

In situ chemical composition measurement of individual cloud residue particles at a mountain site, southern China

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Abstract. To investigate how atmospheric aerosol particles interact with chemical composition of cloud droplets, a ground-based counterflow virtual impactor (GCVI) coupled with a real-time single-particle aerosol mass spectrometer (SPAMS) was used to assess the chemical composition and mixing state of individual cloud residue particles in the Nanling Mountains (1690 m a.s.l.), southern China, in January 2016. The cloud residues were classified into nine particle types: aged elemental carbon (EC), potassium-rich (K-rich), amine, dust, Pb, Fe, organic carbon (OC), sodiumrich (Na-rich) and "Other". The largest fraction of the total cloud residues was the aged EC type (49.3 %), followed by the K-rich type (33.9%). Abundant aged EC cloud residues that mixed internally with inorganic salts were found in air masses from northerly polluted areas. The number fraction (NF) of the K-rich cloud residues increased within southwesterly air masses from fire activities in Southeast Asia. When air masses changed from northerly polluted areas to southwesterly ocean and livestock areas, the amine particles increased from 0.2 to 15.1 % of the total cloud residues. The dust, Fe, Pb, Na-rich and OC particle types had a low contribution (0.5-4.1%) to the total cloud residues. Higher fraction of nitrate (88-89%) was found in the dust and Na-rich cloud residues relative to sulfate (41-42%) and ammonium (15–23%). Higher intensity of nitrate was found in the cloud residues relative to the ambient particles. Compared with nonactivated particles, nitrate intensity decreased in all cloud residues except for dust type. To our knowledge, this study is the first report on in situ observation of the chemical composition and mixing state of individual cloud residue particles in China.

1 Introduction

Aerosol–cloud interactions influence the thermodynamic and radiation balance of the atmosphere (Boucher et al., 2013). Anthropogenic particles can increase the number concentration of small cloud droplets, and, in turn, affect reflectivity and lifetime of clouds (Stier et al., 2005; Lohmann et al., 2007; Rosenfeld et al., 2008). In situ cloud chemical measurements have shown varied chemical composition of cloud water or residues in various regions (Sorooshian et al., 2007a; Roth et al., 2016; Li et al., 2017). Despite a large number of aerosol and cloud studies over the past 20 years, the uncertainty for evaluating radiative forcing due to aerosol–cloud interactions has not been reduced (Seinfeld. et al., 2016). Therefore, it is crucial to assess how atmospheric aerosol particles contribute and interact with cloud droplets.

The ability of aerosol particles to act as cloud condensation nuclei (CCN) is dependent on the size and chemical composition of particles at a given supersaturation (McFiggans et al., 2006). Wiedensohler et al. (2009) found that the enhancement of particles' CCN ability was related to an increase in the average sulfate mass concentration. Dusek et al. (2006) demonstrated that CCN behavior was more affected by aerosol size than chemical composition. Meanwhile, aerosol mixing state also play an important role in the ability of aerosol to act as CCN. It has been reported that freshly emitted elemental carbon (EC) particles generally exhibit low CCN activity, whereas aged EC particles show high CCN activity after experiencing atmospheric processes (Zhang et al., 2008). Pratt et al. (2011) found that number fractions (NFs) of biomass-burning particles internally mixed with ammonium or oxalate increased with the ageing process, which promote their CCN behavior. Laboratory studies have shown that low-solubility organic particles internally mixed with ammonium sulfate would suppress water uptake of mixed particles and thus might affect CCN activity (Wise et al., 2003; Svenningsson et al., 2006; Sjogren et al., 2007). An overprediction of CCN concentration by up to 35 % was estimated based on particle internal mixingstate assumptions (Medina et al., 2007; Collins et al., 2013). The influence of mixing state on aerosol CCN activity varies depending on the proximity to the pollution plume source and/or photochemical ageing activity (Ervens and Volkamer, 2010). More detailed measurements to characterize the mixing state of CCN particles would improve our understanding of aerosol-cloud interactions.

The combined technique of a counterflow virtual impactor (CVI) and an aerosol mass spectrometer (AMS) or other online or offline single-particle instruments is widely used to characterize the chemical composition and/or mixing state of cloud or fog droplet residue particles. These studies were mainly conducted at sites in North America including Wyoming (Pratt et al., 2010a), Ohio (Hayden et al., 2008), Oklahoma (Berg et al., 2009), Florida (Cziczo et al., 2004; Twohy et al., 2005) and California (Coggon et al., 2014); in Europe including Schmücke (Roth et al., 2016; Schneider et al., 2017), Jungfraujoch (Kamphus et al., 2010), Åreskutan (Drewnick et al., 2007) and Scandinavia (Targino et al., 2006); in the Arctic (Zelenyuk et al., 2010); in Central America (Cziczo et al., 2013); in western Africa (Matsuki et al., 2010); and in marine areas (Twohy and Anderson, 2008; Twohy et al., 2009; Shingler et al., 2012).

Over the past three decades, China has undergone rapid economic growth accompanied by increased aerosol emissions. Although scientists have worked to increase our understanding of an emissions inventory and the temporal and spatial variation of atmospheric aerosols in China (X. Y. Zhang et al., 2012), few studies have employed direct observation of the chemical composition and mixing state of cloud or fog droplets. W. Li et al. (2011) utilized transmission electron microscopy to obtain the mixing state of individual ambient particles during cloud events at Mount Tai in northern China. This result showed that sulfate-related salts dominated in larger particles. Bi et al. (2016) used a groundcounterflow virtual impactor (GCVI) coupled with a realtime single-particle aerosol mass spectrometer (SPAMS) to explore the chemical composition and mixing state of individual fog residual particles at ground level in an urban area of southern China. They found an abundance of ECcontaining particles in fog residues.

Here, we present a study on the chemical composition and mixing state of individual cloud residue particles at a mountain site in southern China. The same experimental methods of Bi et al. (2016) were used in this study. The size distribution, chemical composition and mixing state of cloud residues during cloud events are discussed. Moreover, the chemical compositions of ambient and nonactivated particles were also compared with the cloud residues. The aim of this study is to assess the potential effects of anthropogenic aerosols from regional transportation on cloud formation and to investigate the dominant particle types in cloud droplets at a mountain site in southern China.

2 Experimental

2.1 Measurement site

Our measurements were carried out during 15–26 January 2016. The sampling site was located in the Nanling Background Station ($112^{\circ}53'56''$ E, $24^{\circ}41'56''$ N; 1690 m a.s.l.) at the National Air Pollution Monitoring System in southern China (Fig. S1 in the Supplement). This station is located at 200 km north of the metropolitan city Guangzhou and 350 km north of the South China Sea (Fig. S1). This site is also surrounded by a national park forest (273 km²), where there are scarcely any emissions from anthropogenic activities. During the winter monsoon period, air pollution from northern China moves to southern China and crosses the study region (Lee et al., 2005).

2.2 Instrumentation

In this study, a GCVI inlet system (GCVI Model 1205, Brechtel Manufacturing Inc.) was used to sample cloud droplets with a diameter greater than 8 µm. The ambient temperature on average was 6.9 °C (ranging from -7.2 to 11.4 °C) during cloud events. Some mineral dust might trigger heterogeneous ice nucleation at temperature below $-7 \,^{\circ}C$ (Atkinson et al., 2013). Only 20 cloud residues that accounted for 0.08 % of the total cloud residues were detected when the ambient temperature was below $-7 \,^{\circ}\text{C}$ when observed from 06:00 to 08:00 LT on 23 January. Thus, cloud droplets were dominated by liquid water droplets. The measurements of the droplet size spectra in this region performed during the winter of 1999-2001 showed that sizes of cloud droplets ranged from 4 to 25 µm, with an average size of 10 µm and a corresponding liquid water content of 0.11- 0.15 gm^{-3} (Deng et al., 2007). A previous study at another mountain site also showed an average size of $\sim 10 \,\mu m$ (Borys et al., 2000). Hence, it was assumed that size distribution of cloud droplets was mostly above 8 µm in this region. The sampled cloud droplets were passed through an evaporation chamber (air flow temperature at 40 °C), where the associated water was removed and the dry residue particles (with the air flow relative humidity (RH) lower than 30 %) remained. A stream of filtered and heated ambient air (counterflow) was provided by a compressor. The particle transmission efficiency of the cut size (8 µm) was 50 %. The enrichment factor of the particles collected by the GCVI inlet was estimated to be 5.25 based on theoretical calculation (Shingler et al., 2012). Ambient particles were collected through an ambient inlet with a cut-off aerodynamic diameter (d_a) of 2.5 µm when cloud-free periods were present. Nonactivated (interstitial) particles were sampled through the ambient inlet during the cloud events in this study. The ambient or nonactivated-particle inlet was dried using a silica gel diffusion dryer. During cloud-free periods, a ratio of particle concentration measured behind the CVI (below 1 cm^{-3}) to ambient aerosol concentration (2000 cm⁻³) was 0.0005, indicating that instances of particle breakthrough and small particle contamination were absent. The cloud droplet residues, ambient or nonactivated particles were subsequently analyzed by a suite of aerosol measurement devices, including a SPAMS (Hexin Analytical Instrument Co., Ltd., Guangzhou, China), a scanning mobility particle sizer (SMPS; MSP Cooperation) and an Aethalometer (AE-33, Magee Scientific Inc.).

A detailed operational principle of the SPAMS has been described elsewhere (L. Li et al., 2011). Briefly, aerosol particles are drawn into SPAMS through a critical orifice. The particles are focused and aerodynamically sized by two continuous diode Nd:YAG laser beams (532 nm). The particles are subsequently desorbed and ionized by a pulsed laser (266 nm) triggered exactly based on the velocity of the specific particle. The positive and negative ions generated are recorded with the corresponding size of individual particles. Polystyrene latex spheres (Nanosphere Size Standards, Duke Scientific Corp., Palo Alto) of 0.2-2.0 µm in diameter were used to calibrate the sizes of the detected particles. The ambient pressure was 830 hPa (826-842 hPa) during the measurements and the calibration. Particles measured by SPAMS mostly fell within the size range of vacuum aerodynamic diameter (d_{va}) 0.2–2.0 µm (L. Li et al., 2011).

2.3 Definition of cloud events

To reliably identify the presence of cloud events, an upperlimit visibility threshold of 5 km and a lower-limit RH threshold of 95 % were set in the GCVI software (Bi et al., 2016). Three long-lasting cloud events occurred during the periods of 16:00 LT 15 January to 07:00 LT 17 January (cloud I), 20:00 LT 18 January to 12:00 LT 19 January (cloud II) and 17:00 LT 19 January to 13:00 LT 23 January (cloud III), as marked in Fig. 1. In addition, a cloud event occurred at Figure 1. The hourly average variations in meteorological conditions (temperature, relative humidity, visibility, pressure, wind speed and direction) and $PM_{2.5}$.

14:40-15:00 LT on 17 January, but we did not do an analysis due to the short duration. The measured cloud residual concentration was integrated by the SMPS and was then corrected by the enrichment factor and transmission efficiency of the GCVI. The corrected cloud residual concentrations on average were 436, 568 and 544 cm^{-3} for cloud I, cloud II and cloud III, respectively (Fig. S2). From 10:00 LT 21 January to 13:00 LT 23 January, cloud residues and nonactivated particles were alternately sampled with an interval of 1 h. During this period, a ratio of number residues to total number particles (sum of cloud residues and nonactivated particle) on average was 0.43 ± 0.20 . Low levels of PM_{2.5} (~ 12.7 µg m⁻³) exclude the influence of hazy days. A rainfall detector of the GCVI system was also used to exclude rain droplet contamination. When cloud events occurred without precipitation, sampling was automatically triggered by the GCVI control software (Bi et al., 2016).







Figure 2. (a) HYSPLIT back trajectories (72 h) for air masses at 1800 m during the whole sampling period. The white borders and circle refer to the Pearl River Delta (city cluster 1) and Yangtze River mid-reach city clusters (city cluster 2), respectively. The fire data (yellow dots) are available at https://earthdata.nasa.gov/; (b) heights (above sea level) of the air masses as a function of time.

2.4 Particle classification

During this study period, a total of 73 996 particles including 49 322 ambient, 23 611 cloud residual and 1063 nonactivated particles with bipolar mass spectra were chemically analyzed in the size range of d_{va} 0.2–1.9 µm. The sampled particles were firstly classified into 101 clusters using an adaptive resonance theory neural network (ART-2a) with a vigilance factor of 0.75, a learning rate of 0.05 and 20 iterations (Song et al., 1999). Then by manually combining similar clusters, eight major particle types (aged EC, potassiumrich (K-rich), amine, dust, Fe, Pb, organic carbon (OC) and sodium-rich (Na-rich)) with distinct chemical patterns were obtained, which represented ~ 99.9 % of the population of the detected particles. The remaining particles were grouped together as "Other". Assuming that the number of individual particles followed Poisson distribution, standard errors for NF of particle type were estimated (Pratt et al., 2010a).

3 Results and discussion

3.1 Back trajectories and meteorological conditions

Back trajectories in this study were calculated using the Hybrid Single Particle Lagrangian Integrated Trajectory (HYS-PLIT) model. A height of the HYSPLIT model in the study region (a spatial resolution of $0.5^{\circ} \times 0.5^{\circ}$) is averaged 500 ma.s.l., lower than the height of the observed site (1690 ma.s.l.). Thus, a height of 1800 ma.s.l. (approximately 100 m above the observed site) was chosen as an endpoint in the model. The station was mainly affected by southwesterly or northerly air masses in this study (Fig. 2). In addition, the beginning altitude of the southwesterly air masses traversed at lower heights relative to the northerly

air masses (Fig. 2). The southwesterly air masses, accompanied by warm moist airflows, occurred during 15–17 and 20–21 January, which promoted high RH condition (Fig. 1). Conversely, the northerly air masses, associated with cool dry airstreams, occurred during 18 and 23–24 January and led to a decrease in temperature and RH. Note that, on 18– 19 and 22–23 January, the air mass encountered initial mixing of northerly cloud-free air and southwesterly cloudy air. Entrainment of nuclei particles originated from northern air masses might be activated to cloud droplets (Sect. 3.4).

Meteorological conditions were unstable, with high southwesterly flow (~ 6.5 m s^{-1}) during 15–17 and 20–22 January (Fig. 1). The level of PM_{2.5} remained a low value of approximately 3 µg m⁻³ for this time period. A high level of PM_{2.5} (~ 20 µg m⁻³) was observed during 18 January when the northerly flow dominated. Similarly, the average PM_{2.5} value reached 24 µg m⁻³ during 24 January. Although the local northerly and southwesterly flows occurred alternately, the particles were still originated from the northerly air mass for this period (Fig. 2). During 23–24 January, a sharp decrease in temperature (Fig. 1) was observed due to a cold wave associated with a violent northerly flow. The wind speed during the cold wave exceeded the upper-limit speed (~ 12 m s^{-1}) of a wind speed sensor.

3.2 The chemical characterization of cloud droplet residues

Figure 3 shows the average positive and negative mass spectra of the main six particle types. The aged EC particles were characterized by EC cluster ions (e.g., $m/z \ 12C^{\pm}$, $36C_3^{\pm}$, $48C_4^{\pm}$, $60C_5^{\pm}$, ...) and a strong K⁺ ion signal ($m/z \ 39K^+$) as well as a sulfate ion signal ($m/z \ -97HSO_4^-$), and some minor organic markers ($m/z \ 27C_2H_3^+$, $43C_2H_3O^+$; Moffet



Figure 3. Averaged positive and negative mass spectra for the main six particle types (aged EC, K-rich, amine, dust, Fe and Na-rich) of the sampled particles. RPA on the vertical axis refers to relative peak area. On the horizontal axis, m/z represents mass-to-charge ratio.

and Prather, 2009). EC particles mainly originated from combustion processes (Bond et al., 2013). The strong K^+ ion signal in the aged EC particles implies that they partially originated from biomass-burning sources (Bi et al., 2011). The aged EC particle type was the largest fraction (49.3%)of the total cloud residues (Fig. S3). In addition, the NF of the aged EC residues significantly decreased from 54.1 % in the size range of 0.2–1.0 µm to 19.2 % in the size range of 1.1–1.9 µm (Fig. 4). Note that the chemical composition of cloud residues is dependent on the particle size (Roth et al., 2016), and the number reported for each particle type might suffer from the bias related to size-dependent transmission efficiency (Qin et al., 2006). The relative fraction of cloud residues in the 0.1 µm size interval is presented to minimize the influence of size-dependent transmission efficiency of single-particle mass spectrometry (Roth et al., 2016).

The K-rich particles exhibited the highest peak at m/z 39K⁺, mainly combined with sulfate and nitrate $(m/z - 46NO_2^-, -62NO_3^-)$ and presumably derived from biomass and/or biofuel burning sources (Moffet et al., 2008; Pratt et al., 2011; Zhang et al., 2013). An aged time of 81–88 min biomass-burning particles were found to show an increase in the mass fractions of ammonium, sulfate and nitrate (Pratt et al., 2011). In this study, the K-rich parti-



Figure 4. Number fractions for size distribution of the cloud residual types in 100 nm size intervals.

cles would be expected to experience aged process due to strong sulfate and nitrate signals (Hudson et al., 2004; Pratt et al., 2011). Aged biomass-burning particles can participate in cloud droplet formation and show an effective CCN activity (Pratt et al., 2010a). The K-rich particle type, the second largest contributor, accounted for 33.9 % of the total cloud residues (Fig. S3).

The abundant aged soot, EC and biomass-burning particles were often detected in cloud residues (Pratt et al., 2010a; Roth et al., 2016). The contribution of local anthropogenic origins to aged soot and/or biomass-burning particles in cloud or fog residues has been reported in Schmücke (Roth et al., 2016) and Guangzhou city (Bi et al., 2016). At the North Slope of Alaska, the measurement of biomassburning particles in cloud residues mainly resulted from the local vicinity, or sources as far away as Siberia and Asia (Zelenyuk et al., 2010; Hiranuma et al., 2013). Similarly, the majority of aged EC and K-rich cloud residues observed here are expected to originate from long-range transportation due to insignificant sources of local anthropogenic emissions and the fire dots (Fig. 2). At the Jungfraujoch station (3580 ma.s.l.) in Europe, the K-rich (biomass-burning) particles only contributed 3% of the cloud droplets, and the aged EC residuals were insignificant (< 1 %; Kamphus et al., 2010). The Jungfraujoch station is predominantly within the free tropospheric condition, such that the biomass-burning contribution can be expected to be lower than at other sites.

The amine particles were characterized by related amine ion signals at m/z 58C₂H₅NHCH₂⁺ (diethylamine, DEA), $59N(CH_3)_3^+$ (trimethylamine, TMA) and $86C_5H_{12}N^+$ (triethylamine, TEA; Angelino et al., 2001; Moffet et al., 2008). This particle type also contained sulfuric acid ion signals at $m/z - 195 H(HSO_4)_2^-$, indicative of acidic particles (Rehbein et al., 2011). The amine particles represented 3.8 % of the total cloud residues (Fig. S3). Higher NF of the amine residues was detected in the size range from 0.7 to 1.9 µm relative to the size range from 0.2 to 0.6 µm (16.7 vs. 0.4%), as shown in Fig. 4. Aqueous reactions improving the participation of amine have been observed in Guangzhou (G. Zhang et al., 2012) and southern Ontario (Rehbein et al., 2011). A recent study also showed a clear enhancement of amine-containing particles in cloud residues compared to the ambient particles (9 vs. 2%; Roth et al., 2016). It indicates a preferential formation of amine within the cloud, which is in contrast to the observations of Bi et al. (2016). It might suggest that enhancement of particle amine is not only dependent on high RH or fog or cloud processes, but is also sensitive to other parameters, such as the presence of gas-phase amine sources (Rehbein et al., 2011).

The dust particles presented significant ions at m/z $40Ca^+$, $56CaO^+/Fe^+$, $96Ca_2O^+$ and $-76SiO_3^-$ and sulfate as well as nitrate markers. Previous studies showed that dust particles that are internally mixed with sulfate and nitrate are expected to act as CCN (Twohy and Anderson, 2008; Twohy et al., 2009), despite sulfate and nitrate partially forming from in-cloud production. This type contributed 2.9 % of the total cloud residues (Fig. S3). A slight increase in NF of the dust residues was observed in the size range above 0.5 µm relative to that below 0.5 µm (3.0 vs. 1.0%). At Mount Tai in northern China, a high concentration of Ca²⁺ in cloud or fog water was mainly attributed to a sandstorm event during the spring season (Wang et al., 2011). At Mount Heng in southern China, the abundant crust-related elements (e.g., Al) observed in cloud water is due to Asian dust storms that occurred in March-May (Li et al., 2017). Based on the backward trajectory, the site was unlikely affected by sandstorm source in northwestern China during the cloud events. Local dust emissions from anthropogenically disturbing soils or removing vegetation cover can be excluded as a result of forest protection. Additionally, dust residues may have occupied larger CCN (Tang et al., 2016), which cannot be detected by the SPAMS. Hence, a low fraction (2.9%) of dust cloud residue might be due to the limitation of the SPAMS.

The Fe particles, internally mixed with sulfate and nitrate and with typical ions at m/z 56Fe⁺, made up 4.1% of the total cloud residues. Approximately 16% of the Fe cloud residues contained Ca⁺ peak (m/z 40). Predominant Fe ion peaks possibly indicate the contribution from anthropogenic sources (Zhang et al., 2014), especially the northern air masses across iron and steel industrial activities in Yangtze River mid-reach city clusters (Fig. 2). The contribution of anthropogenic and natural Fe-containing particle sources (Moteki et al., 2017) to observed Fe-containing residues is expected. The presence of Fe in the cloud droplets plays an important role in aqueous-phase SO₂ catalytic oxidation in cloud processing (Harris et al., 2013), thus accelerating the sulfate content of Fe-containing particles in cloud processing.

The Na-rich particles were mainly composed of ion peaks at m/z 23Na⁺ and 39K⁺ in the positive mass spectra, and nitrate and sulfate species in the negative mass spectra made up 3.0 % of the total cloud residues. Na-rich particles are formed from varied sources including industrial emissions, sea salt or dry lake beds (Moffet et al., 2008). The NF of Na-rich cloud residues did not increase from continental (northerly) air mass on 19 January to maritime (southwesterly) air mass on 21 January (3.3 vs. 2.4%). However, the related sea salt ion peak area $(m/z, 81/83 \text{ Na}_2^{35}\text{Cl}^+/\text{Na}_2^{37}\text{Cl}^+)$ were enhanced for Na-rich particles originating from maritime air mass relative to continental air mass $(3.8 \pm 2.4 \text{ times})$. The continental air masses crossed industrial areas where the Yangtze River mid-reach city cluster is located (Fig. 2). Industrial emissions were a possible contributor to Na-rich particles under the influence of continental air masses (Wang et al., 2016). This might suggest that the Na-rich particles originated from both industrial emissions and sea salts.

The OC, Pb and Other particle types contributed 0.1–2.3 % to the total cloud residues (Fig. S3). Their average mass spectra can be found in Fig. S4. The OC particles presented dominant intense OC signals (e.g., $m/z \ 27C_2H_3^+$, $37C_3H^+$, $43C_2H_3O^+$ and $51C_4H_3^+$) and abundant sulfate. Presence of K⁺ signal was found in the OC particles, suggesting possible biomass-burning sources (Bi et al., 2011). OC particles might exist in smaller cloud residues (Sellegri et al., 2003a), which cannot be detected by the SPAMS. The Pb particles had typical ions at $m/z \ 208Pb^+$ and were internally mixed with K⁺ and Cl⁻. Previous studies have found that K and Cl internally mixed with Pb particles have a possible origination of waste incineration (Zhang et al., 2009) or iron and steel product

manufacturing facilities (Tsai et al., 2007). Only three particles were found containing calcium, organic carbon, organic nitrogen and phosphate ion signals, implying a possible existence of biological particles (Pratt et al., 2009a). Such particles were classified as the Other type due to low number.

Previous measurements have found that dust, playa salts, sea salt or metal particles were often enriched in larger cloud droplets ($\sim 20 \,\mu\text{m}$; Bator and Collett, 1997; Moore et al., 2004; Pratt et al., 2010b). Organic carbon tended to be enriched in small cloud or fog droplets, extending to $4 \,\mu\text{m}$ (Herckes et al., 2013). The size of cloud droplets were above 8 μm in the present study. Additionally, the particle transmission efficiency increased with increasing cloud droplet size (Shingler et al., 2012). Thus, it partially leads to relatively larger fractions of the observed dust, Na-rich and metal cloud residues, and the smaller fraction of the OC cloud residues in this study.

3.3 Mixing state of secondary species in cloud residues

The high NFs of sulfate-containing particles were found in the K-rich (91%), OC (100%), aged EC (98%), Pb (74%), Fe (93%) and amine (99%) cloud residues, as shown in Fig. 5. Lower NFs of sulfate-containing particles were observed in the Na-rich (41%) and dust (42%) cloud residues. In contrast, nitrate-containing particles contributed 89 and 88 % to the Na-rich and dust cloud residues, respectively. The acid displacement reaction of sea salt chloride (Na-rich particles) by HNO₃ might lead to a depletion of chloride and enhancement of nitrate (Laskin et al., 2012). Similarly, the heterogeneous chemistry of HNO₃ in the dust particles also contributes the preferential enrichment of nitrate (Tang et al., 2016). Moreover, after activation, uptake of gas-phase HNO₃ would increase nitrate level in the cloud residues (Schneider et al., 2017). The nitrate in the cloud residues was thought to be in the form of ammonium nitrate by estimating the ratio of m/z 30 to m/z 46 in AMS data (Drewnick et al., 2007; Hayden et al., 2008). Relative to nitrate, low portions of ammonium $(m/z, 18\text{NH}_4^+)$ in the Na-rich (23 %) and dust (15%) cloud residues suggest that in this region, ammonium nitrate was not a predominant form of nitrate in the two cloud residual types. The Na-rich and dust types were mainly composed of alkaline ion peaks $(m/z, 23Na^+, 39K^+ \text{ and } 40Ca^+)$ in mass spectra (Fig. 3), accompanied by a larger fraction (88-89%) of nitrate. Thus, our data suggest that nitrate might exist in the form of Ca(NO₃)₂, NaNO₃ or KNO₃ in the dust and Na-rich cloud residues. It should be noted that the evaporation chamber of the GCVI may lead to a reduction of ammonium nitrate in the cloud residues (Hayden et al., 2008). The nitrate-containing particles accounted for only 46% of the aged EC cloud residues, which is significantly less than the sulfate-containing particles. Previous field studies have found that aged EC (soot) fog/cloud residues are mainly internally mixed with sulfate (Pratt et al., 2010a; Harris et al., 2014; Bi et al., 2016). Aged EC particles mixed with sulfate



Figure 5. Number fractions of secondary markers associated with the total cloud residues types: sulfate $(m/z, -97\text{HSO}_4^-)$, nitrate $(m/z, -46\text{NO}_2^- \text{ or } -62\text{NO}_3^-)$, ammonium $(m/z, 18\text{NH}_4^+)$, sulfuric acid $(m/z, -195\text{H}(\text{HSO}_4)_2^-)$, TMA $(m/z, 59\text{N}(\text{CH}_3)_3^+)$ and oxalate $(m/z, -89\text{HC}_2\text{O}_4^-)$.

are good CCN (Bi et al., 2016; Roth et al., 2016). Laboratory measurements have also demonstrated that EC particles internally mixed with sulfate showed a high hygroscopic behavior and thus affect CCN ability (McMeeking et al., 2011). High portions (75–86 %) of ammonium-containing particles were observed for the OC and aged EC cloud residues, suggesting that ammonium will mostly be in the form of ammonium sulfate or ammonium nitrate for the two cloud residual types. This result also implies that ammonium-containing particles are preferentially activated or enhanced uptake of gaseous NH₃ to neutralize acidic cloud droplets for the OC and EC types.

Water soluble organics (e.g., amine and oxalate) have previously been measured in cloud water or residues (Sellegri et al., 2003b; Sorooshian et al., 2007a; Pratt et al., 2010a). The presence of TMA (93%) in the amine cloud residues is expected to promote water uptake activity (Sorooshian et al., 2007b). A total of 3410 oxalate-containing particles (m/z, m/z) $-89HC_2O_4^-$) represented 14.4 % of the total cloud residues, which was mainly associated with the K-rich cloud residues $(\sim 70\%)$. Oxalate-containing particles $(\sim 30\%)$ in the metal (Pb, Fe) cloud residues might be in the form of metal oxalate complexes from reactions of in-cloud formation oxalate with metals (Furukawa and Takahashi, 2011). Oxalate can readily partition into the particle phase to form amine salts (Pratt et al., 2009b). It may facilitate the entrainment of oxalate (33%) in the amine residues. A low fraction (4%) of oxalate-containing particles in the OC type is a result of restrictive classification. Classification of the OC particles is mainly based on intense organic carbon ion signals (e.g., m/z $27C_2H_3^+$, $37C_3H^+$, $43C_2H_3O^+$ and $51C_4H_3^+$). The majority of oxalate-containing particles were internally mixed with the K-rich type. Therefore, oxalate was classified as the Krich type and was probably contributed from biomass burning. The K-rich particles could contain an abundance of organics (Pratt et al., 2011); however, the signals of organics



Figure 6. The hourly average variations in the cloud residual and ambient particles during the whole sampling period.

were covered by the potassium due to its high sensitivity to the laser.

3.4 Comparison of cloud residues in different air mass sources

Figure 6 displays the hourly detected particle counts and NF values of the nine types of cloud residues and ambient particles. The NF of the aged EC type showed a very abrupt increase from cloud residues to ambient particles on 17 January. The ambient RH showed an abrupt decrease from nearly 100% at 10:00 LT to 85% at 11:00 LT on 17 January (Fig. 1). The ambient temperature also decreased from 10 °C at 11:00 LT to 4 °C at 18:00 LT on 17 January (Fig. 1). These changes imply that the air mass shifted from southwesterly cloudy air to northerly cloud-free air around noon on 17 January (Fig. 2). The entrained particles that originated from northern air mass might have insufficient supersaturation to be activated as cloud droplets. It resulted in the remarkable increase of the aged EC particles in ambient particles on 17 January (Fig. 6).

The ambient RH increased from 60% at 19:00 LT to nearly 100% at 21:00 LT on 18 January (Fig. 1). The ambient temperature also increased from 1.3 °C at 22:00 LT on 18 January to 3.2 °C at 06:00 LT on 19 January (Fig. 1). These changes imply that the air mass changed from northerly cloud-free air to southwesterly cloudy air at night on 18 January (Fig. 2). During 18-19 January, the cloud residues and ambient particles showed similar chemical characteristics and were dominated by aged EC particles (Fig. 6). A lack of significant variation in the particle types for this period suggests that nuclei particles originated from northerly cloud-free air could be activated to become cloud droplets. When a cloud-free event occurred at 11:00-17:00 LT on 19 January, ambient particles retained a high level of $PM_{2.5}$ (~ 22.7 µg m⁻³). The southwesterly wind flow on 19-20 January was too weak $(\sim 2.75 \,\mathrm{m\,s^{-1}})$ to dilute particles that originated from the northerly air masses (Fig. 1). Additionally, a high RH (90%) air mass at 1500 ma.s.l. gradually moved to northern China from 19 to 20 January (Fig. S5). These changes might have led to similar residual particle types observed from 19 to 20 January, although the site encountered southwesterly cloudy air on 19-20 January (Fig. 2).

As mentioned above, the NF of the cloud residue types significantly changed as the air mass origin varied from northerly to southwesterly. To further investigate the influence of air mass history, we chose to analyze cloud residues that had arrived from a northerly air mass on 18–19 January compared to cloud residues that originated from a southwesterly air mass during the periods of 16–17 and 21–22 January. The detected number of cloud residues for both the northerly and southwesterly air masses are given in Table S1. The southwesterly air masses accompanied by high RH (90%; Fig. S5) may have triggered particles activated to CCN prior to their arrival at the sampling site.

The K-rich type was found to contribute 23.9% to the cloud residues in the northerly air mass, which was significantly lower than its contribution to the southwesterly air mass (51.5%), as shown in Fig. 7. A similarity in averaged mass spectrum of the K-rich residues was found for the southwesterly and northerly air masses (Fig. S6). The considerable increase of K-rich cloud residues suggests a major influence of regional biomass-burning activities. Biomassburning emissions from Southeast Asia, including Myanmar, Vietnam, Laos and Thailand, where abundant fire dots are observed (Fig. 2), could have been transported to the sampling site under a southwesterly air mass (Duncan et al., 2003). In contrast, the aged EC type represented only 23.7 % of the cloud residues under the influence of the southwesterly air mass, which was significantly lower than observations for the northerly air mass (59.9%), as shown in Fig. 7. This result suggests that the northern air mass has a greater influence on the presence of aged EC cloud residues.

An obvious increase in NF of the amine type was observed in the southwesterly air mass (15.1%) compared to the northerly air mass (0.2%), as shown in Fig. 7. These data imply that the sources or formation mechanisms of amine in cloud residues varied in different air masses. The southwesterly air masses arrived from as far as the Bay of Bengal and then traveled through the Southeast Asia region before reach-



Figure 7. Number fractions of the cloud residues, ambient particles and nonactivated particles; (a) cloud residues during northerly air mass; (b) ambient particle during northerly air mass; (c) cloud residues during southwesterly air mass; (d) cloud residues; and (e) nonactivated particles were alternately sampled with an interval of 1 h during the cloud III event. Uncertainties were calculated assuming Poisson statistics for analyzed particles.

ing southern China (Fig. 2). The potential gas amine emissions from ocean (Facchini et al., 2008) and livestock areas (90 million animals; data are available at the website http: //faostat3.fao.org) in the Southeast Asia region might promote the enrichment of amine particles. Furthermore, after activation, the partitioning of the gas amine on cloud droplets may further contribute to the enhanced amine cloud residues (Rehbein et al., 2011), especially for air masses delivered via routes with high RH, as mentioned above (Fig. S5). In contrast, northerly air mass accompanied with dry airstreams may unfavorably induce the partitioning of gas amines into the particle phase (Rehbein et al., 2011).

3.5 Comparison of cloud residues with ambient and nonactivated particles

A direct comparison between cloud residues and ambient particles was limited because of their differences in air mass origins. During the sampling period, the cloud events occurred once the southwesterly air masses were dominant. Hence a comparison between cloud residues and ambient particles cannot be addressed under the influence of southwesterly air masses. Here, we chose 5 h before and after the beginning of the cloud II period in order to compare cloud residues and ambient particles with similar northerly air mass origins, as discussed in Sect. 3.4.

The cloud residues and nonactivated particles were alternately sampled with an interval of 1 h from 21 January to



Figure 8. Mass spectral subtraction plot of the average mass spectrum corresponding to cloud residues minus ambient particles. Positive area peaks correspond to higher abundance in cloud residues, whereas negative area peaks show higher intensity in ambient particles.

23 January. The ambient temperature decreased from 6 °C at 11:00 LT to 0 °C at 23:00 LT on 22 January (Fig. 1). The ambient-particle level (sum of residual and nonactivated particles) showed a clear increase from 156 to 1460 cm⁻³ during this period (Fig. S2). Thus, the data suggest that the initial mixing of northerly cloud-free air and southwesterly cloudy air occurred at around noon on 22 January. A reduction of supersaturation due to entrainment of the dry northern air mass might have insufficient moisture to activate small particles, leading to nonactivated particles above 0.2 µm (Fig. S7; Mertes et al., 2005; Kleinman et al., 2012; Hammer et al., 2014), which can be detected by the SPAMS.

The contribution of K-rich particles in cloud residues (23.9%) slightly decreased relative to ambient particles (30.7%), as shown in Fig. 7. Previous studies have found that there were no significant changes in NF of biomassburning particles for cloud residues relative to ambient particles (Pratt et al., 2010a; Roth et al., 2016). The biomassburning particles internally mixed with soluble species (e.g., sulfate, nitrate and oxalate) enhanced their ability to act as CCN, as discussed in Sect. 3.3. Kamphus et al. (2010) reported that biomass-burning particles account for only 3 % of cloud residues compared with 43 % of ambient particles, and they suspected that biomass-burning particles might exist in the form of tar balls (hydrophobic materials). A slight increase in NF of the aged EC cloud residues (59.9 %) was observed relative to ambient particles (53.8 %), as shown in Fig. 7. In general, freshly emitted EC particles are less hydrophilic and are not active as CCN (Bond et al., 2013). The aged EC particles show a high degree of internal mixing with secondary inorganic compounds in this study (Fig. 5), improving their ability to act as CCN. The remaining particle types showed no clear differences in NF between cloud residues and ambient particles.

When comparing the cloud residues with nonactivated particles, a significant change in NF was found for the aged EC and K-rich type. A higher NF of K-rich particles and a lower NF of EC particles were found for the cloud residues rel-



Figure 9. Mass spectral subtraction plot of the average mass spectrum corresponding to cloud residues minus nonactivated particles. Positive area peaks correspond to higher abundance in cloud residues, whereas negative area peaks show higher intensity in nonactivated particles.

ative to the nonactivated particles (Fig. 7). Entrainment of northerly cloud-free air might lower supersaturation during this period. Aged EC particles may require very high supersaturation to grow into cloud droplets and, thus, only form hydrated nonactivated aerosol (Hallberg et al., 1994).

Figures 8 and 9 show the differences in average mass spectra for cloud residues vs. ambient particles as well as cloud residues vs. nonactivated particles, respectively. Nitrate intensity (average ion peak area) enhanced in the cloud residues when compared to ambient particles. In addition, nitrate-containing particles accounted for 70% of the cloud residues compared to 38% of the ambient particles. Drewnick et al. (2007) suggested that rather than sulfate, high nitrate content in pre-existing particles preferentially acted as cloud droplets. Compared with nitrate-containing ambient particles, the larger sizes of nitrate-containing residues (Fig. S8) possibly reflect the uptake of gaseous HNO₃ during cloud processes (Hayden et al., 2008; Roth et al., 2016). A recent study also confirmed that the update of gaseous HNO₃ is an important contributor for the increased nitrate level in the cloud residuals (Schneider et al., 2017). Interestingly, we observed a decrease in nitrate intensity in all cloud residues except for dust type (Fig. 9), and a large size distribution of nitrate-containing cloud residues (Fig. S7) when compared with nonactivated particles. This result suggests that particle size, rather than nitrate content, plays a more important role in the activation of particles into cloud droplets.

Sulfate intensity enhancement was only observed in the OC cloud residues relative to both ambient and nonactivated particles. Although the in-cloud addition of sulfate can be produced from aqueous Fe-catalyzed or oxidation by H_2O_2/O_3 reactions (Harris et al., 2014), sulfate abundance was found in the Fe cloud residues relative to nonactivated particles, but no enhancement relative to ambient particles was found. Previous studies also showed that the mass or NF of sulfate-containing particles in the cloud residues changed between ambient and nonactivated particles (Drewnick et al., 2007; Twohy and Anderson, 2008; Schneider et al., 2017). However, the reason for these changes remains unclear.

The in-cloud process has been reported to be an important pathway for the production of amine particles (Rehbein et al., 2011; G. Zhang et al., 2012). In this study, no remarkable change in NF of the amine cloud residues was obtained relative to the ambient particles (0.2 vs. 0.2 %), as shown in Fig. 7. The absence of amine species in cloud residues may be partially affected by droplet evaporation in the GCVI (Bi et al., 2016). However, there was a high fraction of the amine cloud residuals when the southwesterly air mass prevailed, as discussed in Sect. 3.4. A lack of gas-phase amines may be the cause of the few amine particles detected in the ambient particles and cloud residues (Rehbein et al., 2011). An increase in NF of cloud residues was observed compared with nonactivated particles (5.2 vs. 0.1%), as shown in Fig. 7. An increase of particle water content facilitates partitioning of gas-phase amine species into the aqueous phase when gasphase amines are present (Rehbein et al., 2011).

4 Conclusions

This study presented an in situ observation of individual cloud residues, nonactivated particles and ambient particles at a mountain site in southern China. The finding shows that the aged EC (49.3%) and K-rich types (33.9%) dominate the cloud residues in a remote area of China, followed by the Fe (4.1%), amine (3.8%), Na-rich (3.0%) and dust (2.9%) types. The OC, Pb and Other types contributed 0.1-2.3 % to the total cloud residues. The observed change in NF of the cloud residue types, influenced by various air masses, highlights the important role of regional transportation in the observed cloud residual chemistry. Amine particles represented from 0.2 to 15.1 % of the total cloud residues, dependent on the air mass history. Sulfate was found to be highly mixed with the K-rich, OC, aged EC, Pb, Fe and amine cloud residues, while nitrate was highly mixed with the Na-rich and dust cloud residues. Compared with nonactivated particles, nitrate intensity decreased in all cloud residues except for dust type, and sulfate intensity enhancement was only observed in the OC and Fe cloud residues.

Data availability. All the data can be obtained by contacting the corresponding author.

The Supplement related to this article is available online at https://doi.org/10.5194/acp-17-8473-2017-supplement.

Competing interests. The authors declare that they have no conflict of interest.

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