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Supplement of

Evidence for an unidentified non-photochemical ground-level source of formaldehyde in the Po Valley with potential implications for ozone production

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1 **Additional consideration of deposition**

2 In the base case scenario, the model is constrained to all measurements with the exception of
3 HCHO. Given the extensive constraints, deposition, emission, and advection are not treated
4 explicitly. While such simplification is commonly performed in box models, the coupling
5 between the surface layer and the overlying atmosphere can drive the behavior of many trace
6 gases while chemistry plays a smaller role. The assumption is that with small enough time steps,
7 constraining calculations to observed mixing ratios is functionally equivalent to having an inflow
8 or outflow rate that accounts for net flux from the surface or emission sources. However, the
9 impact on species that are not constrained could be a function of what the model sources and
10 sinks of a given other species. For example, if an unconstrained HCHO precursor has a high
11 deposition rate which is neglected, the model may overestimate loss to oxidation and create high
12 concentrations of unconstrained oxidation products. This could lead to HCHO production, while
13 deposition would not lead to HCHO formation. Furthermore, reaction with OH would be over-
14 expressed, leading to erroneously high calculated OH reactivity.

15 To investigate the effect of neglecting model deposition, a model scenario which includes dry
16 deposition of O₃, HNO₃, NO_x, H₂O₂, CH₃OOH, peroxyacetyl nitrate, CH₃CHO, HCHO,
17 CH₃CO₃H, HCOOH, isoprene, methyl vinyl ketone, methacrolein, methylacetate, acetic acid,
18 and methanol was constructed. Deposition velocities were calculated using the SCM without
19 imposing any constraint on the soil moisture level. The deposition rates are generally based on
20 species solubility and reactivity according to Wesely (1989), except for species which have been
21 modified according to Ganzeveld and Lelieveld (1995) (O₃, HNO₃, NO_x), and peroxides
22 (Ganzeveld et al., 2006). The base case model domain was extended down to the ground (rather
23 than stopping at 100 m), and a first order deposition loss rate was applied to the surface layer. As
24 observations were not available at these altitudes, the observed averages between 50-150 m are
25 applied to the surface layer. All species except HCHO are constrained to measurements.

26 The results of this model run are shown in Fig. S1. Including deposition has negligible effects on
27 calculated OH reactivity; however, calculated HCHO decreases significantly in the boundary
28 layer, creating an inverted profile. To compensate for this potential depositional sink, an
29 emission rate eight times greater than that used in the main text would be required. This suggests
30 either net deposition and emission occur on separate surfaces, or the existence of a HCHO

1 compensation point at which the surface as a whole acts as a net source rather than a sink.
2 Because calculated OH reactivity is largely unaffected by including deposition and because our
3 base model scenario represents an upward bound on the calculated HCHO mixing ratio, the
4 conclusions stated in the main text are unaffected by the deposition term. Our study represents a
5 lower limit on the non-photochemical HCHO source, and therefore a lower limit on the
6 sensitivity of ozone production to agricultural emissions of HCHO.

7

8 **Model sensitivity to turbulent mixing**

9 The eddy-diffusivities for tracer transport are taken from the SCM which simulates online the
10 turbulent fluxes of momentum, heat and moisture transport according to the local-closure
11 turbulent transport scheme of the climate model ECHAM4 including modifications for non-local
12 contributions to the turbulent kinetic energy budget (Brinkop and Roeckner, 1995) and updated
13 mixing length formulation (Lenderink and Holtslag, 2004). The simulated meteorology
14 resembles the actual observed meteorological conditions quite well. The SCM simulations are
15 constrained with the ECMWF (European Centre for Medium range Weather forecast) re-analysis
16 data for the grid point resembling the location of the SPC site covering (e.g. Ganzeveld et al.,
17 2006). The eddy diffusion constants are shown in context of the HCHO lifetime with respect to
18 photolysis/reaction with OH in Fig. S2.

19 As seen in the case where direct HCHO emission is included in the model, the eddy diffusion
20 constants may not accurately capture the speed of growth of the mixed layer. As a sensitivity
21 analysis, the coefficients are increased uniformly by a factor of 10 in both the base case scenario
22 and the scenario where emissions are included (Fig. S3). While faster mixing better captures the
23 early growth of the mixed layer in the emissions included case, the effect on the base case is
24 minimal. Therefore, we conclude that the uncertainties in eddy diffusion have minimal impact on
25 the conclusion that HCHO is underestimated at the lowest altitudes while OH reactivity is well
26 captured.

27 Finally, as a boundary condition, no mixing occurs between the uppermost model layer and the
28 overlying atmosphere (i.e. all boundaries are reflective). Any mixing would only affect non-
29 constrained species. It is probable that the concentrations of these species (e.g. RO₂, HCHO) is
30 lower in the overlying layer and additional mixing would further dilute HCHO. Because our base

1 case model represents an upper estimate of model HCHO, i.e., a conservative estimate of missing
2 HCHO, including mixing with the upper atmosphere would not affect our conclusions.
3 Additionally, the reflective boundary conditions restrict emitted species to the total model
4 volume. This could artificially inflate the concentrations of emitted species if the actual
5 atmospheric mixing volume is greater than the model volume. According to the eddy diffusion
6 coefficients, mixing significantly decreases at highest model altitude. Therefore, it is unlikely
7 that the reflective boundary significantly inflates the increase in HCHO mixing ratios caused by
8 emissions in the early morning.

9

10 **References**

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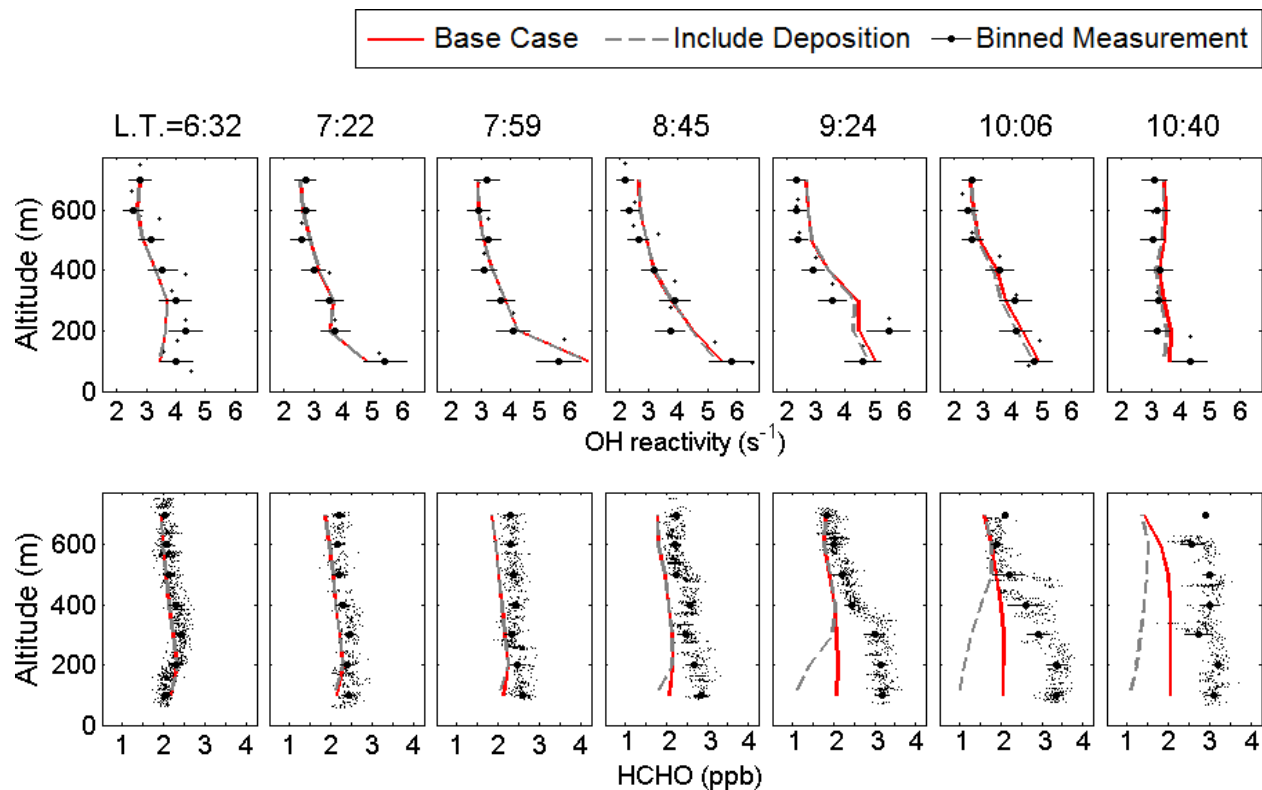
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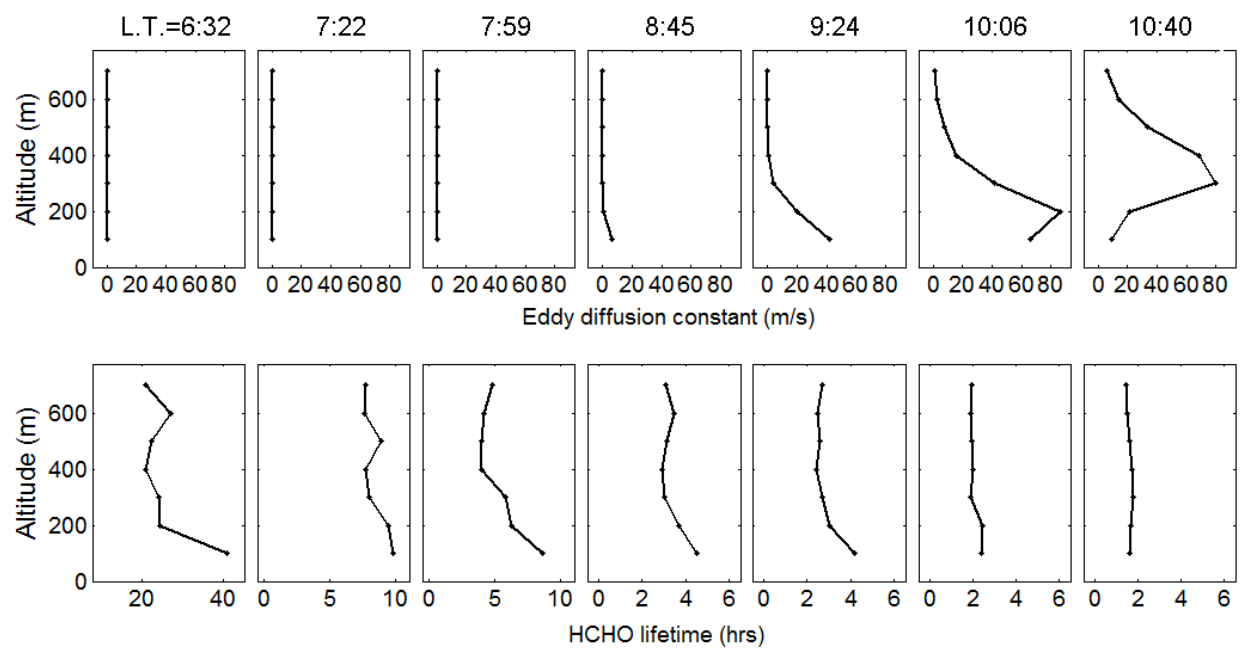
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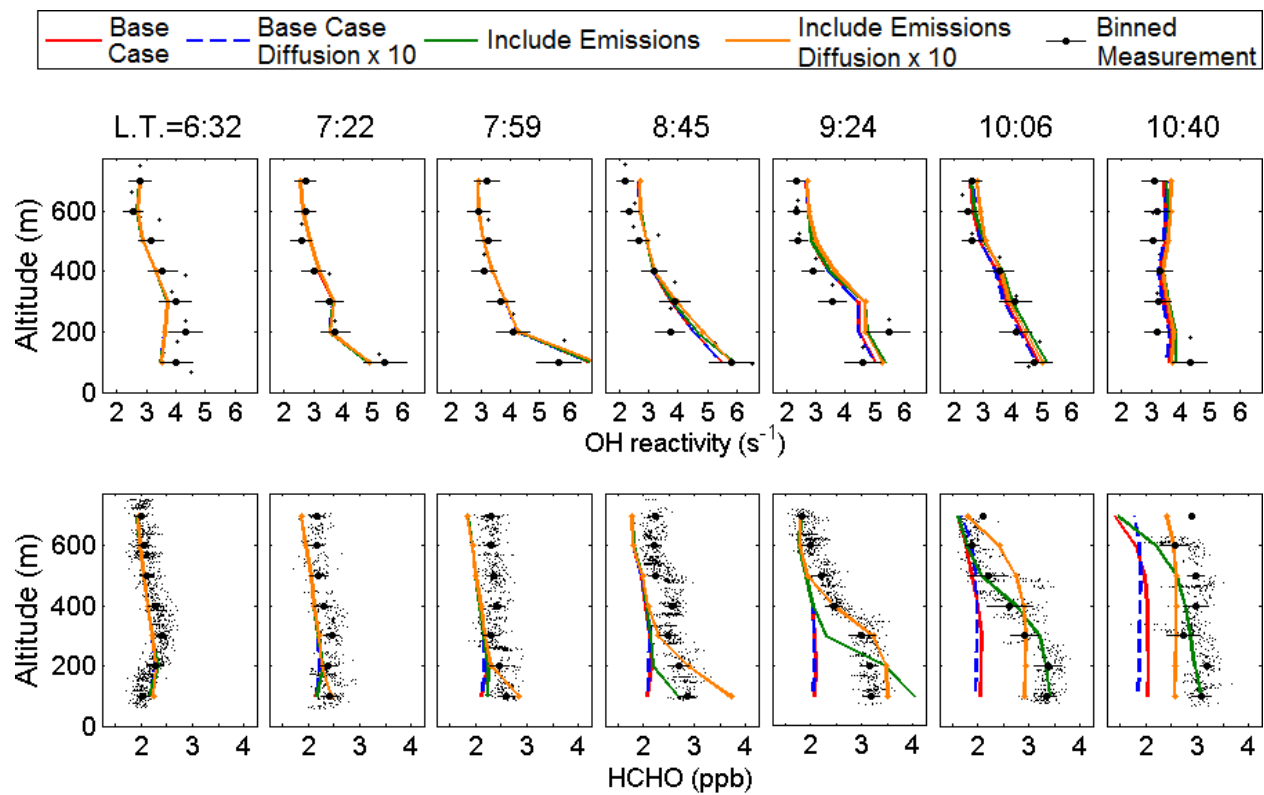


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 2 Figure S1. Measured and calculated OH reactivity and HCHO vertical profiles for every other
 3 Zeppelin ascent. Error bars on OH reactivity represent the measurement precision. Error bars on
 4 HCHO represent the standard deviation of the measurements in the given altitude bin.



1

2 Figure S2. Calculated eddy diffusion constants and HCHO lifetime with respect to
 3 photolysis/reaction with OH.



1
 2 Figure S3. Sensitivity to eddy diffusion constants for calculated OH reactivity and HCHO
 3 vertical profiles in the base case scenario and including direct HCHO emissions. Increased eddy
 4 diffusivity in most cases does not change the average model-measurement discrepancy.