



Supplement of

NO_2 -initiated multiphase oxidation of SO_2 by O_2 on $CaCO_3$ particles

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10 S1. Influence of gas phase diffusion on reactive uptake coefficients

- 11 The Influence of the gas phase diffusion on reactive uptake coefficients was analyzed using the resistor
- 12 model described by Davidovits et al. (2006) and the references therein.

13
$$\frac{1}{\gamma} = \frac{1}{\Gamma_{diff}} + \frac{1}{\alpha} + \frac{1}{\Gamma_{sat} + \Gamma_{rxn}}$$
(1)

14 where Γ_{diff} is the transport coefficient in the gas phase, $1/\Gamma_{\text{diff}}$ is the resistance due to the diffusion in the

- 15 gas phase. Similarly, $1/\Gamma_{sat}$ and $1/\Gamma_{rxn}$ are the resistance due to liquid phase saturation and liquid phase
- 16 reaction, respectively. α is the mass accommodation coefficient of SO₂.
- 17 $1/\Gamma_{diff}$ can be determined using the following equation:

18
$$\frac{1}{\Gamma_{\rm diff}} = \frac{0.75 + 0.238 \,{\rm Kn}}{{\rm Kn}(1 + {\rm Kn})}$$
 (2)

19 where Kn is Knudsen number. Knudsen number is defined as

$$20 \quad \mathrm{Kn} = \frac{\lambda}{a} \; , \tag{3}$$

21 where λ is the mean free path of a molecule in the gas phase and a is the radius of the particle.

22 λ can be derived from

$$23 \qquad \lambda = \frac{3D_g}{c},\tag{4}$$

24 where D_g is the diffusion coefficient in the gas phase and c is the mean molecular velocity.

c is determined by

$$26 \qquad \mathbf{c} = \sqrt{\frac{8RT}{\pi M}} \tag{5}$$

27 where R is the gas constant, T is temperature, and M is the molecular mass of SO₂.

28 $1/\Gamma_{diff}$ was calculated to be 78 and $1/\gamma$ was calculated to be ~8.3×10⁴. $1/\Gamma_{diff}$ only accounted for <0.1%

29 of $1/\gamma$. Therefore, the reactive uptake of SO₂ in this study was not limited by gas phase diffusion.

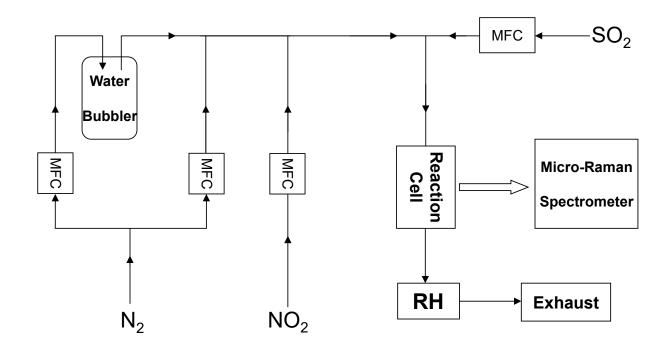
The same conclusion can also be drawn by calculating the gas phase diffusion correction factor for a reactive uptake coefficient according to the method in Pöschl et al. (2007) (Equation 20 in their study, also shown as equation 6 below).

33
$$C_g = \frac{1}{1 + \gamma \frac{0.75}{Kn}}$$
 (6)

34 where C_g is the gas phase diffusion correction factor for a reactive uptake coefficient.

35 References

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46 Figure S1. Schematic diagram of the experimental setup (Zhao et al., 2017). MFC: mass flow controller.

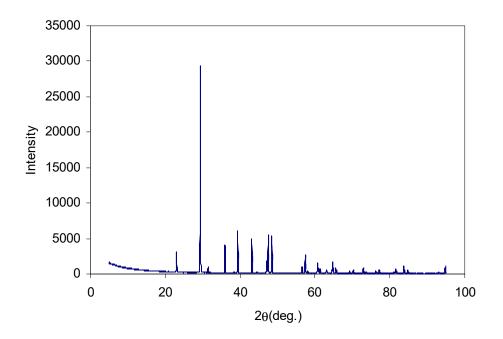


Figure S2. X-ray diffraction spectra of CaCO₃ particles.