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*Supplement of*

## **Investigation of processes controlling summertime gaseous elemental mercury oxidation at midlatitudinal marine, coastal, and inland sites**

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Table S1. Aqueous phase reactions and equilibria of Hg in the box model.

No.	REACTIONS	KINETIC (L mol <sup>-1</sup> s <sup>-1</sup> ) or EQUILIBRIUM CONSTANT	REFERENCE
A1	$\text{Hg}^0(\text{aq}) + \text{O}_3(\text{aq}) \rightarrow \text{HgO}(\text{aq}) + \text{O}_2(\text{aq})$	$2.4 \times 10^9$	Munthe et al., 1992
A2	$\text{Hg}^0(\text{aq}) + \text{OH}(\text{aq}) \rightarrow \text{HgOH}(\text{aq})$	$2.4 \times 10^9$	Gardfeldt et al., 2001
A3	$\text{HgOH}(\text{aq}) + \text{OH}(\text{aq}) \rightarrow \text{Hg}(\text{OH})_2(\text{aq})$	$1.0 \times 10^{10}$	Nazhat and Asmus, 1973
A4	$\text{HgOH}(\text{aq}) + \text{O}_2(\text{aq}) + \text{H}_2\text{O}(\text{aq}) \rightarrow \text{Hg}(\text{OH})_2(\text{aq}) + \text{H}^+ + \text{O}_2^-$	$1.0 \times 10^9$	Nazhat and Asmus, 1973
A5	$\text{Hg}^0(\text{aq}) + \text{OH}(\text{aq}) \rightarrow \text{Hg}^+ + \text{OH}^-$	$2.0 \times 10^9$	Lin and Pehkonen, 1997
A6	$\text{HgO}(\text{aq}) + \text{H}^+ \rightarrow \text{Hg}^{2+} + \text{OH}^-$	$1.0 \times 10^{10}$	Pleijel and Munthe, 1995
A7	$\text{HOCl}(\text{aq}) + \text{Hg}^0(\text{aq}) \rightarrow \text{Hg}^{2+} + \text{Cl}^- + \text{OH}^-$	$2.09 \times 10^6$	Lin and Pehkonen, 1997
A8	$\text{ClO}^- + \text{Hg}^0(\text{aq}) \rightarrow \text{Hg}^{2+} + \text{Cl}^- + \text{OH}^-$	$1.99 \times 10^6$	Lin and Pehkonen, 1997
A9	$\text{HgSO}_3(\text{aq}) \rightarrow \text{Hg}^0(\text{aq}) + \text{S(VI)}$	0.6	Munthe et al., 1991
A10	$\text{Hg}(\text{OH})_2(\text{aq}) \rightarrow \text{Hg}^0(\text{aq}) + \text{products}$	$3.0 \times 10^7$	Pleijel and Munthe, 1995
A11	$\text{Hg}^+ + \text{HO}_2(\text{aq}) \rightarrow \text{Hg}^0(\text{aq}) + \text{O}_2(\text{aq}) + \text{H}^+$	$1.0 \times 10^{10}$	Xie et al., 2008
A12	$\text{Hg}^{2+} + \text{HO}_2(\text{aq}) \rightarrow \text{Hg}^+ + \text{O}_2(\text{aq}) + \text{H}^+$	$1.7 \times 10^4$	Pehkonen and Lin, 1998
AE1	$\text{Hg}^{2+} + \text{SO}_3^{2-} \leftrightarrow \text{HgSO}_3(\text{aq})$	$2.0 \times 10^{13}$	van Loon et al., 2001
AE2	$\text{HgSO}_3(\text{aq}) + \text{SO}_3^{2-} \leftrightarrow \text{Hg}(\text{SO}_3)_2^{2-}$	$1.0 \times 10^{10}$	van Loon et al., 2001
AE3	$\text{Hg}^{2+} + \text{OH}^- \leftrightarrow \text{HgOH}^+$	$3.98 \times 10^{10}$	Smith and Martell, 2004
AE4	$\text{HgOH}^+ + \text{OH}^- \leftrightarrow \text{Hg}(\text{OH})_2(\text{aq})$	$1.58 \times 10^{11}$	Smith and Martell, 2004
AE5	$\text{HgOH}^+ + \text{Cl}^- \leftrightarrow \text{HgOHCl}(\text{aq})$	$2.7 \times 10^7$	Xiao, 1994
AE6	$\text{Hg}^{2+} + \text{Cl}^- \leftrightarrow \text{HgCl}^+$	$2.0 \times 10^7$	Smith and Martell, 2004
AE7	$\text{HgCl}^+ + \text{Cl}^- \leftrightarrow \text{HgCl}_2(\text{aq})$	$5.0 \times 10^6$	Smith and Martell, 2004
AE8	$\text{HgCl}_2(\text{aq}) + \text{Cl}^- \leftrightarrow \text{HgCl}_3^-$	6.7	Smith and Martell, 2004
AE9	$\text{HgCl}_3^- + \text{Cl}^- \leftrightarrow \text{HgCl}_4^{2-}$	13.0	Smith and Martell, 2004
AE10	$\text{Hg}^{2+} + \text{Br}^- \leftrightarrow \text{HgBr}^+$	$1.10 \times 10^9$	Smith and Martell, 2004
AE11	$\text{HgBr}^+ + \text{Br}^- \leftrightarrow \text{HgBr}_2(\text{aq})$	$2.50 \times 10^8$	Smith and Martell, 2004
AE12	$\text{HgBr}_2(\text{aq}) + \text{Br}^- \leftrightarrow \text{HgBr}_3^-$	$1.50 \times 10^2$	Smith and Martell, 2004
AE13	$\text{HgBr}_3^- + \text{Br}^- \leftrightarrow \text{HgBr}_4^{2-}$	23.0	Smith and Martell, 2004

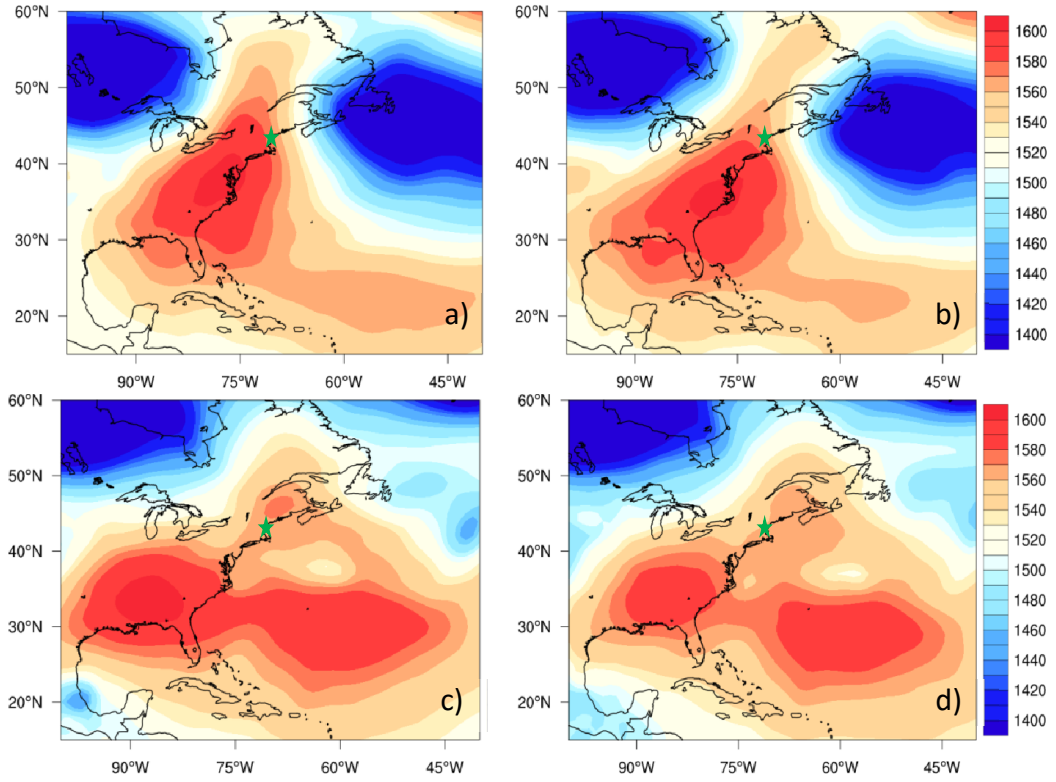


Figure S1. Geopotential height for a) 06/13/2008 08:00 EDT, b) 06/13/2008 14:00 EDT, c) 08/22/2007 14:00 EDT, and d) 08/22/2007 20:00 EDT at 850 hPa, the green star shows the location of TF site.