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Continuous measurements at the urban roadside in an Asian megacity by Aerosol Chemical Speciation Monitor (ACSM): particulate matter characteristics during fall and winter seasons in Hong Kong

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Abstract. Non-refractory submicron aerosol is characterized using an Aerosol Chemical Speciation Monitor (ACSM) in the fall and winter seasons of 2013 on the roadside in an Asian megacity environment in Hong Kong. Organic aerosol (OA), characterized by application of Positive Matrix Factorization (PMF), and sulfate are found to be dominant. Trafficrelated organic aerosol shows good correlation with other vehicle-related species, and cooking aerosol displays clear mealtime concentration maxima and association with surface winds from restaurant areas. Contributions of individual species and OA factors to high NR-PM₁ are analyzed for hourly data and daily data; while cooking emissions in OA contribute to high hourly concentrations, particularly during mealtimes, secondary organic aerosol components are responsible for episodic events and high day-to-day PM concentrations. Clean periods are either associated with precipitation, which reduces secondary OA with a lesser impact on primary organics, or clean oceanic air masses with reduced long-range transport and better dilution of local pollution. Haze events are connected with increases in contribution of secondary organic aerosol, from 30 to 50 % among total non-refractory organics, and the influence of continental air masses.

1 Introduction

The Special Administrative Region of Hong Kong is a global logistics and finance center located at the southeastern edge of the Pearl River Delta (PRD) region, China's largest manufacturing area and one of the world's most densely populated regions. Hong Kong has been plagued by deteriorating air quality, attributed to local emissions from traffic, residential and commercial activity, regional pollution from the PRD and long-range transport (Nie et al., 2013; Wong et al., 2013; Yuan et al., 2013).

High-time-resolution online instruments can characterize ambient aerosols quickly and mitigate the influence of changing environmental conditions. Few real-time studies have been conducted in Hong Kong aside from recent measurement campaigns conducted by high-resolution aerosol mass spectrometer (HR-AMS; Lee et al., 2013; Li et al., 2013, 2015; Huang et al., 2015). Long-term AMS studies tend to be costly and time-consuming due to the complexity of the instrument. The Aerosol Chemical Speciation Monitor (ACSM), whose design is based on the AMS but has been substantially simplified, has seen a growing trend of use due to its comparative ease of operation, robustness and sufficient time resolution (\sim 20–60 min) for studies spanning months or longer (Ng et al., 2011; Sun et al., 2012, 2013a, b; Budisulistiorini et al., 2013; Canonaco et al., 2013; Takahama et al., 2013; Bougiatioti et al., 2014; Petit et al., 2015; Ripoll et al., 2015; Tiitta et al., 2014; Minguillón et al., 2015).

Recently, a high-resolution aerosol mass spectrometer (HR-ToF-AMS) was applied at an urban site in the Shenzhen metropolitan area and a rural site in PRD region during October and November (He et al., 2011; Huang et al., 2011). They found that organic concentration dominates followed by sulfate, which is similar to this study, but the fraction of sulfate at the rural site is larger than that of the urban site. Four organic aerosol (OA) components were identified in urban site including hydrocarbon-like OA (HOA), biomass burning OA (BBOA), low-volatility oxygenated OA (LV-OOA) and semivolatile oxygenated OA (SV-OOA), but only three OA factors without HOA were resolved at the rural site. They both reported an important contribution from BBOA with about 24 % of total OA.

We also have previously deployed HR-ToF-AMS at the supersite of the Hong Kong University of Science and Technology (HKUST) to determine typical variations in submicron species concentrations, overall composition, size distributions, PMF-resolved organic factors and degree of oxygenation. The supersite measurements provided valuable insights into characteristics of mainly of secondary components of submicron particulate matter, with dominance of sulfate and oxygenated organic aerosol species observed (Lee et al., 2013; Li et al., 2013, 2015). Subsequent work was conducted at a downtown location (Mong Kok, MK) in Hong Kong, next to the roadside, in spring 2013 to assess important primary aerosol sources in the inner-city to identify contributions of long-range transport to roadside pollution and to establish characteristic concentration trends at different temporal scales. Cooking aerosol was identified as the dominant component in submicron non-refractory organics, followed by traffic-related emissions (Lee et al., 2015).

Differentiate from previous studies in Hong Kong, this work focuses on the characterization of roadside aerosol during the fall and winter seasons, when the influence of transported air mass is greatest and PM pollution in Hong Kong generally more severe. Episodic haze events were found to be mainly driven by secondary aerosol rather than primary emissions, while hourly high PM concentrations were often driven by cooking aerosol. Statistical methods were employed to show that the correlation of cooking organic aerosol (COA) and HOA to SV-OOA varied under different conditions and period of a day. While HOA showed a stronger relationship to SV-OOA overall, COA can be an important contributor to SV-OOA during mealtimes.

2 Experimental

The roadside measurement data collected from 3 September to 31 December 2013 in MK, an urban area with dense buildings and population in the Kowloon peninsula under the Hong Kong Environmental Protection Department (HKEPD) project (ref. 13-00986) were adopted. The sampling site was next to the roadside air quality monitoring station of HKEPD at the junction of the heavily trafficked Nathan Road and Lai Chi Kok Road (22°19′2″ N, 114°10′06″ E). The distribution of businesses in the vicinity varies, with restaurants mainly to the east, commercial buildings to the south and east, small shops for interior decoration, furniture and electrical goods to the west and residential buildings to the north of the sampling location (Lee et al., 2015). The sampling setup is described in detail in the Supplement, Sect. S1.

Non-refractory PM₁ (NR-PM₁) species (sulfate, nitrate, ammonium, chloride and organics) were measured in situ by an Aerodyne ACSM (SN: 140-154). Other data including meteorological data (wind, temperature, relative humidity, solar irradiation), volatile organic compounds (VOCs) measured by an online gas-chromatography system (GC955-611 and GC955-811, Synspec BV) and standard criteria pollutants (NO_x, SO₂ and PM_{2.5}) were provided by the HKEPD, with equipment details available from the HKEPD air quality reports (Chow, 2013).

The acquired 20-minute-average data were treated according to the general ACSM data analysis protocols established in previous studies (Ng et al., 2011; Sun et al., 2012), using the standard WaveMetrics Igor Pro based data analysis software (version 6.3.5.5) and incorporating calibrations for relative ionization efficiency, collection efficiency and detection limit. Further details on data treatment can be found in Sect. S2.

Factors contributing to organic aerosol were explored using PMF (Paatero and Tapper, 1994; Zhang et al., 2011) with the Igor Pro based PMF evaluation toolkit (PET; UIbrich et al., 2009). In general, PMF can be used to resolve factors as organic aerosol into HOA, COA, SV-OOA, LV-OOA and others. ME-2 analysis with the SOFI tool as applied in several studies may yield additional insights but has not been applied in this study due to its ongoing development (Canonaco et al., 2013; Minguillón et al., 2015). The optimal factor number was determined by inter-comparing factors' mass spectra and time series, correlations between factors and related tracers and correlations with standard mass spectra; solutions with three, four and five factors at f peak = 0 and six factors at f peak = -0.2 were explored, after which the optimal f peak value was determined by repeating the above analysis with varying f peak values.

The four-factor solution (HOA, COA, SV-OOA, LV-OOA) is optimal, with $Q/Q_{exp} = 0.8$ and better differentiation between factor time series ($R_{pr} < 0.6$; Fig. S6 in the Supplement). The factors also correlate well with associated inorganics and external tracers (NO₃, SO₄, NH₄, NO_x; Zhang et al., 2005, 2011; Ulbrich et al., 2009), e.g., HOA with NO_x, SV-OOA with NO₃, LV-OOA with SO₄ and NH₄ (Table S4 in the Supplement). Furthermore, the resolved mass spectra of four factors exhibit good similarity (all uncentered *R* (R_{uc}) > 0.80) with reference source mass spec-



Figure 1. Overview of temporal variation of (a) meteorological factors (relative humidity, temperature and precipitation) and (b) stacked plot of non-refractory PM_1 species (Org, SO₄, NO₃, NH₄ and Chl) and non-stacked plot of organic aerosol components (LV-OOA, SV-OOA, HOA and COA). Five periods – clean period 1 (C1), haze period 1 (H1), haze period 2 (H2), haze period 3 (H3) and clean period 2 (C2) – are highlighted.

tra from the AMS MS database (Ulbrich, I. M., Lechner, M., and Jimenez, J. L., AMS Spectral Database, url: http://cires.colorado.edu/jimenez-group/AMSsd; Ulbrich et al., 2009). PMF diagnostic details are shown in the Supplement (Sect. S3) and Fig. S7.

We note that m/z 60 and 73, important makers of BBOA mass spectra (Aiken et al., 2010; Cubision et al., 2011; Huang et al., 2011), were resolved not only in COA but also in SV-OOA. Their presence in SV-OOA is not the result of artifacts from the PMF analysis but was attributed to the following reasons, with more details shown in the Supplement (Sects. S4–6). Firstly, when PMF was run using only night-time data (between 00:00 and 06:00; local time: UTC+8), i.e., when there is little COA (Fig. S10), these two ions still persist with similar fractional intensities in SV-OOA as at other times. Secondly, increasing the number of PMF factors and adjusting the *f* peak value did not yield a distinct satisfactory BBOA factor. Thirdly, the time series of m/z 60 and 73 show weak correlation with other burning tracers

(EC_residual, CO_residual), with R_{pr} of about 0.2 and 0.4, respectively, but track well with SV-OOA, with R_{pr} of 0.92 and 0.93, respectively (Fig. S12, Table S9).

In terms of the possible sources of m/z 60 and 73, we observe that these two ions showed matching peaks with the COA diurnal profile and good correlations with the sum of the time series of COA and LV-OOA, with $R_{\rm pr}$ of 0.72 and 0.78, respectively. Furthermore, the ratio of the integrated signal at m/z 60 to the total signal in the organic component mass spectrum is 0.48%, which is just slightly higher than the baseline level $(0.3\% \pm 0.06\%)$ observed in environments without biomass burning influence and with secondary OA (SOA) dominance in ambient OA (Cubision et al., 2011). This indicates that these two ions at Mong Kok were mainly imbedded in cooking emissions and background aerosol due to transport rather than in a distinct source with further details shown in the Supplement (Sect. S6). The existence of m/z 60 and 73 in the emissions of Chinese cooking has been reported by He et al. (2010). Combustion of pulverized coal



Figure 2. Average concentration of each chemical composition of **(a)** NR-PM₁ (Org, SO₄, NO₃, NH₄ and Chl) and **(b)** organic aerosol (LV-OOA, SV-OOA, HOA and COA).

for barbecue or hot pot rice is a potential additional source of these two ions (Wang et al., 2013). Additionally, the existence of transported m/z 60 and 73 indicates that SV-OOA at MK is potentially influenced by transported BBOA and coal combustion aerosol.

3 Results and discussion

3.1 Mass concentration and chemical composition

Figure 1a and b display meteorological data (relative humidity, temperature and precipitation) and mass concentrations of non-refractory PM₁ (NR-PM₁) species and OA components, respectively, between September and December 2013. Total NR-PM₁ concentrations vary from 2.1 to $76.4 \,\mu g \, m^{-3}$ with an average of $25.9 \pm 13.0 \,\mu g \, m^{-3}$. ACSM NR-PM₁ concentrations co-vary with that of PM2.5 measured by TEOM $(R^2 = 0.64, \text{ slope} = 0.59; \text{ Fig. S1})$; the low slope value may be caused by the different size cuts of ACSM and TEOM and the presence of refractory materials such as elemental carbon (and to a lesser extent mineral dust and sea salt) which the ACSM cannot detect. Overall, daily PM2.5 concentrations range from 3.7 to 106.0 μ g m⁻³ and are largely (90.0 %) within the 24 h air quality standard of 75 μ g m⁻³ set by the Hong Kong Air Quality Objectives. Days with better air quality ($PM_{2.5} < 35 \,\mu g \,m^{-3}$) are mainly observed in the month of September and in rainy periods of other months. The prevailing winds from the ocean in September not only bring in less polluted air mass but also dilute the local air pollutants compared with other seasons (Yuan et al., 2006; Li et al., 2015). Precipitation has an obvious impact on total NR-PM₁ concentrations but, as we will discuss, has a lesser effect on primary organics.

Overall, NR-PM₁ is dominated by organics and sulfate with relative contributions of 58.2 and 23.3 % and average concentrations of 15.1 ± 8.1 and $6.0 \pm 3.5 \,\mu g \,m^{-3}$, respectively (Fig. 2a). Other inorganic species (ammonium, nitrate and chloride) amount to approximately 20% of NR-PM₁. The dominance of organics and sulfate is consistent with previous online studies in urban areas (e.g., Salcedo et al.,

2006; Aiken et al., 2009; Sun et al., 2012, 2013b) as well as previous filter-based studies in MK (e.g., Louie et al., 2005; Cheng et al., 2010; Huang et al., 2014). The measured composition is consistent with earlier HR-AMS measurements carried out at the same site in spring and summer 2013 (Lee et al., 2015) with very similar overall species distribution but slightly lower measured concentrations as compared to the ACSM, likely due to the fact that sampling took place in different time periods (spring-summer 2013 for the AMS campaign, fall-winter 2013 for the ACSM campaign). In the AMS study, PMF aerosol factors were identified (one additional OOA factor and one additional COA factor). A marked difference is observed in the distribution of primary OA (POA) and SOA; whereas in spring and summer (AMS) POA makes up 65 % of total organics, the reverse is observed for fall and winter (ACSM) when POA only amounts to 42 % overall. A possible reason for this discrepancy is the fact that impacts of regional pollution and long-range transport are usually higher during fall and winter (Yuan et al., 2013; Li et al., 2015), thus contributing more SOA.

Elemental carbon (EC) concentrations are significant at the Mong Kok site but not measurable by ACSM due to its high refractory temperature. EC has been discussed extensively in the previously mentioned filter-based studies and a brief comparison of online elemental carbon/organic carbon (ECOC) measurements to the results of HR-AMS measurements has been presented in an HR-AMS study (Lee et al., 2015). We therefore do not discuss EC in detail in this work.

3.2 OA Components

PMF-resolved four factors, including two primary OA factors (HOA from traffic emissions and COA) and two OOAs: LV-OOA and the less-oxidized SV-OOA (Aiken et al., 2008; Jimenez et al., 2009; Tiitta et al., 2014). The mass spectra are depicted in Fig. 3. The mass concentration of primary OA factors (HOA and COA), a surrogate of local emissions, constitutes 42 % of total organics and is slightly higher than that of LV-OOA (38 %; Fig. 2b). SV-OOA contributes approximately 20 % to total OA and is associated with both the primary organic aerosol sources and LV-OOA (see Sect. 3.2).

3.2.1 Hydrocarbon-like OA

The mass spectrum of HOA is dominated by the $C_nH_{2n-1}^+$ ion series (m/z 27, 41, 55, 69, 83, 97), typical of cycloalkanes or unsaturated hydrocarbon, which account for 27% of total peak intensity in the HOA spectrum. The other prominent group is the $C_nH_{2n+1}^+$ ion series (m/z 29, 43, 57, 71, 85, 99), typical of alkanes and accounting for 26% of the total peak. This mass spectrum is very similar to the standard HOA spectrum with R_{uc} of 0.92, and its fractions of $C_nH_{2n-1}^+$ and $C_nH_{2n+1}^+$ (27, 26%) are consistent with standard ones (= 28, 27%; Ng et al., 2011). This HOA spectrum is also consistent with that resolved by HR-ToF-AMS at the



Figure 3. Mass spectra of resolved OA components (HOA, SV-OOA, LV-OOA, COA) with the corresponding standard spectra (in gray) and the correlation with standard mass spectral profiles available on the AMS MS database (Ulbrich, I. M., Lechner, M., and Jimenez, J. L., AMS Spectral Database). The *x* and *y* axes in the right-hand graphs are mass spectra of resolved factor and the standard, respectively.

HKUST supersite on the dominance of saturated CxHy-type ions, most notably at m/z 43 and 57 (Lee et al., 2013).

HOA has an average concentration of $2.7 \pm 0.98 \,\mu g \,m^{-3}$ (Fig. 1b) and shows strong diurnal variations, including a regular decrease to about $1 \,\mu g \,m^{-3}$ during 00:00–05:00 (Fig. 4h) which is discussed in Sect. 3.3 in detail. In addition, the temporal variation of HOA displays strong correlations with NO_x ($R_{\rm pr} = 0.69$), CO ($R_{\rm pr} = 0.62$) and several VOCs (pentane, toluene, benzene) as shown in Table S10.

The diurnal patterns of vehicle numbers, HOA, NO, NO₂, NO_x and traffic-related VOCs (i-pentane, n-pentane, toluene, octane, benzene, i-butane and n-butane) are depicted in Fig. 5. Vehicle counting on Lai Chi Kok road next to the sampling site spanned 28-31 May 2013 and was provided by HKEPD (Lee et al., 2015). Although these dates are different from our campaign period, they provide a useful reference for the traffic conditions near the site. In general, more gasoline and diesel vehicles are observed during daytime than at night. The decrease of these vehicles during 22:00-04:00 is in agreement with the diurnal profile of HOA (Fig. 4h). However, liquefied petroleum gas (LPG) vehicles, which are usually taxis, show slightly higher numbers during 22:00-04:00 at the site. HOA increases sharply from $1.5 \,\mu g \,m^{-3}$ at about 06:00 to the morning peak of $3.6 \,\mu g \,m^{-3}$ at 09:00 and then persists at high concentrations until midnight, including another peak with $3.9 \,\mu g \, m^{-3}$ at 17:00. The diurnal pattern of HOA is consistent with that of NO_x (NO+NO₂), which is almost exclusively from vehicle emissions. These results are consistent with the traffic conditions at MK with heavy traffic continuously after 06:00 and rush hours from 07:00 to 11:00 and 16:00 to 19:00. NO₂ is the result of direct emission as well as formation from NO, and it increased during daytime to reach a maximum even higher than that of NO at about 17:00. Concentrations of toluene (a fuel additive) and pentane and octane (significant components in exhaust of petrol vehicles; Huang et al., 2011; Wanna et al., 2008) start to increase during the morning rush hour (07:00) and peak between 18:00 and 19:00. HOA and NO_x show a distinct morning peak at $\sim 08:00$ when a small shoulder is also found in the VOCs. Butane, a constituent of LPG, displays a diurnal pattern different from that of HOA, with higher concentrations between 22:00 and 04:00; LPG-fueled taxis are a major means of transport during the nighttime and early morning, and fuel leakage during refueling may contribute to the observed pattern. Furthermore, fuel leakage during refueling of LPG vehicles may contribute more than diesel-fueled vehicular emissions to butane even though the number of diesel fueled vehicles is slightly higher than LPG ones at that time. At last, the sampling site is near a major junction serving a number of district centers (West Kowloon, Sha Tin, Tsim Sha Tsui) and is therefore frequented by taxis.

3.2.2 Cooking-related OA

The most prominent ions of the resolved COA profile at MK were $m/z \,41$ (mainly C₂HO⁺, C₃H₅⁺) and $m/z \,55$

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Figure 4. Diurnal profiles of NR-PM₁ species, OA components and temperature for the entire study with 25th and 75th percentile boxes, 10th and 90th percentile whiskers, mean as colored marker and median as black line in the whisker box.

(mainly C₃H₃O⁺, C₄H₇⁺). Ratios of m/z 41/43 = 1.8 and m/z 55/57 = 2.2, which are distinctly larger than that of HOA at 0.73 and 0.76, respectively (Fig. 4); such ratios have been widely reported for COA in AMS and ACSM studies. For example, Lanz et al. (2010) reported ratios of m/z 41/43 and m/z 55/57 of 0.5 and 0.4 in HOA and 1.2 and 1.2 in COA, respectively, while Sun et al. (2013a) reported 0.5 for these two ratios in HOA and 2.3 for those in COA, respectively.

Figure 6a shows COA concentrations sorted by wind direction in MK. The COA concentration reaches up to $12 \,\mu g \,m^{-3}$, contributing ~ 60 % of total organics, when easterly winds dominate, probably due to the large number of restaurants located on the eastern side of the sampling site (Fig. 6a). In general, COA contributes significantly to the total mass of organic aerosol with an average fraction of 24 % (3.7 $\mu g \,m^{-3}$), in line with the 16–30 % COA contributions found in several cities including London, Manchester, Barcelona, Beijing, Fresno and New York (Allen et al., 2010; Huang et al., 2010; Sun et al., 2013b; Mohr et al., 2012; Ge et al., 2012). Figure 6b and c compare the chemical



Figure 5. Diurnal patterns of vehicle numbers at the Mong Kok site during 28-31 May 2013 and concentrations of HOA, NO_x, NO₂, NO, i-pentane, n-pentane, i-octane, i-butane, n-butane, benzene and toluene during the whole study.

composition of NR-PM₁ and OA during mealtimes (lunch, 12:00-02:00, and dinner, 19:00-21:00) and non-mealtimes (00:00-06:00); the non-meal period is defined by the periods of low concentration ($<2^{\circ} \mu g m^{-3}$) in the COA diurnal pattern. During dinner time, the average concentration of organics increases by about $11 \,\mu g \, m^{-3}$ and its contributions in total NR-PM1 increase to 70%, while the concentrations of other species do not change much (Fig. 6b). As shown in Fig. 6c, the increase in organic concentrations results from the increase in COA from 1.7 to $7.8\,\mu g\,m^{-3}$ (~360% increase), and to a lesser extent increases in SV-OOA (from 1.5 to $4.5 \,\mu g \, m^{-3}$, a $\sim 200 \,\%$ increase) and in HOA (from 1.4 to $3.2\,\mu g\,m^{-3}$, a $\sim 130\,\%$ increase). As shown in Table 1, the average concentration of organics during dinner time is $5 \,\mu g \, m^{-3}$ higher than that during lunch, and this increase is attributed to the increase of COA and SV-OOA mass but not of HOA. This is consistent with the expectation that the cooking activities at MK are higher during dinner than during lunch, while traffic during dinner is comparable to or smaller than that during lunch (Fig. 4f and h). The increase of SV-OOA during dinner time may be the result of enhanced cooking emissions and possibly less evaporation due to lower ambient temperature; contributions from traffic emissions are not likely to be important since there is little increase of HOA during mealtime.

3.2.3 Oxygenated OA

LV-OOA is characterized by the prominent m/z 44 ion (mainly CO_2^+) and minor C_nH_{2n-1} and C_nH_{2n+1} ion series generated by saturated alkanes, alkenes and cycloalkanes.



Figure 6. (a) Wind rose plot of COA concentration. The angle and radius present the wind direction and its probability, respectively, while color indicates COA concentration. (b) The fractional composition of NR-PM₁ species during mealtime (12:00-02:00, 19:00-21:00) and non-mealtime (00:00-06:00). (c) The fractional composition of OA during mealtime and non-mealtime.

Species $\mu g m^{-3}$	Lunch	Dinner	Non-meal					
Org	18.8	23.7	10.3					
SO_4	5.8	6.1	6.3					
NH ₄	2.6	2.9	3.0					
NO ₃	1.4	1.8	1.6					
Chl	0.1	0.2	0.2					
Organic aerosol components								
HOA	3.2	3.2	1.4					
COA	6.2	9.6	1.7					
LV-OOA	5.8	5.4	5.6					
SV-OOA	3.6	5.5	1.5					

Table 1. Average concentrations of NR-PM1 and OA components
during lunch time, dinner time and non-mealtimes.Ox concentration
60SpeciesLunchDinnerNon-meal $\mu g m^{-3}$ 2015



The LV-OOA spectrum correlates well with the standard LV-OOA spectrum (Fig. 3), with a $R_{\rm uc}$ of 0.97. The LV-OOA time series is associated with that of SO_4^{2-} with a $R_{\rm pr}$ of 0.86 (Fig. 1), consistent with reports in the literature (DeCarlo et al., 2010; He et al., 2011; Zhang et al., 2014; Tiitta et al., 2014). The LV-OOA diurnal pattern varies little, suggesting that it is part of the background aerosol, possibly resulting from long-range transport (Li et al., 2013, 2015).

SV-OOA, which is less oxidized than LV-OOA, is marked by the dominant ions of m/z 43 and m/z 44 mainly contributed by C₂H₃O⁺ and CO₂⁺. The mass spectrum of SV-OOA closely resembles that of "standard" SV-OOA with a R_{uc} of 0.87 (Fig. 3). Some marker fragments of COA and HOA, for example m/z 41, 43, 55 and 57, are presented in the SV-OOA mass spectrum. SV-OOA concentrations are also

Figure 7. Variation of the average concentration of OA components (HOA, SV-OOA, LV-OOA and COA) coded by color as a function of binned O_x concentration (ppb) and binned temperature (°C).

weakly associated with those of HOA and their co-emitted precursors (benzene and toluene), with $R_{\rm pr}$ of 0.58, 0.65 and 0.51, respectively. In fact, the correlation between SV-OOA and benzene is better than that of HOA and benzene (0.56). The diurnal pattern of SV-OOA also shows peaks at mealtimes like COA. Lastly, the fraction of signal at m/z 44 (f44 fraction) of SV-OOA at MK is twice that of the standard measured by Q-AMS (Zhang et al., 2014; Tiitta et al., 2014). Together, these results suggest that SV-OOA may be correlated with POA (HOA and COA), possibly due to rapid oxidation of POA to semivolatile gases, which may then form

Period	Mealtime (MT)		Backgrou	and time (BT)	Other time (OT)					
Temperature	LTemp	HTemp	LTemp HTemp		LTemp	HTemp				
Coefficients ^a										
HOA	0.80	0.56	0.70	0.43	0.48	0.23				
COA	0.29	0.15	0.22	0.00	0.31	0.11				
LV-OOA	0.25	0.23	0.23	0.24	0.25	0.28				
Adjusted R^2	0.90	0.81	0.83	0.57	0.85	0.73				
Average concentration ($\mu g m^{-3}$, ppb)										
HOA	3.71	2.85	1.60	1.18	3.51	2.88				
COA	7.34	7.40	1.61	1.54	3.50	3.74				
LV-OOA	5.46	5.57	5.91	5.07	5.85	5.99				
SV-OOA	6.30	3.89	2.68	1.44	4.1	2.39				
O_x (ppb)	83.12	85.23	58.71	53.45	75.06	76.77				

Table 2. Regression of SV-OOA on HOA, COA and LV-OOA and concentrations of OA factors and O_x under high and low temperature (LT and HT) of the three chosen periods (MT, BT and OT).

^a The coefficient of HOA, COA and LV-OOA in the regression equation reconstructing SV-OOA under LTemp (T < 22.5 °C) and HTemp (T > 22.5 °C) during mealtime (12:00–14:00, 19:00–21:00), background time (00:00–06:00) and other time. The average temperature of the whole campaign is 22.5 °C. All entries of coefficients are significant at the 1 % level (two-tailed) except that of HOA/OT, which is significant at the 5 % level.

SV-OOA. However, the variation of the average concentration of SV-OOA as a function of binned LV-OOA concentration in increments and a bin width of $2 \,\mu g \,m^{-3}$ is shown in Fig. S13. The linear, positive relationship between SV-OOA and LV-OOA suggests that non-local formation and subsequent transport may also contribute to the measured SV-OOA at MK. However, it should be mentioned that ACSMresolved organic spectra have been observed to show higher f44 in other studies (Crenn et al., 2015; Frohlich et al., 2015) compared to HR-ToF-AMS measurements due to inherent instrumental uncertainties in the determination of f44. This might have caused the elevated f44 observed in our SV-OOA spectrum.

Figure 7 displays the concentration of different OA factors (coded by color) as a function of binned O_x concentration (ppb) and binned temperature (°C) with a bin width of 15 ppb and 5 °C, respectively. In general, the concentration of all OA factors increases as O_x increases across all temperatures. While it is understood that LV-OOA and SV-OOA are correlated with O_x because they all result from similar photochemical activities, the correlation between HOA and O_x is the result of the good correlation (0.78) between HOA and NO₂, which accounts for 84 % of total O_x . NO₂ is partly emitted directly from vehicles and partly formed by secondary oxidation at MK as discussed in Sect. 3.2.1. Increase in ambient temperature is associated with decrease in HOA and SV-OOA, likely due to evaporation effects and partitioning, but it has no obvious correlations with LV-OOA and COA.

To further assess the relative importance of other OA factors to the resolved SV-OOA, ordinary least squares regressions were conducted. Considering the potential influence of primary OA on the regression results, the whole data set was separated into three time periods consisting of mealtime (MT; 12:00–14:00 and 19:00–21:00) marked by enhanced COA, background time (BT; 00:00–06:00) marked by low POA and other time (OT; 06:00–12:00, 14:00–19:00 and 21:00–24:00). The data of each period were further divided into high/low temperature (HTemp, LTemp = T < 22.5 °C) and high/low O_x (HiO_x, LO_x = O_x < 70 ppb) to reveal impacts of temperature and the degree of oxygenation on the correlations among OA factors.

Tables 2 and 3 show the coefficients of HOA, COA and LV-OOA in the regression equation for the reconstructed SV-OOA and their average concentrations during different periods under high/low temperature and high/low O_x , respectively. The average concentrations of HOA and SV-OOA under HTemp are obviously lower than under LTemp for each period but the concentration of COA and LV-OOA varies little across different temperatures (Table 2). By combining the stronger correlations between HOA and SV-OOA rather than between COA and SV-OOA, a stronger and closer temperature dependence of HOA and SV-OOA was revealed. In addition, the regression coefficients of HOA and COA during each period under HTemp are much smaller than under LTemp, reflecting a weakening of their relationship with SV-OOA as temperature increases.

Consistent with the discussion of Fig. 7, the concentrations of HOA, SV-OOA and LV-OOA except for COA under HiOx are greatly higher than those under LOx for each period (Table 3). Besides, HOA shows an increase correlation with SV-OOA under HiOx due to the more intensive oxidation of HOA precursor to SV-OOA. However, LV-OOA shows a reverse trend with smaller coefficients with SV-OOA. It is



Figure 8. (a) Day-of-week variations of NR-PM₁ species (standard deviation as vertical line) and (b) average diurnal patterns of OA components for weekdays, Saturdays and Sundays.

probable that HiOx conditions favor the conversion of SV-OOA to LV-OOA leading to smaller coefficient of LV-OOA on SV-OOA, although overall most LV-OOA is considered to be from transport.

At last, we also can conclude that HOA overall has a stronger relationship to SV-OOA than COA has, supported by much higher coefficients of HOA than that of COA over all time periods, and temperature and O_x levels. Cooking emissions are not as important to SV-OOA in the BT periods but they can be important during MT periods, indicated by the lowest concentration and correlation with SV-OOA during BT but highest concentration during MT periods.

3.3 Diurnal patterns

The diurnal profiles of NR-PM₁ species and OA components are depicted in Fig. 4. Total organics display a diurnal pattern with two pronounced peaks during 12:00–14:00 and 19:00–21:00, corresponding to the peaks of COA at lunch and dinner time, respectively. In addition, organics increase at about 10:00, which may be related to the increase of local emissions of HOA and COA by 2.3 and 1.1 μ g m⁻³, respectively, from 06:00 to 10:00.

The mass concentration of sulfate (Fig. 4b) does not show any diel variation. It is likely that sulfate, as a regional pollutant, is mainly formed during long-range transport, leading to the lack of a specific diurnal pattern at MK; a similar flat diurnal pattern for sulfate has also been found at the HKUST supersite in Hong Kong (Lee et al., 2013; Li et al., 2015). These results differ significantly from observations in Beijing and Lanzhou in China and Welgegund in South Africa (Sun et al., 2012, 2013b; Xu et al., 2014; Tiitta et al., 2014) where sulfate displays an obvious increase at noontime in summer and wet seasons due to either photochemical reaction or aqueous oxidation of SO₂. The difference may result from the much lower level of sulfur dioxide (SO₂) with an average of 4.6 ppb in MK compared to, for example, ~ 32 ppb in Beijing, where coal combustion leads to a much higher SO₂ concentration (Lin et al., 2011); sulfate and relative humidity (RH) have almost no correlation ($R^2 = 0.06$) in MK, suggesting little importance of local aqueous processing for the formation of sulfate.

Nitrate shows a slight dip around noontime, corresponding to the increase of the ambient temperature (Fig. 4j); evaporative loss of particulate nitrate might outweigh the secondary production of nitrate during this time. The diurnal pattern of ammonium (Fig. 4d) is very similar to that of sulfate, as expected based on their commonly observed association in atmospheric particles. Chloride (Fig. 4e) has rather low concentrations and shows a similar diurnal variation to that of nitrate, likely due to its volatility.

3.4 Day-of-week patterns

Figure 8a shows the average concentration trends on individual days of the week for NR-PM1 species. Figure 8b describes the diurnal patterns of the OA components for weekdays, Saturdays and Sundays, respectively. Because of the small data sets on Saturdays and Sundays, data beyond 1 standard deviation from the mean $(25.9 \pm 13.0 \,\mu g \, m^{-3})$ were removed from the whole data set to remove the influence of episodic events in this analysis. Overall, total NR-PM1 concentrations have no obvious variation (average variation less than 5%) from Monday to Saturday, but they drop by 16% on Sundays compared to Saturdays. This weekend difference is opposite to the result found in Beijing where higher concentrations occurred on Sundays than Saturdays (Sun et al., 2013b). However, some others such as Lough et al. (2006) and Rattigan et al. (2010) reported that both Saturdays and Sundays had obvious traffic emissions reduction due to less human activities on weekends in Los Angeles and New York, respectively.

Organics and secondary inorganics (SO₄, NH₄ and NO₃) contributed 54 and 46%, respectively, to the concentration difference between Sundays and Saturdays in MK. The difference in organics is mainly attributed to the variation of HOA, which shows very similar diurnal variations on Saturdays and weekdays, but has an average decrease of 23% after 07:00 on Sundays. A 37% reduction of traffic-related carbonaceous aerosol on Sundays compared with weekdays in MK has been reported (Huang et al., 2014). In Hong Kong many people work on Saturday, which leads to a traffic pattern similar to normal weekdays. COA shows nearly the same diurnal patterns on all days, and LV-OOA and SV-OOA do not show obvious variations. Overall, local emissions from traffic contribute most to the day-of-week variations in organics.

	Mealtin	ne (MT)	Backgr	ound time (BT)	Other time (OT)				
	LO_X	HiO_X	LO_x	HiO_X	LO _x	HiO _x			
Coefficients ^a									
HOA	0.50	1.13	0.62	0.64 ^b	0.08 ^b	0.52			
COA	0.13	0.14	0.00	0.15	0.14	0.14			
LV-OOA	0.33	0.10 ^b	0.26	0.18	0.34	0.21			
Adjusted R^2	0.73	0.86	0.73	0.80	0.67	0.78			
Average concentration ($\mu g m^{-3}$)									
HOA	2.24	3.41	1.20	2.03	2.11	3.55			
COA	7.31	7.57	1.59	1.73	2.77	3.71			
LV-OOA	3.50	5.92	5.07	7.22	4.06	6.77			
SV-OOA	3.22	5.29	1.85	2.79	1.8	3.56			
Temp (°C)	23.30	23.80	21.48	20.39	22.01	22.74			

Table 3. Regression of SV-OOA on HOA, COA and LV-OOA and concentrations of OA factors and temperature under high and low O_x (HiO_x and LO_x) of four chosen periods (MT, BT and OT).

^a The coefficient of HOA, COA and LV-OOA in the regression equation reconstructing SV-OOA under LO_x (O_x < 70 ppb) and HT (O_x > 70 ppb) during mealtime (12:00–14:00, 19:00–21:00), background time (00:00–6:00) and other time. The average O_x of the whole study is 70 ppb. All entries of coefficients are significant at 1 % level (two-level) except those indicated with ^b, which indicates significance at the 5 % level.



Figure 9. (a) Variations in mass concentration of NR-PM₁ species and OA components as a function of total NR-PM₁ mass loading, and (b) mass fraction of total NR-PM₁ for NR-PM₁ species as a function of total NR-PM₁ mass loading, and (c) mass fraction of total organics for OA components as a function of total NR-PM₁ mass loading. All the mass concentrations and fractions of above species were sorted according to the hourly average NR-PM₁ mass in ascending order. The solid circles represent the average value for each concentration bin with a width of $7 \,\mu g \,m^{-3}$, and the vertical lines represent the standard deviations.

3.5 Contributions of individual species and OA factors to high NR-PM₁

Figure 9a, b and c show the variation in hourly mass concentration of NR-PM1 species and OA components and their mass fractions as a function of hourly total NR-PM1 mass loading, respectively. Below 50 µg m⁻³, all aerosol species display a nearly linear increase with PM₁ mass loading, with slopes of about 0.5 for organics, 0.25 for sulfate and LV-OOA and around 0.1 for nitrate, ammonium, COA, HOA and SV-OOA (Fig. 9a). While the fractions of NH₄ and organics remain relatively stable, sulfate exhibits a decrease and then an increase, and NO₃ and chloride show a gradual increase and then a decrease, respectively, as NR-PM1 increased to $50 \,\mu g \,m^{-3}$ (Fig. 9b, c). Although the mass concentrations of all organic factors increase as NR-PM1 increases, SV-OOA is the only factor that increased in mass fraction. Primary OA components (HOA and COA) and transported OA (LV-OOA) show a decrease in fraction and stable contributions, respectively, as NR-PM₁ increases to $50 \,\mu g \,m^{-3}$, while the contribution of SV-OOA increases sharply from around 5 to 25 % of total organic mass. It suggests that SV-OOA plays an important role as NR-PM₁ increases to $50 \,\mu g \,m^{-3}$ in MK. However, beyond $50 \,\mu g \,m^{-3}$, the mass loadings of SO₄ and organics increase, while those of NH₄, NO₃ and LV-OOA remain almost constant, which differs from the observations in Beijing, where NH₄ and NO₃ kept a linear increase from 50 to about $200 \,\mu g \,m^{-3}$ (Sun et al., 2013b; Zhang et al., 2014). In terms of fractions, only COA and, to a lesser extent, SV-OOA increase as NR-PM1 increases further. In fact, over 80% of the high hourly NR-PM₁ concentrations (>50 μ g m⁻³) are



Figure 10. (a) Variation in mass concentration of NR-PM₁ species and OA components as a function of total NR-PM₁ mass loading, and (b) mass fraction of total NR-PM₁ for NR-PM₁ species as a function of total NR-PM₁ mass loading, and (c) mass fraction of total organics for OA components as a function of total NR-PM₁ mass loading. All the mass concentrations and fractions of above species were sorted according to the daily average NR-PM₁ mass in ascending order. The solid circles represent the average values for each concentration bin with a width of $7 \,\mu g \,m^{-3}$, and the vertical lines represent the standard deviations.

observed during the mealtime periods with enhanced cooking activities.

When the hourly averages in Fig. 9 are replaced by daily averages (Fig. 10), the COA concentration varies little and its fraction does not exhibit an increase but instead decreases significantly with increasing daily NR-PM₁. Contrarily, the fractions of SV-OOA and LV-OOA clearly increase. This analysis suggests that while cooking OA is responsible for the hourly high concentrations during mealtime and potential high hourly PM levels, LV-OOA/SV-OOA is responsible for episodic events and high day-to-day PM levels.

To analyze the difference in particle composition and meteorological conditions among episodic periods and clean periods, three heavy polluted episodes (19–22, 23–26 October and 10–13 December) and two clean periods (17–18 September and 14–18 December), highlighted in Fig. 1, were analyzed. The average concentrations of these chosen periods are larger than 1 standard deviation from the average concentration of the campaign ($25.9 \pm 13.0 \,\mu\text{g m}^{-3}$). The composition, meteorological features (*T* and RH) and oxidation index (O_x and *f* 44) of these five events are shown in Table 4. Clean period 1 (C1) is characterized by low NR-PM₁ concentrations (below 13 μ g m⁻³), prevailing coastal wind (easterly wind), lack of rain, high ambient temperature (~ 28 °C) and high relative humidity (~ 70 %). Another clean period (C2) features continuous precipitation with the coldest and



Figure 11. Mass fraction of hydrocarbon-like organic aerosol (HOA), cooking organic aerosol (COA), semi-volatile oxygenated organic aerosol (SV-OOA) and low-volatility oxygenated organic aerosol (LV-OOA) in color, and the mass concentration of POA and SOA marked by triangles and circles, respectively, during five periods: clean periods (C1 and C2) and haze periods (H1, H2 and H3).

most humid weather condition in the period studied. Haze period 1 (H1) has similar temperature and humidity as C1 but is marked by mixed continental/oceanic winds. From H1 to the following haze period (H2), the observed wind direction shifts to reflect continental transport, with a significant decrease in RH to 36 %. Haze period 3 (H3), just before C2, is also dominated by continental winds but with lower temperatures (~ 19 °C) than during other haze events.

Although the total NR-PM₁ of C1 ($12.2 \,\mu g \, m^{-3}$) and C2 $(11.8 \,\mu g \, m^{-3})$ are both only 25–30% of that during haze periods, they were driven by different mechanisms. The main differences in meteorological conditions between C1 and C2 are the dominance of continental wind rather than coastal wind, much lower temperature and the existence of precipitation in C2. The low concentration of C1 is mainly attributed to easterly wind bringing less air pollutants and diluting local air pollutants. To a lesser extent, it is influenced by both particle evaporation, especially for SV-OOA, and dilution of local emissions during high temperatures, which might be the reason why HOA, COA and SV-OOA in C1 are lower than in C2 despite the lack of rain. The low mass loading of C2 was mainly caused by the wet deposition of precipitation. It dramatically reduces the concentration of secondary species, such as SO₄, NH₄, NO₃, SV-OOA and LV-OOA, but not primary HOA and COA. Compared to the adjacent period H3, the total organic mass reduces by 68% to an average of $8.1 \,\mu\text{g m}^{-3}$ (Table 4). Precipitation effectively removes secondary particles but is less efficient for primary particles that are continuously generated locally.

With similar continental source region as C2, the most severe pollution event H3 occurred during 10–13 December with an average NR-PM₁ of 47.7 μ g m⁻³. The persistent northerly wind continually brought air masses from the PRD region into Hong Kong and lead to a marked mass increase of secondary species of SO₄, NH₄, NO3, LV-OOA and SV-

	Clean p (Cl	eriod 1	Haze period 1 (H1)		Haze period 2 (H2)		Haze period 3 (H3)		Clean period 2 (C2)	
RH (%)	70.8		65.0		36.4		64.8		84.6	
<i>T</i> (°C)	27.6		25.0		23.8		18.7		13.2	
O_x (ppb)	69.6		82.0		99.5		70.4		40.9	
f44	0.114		0.118		0.120		0.108		0.057	
Precip.(mm)	0)	0		0		0		8.9	
Wind	coas	stal	continental/oceanic		continental		continental		continental	
(µg m ⁻³ %)	Conc.	Perc.	Conc.	Perc.	Conc.	Perc.	Conc.	Perc.	Conc.	Perc.
NR-PM ₁	12.2		44.1		39.0		47.7		11.6	
Org	6.7	54.4	25.2	57.2	21.1	54.2	25.1	52.6	8.1	69.6
SO_4	3.8	31.2	11.8	26.8	12.1	30.9	11.4	23.8	1.5	12.8
NH ₄	1.2	9.9	4.4	10.1	4.4	11.3	6.5	13.6	1.1	9.4
NO ₃	0.4	3.5	2.4	5.6	1.3	3.4	4.4	9.2	0.8	7.3
Chl	0.1	1.0	0.2	0.4	0.1	0.2	0.4	0.8	0.1	0.9
HOA	1.2	18.5	3.8	15.1	3.0	14.4	4.2	16.9	2.1	26.2
COA	2.3	34.8	3.7	14.5	3.3	15.5	3.3	13.1	2.6	31.7
LV-OOA	3.0	44.8	11.5	45.4	10.2	48.4	9.9	39.6	1.8	22.0
SV-OOA	0.1	2.0	6.3	25.0	4.5	21.6	7.6	30.4	1.6	20.1

Table 4. Measured and calculated parameters in five chosen periods (C1, H1, H2, H3 and C2).

^a Average of data from clean days (C1 and C2) and hazy days (H1, H2 and H3). C1: 17–18 September; H1: 19–22 October; H2: 23–26 October; H3: 10–13 December; C2: 14–18 December. T: temperature; RH: relative humidity; O_x : odd oxygen ($O_3 + NO_2$) in ppbv; f44: fraction of m/z 44 in organic mass spectra.

OOA. Furthermore, H3 is characterized by the highest mass concentration and relative contribution of nitrate and SV-OOA compared with other haze periods. This is likely due to the average temperature of H3 being 5–6 °C lower than that of other haze events. In addition, although all three haze events have very similar SO₄ mass loading, there is a \sim 50 % increase in NH₄ concentration during the H3 episode, consistent with the increase of nitrate in that period.

The other two haze events are adjacent with influence from both continental and oceanic region in H1 and continental source region in H2. The mixed pattern of source regions during H1 identified as land-sea breeze (Fig. S14) can redistribute PM pollution over the whole PRD region and accumulate air pollutants effectively (Lo et al., 2006; Chan and Yao, 2008; Lee et al., 2013). The pronounced high concentration of LV-OOA and SV-OOA, jointly contributing 70% of total organics, reflects the oxidation of primary emissions in the PRD under such cycles, which is also observed at the suburban HKUST site (Lee et al., 2013). The periodic nitrate peaks in H1 with low concentration in daytime and high concentration in nighttime coincide with temperature changes. During H2 period, the prevailing wind is northwesterly and there is a sharp decrease in relative humidity. It is interesting to note that the dip in RH during H2 coincides with the dip in sulfate, ammonium, nitrate and LV-OOA; this might be caused by decreased aqueous-phase processing and by decreased gasparticle partitioning associated with water uptake under low RH for secondary aerosol particles (Sun et al., 2013a, b).

The fractions of f 44 during these three haze occasions are all lower than that at HKUST (Li et al., 2013), which reflects a larger abundance of the less oxygenated POA at the urban MK site. In addition, the POA concentration (HOA + COA) does not change much between clean periods and haze periods. However, its relative contribution decreases from about 50 % during clean periods to 30 % during haze events because of the pronounced variation of secondary OA as shown in Fig. 11.

4 Conclusions

The characteristics and sources of ambient submicron nonrefractory particulate matter (NR-PM1) were investigated in an urban roadside environment in Hong Kong using an Aerodyne ACSM from September to December, 2013; these are the first ACSM measurements in Hong Kong. Organics and sulfate dominate total NR-PM₁, making up more than 50 and 20% of measured mass concentration, respectively. PMF analysis of organic aerosol mass spectra yielded four characteristic OA factors: HOA, COA, SV-OOA and LV-OOA. Primary OA factors (HOA and COA) from freshly emission contribute 43% of total organics, slightly larger than that of LV-OOA, which is generally transported pollutant in this study, with about 37 % of total organics. SV-OOA contributes about 20% of total organics and is variably correlated with HOA, COA and LV-OOA under different conditions and period of a day. While HOA showed a stronger relationship to

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SV-OOA overall, COA can be an important contributor to SV-OOA during mealtimes. In addition, the transported pollutants reflected by LV-OOA displays a relatively stable correlation with SV-OOA during the different periods (BT, MT and OT).

The mass loadings of traffic-related aerosol (HOA) are consistent with expected traffic count data and correlate well with various vehicle-related VOCs and NO_x . Furthermore, HOA, with an average decrease of 23 % after 07:00 on Sundays, contributes most to the lower organic concentrations on Sundays when compared with other days. Cooking aerosol displays a well-defined diurnal variation with lunch- and dinnertime peaks and contributes on average 40 % of total organics during mealtimes; COA is clearly associated with local easterly winds, which coincides with the placement of nearby restaurant.

The contributions of individual species and OA factors to high NR-PM₁ were analyzed based on hourly data and daily data. It suggests that while cooking is responsible for the hourly high concentrations during mealtimes, LV-OOA/SV-OOA are responsible for episodic events and high daily PM concentration. Three heavily polluted episodes and two clean periods were recorded during sampling and attributed to different meteorological and circulatory conditions. The analysis of clean periods shows that precipitation has an obvious deposition impact on total NR-PM₁ concentrations, but it has a lesser effect on primary organics. Clean ocean wind not only brings in less polluted air mass but also dilutes the local air pollutants. During this campaign, high-PM events are generally related to continental air mass influence or landsee breeze circulatory conditions, which has less influence on primary emissions but significant effects on secondary particles, with a pronounced increase in the secondary OA contribution during haze events (from 30 to 50%).

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