



Supplement of

Comparison of primary aerosol emission and secondary aerosol formation from gasoline direct injection and port fuel injection vehicles

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18 Data correction

19 Wall-loss correction as well as particle and gas dilution corrections were considered in 20 this study. The details of wall-loss correction are introduced by Du et al. (2017). The real-time instruments sampled from the chamber during the whole photo-oxidation experiment, and zero 21 22 air was added to maintain a constant pressure inside the chamber. This led to particle dilution 23 that the sampled particles would not be included in the subsequent measurement, and gas 24 dilution that the sampled gas would not participate in the subsequent photo-oxidation reaction 25 and SOA formation. The particle dilution corrected mass concentration $C_{corr,n+1}$ could be 26 calculated as:

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$$C_{corr,n+1} = C_{n+1} + \sum_{i=1}^{n} (k_{wall} \times C_i) + \sum_{i=1}^{n} (k_{dilu,i} \times C_i)$$
 (1)

where C_{n+1} was the measured particle mass concentration at time n+1, k_{wall} was the wall loss decay constant and $k_{dilu,i}$ was dilution ratio at time i.

30 Then the gas dilution was taken into consideration. The final particle mass concentration 31 $C_{final,n+1}$ could be calculated as:

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$$C_{final,n+1} = C_{corr,n+1} - C_1 + \sum_{i=1}^{n} (\sum_{i=1}^{n+1} k_{dilu,i}) \times (C_{corr,n+1} - C_{corr,n})$$
 (2)

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36 **References**

37 Du, Z., Hu, M., Peng, J., Guo, S., Zheng, R., Zheng, J., Shang, D., Qin, Y., Niu, H., Li, M.,

38 Yang, Y., Lu, S., Wu, Y., Shao, M., and Shuai, S.: Potential of secondary aerosol formation from

39 Chinese gasoline engine exhaust, Journal of environmental sciences, in press.





43 Figure S1. Chemical composition of secondary aerosol formed in the chamber experiment

- 44 (Experiment GDI-1).
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48 Figure S2. Speed profiles of NEDC and BJC driving cycle.

Specifications	Fuel
Density (g mL ⁻¹)	0.7
Rvp (kPa)	55.4
Aromatics (% v/v)	36.7
Olefin (% v/v)	15.4
Ethanol (% v/v)	0.01
Oxygen (% m/m)	0.02
Mn (mg kg ⁻¹)	< 0.1
Sulfur (mg kg ⁻¹)	6
T10 (°C)	55.4
T50(°C)	109.9
T90 (°C)	164.3
Fbp (°C)	194.4

50 Table S1 Details of the fuel used in the experiments.

Compound	Emission facto	Emission factor (ng kg-fuel ⁻¹)	
	GDI	PFI	
Napthalene	0.025	<0.0001	
1-Methylnaphthalene	< 0.0001	< 0.0001	
2-Methylnaphthalene	0.012	0.004	
2,6-Dimethylnaphthalene	0.006	0.003	
Acenaphthylene	0.012	0.009	
Acenapthene	< 0.0001	0.015	
Fluorene	0.003	0.105	
Methyl-fluorene	0.083	0.105	
Dibenzofuran	0.006	0.039	
Retene	0.009	0.011	
9-Methylanthracene	< 0.0001	0.013	
Phenanthrene	0.244	0.069	
Anthracene	0.048	0.018	
Fluoranthene	0.201	0.034	
Pyrene	0.246	0.029	
Methyl-fluoranthene	0.004	0.007	
Benzo[a]anthracene	0.006	0.036	
Chrysene	0.020	0.065	
Methyl-chrysene	< 0.0001	0.007	
Benzo[b]fluoranthene	0.034	0.147	
Benzo[k]fluoranthene	0.041	0.129	
Benzo[e]pyrene	0.028	0.051	
Benzo[a]pyrene	0.012	0.041	
Benzo[ghi]flouranthene	0.095	0.027	
Cyclopenta[cd]pyrene	< 0.0001	0.032	
Dibenzo[a,h]anthracene	< 0.0001	< 0.0001	

53 Table S2 The EFs of Particulate-phase PAHs from GDI and PFI vehicles.

Picene	< 0.0001	< 0.0001
Perylene	0.009	< 0.0001
Benzo[ghi]perylene	< 0.0001	< 0.0001
Indeno[1,2,3-cd]pyrene	< 0.0001	< 0.0001
Coronene	< 0.0001	< 0.0001
Sum PAHs	1.144	0.994