



*Supplement of*

**Principal component analysis of summertime ground site measurements in the Athabasca oil sands with a focus on analytically unresolved intermediate-volatility organic compounds**

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## Descriptions of instrumentation used

A Griffin 450 gas chromatograph equipped with a cylindrical ion trap mass spectrometer and electron impact ionization (GC-ITMS) was used to quantify selected VOCs including o-xylene, decane, undecane, 1,2,3- and 1,2,4-trimethylbenzene (TMB), and several monoterpenes (i.e.,  $\alpha$ -pinene,  $\beta$ -pinene and limonene). The GC-ITMS primary responsibility was the quantification of monoterpenes. The remaining VOCs quantified were chosen because (a) they sufficiently resolved on the analytical column, and (b) response factors could be determined, either because the compounds of interest were part of the VOC standard used in the field (such as the aromatics o-xylene, 1,2,3- and 1,2,4-TMB, see below) or relative response factors were determined post-campaign. Operation, calibration and performance of this instrument have been described elsewhere (Tokarek et al., 2017; Liggio et al., 2016). Briefly, the GC was operated with 30 m (length)  $\times$  0.25 mm (inner diameter)  $\times$  0.25  $\mu$ m (film thickness) DB-5MS analytical column with helium carrier gas. The GC-ITMS sampled from a 3.6 m long stainless-steel inlet with an o.d. of 0.635 cm from a height of 5 m above ground. A 1 m long section of the inlet was heated to 110 °C and optimized to remove interference due to O<sub>3</sub> while avoiding decomposition of alkenes (Tokarek et al., 2017). The GC oven was programmed as follows: hold at 40 °C for 3.00 min, heat at 1.5 °C min<sup>-1</sup> to 70° C (reached at 23.00 min), heat at 5° C min<sup>-1</sup> to 200 °C (reached at 49.00 min) and hold for 4 min (total 53.00 min). This was followed by a 5 min recovery time to allow the oven and pre-concentration trap to cool back to 40 °C. The ion trap mass spectrometer was set to an  $m/z$  range of 50-425. After data reduction, the GC-ITMS generated 10-minute average concentrations of each VOC quantified every hour.

During the campaign, the GC-ITMS was calibrated in the field using an IONICON VOC standard (Table S-1) containing (in addition to VOCs that the GC-ITMS did not detect)  $\alpha$ -pinene and o-xylene at mixing ratios of  $\sim$  1 ppmv and an uncertainty of 5% and 6%, respectively. A commercial calibrator assembly (IONICON, GCU Standard) was used to deliver diluted calibration mixtures. The instrument responses to

the VOC standards were linear ( $R^2 > 0.99$ ). The GC-ITMS was calibrated for other VOCs offline relative to  $\alpha$ -pinene. In the field, there was no noticeable carry-over (i.e., memory effects) of IVOCs, which was occasionally evaluated by flooding the inlet with purified, VOC-free air.

Matrices of ions plotted against retention times for the total ion chromatograms (shown in Figure 2 in the main manuscript) are shown in Fig. S-1. In both cases, the greatest intensity is with masses are associated with alkanes (i.e.,  $m/z$  55, 57, 67, 69, etc.).

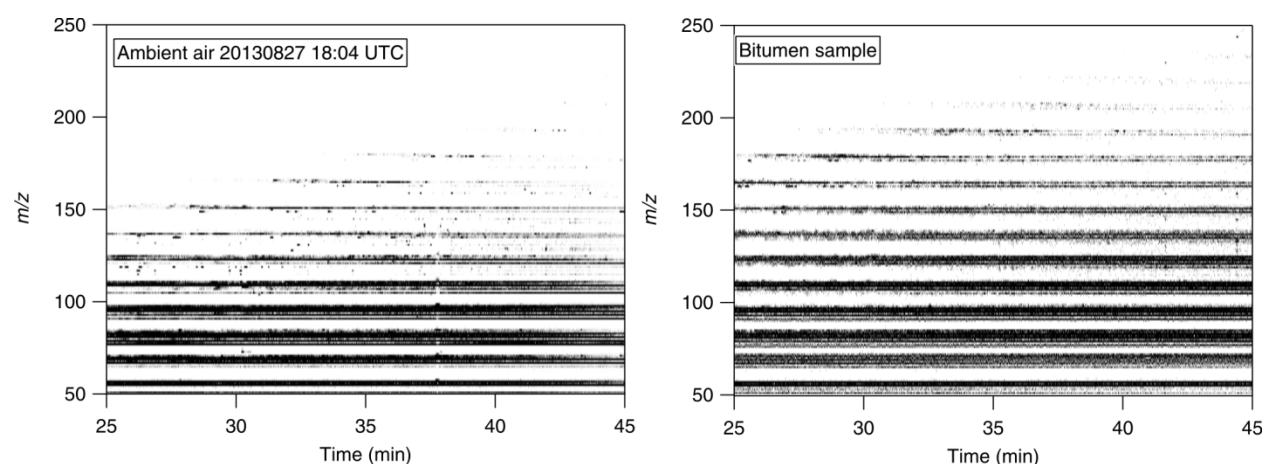


Figure S-1. Scatter of ions as a function of retention time for the total ion chromatograms shown in Figure 2 of the main manuscript. Darker pixels represent a higher intensity than lighter pixels.

Mixing ratios of carbon monoxide (CO), carbon dioxide (CO<sub>2</sub>) and methane (CH<sub>4</sub>) in ambient air were quantified using a commercial cavity ring-down spectrometer (Picarro G2401) (Nara et al., 2012; Chen et al., 2013). Ambient air was sampled from a height of 10 m through 0.635 cm outer diameter (o.d.) perfluoroalkoxyalkane (PFA) Teflon™ tubing and a 47 mm diameter, 1  $\mu$ m pore filter at a flow rate of  $\sim 0.5$  L min<sup>-1</sup>. A scrubber (MgClO<sub>4</sub>) was installed at the base of the sample line to remove water from the air. Operating procedures developed for Canada's greenhouse gas network of monitors across all stations in Canada by the Climate Division of ECCC were followed (ECCC, 2013a). The cavity ring-down spectrometer was calibrated every few days with calibrated standard gas mixtures (Scott-Marrin); a

96 target background mixture ( $\text{CO}_2$  at a mixing ratio of 379.5 parts-per-million by volume (ppmv),  $\text{CH}_4$  at  
97 1.976 ppmv and CO at 198.4 parts-per-billion by volume (ppbv)) and a working mixture ( $\text{CO}_2 = 452.15$   
98 ppmv,  $\text{CH}_4 = 2.988$  ppmv,  $\text{CO} = 494.5$  ppbv). The estimated precision of 1 min data was  $\pm 0.12$  ppmv,  $\pm 0.6$   
99 ppbv, and  $\pm 1.89$  ppbv for  $\text{CO}_2$ ,  $\text{CH}_4$  and CO respectively, while the estimated accuracy was  $< 1$  ppmv,  $< 3$   
100 ppbv, and  $< 4$  ppbv, respectively.

101 Mixing ratios of total odd nitrogen ( $\text{NO}_y \equiv \text{NO} + \text{NO}_2 + \Sigma \text{PAN} + \Sigma \text{AN} + \text{HNO}_3 + \text{HONO} + 2\text{N}_2\text{O}_5 + \text{ClNO}_2 + \dots$ )  
102 were measured by a chemiluminescence analyzer equipped with a heated Molybdenum converter  
103 (Thermo 42i) as described elsewhere (Tokarek et al., 2014; Odame-Ankrah, 2015).

104 The total sulfur (TS) measurements were conducted using a thermal oxidizer (Thermo Scientific Model  
105 CON101) to convert TS to  $\text{SO}_2$  and detected using a pulsed-fluorescence analyzer (Thermo Scientific,  
106 Model 43iTLE).  $\text{SO}_2$  was measured directly with a second analyzer (Thermo Scientific, Model 43iTLE).  
107 Total reduced sulfur (TRS) mixing ratios were calculated by subtracting mixing ratios of  $\text{SO}_2$  from TS.

108 Concentrations of particle-surface bound polycyclic aromatic hydrocarbons (pPAH) were measured using  
109 a photoelectric aerosol sensor (EcoChem Analytics, Model PAS 2000CE) (Wilson et al., 1994; Burtscher et  
110 al., 1982).

111 Two soot-particle aerosol mass spectrometers (SP-AMS, Aerodyne Research, Inc.) (Onasch et al., 2012)  
112 measured non-refractory  $\text{PM}_{10}$  components. Both SP-AMS were high resolution time-of-flight aerosol  
113 mass spectrometers (HR-ToF-AMS) fitted with a diode pumped Nd:YAG 1064 nm laser vaporizer; one SP-  
114 AMS had its oven removed to measure black carbon containing particles only using the laser. Direct  
115 calibrations of rBC using mono-disperse "Regal Black" (Cabot Corp. R400) particles were carried out  
116 three times during the 2013 JOSM intensive study. Positive Matrix Factorization (PMF) was performed to  
117 identify the potential sources of organic aerosol as described in the companion study (Adam et al., in  
118 prep). Factors associated with primary aerosol, i.e., hydrocarbon-like organic aerosol (HOA), a less

oxidized oxygenated organic aerosol factor (LO-OOA) and measured refractory black carbon (rBC) were added as variables for PCA analysis. Mass spectra associated with LO-OOA exhibited H/C, O/C and N/C ratios of  $\sim 1.62$ ,  $\sim 0.36$ , and  $\sim 0.004$ , respectively; while the O/C and N/C ratios are similar to HOA, the H/C ratio of LO-OOA more resembles the more oxidized OOA factor (MO-OOA) (Adam et al., in prep.).

Particle volumes were calculated (assuming spherical particle shapes) from sub- and super-micron size distributions acquired using a scanning mobility particle sizer (SMPS, TSI with a differential mobility analyzer model 3081 and condensation particle counter model 3776;  $PM_1$ ) and a 0.071 cm impactor over the size range of 13.6 nm to 736.5 nm and an Aerodynamic Particle Sizer (APS, TSI 3321;  $PM_{10-1}$ ) over the size range 1.04  $\mu m$  to 10.4  $\mu m$ , respectively. Both instruments were operated at ambient relative humidity. The SMPS sampled through conductive silicon tubing to minimize wall losses due to wall charges. The APS was operated from a container located on top of the trailer and sampled from a 1.6 m tall,  $\frac{1}{2}$  o.d. aluminum tube whose tip was bent into a U-shape.

An ambient ion monitor – ion chromatograph (AIM-IC) (Markovic et al., 2012) was used to measure hourly averaged gas-phase  $NH_3$  and  $PM_{2.5}$  particle-phase (i.e., of particles  $< 2.5 \mu m$  diameter)  $NH_4^+$  concentrations. High time-resolution particle-phase  $NH_4^+$  measurements made by the SP-AMS were scaled by interpolated phase ratios observed by AIM-IC to calculate gas-phase  $NH_3$  concentrations at high time resolution. This approach assumes the same phase ratios for  $PM_{2.5}$  as for  $PM_1$ .

## **Determination of optimum PCA solution**

The full component pattern (before component removal, with rotation, i.e., showing 22 components for 22 variables) obtained for this data set is shown in Table S-2. A common challenge in PCA is the determination of the maximum number of components to retain in the analysis. Several criteria are used for this purpose: the latent root criterion, where only components with eigenvalues greater than 1 are considered significant, the 5% variance criterion, where the last component selected accounts for only a small portion (<5%) of the variance, the 95% cumulative percentage of variance criterion, where the extracted components account for at least 95% of the total variance, and the Scree test. In the latter, the eigenvalues are plotted against the number of components in the order of extraction (Fig. S-2); to avoid including too many components with unique variance, the number of acceptable components is located at the point where this plot becomes horizontal. The latent root criterion is most commonly used, but tends to extract too few components when the number of variables is < 20 (Hair et al., 1998). The Scree test, on the other hand, often requires "some art in administering it" (Cattell, 1966), i.e., is subjective, though generally results in the inclusion of two or three more components than the latent root criterion (Hair et al., 1998).

The maximum component number for each criterion are summarized in Table S-11. The Scree test plot (Fig. S-1) shows two plateaus where the slope becomes approximately horizontal: The first is located at N = 5 and the second at N = 12. The latent root criterion and the <5% variance method suggests a 7-component solution, whereas the >95% percentage of variance criterion suggests using a 10-component solution. Hair et al. (1998) recommend to examine component solutions with differing numbers of components to evaluate which best represents the structure of the variables. In the following, solutions are presented in ascending order of extracted components.

## 5-component solution

As a first attempt at interpretation of the PCA, the first cut-off of the Scree test criterion was chosen (N = 5 variables). The results (after Varimax rotation) are presented in Table S-3.

The 5-component solution accounts for a cumulative variance of 81.0 % after rotation. Communalities for the analysis, i.e., the fraction of total pollutant observations accounted for by the PCA (Otto, 2007), are greater than 70% for 18 variables. The lowest communalities were obtained for gas-phase ammonia (0.40), CO (0.48) and PM<sub>10-1</sub> (0.51). TRS and the IVOCs were also relatively poorly represented (0.63 and 0.73, respectively). All eigenvalues are greater than 1.

The component accounting for most of the variance of the data, component 1, is strongly associated with all of the anthropogenic VOCs (with correlations of  $r > 0.8$ ) and TRS ( $r = 0.76$ ), weakly associated with CH<sub>4</sub> ( $r = 0.62$ ), HOA ( $r = 0.44$ ), LO-OOA ( $r = 0.59$ ), IVOCs ( $r = 0.47$ ), and CO ( $r = 0.53$ ), and poorly associated with NO<sub>y</sub> and TS ( $r = 0.25$  and  $r = 0.28$ , respectively). Component 1 is consistent with tailings ponds emissions with potentially small contributions from nearby facilities (interpreted from weak and poor correlations with rBC ( $r = 0.33$ ) and NO<sub>y</sub> ( $r = 0.25$ )), which would otherwise remain unexplained.

Component 2 is strongly associated with the combustion tracers NO<sub>y</sub> ( $r = 0.83$ ), rBC ( $r = 0.89$ ) and pPAH ( $r = 0.83$ ) and weakly associated with IVOCs ( $r = 0.61$ ), gas-phase ammonia ( $r = 0.34$ ), undecane ( $r = 0.31$ ), and CH<sub>4</sub> (0.38), but poorly and not with CO or CO<sub>2</sub> ( $r = 0.19$  and 0.06, respectively); this component is identified as mine fleet emissions. Component 3 is strongly associated ( $r > 0.9$ ) with the biogenic VOCs and weakly ( $r = 0.55$ ) associated with CO<sub>2</sub> and is identified as a biogenic component.

Component 4 is strongly associated with SO<sub>2</sub> and TS ( $r = 0.93$  and 0.91, respectively) and is consistent with emissions from upgrader facilities. These four components persisted, with little variation, in all solutions with a greater number of selected components (see below).

Component 5 is strongly associated with CO<sub>2</sub> ( $r = 0.71$ ), and weakly associated with PM<sub>10-1</sub> ( $r = 0.57$ ), CH<sub>4</sub> ( $r = 0.53$ ) and CO ( $r = 0.40$ ). We are not aware of a source type that would fit this profile, i.e., combine



184 this particular set of pollutants without also being associated with  $\text{NO}_y$  ( $r = 0.02$ ). This suggests that this  
185 component is an artifact arising from an insufficient number of components used in the analysis and  
186 motivates the inclusion of more components.

#### 188 **6-component solution**

189 A 6-component solution is shown in Table S-4. Satisfying the percentage of variance criterion of the last  
190 component accounting for less than 5% of the variance (4.6% in this case, Table S-2) was selected.  
191 This solution accounts for a total variance of 85.23%. The first four components are essentially  
192 unchanged from the 5-component solution (with the exception of LO-OOA in component 2 becoming  
193 more poorly correlated ( $r = 0.22$ )). Component 5 is strongly associated with IVOCs ( $r = 0.70$ ) and weakly  
194 associated with LO-OOA ( $r = 0.60$ ), and TRS ( $r = 0.56$ ). Component 6 is strongly associated with  $\text{PM}_{10-1}$  ( $r$   
195  $= 0.81$ ) and weakly associated with  $\text{CO}_2$  ( $r = 0.62$ ),  $\text{CH}_4$  ( $r = 0.41$ ), HOA ( $r = 0.30$ ) and  $\text{NH}_3$  ( $r = 0.36$ ) and,  
196 unlike the 5-component solution, not associated with CO.

#### 198 **7-component solution**

199 Next, the latent root criterion gives a 7-component solution. The PCA results (after Varimax rotation) are  
200 presented in Table S-5. The seven components account for a cumulative variance of 88.7% after  
201 rotation. Communalities for the analysis are all greater than 60%, with the lowest communality obtained  
202 for CO (0.61). All eigenvalues are greater than 1.

203 Components 1 through 4 have the same associations with similar  $r$  values as those in the 5-component  
204 analysis, with the only significant exception a poorer association ( $r = 0.20$ ) of component 2 with gas-  
205 phase ammonia.

206 The identifications of components 5 through 7 of the 7-component solution are murky at best.

207 Component 5 is weakly associated with TRS ( $r = 0.56$ ) and IVOCs ( $r = 0.66$ ). Component 6 is strongly

associated with  $\text{PM}_{10-1}$  volume ( $r = 0.89$ ), and weakly with  $\text{CO}_2$  ( $r = 0.54$ ), and  $\text{CH}_4$  ( $r = 0.36$ ) and appears to be combination of a dust component with a source of greenhouse gases, whereas component 7 is strongly associated with gas-phase ammonia ( $r = 0.82$ ) and poorly associated with CO ( $r = 0.29$ ). Both appear to be amalgamations of distinct sources and suggest that too few components were selected. Hair et al. (1998) note that the latent root criterion has a tendency to extract a conservative number of components if the number of variables is  $< 20$ , close to the 22 variables in this analysis, consistent with what is observed here. Hence, the 7-component solution is sub-optimal.

#### **8-component solution**

An 8-component solution is presented in Table S-6. Not satisfying any criterion, it is included here for the sake of completeness. Owing to the inclusion of an additional component, the cumulative variance improved to 91.6%. The greatest improvement was seen for CO, gas-phase ammonia, as well as the IVOCs, whose communalities increased from 0.61, 0.91, and 0.80 (for the 7-component solution) to 0.96, 0.96 and 0.84, respectively.

The main effect of the inclusion of an additional component was the separation of component 7 into two distinct components: one of these was strongly associated with gas-phase ammonia ( $r = 0.92$ ), and the other was strongly associated with CO ( $r = 0.85$ ). A considerable fraction of the CO observed in the region is generated as a byproduct of the photochemical oxidation of hydrocarbons (Shephard et al., 2015); component 8 appears to capture this source, whereas component 1 captures the anthropogenic emissions. The area near the oil sands mining operations is enriched in ammonia, which originates from multiple sources: it is used as a floating agent to separate and recover bitumen from tar and is generated during bitumen upgrading (called hydrotreating) in which N is removed as  $\text{NH}_3$  and can be present as a contaminant in tailing ponds. Other sources, such as agricultural activities, biological decay processes, and smoldering fires are relatively minor in the region (Bytnerowicz et al., 2010). The poor

association of component 2 with ammonia ( $r = 0.22$ ) may capture the use of ammonia as a floating agent, whereas component 8 embodies the remaining sources. Component 5 is strongly associated with IVOCs ( $r = 0.71$ ), and weakly associated with LO-OOA ( $r = 0.65$ ) and TRS ( $r = 0.40$ ). It is unclear if these variables originate from the same source or are forced together as a result of having chosen too few components. Considering that component 7 is split when an additional component is used (see below), the latter is more likely. Component 6 remains strongly associated with  $PM_{10-1}$  volume ( $r = 0.89$ ), and weakly associated with  $CO_2$  ( $r = 0.53$ ), and  $CH_4$  ( $r = 0.35$ ) and is difficult to interpret. Because of the unclear classification of components 5 through 8, the 8-component solution is rejected.

#### **9-component solution**

A 9-component solution is presented in Table S-7. Components 1 through 8 describe sources that are qualitatively similar to those provided by the 8-component solution. Component 9 is strongly associated with TRS ( $r = 0.71$ ) and poorly associated with o-xylene ( $r = 0.30$ ); its profile is consistent with tailings ponds emission, where the presence of naphtha as a diluent gives rise to BTEX emissions and bacteria produce reduced sulfur compounds (Small et al., 2015; Warren et al., 2016). Component 6 is strongly associated with  $PM_{10-1}$  ( $r = 0.89$ ) and weakly associated with  $CO_2$  ( $r = 0.54$ ) and  $CH_4$  ( $r = 0.41$ ). We have decided to reject this solution on the basis that  $< 95\%$  cumulative variance is observed.

#### **10-component solution**

Next, a 10-component solution with cumulative variance of 95.5%, satisfying the 95% criterion, was considered. With this solution, all communalities are  $>0.85$  (Table 3). Component 6 is strongly associated with  $CO_2$  ( $r = 0.77$ ) and weakly associated with  $CH_4$  ( $r = 0.59$ ) but is not associated with other combustion tracers and is identified as inactive open-pit mines (see main text). Component 7 is strongly correlated

with PM<sub>10-1</sub> ( $r = 0.93$ ) and is identified as wind-blown dust. Component 8 and 9 are strongly associated with a single variable each, gas-phase ammonia ( $r = 0.94$ ) and CO ( $r = 0.87$ ), respectively. Component 10 is strongly associated with TRS ( $r = 0.71$ ) and weakly associated with o-xylene ( $r = 0.32$ ). Overall, this component is most consistent with a tailings ponds source, where the presence of naphtha as diluent gives rise to BTEX emissions, and sulfur-reducing bacteria are at work (Small et al., 2015; Warren et al., 2016). Overall, the 10-component solution was judged to be optimal.

### **11-component solution**

The 11-component analysis is presented in Table S-8. Component 10 is now strongly associated with LO-OOA ( $r = 0.72$ ) and weakly with rBC ( $r = 0.34$ ) and has a low eigenvalue of 0.87. This solution is therefore rejected as we believe it contains too many components.

### **PCA without aerosol variables**

A sensitivity test was conducted by which all aerosol species were removed as variables. The results of this sensitivity test are shown in Table S-10 and are presented as a 9-component solution, since the dust component associated with PM<sub>10-1</sub> (component 7 in Table 5) cannot be generated when its main variable is removed.

The pattern in Table S-10 resembles that in Table 5 of the main manuscript, in that the same nine components emerged in both solutions with similar magnitude  $r$  values for each of the variables, including the IVOC signature. The only difference is that components 2 and 3 as well as 5 and 6 have traded places (i.e., the relative magnitudes of their eigenvalues, which were similar in Table 5, have switched), which is inconsequential. Furthermore, the correlation coefficients in Table S-10 are of similar magnitude (i.e., within  $\pm 0.1$ ) as those in Table 5, which suggests that IVOC to SOA conversion does not adversely affect the PCA, likely because of the proximity of the receptor site to sources.

## Discussion of low-eigenvalue components

### Component 6: A non-combustion source of CO<sub>2</sub> and CH<sub>4</sub>

Component 6 of the analysis has a strong association with the greenhouse gases CO<sub>2</sub> ( $r = 0.77$ ) and a weak association with CH<sub>4</sub> ( $r = 0.59$ ) but is not associated with tracers of combustion (i.e., NO<sub>y</sub>, pPAH, rBC) or naphtha (i.e., anthropogenic VOCs).

A significant amount of carbon is stored in bitumen, which, on geological time scales, conduces formation of CO<sub>2</sub> and CH<sub>4</sub> (i.e., natural gas) reservoirs and pools. When bitumen is mined, substantial emissions of CO<sub>2</sub> and, in particular, of CH<sub>4</sub> occur (Johnson et al., 2016). It is unclear, though, to what extent these greenhouse gases are released from "hot spots" (i.e., from a small number of locations) through surface cracks and fissures in the mine faces, or from new material that is exposed and then releases greenhouse gases during material handling, transport and processing (Johnson et al., 2016). The PCA analysis presented here would be more consistent with the "hot spots" hypothesis since component 6 is not associated with NO<sub>y</sub>, PAHs, or CO, which are expected to be emitted by the Diesel machinery involved in surface mining (i.e., active disturbance of the bitumen).

Another potential source contribution to component 6 is the degradation of peat and surface soil. Peatland soils, as they occur in the boreal forest surrounding the AMS 13 site, have long been recognized as important contributors to greenhouse gas fluxes and may also be contributing to component 6 (Miller et al., 2014; Gorham, 1991; Warner et al., 2017). The fixation and/or release of CO<sub>2</sub> as well as consumption and/or production of CH<sub>4</sub> through root, anaerobic and aerobic microbial respiration are dependent on soil conditions such as water table position, temperature, soil pH, and plant community composition (Yavitt et al., 2005; Oertel et al., 2016; Whalen, 2005). Emissions from peat and surface soil that was stripped as part of surface mining is expected to release between

1.1×10<sup>10</sup> and 4.7×10<sup>10</sup> kg stored carbon (Rooney et al., 2012), though it is unclear on what time scale this release will occur. Some of this historical peat material is used for land reclamation. However, a preliminary assessment of greenhouse gas fluxes from such a site gave no indication of significant emissions, at least in the short term (Nwaishi et al., 2016). The bivariate polar plot shows that component 6 is associated with no particular wind direction but with relatively low wind speeds (< 1.5 m/s; Figure S-7C), consistent with a dispersed surface source. Further, when variables associated with secondary processes were added to the analysis (Table 7), component 6 anticorrelates with O<sub>x</sub> (r = -0.41). Dry deposition is a significant O<sub>3</sub> and NO<sub>2</sub>, and therefore O<sub>x</sub>, loss process (Wesely and Hicks, 2000; Zhang et al., 2002).

Overall, we have too little information to constrain soil fluxes for this data set. Considering the large CH<sub>4</sub> and CO<sub>2</sub> concentrations observed in this study, it is more likely that anthropogenic sources dominate over natural soil emissions (Thompson et al., 2017). Future field campaigns at AMS 13 would benefit from N<sub>2</sub>O measurements to constrain contributions of natural sources to greenhouse gas concentrations, such as those produced by microbes in water-logged soil.

#### **Component 7: Wind-blown dust**

Component 7 is correlated with PM<sub>10-1</sub> (r = 0.93) and poorly with CO<sub>2</sub> (r = 0.25), HOA (r = 0.23), and LO-OOA (r = 0.25). In the Athabasca oil sands region, surface mining has created large portions of land whose surface is void of vegetation and is covered by sand and soil particles, which are readily suspended by wind and vehicle traffic. Other mining activities add to the PM<sub>10-1</sub> emissions, including combustion processes, tailings sands, and mine haul roads, though the contributions of each of these to the overall PM<sub>10-1</sub> burden is uncertain (Wang et al., 2015). Recently, Phillips-Smith et al. investigated metal species found in PM<sub>2.5</sub> aerosol at AMS 13 and found haul road dust and soil from mine faces to be

important sources of PM<sub>2.5</sub> (Phillips-Smith et al., 2017) and, likely, PM<sub>10-1</sub> as well. The very poor associations of this component with CO<sub>2</sub> and CH<sub>4</sub> and lack of association with NO<sub>y</sub> ( $r = 0.02$ ) suggest contributions of open mine face soil in addition to dust suspended by vehicles travelling on unpaved roads.

The size range captured by PM<sub>10-1</sub> may also include bioaerosol, including bacteria, fungal spores and plant pollen, which constitute the "natural" background aerosol over vegetated continental regions, typically contributing a few  $\mu\text{g m}^{-3}$  of aerosol mass (Huffman et al., 2010). Considering the large PM<sub>10-1</sub> volumes observed in this work (Table 3), the contribution of bioaerosol is likely minor.

#### **Component 8: Ammonia**

Component 8 is a single variable component strongly associated with NH<sub>3</sub> ( $r = 0.94$ ) but with no other variables.

Bytnerowicz et al. (2010) reported larger concentrations of NH<sub>3</sub> in the oil sands region than the provincial average. More recently, Shephard et al. (2015) reported enhancements of NH<sub>3</sub> in the general area as judged from satellite observations. Both studies hence suggest the existence of anthropogenic sources, though Shephard et al. (2015) speculated that biomass burning can contribute to the ammonia burden in the region. A recent modelling study by Whaley et al. (2018) estimated that around half of near-surface NH<sub>3</sub> during the study was likely from bi-directional exchange (i.e., re-emission from soil and plants).

In the oil sands, NH<sub>3</sub> is used as a floating agent for the separation and recovery of bitumen from tar, during bitumen upgrading in a process called "hydrotreating", and in tailing ponds, which, on occasion, have been contaminated with NH<sub>3</sub> to such a degree that they outgas it (Bytnerowicz et al., 2010).

Ammonia is also used for flue gas de-sulfurization by Syncrude; emission inventories (NPRI, 2013; ECCC, 2013b) suggest their fugitive emissions are the largest anthropogenic source in the region, though it is not clear if all sources are accurately inventoried.

The lack of association of ammonia with other variables in this component and the bivariate polar plots (Figure S-9) are consistent with an  $\text{NH}_3$ -specific source profile, such as fugitive emissions from one or more point sources that emit independently from other activities (i.e., ammonia storage tanks) and natural emissions from soil and trees (Whaley et al., 2018).

#### **Component 9: Incomplete hydrocarbon oxidation**

Component 9 is another single variable component and strongly correlates with CO ( $r = 0.87$ ). The conventional interpretation of CO is as a byproduct of incomplete VOC oxidation, as it is found in fossil fuel combustion exhaust or in biomass burning plumes. Component 9, however, is not associated with  $\text{NO}_y$  ( $r = -0.08$ ) or  $\text{CO}_2$  ( $r = 0.05$ ), which rules out this conventional interpretation.

Recently, Marey et al. (2015) examined the spatial distribution of CO in Northern Alberta using a combination of satellite and ground station data and found that most CO is derived from anthropogenic sources, biomass burning and the photochemical oxidation of methane and other VOCs. During the 2013 JOSM study, there was no obvious (i.e., tracer) evidence for fire emissions impacting the measurements at AMS 13 (Phillips-Smith et al., 2017), though an impact from distant sources (such as fires located 1,000s of km upwind in British Columbia or Washington State) cannot be entirely ruled out. We therefore interpret component 9 as a VOC oxidation product component.



## **Component 10: Dry tailings**

Component 10 is strongly associated with TRS ( $r = 0.71$ ) and weakly with o-xylene ( $r = 0.32$ ) and poorly with IVOCs ( $r = 0.20$ ). This component is qualitatively similar to component 1, in that the presence of o-xylene suggests emission of naphtha, and the presence of TRS suggests anaerobic sulfur reducing bacteria and methanogens as they occur in tailings ponds (Holowenko et al., 2000; Percy, 2013; Quagraine et al., 2005). However, the absence of correlations with  $\text{NO}_y$ , rBC, and CO suggests that this source is not in spatial proximity with a continuously operating combustion source. The much poorer correlations of o-xylene,  $\text{CH}_4$ , and IVOCs than for component 1 suggests that this component is much more "aged", i.e., emits less naphtha and bitumen.

As part of the reclamation process, tailings ponds in the Alberta oil sands region are converted into "composite tailings", which consist of a consolidated alkaline, saline mixture of processed sand, residual bitumen, clay fines, and gypsum ( $\text{CaSO}_4$ ). This mixture settles and releases water, forming shallow pools of surface water (Figure 4J). Due to intensive microbial activity, composite tailings deposits are strong sources of  $\text{H}_2\text{S}$  and, likely, other reduced sulfur species (Warren et al., 2016; Bradford et al., 2017). Composite tailings are a source consistent with the emission profile of component 10. The association with TRS is explained by its production from biological activity and the presence of IVOCs by outgassing from the residual bitumen. Syncrude (the company operating closest to AMS 13) has been undertaking a pilot scale wetland reclamation project in the Athabasca Oil Sands Region to allow the development of a fen wetland above composite tailings (Bradford et al., 2017). Component 10 is hence interpreted as a dry tailings pond component, though the confidence in this interpretation is somewhat marginal as judged, for example, from the low eigenvalue of 0.74.

## 390     **Bivariate polar plots**

391     Bivariate polar plots map a surface using wind direction and wind speed and then model pollutant  
392     concentrations. While PCA is good at showing the temporal distribution of sources, bivariate polar plots  
393     help to show the spatial distribution of sources.

394     Figure S-3 shows a sample of variables associated with component 1. This component appears to  
395     dominate when winds are from the SSE and E and of moderate wind speeds (2-3 m/s).

396     Figure S-4 shows a sample of dominant variables associated with component 2. This component appears  
397     to dominate when winds are from the E at low wind speeds (1-2 m/s). The map appears to track the  
398     location of the Athabasca river and highway 63, corroborating that this source is from vehicular  
399     emissions.

400     Figure S-5 shows a sample of dominant variables associated with component 3. This component appears  
401     to dominate when winds are stagnant and local. This is unsurprising because biogenic emissions are  
402     expected to be emitted in great concentrations locally since our site is surrounded on all sides by forest.

403     Figure S-6 shows a sample of dominant variables associated with component 4 (or with component 2 in  
404     the secondary processes PCA). This component appears to dominate when winds are moderate (2-3  
405     m/s) and from the SE and E.

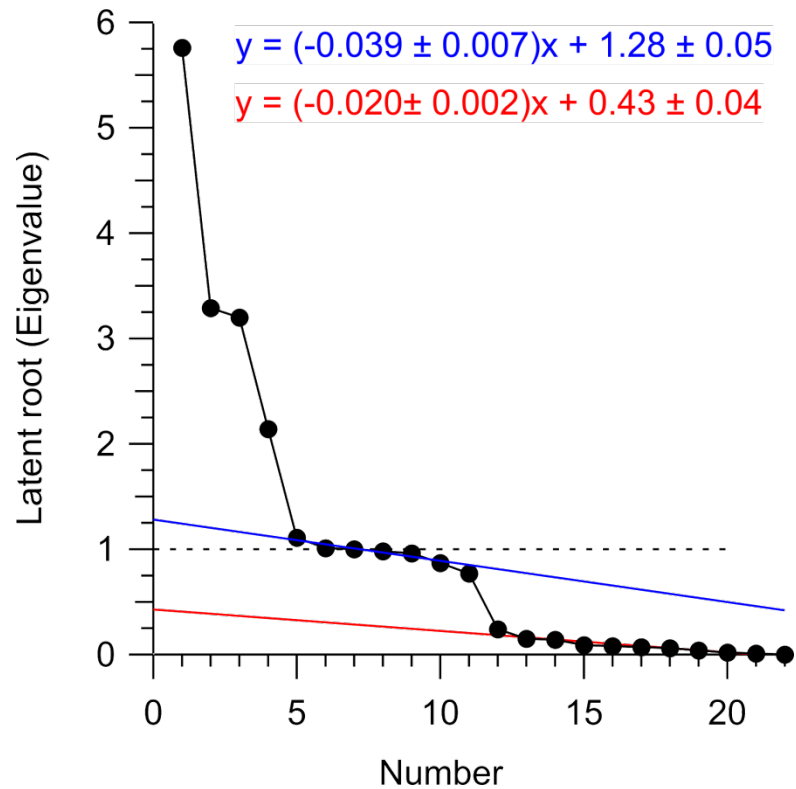
406     Figure S-7 shows a sample of dominant variables associated with component 5. This component appears  
407     to dominate when winds are from the E at moderate wind speeds (2-3 m/s).

408     Figure S-8 shows a sample of dominant variables associated with component 6. This component appears  
409     to dominate when winds are stagnant and local. This suggests that this source is biogenic and may be  
410     due to emissions from trees.

411     Figure S-9 shows a sample of dominant variables associated with component 7. This component appears  
412     to dominate when winds are from the SE and E at moderate wind speeds (1-3 m/s).

413     Figure S-10 shows a sample of dominant variables associated with component 8. This component does

not appear to have a specific direction associated with it and is observed in all directions. This component is observed when winds are at moderate to high speeds (2-4 m/s). Figure S-11 shows a sample of dominant variables associated with component 9. This component is observed when winds are from the S, SE, and E. This component is observed when winds are at low to moderate speeds (1-3 m/s). Figure S-12 shows a sample of dominant variables associated with component 10. This component is observed when winds are from the SSE. This component is observed when winds are around 1.5 m/s. This source is very likely a point source due to its consistency with wind direction and speed.



**Figure S-2.** Scree plot used to consider the number of components to retain. Dashed line represents the latent root criteria (eigenvalues > 1). Blue line represents the first instance of eigenvalues becoming

horizontal. Red line represents the second instance of eigenvalues becoming horizontal.

**Table S-1.** Ionimed Analytical GCU Standard.

Compound	Volume mixing ratio (ppmv)	Uncertainty (%)
Formaldehyde	1.01	±8
Methanol	1.01	±8
Acetonitrile	1.01	±6
Acetaldehyde	1.01	±5
Ethanol	1.01	±8
Acrolein	0.98	±5
Acetone	1.02	±5
Isoprene	0.99	±5
Crotonaldehyde	0.92	±6
2-Butanone	1.01	±5
Benzene	1.01	±5
Toluene	1.02	±5
o-xylene	1.03	±6
Chlorobenzene	1.02	±5
α-pinene	0.93	±5
1,2, Dichlorobenzene	1.03	±7
1,2,4-Trichlorobenzene	1.01	±9

430 **Table S-2.** The component pattern after Varimax rotation. Correlations greater than 0.30 or less than -0.30 are bolded.

	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22
<b><u>Anthropogenic VOCs</u></b>																						
o-xylene	<b>0.89</b>	0.07	0.03	0.10	0.09	0.06	-0.03	0.11	0.15	0.06	0.26	0.11	0.00	0.04	-0.08	0.24	0.00	-0.01	-0.02	0.00	-0.01	0.00
1,2,3 - TMB	<b>0.94</b>	0.15	0.07	0.06	0.05	0.08	-0.01	0.06	0.17	0.01	-0.01	0.02	-0.04	0.01	-0.08	-0.14	0.03	0.00	-0.09	0.00	-0.07	0.00
1,2,4 - TMB	<b>0.94</b>	0.13	0.01	0.11	0.08	0.05	-0.02	0.09	0.17	0.04	0.12	0.03	-0.01	0.02	-0.06	0.03	0.01	0.00	-0.04	0.00	0.09	0.00
decane	<b>0.91</b>	0.22	-0.02	0.15	0.05	0.00	0.04	0.15	0.05	0.15	0.07	-0.02	0.04	-0.01	0.07	-0.03	-0.03	0.02	0.16	0.00	-0.01	0.00
undecane	<b>0.85</b>	0.27	-0.08	0.23	0.08	-0.03	0.06	0.05	0.00	0.20	0.01	-0.10	0.09	0.00	0.26	-0.05	-0.02	0.02	0.02	0.00	0.00	0.00
<b><u>Biogenic VOCs</u></b>																						
α-pinene	-0.03	-0.08	<b>0.98</b>	-0.11	0.02	0.05	-0.08	0.02	0.01	0.00	-0.01	0.02	0.00	0.00	-0.01	0.01	-0.08	-0.01	0.00	0.09	0.00	0.00
β-pinene	-0.02	-0.08	<b>0.97</b>	-0.12	0.01	0.05	-0.08	0.01	0.01	0.01	0.01	-0.02	0.00	-0.02	0.02	0.00	-0.10	0.01	0.01	-0.09	0.00	0.00
limonene	0.08	-0.02	<b>0.92</b>	-0.08	0.06	0.23	-0.11	0.08	0.02	0.07	-0.05	0.02	-0.03	0.03	-0.02	0.00	0.23	0.00	-0.01	0.00	0.00	0.00
<b><u>Combustion tracers</u></b>																						
NO <sub>y</sub>	0.26	<b>0.80</b>	-0.25	0.21	0.03	-0.05	0.10	0.19	-0.04	0.07	0.04	0.01	<b>0.34</b>	0.00	0.04	0.00	-0.01	-0.01	0.01	0.00	0.00	0.00
rBC	<b>0.31</b>	<b>0.80</b>	0.03	0.05	0.08	0.07	0.11	0.24	0.12	<b>0.34</b>	-0.03	0.02	-0.02	-0.05	0.02	-0.01	0.00	0.22	0.01	0.00	0.00	0.00
CO	<b>0.41</b>	0.18	0.04	0.02	0.08	0.07	0.05	0.03	<b>0.88</b>	0.06	0.00	0.02	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
CO <sub>2</sub>	0.10	0.09	<b>0.46</b>	-0.12	0.23	<b>0.82</b>	-0.14	-0.03	0.07	0.00	-0.05	0.02	-0.01	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
<b><u>Aerosol species</u></b>																						
pPAH	0.07	<b>0.94</b>	-0.08	-0.11	0.02	0.06	0.14	0.00	0.09	-0.15	0.00	0.01	-0.14	-0.07	-0.02	0.00	0.00	-0.10	0.00	0.00	0.00	0.00
PM <sub>10-1</sub>	0.18	0.13	0.07	0.10	<b>0.94</b>	0.16	-0.03	0.04	0.07	0.10	0.08	0.02	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
HOA	<b>0.41</b>	<b>0.74</b>	0.02	0.11	0.21	0.11	-0.04	0.13	0.15	0.19	0.10	0.05	0.01	<b>0.35</b>	0.00	0.01	0.01	-0.01	0.00	0.00	0.00	0.00
LO-OOA	<b>0.45</b>	0.17	0.13	0.25	0.19	0.01	-0.04	0.28	0.10	0.73	0.16	0.02	0.01	0.03	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00
<b><u>Sulfur</u></b>																						
TS	0.25	0.04	-0.16	<b>0.94</b>	0.07	-0.05	-0.02	0.02	0.01	0.09	0.14	0.00	0.01	0.01	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.01
SO <sub>2</sub>	0.11	0.02	-0.15	<b>0.98</b>	0.04	-0.04	-0.03	-0.02	0.01	0.05	-0.05	-0.01	0.01	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	-0.01
TRS	<b>0.57</b>	0.05	-0.08	0.11	0.14	-0.05	0.03	0.16	-0.01	0.13	<b>0.77</b>	0.02	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
<b><u>Other</u></b>																						
IVOCs	<b>0.34</b>	<b>0.34</b>	0.12	-0.03	0.05	-0.02	-0.02	<b>0.84</b>	0.04	0.18	0.13	0.01	0.01	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
NH <sub>3</sub>	0.01	0.19	-0.23	-0.04	-0.03	-0.09	<b>0.95</b>	-0.01	0.04	-0.01	0.01	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
CH <sub>4</sub>	<b>0.60</b>	<b>0.39</b>	0.10	-0.05	0.14	0.44	0.00	0.06	0.16	0.08	0.09	<b>0.46</b>	0.01	0.03	-0.02	0.01	0.00	0.00	0.00	0.00	0.00	0.00
Eigenvalues	5.76	3.29	3.20	2.14	1.11	1.01	1.00	0.98	0.96	0.87	0.77	0.24	0.15	0.14	0.09	0.08	0.07	0.06	0.04	0.02	0.01	0.00
% of variance	26.17	14.97	14.55	9.75	5.06	4.60	4.53	4.46	4.34	3.94	3.51	1.09	0.69	0.63	0.42	0.37	0.33	0.28	0.17	0.08	0.06	0.00
% cum. var.	26.17	41.14	55.69	65.43	70.49	75.09	79.62	84.07	88.42	92.36	95.87	96.97	97.66	98.29	98.71	99.08	99.41	99.69	99.86	99.94	100.0	100

431

432 **Table S-3.** The pattern after Varimax rotation with 5 components selected.

	1	2	3	4	5	Communalities
<b><u>Anthropogenic VOCs</u></b>						
o-xylene	<b>0.94</b>	0.08	0.03	0.09	0.15	0.93
1,2,3 - TMB	<b>0.90</b>	0.14	0.04	0.01	0.23	0.89
1,2,4 - TMB	<b>0.95</b>	0.14	-0.01	0.09	0.18	0.97
decane	<b>0.91</b>	0.27	-0.03	0.16	0.05	0.93
undecane	<b>0.82</b>	<b>0.31</b>	-0.10	0.26	0.05	0.84
<b><u>Biogenic VOCs</u></b>						
α-pinene	-0.03	-0.05	<b>0.94</b>	-0.15	0.04	0.91
β-pinene	-0.02	-0.06	<b>0.94</b>	-0.15	0.03	0.90
limonene	0.07	0.02	<b>0.94</b>	-0.10	0.18	0.93
<b><u>Combustion tracers</u></b>						
NO <sub>y</sub>	0.25	<b>0.83</b>	-0.29	0.22	0.02	0.89
rBC	<b>0.33</b>	<b>0.89</b>	0.04	0.07	0.13	0.92
CO	<b>0.53</b>	0.19	-0.02	-0.08	<b>0.40</b>	0.48
CO <sub>2</sub>	0.07	0.06	<b>0.55</b>	-0.13	<b>0.71</b>	0.83
<b><u>Aerosol species</u></b>						
pPAH	0.01	<b>0.83</b>	-0.20	-0.20	0.27	0.84
PM <sub>10-1</sub>	0.21	0.19	0.15	0.29	<b>0.57</b>	0.51
HOA	<b>0.44</b>	<b>0.75</b>	0.03	0.15	<b>0.32</b>	0.88
LO-OOA	<b>0.59</b>	<b>0.37</b>	0.28	0.41	-0.06	0.74
<b><u>Sulfur</u></b>						
TS	0.28	0.04	-0.18	<b>0.91</b>	0.01	0.94
SO <sub>2</sub>	0.10	0.01	-0.18	<b>0.93</b>	0.05	0.91
TRS	<b>0.76</b>	0.10	-0.04	0.17	-0.13	0.63
<b><u>Other</u></b>						
IVOCs	<b>0.47</b>	<b>0.61</b>	0.28	0.03	-0.25	0.73
NH <sub>3</sub>	0.02	<b>0.34</b>	<b>-0.47</b>	-0.21	-0.14	0.40
CH <sub>4</sub>	<b>0.62</b>	<b>0.38</b>	0.12	-0.09	<b>0.53</b>	0.84
<b>Eigenvalues</b>	6.42	3.79	3.56	2.34	1.70	
<b>% of variance</b>	29.20	17.25	16.20	10.63	7.74	
<b>Cumulative variance</b>	29.20	46.45	62.65	73.28	81.01	

433

434 **Table S-4.** The pattern after Varimax rotation with 6 components selected.

	1	2	3	4	5	6	Communalities
<b><u>Anthropogenic VOCs</u></b>							
o-xylene	<b>0.92</b>	0.04	0.01	0.08	0.24	0.13	0.93
1,2,3 - TMB	<b>0.94</b>	0.16	0.08	0.07	0.06	0.05	0.92
1,2,4 - TMB	<b>0.95</b>	0.13	0.00	0.11	0.16	0.08	0.97
decane	<b>0.88</b>	0.23	-0.02	0.18	0.28	0.00	0.94
undecane	<b>0.79</b>	0.28	-0.09	0.29	0.25	0.00	0.85
<b><u>Biogenic VOCs</u></b>							
$\alpha$ -pinene	-0.02	-0.09	<b>0.96</b>	-0.12	0.04	0.01	0.95
$\beta$ -pinene	-0.02	-0.10	<b>0.96</b>	-0.12	0.05	0.01	0.94
limonene	0.09	-0.01	<b>0.95</b>	-0.09	0.04	0.15	0.95
<b><u>Combustion tracers</u></b>							
NO <sub>y</sub>	0.22	<b>0.81</b>	-0.27	0.23	0.25	0.00	0.89
rBC	<b>0.31</b>	<b>0.85</b>	0.07	0.08	0.28	0.10	0.92
CO	<b>0.64</b>	0.29	0.09	0.02	-0.23	0.09	0.56
CO <sub>2</sub>	0.17	0.12	<b>0.57</b>	-0.17	-0.23	<b>0.62</b>	0.84
<b><u>Aerosol species</u></b>							
pPAH	0.09	<b>0.90</b>	-0.10	-0.14	-0.10	0.05	0.87
PM <sub>10-1</sub>	0.19	0.14	0.03	0.12	0.17	<b>0.81</b>	0.76
HOA	<b>0.44</b>	<b>0.73</b>	0.04	0.13	0.22	<b>0.30</b>	0.88
LO-OOA	<b>0.46</b>	0.22	0.17	<b>0.32</b>	<b>0.60</b>	0.21	0.79
<b><u>Sulfur</u></b>							
TS	0.25	0.02	-0.18	<b>0.92</b>	0.12	0.06	0.97
SO <sub>2</sub>	0.10	0.03	-0.15	<b>0.97</b>	-0.02	0.02	0.98
TRS	<b>0.62</b>	-0.04	-0.17	0.06	<b>0.56</b>	0.14	0.75
<b><u>Other</u></b>							
IVOCs	<b>0.31</b>	<b>0.43</b>	0.17	-0.06	<b>0.70</b>	0.02	0.80
NH <sub>3</sub>	0.06	<b>0.41</b>	<b>-0.36</b>	-0.11	-0.11	<b>-0.36</b>	0.46
CH <sub>4</sub>	<b>0.68</b>	<b>0.42</b>	0.15	-0.10	0.00	<b>0.41</b>	0.84
<b>Eigenvalues</b>	6.09	3.60	3.47	2.25	1.76	1.58	
<b>% of variance</b>	27.70	16.38	15.78	10.22	7.98	7.18	
<b>Cumulative variance</b>	27.70	44.07	59.85	70.08	78.06	85.23	

435

436 **Table S-5.** The pattern after Varimax rotation with 7 components selected.

	1	2	3	4	5	6	7	Communalities
<b><u>Anthropogenic VOCs</u></b>								
o-xylene	<b>0.93</b>	0.07	0.01	0.08	0.21	0.12	-0.05	0.93
1,2,3 - TMB	<b>0.94</b>	0.18	0.08	0.06	0.03	0.03	-0.03	0.93
1,2,4 - TMB	<b>0.96</b>	0.15	0.00	0.11	0.14	0.07	-0.02	0.98
decane	<b>0.88</b>	0.26	-0.02	0.18	0.26	0.00	0.01	0.94
undecane	<b>0.79</b>	<b>0.30</b>	-0.09	0.28	0.23	0.00	0.03	0.85
<b><u>Biogenic VOCs</u></b>								
$\alpha$ -pinene	-0.03	-0.09	<b>0.96</b>	-0.11	0.04	0.00	-0.05	0.96
$\beta$ -pinene	-0.02	-0.10	<b>0.96</b>	-0.11	0.06	0.01	-0.05	0.95
limonene	0.09	0.01	<b>0.95</b>	-0.08	0.03	0.12	-0.13	0.95
<b><u>Combustion tracers</u></b>								
NO <sub>y</sub>	0.22	<b>0.83</b>	-0.28	0.22	0.20	-0.02	0.06	0.91
rBC	<b>0.31</b>	<b>0.85</b>	0.07	0.08	0.25	0.11	0.13	0.92
CO	<b>0.62</b>	0.22	0.13	0.03	-0.21	0.19	0.29	0.61
CO <sub>2</sub>	0.17	0.17	<b>0.56</b>	-0.18	-0.28	<b>0.54</b>	-0.27	0.84
<b><u>Aerosol species</u></b>								
pPAH	0.08	<b>0.89</b>	-0.10	-0.15	-0.14	0.03	0.15	0.88
PM <sub>10-1</sub>	0.18	0.12	0.06	0.13	0.18	<b>0.89</b>	0.01	0.89
HOA	<b>0.44</b>	<b>0.76</b>	0.03	0.12	0.16	0.27	-0.03	0.89
LO-OOA	<b>0.46</b>	0.23	0.18	<b>0.33</b>	<b>0.59</b>	0.25	0.01	0.81
<b><u>Sulfur</u></b>								
TS	0.25	0.04	-0.18	<b>0.92</b>	0.12	0.06	-0.02	0.97
SO <sub>2</sub>	0.10	0.04	-0.15	<b>0.97</b>	-0.02	0.02	-0.03	0.98
TRS	<b>0.62</b>	-0.03	-0.16	0.07	<b>0.56</b>	0.19	0.02	0.77
<b><u>Other</u></b>								
IVOCs	<b>0.32</b>	<b>0.48</b>	0.16	-0.07	<b>0.66</b>	0.00	-0.05	0.80
NH <sub>3</sub>	0.01	0.20	-0.26	-0.06	0.00	-0.05	<b>0.89</b>	0.91
CH <sub>4</sub>	<b>0.68</b>	<b>0.45</b>	0.15	-0.11	-0.05	<b>0.36</b>	-0.10	0.85
<b>Eigenvalues</b>	6.09	3.65	3.42	2.23	1.61	1.46	1.03	
<b>% of variance</b>	27.70	16.60	15.57	10.14	7.33	6.66	4.70	
<b>Cumulative variance</b>	27.70	44.30	59.86	70.01	77.34	84.00	88.70	



437 **Table S-6.** The pattern after Varimax rotation with 8 components selected.

	1	2	3	4	5	6	7	8	Communalities
<b><u>Anthropogenic VOCs</u></b>									
o-xylene	<b>0.93</b>	0.08	0.03	0.07	0.14	0.13	-0.03	0.11	0.93
1,2,3 - TMB	<b>0.91</b>	0.19	0.09	0.06	-0.01	0.02	-0.04	0.21	0.93
1,2,4 - TMB	<b>0.95</b>	0.16	0.01	0.11	0.09	0.07	-0.03	0.18	0.98
decane	<b>0.90</b>	0.26	-0.01	0.17	0.19	0.01	0.03	0.08	0.95
undecane	<b>0.82</b>	<b>0.30</b>	-0.07	0.28	0.15	0.01	0.06	0.03	0.87
<b><u>Biogenic VOCs</u></b>									
$\alpha$ -pinene	-0.04	-0.10	<b>0.97</b>	-0.11	0.06	0.00	-0.05	0.00	0.96
$\beta$ -pinene	-0.02	-0.10	<b>0.96</b>	-0.11	0.06	0.00	-0.04	-0.01	0.96
limonene	0.07	0.00	<b>0.95</b>	-0.08	0.06	0.11	-0.14	0.05	0.95
<b><u>Combustion tracers</u></b>									
NO <sub>y</sub>	0.25	<b>0.83</b>	-0.26	0.22	0.19	-0.01	0.10	-0.08	0.92
rBC	0.28	<b>0.83</b>	0.06	0.07	<b>0.33</b>	0.10	0.09	0.17	0.93
CO	<b>0.42</b>	0.18	0.04	0.01	0.07	0.11	0.06	<b>0.85</b>	0.96
CO <sub>2</sub>	0.13	0.19	<b>0.58</b>	-0.17	-0.28	<b>0.53</b>	-0.27	0.10	0.86
<b><u>Aerosol species</u></b>									
pPAH	0.06	<b>0.91</b>	-0.08	-0.14	-0.11	0.03	0.16	0.06	0.89
PM <sub>10-1</sub>	0.18	0.12	0.07	0.12	0.16	<b>0.89</b>	0.01	0.06	0.89
HOA	<b>0.42</b>	<b>0.75</b>	0.03	0.12	0.21	0.26	-0.06	0.15	0.89
LO-OOA	<b>0.46</b>	0.19	0.15	<b>0.30</b>	<b>0.65</b>	0.24	-0.04	0.15	0.87
<b><u>Sulfur</u></b>									
TS	0.28	0.03	-0.18	<b>0.92</b>	0.09	0.07	-0.01	-0.02	0.97
SO <sub>2</sub>	0.11	0.04	-0.15	<b>0.97</b>	-0.01	0.02	-0.04	0.03	0.98
TRS	<b>0.72</b>	-0.03	-0.13	0.06	<b>0.40</b>	0.23	0.11	-0.20	0.80
<b><u>Other</u></b>									
IVOCs	<b>0.35</b>	<b>0.43</b>	0.13	-0.09	<b>0.71</b>	0.00	-0.07	0.01	0.84
NH <sub>3</sub>	0.01	0.22	-0.24	-0.05	-0.05	-0.04	<b>0.92</b>	0.05	0.96
CH <sub>4</sub>	<b>0.65</b>	<b>0.47</b>	0.17	-0.11	-0.08	<b>0.35</b>	-0.10	0.16	0.86
<b>Eigenvalues</b>	5.99	3.58	3.40	2.20	1.52	1.43	1.03	1.00	
<b>% of variance</b>	27.23	16.28	15.44	10.01	6.90	6.52	4.70	4.53	
<b>Cumulative variance</b>	27.23	43.51	58.95	68.96	75.86	82.37	87.08	91.61	

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439 **Table S-7.** The pattern after Varimax rotation with 9 components selected.

	1	2	3	4	5	6	7	8	9	Communalities
<b><u>Anthropogenic VOCs</u></b>										
o-xylene	<b>0.89</b>	0.09	0.03	0.09	0.11	0.11	-0.05	0.16	<b>0.30</b>	0.95
1,2,3 - TMB	<b>0.93</b>	0.16	0.08	0.05	0.05	0.05	-0.02	0.17	-0.02	0.94
1,2,4 - TMB	<b>0.94</b>	0.15	0.01	0.11	0.11	0.08	-0.02	0.18	0.12	0.98
decane	<b>0.91</b>	0.22	-0.03	0.16	0.25	0.04	0.05	0.03	0.03	0.97
undecane	<b>0.85</b>	0.25	-0.10	0.25	0.24	0.06	0.10	-0.05	-0.08	0.94
<b><u>Biogenic VOCs</u></b>										
$\alpha$ -pinene	-0.04	-0.09	<b>0.97</b>	-0.10	0.05	0.00	-0.06	0.02	0.02	0.97
$\beta$ -pinene	-0.03	-0.10	<b>0.97</b>	-0.11	0.05	0.00	-0.05	0.00	0.02	0.96
limonene	0.09	-0.01	<b>0.94</b>	-0.09	0.08	0.13	-0.13	0.03	-0.06	0.95
<b><u>Combustion tracers</u></b>										
NO <sub>y</sub>	0.26	<b>0.82</b>	-0.26	0.22	0.22	0.00	0.11	-0.09	0.02	0.92
rBC	<b>0.31</b>	<b>0.79</b>	0.04	0.05	<b>0.41</b>	0.12	0.12	0.12	-0.10	0.94
CO	<b>0.42</b>	0.19	0.05	0.02	0.08	0.10	0.06	<b>0.87</b>	-0.02	0.98
CO <sub>2</sub>	0.16	0.17	<b>0.56</b>	-0.18	-0.22	<b>0.56</b>	-0.25	0.06	-0.17	0.86
<b><u>Aerosol species</u></b>										
pPAH	0.06	<b>0.93</b>	-0.07	-0.12	-0.11	0.01	0.14	0.09	0.03	0.93
PM <sub>10-1</sub>	0.16	0.11	0.06	0.12	0.17	<b>0.89</b>	0.02	0.06	0.10	0.89
HOA	<b>0.41</b>	<b>0.75</b>	0.03	0.13	0.23	0.25	-0.06	0.16	0.08	0.90
LO-OOA	<b>0.46</b>	0.14	0.13	0.28	0.70	0.26	-0.01	0.10	0.05	0.90
<b><u>Sulfur</u></b>										
TS	0.25	0.04	-0.17	<b>0.93</b>	0.08	0.06	-0.02	0.00	0.13	0.99
SO <sub>2</sub>	0.11	0.03	-0.15	<b>0.97</b>	0.01	0.02	-0.03	0.01	-0.05	0.99
TRS	<b>0.59</b>	0.05	-0.09	0.11	0.24	0.14	0.04	-0.04	<b>0.71</b>	0.96
<b><u>Other</u></b>										
IVOCs	<b>0.32</b>	<b>0.41</b>	0.12	-0.09	<b>0.70</b>	-0.01	-0.08	0.02	0.20	0.84
NH <sub>3</sub>	0.01	0.21	-0.24	-0.05	-0.04	-0.04	<b>0.93</b>	0.04	0.01	0.97
CH <sub>4</sub>	<b>0.65</b>	<b>0.47</b>	0.16	-0.10	-0.06	<b>0.36</b>	-0.10	0.17	0.07	0.86
<b>Eigenvalues</b>	5.84	3.44	3.37	2.19	1.58	1.47	1.03	0.95	0.74	
<b>% of variance</b>	26.54	15.63	15.30	9.98	7.19	6.66	4.69	4.31	3.38	
<b>Cumulative variance</b>	26.54	42.17	57.47	67.44	74.63	81.29	85.98	90.29	93.67	

441 **Table S-8.** The factor pattern after Varimax rotation with 11 factors selected.

	Factor 1	Factor 2	Factor 3	Factor 4	Factor 5	Factor 6	Factor 7	Factor 8	Factor 9	Factor 10	Factor 11	Commu- nalities
<b><u>Anthropogenic VOCs</u></b>												
o-xylene	<b>0.88</b>	0.08	0.03	0.10	0.13	0.07	-0.04	0.17	0.11	0.04	<b>0.32</b>	0.95
1,2,3 - TMB	<b>0.94</b>	0.16	0.07	0.05	0.11	0.05	-0.01	0.18	0.06	-0.01	-0.02	0.96
1,2,4 - TMB	<b>0.94</b>	0.15	0.01	0.11	0.08	0.08	-0.02	0.18	0.09	0.03	0.13	0.99
decane	<b>0.92</b>	0.24	-0.02	0.15	0.00	0.05	0.04	0.04	0.13	0.16	0.05	0.97
undecane	<b>0.87</b>	0.29	-0.08	0.22	-0.06	0.09	0.05	-0.05	0.03	0.22	-0.05	0.96
<b><u>Biogenic VOCs</u></b>												
$\alpha$ -pinene	-0.03	-0.08	<b>0.98</b>	-0.11	0.04	0.01	-0.08	0.02	0.02	0.00	0.00	0.98
$\beta$ -pinene	-0.02	-0.08	<b>0.98</b>	-0.12	0.02	0.02	-0.07	0.00	0.00	0.02	0.01	0.98
limonene	0.08	-0.02	<b>0.93</b>	-0.08	0.24	0.05	-0.11	0.03	0.09	0.06	-0.05	0.95
<b><u>Combustion tracers</u></b>												
NO <sub>y</sub>	0.27	<b>0.83</b>	-0.26	0.21	-0.04	0.03	0.10	-0.08	0.18	0.07	0.01	0.92
rBC	<b>0.30</b>	<b>0.81</b>	0.04	0.04	0.09	0.07	0.12	0.12	0.23	<b>0.34</b>	-0.05	0.95
CO	<b>0.41</b>	0.19	0.04	0.02	0.08	0.08	0.05	<b>0.87</b>	0.03	0.06	-0.01	0.99
CO <sub>2</sub>	0.09	0.08	<b>0.48</b>	-0.12	<b>0.77</b>	0.25	-0.14	0.05	-0.04	-0.01	-0.09	0.95
<b><u>Aerosol species</u></b>												
pPAH	0.06	<b>0.93</b>	-0.07	-0.12	0.07	0.02	0.14	0.10	-0.02	-0.20	0.01	0.95
PM <sub>10-1</sub>	0.18	0.14	0.08	0.10	0.17	<b>0.94</b>	-0.03	0.07	0.04	0.09	0.08	1.00
HOA	<b>0.40</b>	<b>0.77</b>	0.03	0.11	0.14	0.20	-0.07	0.16	0.09	0.19	0.13	0.92
LO-OOA	<b>0.45</b>	0.19	0.13	0.25	0.03	0.19	-0.04	0.11	0.27	<b>0.72</b>	0.16	0.98
<b><u>Sulfur</u></b>												
TS	0.26	0.05	-0.16	<b>0.93</b>	-0.05	0.07	-0.02	0.01	0.02	0.08	0.13	0.26
SO <sub>2</sub>	0.12	0.03	-0.15	<b>0.98</b>	-0.04	0.04	-0.03	0.01	-0.02	0.05	-0.05	0.12
TRS	<b>0.58</b>	0.06	-0.08	0.10	-0.04	0.14	0.03	-0.03	0.16	0.13	<b>0.74</b>	0.58
<b><u>Other</u></b>												
IVOCs	<b>0.34</b>	<b>0.37</b>	0.13	-0.03	-0.01	0.05	-0.03	0.03	<b>0.82</b>	0.18	0.12	1.00
NH <sub>3</sub>	0.01	0.20	-0.24	-0.04	-0.08	-0.03	<b>0.94</b>	0.04	-0.02	-0.01	0.01	1.00
CH <sub>4</sub>	<b>0.59</b>	<b>0.40</b>	0.10	-0.06	<b>0.59</b>	0.10	0.00	0.17	0.05	0.07	0.15	0.93
<b>Eigenvalues</b>	5.75	3.43	3.24	2.13	1.12	1.10	0.99	0.96	0.93	0.87	0.79	
<b>% var.</b>	26.14	15.61	14.72	9.66	5.11	4.99	4.51	4.35	4.22	3.95	3.60	
<b>% Cum. var.</b>	26.14	41.74	56.46	66.12	71.23	76.22	80.73	85.09	89.31	93.26	96.86	

442 **Table S-9.** The pattern with mixing height included after Varimax rotation with 10 components.

	1	2	3	4	5	6	7	8	9	10	Communi- nalities
<b><u>Anthropogenic VOCs</u></b>											
o-xylene	<b>0.89</b>	0.04	0.03	0.10	0.29	0.10	-0.01	-0.06	0.17	0.16	0.95
1,2,3 - TMB	<b>0.94</b>	0.17	0.10	0.04	-0.04	0.01	-0.03	-0.04	0.17	0.06	0.95
1,2,4 - TMB	<b>0.94</b>	0.13	0.03	0.10	0.11	0.09	-0.02	-0.06	0.18	0.07	0.98
decane	<b>0.92</b>	0.25	0.03	0.15	0.11	0.11	0.01	0.04	0.03	-0.04	0.97
undecane	<b>0.87</b>	<b>0.31</b>	-0.05	0.23	-0.03	0.17	0.03	0.10	-0.05	-0.11	0.96
<b><u>Biogenic VOCs</u></b>											
$\alpha$ -pinene	-0.02	-0.08	<b>0.96</b>	-0.10	0.03	0.01	-0.05	-0.11	0.01	0.02	0.96
$\beta$ -pinene	-0.01	-0.08	<b>0.96</b>	-0.11	0.04	0.02	-0.05	-0.12	-0.01	0.02	0.96
limonene	0.11	0.02	<b>0.95</b>	-0.08	0.04	0.02	-0.11	-0.06	0.03	0.12	0.96
<b><u>Combustion tracers</u></b>											
NO <sub>y</sub>	0.21	<b>0.86</b>	-0.25	0.21	0.11	0.06	0.10	0.01	-0.08	-0.02	0.92
rBC	0.29	<b>0.89</b>	0.12	0.02	0.10	0.19	0.03	0.09	0.10	-0.01	0.95
CO	<b>0.43</b>	0.20	0.04	0.01	0.00	0.08	0.02	0.05	<b>0.86</b>	0.07	0.98
CO <sub>2</sub>	0.15	0.17	<b>0.56</b>	-0.13	-0.12	0.13	-0.14	-0.12	0.08	<b>0.68</b>	0.91
<b><u>Aerosol species</u></b>											
pPAH	0.01	<b>0.86</b>	-0.09	-0.13	-0.08	-0.03	0.23	-0.18	0.12	0.17	0.90
PM <sub>10-1</sub>	<b>0.31</b>	0.22	0.05	0.15	0.12	<b>0.88</b>	0.04	-0.01	0.08	0.10	0.97
HOA	<b>0.45</b>	<b>0.79</b>	0.02	0.14	0.16	0.15	-0.02	-0.02	0.16	0.10	0.93
LO-OOA	<b>0.52</b>	<b>0.30</b>	0.21	0.26	<b>0.37</b>	<b>0.36</b>	-0.17	0.27	0.04	-0.18	0.88
<b><u>Sulfur</u></b>											
TS	0.27	0.06	-0.16	<b>0.93</b>	0.10	0.10	-0.03	0.04	-0.01	-0.02	1.00
SO <sub>2</sub>	0.11	0.05	-0.14	<b>0.97</b>	-0.06	0.06	-0.05	0.05	0.01	-0.05	0.99
TRS	<b>0.64</b>	0.01	-0.12	0.11	<b>0.63</b>	0.20	0.09	-0.04	-0.09	0.12	0.91
<b><u>Other</u></b>											
IVOCs	0.28	<b>0.50</b>	0.22	-0.07	<b>0.66</b>	0.08	-0.13	0.04	0.06	-0.18	0.87
NH <sub>3</sub>	0.00	0.22	-0.20	-0.07	-0.03	0.03	<b>0.92</b>	0.11	0.02	-0.06	0.96
CH <sub>4</sub>	<b>0.64</b>	<b>0.43</b>	0.15	-0.07	0.09	0.10	0.00	-0.07	0.17	<b>0.50</b>	0.92
Mixing height	-0.04	-0.07	<b>-0.35</b>	0.07	0.01	0.00	0.12	<b>0.90</b>	0.04	-0.07	0.96
<b>Eigenvalues</b>	6.06	3.81	3.51	2.16	1.19	1.12	1.03	1.02	0.94	0.92	
<b>% var.</b>	26.35	16.57	15.27	9.39	5.19	4.86	4.48	4.42	4.08	3.98	
<b>% Cum. var.</b>	26.35	42.92	58.19	67.58	72.77	77.63	82.11	86.53	90.61	94.60	

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445 **Table S-10.** The pattern without aerosol variables after Varimax rotation with 9 components.

	1	2	3	4	5	6	7	8	9	Communi- nalities
<b><u>Anthropogenic VOCs</u></b>										
o-xylene	<b>0.88</b>	0.02	0.04	0.09	0.12	0.10	-0.03	0.17	<b>0.32</b>	0.94
1,2,3 - TMB	<b>0.94</b>	0.07	0.12	0.04	0.11	0.03	-0.02	0.18	-0.02	0.95
1,2,4 - TMB	<b>0.94</b>	0.01	0.11	0.10	0.08	0.08	-0.02	0.19	0.13	0.98
decane	<b>0.93</b>	-0.01	0.20	0.16	0.01	0.18	0.04	0.04	0.06	0.97
undecane	<b>0.89</b>	-0.07	0.26	0.25	-0.03	0.12	0.06	-0.04	-0.03	0.94
<b><u>Biogenic VOCs</u></b>										
$\alpha$ -pinene	-0.03	<b>0.97</b>	-0.08	-0.12	0.06	0.01	-0.08	0.02	0.00	0.98
$\beta$ -pinene	-0.02	<b>0.97</b>	-0.08	-0.12	0.05	0.00	-0.08	0.00	0.01	0.98
limonene	0.08	<b>0.92</b>	-0.04	-0.08	0.27	0.10	-0.11	0.03	-0.06	0.95
<b><u>Combustion tracers</u></b>										
NO <sub>y</sub>	0.30	-0.25	<b>0.81</b>	0.23	-0.03	0.24	0.09	-0.06	0.03	0.92
rBC	<b>0.34</b>	0.04	<b>0.78</b>	0.08	0.12	<b>0.37</b>	0.11	0.13	-0.04	0.92
CO	<b>0.42</b>	0.04	0.16	0.03	0.10	0.05	0.05	<b>0.87</b>	-0.01	0.98
CO <sub>2</sub>	0.10	<b>0.46</b>	0.06	-0.10	<b>0.84</b>	-0.02	-0.13	0.06	-0.05	0.96
<b><u>Aerosol species</u></b>										
pPAH	0.07	-0.07	<b>0.94</b>	-0.13	0.08	-0.06	0.11	0.12	0.03	0.95
<b><u>Sulfur</u></b>										
TS	0.26	-0.15	0.03	<b>0.94</b>	-0.05	0.03	-0.02	0.01	0.14	1.00
SO <sub>2</sub>	0.12	-0.14	0.02	<b>0.98</b>	-0.05	-0.03	-0.03	0.02	-0.04	0.99
TRS	<b>0.59</b>	-0.07	0.04	0.14	-0.01	0.19	0.02	-0.02	<b>0.75</b>	0.97
<b><u>Other</u></b>										
IVOCs	<b>0.35</b>	0.13	<b>0.32</b>	-0.03	0.00	<b>0.84</b>	-0.03	0.05	0.15	0.98
NH <sub>3</sub>	0.01	-0.23	0.21	-0.05	-0.10	-0.01	<b>0.94</b>	0.05	0.01	1.00
CH <sub>4</sub>	<b>0.61</b>	0.09	<b>0.36</b>	-0.05	<b>0.59</b>	0.08	0.00	0.18	0.16	0.92
<b>Eigenvalues</b>	5.54	3.16	2.60	2.08	1.20	1.03	0.97	0.94	0.77	
<b>% var.</b>	29.15	16.63	13.68	10.96	6.33	5.40	5.11	4.96	4.03	
<b>% Cum. var.</b>	29.15	45.79	59.46	70.43	76.76	82.16	87.26	92.23	96.25	

446

447

448 **Table S-11.** Criteria for number of components extracted by PCA.

Criterion	Number of components extracted	% Variance explained (after rotation)
Scree test 1	5	81.0%
< 5% variance	6	85.2%
Latent root	7	88.7%
≥ 95% cumulative variance	10	95.5%
Scree test 2	11	96.9%

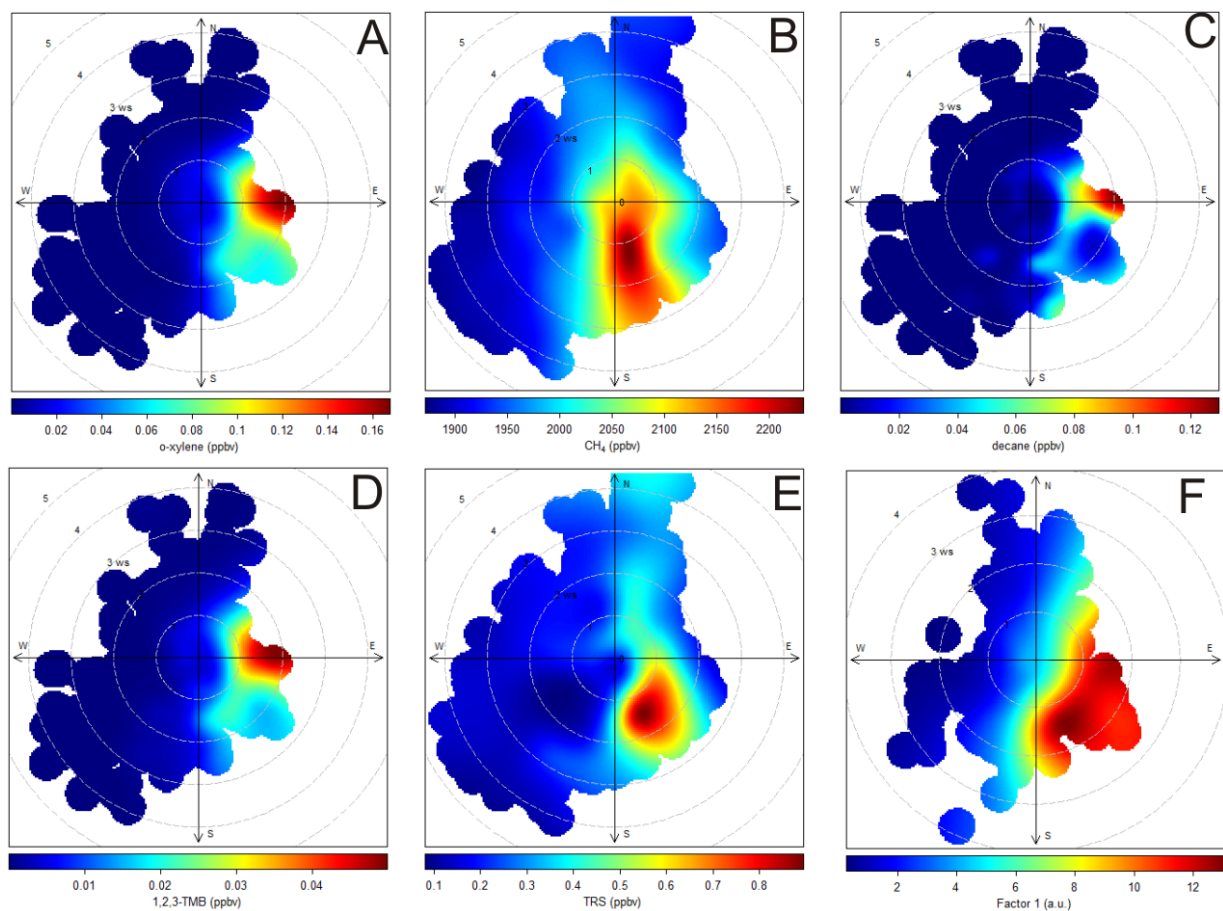
449

450

451 **Table S-12.** Association of IVOCs with relevant components.

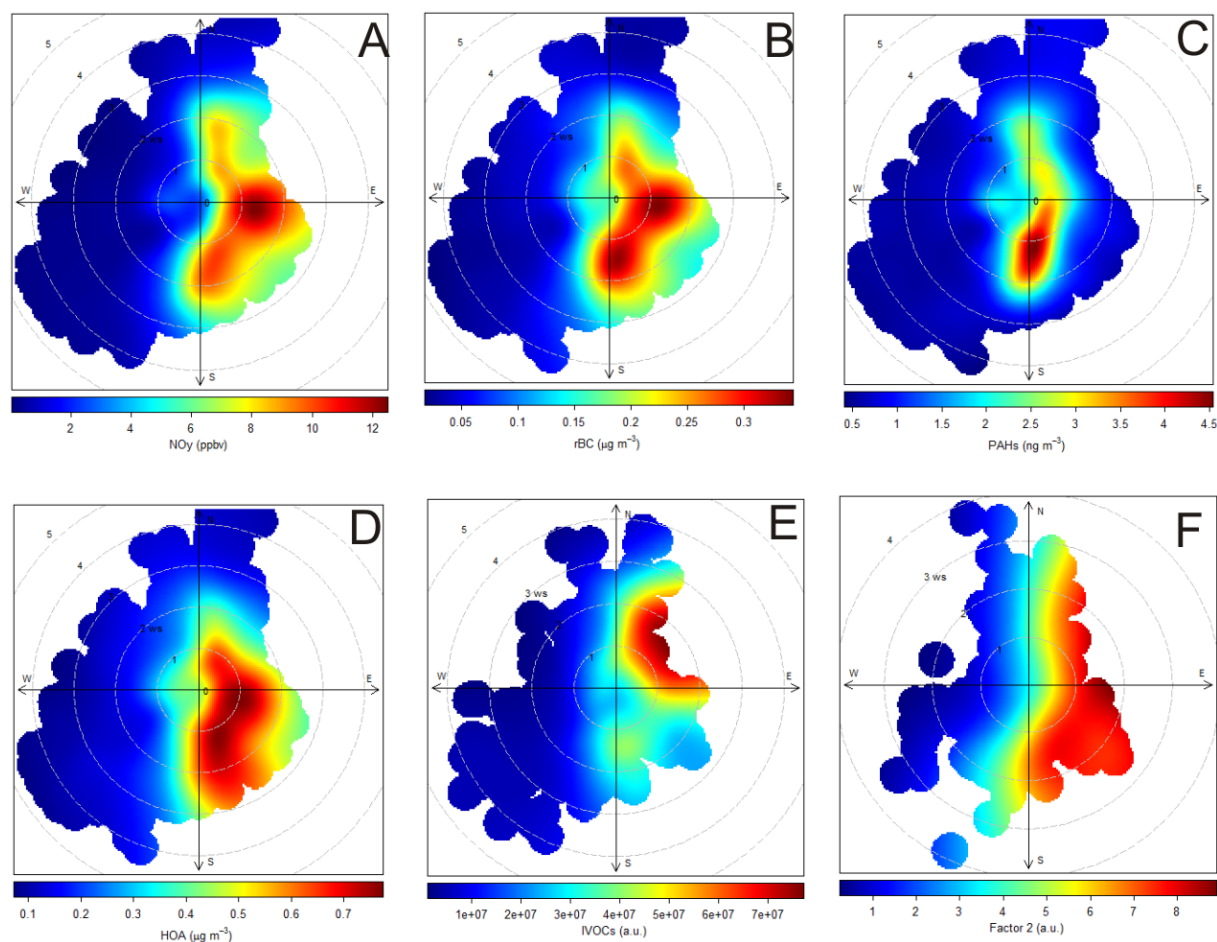
# of components in solution	Oil sands surface mining facilities (Component 1)	Mine fleet and operations (Component 2)	Mine face (Component 5)
5	0.47	0.61	n/a
6	0.31	0.43	n/a
7	0.32	0.48	0.66
8	0.35	0.43	0.71
9	0.32	0.41	0.70
10	0.31	0.39	0.74
11	0.34	0.37	n/a

452



**Figure S-3.** Bivariate polar plots associated with component 1 for the optimum primary pollutant solution (Table 5.). (A) o-xylene, (B) CH<sub>4</sub>, (C) decane, (D) 1, 2, 3-TMB, (E) TRS, (F) and component 1.

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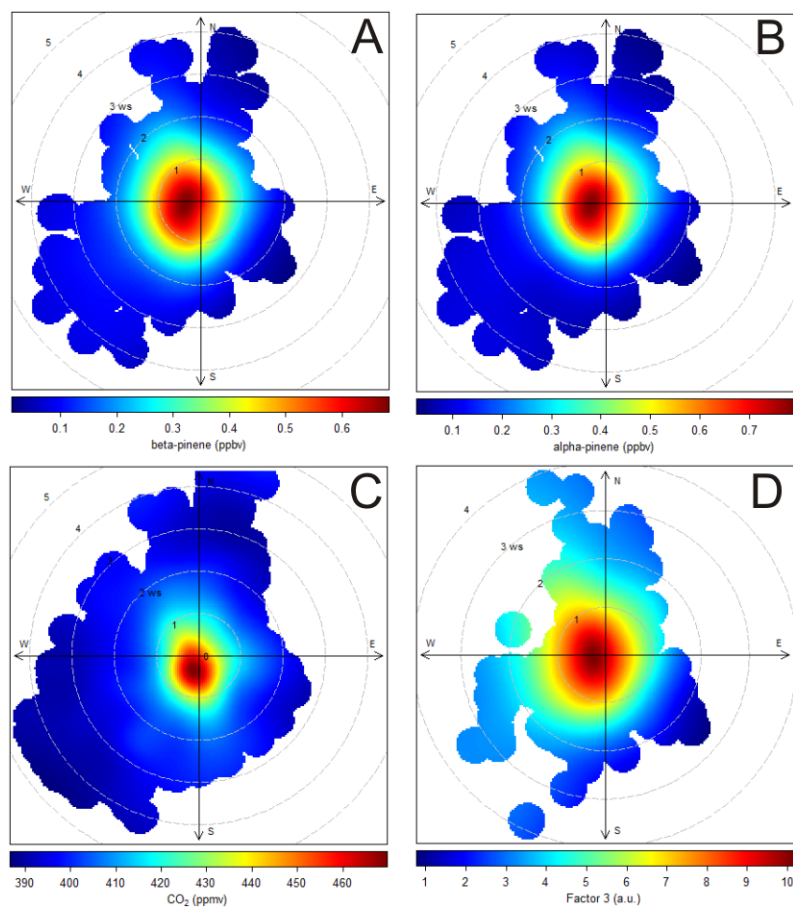
459

**Figure S-4.** Bivariate polar plots associated with component 2 for the optimum primary pollutant solution (Table 5.). (A)  $\text{NO}_y$ , (B) rBC, (C) PAHs, (D) HOA, (E) IVOCs, (F) and component 2.

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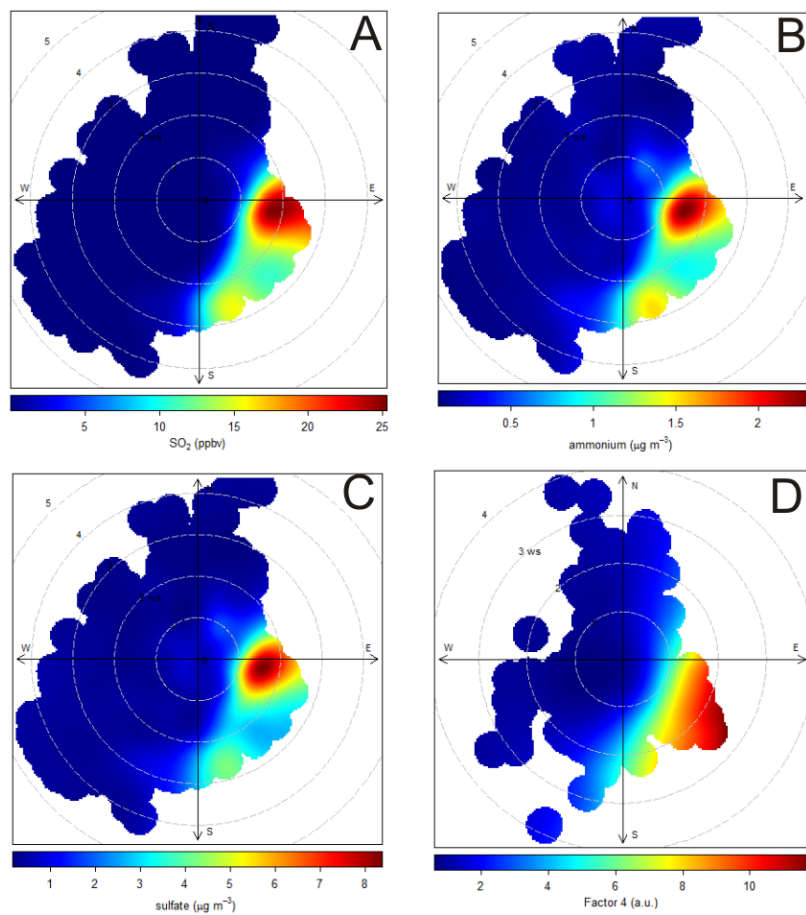


463

464 **Figure S-5.** Bivariate polar plots associated with component 3 for the optimum primary pollutant  
 465 solution (Table 5.). **(A)**  $\beta$ -pinene, **(B)**  $\alpha$ -pinene, **(C)** CO<sub>2</sub>, **(D)** and component 3.

466

467

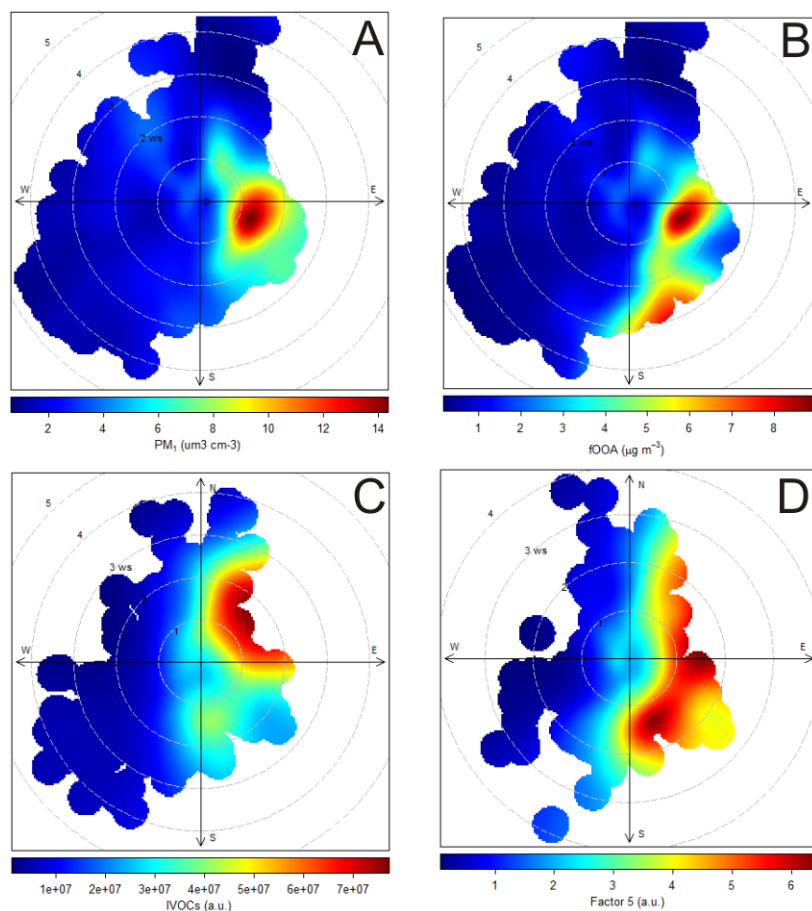


468

469 **Figure S-6.** Bivariate polar plots associated with component 4 for the optimum secondary pollutant  
 470 solution (Table 7). **(A)** SO<sub>2</sub>, **(B)** NH<sub>4</sub><sup>+</sup><sub>(p)</sub>, **(C)** SO<sub>4</sub><sup>2-</sup><sub>(p)</sub>, **(D)** and component 4.

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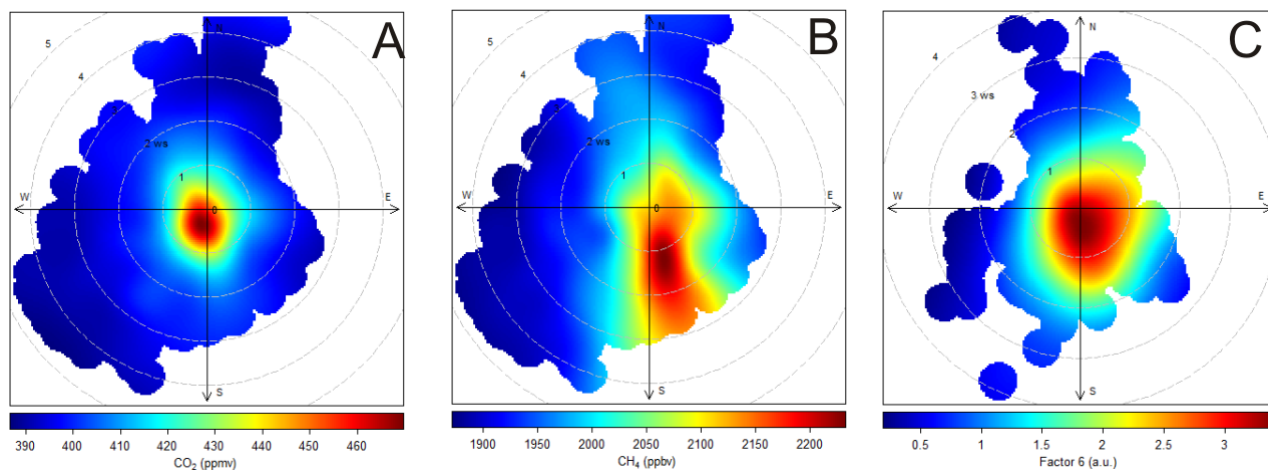


473

474 **Figure S-7.** Bivariate polar plots associated with component 5 for the optimum secondary pollutant  
 475 solution (Table 7). **(A)** PM<sub>1</sub> (11-component solution), **(B)** LO-OOA, **(C)** IVOCs, and **(D)** component 5.

476

477

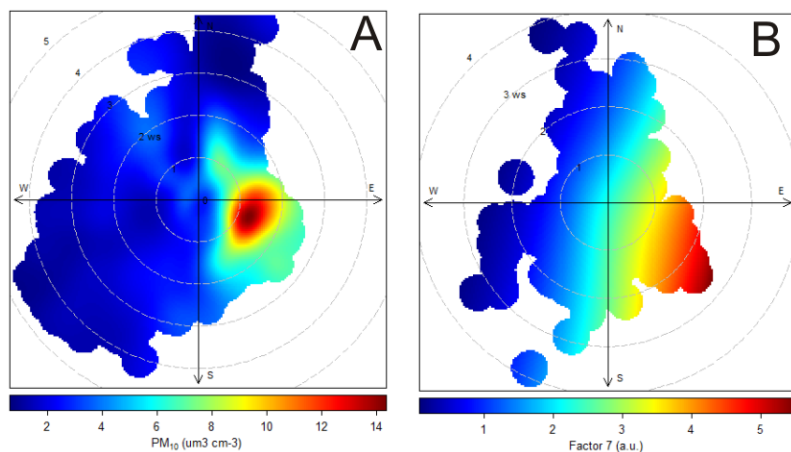


478

479 **Figure S-8.** Bivariate polar plots associated with component 6 for the optimum primary pollutant  
480 solution (Table 5). **(A)** CO<sub>2</sub>, **(B)** CH<sub>4</sub>, and **(C)** component 6.

481

482

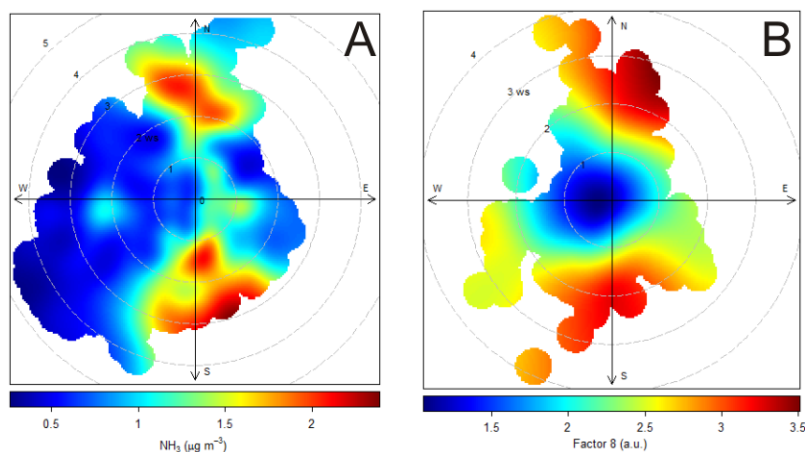


483

484 **Figure S-9.** Bivariate polar plots associated with component 7 for the optimum primary pollutant  
485 solution (Table 5). **(A)** PM<sub>10-1</sub>, **(B)** and component 7.

486

487

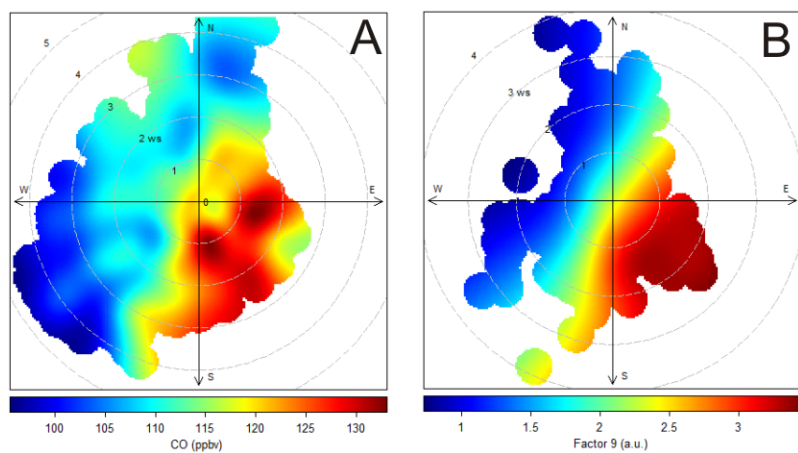


488

489 **Figure S-10.** Bivariate polar plots associated with component 8 for the optimum primary pollutant  
 490 solution (Table 5). **(A)**  $\text{NH}_3$ , **(B)** and component 8.

491

492

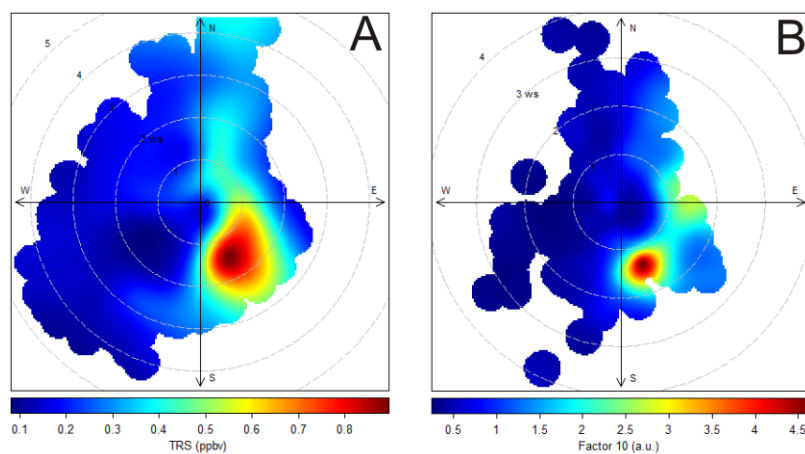


493

494 **Figure S-11.** Bivariate polar plots associated with component 9 for the optimum primary pollutant  
 495 solution (Table 5). **(A)** CO, and **(B)** component 9.

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497



498

499 **Figure S-12.** Bivariate polar plots associated with component 10 for the optimum primary pollutant  
500 solution (Table 5). **(A)** TRS, **(B)** and component 10.

501

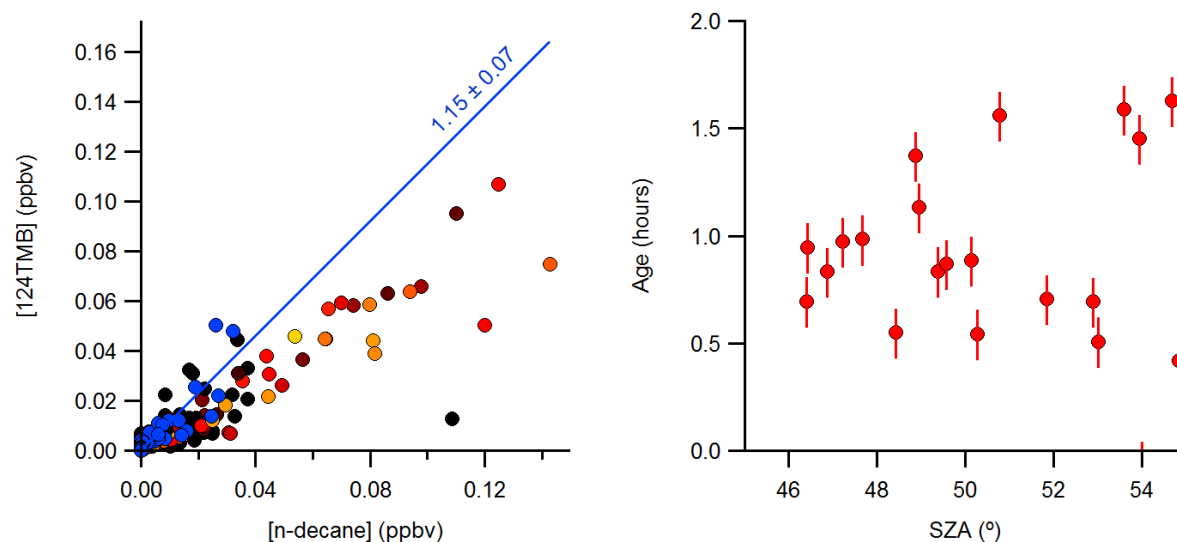
## Estimate of photochemical age

Photochemical age was calculated using the method outlined by Borbon et al. (2013), but substituting n-decane for benzene since the latter was not quantified. The photochemical age of an air mass,  $\Delta t$  was calculated from the observed concentrations of 124-trimethylbenzene (124TMB) and n-decane using:

$$\Delta t = \frac{1}{[\text{OH}] \times (k_{124\text{TMB}} - k_{\text{decane}})} \times \left[ \ln \left( \frac{[\text{124TMB}]}{[\text{decane}]} \right)_{t=0} - \ln \left( \frac{[\text{124TMB}]}{[\text{decane}]} \right) \right] \quad (\text{S-1})$$

where  $k_{124\text{TMB}} = 3.25 \times 10^{-11} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$  and  $k_{\text{decane}} = 1.10 \times 10^{-11} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$  are rate coefficients for reaction of OH with 124-TMB and n-decane (at 298 K), respectively, whose values were taken from Seinfeld and Pandis (2006). The ratio of [124TMB] to [decane] at the point of emission (time  $t = 0$ ) was estimated from a plot of [124TMB] to [n-decane] (Figure S-13, left-hand side) and a straight-line fit to the nocturnal data (assumed to be unaffected by oxidation and shown in blue color). The slope of this line was  $1.15 \pm 0.07$  ( $r^2 = 0.84$ ). Daytime data (color-coded by solar zenith angle, SZA) exhibit lower ratios of [124TMB]/[decane] as a result of the faster oxidation of 124TMB by OH.

Shown in Figure S-13 on the right-hand side is a plot of the photochemical age, calculated using equation (S-1) and an assumed [OH] of  $7 \times 10^6 \text{ molecules cm}^{-3}$  taken from Liggio et al. (2016), as a function of SZA (filtered for peak OH of 11:00 and 16:00 local time). The error bars indicate ages calculated using emission factors of 1.08 and 1.22, respectively. The average ( $\pm 1$  standard deviation) photochemical age is  $1.0 \pm 0.4 \text{ hr}$ . This photochemical age applies mainly to component 1; we assume that the photochemical ages of sources associated with other components were similar.



**Figure S-13. (A)** Plot of 124TMB mixing ratios against mixing ratios of n-decane, color-coded by solar zenith angle. The blue data points were collected at night. **(B)** Photochemical age calculating using equation S-1 plotted as a function of solar zenith angle.

In their analysis of IVOC photochemical aging, Zhao et al. (2014) estimated an average  $k_{OH}$  for diesel-exhaust IVOCs of  $1.8 \times 10^{-11} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$  (though their estimated rate coefficients varied and increased slightly with volatility bin between about 1 and  $3 \times 10^{-11} \text{ cm}^3 \text{ molecule}^{-1} \text{ s}^{-1}$ ). From this, we calculate a pseudo first-order lifetime of 130 min (2.17 hr) with respect to IVOC oxidation by OH during daytime. Using a photochemical age of  $1.0 \pm 0.4$  hr, we calculate that between 25% and 50% of the emitted IVOC is (potentially) oxidized during daytime. Photochemical aging will affect data collected during the daytime hours (from ~11 am to ~4 pm) or ~25% of the data (56 out of 218 data points) used in the PCA and likely resulted in partial conversion of IVOCs to SOA.



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