

The impacts of regional shipping emissions on the chemical characteristics of coastal submicron aerosols near Houston, TX

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Abstract. The air quality of the Texas Gulf Coast region historically has been influenced heavily by regional shipping emissions. However, the effects of the recently established North American Emissions Control Area on aerosol concentrations and properties in this region are presently unknown. In order to better understand the current sources and processing mechanisms influencing coastal aerosol near Houston. a high-resolution time-of-flight aerosol mass spectrometer (HR-ToF-AMS) was deployed for 3 weeks at a coastal location during May-June 2016. Total mass loadings of organic and inorganic non-refractory aerosol components during onshore flow periods were similar to those published before establishment of the regulations. Based on estimated methanesulfonic acid (MSA) mass loadings and published biogenic MSA/non-sea-salt sulfate (nss-SO₄) ratios, an average of over 75 % of the observed nss-SO₄ was from anthropogenic sources, predominantly shipping emissions. Mass spectral analysis indicated that for periods with similar backwardtrajectory-averaged meteorological conditions, air masses influenced by shipping emissions had an increased mass fraction of ions related to carboxylic acids and larger oxygen-tocarbon ratios than those that avoided shipping lanes, suggesting that shipping emissions increase marine organic aerosol (OA) oxidation state. Amine fragment mass loadings were correlated positively with anthropogenic nss-SO₄ during on-

shore flow, implying anthropogenic-biogenic interaction in marine OA production. Model calculations also suggest that advection of shipping-derived aerosol may enhance inland aqueous-phase secondary OA production. These results imply a continuing role of shipping emissions on aerosol properties over the Gulf of Mexico and suggest that further regulation of shipping fuel sulfur content will reduce coastal submicron aerosol mass loadings near Houston.

1 Introduction

Seaborne trade is a relatively inexpensive and efficient mechanism for transporting goods across the globe (IMO, 2012). As a result, such transportation is thought to account for more than 90 % of global trade volume (Eyring et al., 2010; IMO, 2012) and has been growing rapidly in the past 2 decades (Lack et al., 2009; Eyring et al., 2010; Tournadre, 2014; Johansson et al., 2017). As large commercial shipping vessels historically have had little or inconsistent regulation in international waters, they frequently burn low-quality residual fuel oils, leading to considerable emissions of sulfur dioxide (SO₂), nitrogen oxides (NO_x), and particulate matter (PM) (Lack et al., 2009; Murphy et al., 2009; Czech et al., 2017). Recently, increasing attention has been paid to the impact of these emissions on ambient PM mass loadings in coastal areas, with notable contributions in Europe (Viana et al., 2014, and references therein; Aksoyoglu et al., 2016), Asia (Zhao et al., 2012; Liu et al., 2016) and the United States (Vutukuru and Dabdub, 2008; Agrawal et al., 2009). Coastal populations exposed to these emissions are subsequently affected by numerous negative health impacts. Corbett et al. (2007) estimated that shipping activity was responsible for 60 000 global premature mortalities annually. More recent studies have confirmed links between shipping emissions and increased hospitalizations (Tian et al., 2013).

The Port of Houston is the second largest in the United States (U.S.) by tonnage (Port of Houston, 2017), and the Gulf of Mexico has a high density of marine vessel emissions relative to many other marine locations (Tournadre, 2014; Johansson et al., 2017); however, relatively little research has aimed to characterize the impact of shipping emissions on Houston air quality. During the Texas Air Quality Study and Gulf of Mexico Atmospheric Composition and Climate Study 2006, measurements on board the R/V Brown were used to characterize aerosol sources over the Gulf of Mexico (Bates et al., 2008; Russell et al., 2009). Measured submicron aerosol sulfate (SO₄) mass loadings during periods of onshore flow were significantly larger than expected for a marine environment, leading Bates et al. (2008) to conclude that shipping emissions contributed heavily to total submicron aerosol mass. Russell et al. (2009) further determined that an "oil combustion/refining" organic factor accounted for 33 %-68 % of organic aerosol (OA) mass during onshore flow periods. Using a large-scale three-dimensional air quality model, Caiazzo et al. (2013) calculated that in 2005 marine vessel emissions increased annual average PM mass loadings across the Texas Gulf Coast by ~ 0.5 to $1 \,\mu g \, m^{-3}$, leading to 645 estimated premature mortalities in Texas.

Recent concerns over the health impacts of marine vessel emissions led to the establishment of the North American Emissions Control Area (ECA; U.S. Environmental Protection Agency (EPA), 2010). Prior to establishment of the ECA, multiple studies demonstrated that shipping emissions of PM were related to fuel sulfur content (FSC) (Kasper et al., 2007; Lack et al., 2009, and references therein), leading to the requirement that shipping vessels within 200 nautical miles of the U.S. and Canadian coast reduce their FSC from the commonly utilized 3%-4% (by mass) to only 1%. In 2015, the limit was reduced to 0.1 % (Zetterdahl et al., 2017). In order to comply with these regulations, marine vessels typically switch from low-grade heavy fuel oil to marine gas oil or marine diesel oil at the ECA boundary; however, low-FSC residual fuels have also recently become available (Wan et al., 2016; Czech et al., 2017). Numerous studies have demonstrated that such fuel switching dramatically reduces emissions of SO₂, SO₄, primary OA (POA), and black carbon (Lack et al., 2011; Browning et al., 2012; Tao et al., 2013; Zetterdahl et al., 2017).

Using the U.S. Interagency Monitoring of Protected Visual Environments network and positive matrix factorization (PMF) modeling, Kotchenruther (2016) determined that the average decrease in annual PM2.5 (that with diameters less than or equal to 2.5 µm) from residual fuel combustion (i.e., shipping emissions) in U.S. coastal locations due to establishment of the ECA (i.e., pre-2012 to 2016) was 74.1 %. However, at two sites along the Gulf Coast (located in Louisiana and Florida), the average reduction was only 35%-50% (Kotchenruther, 2016). While the reason for the difference between the gulf sites and the rest of the country is currently unclear, it is nevertheless evident that the implementation of the ECA may have drastically changed the speciation and total mass loading of aerosol over the Gulf of Mexico, presenting the need for further research on this source.

Shipping emissions also may have numerous secondary effects on marine aerosol. Models indicate that shippingrelated NO_x emissions likely elevate hydroxyl radical (OH) concentrations within the marine boundary layer (MBL) (Chen et al., 2005; Kim et al., 2009, 2013), potentially impacting the oxidation state of marine OA. Furthermore, production of the two most commonly identified components of marine secondary OA (SOA), methanesulfonic acid (MSA) and dimethyl/diethylamines (Facchini et al., 2008; Claeys et al., 2009; Rinaldi et al., 2010), may be enhanced in the presence of shipping emissions (Gaston et al., 2010; Sorooshian et al., 2015a). Finally, shipping-related SO₄ should increase submicron mass loadings of aerosol liquid water (ALW), which may subsequently impact aqueous processing of water-soluble organics (Carlton and Turpin, 2013). These effects are difficult to model on a global scale due to the complexities of accurately simulating the photochemistry and physical transport of shipping plumes (Kim et al., 2009), making field measurements useful to evaluate these hypotheses.

In the present study, 3 weeks of coastal air measurements were performed near Houston, TX, to investigate the impact of marine vessel emissions on ambient aerosol mass and composition. Specific focus was placed on apportioning anthropogenic and biogenic sources of SO_4 ; attributing anthropogenic SO_4 to marine vessel emissions; investigating links between marine vessel emissions and measured OA; and exploring whether these emissions appear to influence OA composition, amine/MSA aerosol formation, or ALW.

2 Experimental methods

2.1 Sampling site characterization

Atmospheric measurements were conducted during 24 May– 14 June 2016, at a private coastal home southwest of Galveston, Texas (29.074° N, 95.125° W). Figure 1 presents an overview of the sampling location. The site is approximately



Figure 1. Map depicting the Houston region, the coastal study site (star), and the location of recent stationary campaigns that characterized aerosol dynamics in Houston.

75 km directly south of the Houston Ship Channel (HSC) and is therefore a similar distance from Houston's urban core. In addition, the primary inlet to Galveston Bay used for commercial shipping is about 45 km to the northeast. The nearest road - Highway 257, just north of the site - connects the cities of Galveston and Freeport, TX, and receives relatively little traffic. As a result, this location is likely to be less influenced by primary anthropogenic emissions than recent campaigns in Houston that took place closer to the urban core (Cleveland et al., 2012; Bean et al., 2016; Leong et al., 2017; Wallace et al., 2018). Instruments including a high-resolution time-of-flight aerosol mass spectrometer (HR-ToF-AMS, Aerodyne, Inc.) and those measuring traces gases and meteorological parameters were housed inside the University of Houston/Rice University Mobile Air Quality Laboratory (MAQL), which was stationed outside of the private home and has been described previously (Leong et al., 2017). As measurements were performed during a single 3week period, presented results should be viewed with this limitation in mind. Ultimately, longer-time-frame studies are warranted to test whether results presented here are consistently observed.

2.2 HR-ToF-AMS operation

The chemical composition of non-refractory submicron PM (NR-PM₁) was determined through the use of a HR-ToF-AMS (DeCarlo et al., 2006). Numerous detailed descriptions of HR-ToF-AMS operation can be found elsewhere (De-Carlo et al., 2006; Canagaratna et al., 2007). Air flow was drawn into the HR-ToF-AMS through a 2.5 µm cut diameter Teflon[®]-coated cyclone located on top of the MAQL mast approximately 6 m above ground level. Incoming air is transmitted through a 100 µm critical orifice, after which particles are focused into a beam through the use of an aerodynamic lens and accelerated under high vacuum (10^{-5} Torr) into the sizing chamber. After passing the sizing chamber, non-refractory chemical components are flash-vaporized at approximately 600 °C and ionized at 70 eV. Ionized mass fragments are then directed into the time-of-flight mass spectral detection region. For this study, the HR-ToF-AMS was operated in V mode (higher signal, less mass-to-charge (m/z)) resolution compared to the alternative W mode), and data were collected over 80 s intervals. A Nafion dryer was placed upstream of the HR-ToF-AMS inlet to maintain a sampling line relative humidity (RH) below 40 %.

2.3 HR-ToF-AMS data analysis

The HR-ToF-AMS data were analyzed with the SQUIRREL v 1.57I and PiKA v 1.16I (D. Sueper, University of Colorado-Boulder) software packages within Igor Pro (Wavemetrics, Inc.). The collection efficiency (CE) of the HR-ToF-AMS, which is influenced by sampling line RH as well as particle composition, was determined using the compositiondependent calculator within the SQUIRREL and PiKA software packages (Middlebrook et al., 2011). This method produced a CE of 0.5 for the majority of the campaign (89%) of the time). High-resolution analysis was performed on each ion in the m/z range 10–125, and elemental analysis of organic composition was performed using the improvedambient method (Canagaratna et al., 2015). The ionization efficiency of the HR-ToF-AMS with respect to nitrate (NO₃) was calibrated before and after the campaign using 350 nm ammonium nitrate (NH₄NO₃) particles following standard procedures. In order to calculate campaign-averaged detection limits, filtered air was sampled every 2 days for approximately 30 min at a time, and the detection limit was calculated as 3 times the standard deviation of the filter measurements. Detection limits are provided in Table S1 in the Supplement.

2.4 Positive matrix factorization

Positive matrix factorization analysis (Paatero and Tapper, 1994) was performed on the high-resolution HR-ToF-AMS

mass spectral dataset in order to further investigate potential sources and transformation processes of measured OA. The PMF technique has been applied extensively in urban (Ulbrich et al., 2009; Ng et al., 2010), rural/downwind (Crippa et al., 2014, and references therein), and coastal locations (Hildebrandt et al., 2010, 2011; Schmale et al., 2013) to characterize classes of compounds that constitute OA. The PMF model assumes that the time series of organic mass spectra can be divided into a number of temporally unvarying components. These components, defined by their fixed mass spectra, contribute varying amounts of organic mass to the total organic signal at each time. Details on PMF and the resulting factors are included in the Supplement.

2.5 HYSPLIT backward-trajectory calculation

Analysis of air mass history is often a useful tool for characterizing likely sources and processes affecting measured aerosol composition. As a result, 120h backward trajectories were calculated at heights of 100, 200, 300, 400, and 500 m using the Hybrid Single-Particle Lagrangian Integrated Trajectory (HYSPLIT) model (Draxler and Rolph, 2003) for every hour during the campaign. Meteorological data at a resolution of $12 \text{ km} \times 12 \text{ km}$ were obtained from the North American Mesoscale Forecast System (NAMS) archive (https://ready.arl.noaa.gov/archives.php, last access: 30 August 2018) through the HYSPLIT software. Recorded meteorological parameters such as solar flux $(W m^{-2})$, mixing-layer depth, and precipitation were averaged for each trajectory to provide insight into influences of photochemistry, mixing, and possible wet deposition during transport. In addition, the overall length of each 5-day trajectory was used to represent an average wind speed, as knowledge of historical wind speed is important for predictions concerning the influence of POA particles in ocean environments (Zorn et al., 2008; Russell et al., 2010; Ovadnevaite et al., 2011).

2.6 Weighted potential source contribution function

In order to provide further insight into likely aerosol source regions during the campaign, the weighted potential source contribution function (WPSCF) was applied to the dataset. The WPSCF combines measured mass loadings of atmospheric species with air mass trajectories to determine probable source locations. The WPSCF method has been used to study regional sources of air pollutants at different receptor sites (Hopke et al., 1995; Zhu et al., 2011; Guo et al., 2014). For this study, the WPSCF analysis utilized HYSPLIT trajectories described previously. The spatial area covered by the trajectories was divided into a grid of $0.5^{\circ} \times 0.5^{\circ}$ cells, and the number of trajectory segment end points located in each cell for five different starting heights (100, 200, 300, 400, and 500 m) was determined. Incorporation of multiple starting heights accounts for the general clockwise rotation of air mass backward trajectories with altitude. While these cell sizes are relatively small (typical values $\sim 0.5-2^{\circ}$), this study was particularly focused on attribution of measured SO₄ to specific locations within the Gulf of Mexico (i.e., shipping lanes), which requires a small cell size.

In order to calculate the WPSCF value for each cell, the total potential source contribution function (TPSCF) value is first calculated and then weighted. The value of the TP-SCF function for a specific grid cell (i, j) is calculated using (Hopke et al., 1995; Guo et al., 2014)

$$\text{TPSCF}_{i,j} = \frac{\sum m_{i,j}^k}{\sum n_{i,j}^k},\tag{1}$$

where $n_{i,j}^k$ represents the total number of trajectory segment end points located within cell *i*, *j* for height *k*, while $m_{i,j}^k$ represents the number of these end points that also correspond to measured values of a specific species above a critical value, in this case the 75th percentile (Hopke et al., 1995; Guo et al., 2014).

In the case of highly variable air mass trajectories or strong local sources during a campaign, distant grid cells that were intersected by only a small number of trajectories may be incorrectly assumed to represent likely sources. To prevent this, a weighting function is applied to TPSCF values based on the $n_{i,j}^k$ value, with higher weight given to cells that were intersected by more trajectories. The weighting method, based on the power of the number of trajectories at a specific height, is (Guo et al., 2014)

$$W\left(\sum n_{i,j}^{k}\right) = \left\{ \begin{array}{l} 1, T^{0.7} < \sum n_{i,j}^{k} \\ 0.7, T^{0.56} < \sum n_{i,j}^{k} \le T^{0.7} \\ 0.42, T^{0.42} < \sum n_{i,j}^{k} \le T^{0.56} \\ 0.17, \sum n_{i,j}^{k} \le T^{0.42} \end{array} \right\},$$
(2)

where *T* represents the total number of trajectories calculated at each specific height. The WPSCF value is then calculated by applying the relevant weights to each cell.

$$WPSCF_{i,j} = W_{i,j} \times TPSCF_{i,j}$$
(3)

2.7 MSA calibration

Methanesulfonic acid is widely regarded as a robust indicator of SOA production from marine sources (Facchini et al., 2008; Crippa et al., 2013; Schmale et al., 2013; Ovadnevaite et al., 2014). In addition, MSA is often the most abundant identifiable component of marine OA (Facchini et al., 2008; Claeys et al., 2009). Recent work has identified that MSA mass loadings can be quantified in near-real time by the HR-ToF-AMS provided that accurate instrument-specific calibrations are performed (Zorn et al., 2008; Ovadenvaite et al., 2014; Huang et al., 2017). As MSA fragments into both organic and inorganic SO₄-containing ions within the HR-ToF-AMS, accurate mass prediction requires reconstruction of the compound based on knowledge of the fragmentation pattern in the specific instrument being used (Zorn et al., 2008). As such, calibrations were performed following the procedure of Ovadnevaite et al. (2014).

A 0.02 % aqueous solution of MSA (Sigma-Aldrich, >99.0 % purity) was nebulized by a TSI, Inc. atomizer (model 3076) and passed through a differential mobility analyzer (BMI, Inc.) to size-select particles 300 nm in mobility diameter. These particles were then measured by the HR-ToF-AMS. Mass spectra from two separate calibrations are provided in Fig. S13. While MSA fragments into a variety of ions within the HR-ToF-AMS (CH₃⁺, CHS⁺, CH₃SO₂⁺, SO^+ , SO_2^+ , etc.), the $CH_3SO_2^+$ ion is thought to originate almost exclusively from MSA, as other organosulfate standards measured by the HR-ToF-AMS show negligible contributions to $CH_3SO_2^+$ (Huang et al., 2015). Therefore, MSA mass loadings during the campaign were calculated based on the ratio of this ion to the total MSA mass measured during the calibrations (Huang et al., 2015, 2017). The average ratio measured during the two calibrations (18.18) was similar to that determined from the calibration of Huang et al. (2017) (23.81).

2.8 Distinction between anthropogenic and biogenic nss-SO₄

The MSA measurements also allow estimation of the relative contributions of biogenic and anthropogenic (primarily due to shipping) sources of non-sea-salt sulfate (nss-SO₄) in marine environments. In many studies attempting to apportion the impact of shipping emissions on measured aerosol mass, ratios of trace metals specific to heavy-fuel-oil combustion are used as tracers (Zhao et al., 2012; Viana et al., 2014; Kotchenruther, 2016). However, in cases where such data are unavailable, biogenic sulfur sources, based on the oxidation chemistry of dimethyl sulfide (DMS), should produce a latitude-specific biogenic MSA/nss-SO₄ ratio, presenting a metric to apportion biogenic and anthropogenic nss-SO₄ (Jung et al., 2014). Specifically, DMS oxidation, which primarily occurs through initial reaction with OH, can proceed through either an abstraction or addition pathway (Hynes et al., 1986). The addition pathway, which is favored at lower temperatures prevalent at higher latitudes, mainly produces dimethylsulfoxide and MSA. The abstraction pathway, favored in higher temperatures, primarily produces SO₂ and therefore eventually nss-SO₄ (Hynes et al., 1986; Jung et al., 2014). As a result, previous long-distance remote transoceanic cruises have observed significant latitudinal gradients in the MSA/nss-SO₄ ratio in both the Atlantic and Pacific oceans (Jung et al., 2014; Huang et al., 2017), with consistently larger values at high latitudes.

As nss-SO₄ measured in marine environments is often produced by a combination of anthropogenic and biogenic sources, multiple linear regression (MLR) analysis is often used to extract the biogenic MSA/nss-SO₄ ratio from ambient marine aerosol data. The MLR technique assumes that marine nss-SO₄ is produced from a biogenic source, which can be traced with MSA mass loadings (used as one predictor variable), and an anthropogenic source, which can be traced using concentrations of heavy metals emitted by shipping vessels (e.g., antimony) (used as the second predictor variable) (Savoie et al., 2002). Previously published agreement between measured and predicted nss-SO₄ using the MLR method was robust ($R^2 > 0.7$) (Savoie et al., 2002). For this study, the biogenic MSA/nss-SO₄ ratio (0.053) determined by Savoie et al. (2002) using MLR at Bermuda was utilized to apportion biogenic versus anthropogenic sources of nss- SO_4 , as Bermuda is the closest location to our sampling site in terms of latitude (32° N at Bermuda versus 29° N at our sampling site). In addition to being collected at the closest location to our sampling site, the ratio extracted at Bermuda is the lowest reported ratio in the literature, to the authors' knowledge. A more recent study by Lin et al. (2012) quantified the biogenic MSA/nss-SO₄ ratio of submicron marine aerosol using sulfur isotopic data at varying latitudes across the Atlantic Ocean. The authors reported ratios similar to or larger than 0.053 at all sampled latitudes. As a result, the application of other published ratios will only increase the fraction of nss-SO₄ attributed to anthropogenic sources.

However, as the biogenic MSA/nss-SO₄ ratio was originally determined using samples of total suspended particulate matter (i.e., no size cutoff) application of the ratio to PM₁ data should produce an upper-limit estimate of the anthropogenic fraction of marine nss-SO₄. Briefly, as previous field and laboratory studies have noted that MSA solubility decreases with solution acidity (Kerminen et al., 1997; Jung et al., 2014), the presence of acidic sulfate aerosol can shift the size distribution of MSA towards larger, more alkaline particles relative to sulfate (a stronger acid) (Jung et al., 2014). This effect, if substantial, could cause the HR-ToF-AMS to report a lower observed MSA/nss-SO₄ ratio than would be observed by an instrument measuring both submicron and super-micron PM. However, Saltzman et al. (1983) found that the size distributions of MSA and SO₄ measured in the Gulf of Mexico were quite similar, with around 75 % of MSA and approximately 87 % of nss-SO₄ contained within submicron particles, suggesting that the overall uncertainty resulting from this effect is small.

Quantification of anthropogenic nss-SO₄ also requires that the contribution of sea salt sulfate (ss-SO₄) be determined. Using laboratory calibrations, the ss-SO₄ mass loading measured by the HR-ToF-AMS when sampling a sea salt standard (Lake Products Co., ASTM D1141) is approximately $26 \pm 2\%$ of the corresponding chloride mass loading. Using this ratio and measured chloride mass loadings during the campaign, ss-SO₄ contributed only $0.4 \pm 0.4\%$ of the total SO₄ mass loading. Therefore, to produce an estimate of anthropogenic nss-SO₄, the HR-ToF-AMS estimate of MSA is divided by the biogenic MSA/nss-SO₄ ratio published by Savoie et al. (2002) to produce a "biogenic" mass loading of nss-SO₄. This amount is subtracted from the total nss-SO₄

Period	NR-PM ₁	OA	SO ₄	NH ₄	NO ₃	Chl.	O ₃	NO _x	СО
Marine	3.8 ± 2.0	0.7 ± 0.8	2.4 ± 1.1	0.7 ± 0.3	0.02 ± 0.01	0.02 ± 0.01	31.1 ± 11.9	0.4 ± 1.2	111.5 ± 16.5
Frontal/LP	2.6 ± 2.1	1.0 ± 0.9	1.3 ± 1.2	0.3 ± 0.4	0.04 ± 0.03	0.02 ± 0.01	43.8 ± 11.1	1.0 ± 1.3	107.2 ± 14.4
Continental	9.9 ± 2.9	7.2 ± 2.8	1.9 ± 0.7	0.6 ± 0.2	0.1 ± 0.1	0.02 ± 0.01	52.7 ± 12.7	1.3 ± 1.8	141.3 ± 26.2

Table 1. Aerosol ($\mu g m^{-3}$) and trace gas (ppbv) characteristics (avg. \pm SD) measured during each distinct period type of the campaign.

Table 2. Elemental composition of each PMF factor and contributions to total OA during each period of the campaign.

	Elem	nental and	alysis	Mar	ine period	Frontal/LP period		Continental period	
	0 : C	H : C	N : C	%	μgm^{-3}	%	μgm^{-3}	%	μgm^{-3}
OOA – 1	1.16	1.29	0.013	21	0.14	65	0.63	15	1.10
OOA – 2	0.79	1.41	0.007	11	0.08	6	0.06	72	5.21
OOA – 3	0.76	1.44	0.077	55	0.40	11	0.11	2	0.15
SV – OOA	0.43	1.77	0.013	3	0.02	9	0.09	8	0.60
HOA	0.08	1.89	0.002	7	0.05	9	0.09	2	0.16

measured by the HR-ToF-AMS, and the remaining nss-SO₄ is assumed to be anthropogenic. This technique could lead to underprediction of sea-salt sulfate mass loadings in the case of substantial chloride displacement by nitrate in the marine boundary layer. However, the near-detection-limit concentrations of nitrate measured during the marine period of the study $(0.02\pm0.01 \,\mu g \, m^{-3})$ suggest that this possibility has at most a negligible influence on our results. Multiple lines of evidence described in Sect. 3.2 support the use of this technique.

2.9 Ancillary measurements

A variety of trace gases and meteorological parameters were measured during the campaign. All trace gas and meteorological data were measured with a 5 min averaging time. Individual NO_x species (nitric oxide and nitrogen dioxide (NO₂)) and total reactive gas-phase nitrogen were measured using high-sensitivity chemiluminescence monitors (AQD, Inc.). Ozone (O₃) mixing ratios were measured with an ultraviolet absorption instrument (2BTech, Inc., model 205), and carbon monoxide mixing ratios were measured using highresolution cavity-enhanced direct-absorption spectroscopy (Los Gatos Research, Inc.). Sulfur dioxide was measured with a pulsed fluorescence analyzer (ThermoFischer Scientific, model 43i). Ambient temperature, pressure, wind speed, and wind direction were measured using an RM Young meteorological station.

3 Results and discussion

3.1 Campaign overview

Figure 2 displays the speciated aerosol mass loadings, PMF factor contributions, important trace gas concentrations, and

meteorological conditions encountered during the campaign. Overall, the average NR-PM1 mass loading was $4.66 \pm$ $3.17 \,\mu\text{g}\,\text{m}^{-3}$ (1 standard deviation) and was dominated at times by either SO₄ (44 % on average) or OA (42 %). As the measurements were performed in the early summer in close proximity to the coast, RH was relatively high (average: 81%), conditions were generally sunny, and temperatures were warm (average: 27.3 °C). Examination of the aerosol time series data reveals three distinct period types. Marine periods are characterized by consistent onshore flow, while continental periods are characterized by offshore flow or daily land and sea breeze circulation patterns. A 3-day period influenced by the passage of two cold fronts and a lowpressure (LP) system that produced heavy cloud cover, intermittent rain, and a distinct aerosol diurnal profile was termed "frontal/LP."

Each of these periods contained a unique dominant PMF factor resembling low-volatility oxygenated organic aerosol (OOA) (Ng et al., 2010), denoted as OOA-1, OOA-2, or OOA-3 (Fig. 3). An overview of the average aerosol and trace gas characteristics during each period is provided in Table 1, and a comparison to previous campaigns in the Houston region is shown in Fig. S14. While the extracted PMF factors are highlighted briefly below and summarized in Table 2, more detailed factor descriptions are included in the Supplement.

The majority of the campaign (~ 12 days total), characterized by onshore flow conditions with wind directions generally between 120° and 240° , was classified as "marine." During these periods, which encompass 24 May–1 June and 10–14 June, aerosol mass loadings were relatively stable from day to day. Interestingly, average observed mass loadings were much larger in the first portion of the marine period (4.69 µg m⁻³) (24 May–1 June) than in the second (2.71 µg m⁻³) (10–14 June), despite similar local wind di-



Figure 2. From top to bottom, time series of major NR-PM₁ species, extracted PMF factors, O_3 and NO_2 , and meteorological variables (wind direction, wind speed, RH, and temperature) measured during the campaign. Dotted lines distinguish distinct time period types described in Sect. 3.1.

rection, O₃, and meteorological conditions, implying that air mass history has a large influence on marine aerosol loadings. The observation of SO₄ mass loadings much larger than $1 \,\mu g \,m^{-3}$, which is generally the maximum observed in remote marine locations, even during periods of high biological activity (Zorn et al., 2008; Rinaldi et al., 2010; Schmale et al., 2013; Ovadnevaite et al., 2014), supports either a major anthropogenic aerosol source in the Gulf of Mexico or an influence from continental recirculation; however, backwardtrajectory analysis combined with an investigation of relevant source magnitudes, described in Sect. 3.3, points to a marine anthropogenic source. The average mass loading of SO₄ plus NH_4 (3.04 µg m⁻³) was similar to that measured during onshore flow by Bates et al. (2008) $(3 \,\mu g \, m^{-3})$, despite recent regulations on shipping emissions, while measured OA mass loadings were larger during this study $(0.72 \,\mu g \, m^{-3} \, versus)$ $0.38 \,\mu\text{g}\,\text{m}^{-3}$ from Bates et al., 2008). The OA, which constituted 21 % of total mass, was highly oxidized (average oxygen-to-carbon ratio, O: C = 0.73), consistent with previous measurements of marine aerosols (Russell et al., 2009; Chang et al., 2011; Schmale et al., 2013). The average mass fraction of m/z44 (f_{44}), a metric used to describe the extent of OA oxidation, was 0.15, a value very similar to that observed by Russell et al. (2009) during marine flow conditions (0.16), suggesting that on average the oxidation state of marine OA over the Gulf of Mexico has not changed substantially since ECA implementation. However, as these comparisons are based on only 3–4 weeks of data collected during each of the two time periods, the results should be interpreted with this limitation in mind.

The light winds observed during the campaign suggest that little of the measured marine OA was the result of organicenriched sea spray, as this production pathway generally requires significant whitecap coverage, which is typically only observed above wind speeds of 7–8 m s⁻¹ (Gantt et al., 2011; Shank et al., 2012; Ovadnevaite et al., 2011; Schmale et al., 2013; Frossard et al., 2014). Local wind speeds were virtually never above 8 m s⁻¹ (Fig. 2), and 5-day-averaged



Figure 3. Diurnal variation of NR-PM₁ component average concentrations (\mathbf{a} - \mathbf{c}) and PMF factors (\mathbf{d} - \mathbf{f}) during the marine period (\mathbf{a} , \mathbf{d}), the frontal/LP period (\mathbf{b} , \mathbf{e}), and the continental period (\mathbf{c} , \mathbf{f}). The legends above (\mathbf{a} - \mathbf{c}) describe how mass loadings of specific components were adjusted to fit the figure. Note the different *y*-axis ranges applicable to each period type.

wind speeds calculated using total trajectory lengths were only > 8 m s⁻¹ for 4% of the marine period. Potential major sources of OA therefore include secondary production through processing of biogenic volatile organic compounds (VOCs), as well as primary and secondary production from shipping emissions (Lack et al., 2009; Coggon et al., 2012). This hypothesis is supported by the fact that marine OA composition was dominated by a highly oxidized PMF factor, OOA-3 (O : C = 0.77; 55% of OA on average) (Fig. S5), which was moderately correlated with SO₄ ($R^2 = 0.55$) and displayed little diurnal variation.

Two storms occurred during the sampling campaign. The first (27 May) caused a loss of power to the HR-ToF-AMS and the data gap shown in Fig. 2; the second, denoting the beginning of the frontal/LP period, caused a rapid reduction in aerosol mass that was followed by 3 days of markedly different aerosol characteristics, despite initially similar wind directions. Diurnal profiles of virtually all NR-PM₁ species during the frontal/LP period are distinct from the preceding marine period (Fig. 3) and show maximum concentrations at night. Satellite images of the area show the arrival

of a large-scale frontal system on 2 June and the presence of heavy cloud cover through 5 June (Fig. S15). The O : C ratio during this period is the highest of the campaign, which, combined with the strong correlation between diurnal trends of OA and SO₄ ($R^2 = 0.78$), suggests measured OA represents regional background OA that is diluted with the rise of the boundary layer in the morning. The dominant PMF factor extracted during this period (OOA-1) had an O : C ratio (1.15) similar to the most aged OA observed in urban areas (Hayes et al., 2013), implying an influence of extensive atmospheric processing during transport (Ortega et al., 2016), aqueous processing of highly oxidized water-soluble organics (e.g., glyoxal O : C = 1) (Chhabra et al., 2010), or some combination of the two.

The third identified period, which occurred from 6 to 9 June, shows evidence of continentally influenced air masses and a multi-day increase of NR-PM₁ following passage of the frontal system. The organic-to-SO₄ ratio shifts from a value of 0.34 during the marine period, typical of marine environments (Coggon et al., 2012), to an average value of 3.08, highlighting the predominance of OA sources within

the Houston region. Local wind direction measured from 6 to 8 June appears to show a land-sea breeze type circulation pattern, and midday O₃ concentrations during this period reach the highest levels of the campaign (Fig. 2). Diurnal profiles of NR-PM₁ species highlight the influence of local photochemistry and/or boundary layer dynamics (i.e., downward mixing of aged OA into the growing boundary layer) on aerosol mass loadings (Fig. 3). OOA-2, the dominant PMF factor during this period (72%), displays a diurnal profile similar to a previously extracted OOA factor in Houston's urban core (Cleveland et al., 2012) but is much more oxidized (O : C = 0.79 in this study versus 0.46 in Cleveland et al., 2012), highlighting the effect of aging during transport. Plotting mass loadings of OOA-2 against ambient CO concentrations produces a slope of $\sim 150 \,\mu g \, m^{-3} \, ppmv^{-1} \, dur$ ing this period. This value is similar to previous aircraft measurements of aged industrial plumes in Houston (Bahreini et al., 2009; Wood et al., 2010). Modeling results have suggested that biogenic VOCs contribute little OA during Houston industrial plume transport (Bahreini et al., 2009) except in the case of advection into the forested north of Houston (Brown et al., 2013), which suggests a likely anthropogenic origin of OOA-2.

3.2 Analysis of MSA mass loadings

The time series of calculated MSA mass loadings is shown in Fig. 4, as are concentrations determined for the three distinct periods described previously and comparisons with literature values. Overall during the marine period, MSA mass ranged from ~ 0 to $0.07\,\mu g\,m^{-3}$ and showed moderate correlation with nss-SO₄ ($R^2 = 0.46$) and weak correlation with OA ($R^2 = 0.12$), suggesting major additional sources of both nss-SO₄ and OA over the MBL. While previous MSA measurements in the Gulf of Mexico are sparse, Saltzman et al. (1983) recorded submicron mass loadings of 0.022- $0.066 \,\mu g \,m^{-3}$ in Miami, in agreement with our results. In addition, our results align well with previous submicron measurements taken at lower latitudes in both the Atlantic and Pacific oceans, as well as with measurements taken at higher latitudes while sampling tropical air masses (Fig. 4) (Zorn et al., 2008; Ovadnevaite et al., 2014; Huang et al., 2017).

On average, MSA accounts for only 3.2% of submicron OA during marine periods, a value much lower than observed in previous coastal measurements with a HR-ToF-AMS. For instance, Crippa et al. (2013) reported that MSA accounted for approximately 20% of submicron OA in Paris when air masses traveled from marine locations. At Mace Head, Ireland, MSA represented 12.5%–18% of submicron OA during May and June when air masses traveled from the tropics (Ovadnevaite et al., 2014). However, before establishment of the ECA, Russell et al. (2009) found that during onshore flow in the Gulf of Mexico between 52% and 89% of organic mass could be attributed to oil combustion/refining and wood-smoke-related sources. Therefore, the small MSA mass fraction observed here is likely the result of strong remaining anthropogenic OA sources over the gulf. This hypothesis is supported by the relatively weak correlation between the dominant marine PMF factor (OOA-3) and MSA ($R^2 = 0.41$). For comparison, the distinctly biogenic marine PMF factor extracted by Crippa et al. (2013) in Paris correlated strongly with MSA ($R^2 = 0.84$).

Quantification of the MSA/OA ratio permits a rough calculation of the contribution of biogenic sources to total marine OA. Using an assumption that MSA should only represent 5 %-10 % of total biogenic OA under pristine conditions over the Gulf of Mexico (a low estimate based on the previous observations discussed), calculation of the mass fraction of biogenic OA based on this ratio (MSA/biogenic OA = 0.05 - 0.1) and the measured MSA/total OA ratio (MSA / total OA = 0.032) implies biogenic sources only produce $\sim 32\%$ -64% of total measured marine OA. In addition, vanadium, an element common to shipping emissions, is thought to act as a catalyst to MSA formation (Gaston et al., 2010). As oil combustion emissions were responsible for a major fraction of OA over the Gulf of Mexico in the past (Russell et al., 2009), this type of catalytic process may be enhancing MSA production relative to more pristine locations at similar latitudes. This further suggests that the assumption that MSA accounts for only 5 %-10 % of biogenic OA is likely a low estimate.

MSA mass loadings were positively, though only slightly, correlated with trajectory-averaged solar flux ($R^2 = 0.12$) and negatively correlated with trajectory length (i.e., wind speed) ($R^2 = 0.16$). The lack of a strong correlation with these parameters is partly due to the fact that DMS is emitted primarily in regions with high concentrations of biological organisms in the sea-surface layer, which typically occur close to the coast. Therefore, emissions are not uniform across the Gulf of Mexico (Sorooshian et al., 2009). Often, high MSA mass loadings are linked to specific locations of high biological activity through analysis of backward trajectories and comparison to chlorophyll-a levels (Sorooshian et al., 2009, 2015a; Gaston et al., 2010; Schmale et al., 2013; Huang et al., 2017). While the accuracy of satellite-derived measures of chlorophyll a as an indicator of DMS production potential is still under debate (Sorooshian et al., 2009; Huang et al., 2017), the data here support a link between oceanic chlorophyll a and MSA mass loadings, as a peak in MSA mass is observed on 11 June, when backward-trajectory analysis indicates air masses slowly traveled over the nutrientrich waters close to the coast and near the mouth of the Rio Grande (Fig. S16).

3.3 Quantifying anthropogenic contributions to marine aerosol mass

The average MSA/nss-SO₄ ratio measured during the marine period was 0.012. Applying the biogenic MSA/nss-SO₄ ratio determined by Savoie et al. (2002) indicates that



Figure 4. (a) Comparison of average MSA mass loadings measured during each of the three major periods of the campaign with previously published values measured in Florida; on an Atlantic Cruise; and at Mace Head, Ireland. (b) Time series of MSA and nss-SO₄. Black dashed lines denote boundaries of distinct time period types. Inset graph shows the correlation of total OA and SO₄ with MSA during the marine period. (c) Hourly averages of the estimated fraction of nss-SO₄ attributed to anthropogenic sources. Inset pie charts depict anthropogenic and biogenic contributions to nss-SO₄ (left) and total inorganic NR-PM₁ (right) during the marine period. Mass loadings of nitrate and chloride comprise less than 2 % of total inorganic aerosol.

an average of 77 % of nss-SO₄ ($1.8 \,\mu g \, m^{-3}$) is the result of anthropogenic sources during onshore flow (Fig. 4c). This value likely represents an upper limit (see Sect. 2.8). Furthermore, as the partitioning of gaseous ammonia (NH₃) to the aerosol phase is driven by the neutralization of acidic SO₄, mass loadings of nss-SO₄ and NH₄ are highly correlated during the marine period ($R^2 = 0.97$). As a result, an estimate of an "anthropogenic" mass loading of NH₄ can be calculated based on the NH4/nss-SO4 ratio and the calculated anthropogenic fraction of nss-SO₄. The classification of a fraction of NH₄ as "anthropogenic" therefore refers to the necessity of an anthropogenic species (in this case nss-SO₄) for the production of NH₄ aerosol, rather than anthropogenic NH₃ emissions. By applying this method and combining anthropogenic NH₄ and nss-SO₄ mass loadings, anthropogenic sources contribute 73 % of total inorganic NR- PM_1 (2.3 µg m⁻³) on average during marine flow conditions.

Multiple lines of evidence support the use of MSA measurements coupled with the biogenic MSA/nss-SO₄ ratio to apportion anthropogenic and biogenic nss-SO₄. Using measurements of chlorophyll-a concentrations, wind speeds measured on board the R/V Brown, and the wind speedtransfer velocity relationship determined by Nightingale et al. (2000), Bates et al. (2008) estimated that the DMS flux from the Gulf of Mexico was capable of producing between 0.2 and 0.4 μ g m⁻³ of biogenic nss-SO₄. For comparison, the average mass loading of biogenic nss-SO₄ calculated using the biogenic MSA/nss-SO₄ ratio during marine periods in our study is $0.54 \,\mu g \, m^{-3}$, in relatively good agreement with those results. Furthermore, Fig. 5 shows the WPSCF analysis of anthropogenic nss-SO₄ and MSA, the Automated Mutual Assistance Vessel Rescue System (AMVER) shipping spatial proxy map (Wang et al., 2008), and chlorophyll-a levels derived from MODIS satellite measurements. The use of the



Figure 5. WPSCF plots of MSA (top left) and anthropogenic nss-SO₄ (top right) for the marine period of the study, along with the chlorophyll-*a* concentration observed by the NASA MODIS satellite (bottom left) and AMVER shipping spatial proxy map (bottom right). Warmer colors in the WPSCF grid cells indicate higher source probability. The color of each $0.1^{\circ} \times 0.1^{\circ}$ grid cell in the AMVER map is based on the corresponding "shipping emissions allocation factor" (SEAF) value (Wang et al., 2008). The hatched region extending from the coasts in each panel represents the approximate area encompassed by the ECA. Note that the ECA boundary is narrowed on the southeastern coast of Florida due to proximity to the territorial waters of Cuba and the Bahamas.

biogenic MSA/nss-SO₄ ratio is qualitatively supported by the relatively distinct WPSCF results of anthropogenic nss-SO₄ and MSA and by the agreement between the anthropogenic nss-SO₄ WPSCF map and the region of high shipping traffic indicated by the AMVER inventory. The highprobability region of anthropogenic nss-SO₄ is located predominately outside of the ECA boundary where shipping lanes converge, while the MSA high-probability region is largely within the ECA where surface chlorophyll-*a* concentrations are elevated (Fig. 5).

While point-source emissions in Florida or long-range transport could contribute to the anthropogenic nss-SO₄ measured during this study, further analysis suggests these sources are minor in comparison to marine vessel emissions. According to the National Emissions Inventory (NEI), $\sim 160\,000\,t$ of SO₂ were emitted in Florida in 2014 (U.S.

EPA, 2014). However, only $\sim 30\,000\,t$ ($\sim 19\,\%$) were emitted in the southern peninsular region indicated as a potential source by the WPSCF analysis (south of 28° N) (Fig. S17). While point-source-distributed NEI data are not yet available for 2016, EPA statewide average data suggest that Florida SO₂ emissions were approximately half of those in 2014 $(\sim 80\,000\,t)$, with the change almost entirely due to a 75 % reduction in emissions from electricity-generating stations (U.S. EPA, 2017). If emissions from individual electricitygenerating stations south of 28° N have been similarly reduced, only $\sim 20\,000$ tons of SO₂ were emitted in the southern peninsular region in 2016. For comparison, recent emissions inventories predict that marine vessels emit as much as 75 000 t of SO₂ annually in the Gulf of Mexico after accounting for the ECA, nearly 4 times as much as the geographically relevant Florida emissions (Johansson et al., 2017).

Offshore oil and gas extraction operations represent another source of SO₂ emissions within the Gulf of Mexico. As can be seen from Fig. 5, all currently operating offshore sites are located within the ECA (Wilson et al., 2017). Under the Clean Air Act, regulation of currently operating offshore oil and gas sites is handled by the Department of the Interior's Bureau of Ocean Energy Management (BOEM). According to the BOEM 2014 Gulfwide Emission Inventory Study, annual SO₂ emissions from platform and non-platform oil and gas operations totaled 7150 t (Wilson et al., 2017). Within the same study region, SO₂ emissions from commercial marine vessels were estimated at 48 215 t (Wilson et al., 2017). The discrepancy between marine vessel SO₂ emissions reported in the BOEM inventory (48215t) and that of Johansson et al. (2017) (75 000 t) is due to the limited spatial scope of the BOEM study (only the western Gulf of Mexico was considered). While non-negligible, offshore operations therefore represent less than $\sim 15 \%$ of SO₂ emissions in the western Gulf of Mexico and less than $\sim 10\,\%$ of total SO_2 emissions in the entire Gulf of Mexico.

In terms of the contribution from long-range transport, air masses that originated in Europe or Africa required 15 days of transit or more to reach the measurement site based on HYSPLIT modeling. Assuming that sulfur compounds have a lifetime of $\sim 5-7$ days in the MBL (Faloona, 2009), 89 %–95 % of the original sulfur in these air masses would be lost prior to measurement. This agrees with the finding by Bates et al. (2008) that only a small fraction of SO₄ measured in the Gulf of Mexico was contributed by African dust during measurements in 2006.

Volcanic sources of SO_2 have occasionally contributed significantly to nss- SO_4 in marine regions during previous campaigns (Jung et al., 2014). Any nss- SO_4 produced by volcanic emissions would be apportioned to anthropogenic sources due to the apportionment technique used (i.e., volcanoes would not be expected to produce substantial MSA, leading to a depression of the MSA/nss- SO_4 ratio). Therefore, the influence of volcanic emissions would incorrectly increase the fraction of measured nss- SO_4 attributed to anthropogenic sources. However, the only relevant volcanoes in the area are along the Caribbean islands, and backwardtrajectory analysis reveals that the largest measured mass loadings of nss-SO₄ correspond to air masses that passed far north of them. It therefore appears that the vast majority of measured anthropogenic nss-SO₄ was emitted by marine vessels rather than other sources.

These results contrast with those from the previous model study of Lauer et al. (2007), who predicted using a global model that shipping contributes only $\sim 30\%$ of submicron SO₄ over the Gulf of Mexico using the AMVER-distributed shipping inventory from Eyring et al. (2005) and as little as 15% or less using the International Comprehensive Ocean-Atmospheric Data Set-distributed inventory from Corbett and Kohler (2003) on an annual basis. Multiple lines of evidence suggest that the discrepancies observed between our results and previous modeling results are not simply due to the timing of our measurements. For instance, while less shipping-related SO₄ is likely produced in the fall/winter due to the reduction in photochemical activity during that time, conversion of biologically emitted SO₂ into nss-SO₄ should have the same photochemical dependence. Furthermore, the SO₂ yield from DMS oxidation, the major biological nss-SO₄ production pathway, is reduced in the winter due to the temperature dependence of DMS oxidation chemistry, as previously explained (Sect. 2.8) (Jung et al., 2014). Finally, data from the Port of Houston suggest that shipping traffic (estimated by the number of 20 ft equivalent cargo units (TEUs) processed at the port) is only reduced by $\sim 10\%$ in the winter (Fig. S18) (Port of Houston, 2017). However, a portion of this discrepancy is likely attributable to the fact that the marine period of our study only encompasses onshore flow conditions, whereas the annual average calculated by Lauer et al. (2007) also incorporates periods of offshore flow, when continental emissions act as source of nss-SO₄ to the MBL over the Gulf of Mexico.

Quantification of anthropogenic nss-SO₄ allows for a more detailed apportionment of marine OA than was possible based on MSA alone. While the correlation between total nss-SO₄ and OOA-3 (the dominant marine OA factor) is moderate ($R^2 = 0.55$), anthropogenic nss-SO₄ is strongly correlated with OOA-3 ($R^2 \ge 0.78$) (Fig. S12), suggesting OOA-3 is coupled to shipping emissions either directly (e.g., SOA from marine vessel VOCs or oxidation of evaporated marine POA) or indirectly (e.g., increased uptake of watersoluble gases over the MBL due to increased ALW). Substantial processing of OOA-3 during transport leads to the removal of major mass spectral tracers; however, there is some evidence for a contribution from naphthalene OA (discussed in the Supplement), which is a dominant, commonly measured VOC emitted by major commercial shipping vessels (Agrawal et al., 2008; Murphy et al., 2009; Czech et al., 2017). Assuming, as a strictly upper-bound estimate, that OOA-3 production is entirely dependent on shipping emis-

sions, anthropogenic sources contributed 71 % of total NR-PM₁ ($2.7 \,\mu g \,m^{-3}$) on average during the marine period.

Comparing the submicron mass loadings of nss-SO₄ and NH₄ measured during marine periods by Bates et al. (2008) (pre-ECA) $(3 \mu g m^{-3})$ to those measured during our study $(3.04 \,\mu g \,m^{-3})$ suggests ECA implementation has had a negligible effect on aerosol mass. However, the amount of shipping traffic within the Gulf of Mexico, estimated with the total number of loaded TEUs processed at the ports of Houston, TX; Galveston, TX; Freeport, TX; New Orleans, LA; and Mobile, AL, has increased by approximately 42 % since 2006 (Fig. S19) (U.S. Army Corps of Engineers, Navigation Data Center, 2016), suggesting that emissions reductions per vessel within the ECA boundary may have been offset by increased traffic. On a yearly basis, the estimated increase in shipping traffic since 2006 (4.6% per year) is similar to the annual growth in seaborne trade observed between 2002 and 2007 (5.2 %) and is within the range of growth predicted through 2050 (3.6 %–5.9 %) (Corbett et al., 2007; Eyring et al., 2010). In support of these rapid growth estimates, Tournadre (2014) recently concluded that shipping traffic in the Atlantic Ocean nearly doubled between 2006 and 2012, corresponding to an average annual increase of ~ 8 %. While it is also possible that the specific meteorological conditions encountered during this campaign (e.g., air mass trajectories, average wind speeds) were more conducive to the accumulation of anthropogenic nss-SO₄ than during the study of Bates et al. (2008), this is unlikely to be the dominant reason for the little change observed since ECA implementation. Therefore, our results suggest that the ECA has reduced shipping emissions on a per-vessel basis, as there has been little change in shipping-related aerosol despite significant growth in the shipping trade. These results also provide justification for further limits on FSC, which are expected to be implemented in 2020 and require a reduction of FSC to a maximum of 0.5 % globally (Kotchenruther, 2016). However, longer-term observational or modeling studies are nevertheless needed to more definitively quantify the effect of the ECA.

3.4 Relationship between shipping emissions and OA oxidation state

In order to obtain a quantitative measure of the difference in OA composition between air masses influenced by shipping emissions and those lacking such influence, a 12 h period was isolated on 10–11 June, when the site encountered air masses that had been inside the ECA boundary but over the ocean (i.e., within 200 nautical miles of the coast) for virtually their entire 5-day history (Fig. S20). Assuming ECA compliance, these air masses should receive only a small fraction of the particulate and SO₂ emissions encountered by those originating outside the boundary (Lack et al., 2009, 2011; Browning et al., 2012). Based on air mass history and the accompanying mass spectral analysis described below, we classified OA measured during this period as "marine-biogenic".

For comparison, we distinguished a second 24 h period (30 May) that had similar 5-day backward-trajectory-averaged meteorological conditions to the biogenic period (faster average wind speed and comparable average solar flux) but had trajectories that originated outside the ECA boundary and passed through the high-intensity shipping region. The shipping-influenced period had notably larger mass loadings of anthropogenic nss-SO₄ (2.24 versus 1.09 μ g m⁻³) and OA (1.04 versus 0.29 μ g m⁻³).

Figure 6 presents the average OA mass spectra determined for each of these periods. In the shipping-influenced air masses, measured OA is highly processed, with a much larger f_{44} (0.20) (a marker of carboxylic acids) than is typical of marine biogenic OA ($\sim 0.08-0.14$) (Chang et al., 2011; Coggon et al., 2012, 2014; Crippa et al., 2013) and a composition dominated by oxygenated species (66%). In contrast, OA measured during the period of minor shipping influence is notably less aged and contains numerous indicators of a marine biogenic source. For instance, prominent non-oxygenated spectral fragments are observed at m/z27, 39, 41, 43, 55, and 67 (Fig. S21), implying the presence of alkenes, cycloalkenes, cycloalkanes, and dienes, in agreement with Ovadnevaite et al. (2011, 2014) for marine OA measured at Mace Head, Ireland, and by Bates et al. (2012) in physically generated sea spray aerosol. A relatively significant contribution from m/z79 (CH₃SO₂) (~1%) is also apparent, and as a result MSA contributes 9.3 % of total OA, a value 3 times larger than the average during the marine period and in closer agreement with previous measurements in remote marine regions (Ovadnevaite et al., 2014). Furthermore, a prominent signal from the CHO⁺ ion, an aldehyde tracer, is observed (~ 7 %), which is uncharacteristic of aged urban emissions (Ng et al., 2010) but has been observed in the mass spectra of numerous biogenic SOAs from both chamber experiments and ambient measurements (Shilling et al., 2009; Chhabra et al., 2010; Slowik et al., 2010; Setyan et al., 2012) and from marine biogenic OA specifically (Chang et al., 2011; Crippa et al., 2013; Coggon et al., 2014). Ultimately, the biogenic period spectra correlate well with the marine biogenic factor extracted by Chang et al. (2011) over the Arctic Ocean ($R^2 = 0.78$) as well as with the marine OA factor extracted by Crippa et al. (2013) in Paris (\mathbb{R}^2) 0.68), while the shipping-influenced period spectra correlate extremely well with the continental factor extracted by Chang et al. $(2011) (R^2 = 0.95)$.

The mass spectra from the shipping-influenced period have notably larger signals from m/z44 and m/z28 than the biogenic period, suggesting a larger amount of atmospheric processing that converted OA components into organic acids (Chhabra et al., 2011). Numerous remote marine studies have shown that, on average, the oxidation state of marine aerosol varies only slightly in the absence of anthropogenic influences (Gantt and Meskhidze, 2013; Wozniak et al., 2014). In this case, the absolute difference in the O : C ratio between the two scenarios is 0.29 (0.90 for the shipping-



Figure 6. Average OA mass spectra measured during (**a**) a period heavily influenced by shipping emissions (30 May 2016) and (**b**) the "marine-biogenic" period when air masses traveled within the ECA for nearly their entire 5-day history. The overall organic fragment composition measured during each period is shown in the corresponding pie charts.

influenced period versus 0.61 for the marine-biogenic period), implying a major impact of shipping on related OA chemical and potentially physical properties. While primary marine aerosol particles can have high O : C ratios (\sim 1) due to the significant mass fraction of carbohydrate components in dissolved organic matter (Russell et al., 2010), the low trajectory-averaged wind speeds and high f_{44} suggest that OA measured during the shipping-influenced period is not primary (Frossard et al., 2014).

Using the function developed by Duplissy et al. (2011) to describe the relationship between OA oxidation state (represented by the mass fraction of m/z44) and hygroscopicity, the calculated organic hygroscopic growth factor (K_{org}) for the shipping-influenced period is 3 times larger (0.31 versus 0.101) than that calculated for the marine-biogenic period. Therefore, despite the fact that freshly emitted in-plume shipping aerosol is thought to have a suppressed hygroscopic growth factor relative to background marine aerosol (Murphy et al., 2009), our results suggest that extensive aging during transport near shipping lanes (presumably due to increased oxidant levels) may lead to an eventual increase in bulk marine OA hygroscopicity relative to aerosol unaffected by shipping emissions. This hypothesis is supported by the relatively strong correlation observed between daily anthropogenic nss-SO₄ and the organic hygroscopicity factor $(R^2 = 0.64)$ calculated using the Duplissy et al. (2011) method during the marine period (Fig. 7).

Figure 8 displays marine OA plotted on the f_{44} -versus f_{43} triangle diagram (Ng et al., 2010) to describe OA aging. Less oxidized OA typically occupies a wide space at the



Figure 7. Correlation between daily averaged anthropogenic nss- SO_4 and the OA hygroscopicity factor calculated using the method developed by Duplissy et al. (2011).

bottom of the plot, indicative of variable ambient OA mass spectra, while aging causes movement diagonally upward, as mass spectra become more similar with age (Ng et al., 2010). Figure 8 highlights that OA oxidation is greatly influenced by a combination of physical air mass history and meteorology. Three specific days demonstrate these influences particularly well. On 24 May, backward-trajectory analysis reveals that air masses passed directly over the region of



Figure 8. (a) The f_{44} : f_{43} diagram highlighting different influences on OA oxidation state during the marine period. Circles represent daily average values, and bars indicate standard deviations. Periods with HOA mass loadings greater than twice the median value during the marine period were removed from the analysis. Two additional values from published marine studies are shown as a reference. (b) Map of Gulf of Mexico showing hourly 5-day back trajectories calculated for each of the 3 days identified in the f_{44} : f_{43} diagram. The AMVER shipping emissions spatial proxy map (Wang et al., 2008) is shown for reference. The hatched region extending from the coasts represents the approximate area encompassed by the ECA. Note that the ECA boundary is narrowed on the southeastern coast of Florida due to proximity to the territorial waters of Cuba and the Bahamas.

major shipping influence, resulting in a substantial amount of nss-SO₄ aerosol and highly oxidized OA. In contrast, on 11 June – despite the fact that trajectory-averaged wind speeds were lower and solar flux was comparable, suggesting meteorological conditions were more conducive to OA processing and elevated aerosol mass loadings – air masses largely missed the high-intensity shipping region (and remained largely within the ECA), resulting in less nss-SO₄ and less oxidized OA. On 13 June, arriving air masses had faster average wind speeds and avoided shipping lanes, resulting in an extreme case of very little nss-SO₄ and only minor processing. Based on this information, it appears that air mass transit within major shipping corridors is associated with more processed organic aerosol relative to background marine conditions in the Gulf of Mexico.

There are multiple ways in which the presence of shipping emissions could increase the rate of OA processing. While peak daytime concentrations of OH of 6×10^{6} –1 × $10^7 \,\mathrm{mol}\,\mathrm{cm}^{-3}$ are relatively consistent throughout the clean MBL (Raper et al., 2001; Vaughan et al., 2012), modeling results by Chen et al. (2005) and Kim et al. (2013) indicate that, within individual shipping plumes, OH concentrations are elevated by a factor of 1.2 to 2.7 and OH concentrations can remain elevated up to 140 km behind an individual shipping vessel. Significant NO₂ levels within the plume also increase concentrations of nitrate radical to several ppty, even during the daytime, which would hypothetically result in rapid oxidation of any unsaturated VOCs or components of primary marine OA (Myriokefalitakis et al., 2010; Bates et al., 2012; Kim et al., 2013). Additionally, elevated production of nss-SO₄ aerosol increases ambient ALW mass, increasing the partitioning medium available to small, water-soluble organic gases (WSOGs) produced from both biogenic and anthropogenic sources (e.g., glyoxal, methylglyoxal, acetaldehyde) and processed in the aqueous phase into highly oxidized species (such as glyoxylic acid/glyoxylate, O: C = 1.5, or oxalic acid/oxalate, O: C = 2) (Ervens et al., 2011; Ge et al., 2012).

However, it should be noted that, unlike SO₂ emissions/concentrations, the presence of the ECA may not substantially reduce concentrations of atmospheric oxidants within shipping plumes. Numerous studies have linked shipping-related perturbations of marine OH radical concentrations to vessel NO_x emissions (e.g., Lawrence and Crutzen, 1999; Kim et al., 2013). While Tier III NO_x emissions standards implemented in the ECA in 2016 are projected to slowly reduce emissions over time (as standards only apply to vessels constructed in 2016 and later), Browning et al. (2012) have shown that the sole switching from high-sulfur HFO to MGO at the ECA border in the Gulf of Mexico only produced NO_x reductions of 1 % - 6 %. As a result, the effect of shipping emissions on marine OA processing likely depends more on whether air masses align with shipping lanes during transit than whether such transit occurs within or outside of the current ECA.

3.5 Relationship between shipping emissions and major marine OA components: amines and MSA

While MSA and alkylamines, such as dimethylamine (DMA) and diethylamine specifically, are frequently observed over the MBL and are linked to biogenic emissions (Murphy et al., 2007; Facchini et al., 2008; Sorooshian et al., 2009), the partitioning dynamics of each are influenced by shipping emissions. For instance, recent single-particle measurements in California reveal a possible catalytic role of vanadium in MSA formation (Gaston et al., 2010), while gaseous alkylamines typically undergo neutralization reactions with sulfuric or nitric acids to form aminium salts (Murphy et al., 2007). However, previous studies have produced conflicting results about whether biogenic marine SOA mass is maximized in clean or polluted environments. For instance, Sorooshian et al. (2009) and Facchini et al. (2008) both noted that mass loadings of amines and MSA were largest in clean rather than polluted air masses, supporting their attribution to biogenic sources; however, Sorooshian et al. (2015a) observed similar size distributions of MSA and vanadium along the California coast, while Youn et al. (2015) reported noticeable long-term correlations between amines and SO₄. Myriokefalitakis et al. (2010) suggested that, on a global basis, modeled marine SOA originates almost entirely from either DMS oxidation (i.e., MSA-related) (\sim 78%) or formation of dialkyl amine salts ($\sim 21\%$), highlighting the importance of understanding anthropogenic influences on their production in areas influenced heavily by ship traffic. To quantify a lower-bound ambient amine signal from this coastal dataset, individual mass spectral fragments typical of alkyl amines identified in previous HR-ToF-AMS studies – specifically those at m/z27 (CHN⁺), 30 (CH₄N⁺), 44 $(C_2H_6N^+)$, 56 $(C_3H_6N^+)$, 58 $(C_3H_8N^+)$, and 72 $(CH_4N_4^+)$ – were combined (Murphy et al. 2007; Hildebrandt et al., 2011; Sun et al., 2011).

Figure 9 highlights that hourly averaged amine mass loadings correlate well with anthropogenic nss-SO₄ ($R^2 = 0.63$), while MSA mass loadings show a noticeably weaker relationship ($R^2 = 0.30$). A strong correlation between MSA and anthropogenic nss-SO₄ would indicate either that the biogenic nss-SO₄ fraction had been underpredicted or that a strong catalytic effect on MSA production was occurring.

The correlation between anthropogenic nss-SO₄ and amines is consistent with those observed by Youn et al. (2015) for DMA and SO₄ in Tucson, AZ, in 2013 (r > 10.72). Amines also display a positive relationship with NH₄ $(R^2 = 0.61$, similarly to nss-SO₄), in agreement with the fact that throughout the campaign NR-PM1 was never fully neutralized by the small ammonia sources that exist over the MBL. This is highlighted by the fact that the average neutralization ratio, the molar ratio of ammonium to the sum of sulfate and nitrate $([NH_4^+]/(2 \times [SO_4^{2-}] + [NO_3^-]))$, was only 0.74, resulting in a consistent pathway for amine SOA formation through aqueous dissolution and partial neutralization of the acidic nss-SO₄ aerosol. Furthermore, the correlation between amines and anthropogenic nss-SO₄ is much stronger than correlations with average wind speed, solar flux, and mixing-layer depth ($R^2 = 0.04, 0.18$, and 0.10, respectively), suggesting that anthropogenic emissions play a larger role in amine aerosol formation than meteorology.

The link between shipping emissions and amine formation also is supported by the high nitrogen-to-carbon ratio (N : C) of the dominant marine PMF factor, OOA-3 (N : C = 0.074),

a value larger than that observed in aged marine OA (N : C \sim 0.04) (Schmale et al., 2013) and amine-related urban PMF factors extracted in Pasadena, CA (N : C = 0.052) (Hayes et al., 2013), and New York City (N : C = 0.053) (Sun et al., 2011), but similar to a biogenic MSA-related factor extracted at Bird Island near Antarctica (N : C = 0.08) (Schmale et al., 2013). It is likely that this anthropogenic-biogenic link can be extrapolated to other areas where marine biogenic and shipping emissions coexist. As a result, amines measured in heavily trafficked marine environments should not be interpreted exclusively as products of a purely biogenic SOA formation pathway.

3.6 Anthropogenic ALW and potential influences on SOA formation

While the impacts of shipping emissions on marine cloud condensation nuclei concentrations, cloud formation processes (e.g., ship tracks), and cloud properties (e.g., increased albedo, increased droplet acidity) have been extensively studied in the recent past (e.g., Coakley et al., 1987; Durkee et al., 2000; Coggon et al., 2012; Peters et al., 2012; Sorooshian et al., 2015b), there have been few measurement-based analyses of the role of these emissions on the production of ALW and subsequent OA processing in coastal marine environments. The injection of SO₂ and production of SO₄ aerosol mass in the marine boundary layer, combined with high ambient relative humidity levels, will lead to a substantial mass concentration of ALW associated with submicron anthropogenic particles, even in cloud-free conditions. As organic gases capable of partitioning to ALW are present throughout the MBL (Sinreich et al., 2010), have elevated concentrations near the coasts (Fu et al., 2008, 2011), and contribute significantly to aerosol mass in both rural and urban areas within the southeastern United States (Li et al., 2015), the production of this partitioning medium over the marine boundary layer may impact SOA production. As a result, measurement-based modeling of ALW is needed to inform understanding of how future changes to shipping sulfur emissions might influence SOA formation in coastal environments.

On average, ALW mass loadings of $5.21 \pm 4.62 \,\mu g \,m^{-3}$ are modeled using ISORROPIA II (Fountoukis and Nenes, 2007) during the marine period, representing on average 58% of total NR-PM₁ particle mass. This value is slightly larger than the average determined for the HSC region in September–October 2006 by Nguyen et al. (2016) ($4.6 \,\mu g \,m^{-3}$), presumably because of higher average RH along the coast and similar total inorganic mass loadings, and is larger than the average values reported for every major city in North America analyzed by Nguyen et al. (2016). If anthropogenic nss-SO₄ were eliminated completely, average ALW mass loadings associated with NR-PM₁ aerosol would ultimately be reduced by 66.4%. As a result, the majority of NR-PM₁-associated ALW over the Gulf of Mexico appears



Figure 9. Observed correlation between anthropogenic nss-SO₄ and (left) the sum of measured alkyl amine fragments and (right) MSA.

to be controllable. While concentrations of WSOG over the MBL are relatively small (Sinreich et al., 2010), advection of this ALW inland may have large impacts on nearby SOA formation where precursor sources are more prevalent.

Multiple modeling studies have suggested that small WSOGs - specifically glyoxal, methylglyoxal, and isoprene epoxides (IEPOX) - contribute heavily to SOA mass in the Houston region (Li et al., 2015; Ying et al., 2015). Li et al. (2015) found that these three compounds were responsible for nearly 80 % of total SOA mass loadings at a downtown site during a simulated period in 2006, with the largest fraction ($\sim 30 \%$ –50 %) contributed by IEPOX. The authors also showed that these species dominate SOA mass across the Gulf Coast, with significant total loadings ($\sim 4 \,\mu g \, m^{-3}$ or greater) from the southern end of Texas to the Florida panhandle. Ying et al. (2015) used the Community Multiscale Air Quality model (CMAQ) to characterize biogenic and anthropogenic contributions to glyoxal and methylglyoxal SOA and found that isoprene is a major contributor to both (47 %of glyoxal and 82 % of methylglyoxal SOA, specifically). As these compounds are precursors to aqueous SOA (aq-SOA) formation, anthropogenic impacts on ALW represent another potential anthropogenic-biogenic link in SOA production (Carlton and Turpin, 2013).

The aqSOA formation from these WSOGs ultimately depends on both uptake into ALW and subsequent reactions to produce low-volatility organic acids or high-molecularweight oligomeric products (McNeill, 2015). However, as uptake of OH, the dominant aqueous phase oxidant, is typically surface-limited (Ervens et al., 2014), large-scale models often simplify this process by assuming that aqSOA formation is irreversible and surface-controlled (Li et al., 2015; Ying et al., 2015), representing SOA production rate by

$$\frac{\mathrm{d}M_{a,i}}{\mathrm{d}t} = \frac{1}{4}\gamma_i v_i A M_i,\tag{4}$$

where $M_{a,i}$ is the aerosol-phase mass concentration of species *i* (µg m⁻³), γ_i is its reactive uptake coefficient, v_i

is its gas-phase thermal velocity $(m s^{-1})$, *A* is the ambient aerosol surface area (SA) concentration $(m^2 m^{-3})$, and M_i is the mass concentration of the species in the gas phase $(\mu g m^{-3})$. As a fraction of WSOG partitioning is reversible (Chhabra et al., 2010; Wong et al., 2015), and SOA formation may be more dependent on the particle-phase reaction rate than simply the particle SA (Budisulistiorini et al., 2017), this estimated production rate likely represents an upper limit; however, published CMAQ results for OA mass loadings in the Houston area calculated in this manner agree well with observations (Li et al., 2015; Ying et al., 2015).

In order to approximate the effect of anthropogenic marine aerosol (AMA) on WSOG aqSOA production, we modeled aqSOA formation from isoprene-derived glyoxal, methylglyoxal, and IEPOX in the Houston area using the equation above and a previously developed 0-D model including a semi-explicit isoprene oxidation mechanism (Schulze et al., 2017). This model assumes that air masses rich in AMA advect over the urban core of Houston, where the added SA due to anthropogenic emissions over the ocean increases SA-dependent aqSOA production rates. Total model SA was quantified by combining the dry-mass size distribution measured by the HR-ToF-AMS at the coastal site during onshore flow and the mass added by NR-PM1-associated ALW; however, a correction was applied to account for SA loss due to deposition and ALW evaporation during transport to the HSC. A detailed description of all model assumptions (e.g., boundary layer height, aerosol deposition, and ALW evaporation during transport) is provided in the SI. Average diurnal isoprene, O_3 , and NO_x concentrations measured by five monitors within the HSC during the marine period were used as model constraints (Fig. S22). Diurnal OH concentrations were taken from measurements in downtown Houston during the SHARP 2009 campaign (Ren et al., 2013).

A diurnal model run was first performed using the total corrected marine aerosol SA to predict aqSOA formation in the HSC. This procedure isolates aqSOA production due to marine aerosol SA specifically. In order to produce an upperbound estimate of the effect of AMA, a second run was performed with all SA contributions from anthropogenic species removed (i.e., anthropogenic nss-SO₄, NH₄, and ALW), and the difference in aqSOA was calculated. A lower-bound estimate was calculated with a model run that only removed the SA contribution of anthropogenic ALW. In order to ensure conservative results, OA was assumed to be entirely biogenic for the purposes of this calculation.

To compare this effect with SOA production from locally emitted anthropogenic VOCs (AVOC-SOA) in Houston, gasphase AVOC data measured during the marine period (concentrations of 16 alkanes, 7 alkenes, and 9 aromatics) were obtained from the same monitoring sites around the HSC (Fig. S22). Estimates of SOA production rates from these 32 VOCs were calculated using the volatility basis set approach utilized in Tsimpidi et al. (2010). In this mechanism, organic condensable gases produced from initial VOC oxidation are allowed to undergo further aging to produce lower-volatility products (Tsimpidi et al., 2010; Hayes et al., 2015). A more detailed description of this process is provided in the Supplement.

Figure 10 shows that, on a daily basis, aqSOA production attributable to isoprene WSOG reactive uptake is primarily due to methylglyoxal rather than IEPOX, implying "high-NO_x" rather than "low-NO_x" ambient conditions (Budisulistiorini et al., 2017). Assuming high-NO_x conditions, the modeled effect of AMA on aqSOA production in the HSC is equivalent to 6 %-23 % of potential daily SOA production from AVOCs measured locally. Using data from the monitor with the highest isoprene concentrations (Haden Road; Fig. S24), we predict that the AMA effect may constitute as much as 11 %-43 % of total AVOC-SOA production, implying strong spatial variability in the relative contribution of this effect. Modeled AVOC-SOA production peaks in the early afternoon, consistent with the fact that the aging of condensable gases formed by measured VOCs produces the majority ($\sim 80\%$) of modeled AVOC-SOA.

Recent studies have revealed that a substantial fraction of SOA formation in urban environments may be produced by primary anthropogenic semi-volatile/intermediate volatility VOCs (P-S/IVOCs) co-emitted with typical VOCs or evaporated during POA dilution but not typically measured (Hayes et al., 2015, and references therein). In Los Angeles, for instance, Hayes et al. (2015) predicted that P-S/IVOCs comprise between 44 % and 92 % of total modeled SOA depending on the specific SOA formation mechanism used. As a result, the relative magnitude of the AMA effect may be somewhat overestimated here. Still, the AMA effect is responsible for $0.2-0.35 \,\mu\text{g}\,\text{m}^3$ of ambient aqSOA according to the model calculations, which represents 4 %-6 % of ambient OA measured by Cleveland et al. (2012) near downtown, $\sim 3 \% - 5 \%$ of OA modeled using CMAQ by Li et al. (2015), and as high as ~ 10 %–17 % of average OA measured in Houston's urban core by Leong et al. (2017). Furthermore, as AVOCs such as benzene and acetylene are known to produce gly-



Figure 10. (a) Diurnal SOA production rate calculated from anthropogenic VOCs (low estimate) measured by monitors in the HSC (filled) and the modeled-estimated production rate from WSOGs (glyoxal, methylglyoxal, IEPOX) attributable to the effect of anthropogenic marine aerosol (AMA) (hatched). (b) Total daily SOA production from WSOGs (upper and lower estimates) and HSC AVOCs (upper and lower estimates). WSOG aqSOA production is characterized by individual species. Note that IEPOX production is negligible. AVOC SOA production is characterized by lumped VOC species defined in Tsimpidi et al. (2010). These species are further described in the Supplement. Upper and lower estimates of AVOC SOA production are based on the assumed background OA mass loading, as described in the Supplement. High-NO_x product yields were used for both AVOC estimates.

oxal and methylglyoxal with high yields (Fu et al., 2008), the total OA mass attributable to AMA through this pathway (on an absolute rather than relative basis) may actually be larger than predicted here. Our results therefore suggest that future reductions in marine nss-SO₄ may reduce aqSOA formation in both urban (e.g., Houston) and forested regions across the Gulf Coast.

4 Conclusions

Three weeks of continuous measurements with an HR-ToF-AMS at a coastal location near Houston, TX, were used to gain further insight into the impact of shipping emissions on coastal aerosol properties. Measured mass loadings of inorganic NR-PM₁ components were similar to those reported before establishment of the ECA within the Gulf of Mexico; however, data from nearby ports suggest that this is the result of growth in the shipping trade rather than regulatory ineffectiveness on a per-vessel basis. Using MSA calibrations and published biogenic MSA/nss-SO₄ ratios, we predict that over 70 % of inorganic marine NR-PM₁ is anthropogenic rather than biogenic. Source apportionment using PMF revealed that the dominant marine OA factor (OOA-3) is highly correlated with calculated anthropogenic nss-SO₄ ($R^2 \ge 0.78$), supporting a link between shipping emissions and SOA production. Assuming, as an upper-bound estimate, that OOA-3 production is entirely dependent on shipping emissions, anthropogenic sources contribute over 70% of total measured NR-PM₁ during onshore flow, despite the regulations. This suggests that the proposed future global decrease in shipping FSC (decrease to 0.5%) should substantially reduce PM levels over the Gulf of Mexico. However, longer-term studies are warranted to fully investigate this possibility.

Shipping emissions were also found to have numerous secondary effects on OA composition. Detailed backwardtrajectory and mass spectral analysis revealed that air mass transit within shipping lanes leads to more processed (i.e., oxidized) OA than is encountered in "clean" marine air masses, and calculations suggest that this aging increases OA hygroscopicity. In addition, marine alkyl amine aerosol formation in the Gulf of Mexico appears to depend on ambient anthropogenic nss-SO₄ mass, implying that marine amine aerosol cannot be viewed as purely biogenic in heavily trafficked marine environments. OOA-3 was found to have a larger N : C ratio than is typical of aged marine components, supporting this link. Finally, modeling suggests that inland advection of shipping-related nss-SO₄ and related ALW may enhance aqSOA formation and produce 4 % to 17 % of OA in the urban core of Houston during marine flow for the conditions considered. Future detailed three-dimensional modeling studies are recommended to better quantify this effect.

Data availability. The compiled datasets used to produce each figure within the paper are available as Igor Pro files upon request.

Supplement. The supplement related to this article is available online at: https://doi.org/10.5194/acp-18-14217-2018-supplement.

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