Mar 10, 2016. As shown in Figure
S10C, the estimated diurnal curve of OH is comparable with that measured at Huairou.
Fig. S7 shows the calculated photolysis rates.

50 Supplementary figures



52 Figure S1. Location of AHL/BUCT observation station. The map was

made from Wemap and © Google Earth.





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55 Figure S2. Hourly averaged (A)-(F) concentration of pollutants from Feb 1 to Jun 30,

56 2018.





58 Figure S3. The monthly cumulative frequency of PM_{2.5} and HONO and the monthly

59 mean concentration of $PM_{2.5}$ and HONO.



Figure S4. (A)-(B) monthly Windrose-PBL plots, and monthly averaged (C) UVB
intensity, mass concentration and (D) fraction of individual component in NR-PM_{2.5}
composition from Feb to Jun, 2018.



Figure S5. Transport of air mass during Chinese New Year based on back trajectory
analysis (A) at 100 and (B) 500 m height; (C) Diurnal variation of NR-PM_{2.5} normalized
to CO concentration from Feb 1 to March 31; (D) Hourly averaged wind speed variation
in the 12th episode; (E) Correlation of the concentration increment of individual
component and consumed HONO normalized to CO in the daytime.







Figure S6. Distribution of wind speed when the PM_{2.5} concentration was larger than 50







Figure S8. Correlation of the charge between inorganic anions and cations in non-

83 refractory PM_{2.5} in Beijing.

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Figure S9. Correlation of measured HONO concentration with NO_x concentration.



Figure S10. (A) Measured and predicted J_{NO2} and (B) the correlation between measured and predicted J_{NO2} from Aug. 15 to Sep. 16; (C) calculated diurnal curve of OH concentration based on J_{O1D} compared with that measured at Huairou (60 km northeast from BUCT) from Jan 11 to Mar 10, 2016; (D) OH concentrations estimated using $c_{OH}=J_{O1D}\times2\times10^{11}$ (Tan et al., 2019).





Supplementary tables

| | Fraction of NR-PM _{2.5} (%) | | | | |
|-----------|--------------------------------------|-----------------|-----------------|-----------------|-----------------|
| Component | or Concentration of gaseous | Feb | Mar | Apr | May |
| | pollutants (ppbv) | | | | |
| | Feb (12.2±2.9) | | | | |
| | Mar (14.2±2.8) | Significant | | | |
| Ammonium | Apr (14.0±4.0) | Significant | Not significant | | |
| | May (11.6±4.6) | Not significant | Significant | Significant | |
| | Jun (12.2±5.2) | Not significant | Significant | Significant | Not significant |
| | Feb (7.7±6.1) | | | | |
| | Mar (4.4±2.6) | Significant | | | |
| Chloride | Apr (1.1±1.2) | Significant | Significant | | |
| | May (0.7±1.1) | Significant | Significant | Not significant | |
| | Jun (0.3±0.2) | Significant | Significant | Significant | Not significant |
| | | | | | |
| | Mar (26.7±8.8) | Significant | | _ | |
| Nitrate | | | | | _ |
| | May (17.3±11.8) | Not significant | | | |
| | Jun (16.7±12.8) | Not significant | Significant | Significant | Not significant |
| | Feb (47.9±10.7) | | | | |
| | Mar (45.9±10.2) | Not significant | | | |
| Organic | Apr (46.5±14.2) | Not significant | Significant | | _ |
| | May (52.9±17.0) | Not significant | Significant | Significant | |
| | Jun (52.6±18.7) | Significant | Significant | Significant | Not significant |

Table S1. ANOVA statistics analysis for the monthly mean fraction of the individual component in NR-PM_{2.5} and HONO concentration.

| | Feb (16.0±9.1) | | | | |
|---------|-------------------|-----------------|-----------------|-----------------|-----------------|
| | Mar (8.8±5.4) | Significant | | | |
| Sulfate | Apr (16.4±8.2) | Not significant | Significant | | |
| | May (17.5±6.6) | Significant | Significant | Not significant | |
| | Jun (18.2±8.0) | Significant | Significant | Significant | Not significant |
| | Feb (3.0±2.8) | | | | |
| | Mar (4.6±3.1) | Significant | | | |
| BC | Apr (3.2±2.6) | Not significant | Significant | | |
| | May (2.8±2.1) | Not significant | Significant | Not significant | |
| | Jun (2.6±1.5) | Significant | Significant | Significant | Not significant |
| | Feb (0.73±0.70) | | | | |
| | Mar (1.53±1.25) | Significant | | | |
| HONO | Apr (1.38±1.35) | Significant | Not significant | | |
| | May (1.31±1.00) | Significant | Significant | Not significant | |
| | Jun (1.35±0.80) | Significant | Significant | Not significant | Not significant |
| | Feb (20.4±17.3) | | | | |
| | Mar (40.5±24.0) | Significant | | | |
| NOx | Apr (22.8±18.6) | Not significant | Significant | | |
| | May (25.0±15.9) | Significant | Significant | Not significant | |
| | Jun (19.0±12.1) | Not significant | Significant | Significant | Significant |
| | Feb (3.8±3.3) | | | | |
| | Mar (12.1±13.0) | Significant | | | |
| SO_2 | Apr (2.8±2.4) | Significant | Significant | | |
| | May (1.8±1.7) | Significant | Significant | Not significant | |
| | Jun (1.3±1.2) | Significant | Significant | Significant | Not significant |
| СО | Feb (959.6±554.6) | | | | |

| | Mar (1075.0±571.8) Apr (546.6±378.1) | Significant Significant | Significant | | |
|----------------|---|----------------------------|-------------|-----------------|-----------------|
| | May (554.1±336.9) | Significant | Significant | Not significant | |
| | Jun (583.4±286.2) | Significant | Significant | Not significant | Not significant |
| | Feb (22.6±14.6) | | | | |
| | Mar (23.8±19.2) | Not significant | | | |
| O ₃ | Apr (43.5±29.0) | Significant | Significant | | |
| | May (42.5±28.3) | Significant | Significant | Not significant | |
| | Jun (57.2±30.7) | Significant | Significant | Significant | Significant |

Note: "Significant" or "Not significant" denotes that the difference of the monthly mean fractions or concentrations is significant or not significant at the 0.05 level.

| Enicodo | | UONO | Average | NR-PM _{2.5} Concentration (%) | | | | | | | | | |
|---------|-----------|-----------------|-------------------|--|-----------------------|----------|------------------|----------------|------------------|----------------|------------------|----------------|------------------|
| Episode | Duration | (null) | PM _{2.5} | Ch | loride | ١ | Nitrate | O | rganic | Sulp | ohate | Amr | nonium |
| 100. | | (ррв) | concentration | (%) | (µg m ⁻³) | (%) | $(\mu g m^{-3})$ | (%) | $(\mu g m^{-3})$ | (%) | $(\mu g m^{-3})$ | (%) | $(\mu g m^{-3})$ |
| 1 | Feb 2-5 | 0.38 ± 0.28 | 9.3±4.5 | 4.0±2.3 | 0.26±0.39 | 12.3±5.6 | 0.80±1.17 | 51.1±10.0 | 2.68±3.00 | 20.6±9.2 | 0.69±0.24 | 12.0±3.2 | 0.54 ± 0.49 |
| 2 | Feb 8-9 | 0.90 ± 0.72 | 44.5±3.5 | 6.3±2.9 | 1.59±1.46 | 15.8±7.9 | 4.20±3.87 | 49.9±4.8 | 9.63±7.64 | 17.3±8.8 | 2.31±1.42 | 10.8 ± 1.0 | 2.14±1.69 |
| 3 | Feb 10-12 | 0.31±0.40 | 9.0±0.8 | 5.2±3.5 | 0.18±0.22 | 6.8±3.9 | 0.30 ± 0.44 | 48.6±10.6 | 1.75±1.72 | 28.1±11.5 | 0.74 ± 0.38 | 11.2±2.5 | 0.35 ± 0.23 |
| 4 | Feb 16-19 | 1.38 ± 0.86 | 101.5±26.8 | 15.5±4.2 | 9.04±4.94 | 25.0±4.1 | 13.15±7.73 | 32.2±3.8 | 18.21±8.25 | 14.4±3.7 | 7.82±4.39 | 12.9±1.5 | 6.85±3.78 |
| 5 | Feb 21-24 | 0.64 ± 0.58 | 24.3±7.0 | 5.5±4.1 | 0.60±0.51 | 14.9±6.3 | 1.80 ± 1.38 | 56.3±10.0 | 5.83±2.94 | 11.8 ± 5.0 | 1.17±0.67 | 11.6±2.8 | 1.24 ± 0.77 |
| 6 | Feb 25-28 | 0.87 ± 0.64 | 108.8±42.9 | 5.2±1.4 | 2.94±1.97 | 27.1±3.9 | 15.3±8.77 | 42.5±6.8 | 22.83±9.68 | 10.4±3.8 | 6.44±5.78 | 14.7±1.8 | 8.34±5.30 |
| 7 | Mar 2-3 | 1.41 ± 0.84 | 120.0±47.0 | 8.3±2.2 | 4.23±1.72 | 26.5±4.8 | 15.29±9.44 | 44.4±6.2 | 23.40±10.49 | 7.2±1.9 | 4.36±3.37 | 13.5±1.9 | 7.74±4.76 |
| 8 | Mar 8-10 | 1.36±0.89 | 88.7±34.2 | 4.8±1.8 | 1.87±1.09 | 28.3±5.2 | 11.00±6.20 | 43.0±7.0 | 15.65±7.15 | 9.0±2.8 | 3.10±1.42 | 14.9±2.0 | 5.58 ± 2.92 |
| 9 | Mar 11-14 | 2.27±1.68 | 170.3±75.4 | 3.5±0.9 | 2.48±1.32 | 34.8±4.3 | 28.32±19.09 | 36.8 ± 5.0 | 27.90±15.78 | 8.1±1.8 | 6.60±4.72 | 16.8±1.5 | 13.57±8.99 |
| 10 | Mar 16-19 | 1.88 ± 1.38 | 66.0±25.7 | 3.8±1.7 | 1.99±1.18 | 30.2±6.3 | 17.40±12.45 | 35.9±2.8 | 20.87±10.52 | 13.5±5.1 | 7.00±4.92 | 16.5±1.0 | 9.17±5.86 |
| 11 | Mar 21-23 | 1.41±0.72 | 83.7±22.1 | 5.3±2.8 | 2.54±2.30 | 31.5±3.8 | 12.23±5.22 | 45.1±6.7 | 18.02 ± 5.46 | 4.4±1.0 | 1.67±0.92 | 13.7±1.6 | 5.38 ± 2.08 |
| 12 | Mar 25-27 | 2.22±1.34 | 129.5±51.9 | 2.0±0.7 | 0.94±0.64 | 35.3±3.6 | 16.32±9.90 | 41.5±5.4 | 20.46±10.18 | 5.7±1.2 | 2.56±1.68 | 15.6±1.6 | 7.11±4.37 |

Tab. S2. Mean concentrations of HONO and PM_{2.5} in selected episodes

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Table S3. The summary of the HONO/NOx ratio from vehicles in this study and the reported emission ratio of HONO/NOx from vehicles in China.

| No. | Time | | ΔΝΟ/ΔΝΟχ | $R_{\Delta \rm NO/\Delta \rm NOx}$ | ΔΗΟΝΟ/ΔΝΟχ | $R_{\Delta \mathrm{HONO}/\Delta \mathrm{NOx}}$ |
|----------------------------------|-------------------------|---|---|------------------------------------|-------------------------|--|
| 1 | 2018/2/6 5:00-8:00 | | 1.00 | 0.99 | 1.3% | 0.92 |
| 2 | 2018/2/8 5:00-8:00 | | 0.94 | 0.99 | 1.8% | 0.96 |
| 3 | 3 2018/3/3 5:00-8:00 | | 0.98 | 0.99 | 2.4% | 0.96 |
| 4 | 2018/3/13 5:00- 8:00 | | 1.00 | 0.99 | 1.4% | 0.86 |
| 5 | 2018/4/15 5:00- 7:00 | | 0.82 | 0.82 0.97 | | 0.99 |
| | Me | ean | 0.95 ± 0.08 | - | 1.8±0.5% | - |
| Tim | | Place | Mathada | Δł | HONO/ANOx | Deference |
| 1 111 | le | Flace | Methods | Range | e Mean | - Kelelelice |
| 2015/9 2016/8 | /1- /31 | Ji'nan, Shandong | Empirical analysis of field data | s 0.19% 0.87% | 0.53±0.20% | (Li et al., 2018) |
| 2011/8/3- 2012/5/31 Hongko | | Hongkong | Empirical analysis of field data | s 0.5% 1.6% | 1.2±0.4% | (Xu et al., 2015) |
| 2015/3/11- 2015/3/21 Hongkong | | Tunnel experimen | ıt - | 1.24±0.35% | (Liang et al., 2017) | |
| 2014 Beiji | | Beijing | Tunnel experimen | ıt – | 2.1% | (Yang et al., 2014) |
| 2017 | 7 | Beijing | Chassis dynamometer test | 0.03% t 0.42% | 0.18% | (Liu et al., 2017) |
| 2016/12 2016/12 | 2/16- 2/24 | Beijing | Empirical analysis of field data | - | 1.3% | (Zhang et al., 2018) |
| 2016/12/7- 2016/12/13 Beijing | | Low limit correlation of field data | 1 - | 1.41% | (Meng et al., 2019) | |
| 2018/2 2018/6 | /1- /30 | Beijing | Low limit correlation of field data | 1 - | 1.17% | This study |
| 2018/2 2018/6 | /1- /30 | Beijing | Empirical analysis of field data | 1.3-2.4 | % 1.8±0.5% | This study |

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4 **References:**

5 Fröhlich, R., Cubison, M. J., Slowik, J. G., Bukowiecki, N., Prévôt, A. S. H., Baltensperger, U., Schneider,

6 J., Kimmel, J. R., Gonin, M., Rohner, U., Worsnop, D. R., and Jayne, J. T.: The ToF-ACSM: a portable

aerosol chemical speciation monitor with TOFMS detection, Atmos. Meas. Tech., 6, 3225-3241,
10.5194/amt-6-3225-2013, 2013.

9 Li, D., Xue, L., Wen, L., Wang, X., Chen, T., Mellouki, A., Chen, J., and Wang, W.: Characteristics and

10 sources of nitrous acid in an urban atmosphere of northern China: Results from 1-yr continuous

11 observations, Atmos. Environ., 182, 296-306, https://doi.org/10.1016/j.atmosenv.2018.03.033, 2018.

12 Liang, Y., Zha, Q., Wang, W., Cui, L., Lui, K. H., Ho, K. F., Wang, Z., Lee, S.-c., and Wang, T.: Revisiting

- 13 nitrous acid (HONO) emission from on-road vehicles: A tunnel study with a mixed fleet, J. Air Waste
- 14 Manage. Assoc., 67, 797-805, 10.1080/10962247.2017.1293573, 2017.
- 15 Liu, Y., Lu, K., Ma, Y., Yang, X., Zhang, W., Wu, Y., Peng, J., Shuai, S., Hu, M., and Zhang, Y.: Direct
- 16 emission of nitrous acid (HONO) from gasoline cars in China determined by vehicle chassis
- dynamometer experiments, Atmos. Environ., 169, 89-96, 10.1016/j.atmosenv.2017.07.019, 2017.
- 18 Meng, F., Qin, M., Tang, K., Duan, J., Fang, W., Liang, S., Ye, K., Xie, P., Sun, Y., Xie, C., Ye, C., Fu, P.,
- 19 Liu, J., and Liu, W.: High resolution vertical distribution and sources of HONO and NO2 in the nocturnal
- boundary layer in urban Beijing, China, Atmos. Chem. Phys. Discuss., 2019, 1-34, 10.5194/acp-2019613, 2019.
- 22 Tan, Z. F., Lu, K. D., Jiang, M. Q., Su, R., Wang, H. L., Lou, S. R., Fu, Q. Y., Zhai, C. Z., Tan, Q. W.,
- 23 Yue, D. L., Chen, D. H., Wang, Z. S., Xie, S. D., Zeng, L. M., and Zhang, Y. H.: Daytime atmospheric
- 24 oxidation capacity in four Chinese megacities during the photochemically polluted season: a case study
- 25 based on box model simulation, Atmos. Chem. Phys., 19, 3493-3513, 10.5194/acp-19-3493-2019, 2019.
- 26 Tong, S., Hou, S., Zhang, Y., Chu, B., Liu, Y., He, H., Zhao, P., and Ge, M.: Exploring the nitrous acid
- 27 (HONO) formation mechanism in winter Beijing: direct emissions and heterogeneous production in
- 28 urban and suburban areas, Faraday Discuss., 189, 213-230, 10.1039/c5fd00163c, 2016.
- 29 Williams, L. R., Gonzalez, L. A., Peck, J., Trimborn, D., McInnis, J., Farrar, M. R., Moore, K. D., Jayne,
- 30 J. T., Robinson, W. A., Lewis, D. K., Onasch, T. B., Canagaratna, M. R., Trimborn, A., Timko, M. T.,
- 31 Magoon, G., Deng, R., Tang, D., de la Rosa Blanco, E., Prevot, A. S. H., and Worsnop, D. R.:
- 32 Characterization of an aerodynamic lens for transmitting particles greater than 1 micrometer in diameter
- into the Aerodyne aerosol mass spectrometer, Atmos. Meas. Tech., 6, 3271-3280, 10.5194/amt-6-3271 2013, 2013.
- 35 Xu, Z., Wang, T., Wu, J., Xue, L., Chan, J., Zha, Q., Zhou, S., Louie, P. K. K., and Luk, C. W. Y.: Nitrous
- acid (HONO) in a polluted subtropical atmosphere: Seasonal variability, direct vehicle emissions and
- 37 heterogeneous production at ground surface, Atmos. Environ., 106, 100-109,
- 38 10.1016/j.atmosenv.2015.01.061, 2015.
- 39 Yang, Q., Su, H., Li, X., Cheng, Y., Lu, K., Cheng, P., Gu, J., Guo, S., Hu, M., Zeng, L., Zhu, T., and
- 40 Zhang, Y.: Daytime HONO formation in the suburban area of the megacity Beijing, China, Sci. China-
- 41 Chem., 57, 1032-1042, 10.1007/s11426-013-5044-0, 2014.
- 42 Zhang, W., Tong, S., Ge, M., An, J., Shi, Z., Hou, S., Xia, K., Qu, Y., Zhang, H., Chu, B., Sun, Y., and
- 43 He, H.: Variations and sources of nitrous acid (HONO) during a severe pollution episode in Beijing in
- 44 winter 2016, Scie. Total Environ., 648, 253-262, 10.1016/j.scitotenv.2018.08.133, 2018.
- 45