

The Chemical Mechanism of MECCA

KPP version: 2.2.1_rs5

MECCA version: 3.3

Date: June 30, 2014.

Selected reactions:

“Tr && G && !S && !Cl && !Br && !I && !Hg”

Number of aerosol phases: 0

Number of species in selected mechanism:

Gas phase: 395

Aqueous phase: 0

All species: 395

Number of reactions in selected mechanism:

Gas phase (Gnn): 979

Aqueous phase (Annn): 0

Henry (Hnnn): 0

Photolysis (Jnn): 102

Heterogeneous (HETnnn): 0

Equilibria (EQnn): 0

Isotope exchange (DGnn): 0

Dummy (Dnn): 0

All equations: 1081

Table 1: Gas phase reactions

#	labels	reaction	rate coefficient	reference
G1000	StTrG	$\text{O}_2 + \text{O}(^1\text{D}) \rightarrow \text{O}(^3\text{P}) + \text{O}_2$	$3.2\text{E-}11*\text{EXP}(70./\text{temp})$	Sander et al. (2003)
G1001	StTrG	$\text{O}_2 + \text{O}(^3\text{P}) \rightarrow \text{O}_3$	$6.\text{E-}34*((\text{temp}/300.)**(-2.4))*\text{cair}$	Sander et al. (2003)
G2100	StTrG	$\text{H} + \text{O}_2 \rightarrow \text{HO}_2$	$\text{k_3rd}(\text{temp}, \text{cair}, 5.7\text{E-}32, 1.6, 7.5\text{E-}11, 0., 0.6)$	Sander et al. (2003)
G2104	StTrG	$\text{OH} + \text{O}_3 \rightarrow \text{HO}_2$	$1.7\text{E-}12*\text{EXP}(-940./\text{temp})$	Sander et al. (2003)
G2105	StTrG	$\text{OH} + \text{H}_2 \rightarrow \text{H}_2\text{O} + \text{H}$	$2.8\text{E-}12*\text{EXP}(-1800./\text{temp})$	Sander et al. (2006)
G2107	StTrG	$\text{HO}_2 + \text{O}_3 \rightarrow \text{OH}$	$1.\text{E-}14*\text{EXP}(-490./\text{temp})$	Sander et al. (2003)
G2109	StTrG	$\text{HO}_2 + \text{OH} \rightarrow \text{H}_2\text{O}$	$4.8\text{E-}11*\text{EXP}(250./\text{temp})$	Sander et al. (2003)
G2110	StTrG	$\text{HO}_2 + \text{HO}_2 \rightarrow \text{H}_2\text{O}_2$	k_HO2_HO2	Christensen et al. (2002), Kircher and Sander (1984)*
G2111	StTrG	$\text{H}_2\text{O} + \text{O}(^1\text{D}) \rightarrow 2 \text{OH}$	$2.2\text{E-}10$	Sander et al. (2003)
G2112	StTrG	$\text{H}_2\text{O}_2 + \text{OH} \rightarrow \text{H}_2\text{O} + \text{HO}_2$	$2.9\text{E-}12*\text{EXP}(-160./\text{temp})$	Sander et al. (2003)
G3101	StTrG	$\text{N}_2 + \text{O}(^1\text{D}) \rightarrow \text{O}(^3\text{P}) + \text{N}_2$	$1.8\text{E-}11*\text{EXP}(110./\text{temp})$	Sander et al. (2003)
G3103	StTrGN	$\text{NO} + \text{O}_3 \rightarrow \text{NO}_2 + \text{O}_2$	$3.\text{E-}12*\text{EXP}(-1500./\text{temp})$	Sander et al. (2003)
G3106	StTrGN	$\text{NO}_2 + \text{O}_3 \rightarrow \text{NO}_3 + \text{O}_2$	$1.2\text{E-}13*\text{EXP}(-2450./\text{temp})$	Sander et al. (2003)
G3108	StTrGN	$\text{NO}_3 + \text{NO} \rightarrow 2 \text{NO}_2$	$1.5\text{E-}11*\text{EXP}(170./\text{temp})$	Sander et al. (2003)
G3109	StTrGN	$\text{NO}_3 + \text{NO}_2 \rightarrow \text{N}_2\text{O}_5$	k_NO3_NO2	Sander et al. (2003)*
G3110	StTrGN	$\text{N}_2\text{O}_5 \rightarrow \text{NO}_2 + \text{NO}_3$	$\text{k_NO3_NO2}/(3.\text{E-}27*\text{EXP}(10990./\text{temp}))$	Sander et al. (2003)*
G3200	TrG	$\text{NO} + \text{OH} \rightarrow \text{HONO}$	$\text{k_3rd}(\text{temp}, \text{cair}, 7.\text{E-}31, 2.6, 3.6\text{E-}11, 0.1, 0.6)$	Sander et al. (2003)
G3201	StTrGN	$\text{NO} + \text{HO}_2 \rightarrow \text{NO}_2 + \text{OH}$	$3.5\text{E-}12*\text{EXP}(250./\text{temp})$	Sander et al. (2003)
G3202	StTrGN	$\text{NO}_2 + \text{OH} \rightarrow \text{HNO}_3$	$\text{k_3rd}(\text{temp}, \text{cair}, 1.48\text{E-}30, 3., 2.58\text{E-}11, 0., 0.6)$	Mollner et al. (2010)
G3203	StTrGN	$\text{NO}_2 + \text{HO}_2 \rightarrow \text{HNO}_4$	k_NO2_HO2	Sander et al. (2003)*
G3204	TrGN	$\text{NO}_3 + \text{HO}_2 \rightarrow \text{NO}_2 + \text{OH} + \text{O}_2$	$3.5\text{E-}12$	Sander et al. (2003)
G3205	TrG	$\text{HONO} + \text{OH} \rightarrow \text{NO}_2 + \text{H}_2\text{O}$	$1.8\text{E-}11*\text{EXP}(-390./\text{temp})$	Sander et al. (2003)
G3206	StTrGN	$\text{HNO}_3 + \text{OH} \rightarrow \text{H}_2\text{O} + \text{NO}_3$	k_HNO3_OH	Sander et al. (2003)*
G3207	StTrGN	$\text{HNO}_4 \rightarrow \text{NO}_2 + \text{HO}_2$	$\text{k_NO2_HO2}/(2.1\text{E-}27*\text{EXP}(10900./\text{temp}))$	Sander et al. (2003)*
G3208	StTrGN	$\text{HNO}_4 + \text{OH} \rightarrow \text{NO}_2 + \text{H}_2\text{O}$	$1.3\text{E-}12*\text{EXP}(380./\text{temp})$	Sander et al. (2003)
G4101	StTrG	$\text{CH}_4 + \text{OH} \rightarrow \text{CH}_3\text{O}_2 + \text{H}_2\text{O}$	$1.85\text{E-}20*\text{EXP}(2.82*\log(\text{temp})-987./\text{temp})$	Atkinson (2003)
G4102	TrG	$\text{CH}_3\text{OH} + \text{OH} \rightarrow \text{HCHO} + \text{HO}_2$	$7.3\text{E-}12*\text{EXP}(-620./\text{temp})$	Sander et al. (2003)
G4103a	StTrG	$\text{CH}_3\text{O}_2 + \text{HO}_2 \rightarrow \text{CH}_3\text{OOH}$	$4.1\text{E-}13*\text{EXP}(750./\text{temp})/(1.+1./497.7*\text{EXP}(1160./\text{temp}))$	Sander et al. (2003)*

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4103b	StTrG	$\text{CH}_3\text{O}_2 + \text{HO}_2 \rightarrow \text{HCHO} + \text{H}_2\text{O} + \text{O}_2$	$4.1\text{E-}13 \cdot \text{EXP}(750./\text{temp}) / (1. + 497.7 \cdot \text{EXP}(-1160./\text{temp}))$	Sander et al. (2003)*
G4104	StTrGN	$\text{CH}_3\text{O}_2 + \text{NO} \rightarrow \text{HCHO} + \text{NO}_2 + \text{HO}_2$	$2.8\text{E-}12 \cdot \text{EXP}(300./\text{temp})$	Sander et al. (2003)
G4105	TrGN	$\text{CH}_3\text{O}_2 + \text{NO}_3 \rightarrow \text{HCHO} + \text{HO}_2 + \text{NO}_2$	$1.3\text{E-}12$	Atkinson et al. (1999)
G4106a	StTrG	$\text{CH}_3\text{O}_2 \rightarrow \text{HCHO} + \text{HO}_2$	$2. \cdot \text{R02} \cdot 9.5\text{E-}14 \cdot \text{EXP}(390./\text{temp}) / (1. + 1. / 26.2 \cdot \text{EXP}(1130./\text{temp}))$	Sander et al. (2003)
G4106b	StTrG	$\text{CH}_3\text{O}_2 \rightarrow .5 \text{ HCHO} + .5 \text{ CH}_3\text{OH}$	$2. \cdot \text{R02} \cdot 9.5\text{E-}14 \cdot \text{EXP}(390./\text{temp}) / (1. + 26.2 \cdot \text{EXP}(-1130./\text{temp}))$	Sander et al. (2003)
G4107	StTrG	$\text{CH}_3\text{OOH} + \text{OH} \rightarrow .6 \text{ CH}_3\text{O}_2 + .4 \text{ HCHO} + .4 \text{ OH} + \text{H}_2\text{O}$	k_CH300H_OH	see note
G4108	StTrG	$\text{HCHO} + \text{OH} \rightarrow \text{CO} + \text{H}_2\text{O} + \text{HO}_2$	$9.52\text{E-}18 \cdot \text{EXP}(2.03 \cdot \log(\text{temp}) + 636./\text{temp})$	Sivakumaran et al. (2003)
G4109	TrGN	$\text{HCHO} + \text{NO}_3 \rightarrow \text{HNO}_3 + \text{CO} + \text{HO}_2$	$3.4\text{E-}13 \cdot \text{EXP}(-1900./\text{temp})$	Sander et al. (2003)*
G4110	StTrG	$\text{CO} + \text{OH} \rightarrow \text{H} + \text{CO}_2$	$1.57\text{E-}13 + \text{cair} \cdot 3.54\text{E-}33$	McCabe et al. (2001)
G4111	TrG	$\text{HCOOH} + \text{OH} \rightarrow \text{CO}_2 + \text{HO}_2 + \text{H}_2\text{O}$	$4.5\text{E-}13$	IUPAC (2013)
G4112e	TrGC	$\text{HCHO} + \text{HO}_2 \rightarrow \text{HOCH}_2\text{O}_2$	$7.7\text{E-}15 \cdot \text{EXP}(625./\text{temp})$	IUPAC (2013)
G4113e	TrGC	$\text{HOCH}_2\text{O}_2 \rightarrow \text{HCHO} + \text{HO}_2$	$2.\text{E}12 \cdot \text{EXP}(-7000./\text{temp})$	IUPAC (2013)
G4114e	TrGC	$\text{HOCH}_2\text{O}_2 + \text{HO}_2 \rightarrow .5 \text{ HOCH}_2\text{OOH} + .5 \text{ HCOOH} + .2 \text{ OH} + .2 \text{ HO}_2 + .3 \text{ H}_2\text{O}$	$5.6\text{E-}15 \cdot \text{EXP}(2300./\text{temp})$	Jenkin et al. (2007)
G4115e	TrGC	$\text{HOCH}_2\text{O}_2 + \text{NO} \rightarrow \text{NO}_2 + \text{HO}_2 + \text{HCOOH}$	$2.8\text{E-}12 \cdot \text{EXP}(300./\text{temp})$	Sander et al. (2003)
G4116e	TrGC	$\text{HOCH}_2\text{O}_2 + \text{NO}_3 \rightarrow \text{NO}_2 + \text{HO}_2 + \text{HCOOH}$	$1.2\text{E-}12$	see note
G4117e	TrGC	$\text{HOCH}_2\text{O}_2 \rightarrow \text{HCOOH} + .62 \text{ HO}_2$	$1.4\text{E-}12 \cdot \text{R02}$	see note
G4118e	TrGC	$\text{HOCH}_2\text{OOH} + \text{OH} \rightarrow \text{HOCH}_2\text{O}_2 + \text{H}_2\text{O}$	$0.6 \cdot \text{k_CH300H_OH} + \text{krohro}$	see note
G4119e	TrGC	$\text{HOCH}_2\text{OOH} + \text{OH} \rightarrow \text{OH} + \text{HCOOH} + \text{H}_2\text{O}$	k_s*fsoh*fsooh	see note
G4200	TrGC	$\text{C}_2\text{H}_6 + \text{OH} \rightarrow \text{C}_2\text{H}_5\text{O}_2 + \text{H}_2\text{O}$	$1.49\text{E-}17 \cdot \text{temp} \cdot \text{temp} \cdot \text{EXP}(-499./\text{temp})$	Atkinson (2003)
G4201e	TrGC	$\text{C}_2\text{H}_4 + \text{O}_3 \rightarrow \text{HCHO} + .63 \text{ CO} + .13 \text{ HO}_2 + 0.37 \text{ HOCH}_2\text{OOH} + .13 \text{ OH}$	$1.2\text{E-}14 \cdot \text{EXP}(-2630./\text{temp})$	Sander et al. (2003)*
G4202	TrGC	$\text{C}_2\text{H}_4 + \text{OH} \rightarrow \text{HOCH}_2\text{CH}_2\text{O}_2 + \text{H}_2\text{O}$	k_3rd(temp, cair, 1.E-28, 0.8, 8.8E-12, 0., 0.6)	Sander et al. (2003)
G4203	TrGC	$\text{C}_2\text{H}_5\text{O}_2 + \text{HO}_2 \rightarrow \text{C}_2\text{H}_5\text{OOH}$	$7.5\text{E-}13 \cdot \text{EXP}(700./\text{temp})$	Sander et al. (2003)
G4204	TrGNC	$\text{C}_2\text{H}_5\text{O}_2 + \text{NO} \rightarrow \text{CH}_3\text{CHO} + \text{HO}_2 + \text{NO}_2$	$2.6\text{E-}12 \cdot \text{EXP}(365./\text{temp})$	Sander et al. (2003)
G4205	TrGNC	$\text{C}_2\text{H}_5\text{O}_2 + \text{NO}_3 \rightarrow \text{CH}_3\text{CHO} + \text{HO}_2 + \text{NO}_2$	$2.3\text{E-}12$	Atkinson et al. (1999)
G4206	TrGC	$\text{C}_2\text{H}_5\text{O}_2 \rightarrow .98 \text{ CH}_3\text{CHO} + .38 \text{ HO}_2 + .02 \text{ HOCH}_2\text{CH}_2\text{O}_2$	$3.1\text{E-}13 \cdot \text{R02}$	Rickard and Pascoe (2009)*
G4207	TrGC	$\text{C}_2\text{H}_5\text{OOH} + \text{OH} \rightarrow .43 \text{ C}_2\text{H}_5\text{O}_2 + .43 \text{ H}_2\text{O} + .57 \text{ CH}_3\text{CHO} + .57 \text{ OH}$	$0.6 \cdot \text{k_CH300H_OH} + 8.01\text{E-}12$	see note
G4208ea	TrGC	$\text{CH}_3\text{CHO} + \text{OH} \rightarrow \text{CH}_3\text{C(O)OO} + \text{H}_2\text{O}$	$4.4\text{E-}12 \cdot \text{EXP}(365./\text{temp}) \cdot 0.95$	Atkinson et al. (2006)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4208eb	TrGC	$\text{CH}_3\text{CHO} + \text{OH} \rightarrow .84 \text{HCOCH}_2\text{O}_2 + .1 \text{HCHO} + .1 \text{CO}$ $+ .06 \text{GLYOX} + .16 \text{OH} + \text{H}_2\text{O}$	$4.4\text{E-}12*\text{EXP}(365./\text{temp})*0.05$	Atkinson et al. (2006)
G4209	TrGNC	$\text{CH}_3\text{CHO} + \text{NO}_3 \rightarrow \text{CH}_3\text{C(O)OO} + \text{HNO}_3$	KN03AL	Sander et al. (2003)
G4210e	TrGC	$\text{CH}_3\text{COOH} + \text{OH} \rightarrow \text{CH}_3\text{O}_2 + \text{CO}_2 + \text{H}_2\text{O}$	$4.2\text{E-}14*\text{exp}(850./\text{temp})$	IUPAC (2013)
G4211et1	TrGC	$\text{CH}_3\text{C(O)OO} + \text{HO}_2 \rightarrow \text{OH} + \text{CH}_3\text{O}_2 + \text{CO}_2$	KAPH02*0.70	Taraborrelli (2014)*
G4211et2	TrGC	$\text{CH}_3\text{C(O)OO} + \text{HO}_2 \rightarrow \text{CH}_3\text{C(O)OOH}$	KAPH02*0.12	Taraborrelli (2014)*
G4211et3	TrGC	$\text{CH}_3\text{C(O)OO} + \text{HO}_2 \rightarrow \text{CH}_3\text{COOH} + \text{O}_3$	KAPH02*0.18	Taraborrelli (2014)*
G4212	TrGNC	$\text{CH}_3\text{C(O)OO} + \text{NO} \rightarrow \text{CH}_3\text{O}_2 + \text{CO}_2 + \text{NO}_2$	$8.1\text{E-}12*\text{EXP}(270./\text{temp})$	Tyndall et al. (2001)
G4213	TrGNC	$\text{CH}_3\text{C(O)OO} + \text{NO}_2 \rightarrow \text{PAN}$	k_CH3C03_N02	Tyndall et al. (2001)*
G4214	TrGNC	$\text{CH}_3\text{C(O)OO} + \text{NO}_3 \rightarrow \text{CH}_3\text{O}_2 + \text{NO}_2 + \text{CO}_2$	4.E-12	Canosa-Mas et al. (1996)
G4217	TrGC	$\text{CH}_3\text{C(O)OO} \rightarrow .7 \text{CH}_3\text{O}_2 + .7 \text{CO}_2 + .3 \text{CH}_3\text{COOH}$	$1.00\text{E-}11*\text{R02}$	Rickard and Pascoe (2009)
G4218	TrGC	$\text{CH}_3\text{C(O)OOH} + \text{OH} \rightarrow \text{CH}_3\text{C(O)OO} + \text{H}_2\text{O}$	$0.6*k_{\text{CH300H_OH}}$	Rickard and Pascoe (2009)*
G4220	TrGNC	$\text{PAN} + \text{OH} \rightarrow \text{HCHO} + \text{CO} + \text{NO}_2 + \text{H}_2\text{O}$	$9.50\text{E-}13*\text{EXP}(-650./\text{temp})$	Rickard and Pascoe (2009)
G4221	TrGNC	$\text{PAN} \rightarrow \text{CH}_3\text{C(O)OO} + \text{NO}_2$	k_PAN_M	Sander et al. (2003)*
G4223e	TrGC	$\text{HOCH}_2\text{CHO} + \text{OH} \rightarrow .84 \text{HOCH}_2\text{CO} + .16 \text{HOCHCHO}$ $+ .2 \text{HO}_2 + \text{H}_2\text{O}$	8.00E-12	Rickard and Pascoe (2009)
G4224e	TrGNC	$\text{HOCH}_2\text{CHO} + \text{NO}_3 \rightarrow \text{HOCH}_2\text{CO} + \text{HNO}_3$	KN03AL	Rickard and Pascoe (2009)
G4255et2	TrGC	$\text{HOCH}_2\text{CO} \rightarrow \text{HOCH}_2\text{CO}_3$	KDEC*.97	Taraborrelli (2014)*
G4255et3	TrGC	$\text{HOCH}_2\text{CO} \rightarrow \text{OH} + \text{HCHO} + \text{CO}$	KDEC*.03	Taraborrelli (2014)*
G4256et2	TrGC	$\text{HOCHCHO} \rightarrow \text{GLYOX} + \text{HO}_2$	KDEC	Taraborrelli (2014)
G4225	TrGC	$\text{HOCH}_2\text{CO}_3 \rightarrow .7 \text{HCHO} + .7 \text{CO}_2 + .7 \text{HO}_2 + .3$ $\text{HOCH}_2\text{CO}_2\text{H}$	$1.00\text{E-}11*\text{R02}$	Rickard and Pascoe (2009)
G4226ea	TrGC	$\text{HOCH}_2\text{CO}_3 + \text{HO}_2 \rightarrow \text{HCHO} + \text{HO}_2 + \text{OH} + \text{CO}_2$	KAPH02*rco3_oh	Taraborrelli (2014)*
G4226eb	TrGC	$\text{HOCH}_2\text{CO}_3 + \text{HO}_2 \rightarrow \text{HOCH}_2\text{CO}_3\text{H}$	KAPH02*rco3_ooh	Taraborrelli (2014)*
G4226ec	TrGC	$\text{HOCH}_2\text{CO}_3 + \text{HO}_2 \rightarrow \text{HOCH}_2\text{CO}_2\text{H} + \text{O}_3$	KAPH02*rco3_o3	Taraborrelli (2014)*
G4227	TrGNC	$\text{HOCH}_2\text{CO}_3 + \text{NO} \rightarrow \text{NO}_2 + \text{HO}_2 + \text{HCHO} + \text{CO}_2$	KAPNO	Rickard and Pascoe (2009)
G4228	TrGNC	$\text{HOCH}_2\text{CO}_3 + \text{NO}_2 \rightarrow \text{PHAN}$	k_CH3C03_N02	Rickard and Pascoe (2009)
G4229	TrGNC	$\text{HOCH}_2\text{CO}_3 + \text{NO}_3 \rightarrow \text{NO}_2 + \text{HO}_2 + \text{HCHO} + \text{CO}_2$	KR02N03*1.60	Rickard and Pascoe (2009)
G4230e	TrGC	$\text{HOCH}_2\text{CO}_2\text{H} + \text{OH} \rightarrow .09 \text{HCHO} + .09 \text{CO}_2 + .91$ $\text{HCOCO}_2\text{H} + \text{HO}_2 + \text{H}_2\text{O}$	$\text{kco2h}+\text{ks}*\text{fsoh}*\text{fco2h}$	Taraborrelli (2014)
G4231ea	TrGC	$\text{HOCH}_2\text{CO}_3\text{H} + \text{OH} \rightarrow \text{HOCH}_2\text{CO}_3 + \text{H}_2\text{O}$	$0.6*k_{\text{CH300H_OH}}$	Taraborrelli (2014)
G4231eb	TrGC	$\text{HOCH}_2\text{CO}_3\text{H} + \text{OH} \rightarrow \text{HCOCO}_3\text{H} + \text{HO}_2$	$\text{ks}*\text{fsoh}*\text{fco2h}$	Taraborrelli (2014)
G4232	TrGNC	$\text{PHAN} \rightarrow \text{HOCH}_2\text{CO}_3 + \text{NO}_2$	k_PAN_M	Rickard and Pascoe (2009)
G4233	TrGNC	$\text{PHAN} + \text{OH} \rightarrow \text{HCHO} + \text{CO} + \text{NO}_2 + \text{H}_2\text{O}$	1.12E-12	Rickard and Pascoe (2009)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4234e	TrGC	GLYOX + OH \rightarrow 1.2 CO + .6 HO ₂ + .4 HCOCO3A + H ₂ O	3.1E-12*EXP(340./temp)	IUPAC (2013)
G4235e	TrGNC	GLYOX + NO ₃ \rightarrow 1.2 CO + .6 HO ₂ + .4 HCOCO3A + HNO ₃	KN03AL	Rickard and Pascoe (2009)
G4235et2	TrGNC	HCOCO3A \rightarrow 1.5 CO + .5 HO ₂ + .5 OH + .5 CO ₂	KDEC	Taraborrelli (2014)
G4236	TrGC	HCOCO ₃ \rightarrow .7 CO + .7 HO ₂ + .7 CO ₂ + .3 HCOCO ₂ H	1.00E-11*R02	Rickard and Pascoe (2009)
G4237e	TrGC	HCOCO ₃ + HO ₂ \rightarrow HO ₂ + CO + CO ₂ + OH	KAPH02	Feierabend et al. (2008), Taraborrelli (2014)
G4238	TrGNC	HCOCO ₃ + NO \rightarrow HO ₂ + CO + NO ₂ + CO ₂	KAPN0	Rickard and Pascoe (2009)
G4239	TrGNC	HCOCO ₃ + NO ₃ \rightarrow HO ₂ + CO + NO ₂ + CO ₂	KR02N03*1.60	Rickard and Pascoe (2009)
G4239t2	TrGNC	HCOCO ₃ + NO ₂ \rightarrow HO ₂ + CO + NO ₃ + CO ₂	k_CH3C03_N02	Orlando and Tyndall (2001), Taraborrelli (2014)*
G4240	TrGC	HCOCO ₂ H + OH \rightarrow CO + HO ₂ + CO ₂ + H ₂ O	kco2h+kt*fo*fco2h	Taraborrelli (2014)
G4241	TrGC	HCOCO ₃ H + OH \rightarrow .2 HCOCO ₃ + .8 CO + .8 OH + .8 CO ₂ + H ₂ O	0.6*k_CH300H_OH+kt*fo*fco2h	Taraborrelli (2014)
G4242	TrGC	HOCH ₂ CH ₂ O ₂ \rightarrow .6 HOCH ₂ CH ₂ O + .2 HOCH ₂ CHO + .2 ETHGLY	2.00E-12*R02	Rickard and Pascoe (2009)
G4244	TrGC	HOCH ₂ CH ₂ O ₂ + HO ₂ \rightarrow HYETHO2H	2.00E-13*EXP(1250./temp)	Rickard and Pascoe (2009)
G4243	TrGNC	HOCH ₂ CH ₂ O ₂ + NO \rightarrow .24875 HO ₂ + .4975 HCHO + .74625 HOCH ₂ CH ₂ O + .995 NO ₂ + .005 ETHOHNO3	KR02N0	Orlando et al. (1998b)*
G4245	TrGNC	ETHOHNO3 + OH \rightarrow .93 NO3CH2CHO + .93 HO ₂ + .07 HOCH ₂ CHO + .07 NO ₂ + H ₂ O	ks*(fsoh*fch2ono2+fono2*fpch2oh)+krohro	Taraborrelli (2014)
G4246a	TrGC	HYETHO2H + OH \rightarrow HOCH ₂ CH ₂ O ₂ + H ₂ O	0.6*k_CH300H_OH	Rickard and Pascoe (2009)
G4246b	TrGC	HYETHO2H + OH \rightarrow HOCH ₂ CHO + OH + H ₂ O	ks*fsoh*fpch2oh	Taraborrelli (2014)
G4246c	TrGC	HYETHO2H + OH \rightarrow HOOCH2CHO + HO ₂ + H ₂ O	ks*fsoh*fpch2oh+krohro	Taraborrelli (2014)
G4247a	TrGC	HOCH ₂ CH ₂ O \rightarrow HO ₂ + HOCH ₂ CHO	6.00E-14*EXP(-550./temp)*C(ind_02)	Orlando et al. (1998b)
G4247b	TrGC	HOCH ₂ CH ₂ O \rightarrow HO ₂ + HCHO + HCHO	9.50E+13*EXP(-5988./temp)	Orlando et al. (1998b)
G4248	TrGC	ETHGLY + OH \rightarrow HOCH ₂ CHO + HO ₂ + H ₂ O	2*ks*fsoh*fpch2oh+2*krohro	Taraborrelli (2014)
G4249e	TrGC	HCOCH2O2 \rightarrow 0.6 HCHO + 0.6 CO + 0.6 HO ₂ + 0.2 GLYOX + 0.2 HOCH ₂ CHO	2.00E-12*R02	Taraborrelli (2014)
G4250e	TrGC	HCOCH2O2 + HO ₂ \rightarrow 0.85 HOOCH2CHO + 0.15 HCHO + 0.15 CO + 0.15 HO ₂ + 0.15 OH	KR02H02*0.387	Taraborrelli (2014)
G4251e	TrGC	HCOCH2O2 + NO \rightarrow NO ₂ + HCHO + CO + HO ₂	KR02N0	Taraborrelli (2014)
G4252e	TrGC	HCOCH2O2 + NO ₃ \rightarrow HCHO + CO + HO ₂ + NO ₂	KR02N03	Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4253e	TrGC	$\text{HOOCH}_2\text{CHO} + \text{OH} \rightarrow .71 \text{ OH} + .31 \text{ HCHO} + .31 \text{ CO} + .40 \text{ GLYOX} + .29 \text{ HCOCH}_2\text{O}_2$	$0.6 \cdot k_{\text{CH300H_OH}} + k_{\text{sfsooh}} \cdot f_{\text{cho}} + .8 \cdot 8.E-12$	Taraborrelli (2014)
G4254e	TrGNC	$\text{HOOCH}_2\text{CHO} + \text{NO}_3 \rightarrow \text{OH} + \text{HCHO} + \text{CO} + \text{HNO}_3$	KN03AL	Rickard and Pascoe (2009)
G4257e	TrGC	$\text{NO}_3\text{CH}_2\text{CHO} + \text{OH} \rightarrow \text{HCHO} + \text{CO}_2 + \text{NO}_2 + \text{H}_2\text{O}$	$1.E-11$	Paulot et al. (2009a), Taraborrelli (2014)
G4258e	TrGNC	$\text{HOOCH}_2\text{CO}_3 + \text{NO} \rightarrow \text{NO}_2 + \text{OH} + \text{HCHO} + \text{CO}_2$	KAPNO	Taraborrelli (2014)
G4259e	TrGNC	$\text{HOOCH}_2\text{CO}_3 + \text{NO}_3 \rightarrow \text{NO}_2 + \text{OH} + \text{HCHO} + \text{CO}_2$	KR02NO3*1.60	Taraborrelli (2014)
G4260e	TrGC	$\text{HOOCH}_2\text{CO}_3 + \text{HO}_2 \rightarrow 2 \text{ OH} + \text{HCHO} + \text{CO}_2$	KAPH02*rc03_oh	Taraborrelli (2014)
G4260et2	TrGC	$\text{HOOCH}_2\text{CO}_3 + \text{HO}_2 \rightarrow \text{HOOCH}_2\text{CO}_3\text{H}$	KAPH02*rc03_ooh	Taraborrelli (2014)*
G4260et3	TrGC	$\text{HOOCH}_2\text{CO}_3 + \text{HO}_2 \rightarrow \text{HOOCH}_2\text{CO}_2\text{H} + \text{O}_3$	KAPH02*rc03_o3	Taraborrelli (2014)*
G4261e	TrGC	$\text{HOOCH}_2\text{CO}_3 \rightarrow .7 \text{ OH} + .7 \text{ HCHO} + .7 \text{ CO}_2 + .3 \text{ HOOCH}_2\text{CO}_2\text{H}$	$1.00E-11 \cdot R02$	Taraborrelli (2014)
G4262e	TrGC	$\text{HOOCH}_2\text{CO}_3\text{H} + \text{OH} \rightarrow \text{HOOCH}_2\text{CO}_3 + \text{H}_2\text{O}$	$2 \cdot 0.6 \cdot k_{\text{CH300H_OH}}$	Taraborrelli (2014)
G4263e	TrGC	$\text{HOOCH}_2\text{CO}_3\text{H} + \text{OH} \rightarrow \text{HCOCO}_3\text{H} + \text{OH} + \text{H}_2\text{O}$	$k_{\text{sfsooh}} \cdot f_{\text{co2h}}$	Taraborrelli (2014)
G4265e	TrGC	$\text{HOOCH}_2\text{CO}_2\text{H} + \text{OH} \rightarrow \text{HCOCO}_2\text{H} + \text{OH} + \text{H}_2\text{O}$	$k_{\text{sfsooh}} \cdot f_{\text{co2h}} + k_{\text{co2h}}$	Taraborrelli (2014)
G4266e	TrGC	$\text{CH}_2\text{CO} + \text{OH} \rightarrow .6 \text{ HCHO} + .6 \text{ HO}_2 + .6 \text{ CO} + .4 \text{ HOOCH}_2\text{CO}_2\text{H}$	$2.8E-12 \cdot \exp(510./\text{temp})$	Baulch et al. (2005), Taraborrelli (2014)*
G4267e	TrGC	$\text{CH}_3\text{CHOHOOH} + \text{OH} \rightarrow \text{CH}_3\text{COOH} + \text{OH}$	$k_{\text{tftooh}} \cdot f_{\text{ftoh}} + k_{\text{rohro}}$	Taraborrelli (2014)
G4268e	TrGC	$\text{CH}_3\text{CHOHOOH} + \text{OH} \rightarrow \text{CH}_3\text{CHOHO}_2$	$0.6 \cdot k_{\text{CH300H_OH}}$	Taraborrelli (2014)
G4269e	TrGC	$\text{CH}_3\text{CHOHO}_2 \rightarrow \text{CH}_3\text{CHO} + \text{HO}_2$	$3.46E12 \cdot \exp(-12500./(1.98 \cdot \text{temp}))$	Hermans et al. (2005), Taraborrelli (2014)
G4270e	TrGC	$\text{CH}_3\text{CHO} + \text{HO}_2 \rightarrow \text{CH}_3\text{CHOHO}_2$	$3.46E12 \cdot \exp(-12500./(1.98 \cdot \text{temp})) / (6.34E26 \cdot \exp(-14700./(1.98 \cdot \text{temp})))$	Hermans et al. (2005), Taraborrelli (2014)
G4271e	TrGC	$\text{CH}_3\text{CHOHO}_2 + \text{HO}_2 \rightarrow .5 \text{ CH}_3\text{CHOHOOH} + .3 \text{ CH}_3\text{COOH} + .2 \text{ CH}_3\text{O}_2 + .2 \text{ HCOOH} + .2 \text{ OH}$	$5.6E-15 \cdot \exp(2300./\text{temp})$	Taraborrelli (2014)
G4272e	TrGC	$\text{CH}_3\text{CHOHO}_2 \rightarrow \text{CH}_3\text{O}_2 + \text{HCOOH} + \text{OH}$	$1.4E-12 \cdot R02$	Taraborrelli (2014)
G4273e	TrGC	$\text{CH}_3\text{CHOHO}_2 + \text{NO} \rightarrow \text{CH}_3\text{O}_2 + \text{HCOOH} + \text{OH} + \text{NO}_2$	KR02NO	Taraborrelli (2014)
G4300	TrGC	$\text{C}_3\text{H}_8 + \text{OH} \rightarrow .736 \text{ iC}_3\text{H}_7\text{O}_2 + .264 \text{ C}_2\text{H}_5\text{O}_2 + .264 \text{ CO}_2 + .264 \text{ HO}_2 + \text{H}_2\text{O}$	$1.55E-17 \cdot \text{temp} \cdot \text{temp} \cdot \exp(-61./\text{temp})$	Rickard and Pascoe (2009)*
G4301et2	TrGC	$\text{C}_3\text{H}_6 + \text{O}_3 \rightarrow .0855 \text{ CH}_3\text{CHOHOOH} + .4389 \text{ CH}_3\text{CHO} + .4389 \text{ H}_2\text{O}_2 + .0456 \text{ CH}_3\text{COOH} + .285 \text{ HCOCH}_2\text{O}_2 + .0855 \text{ CH}_4 + .0855 \text{ CO}_2 + .0342 \text{ CH}_2\text{CO} + .0513 \text{ CH}_3\text{OH} + .0228 \text{ CH}_3\text{C(O)OO} + .57 \text{ HCHO} + .2709 \text{ CO} + .0688 \text{ HO}_2 + .1591 \text{ HOCH}_2\text{OOH} + .43 \text{ CH}_3\text{CHO} + .3766 \text{ OH}$	$6.5E-15 \cdot \exp(-1900./\text{temp})$	Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4302	TrGC	$\text{C}_3\text{H}_6 + \text{OH} \rightarrow \text{HYPROPO2}$	$\text{k_3rd}(\text{temp}, \text{cair}, 8.\text{E-}27, 3.5, 3.\text{E-}11, 0., 0.5)$	Atkinson et al. (1999)
G4303	TrGNC	$\text{C}_3\text{H}_6 + \text{NO}_3 \rightarrow \text{PRONO3BO2}$	$4.6\text{E-}13 * \text{EXP}(-1155./\text{temp})$	Atkinson et al. (1999)
G4304	TrGC	$\text{iC}_3\text{H}_7\text{O}_2 + \text{HO}_2 \rightarrow \text{iC}_3\text{H}_7\text{OOH}$	$1.9\text{E-}13 * \text{EXP}(1300./\text{temp})$	Atkinson (1997)*
G4305	TrGNC	$\text{iC}_3\text{H}_7\text{O}_2 + \text{NO} \rightarrow .96 \text{CH}_3\text{COCH}_3 + .96 \text{HO}_2 + .96 \text{NO}_2 + .04 \text{iC}_3\text{H}_7\text{ONO}_2$	$2.7\text{E-}12 * \text{EXP}(360./\text{temp})$	Atkinson et al. (1999)
G4306	TrGC	$\text{iC}_3\text{H}_7\text{O}_2 \rightarrow \text{CH}_3\text{COCH}_3 + .8 \text{HO}_2$	$4.\text{E-}14 * \text{R02}$	Rickard and Pascoe (2009)*
G4307	TrGC	$\text{iC}_3\text{H}_7\text{OOH} + \text{OH} \rightarrow .27 \text{iC}_3\text{H}_7\text{O}_2 + .73 \text{CH}_3\text{COCH}_3 + .73 \text{OH} + \text{H}_2\text{O}$	$1.66\text{E-}11 + 0.6 * \text{k_CH300H_OH}$	Rickard and Pascoe (2009)*
G4311	TrGC	$\text{CH}_3\text{COCH}_3 + \text{OH} \rightarrow \text{CH}_3\text{COCH}_2\text{O}_2 + \text{H}_2\text{O}$	$1.33\text{E-}13 + 3.82\text{E-}11 * \text{EXP}(-2000./\text{temp})$	Sander et al. (2003)
G4312e	TrGC	$\text{CH}_3\text{COCH}_2\text{O}_2 + \text{HO}_2 \rightarrow .15 \text{OH} + .15 \text{CH}_3\text{C(O)OO} + .15 \text{HCHO} + .85 \text{CH}_3\text{COCH}_2\text{O}_2\text{H}$	$8.6\text{E-}13 * \text{EXP}(700./\text{temp})$	Taraborrelli (2014)
G4313	TrGNC	$\text{CH}_3\text{COCH}_2\text{O}_2 + \text{NO} \rightarrow \text{CH}_3\text{C(O)OO} + \text{HCHO} + \text{NO}_2$	$2.9\text{E-}12 * \text{EXP}(300./\text{temp})$	Sander et al. (2003)
G4314	TrGC	$\text{CH}_3\text{COCH}_2\text{O}_2 \rightarrow .6 \text{CH}_3\text{C(O)OO} + .6 \text{HCHO} + .2 \text{MGLYOX} + .2 \text{CH}_3\text{COCH}_2\text{OH}$	$7.5\text{E-}13 * \text{EXP}(500./\text{temp}) * 2. * \text{R02}$	Tyndall et al. (2001)
G4321	TrGNC	$\text{CH}_3\text{COCH}_2\text{O}_2 + \text{NO}_3 \rightarrow \text{CH}_3\text{C(O)OO} + \text{HCHO} + \text{NO}_2$	KR02N03	Rickard and Pascoe (2009)
G4315a	TrGC	$\text{CH}_3\text{COCH}_2\text{O}_2\text{H} + \text{OH} \rightarrow \text{CH}_3\text{COCH}_2\text{O}_2 + \text{H}_2\text{O}$	$0.6 * \text{k_CH300H_OH}$	Rickard and Pascoe (2009)*
G4315b	TrGC	$\text{CH}_3\text{COCH}_2\text{O}_2\text{H} + \text{OH} \rightarrow \text{MGLYOX} + \text{OH} + \text{H}_2\text{O}$	ks*fsooh*fco	Taraborrelli (2014)
G4316e	TrGC	$\text{CH}_3\text{COCH}_2\text{OH} + \text{OH} \rightarrow \text{CH}_3\text{COCHOH} + \text{H}_2\text{O}$	$1.60\text{E-}12 * \text{EXP}(305./\text{temp})$	Taraborrelli (2014)
G4336ea	TrGC	$\text{CH}_3\text{COCHOH} \rightarrow \text{MGLYOX} + \text{HO}_2$	0.8485	Taraborrelli (2014)
G4317e	TrGC	$\text{MGLYOX} + \text{OH} \rightarrow .4 \text{CH}_3\text{O}_2 + .6 \text{CH}_3\text{C(O)OO} + 1.4 \text{CO}$	$1.9\text{E-}12 * \text{EXP}(575./\text{temp})$	Baeza-Romero et al. (2007), IU-PAC (2013)
G4331	TrGNC	$\text{MGLYOX} + \text{NO}_3 \rightarrow \text{CH}_3\text{C(O)OO} + \text{CO} + \text{HNO}_3$	$\text{KN03AL} * 2.4$	Rickard and Pascoe (2009)
G4320	TrGNC	$\text{iC}_3\text{H}_7\text{ONO}_2 + \text{OH} \rightarrow \text{CH}_3\text{COCH}_3 + \text{NO}_2$	$6.2\text{E-}13 * \text{EXP}(-230./\text{temp})$	Atkinson et al. (1999)
G4322	TrGC	$\text{HYPROPO2} \rightarrow \text{CH}_3\text{CHO} + \text{HCHO} + \text{HO}_2$	$8.80\text{E-}13 * \text{R02}$	Rickard and Pascoe (2009)
G4323	TrGC	$\text{HYPROPO2} + \text{HO}_2 \rightarrow \text{HYPROPO2H}$	$\text{KR02H02} * 0.520$	Rickard and Pascoe (2009)
G4324	TrGNC	$\text{HYPROPO2} + \text{NO} \rightarrow \text{CH}_3\text{CHO} + \text{HCHO} + \text{HO}_2 + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009)
G4325	TrGNC	$\text{HYPROPO2} + \text{NO}_3 \rightarrow \text{CH}_3\text{CHO} + \text{HCHO} + \text{HO}_2 + \text{NO}_2$	KR02N03	Rickard and Pascoe (2009)
G4326a	TrGC	$\text{HYPROPO2H} + \text{OH} \rightarrow \text{HYPROPO2}$	$0.6 * \text{k_CH300H_OH}$	Rickard and Pascoe (2009)
G4326b	TrGC	$\text{HYPROPO2H} + \text{OH} \rightarrow \text{CH}_3\text{COCH}_2\text{OH} + \text{OH}$	$\text{ks*fsoh*fpch2oh} + \text{kt*fsooh*fpch2oh}$	Taraborrelli (2014)
G4327	TrGNC	$\text{PRONO3BO2} + \text{HO}_2 \rightarrow \text{PR2O2HNO3}$	$\text{KR02H02} * 0.520$	Rickard and Pascoe (2009)
G4328	TrGNC	$\text{PRONO3BO2} + \text{NO} \rightarrow \text{NOA} + \text{HO}_2 + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009)
G4329	TrGNC	$\text{PRONO3BO2} + \text{NO}_3 \rightarrow \text{NOA} + \text{HO}_2 + \text{NO}_2$	KR02N03	Rickard and Pascoe (2009)
G4330a	TrGNC	$\text{PR2O2HNO3} + \text{OH} \rightarrow \text{PRONO3BO2}$	$1.90\text{E-}12 * \text{EXP}(190./\text{temp})$	Rickard and Pascoe (2009)
G4330b	TrGNC	$\text{PR2O2HNO3} + \text{OH} \rightarrow \text{NOA} + \text{OH}$	kt*fsooh*fch2ono2	Rickard and Pascoe (2009)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4332	TrGNC	$\text{NOA} + \text{OH} \rightarrow \text{MGLYOX} + \text{NO}_2$	$\text{ks*fco*fono2+kp*fco}$	Taraborrelli (2014)
G4333e	TrGC	$\text{HOCH}_2\text{COCHO} + \text{OH} \rightarrow .8609 \text{HOCH}_2\text{CO} + .8609 \text{CO} + .1391 \text{HCOCOCHO} + .1391 \text{HO}_2$	$1.9\text{E-}12*\text{EXP}(575./\text{temp})+\text{ks*fsoh*fco}$	Taraborrelli (2014)
G4334e	TrGNC	$\text{HOCH}_2\text{COCHO} + \text{NO}_3 \rightarrow \text{HOCH}_2\text{CO} + \text{CO} + \text{HNO}_3$	KN03AL*2.4	Taraborrelli (2014)
G4337e	TrGC	$\text{CH}_3\text{COCO}_2\text{H} + \text{OH} \rightarrow \text{CH}_3\text{C(O)OO} + \text{H}_2\text{O} + \text{CO}_2$	$4.9\text{E-}14*\text{EXP}(276./\text{temp})$	Mellouki and Mu (2003), Taraborrelli (2014)
G4338e	TrGC	$\text{HOCH}_2\text{COCH}_2\text{O}_2 \rightarrow \text{HCHO} + \text{HOCH}_2\text{CO}$	$\text{R02*2.0E-}12$	Taraborrelli (2014)
G4339e	TrGC	$\text{HOCH}_2\text{COCH}_2\text{O}_2 + \text{HO}_2 \rightarrow .15 \text{HCHO} + .15 \text{HOCH}_2\text{CO} + .15 \text{OH} + .85 \text{HOCH}_2\text{COCH}_2\text{OOH}$	KR02H02*0.520	Taraborrelli (2014)
G4340e	TrGC	$\text{HOCH}_2\text{COCH}_2\text{O}_2 + \text{NO} \rightarrow \text{HCHO} + \text{HOCH}_2\text{CO} + \text{NO}_2$	KR02N0	Taraborrelli (2014)
G4341e	TrGC	$\text{HOCH}_2\text{COCH}_2\text{OOH} + \text{OH} \rightarrow \text{HOCH}_2\text{COCHO} + \text{OH}$	ks*fsooh*fco	Taraborrelli (2014)
G4342e	TrGC	$\text{HOCH}_2\text{COCH}_2\text{OOH} + \text{OH} \rightarrow \text{HOCH}_2\text{COCH}_2\text{O}_2$	$.6*k_{\text{CH300H_OH}}$	Taraborrelli (2014)
G4343e	TrGC	$\text{HOCH}_2\text{COCH}_2\text{OOH} + \text{OH} \rightarrow \text{HCOCOCH}_2\text{OOH} + \text{HO}_2$	$0.9295*1.60\text{E-}12*\text{EXP}(305./\text{temp})$	Taraborrelli (2014)*
G4344e	TrGC	$\text{HCOCOCH}_2\text{O}_2 \rightarrow 0.6 \text{HCOCO}_3\text{A} + 0.6 \text{HCHO} + 0.2 \text{HCOCOCHO} + 0.2 \text{HOCH}_2\text{COCHO}$	$2.00\text{E-}12*\text{R02}$	Taraborrelli (2014)
G4345e	TrGC	$\text{HCOCOCH}_2\text{O}_2 + \text{NO} \rightarrow \text{HCOCO}_3\text{A} + \text{HCHO} + \text{NO}_2$	KR02N0	Taraborrelli (2014)
G4346e	TrGC	$\text{HCOCOCH}_2\text{O}_2 + \text{HO}_2 \rightarrow 0.85 \text{HCOCOCH}_2\text{OOH} + 0.15 \text{HCOCO}_3\text{A} + 0.15 \text{HCHO} + 0.15 \text{OH}$	KR02H02*0.520	Taraborrelli (2014)
G4347e	TrGC	$\text{HCOCOCH}_2\text{O}_2 + \text{NO}_3 \rightarrow \text{HCOCO}_3\text{A} + \text{HCHO} + \text{NO}_2$	KR02N03	Taraborrelli (2014)
G4348e	TrGC	$\text{HCOCOCH}_2\text{OOH} + \text{OH} \rightarrow \text{HOOCH}_2\text{CO}_3 + \text{CO} + \text{H}_2\text{O}$	kt*fco*fco	Taraborrelli (2014)
G4349e	TrGC	$\text{HCOCOCH}_2\text{OOH} + \text{OH} \rightarrow \text{HCOCOCHO} + \text{OH} + \text{H}_2\text{O}$	ks*fsooh*fco	Taraborrelli (2014)
G4350e	TrGC	$\text{HCOCOCH}_2\text{OOH} + \text{OH} \rightarrow \text{HCOCOCH}_2\text{O}_2 + \text{H}_2\text{O}$	$0.6*k_{\text{CH300H_OH}}$	Taraborrelli (2014)
G4351e	TrGC	$\text{HCOCOCH}_2\text{OOH} + \text{NO}_3 \rightarrow \text{HOOCH}_2\text{CO}_3 + \text{CO} + \text{HNO}_3$	KN03AL*2.4	Taraborrelli (2014)
G4352e	TrGC	$\text{HCOCOCHO} + \text{OH} \rightarrow \text{HCOCO}_3\text{A} + \text{CO}$	$2*kt*fco*fo$	Taraborrelli (2014)
G4353e	TrGC	$\text{CH}_3\text{CHCO} + \text{OH} \rightarrow .72 \text{CO} + .72 \text{CH}_3\text{CHO} + .72 \text{HO}_2 + .21 \text{CH}_3\text{COCO}_2\text{H} + .07 \text{CH}_3\text{CHO} + .07 \text{HO}_2 + .07 \text{CO}_2$	$7.6\text{E-}11$	Hatakeyama et al. (1985), Taraborrelli (2014)*
G4354e	TrGC	$\text{HCOCCH}_3\text{CO} + \text{OH} \rightarrow \text{CO} + \text{CH}_3\text{COCHOH}$	$1\text{E-}10*\text{acho}$	Hatakeyama et al. (1985), Taraborrelli (2014)*
G4355e	TrGC	$\text{CH}_3\text{COCHCO} + \text{OH} \rightarrow \text{CO} + \text{CH}_3\text{COCHOH}$	$7.6\text{E-}11*\text{acoch3}$	Hatakeyama et al. (1985), Taraborrelli (2014)*
G4400	TrGC	$\text{nC}_4\text{H}_{10} + \text{OH} \rightarrow \text{LC}_4\text{H}_9\text{O}_2 + \text{H}_2\text{O}$	$1.81\text{E-}17*\text{temp*temp*EXP}(114./\text{temp})$	Atkinson (2003)*
G4401	TrGC	$\text{LC}_4\text{H}_9\text{O}_2 \rightarrow 0.254 \text{CO}_2 + 0.5552 \text{MEK} + 0.5552 \text{HO}_2 + 0.3178 \text{CH}_3\text{CHO} + 0.4448 \text{C}_2\text{H}_5\text{O}_2$	$2.5\text{E-}13*\text{R02}$	Rickard and Pascoe (2009)*
G4402	TrGC	$\text{LC}_4\text{H}_9\text{O}_2 + \text{HO}_2 \rightarrow \text{LC}_4\text{H}_9\text{OOH}$	KR02H02*0.625	Rickard and Pascoe (2009)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4403	TrGNC	LC ₄ H ₉ O ₂ + NO → 0.9172 NO ₂ + 0.233 CO ₂ + 0.5092 MEK + 0.5092 HO ₂ + 0.2915 CH ₃ CHO + 0.408 C ₂ H ₅ O ₂ + 0.0828 LC ₄ H ₉ NO ₃	KR02NO	Rickard and Pascoe (2009)*
G4404	TrGC	LC ₄ H ₉ OOH + OH → 0.2285796 LC ₄ H ₉ O ₂ + 0.7117253 MEK + 0.1193902 CO ₂ + 0.0596951 C ₂ H ₅ O ₂ + 0.7714204 OH + H ₂ O	2.636E-11	Rickard and Pascoe (2009)*
G4405e	TrGC	MVK + O ₃ → .87 MGLYOX + 0.5481 CO + 0.1392 HO ₂ + 0.1392 OH + 0.3219 HOCH ₂ OOH + .13 HCHO + 0.04680 OH + 0.04680 CO + 0.07280 CH ₃ C(O)OO + .026 CH ₃ CHO + .026 CO ₂ + .026 HCHO + .026 HO ₂ + 0.02402 MGLYOX + 0.02402 H ₂ O ₂ + 0.007176 CH ₃ COCO ₂ H	8.5E-16*EXP(-1520./temp)	Taraborrelli (2014)
G4406e	TrGC	MVK + OH → LHMVKABO ₂	2.6E-12*EXP(610./temp)	Taraborrelli (2014)*
G4413	TrGC	MEK + OH → LMEKO ₂ + H ₂ O	3.24E-18*temp*temp*EXP(414./temp)	Rickard and Pascoe (2009)*
G4414ea	TrGC	LMEKO ₂ + HO ₂ → LMEKOOH	KR02H02*0.625*rcoch2o2_ooh	Taraborrelli (2014)
G4414eb	TrGC	LMEKO ₂ + HO ₂ → 0.538 HCHO + 0.538 CO ₂ + 0.459 HOCH ₂ CH ₂ O ₂ + 0.079 C ₂ H ₅ O ₂ + 0.462 CH ₃ C(O)OO + 0.462 CH ₃ CHO + OH	KR02H02*0.625*rcoch2o2_oh	Taraborrelli (2014)
G4415	TrGNC	LMEKO ₂ + NO → 0.538 HCHO + 0.538 CO ₂ + 0.459 HOCH ₂ CH ₂ O ₂ + 0.079 C ₂ H ₅ O ₂ + 0.462 CH ₃ C(O)OO + 0.462 CH ₃ CHO + NO ₂	KR02NO	Rickard and Pascoe (2009)*
G4416	TrGC	LMEKOOH + OH → 0.40851 CH ₃ COCH ₂ O ₂ + 0.350196 BIACET + 0.807212 OH + 0.048506 C ₂ H ₅ O ₂ + 0.505522 CO ₂ + 0.192788 LMEKO ₂ + H ₂ O	3.786E-11	Rickard and Pascoe (2009)*
G4417	TrGNC	LC ₄ H ₉ NO ₃ + OH → 0.91423 MEK + 0.08577 C ₂ H ₅ O ₂ + 0.17154 CO ₂ + NO ₂ + H ₂ O	9.598E-13	Rickard and Pascoe (2009)*
G4418	TrGNC	MPAN + OH → CH ₃ COCH ₂ OH + CO + NO ₂	3.2E-11	Orlando et al. (2002)
G4419	TrGNC	MPAN → MACO ₃ + NO ₂	k_PAN_M	see note
G4420	TrGC	LMEKO ₂ → 0.538 HCHO + 0.538 CO ₂ + 0.459 HOCH ₂ CH ₂ O ₂ + 0.079 C ₂ H ₅ O ₂ + 0.462 CH ₃ C(O)OO + 0.462 CH ₃ CHO	1.483E-12*R02	Rickard and Pascoe (2009)*
G4421e	TrGC	MACR + OH → .45 MACO ₃ + .55 MACRO ₂	8.E-12*EXP(380./temp)	Orlando et al. (1999b), Taraborrelli (2014)
G4422e	TrGC	MACR + O ₃ → 0.5481 CO + 0.1392 HO ₂ + 0.1392 OH + 0.3219 HOCH ₂ OOH + .87 MGLYOX + .13 HCHO + .13 OH + .065 HCOCOCH ₂ O ₂ + .065 CO + .065 CH ₃ C(O)OO	1.36E-15*EXP(-2112./temp)	Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4423	TrGNC	MACR + NO ₃ → MACO3 + HNO ₃	KN03AL*2.0	Rickard and Pascoe (2009)
G4424e	TrGC	MACO3 → .7 MACO2 + .3 MACO2H	1.00E-11*R02	Taraborrelli (2014)
G4425e	TrGC	MACO3 + HO ₂ → MACO2 + OH	KAPH02*rco3_oh	Taraborrelli (2014)
G4425et2	TrGC	MACO3 + HO ₂ → MACO3H	KAPH02*rco3_ooh	Taraborrelli (2014)
G4425et3	TrGC	MACO3 + HO ₂ → MACO2H + O ₃	KAPH02*rco3_o3	Taraborrelli (2014)
G4426e	TrGNC	MACO3 + NO → MACO2 + NO ₂	8.70E-12*EXP(290./temp)	Taraborrelli (2014)
G4427	TrGNC	MACO3 + NO ₂ → MPAN	k_CH3C03_N02	Rickard and Pascoe (2009)
G4428e	TrGNC	MACO3 + NO ₃ → MACO2 + NO ₂	KR02N03*1.60	Taraborrelli (2014)
G4429e	TrGC	MACRO2 → .7 CH ₃ COCH ₂ OH + .7 HCHO + .7 HO ₂ + .3 MACROH	9.20E-14*R02	Taraborrelli (2014)
G4430e	TrGC	MACRO2 + HO ₂ → MACRO + OH	KR02H02*0.625*rcoch2o2_oh	Taraborrelli (2014)
G4430et2	TrGC	MACRO2 + HO ₂ → MACROOH	KR02H02*0.625*rcoch2o2_ooh	Taraborrelli (2014)
G4431e	TrGNC	MACRO2 + NO → .85 MACRO + .85 NO ₂ + .15 MACRN	KR02N0	Taraborrelli (2014)
G4432e	TrGNC	MACRO2 + NO ₃ → MACRO + NO ₂	KR02N03	Taraborrelli (2014)
G4433ea	TrGC	MACROOH + OH → MACRO2	0.6*k_CH300H_OH	Taraborrelli (2014)
G4433eb	TrGC	MACROOH + OH → CO + OH + CH ₃ COCH ₂ OH	kt*fo*ftch2oh*falk	Taraborrelli (2014)
G4433ec	TrGC	MACROOH + OH → CO + MGLYOX + HO ₂	ks*fsch*fpch2oh + krohro	Taraborrelli (2014)
G4434e	TrGC	MACROH + OH → CH ₃ COCH ₂ OH + CO + HO ₂	kt*fo*ftch2oh*falk	Taraborrelli (2014)
G4434et2	TrGC	MACRO → .885 CH ₃ COCH ₂ OH + .885 CO + .115 MGLYOX + .115 HCHO + HO ₂	KDEC	Taraborrelli (2014)
G4435e	TrGC	MACO2H + OH → CH ₃ COCH ₂ OH + HO ₂ + CO ₂	(kadt+kadp)*aco2h+kco2h	Taraborrelli (2014)
G4436e	TrGC	MACO3H + OH → CH ₃ COCH ₂ OH + CO ₂ + OH	0.6*k_CH300H_OH+(kadt+kadp)*aco2h	Taraborrelli (2014)
G4437e	TrGC	LHMKABO2 → .024 CO2H3CHO + .072 CH3COCHOH + .072 HCHO + .5280 CH ₃ C(O)OO + .5280 HOCH ₂ CHO + .176 BIACETOH + .2 HO12CO3C4	1.014E-12*R02	Taraborrelli (2014)*
G4438e	TrGC	LHMKABO2 + HO ₂ → OH + HOCH ₂ CHO + CH ₃ C(O)OO	KR02H02*0.625*.88*rcoch2o2_oh	Taraborrelli (2014)
G4438et2	TrGC	LHMKABO2 + HO ₂ → LHMKABOOH	KR02H02*0.625*(.12+.88*rcoch2o2_ooh)	Taraborrelli (2014)
G4439ea	TrGNC	LHMKABO2 + NO → .12 CH3COCHOH + .88 HOCH ₂ CHO + .88 CH ₃ C(O)OO + .12 HCHO + NO ₂	KR02N0*(1.-0.11)	Taraborrelli (2014)*
G4439eb	TrGNC	LHMKABO2 + NO → HMVKNO3	KR02N0*0.11	Taraborrelli (2014)
G4440e	TrGNC	LHMKABO2 + NO ₃ → .12 MGLYOX + .88 HOCH ₂ CHO + .88 CH ₃ C(O)OO + .12 HCHO + .12 HO ₂ + NO ₂	KR02N03	Taraborrelli (2014)*

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4441e	TrGC	LHMKABOOH + OH \rightarrow .12 CO ₂ H ₃ CHO + .88 BIACETOH + OH	$0.6 \cdot k_{\text{CH300H_OH}} + .12 \cdot k_{\text{fsooh}} \cdot f_{\text{pch2oh}} + .88 \cdot k_{\text{ftooh}} \cdot f_{\text{pch2oh}} \cdot f_{\text{co}}$	Taraborrelli (2014)*
G4449e	TrGC	CO ₂ H ₃ CHO + OH \rightarrow CO ₂ H ₃ CO ₃	$k_{\text{tfo}} \cdot f_{\text{alk}}$	Taraborrelli (2014)
G4449et2	TrGC	CO ₂ H ₃ CHO + OH \rightarrow CH ₃ COCOCHO + HO ₂ + H ₂ O	$k_{\text{tfc}} \cdot f_{\text{toh}} \cdot f_{\text{cho}}$	Taraborrelli (2014)
G4450	TrGNC	CO ₂ H ₃ CHO + NO ₃ \rightarrow CO ₂ H ₃ CO ₃ + HNO ₃	KN03AL*4.0	Rickard and Pascoe (2009)
G4451e	TrGC	CO ₂ H ₃ CO ₃ \rightarrow CH ₃ COCHOH + CO ₂	1.00E-11*R02	Taraborrelli (2014)
G4452e	TrGC	CO ₂ H ₃ CO ₃ + HO ₂ \rightarrow OH + CH ₃ COCHOH + CO ₂	KAPH02*rco3_oh	Taraborrelli (2014)
G4452et2	TrGC	CO ₂ H ₃ CO ₃ + HO ₂ \rightarrow CO ₂ H ₃ CO ₂ H + O ₃	KAPH02*rco3_o3	Taraborrelli (2014)
G4452et3	TrGC	CO ₂ H ₃ CO ₃ + HO ₂ \rightarrow CO ₂ H ₃ CO ₃ H	KAPH02*rco3_ooh	Taraborrelli (2014)
G4453e	TrGNC	CO ₂ H ₃ CO ₃ + NO \rightarrow CH ₃ COCHOH + NO ₂ + CO ₂	KAPNO	Taraborrelli (2014)
G4454e	TrGNC	CO ₂ H ₃ CO ₃ + NO ₃ \rightarrow CH ₃ COCHOH + NO ₂ + CO ₂	KR02NO3*1.60	Taraborrelli (2014)
G4455	TrGC	CO ₂ H ₃ CO ₃ H + OH \rightarrow 0.5127 CO ₂ H ₃ CO ₃ + 0.4873 CH ₃ C(O)OO + 0.4873 CO + 0.4873 CO ₂ + 0.4873 OH	$k_{\text{tfc}} \cdot f_{\text{co2h}} \cdot f_{\text{co}} \cdot f_{\text{toh}} + 0.6 \cdot k_{\text{CH300H_OH}}$	Taraborrelli (2014)*
G4455t2	TrGC	CO ₂ H ₃ CO ₂ H + OH \rightarrow CH ₃ COCOCO ₂ H + HO ₂	$k_{\text{tfc}} \cdot f_{\text{co2h}} \cdot f_{\text{co}} \cdot f_{\text{toh}} + k_{\text{co2h}}$	Taraborrelli (2014)
G4456a	TrGC	HO ₁₂ CO ₃ C ₄ + OH \rightarrow BIACETOH + HO ₂	$k_{\text{tftoh}} \cdot f_{\text{alk}} \cdot f_{\text{co}}$	Taraborrelli (2014)
G4456b	TrGC	HO ₁₂ CO ₃ C ₄ + OH \rightarrow CO ₂ H ₃ CHO + HO ₂	$k_{\text{sfsoh}} \cdot f_{\text{alk}}$	Taraborrelli (2014)
G4457e	TrGC	MACO ₂ \rightarrow .65 CH ₃ O ₂ + .65 CO + .65 HCHO + .35 OH + .35 CH ₃ COCH ₂ O ₂ + CO ₂	KDEC	Taraborrelli (2014)
G4458e	TrGC	LHMKABO ₂ \rightarrow .88 MGLYOX + .88 HCHO + .12 HOOCH ₂ CHO + .12 CH ₃ C(O)OO + OH	KHSD	Taraborrelli (2014)
G4459e	TrGNC	MACRO ₂ \rightarrow MGLYOX + HCHO + OH	KHSB	Taraborrelli (2014)
G4460e	TrGNC	HMVKNO ₃ + OH \rightarrow .7 MGLYOX + .7 HCOOH + .7 NO ₃ + .3 CO ₂ H ₃ CHO + .3 NO ₂ + H ₂ O	5.6E-12	Taraborrelli (2014)
G4461e	TrGC	MACRN + OH \rightarrow .08 CH ₃ COOH + .08 HCHO + .15 NO ₃ + .07 HCOOH + .07 MGLYOX + .85 CH ₃ COCH ₂ OH + .85 NO ₃ + .93 CO ₂ + H ₂ O	5.E-11	Taraborrelli (2014)
G4462e	TrGC	EZCH ₃ CO ₂ CHCHO \rightarrow .9 CH ₃ COCHCO + .1 CH ₃ C(O)OO + .01 GLYOX + .18 CO + .09 HO ₂ + OH	k16HS	Taraborrelli (2014)*
G4463e	TrGC	EZCHOCCH ₃ CHO ₂ \rightarrow HCOCCH ₃ CO + OH	K16HS	Taraborrelli (2014)*
G4500e	TrGC	C ₅ H ₈ + O ₃ \rightarrow .3508 MACR + 0.01518 MACO ₂ H + .2440 MVK + .7085 HCHO + .11 HOCH ₂ OOH + .1275 C ₃ H ₆ + .1575 CH ₃ C(O)OO + .0510 CH ₃ O ₂ + 0.2625 HO ₂ + .27 OH + .09482 H ₂ O ₂ + .255 CO ₂ + .522 CO + 0.07182 HCHO + .03618 HCOCH ₂ O ₂ + .01782 CO	$1.03\text{E-}14 \cdot \text{EXP}(-1995./\text{temp})$	Taraborrelli (2014)*

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4501e	TrGC	$C_5H_8 + OH \rightarrow .63 \text{ ISOPAB} + .30 \text{ ISOPCD} + .07 \text{ LISOPEFO2}$	$2.7E-11*EXP(390./temp)*(1.-iseg)$	Taraborrelli (2014)*
G4502	TrGNC	$C_5H_8 + NO_3 \rightarrow NISOP02$	$3.15E-12*EXP(-450./temp)$	Rickard and Pascoe (2009)
G4503e	TrGC	$ISOPAB + O_2 \rightarrow LISOPACO2$	$5.530E-13$	Taraborrelli (2014)*
G4504e	TrGC	$ISOPAB + O_2 \rightarrow ISOPBO2$	$3.E-12$	Taraborrelli (2014)*
G4505e	TrGC	$ISOPCD + O_2 \rightarrow DLISOPACO2$	$6.780E-13$	Taraborrelli (2014)*
G4506e	TrGC	$ISOPCD + O_2 \rightarrow ISOPDO2$	$3.E-12$	Taraborrelli (2014)*
G4507e	TrGC	$LISOPACO2 \rightarrow ISOPAB + O_2$	$3.1E12*exp(-7900./temp)*.6+$ $7.8E13*exp(-8600./temp)*.4$	Taraborrelli (2014)*
G4508e	TrGC	$ISOPBO2 \rightarrow ISOPAB + O_2$	$3.7E14*exp(-9570./temp)$ $+4.2E14*exp(-9970./temp)$	Taraborrelli (2014)*
G4509e	TrGC	$DLISOPACO2 \rightarrow ISOPCD + O_2$	$5.65E12*exp(-8410./temp)*.42+$ $1.4E14*exp(-9110./temp)*.58$	Taraborrelli (2014)*
G4510e	TrGC	$ISOPDO2 \rightarrow ISOPCD + O_2$	$5.0E14*exp(-10120./temp)$ $+8.25E14*exp(-10220/temp)$	Taraborrelli (2014)*
G4511e	TrGC	$LISOPACO2 \rightarrow ZCODC23DBCOOH + HO_2$	K16HS	Taraborrelli (2014)*
G4512e	TrGC	$DLISOPACO2 \rightarrow ZCODC23DBCOOH + HO_2$	K16HS	Taraborrelli (2014)*
G4513et3	TrGC	$LISOPACO2 \rightarrow .9 \text{ LHC4ACCHO} + .8 \text{ HO}_2 + .1 \text{ ISOPAOH}$	$2.4E-12*R02$	Rickard and Pascoe (2009)
G4514t2	TrGC	$LISOPACO2 + HO_2 \rightarrow LISOPACOOH$	$.706*KR02H02$	Rickard and Pascoe (2009)
G4515et2	TrGNC	$LISOPACO2 + NO \rightarrow 0.95 \text{ LHC4ACCHO} + 0.95 \text{ HO}_2 + 0.95 \text{ NO}_2 + .05 \text{ LISOPACNO3}$	KR02N0	Lockwood et al. (2010),Taraborrelli (2014)
G4506et3	TrGNC	$LISOPACO2 + NO_3 \rightarrow \text{LHC4ACCHO} + HO_2 + NO_2$	KR02N03	Rickard and Pascoe (2009)
G4507et3	TrGC	$DLISOPACO2 \rightarrow .9 \text{ LHC4ACCHO} + .8 \text{ HO}_2 + .1 \text{ ISOPAOH}$	$2.4E-12*R02$	Rickard and Pascoe (2009)
G4511et3	TrGC	$DLISOPACO2 + HO_2 \rightarrow LISOPACOOH$	$.706*KR02H02$	Rickard and Pascoe (2009)
G4512et3	TrGNC	$DLISOPACO2 + NO \rightarrow 0.95 \text{ LHC4ACCHO} + 0.95 \text{ HO}_2 + 0.95 \text{ NO}_2 + .05 \text{ LISOPACNO3}$	KR02N0	Lockwood et al. (2010),Taraborrelli (2014)
G4513et4	TrGNC	$DLISOPACO2 + NO_3 \rightarrow \text{LHC4ACCHO} + HO_2 + NO_2$	KR02N03	Rickard and Pascoe (2009)
G4514e	TrGC	$LISOPACOOH + OH \rightarrow LISOPACO2$	$0.6*k_CH300H_OH$	Taraborrelli (2014)
G4514et2	TrGC	$LISOPACOOH + OH \rightarrow ZCODC23DBCOOH + HO_2$	$ks*fallyl*fsoh$	Taraborrelli (2014)
G4514et3	TrGC	$LISOPACOOH + OH \rightarrow \text{LHC4ACCHO} + OH$	$ks*fsooh*fallyl+ \text{ krohro}$	Taraborrelli (2014)
G4514et4	TrGC	$LISOPACOOH + OH \rightarrow \text{IEPOX} + OH$	$(kadt+kads)*ach2oh*ach2ooh$	Taraborrelli (2014)
G4515	TrGC	$ISOPAOH + OH \rightarrow \text{LHC4ACCHO} + HO_2$	$(kadt+kads)*ach2oh*ach2oh+$ $ks*fsoh*fallyl+krohro$	Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4516e	TrGNC	LISOPACNO3 + OH \rightarrow LISOPACNO3O2	9.5E-11	Paulot et al. (2009a),Taraborrelli (2014)
G4517e	TrGC	ISOPBO2 \rightarrow .8 MVK + .8 HCHO + .8 HO ₂ + .2 ISOPBOH	8.E-13*R02	Rickard and Pascoe (2009)
G4518	TrGC	ISOPBO2 + HO ₂ \rightarrow ISOPBOOH	.706*KR02H02	Rickard and Pascoe (2009)
G4519e	TrGNC	ISOPBO2 + NO \rightarrow .947 MVK + .947 HCHO + .947 HO ₂ + .947 NO ₂ + .053 ISOPBNO3	KR02N0	Lockwood et al. (2010),Taraborrelli (2014)
G4520e	TrGNC	ISOPBO2 + NO ₃ \rightarrow MVK + .75 HCHO + .75 HO ₂ + .25 CH ₃ O ₂ + NO ₂	KR02N03	Rickard and Pascoe (2009)
G4521ea	TrGC	ISOPBOOH + OH \rightarrow IEPOX + OH	(kads+kadp)*ach2ooh	Paulot et al. (2009b),Taraborrelli (2014)
G4521eb	TrGC	ISOPBOOH + OH \rightarrow ISOPBO2	0.6*k_CH300H_OH	Taraborrelli (2014)
G4521ec	TrGC	ISOPBOOH + O ₃ \rightarrow 0.1368 MACROOH + 0.1368 H ₂ O ₂ + 0.2280 HO ₂ + 0.4332 CH ₃ COCH ₂ OH + 0.2280 CO ₂ + 0.6384 OH + 0.2052 CO + .57 HCHO + .43 MACROOH + 0.06880 HO ₂ + 0.06880 OH + 0.2709 CO + 0.1591 HOCH ₂ OOH	1.E-17	Taraborrelli (2014)*
G41911	TrGC	ISOPBOOH + OH \rightarrow MGLYOX + HOCH ₂ CHO	krohro+ks*falk*fsoh	Taraborrelli (2014)
G4522e	TrGC	ISOPBOH + OH \rightarrow MVK + .75 HCHO + .75 HO ₂ + .25 CH ₃ O ₂	ks*falk*fsoh+(kadp+kads)*ach2oh	Taraborrelli (2014)
G4523e	TrGNC	ISOPBNO3 + OH \rightarrow ISOPBDNO3O2	1.3E-11	Paulot et al. (2009a),Taraborrelli (2014)
G4524	TrGC	ISOPDO2 \rightarrow .8 MACR + .8 HCHO + .8 HO ₂ + .1 HCOC5 + .1 ISOPDOH	2.9E-12*R02	Rickard and Pascoe (2009)
G4525	TrGC	ISOPDO2 + HO ₂ \rightarrow ISOPDOOH	.706*KR02H02	Rickard and Pascoe (2009)
G4526e	TrGNC	ISOPDO2 + NO \rightarrow .85 MACR + .85 HCHO + .85 HO ₂ + .85 NO ₂ + .15 ISOPDNO3	KR02N0	Lockwood et al. (2010),Taraborrelli (2014)
G4527	TrGNC	ISOPDO2 + NO ₃ \rightarrow MACR + HCHO + HO ₂ + NO ₂	KR02N03	Rickard and Pascoe (2009)
G4528ea	TrGC	ISOPDOOH + OH \rightarrow IEPOX + OH	(kadt+kadp)*ach2ooh	Paulot et al. (2009b),Taraborrelli (2014)
G4528eb	TrGC	ISOPDOOH + OH \rightarrow ISOPDO2	0.6*k_CH300H_OH	Taraborrelli (2014)
G4528ec	TrGC	ISOPDOOH + OH \rightarrow HCOC5 + OH	kt*ftooh*fallyl*fpch2oh	Taraborrelli (2014)
G4528ed	TrGC	ISOPDOOH + OH \rightarrow CH ₃ COCH ₂ OH + GLYOX + OH	ks*fpch2oh*fsoh	Taraborrelli (2014)
G45222	TrGC	ISOPDOOH + O ₃ \rightarrow 1.393 OH + BIACETOH + .67 HCHO + 0.05280 HO ₂ + 0.2079 CO + 0.1221 HOCH ₂ OOH	1.E-17	Taraborrelli (2014)*

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4529e	TrGC	ISOPDOH + OH \rightarrow HCOC5 + HO ₂	2.*krohro+(kt*ftoh*fallyl+ks*fsoh) *fpch2oh+(kadt+kadp)*ach2oh	Taraborrelli (2014)
G4530e	TrGNC	ISOPDNO3 + OH \rightarrow ISOPBDNO3O2	1.3E-11	Paulot et al. (2009a), Taraborrelli (2014)
G4531	TrGNC	NISOPO2 \rightarrow .8 NC4CHO + .6 HO ₂ + .2 LISOPACNO3	1.3E-12*R02	Rickard and Pascoe (2009)
G4532	TrGNC	NISOPO2 + HO ₂ \rightarrow NISOPOOH	.706*KR02H02	Rickard and Pascoe (2009)
G4533	TrGNC	NISOPO2 + NO \rightarrow NC4CHO + HO ₂ + NO ₂	KR02N0	Rickard and Pascoe (2009)
G4534	TrGNC	NISOPO2 + NO ₃ \rightarrow NC4CHO + HO ₂ + NO ₂	KR02N03	Rickard and Pascoe (2009)
G4535	TrGNC	NISOPOOH + OH \rightarrow NC4CHO + OH	1.03E-10	Rickard and Pascoe (2009)
G4536	TrGNC	NC4CHO + OH \rightarrow LNISO3	4.16E-11	Rickard and Pascoe (2009)
G4537e	TrGNC	NC4CHO + O ₃ \rightarrow .27 NOA + .027 HCOCO ₂ H + .0162 GLYOX + .0162 H ₂ O ₂ + .1458 HCOCO3A + .0405 HCOOH + .0405 CO + .8758 OH + .365 MGLYOX + .73 NO ₂ + 0.7705 HCHO + .4055 CO ₂ + .73 GLYOX	2.40E-17	Taraborrelli (2014)*
G4538	TrGNC	NC4CHO + NO ₃ \rightarrow LNISO3 + HNO ₃	KN03AL*4.25	Rickard and Pascoe (2009)
G4539	TrGNC	LNISO3 + HO ₂ \rightarrow LNISOOH	.5*.706*KR02H02 + .5*KAPH02	Rickard and Pascoe (2009)
G4540e	TrGNC	LNISO3 + NO \rightarrow NOA + .5 HOCHCHO + .5 CO + .5 HO ₂ + NO ₂ + .5 CO ₂	.5*KAPNO +.5*KR02N0	Rickard and Pascoe (2009)
G4541e	TrGNC	LNISO3 + NO ₃ \rightarrow NOA + .5 HOCHCHO + .5 CO + .5 HO ₂ + NO ₂ + .5 CO ₂	1.3*KR02N03	Rickard and Pascoe (2009)
G4542	TrGNC	LNISOOH + OH \rightarrow LNISO3	2.65E-11	Rickard and Pascoe (2009)
G4543e	TrGC	LHC4ACCHO + OH \rightarrow LC578O2	(kadtertprim+kads)*acho*ach2oh	Taraborrelli (2014)
G4543et2	TrGC	LHC4ACCHO + OH \rightarrow LHC4ACCO3	kcho	Taraborrelli (2014)
G4543et3	TrGC	LHC4ACCHO + OH \rightarrow ZCODC23DBCOD + HO ₂	ks*fsoh*fallyl	Taraborrelli (2014)
G4544	TrGC	LHC4ACCHO + O ₃ \rightarrow .2225 CH ₃ C(O)OO + .89 CO + .0171875 HOCH ₂ CO ₂ H + .075625 H ₂ O ₂ + .0171875 HCOCO ₂ H + .2775 CH ₃ COCH ₂ OH + .6675 HO ₂ + .2603125 GLYOX + .2225 HCHO + .89 OH + .2603125 HOCH ₂ CHO + .5 MGLYOX	2.40E-17	Rickard and Pascoe (2009)
G4545	TrGNC	LHC4ACCHO + NO ₃ \rightarrow LHC4ACCO3 + HNO ₃	KN03AL*4.25	Rickard and Pascoe (2009)
G4546e	TrGC	LC578O2 \rightarrow .25 CH ₃ COCH ₂ OH + .75 MGLYOX + .25 HOCHCHO + .75 HOCH ₂ CHO + .75 HO ₂	9.20E-14*R02	Rickard and Pascoe (2009)
G4547e	TrGC	LC578O2 + HO ₂ \rightarrow MGLYOX + HOCH ₂ CHO + OH	KR02H02*0.706*rcoch2o2_oh	Rickard and Pascoe (2009)
G4547et2	TrGC	LC578O2 + HO ₂ \rightarrow LC578OOH	KR02H02*0.706*rcoch2o2_ooh	Rickard and Pascoe (2009)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4548e	TrGNC	$\text{LC578O2} + \text{NO} \rightarrow .25 \text{CH}_3\text{COCH}_2\text{OH} + .75 \text{MGLYOX} + .25 \text{HOCHCHO} + .75 \text{HOCH}_2\text{CHO} + .75 \text{HO}_2 + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009)
G4549e	TrGNC	$\text{LC578O2} + \text{NO}_3 \rightarrow .25 \text{CH}_3\text{COCH}_2\text{OH} + .75 \text{MGLYOX} + .25 \text{HOCHCHO} + .75 \text{HOCH}_2\text{CHO} + .75 \text{HO}_2 + \text{NO}_2$	KR02N03	Rickard and Pascoe (2009)
G4586e	TrGC	$\text{LC578O2} \rightarrow .25 \text{CH}_3\text{COCH}_2\text{OH} + .75 \text{MGLYOX} + .25 \text{HOCH}_2\text{CHO} + .75 \text{HOCH}_2\text{CHO} + \text{HO}_2 + \text{OH}$	KHSB	Taraborrelli (2014)
G4550e	TrGC	$\text{LC578OOH} + \text{OH} \rightarrow \text{LC578O2}$	$0.6 * k_{\text{CH300H_OH}}$	Taraborrelli (2014)*
G4550et2	TrGC	$\text{LC578OOH} + \text{OH} \rightarrow \text{C1ODC2OOHC4OD} + \text{HO}_2$	$kt * fo * ftch2oh * falk +$ $kt * ftoh * fpch2oh * fpch2oh +$ $ks * fsoh * fpch2oh$	Taraborrelli (2014)*
G4551e	TrGC	$\text{LHC4ACCO3} \rightarrow .3 \text{LHC4ACCO2H} + .7 \text{OH} + .35 \text{MACRO2} + .35 \text{LHMKABO2} + .7 \text{CO}_2$	$1.00\text{E-}11 * R02$	Taraborrelli (2014)*
G4552e	TrGC	$\text{LHC4ACCO3} + \text{HO}_2 \rightarrow 2 \text{OH} + .5 \text{MACRO2} + .5 \text{LHMKABO2} + \text{CO}_2$	$\text{KAPH02} * rco3_oh$	Taraborrelli (2014)*
G4552et2	TrGC	$\text{LHC4ACCO3} + \text{HO}_2 \rightarrow \text{LHC4ACCO3H}$	$\text{KAPH02} * rco3_ooh$	Taraborrelli (2014)
G4552et3	TrGC	$\text{LHC4ACCO3} + \text{HO}_2 \rightarrow \text{LHC4ACCO2H} + \text{O}_3$	$\text{KAPH02} * rco3_o3$	Taraborrelli (2014)
G4553e	TrGNC	$\text{LHC4ACCO3} + \text{NO} \rightarrow .5 \text{MACRO2} + .5 \text{LHMKABO2} + \text{NO}_2 + \text{CO}_2$	KAPN0	Taraborrelli (2014)*
G4554	TrGNC	$\text{LHC4ACCO3} + \text{NO}_2 \rightarrow \text{LC5PAN1719}$	$k_{\text{CH3C03_N02}}$	Rickard and Pascoe (2009)
G4555e	TrGNC	$\text{LHC4ACCO3} + \text{NO}_3 \rightarrow .5 \text{MACRO2} + .5 \text{LHMKABO2} + \text{NO}_2 + \text{CO}_2$	$1.6 * \text{KR02N03}$	Taraborrelli (2014)*
G4556e	TrGC	$\text{LHC4ACCO2H} + \text{OH} \rightarrow \text{OH} + .5 \text{MACRO2} + .5 \text{LHMKABO2} + \text{CO}_2$	$2.52\text{E-}11$	Taraborrelli (2014)
G4557	TrGC	$\text{LHC4ACCO3H} + \text{OH} \rightarrow \text{LHC4ACCO3}$	$2.88\text{E-}11$	Rickard and Pascoe (2009)
G4558	TrGNC	$\text{LC5PAN1719} \rightarrow \text{LHC4ACCO3} + \text{NO}_2$	$k_{\text{PAN_M}}$	Rickard and Pascoe (2009)
G4559	TrGNC	$\text{LC5PAN1719} + \text{OH} \rightarrow .5 \text{MACROH} + .5 \text{HO12CO3C4} + \text{CO} + \text{NO}_2$	$2.52\text{E-}11$	Rickard and Pascoe (2009)
G4560a	TrGC	$\text{HCOC5} + \text{OH} \rightarrow \text{C59O2}$	$3.81\text{E-}11$	Rickard and Pascoe (2009)
G4560eb	TrGC	$\text{HCOC5} + \text{O}_3 \rightarrow \text{BIACETOH} + .335 \text{H}_2\text{O}_2 + 0.67 \text{HCHO} + 0.2079 \text{CO} + 0.1221 \text{HOCH2OOH} + 0.05280 \text{OH}$	$7.51\text{E-}16 * \text{EXP}(-1521./\text{temp})$	Taraborrelli (2014)
G4561	TrGC	$\text{C59O2} \rightarrow \text{CH}_3\text{COCH}_2\text{OH} + \text{HOCH2CO}$	$9.20\text{E-}14 * R02$	Taraborrelli (2014)
G4562e	TrGC	$\text{C59O2} + \text{HO}_2 \rightarrow \text{OH} + \text{CH}_3\text{COCH}_2\text{OH} + \text{HOCH2CO}$	$\text{KR02H02} * 0.706 * rcoch2o2_oh$	Taraborrelli (2014)
G4562et2	TrGC	$\text{C59O2} + \text{HO}_2 \rightarrow \text{C59OOH}$	$\text{KR02H02} * 0.706 * rcoch2o2_ooh$	Taraborrelli (2014)
G4563	TrGNC	$\text{C59O2} + \text{NO} \rightarrow \text{CH}_3\text{COCH}_2\text{OH} + \text{HOCH2CO} + \text{NO}_2$	KR02N0	Taraborrelli (2014)
G4564	TrGNC	$\text{C59O2} + \text{NO}_3 \rightarrow \text{CH}_3\text{COCH}_2\text{OH} + \text{HOCH2CO} + \text{NO}_2$	KR02N03	Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4565	TrGC	$\text{C59OOH} + \text{OH} \rightarrow \text{C59O2}$	9.7E-12	Taraborrelli (2014)
G4566e	TrGC	$\text{IEPOX} + \text{OH} \rightarrow \text{LC578O2} + \text{H}_2\text{O}$	$5.78\text{E-}11 * \text{EXP}(-400/\text{temp})$	Paulot et al. (2009b), Taraborrelli (2014)
G4567e	TrGC	$\text{ISOPBO2} \rightarrow \text{MVK} + \text{HCHO} + \text{OH}$	KHSB	Taraborrelli (2014)
G4568e	TrGC	$\text{ISOPDO2} \rightarrow \text{MACR} + \text{HCHO} + \text{OH}$	KHSD	Taraborrelli (2014)
G4577ea	TrGC	$\text{ZCODC23DBCOOH} + \text{OH} \rightarrow .6 \text{ C1ODC2O2C4OOH} + .4 \text{ C1OOHC2O2C4OD}$	$\text{kadt} * \text{acho} * \text{ach2ooh}$	Taraborrelli (2014)
G4577eb	TrGC	$\text{ZCODC23DBCOOH} + \text{OH} \rightarrow .6 \text{ C1ODC3O2C4OOH} + .4 \text{ C1OOHC3O2C4OD}$	$\text{kads} * \text{acho} * \text{ach2ooh}$	Taraborrelli (2014)
G4577e	TrGC	$\text{ZCODC23DBCOOH} + \text{OH} \rightarrow \text{ZCO3HC23DBCOD}$	$\text{kt} * \text{fo} * \text{falk} + 0.6 * \text{k_CH3OOH_OH}$	Taraborrelli (2014)
G4577et2	TrGC	$\text{ZCODC23DBCOOH} + \text{OH} \rightarrow \text{OH} + \text{ZCODC23DBCOD}$	$\text{ks} * \text{fsooh} * \text{fallyl}$	Taraborrelli (2014)
G4577et3	TrGC	$\text{ZCODC23DBCOOH} + \text{O}_3 \rightarrow .4672 \text{ OH} + .2336 \text{ HCOCOCH2O2} + .2336 \text{ CO} + .2336 \text{ CH}_3\text{C(O)OO} + .4672 \text{ HOOCH2CHO} + .1728 \text{ MGLYOX} + .1901 \text{ OH} + .0864 \text{ GLYOX} + .02765 \text{ HOOCH2CHO} + .02765 \text{ H}_2\text{O}_2 + .02592 \text{ CH}_3\text{OOH} + .02592 \text{ CO}_2 + .01037 \text{ HCOCO3A} + .01555 \text{ HOCH2OOH} + .01555 \text{ CO} + .006912 \text{ HOOCH2CO3} + .2628 \text{ OH} + .1314 \text{ MGLYOX} + .1314 \text{ OH} + .1314 \text{ HCOCOCH2OOH} + 0.2628 \text{ GLYOX} + .0972 \text{ CH}_3\text{COCH}_2\text{O}_2\text{H} + .00972 \text{ HCOCO}_2\text{H} + .005832 \text{ GLYOX} + .005832 \text{ H}_2\text{O}_2 + .05249 \text{ OH} + .05249 \text{ HCOCO3A} + .01458 \text{ HCHO} + .01458 \text{ CO}_2 + .01458 \text{ HCOOH} + .01458 \text{ CO}$	2.4E-17	Taraborrelli (2014)*
G4578e	TrGC	$\text{C1OOHC2O2C4OD} \rightarrow .78 \text{ CH}_3\text{COCH}_2\text{O}_2\text{H} + .78 \text{ HOCHCHO} + .22 \text{ CO2H3CHO} + .22 \text{ HCHO} + .22 \text{ OH}$	$8.00\text{E-}13 * \text{R02}$	Taraborrelli (2014)
G4579e	TrGC	$\text{C1OOHC2O2C4OD} + \text{NO} \rightarrow .78 \text{ CH}_3\text{COCH}_2\text{O}_2\text{H} + .78 \text{ HOCHCHO} + .22 \text{ CO2H3CHO} + .22 \text{ HCHO} + .22 \text{ OH} + \text{NO}_2$	KR02N0	Taraborrelli (2014)
G4580e	TrGC	$\text{C1OOHC2O2C4OD} + \text{HO}_2 \rightarrow \text{C1OOHC2OOHC4OD}$	$\text{KR02H02} * 0.706$	Taraborrelli (2014)
G4580ea	TrGC	$\text{C1OOHC2O2C4OD} \rightarrow \text{CH}_3\text{COCH}_2\text{O}_2\text{H} + \text{GLYOX} + \text{OH}$	KHSB	Taraborrelli (2014)
G4581e	TrGC	$\text{C1ODC2O2C4OOH} \rightarrow \text{OH} + \text{C1ODC2OOHC4OD}$	K15HSDHB	Taraborrelli (2014)
G4581et2	TrGC	$\text{C1OOHC2OOHC4OD} + \text{OH} \rightarrow \text{C1ODC2OOHC4OD} + \text{OH}$	$\text{ks} * \text{fsooh} * \text{fpch2oh}$	Taraborrelli (2014)
G4581et3	TrGC	$\text{C1OOHC2OOHC4OD} + \text{OH} \rightarrow \text{CH}_3\text{COCH}_2\text{O}_2\text{H} + \text{OH} + 2 \text{ CO} + 2 \text{ HO}_2$	$\text{kt} * \text{ftoh} * \text{fpch2oh} * \text{fpch2oh}$	Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4581et4	TrGC	$\text{C1OOHC2OOHC4OD} + \text{OH} \rightarrow \text{C1OOHC2O2C4OD}$	$0.6 \cdot k_{\text{CH3OOH_OH}}$	Taraborrelli (2014)
G4581et6	TrGC	$\text{C1ODC3O2C4OOH} \rightarrow \text{MGLYOX} + \text{HOOCH2CHO} + \text{HO}_2$	$2.90\text{E-}12 \cdot R_{02}$	Taraborrelli (2014)
G4581et7	TrGC	$\text{C1ODC3O2C4OOH} + \text{NO} \rightarrow \text{MGLYOX} + \text{HOOCH2CHO} + \text{HO}_2 + \text{NO}_2$	K_{R02N0}	Taraborrelli (2014)
G4581et8	TrGC	$\text{C1ODC3O2C4OOH} + \text{HO}_2 \rightarrow .5 \text{ CH}_3\text{C(O)OO} + .5 \text{ CO} + .5 \text{ MGLYOX} + .5 \text{ HO}_2 + \text{HOOCH2CO3}$	$K_{R02H02} \cdot 0.706$	Taraborrelli (2014)*
G4581et9	TrGC	$\text{C1ODC3O2C4OOH} \rightarrow \text{MGLYOX} + \text{OH} + \text{HOOCH2CHO}$	KHSD	Taraborrelli (2014)
G4581et10	TrGC	$\text{C1OOHC3O2C4OD} \rightarrow .625 \text{ MGLYOX} + 2 \text{ CO} + 1.625 \text{ HO}_2 + .375 \text{ CH}_3\text{C(O)OO} + .375 \text{ CO}_2 + \text{OH}$	$K_{15HSDHB}$	Taraborrelli (2014)*
G4582e	TrGC	$\text{LHC4ACCO3} \rightarrow \text{ZCO3HC23DBCOD} + \text{HO}_2$	K_{16HS}	Taraborrelli (2014)*
G4583e	TrGC	$\text{ZCODC23DBCOD} + \text{OH} \rightarrow \text{C1ODC2O2C4OD}$	$2 \cdot kt \cdot fo \cdot falk + (kadt + kads) \cdot acho \cdot acho$	Taraborrelli (2014)
G4584e	TrGC	$\text{C1ODC2O2C4OD} + \text{HO}_2 \rightarrow \text{OH} + \text{MGLYOX} + \text{HOCHCHO}$	$K_{R02H02} \cdot 0.706 \cdot r_{\text{coch2o2_oh}}$	Taraborrelli (2014)
G4584et2	TrGC	$\text{C1ODC2O2C4OD} + \text{HO}_2 \rightarrow \text{C1ODC2OOHC4OD}$	$K_{R02H02} \cdot 0.706 \cdot r_{\text{coch2o2_ooh}}$	Taraborrelli (2014)
G4585e	TrGC	$\text{C1ODC2O2C4OD} + \text{NO} \rightarrow \text{NO}_2 + \text{MGLYOX} + \text{HOCHCHO}$	K_{R02N0}	Taraborrelli (2014)
G4585et2	TrGC	$\text{C1ODC2O2C4OD} \rightarrow \text{MGLYOX} + \text{HOCHCHO}$	$8.00\text{E-}13 \cdot R_{02}$	Taraborrelli (2014)
G4585et3	TrGC	$\text{C1ODC2OOHC4OD} + \text{OH} \rightarrow \text{MGLYOX} + 2 \text{ CO} + .5 \text{ OH}$	$2 \cdot kt \cdot fo \cdot ftch2oh \cdot falk + kt \cdot fto \cdot fcho \cdot fpch2oh$	Taraborrelli (2014)*
G4587e	TrGC	$\text{LISOPACNO3O2} + \text{NO} \rightarrow .21 \text{ NOA} + .21 \text{ HOCH}_2\text{CHO} + .21 \text{ HO}_2 + .49 \text{ HO12CO3C4} + .49 \text{ HCHO} + .49 \text{ NO}_2 + .045 \text{ HMKVNO3} + .045 \text{ HCHO} + .255 \text{ CH}_3\text{COCH}_2\text{OH} + .255 \text{ NO3CH2CHO} + .225 \text{ H}_2\text{O}_2 + \text{NO}_2$	K_{R02N0}	Taraborrelli (2014)
G4587et2	TrGC	$\text{LISOPACNO3O2} \rightarrow .21 \text{ NOA} + .21 \text{ HOCH}_2\text{CHO} + .21 \text{ HO}_2 + .49 \text{ HO12CO3C4} + .49 \text{ HCHO} + .49 \text{ NO}_2 + .045 \text{ HMKVNO3} + .045 \text{ HCHO} + .255 \text{ CH}_3\text{COCH}_2\text{OH} + .255 \text{ NO3CH2CHO} + .225 \text{ H}_2\text{O}_2$	$8.00\text{E-}13 \cdot R_{02} + K_{R02H02} \cdot 0.706 \cdot c(\text{ind_H02})$	Taraborrelli (2014)
G4587et3	TrGC	$\text{ISOPBDNO3O2} + \text{NO} \rightarrow .6 \text{ CH}_3\text{COCH}_2\text{OH} + .6 \text{ HOCH}_2\text{CHO} + .26 \text{ MACRN} + .14 \text{ HMKVNO3} + .4 \text{ HCHO} + .4 \text{ HO}_2 + 1.6 \text{ NO}_2$	K_{R02N0}	Taraborrelli (2014)
G4587et4	TrGC	$\text{ISOPBDNO3O2} \rightarrow .6 \text{ CH}_3\text{COCH}_2\text{OH} + .6 \text{ HOCH}_2\text{CHO} + .26 \text{ MACRN} + .14 \text{ HMKVNO3} + .4 \text{ HCHO} + .4 \text{ HO}_2 + .6 \text{ NO}_2$	$2.9\text{E-}12 \cdot R_{02} + K_{R02H02} \cdot 0.706 \cdot c(\text{ind_H02})$	Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G4588e	TrGNC	LISOPACNO3 + O ₃ → .8704 OH + .365 HO ₂ + .73 MGLYOX + .4325 NO ₃ CH ₂ CHO + .135 CH ₃ COCH ₂ OH + .0675 GLYOX + .4325 NO ₂ + .0891 H ₂ O ₂ + .135 NOA + .0675 HOCHCHO + .3866 HOCH ₂ CHO + .0405 CH ₃ OH + .0405 CO + .0054 HOCH ₂ CO	4.E-16	Taraborrelli (2014)
G4599e	TrGC	LISOPACOOH + O ₃ → 1.3272 OH + 0.36986 HO ₂ + .0432 H ₂ O ₂ + 0.23002 CO + .2025 CH ₃ OOH + .01215 HOCH ₂ OOH + 0.3704 HCHO + .00405 CH ₃ OH + .0405 CO ₂ + .1825 HOCH ₂ COCH ₂ O ₂ + .365 MGLYOX + .3866 HOOCH ₂ CHO + .135 CH ₃ COCH ₂ OH + .0675 GLYOX + .00324 HCOCO ₃ A + .3866 HOCH ₂ CHO + .135 CH ₃ COCH ₂ O ₂ H + .0675 HOCHCHO + .0054 HOCH ₂ CO	4.829E-16	Taraborrelli (2014)
G4598et3	TrGC	ZCO3HC23DBCOD + OH → .62 CO ₂ H ₃ CHO + .62 OH + .62 CO ₂ + .38 MGLYOX + .38 HCOCO ₃ H + .38 HO ₂	kadt*acho*aco2h	Taraborrelli (2014)*
G4598et4	TrGC	ZCO3HC23DBCOD + OH → .62 CH ₃ COCO ₃ H + 1.24 CO + 1.24 HO ₂ + .38 CH ₃ COCHOH + .38 CO + .38 HO ₂ + .38 OH + .38 CO ₂	kads*acho*aco2h	Taraborrelli (2014)*
G41311	TrGC	LISOPEFO2 → .7143 MACR + .2857 MVK + HCHO + HO ₂	2.40E-12*R02	Taraborrelli (2014)
G41341	TrGC	LISOPEFO2 + NO → .7143 MACR + .2857 MVK + HCHO + HO ₂ + NO ₂	KR02N0	Taraborrelli (2014)
G41351t2	TrGC	LISOPEFO2 + HO ₂ → .7143 MACR + .2857 MVK + HCHO + HO ₂	KR02H02*0.706	Taraborrelli (2014)
G41361	TrGC	LISOPEFO2 + NO ₃ → .7143 MACR + .2857 MVK + HCHO + HO ₂ + NO ₂	KR02N03	Taraborrelli (2014)
G41378	TrGC	LISOPEFO2 → PEROXYRINGC2O2	9.39E9*EXP(-7322/temp)	Taraborrelli (2014)
G41341t2	TrGC	LISOPEFO2 → .7143 MACR + .2857 MVK + HCHO + OH	.7143*KHSD+.2857*KHSB	Taraborrelli (2014)
G413112	TrGC	PEROXYRINGC2O2 → CH ₃ COCH ₂ OOCH ₂ CHO + HO ₂	8.00E-13*R02	Taraborrelli (2014)
G413416	TrGC	PEROXYRINGC2O2 + NO → CH ₃ COCH ₂ OOCH ₂ CHO + HO ₂ + NO ₂	KR02N0	Taraborrelli (2014)*
G413519	TrGC	PEROXYRINGC2O2 + HO ₂ → PEROXYRINGC2OOH	KR02H02*0.706	Taraborrelli (2014)
G413618	TrGC	PEROXYRINGC2O2 + NO ₃ → CH ₃ COCH ₂ OOCH ₂ CHO + HO ₂ + NO ₂	KR02N03	Taraborrelli (2014)
G413417	TrGC	PEROXYRINGC2O2 → CH ₃ COCH ₂ OOCH ₂ CHO + OH	KHSB	Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G413619	TrGC	PEROXYRINGC2OOH + OH → PEROXYRINGC2O2	0.6*k_CH300H_OH	Taraborrelli (2014)
G413621	TrGC	PEROXYRINGC2OOH + OH → MGLYOX + 2 CO + 2 HO ₂	kt*ftoh*falk*fpch2oh	Taraborrelli (2014)*
G413622	TrGC	PEROXYRINGC2OOH + OH → .8405 HCHO + .8405 OH + .8405 CO2H3CHO + .1595 C1ODC2OOHC4OD + .1595 HO ₂	ks*fsoh*falk+ks*fsooh*falk	Taraborrelli (2014)
G4136	TrGC	CH3COCH2OOCH2CHO + OH → CH ₃ C(O)OO + CO ₂ + 2 HCHO	kt*fo*ftch2oh	Taraborrelli (2014)*
G413610	TrGC	CH3COCH2OOCH2CHO + OH → GLYOX + CH ₃ C(O)OO + HCHO	ks*fcho*fsooh	Taraborrelli (2014)
G41361t2	TrGC	CH3COCH2OOCH2CHO + OH → MGLYOX + GLYOX + HO ₂	ks*fco*fsooh	Taraborrelli (2014)
G45mbo1	TrGC	MBO + OH → LMBOABO2	8.1E-12*EXP(610/TEMP)	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo2	TrGC	MBO + O ₃ → HCHO + .16 CH ₃ COCH ₃ + .16 HO ₂ + .16 CO + .16 OH + .84 MBOOO	1.0E-17*0.57	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo3	TrGC	MBO + O ₃ → IBUTALOH + .63 CO + .37 HOCH2OOH + .16 OH + .16 HO ₂	1.0E-17*0.43	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo4	TrGC	MBO + NO ₃ → LNMBOABO2	4.6E-14*EXP(-400/TEMP)	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo5	TrGC	LMBOABO2 + HO ₂ → LMBOABOOH	KR02H02*0.706	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo6	TrGC	LMBOABO2 + NO → LMBOABNO3	KR02N0*(0.064+0.026)/2.	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo7	TrGC	LMBOABO2 + NO → HOCH ₂ CHO + CH ₃ COCH ₃ + HO ₂ + NO ₂	KR02N0*(0.936+0.974)/2*.67	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo8	TrGC	LMBOABO2 + NO → IBUTALOH + HCHO + HO ₂ + NO ₂	KR02N0*(0.936+0.974)/2*.33	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo9	TrGC	LMBOABO2 → HOCH ₂ CHO + CH ₃ COCH ₃ + HO ₂	8.8E-13*R02*.67	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo10	TrGC	LMBOABO2 → IBUTALOH + HCHO + HO ₂	8.8E-13*R02*.33	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo11	TrGC	LMBOABOOH + OH → MBOACO + OH	.67*2.93E-11+.33*2.05E-12	Rickard and Pascoe (2009), Taraborrelli (2014)*

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G45mbo12	TrGC	LMBOABOOH + OH \rightarrow LMBOABO2	.6*k_CH300H_OH	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo13	TrGC	LMBOABOOH + h ν \rightarrow HOCH ₂ CHO + CH ₃ COCH ₃ + HO ₂ + OH	1.14*jx(ip_CH300H)*.67	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo14	TrGC	LMBOABOOH + h ν \rightarrow IBUTALOH + HCHO + HO ₂ + OH	1.14*jx(ip_CH300H)*.33	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo15	TrGC	LMBOABNO3 + OH \rightarrow MBOACO + NO ₂	.67*1.75E-12+.33*2.69E-12	Rickard and Pascoe (2009), Taraborrelli (2014)*
G45mbo16	TrGC	MBOACO + OH \rightarrow MBOCOCO + HO ₂	3.79E-12	Rickard and Pascoe (2009)
G45mbo17	TrGC	MBOACO + h ν \rightarrow HCHO + HO ₂ + IPRHOCO3	J_ACETOL	Rickard and Pascoe (2009)
G45mbo18	TrGC	MBOCOCO + OH \rightarrow CO + IPRHOCO3	1.38E-11	Rickard and Pascoe (2009)
G45mbo19	TrGC	MBOCOCO + h ν \rightarrow CO + HO ₂ + IPRHOCO3	jx(ip_MGLY0X)	Rickard and Pascoe (2009)
G45mbo20	TrGC	IBUTALOH + OH \rightarrow IPRHOCO3	1.4E-11	Rickard and Pascoe (2009)
G45mbo21	TrGC	IBUTALOH + h ν \rightarrow CH ₃ COCH ₃ + HO ₂ + HO ₂ + CO	J_ACETOL	Rickard and Pascoe (2009)
G45mbo22	TrGC	IPRHOCO3 + HO ₂ \rightarrow CH ₃ COCH ₃ + HO ₂ + OH	KAPH02*rco3_oh	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo23	TrGC	IPRHOCO3 + HO ₂ \rightarrow IPRHOCO2H + O ₃	KAPH02*rco3_o3	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo24	TrGC	IPRHOCO3 + HO ₂ \rightarrow IPRHOCO3H	KAPH02*rco3_ooh	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo25	TrGC	IPRHOCO3 + NO \rightarrow CH ₃ COCH ₃ + HO ₂ + NO ₂	KAPNO	Rickard and Pascoe (2009)
G45mbo26	TrGC	IPRHOCO3 + NO ₂ \rightarrow C4PAN5	k_CH3C03_NO2	Rickard and Pascoe (2009)
G45mbo27	TrGC	IPRHOCO3 + NO ₃ \rightarrow CH ₃ COCH ₃ + HO ₂ + NO ₂	KR02N03*1.74	Rickard and Pascoe (2009)
G45mbo28	TrGC	IPRHOCO3 \rightarrow CH ₃ COCH ₃ + HO ₂	1.00E-11*R02*0.7	Rickard and Pascoe (2009)
G45mbo29	TrGC	IPRHOCO3 \rightarrow IPRHOCO2H	1.00E-11*R02*0.3	Rickard and Pascoe (2009)
G45mbo30	TrGC	IPRHOCO2H + OH \rightarrow CH ₃ COCH ₃ + HO ₂	1.72E-12	Rickard and Pascoe (2009)
G45mbo31	TrGC	IPRHOCO3H + h ν \rightarrow CH ₃ COCH ₃ + HO ₂ + OH	1.14*jx(ip_CH300H)	Rickard and Pascoe (2009)
G45mbo32	TrGC	OH + IPRHOCO3H \rightarrow IPRHOCO3	4.80E-12	Rickard and Pascoe (2009)
G45mbo33	TrGC	C4PAN5 \rightarrow IPRHOCO3 + NO ₂	K_PAN_M	Rickard and Pascoe (2009)
G45mbo34	TrGC	C4PAN5 + OH \rightarrow CH ₃ COCH ₃ + CO + NO ₂	4.75E-13	Rickard and Pascoe (2009)
G45mbo35	TrGC	LNMB0ABO2 + HO ₂ \rightarrow LNMB0ABOOH	KR02H02*0.706	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo36	TrGC	LNMB0ABO2 + NO \rightarrow .65 NO3CH2CHO + .65 CH ₃ COCH ₃ + .65 HO ₂ + .35 IBUTALOH + .35 HCHO + .35 NO ₂ + NO ₂	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G45mbo37	TrGC	$\text{LNMBOABO2} + \text{NO}_3 \rightarrow .65 \text{ NO3CH2CHO} + .65 \text{ CH}_3\text{COCH}_3 + .65 \text{ HO}_2 + .35 \text{ IBUTALOH} + .35 \text{ HCHO} + .35 \text{ NO}_2 + \text{NO}_2$	KR02N03	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo38	TrGC	$\text{LNMBOABO2} \rightarrow .65 \text{ NO3CH2CHO} + .65 \text{ CH}_3\text{COCH}_3 + .65 \text{ HO}_2 + .35 \text{ IBUTALOH} + .35 \text{ HCHO} + .35 \text{ NO}_2$	$8.8\text{E-}13 \cdot \text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo39	TrGC	$\text{LNMBOABOOH} + \text{OH} \rightarrow .65 \text{ C4MCONO3OH} + .35 \text{ NMBOBCO} + \text{OH}$	$.65 \cdot 4.89\text{E-}12 + .35 \cdot 2.52\text{E-}12$	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo40	TrGC	$\text{LNMBOABOOH} + \text{OH} \rightarrow \text{LNMBOABO2}$	$.6 \cdot k_{\text{CH300H_OH}}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo41	TrGC	$\text{LNMBOABOOH} + h\nu \rightarrow \text{NO3CH2CHO} + \text{CH}_3\text{COCH}_3 + \text{HO}_2 + \text{OH}$	$1.14 \cdot j_x(\text{ip_CH300H}) \cdot .65$	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo42	TrGC	$\text{LNMBOABOOH} + h\nu \rightarrow \text{IBUTALOH} + \text{HCHO} + \text{NO}_2 + \text{OH}$	$1.14 \cdot j_x(\text{ip_CH300H}) \cdot .35$	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo43	TrGC	$\text{C4MCONO3OH} + \text{OH} \rightarrow \text{CH}_3\text{COCH}_3 + \text{HCHO} + \text{CO}_2 + \text{NO}_2$	$1.23\text{E-}12$	Rickard and Pascoe (2009), Taraborrelli (2014)*
G45mbo44	TrGC	$\text{NMBOBCO} + \text{OH} \rightarrow \text{NC4OHCO3}$	$4.26\text{E-}12$	Rickard and Pascoe (2009)
G45mbo45	TrGC	$\text{NC4OHCO3} + \text{HO}_2 \rightarrow \text{IBUTALOH} + \text{NO}_2 + \text{OH}$	KAPH02* rco3_oh	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo46	TrGC	$\text{NC4OHCO3} + \text{HO}_2 \rightarrow \text{NC4OHCO3H}$	KAPH02*($\text{rco3_o3} + \text{rco3_ooh}$)	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo47	TrGC	$\text{NC4OHCO3} + \text{NO} \rightarrow \text{IBUTALOH} + \text{NO}_2 + \text{NO}_2$	KAPN0	Rickard and Pascoe (2009)
G45mbo48	TrGC	$\text{NC4OHCO3} + \text{NO}_2 \rightarrow \text{NC4OHCPAN}$	$k_{\text{CH3C03_N02}}$	Rickard and Pascoe (2009)
G45mbo49	TrGC	$\text{NC4OHCO3} + \text{NO}_3 \rightarrow \text{IBUTALOH} + \text{NO}_2 + \text{NO}_2$	$\text{KR02N03} \cdot 1.74$	Rickard and Pascoe (2009)
G45mbo50	TrGC	$\text{NC4OHCO3} \rightarrow \text{IBUTALOH} + \text{NO}_2$	$1.00\text{E-}11 \cdot \text{R02}$	Rickard and Pascoe (2009)
G45mbo51	TrGC	$\text{NC4OHCO3H} + \text{OH} \rightarrow \text{NC4OHCO3}$	$4.50\text{E-}12$	Rickard and Pascoe (2009)
G45mbo52	TrGC	$\text{NC4OHCO3H} \rightarrow \text{IBUTALOH} + \text{NO}_2 + \text{OH}$	$1.14 \cdot j_x(\text{ip_CH300H})$	Rickard and Pascoe (2009)
G45mbo53	TrGC	$\text{NC4OHCPAN} + \text{OH} \rightarrow \text{IBUTALOH} + \text{CO} + \text{NO}_2 + \text{NO}_2$	$1.27\text{E-}12$	Rickard and Pascoe (2009)
G45mbo54	TrGC	$\text{NC4OHCPAN} \rightarrow \text{NC4OHCO3} + \text{NO}_2$	K_PAN_M	Rickard and Pascoe (2009)
G45mbo55	TrGC	$\text{MBOOO} \rightarrow \text{IPRHOCO2H}$	$1.60\text{E-}17 \cdot \text{C}(\text{ind_H2O}) \cdot (0.08 + 0.15)$	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo56	TrGC	$\text{MBOOO} \rightarrow \text{IBUTALOH} + \text{H}_2\text{O}_2$	$1.60\text{E-}17 \cdot \text{C}(\text{ind_H2O}) \cdot 0.77$	Rickard and Pascoe (2009), Taraborrelli (2014)
G45mbo57	TrGC	$\text{MBOOO} + \text{CO} \rightarrow \text{IBUTALOH}$	$1.20\text{E-}15$	Rickard and Pascoe (2009)
G45mbo58	TrGC	$\text{MBOOO} + \text{NO} \rightarrow \text{IBUTALOH} + \text{NO}_2$	$1.00\text{E-}14$	Rickard and Pascoe (2009)
G45mbo59	TrGC	$\text{MBOOO} + \text{NO}_2 \rightarrow \text{IBUTALOH} + \text{NO}_3$	$1.00\text{E-}15$	Rickard and Pascoe (2009)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G45mbo60	TrGC	MBOOO + SO ₂ → IBUTALOH + H ₂ SO ₄	7.00E-14	Rickard and Pascoe (2009)
G410apin1	TrGC	APINENE + OH → LAPINABO2	1.47E-11*EXP(467/TEMP)*(.50+.25)	Vereecken et al. (2007), Taraborrelli (2014)*
G410apin2	TrGC	APINENE + OH → MENTHEN6ONE + HO ₂	1.47E-11*EXP(467/TEMP)*.25*.60	Vereecken et al. (2007), Taraborrelli (2014)*
G410apin3	TrGC	APINENE + OH → LVROO6R1O2	1.47E-11*EXP(467/TEMP)*.25*.40	Vereecken et al. (2007), Taraborrelli (2014)*
G410apin4	TrGC	LAPINABO2 + NO → PINAL + HO ₂ + NO ₂	KR02N0*0.770	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apin5	TrGC	LAPINABO2 + NO → LAPINABNO3	KR02N0*0.230	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apin6	TrGC	LAPINABO2 + HO ₂ → LAPINABOOH	KR02H02*0.914	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apin7	TrGC	LAPINABO2 → PINAL + HO ₂	R02*(.33*9.20E-14+.67*8.80E-13)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apin9	TrGC	LAPINABOOH + OH → .33 LAPINABO2 + .67 C96CO3	.33*1.83E-11+.67*3.28E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apin10	TrGC	LAPINABOOH + hν → PINAL + HO ₂ + OH	1.14*jx(ip_CH300H)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apin11	TrGC	LAPINABNO3 + OH → .33 PINAL + .67 C96CO3 + NO ₂	.33*5.50E-12+.67*3.64E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apin12	TrGC	MENTHEN6ONE + OH → OHMENTHEN6ONEO2	(kads+kadt)*acoch3	Vereecken et al. (2007), Taraborrelli (2014)
G410apin13	TrGC	MENTHEN6ONE + hν → LVRO6R1O2 + OH	1.14*jx(ip_CH300H)	Vereecken et al. (2007), Taraborrelli (2014)
G410apin14	TrGC	OHMENTHEN6ONEO2 + NO →	KR02N0	Vereecken et al. (2007), Taraborrelli (2014)
G410apin15	TrGC	LV2OHMENTHEN6ONE + HO ₂ + NO ₂ OHMENTHEN6ONEO2 + HO ₂ →	KR02H02*0.914	Vereecken et al. (2007), Taraborrelli (2014)
G410apin16	TrGC	LV2OHMENTHEN6ONE OHMENTHEN6ONEO2 → LV2OHMENTHEN6ONE +	R02*9.20E-14	Vereecken et al. (2007), Taraborrelli (2014)
G410apin17	TrGC	HO ₂ LV2OHMENTHEN6ONE + OH → LCARBON	1E-11	Vereecken et al. (2007), Taraborrelli (2014)
G410apin18	TrGC	LV2OHMENTHEN6ONE + hν → LCARBON + OH	1.14*jx(ip_CH300H)	Vereecken et al. (2007), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apin18t2	TrGC	PINAL + OH \rightarrow C96CO3	4.20E-11*0.772	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apin19	TrGC	PINAL + OH \rightarrow PINALO2	4.20E-11*0.228	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apin20	TrGC	PINAL \rightarrow C96O2 + CO + HO ₂	jx(ip_HOCH2CHO)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apin21	TrGC	PINAL + NO ₃ \rightarrow C96CO3 + HNO ₃	3.80E-14	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apin	TrGC	C96CO3 \rightarrow 0.3 PINONIC + 0.7 C96O2	1.00E-11*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint2	TrGC	C96CO3 + HO ₂ \rightarrow PERPINONIC	KAPH02*rco3_ooh	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint3	TrGC	C96CO3 + HO ₂ \rightarrow PINONIC + O ₃	KAPH02*rco3_o3	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint4	TrGC	C96CO3 + HO ₂ \rightarrow C96O2 + OH	KAPH02*rco3_oh	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint5	TrGC	C96CO3 + NO ₂ \rightarrow C10PAN2	k_CH3C03_N02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint6	TrGC	C96CO3 + NO \rightarrow C96O2 + NO ₂	KAPN0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint7	TrGC	C96CO3 + NO ₃ \rightarrow C96O2 + NO ₂	KR02N03*1.60	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint8	TrGC	C10PAN2 \rightarrow C96CO3 + NO ₂	k_PAN_M	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint9	TrGC	C10PAN2 + OH \rightarrow NORPINAL + CO + NO ₂	3.66E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint10	TrGC	C96O2 \rightarrow C97O2	1.30E-12*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint11	TrGC	C96O2 + NO \rightarrow C96NO3	KR02N0*0.157	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint12	TrGC	C96O2 + HO ₂ \rightarrow C96OOH	KR02H02*0.890	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint13	TrGC	C96O2 + NO \rightarrow C97O2 + NO ₂	KR02N0*0.843	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint14	TrGC	$\text{C96NO3} + h\nu \rightarrow \text{C97O2} + \text{NO}_2$	J_IC3H7N03+J_ACETOL	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint15	TrGC	$\text{C96NO3} + \text{OH} \rightarrow \text{NORPINAL} + \text{NO}_2$	2.88E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint16	TrGC	$\text{C96OOH} + h\nu \rightarrow \text{C97O2} + \text{OH}$	1.14*jx(ip_CH300H)+J_ACETOL	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint17	TrGC	$\text{C96OOH} + \text{OH} \rightarrow \text{C96O2}$	1.90E-12*EXP(190/TEMP)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint18	TrGC	$\text{C96OOH} + \text{OH} \rightarrow \text{NORPINAL} + \text{OH}$	1.30E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint19	TrGC	$\text{C97O2} \rightarrow \text{C98O2}$	6.70E-15*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint20	TrGC	$\text{C97O2} + \text{NO} \rightarrow \text{C98O2} + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint21	TrGC	$\text{C97O2} + \text{HO}_2 \rightarrow \text{C97OOH}$	KR02H02*0.890	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint22	TrGC	$\text{C97OOH} + h\nu \rightarrow \text{C98O2} + \text{OH}$	1.14*jx(ip_CH300H)+J_ACETOL	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint23	TrGC	$\text{C97OOH} + \text{OH} \rightarrow \text{C97O2}$	1.05E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint24	TrGC	$\text{C98O2} \rightarrow \text{C614O2} + \text{CH}_3\text{COCH}_3$	6.70E-15*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint25	TrGC	$\text{C98O2} + \text{NO} \rightarrow \text{C98NO3}$	KR02N0*0.118	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint26	TrGC	$\text{C98O2} + \text{NO} \rightarrow \text{C614O2} + \text{CH}_3\text{COCH}_3 + \text{NO}_2$	KR02N0*0.882	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint27	TrGC	$\text{C98O2} + \text{HO}_2 \rightarrow \text{C98OOH}$	KR02H02*0.890	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint28	TrGC	$\text{C98OOH} + h\nu \rightarrow \text{C614O2} + \text{CH}_3\text{COCH}_3 + \text{OH}$	1.14*jx(ip_CH300H)+2.15*jx(ip_MGLY0X)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint29	TrGC	$\text{C98OOH} + \text{OH} \rightarrow \text{C98O2}$	2.05E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint30	TrGC	$\text{PINONIC} + \text{OH} \rightarrow \text{C96O2}$	6.65E-12	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint31	TrGC	PINONIC + $h\nu \rightarrow$ C96O2 + HO ₂	J_ACETOL	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint32	TrGC	NORPINAL + OH \rightarrow C85CO3	2.64E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint33	TrGC	NORPINAL + $h\nu \rightarrow$ C85O2 + CO + HO ₂	jx(ip_HOCH2CHO)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint34	TrGC	NORPINAL + NO ₃ \rightarrow C85CO3 + HNO ₃	KN03AL*8.5	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint35	TrGC	PERPINONIC + $h\nu \rightarrow$ C96O2 + OH	1.14*jx(ip_CH300H)+J_ACETOL	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint36	TrGC	PERPINONIC + OH \rightarrow C96CO3	9.73E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint37	TrGC	C85CO3 \rightarrow C85O2	1.00E-11*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint38	TrGC	C85CO3 + NO \rightarrow C85O2 + NO ₂	KAPNO	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint39	TrGC	C85CO3 + NO ₂ \rightarrow C9PAN2	k_CH3CO3_NO2	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint40	TrGC	C85CO3 + HO ₂ \rightarrow C85CO3H	KAPH02*(rco3_ooh+rco3_o3)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint41	TrGC	C85CO3 + HO ₂ \rightarrow C85O2 + OH	KAPH02*rco3_oh	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint42	TrGC	C85O2 \rightarrow C86O2	6.70E-15*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint43	TrGC	C85O2 + HO ₂ \rightarrow C85OOH	KR02H02*0.859	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint44	TrGC	C85O2 + NO \rightarrow C86O2 + NO ₂	KR02NO	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint45	TrGC	C9PAN2 \rightarrow C85CO3 + NO ₂	k_PAN_M	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint46	TrGC	C9PAN2 + OH \rightarrow C85OOH + CO + NO ₂	6.60E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint47	TrGC	C85CO3H \rightarrow C85O2 + OH	1.14*jx(ip_CH300H)+J_ACETOL	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint48	TrGC	$\text{C85CO3H} + \text{OH} \rightarrow \text{C85CO3}$	1.02E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint49	TrGC	$\text{C85OOH} + h\nu \rightarrow \text{C86O2} + \text{OH}$	1.14*jx(ip_CH300H)+J_ACETOL	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint50	TrGC	$\text{C85OOH} + \text{OH} \rightarrow \text{C85O2}$	1.29E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint51	TrGC	$\text{C86O2} \rightarrow \text{C511O2} + \text{CH}_3\text{COCH}_3$	6.70E-15*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint52	TrGC	$\text{C86O2} + \text{NO} \rightarrow \text{C511O2} + \text{CH}_3\text{COCH}_3 + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint53	TrGC	$\text{C86O2} + \text{HO}_2 \rightarrow \text{C86OOH}$	KR02H02*0.859	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint54	TrGC	$\text{C86OOH} + h\nu \rightarrow \text{C511O2} + \text{CH}_3\text{COCH}_3 + \text{OH}$	1.14*jx(ip_CH300H)+ jx(ip_HOCH2CHO)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint55	TrGC	$\text{C86OOH} + \text{OH} \rightarrow \text{C86O2}$	3.45E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint56	TrGC	$\text{PINALO2} + \text{HO}_2 \rightarrow \text{PINALOOH}$	KR02H02*0.914	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint57	TrGC	$\text{PINALO2} + \text{NO} \rightarrow \text{PINALNO3}$	KR02N0*0.050	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint58	TrGC	$\text{PINALO2} + \text{NO} \rightarrow \text{C106O2} + \text{NO}_2$	KR02N0*0.950	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint59	TrGC	$\text{PINALO2} \rightarrow \text{C106O2}$	6.70E-15*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint60	TrGC	$\text{PINALOOH} + \text{OH} \rightarrow \text{PINALO2}$	2.75E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint61	TrGC	$\text{PINALOOH} + h\nu \rightarrow \text{C106O2} + \text{OH}$	1.14*jx(ip_CH300H)+jx(ip_HOCH2CHO)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint62	TrGC	$\text{PINALNO3} + \text{OH} \rightarrow \text{CO235C6CHO} + \text{CH}_3\text{COCH}_3 + \text{NO}_2$	2.25E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint63	TrGC	$\text{PINALNO3} + h\nu \rightarrow \text{C106O2} + \text{NO}_2$	J_IC3H7N03+jx(ip_HOCH2CHO)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint64	TrGC	$\text{C106O2} + \text{HO}_2 \rightarrow \text{C106OOH}$	KR02H02*0.914	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint65	TrGC	$\text{C106O2} + \text{NO} \rightarrow \text{C106NO3}$	$\text{KR02N0} \times 0.125$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint66	TrGC	$\text{C106O2} + \text{NO} \rightarrow \text{C716O2} + \text{CH}_3\text{COCH}_3 + \text{NO}_2$	$\text{KR02N0} \times 0.875$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint67	TrGC	$\text{C106O2} \rightarrow \text{C716O2} + \text{CH}_3\text{COCH}_3$	$6.70\text{E-}15 \times \text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint68	TrGC	$\text{C106OOH} + \text{OH} \rightarrow \text{C106O2}$	$8.01\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint69	TrGC	$\text{C106OOH} + h\nu \rightarrow \text{C716O2} + \text{CH}_3\text{COCH}_3 + \text{OH}$	$1.14 \times \text{jx}(\text{ip_CH300H}) + \text{jx}(\text{ip_HOCH2CHO})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint70	TrGC	$\text{C106NO3} + \text{OH} \rightarrow \text{CO235C6CHO} + \text{CH}_3\text{COCH}_3 + \text{NO}_2$	$7.03\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint71	TrGC	$\text{C106NO3} + h\nu \rightarrow \text{C716O2} + \text{CH}_3\text{COCH}_3 + \text{NO}_2$	$\text{J_IC3H7N03} + \text{jx}(\text{ip_HOCH2CHO})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint72	TrGC	$\text{CO235C6CHO} + \text{NO}_3 \rightarrow \text{CO235C6CO3} + \text{HNO}_3$	$\text{KN03AL} \times 5.5$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint73	TrGC	$\text{CO235C6CHO} + \text{OH} \rightarrow \text{CO235C6CO3}$	$6.70\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint74	TrGC	$\text{CO235C6CHO} + h\nu \rightarrow \text{CHOC3COCO3} + \text{CH}_3\text{C(O)OO}$	$2.15 \times \text{jx}(\text{ip_MGLY0X})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint75	TrGC	$\text{CO235C6CO3} + \text{HO}_2 \rightarrow \text{C235C6CO3H}$	$\text{KAPH02} \times (\text{rco3_ooh} + \text{rco3_o3})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint76	TrGC	$\text{CO235C6CO3} + \text{HO}_2 \rightarrow \text{CO235C6O2} + \text{OH}$	$\text{KAPH02} \times \text{rco3_oh}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint77	TrGC	$\text{CO235C6CO3} + \text{NO} \rightarrow \text{CO235C6O2} + \text{NO}_2$	KAPN0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint78	TrGC	$\text{CO235C6CO3} + \text{NO}_2 \rightarrow \text{C7PAN3}$	k_CH3CO3_N02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint79	TrGC	$\text{CO235C6CO3} \rightarrow \text{CO235C6O2}$	$1.00\text{E-}11 \times \text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint80	TrGC	$\text{C235C6CO3H} + \text{OH} \rightarrow \text{CO235C6CO3}$	$4.75\text{E-}12$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint81	TrGC	$\text{C235C6CO3H} + h\nu \rightarrow \text{CO235C6O2} + \text{OH}$	$1.14 \times \text{jx}(\text{ip_CH300H}) + 2.15 \times \text{jx}(\text{ip_MGLY0X})$	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint82	TrGC	$\text{CO235C6O2} + \text{HO}_2 \rightarrow \text{CO235C6OOH}$	$\text{KR02H02} \times 0.770$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint83	TrGC	$\text{CO235C6O2} + \text{NO} \rightarrow \text{CO23C4CO3} + \text{HCHO} + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint84	TrGC	$\text{CO235C6O2} \rightarrow \text{CO23C4CO3} + \text{HCHO}$	$2.00\text{E-12} \times \text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint85	TrGC	$\text{C7PAN3} + \text{OH} \rightarrow \text{CO235C5CHO} + \text{CO} + \text{NO}_2$	8.83E-13	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint86	TrGC	$\text{C7PAN3} \rightarrow \text{CO235C6CO3} + \text{NO}_2$	k_PAN_M	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint87	TrGC	$\text{CO235C6OOH} + \text{OH} \rightarrow \text{CO235C6O2}$	1.01E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint88	TrGC	$\text{CO235C6OOH} + h\nu \rightarrow \text{CO23C4CO3} + \text{HCHO} + \text{OH}$	$1.14 \times \text{jx(ip_CH300H)} + 2.15 \times \text{jx(ip_MGLY0X)}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint89	TrGC	$\text{C716O2} + \text{HO}_2 \rightarrow \text{C716OOH}$	$\text{KR02H02} \times 0.820$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint90	TrGC	$\text{C716O2} + \text{NO} \rightarrow \text{CO13C4CHO} + \text{CH}_3\text{C(O)OO} + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint91	TrGC	$\text{C716O2} \rightarrow \text{CO13C4CHO} + \text{CH}_3\text{C(O)OO}$	$8.80\text{E-13} \times \text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint92	TrGC	$\text{C716OOH} + \text{OH} \rightarrow \text{CO235C6CHO} + \text{OH}$	1.20E-10	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint93	TrGC	$\text{C716OOH} + h\nu \rightarrow \text{CO13C4CHO} + \text{CH}_3\text{C(O)OO} + \text{OH}$	$1.14 \times \text{jx(ip_CH300H)} + \text{jx(ip_HOCH2CHO)}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint94	TrGC	$\text{C511O2} \rightarrow \text{CH}_3\text{C(O)OO} + \text{HCOCH2CHO}$	$8.80\text{E-13} \times \text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint95	TrGC	$\text{C511O2} + \text{NO} \rightarrow \text{CH}_3\text{C(O)OO} + \text{HCOCH2CHO} + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint96	TrGC	$\text{C511O2} + \text{HO}_2 \rightarrow \text{C511OOH}$	$\text{KR02H02} \times 0.706$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint97	TrGC	$\text{C511OOH} + h\nu \rightarrow \text{CH}_3\text{C(O)OO} + \text{HCOCH2CHO} + \text{OH}$	$1.14 \times \text{jx(ip_CH300H)} + \text{jx(ip_HOCH2CHO)}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint98	TrGC	$\text{C511OOH} + \text{OH} \rightarrow \text{C511O2}$	7.49E-11	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint99	TrGC	$\text{HCOCH}_2\text{CHO} + \text{OH} \rightarrow \text{HCOCH}_2\text{CO}_3$	4.29E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint100	TrGC	$\text{HCOCH}_2\text{CHO} + h\nu \rightarrow \text{HCOCH}_2\text{O}_2 + \text{HO}_2 + \text{CO}$	jx(ip_H0CH2CHO)*2	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint101	TrGC	$\text{HCOCH}_2\text{CHO} + \text{NO}_3 \rightarrow \text{HCOCH}_2\text{CO}_3 + \text{HNO}_3$	2*KN03AL*2.4	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint102	TrGC	$\text{C}_6\text{H}_{14}\text{O}_2 \rightarrow \text{CO}_2\text{C}_4\text{CHO} + \text{HCHO} + \text{HO}_2$	8.80E-13*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint103	TrGC	$\text{C}_6\text{H}_{14}\text{O}_2 + \text{NO} \rightarrow \text{C}_6\text{H}_{14}\text{NO}_3$	KR02N0*0.098	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint104	TrGC	$\text{C}_6\text{H}_{14}\text{O}_2 + \text{NO} \rightarrow \text{CO}_2\text{C}_4\text{CHO} + \text{HCHO} + \text{HO}_2 + \text{NO}_2$	KR02N0*0.902	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint105	TrGC	$\text{C}_6\text{H}_{14}\text{O}_2 + \text{HO}_2 \rightarrow \text{C}_6\text{H}_{14}\text{OOH}$	KR02H02*0.770	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint106	TrGC	$\text{C}_6\text{H}_{14}\text{NO}_3 + h\nu \rightarrow \text{CO}_2\text{C}_4\text{CHO} + \text{HCHO} + \text{HO}_2 + \text{NO}_2$	2.15*jx(ip_MGLY0X)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint107	TrGC	$\text{C}_6\text{H}_{14}\text{NO}_3 + \text{OH} \rightarrow \text{C}_6\text{H}_{14}\text{CO} + \text{NO}_2$	7.11E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint108	TrGC	$\text{C}_6\text{H}_{14}\text{OOH} + h\nu \rightarrow \text{CO}_2\text{C}_4\text{CHO} + \text{HCHO} + \text{HO}_2 + \text{OH}$	1.14*jx(ip_CH300H)+2.15*jx(ip_MGLY0X)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint109	TrGC	$\text{C}_6\text{H}_{14}\text{OOH} + \text{OH} \rightarrow \text{C}_6\text{H}_{14}\text{CO} + \text{OH}$	8.69E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint110	TrGC	$\text{C}_6\text{H}_{14}\text{CO} + h\nu \rightarrow \text{CH}_3\text{COCOC}_2\text{H}_5 + \text{HOCH}_2\text{CO}_3$	J_ACETOL	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint111	TrGC	$\text{C}_6\text{H}_{14}\text{CO} + \text{OH} \rightarrow \text{CO}_2\text{C}_5\text{CHO} + \text{HO}_2$	3.22E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint112	TrGC	$\text{CH}_3\text{COCOC}_2\text{H}_5 \rightarrow \text{CH}_3\text{C}(\text{O})\text{OO} + \text{HCHO} + \text{CO}$	2.00E-12*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint113	TrGC	$\text{CH}_3\text{COCOC}_2\text{H}_5 + \text{HO}_2 \rightarrow \text{CH}_3\text{COCOC}_2\text{H}_5\text{OOH}$	KR02H02*0.625	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint114	TrGC	$\text{CH}_3\text{COCOC}_2\text{H}_5 + \text{NO} \rightarrow \text{CH}_3\text{C}(\text{O})\text{OO} + \text{HCHO} + \text{CO} + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint115	TrGC	$\text{CH}_3\text{COCOC}_2\text{H}_5\text{OOH} + \text{OH} \rightarrow \text{CH}_3\text{COCOC}_2\text{H}_5 + \text{OH}$	ks*fco*fsooh	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint116	TrGC	$\text{CH}_3\text{COCOCH}_2\text{OOH} + \text{OH} \rightarrow \text{CH}_3\text{COCOCH}_2\text{O}_2$	$.6 * k_{\text{CH300H_OH}}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint117	TrGC	$\text{CH}_3\text{COCOCH}_2\text{OOH} + h\nu \rightarrow \text{CH}_3\text{COCOCHO} + \text{OH} + \text{HO}_2$	$1.14 * jx(ip_{\text{CH300H}})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint118	TrGC	$\text{CH}_3\text{COCOCH}_2\text{OOH} + h\nu \rightarrow \text{CH}_3\text{C(O)OO} + \text{CO} + \text{OH} + \text{HCHO}$	J_{ACETOL}	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint119	TrGC	$\text{CH}_3\text{COCOCH}_2\text{OOH} + h\nu \rightarrow \text{CH}_3\text{C(O)OO} + \text{HCOCO}_3$	$2.15 * jx(ip_{\text{MGLYOX}})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint120	TrGC	$\text{CO}_2\text{C}_5\text{CHO} + \text{OH} \rightarrow \text{CO}_2\text{C}_4\text{CO}_3 + \text{CO}$	$1.33\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint121	TrGC	$\text{CO}_2\text{C}_5\text{CHO} + h\nu \rightarrow \text{CO}_2\text{C}_4\text{CO}_3 + \text{CO} + \text{HO}_2$	$jx(ip_{\text{MGLYOX}})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint122	TrGC	$\text{CO}_2\text{C}_5\text{CHO} + \text{NO}_3 \rightarrow \text{CO}_2\text{C}_4\text{CO}_3 + \text{CO} + \text{HNO}_3$	$\text{KN03AL} * 5.5$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint123	TrGC	$\text{CO}_2\text{C}_4\text{CHO} + \text{OH} \rightarrow \text{CO}_2\text{C}_4\text{CO}_3$	$6.65\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint124	TrGC	$\text{CO}_2\text{C}_4\text{CHO} + h\nu \rightarrow \text{CH}_3\text{COCOCH}_2\text{O}_2 + \text{HO}_2 + \text{CO}$	$jx(ip_{\text{H0CH2CHO}})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint125	TrGC	$\text{CO}_2\text{C}_4\text{CHO} + h\nu \rightarrow \text{CH}_3\text{C(O)OO} + \text{HCOCH}_2\text{CO}_3$	$2.15 * jx(ip_{\text{MGLYOX}})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint126	TrGC	$\text{CO}_2\text{C}_4\text{CHO} + \text{NO}_3 \rightarrow \text{CO}_2\text{C}_4\text{CO}_3 + \text{HNO}_3$	$\text{KN03AL} * 5.5$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint127	TrGC	$\text{CO}_2\text{C}_4\text{CO}_3 \rightarrow \text{CH}_3\text{COCOCH}_2\text{O}_2$	$1.00\text{E-}11 * R_{02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint128	TrGC	$\text{CO}_2\text{C}_4\text{CO}_3 + \text{NO} \rightarrow \text{CH}_3\text{COCOCH}_2\text{O}_2 + \text{NO}_2$	KAPNO	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint129	TrGC	$\text{CO}_2\text{C}_4\text{CO}_3 + \text{NO}_2 \rightarrow \text{C}_5\text{PAN}_9$	$k_{\text{CH3CO3_NO2}}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint130	TrGC	$\text{CO}_2\text{C}_4\text{CO}_3 + \text{HO}_2 \rightarrow \text{CO}_2\text{C}_4\text{CO}_3\text{H}$	$\text{KAPH02} * (rco3_{\text{ooh}} + rco3_{\text{o3}})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint131	TrGC	$\text{CO}_2\text{C}_4\text{CO}_3 + \text{HO}_2 \rightarrow \text{CH}_3\text{COCOCH}_2\text{O}_2 + \text{OH}$	$\text{KAPH02} * rco3_{\text{oh}}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint132	TrGC	$\text{CO}_2\text{C}_4\text{CO}_3\text{H} + h\nu \rightarrow \text{CH}_3\text{COCOCH}_2\text{O}_2 + \text{OH}$	$1.14 * jx(ip_{\text{CH300H}}) + jx(ip_{\text{H0CH2CHO}})$	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint133	TrGC	$\text{CO23C4CO3H} + h\nu \rightarrow \text{CO23C4CO3}$	4.23E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint134	TrGC	$\text{C5PAN9} \rightarrow \text{CO23C4CO3} + \text{NO}_2$	k_PAN_M	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint135	TrGC	$\text{C5PAN9} + \text{OH} \rightarrow \text{CO23C3CHO} + \text{CO} + \text{NO}_2$	3.12E-13	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint136	TrGC	$\text{HCOCH2CO3} \rightarrow 0.7 \text{HCOCH2O2} + 0.3 \text{HCOCH2CO2H}$	1.00E-11*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint137	TrGC	$\text{HCOCH2CO3} + \text{NO} \rightarrow \text{HCOCH2O2} + \text{NO}_2$	KAPN0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint138	TrGC	$\text{HCOCH2CO3} + \text{NO}_2 \rightarrow \text{C3PAN2}$	k_CH3CO3_NO2	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint139	TrGC	$\text{HCOCH2CO3} + \text{HO}_2 \rightarrow \text{HCOCH2CO3H}$	KAPH02*rco3_ooh	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint140	TrGC	$\text{HCOCH2CO3} + \text{HO}_2 \rightarrow \text{HCOCH2CO2H} + \text{O}_3$	KAPH02*rco3_o3	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint141	TrGC	$\text{HCOCH2CO3} + \text{HO}_2 \rightarrow \text{HCOCH2O2} + \text{CO}_2 + \text{OH}$	KAPH02*rco3_oh	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint142	TrGC	$\text{C3PAN2} \rightarrow \text{HCOCH2CO3} + \text{NO}_2$	k_PAN_M	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint143	TrGC	$\text{C3PAN2} + \text{OH} \rightarrow \text{GLYOX} + \text{CO} + \text{NO}_2$	2.10E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint144	TrGC	$\text{HCOCH2CO2H} + \text{OH} \rightarrow \text{HCOCH2O2}$	2.14E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint145	TrGC	$\text{HCOCH2CO2H} + h\nu \rightarrow \text{HCOCH2O2} + \text{HO}_2$	jx(ip_HOCH2CHO)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint146	TrGC	$\text{APINENE} + \text{O}_3 \rightarrow \text{APINBOO}$	$1.01\text{E-15} \cdot \text{EXP}(-732/\text{TEMP}) \cdot 0.50 \cdot 0.18$	Capouet et al. (2008)
G410apint147	TrGC	$\text{APINENE} + \text{O}_3 \rightarrow \text{PINONIC}$	$1.01\text{E-15} \cdot \text{EXP}(-732/\text{TEMP}) \cdot 0.50 \cdot 0.16$	Capouet et al. (2008)
G410apint148	TrGC	$\text{APINENE} + \text{O}_3 \rightarrow \text{OH} + \text{NORPINAL} + \text{CO} + \text{HO}_2$	$1.01\text{E-15} \cdot \text{EXP}(-732/\text{TEMP}) \cdot 0.50 \cdot 0.66$	Capouet et al. (2008)
G410apint149	TrGC	$\text{APINENE} + \text{O}_3 \rightarrow \text{APINAOO}$	$1.01\text{E-15} \cdot \text{EXP}(-732/\text{TEMP}) \cdot 0.50 \cdot 0.12$	Capouet et al. (2008)
G410apint150	TrGC	$\text{APINENE} + \text{O}_3 \rightarrow \text{OH} + \text{C109O2}$	$1.01\text{E-15} \cdot \text{EXP}(-732/\text{TEMP}) \cdot 0.50 \cdot (.22 + .66)$	Capouet et al. (2008)*
G410apint151	TrGC	$\text{APINAOO} \rightarrow \text{PINAL} + \text{H}_2\text{O}_2$	$1.00\text{E-17} \cdot c(\text{ind_H2O})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint152	TrGC	$\text{APINAOO} + \text{CO} \rightarrow \text{PINAL}$	1.20E-15	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint153	TrGC	APINAOO + NO \rightarrow PINAL + NO ₂	1.00E-14	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint154	TrGC	APINAOO + NO ₂ \rightarrow PINAL + NO ₃	1.00E-15	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint155	TrGC	APINAOO + SO ₂ \rightarrow PINAL + H ₂ SO ₄	7.00E-14	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint156	TrGC	APINBOO \rightarrow PINONIC	1.00E-17*c(ind_H2O)*(0.08+0.15)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint157	TrGC	APINBOO \rightarrow PINAL + H ₂ O ₂	1.00E-17*c(ind_H2O)*0.77	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint158	TrGC	APINBOO + CO \rightarrow PINAL	1.20E-15	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint159	TrGC	APINBOO + NO \rightarrow PINAL + NO ₂	1.00E-14	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint160	TrGC	APINBOO + NO ₂ \rightarrow PINAL + NO ₃	1.00E-15	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint161	TrGC	APINBOO + SO ₂ \rightarrow PINAL + H ₂ SO ₄	7.00E-14	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint162	TrGC	C109O2 \rightarrow C89CO3 + HCHO	2.00E-12*R02	Rickard and Pascoe (2009), Taraborrelli (2014)*
G410apint163	TrGC	C109O2 + NO \rightarrow C89CO3 + HCHO + NO ₂	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint164	TrGC	C109O2 + HO ₂ \rightarrow C109OOH	KR02H02*0.914	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint165	TrGC	C109OOH \rightarrow C89CO3 + HCHO + OH	1.14*jx(ip_CH300H)+jx(ip_HOCH2CHO)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint166	TrGC	C109OOH + OH \rightarrow C109CO + OH	5.47E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint167	TrGC	C109OOH + h ν \rightarrow C89CO3 + HCHO + OH	J_ACETOL	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint168	TrGC	C109CO + OH \rightarrow C89CO3 + CO	5.47E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint169	TrGC	C109CO + h ν \rightarrow C89CO3 + CO + HO ₂	jx(ip_MGLY0X)+jx(ip_HOCH2CHO)	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint170	TrGC	$\text{C89CO3} \rightarrow .56 \text{ C811CO3} + .14 \text{ C89O2} + 0.3 \text{ C89CO2H}$	$1.00\text{E-}11 \cdot \text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint171	TrGC	$\text{C89CO3} + \text{HO}_2 \rightarrow \text{C89CO3H}$	KAPH02*rc03_ooh	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint172	TrGC	$\text{C89CO3} + \text{HO}_2 \rightarrow \text{C89CO2H} + \text{O}_3$	KAPH02*rc03_o3	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint173	TrGC	$\text{C89CO3} + \text{HO}_2 \rightarrow .80 \text{ C811CO3} + 0.20 \text{ C89O2} + \text{OH}$	KAPH02*rc03_oh	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint174	TrGC	$\text{C89CO3} + \text{NO}_2 \rightarrow \text{C89PAN}$	k_CH3C03_N02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint175	TrGC	$\text{C89CO3} + \text{NO} \rightarrow 0.80 \text{ C811CO3} + 0.20 \text{ C89O2} + \text{NO}_2$	KAPN0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint176	TrGC	$\text{C89CO2H} + \text{OH} \rightarrow 0.80 \text{ C811CO3} + 0.20 \text{ C89O2}$	$2.69\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint177	TrGC	$\text{C89CO2H} + h\nu \rightarrow 0.80 \text{ C811CO3} + 0.20 \text{ C89O2} + \text{HO}_2$	jx(ip_H0CH2CH0)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint178	TrGC	$\text{C89CO3H} \rightarrow 0.80 \text{ C811CO3} + 0.20 \text{ C89O2} + \text{OH}$	$1.14 \cdot \text{jx(ip_CH300H)} + \text{jx(ip_H0CH2CH0)}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint179	TrGC	$\text{C89CO3H} + \text{OH} \rightarrow \text{C89CO3}$	$3.00\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint180	TrGC	$\text{C89PAN} \rightarrow \text{C89CO3} + \text{NO}_2$	k_PAN_M	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint181	TrGC	$\text{C89PAN} + \text{OH} \rightarrow \text{CH}_3\text{COCH}_3 + \text{CO13C4CHO} + \text{CO} + \text{NO}_2$	$2.52\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint182	TrGC	$\text{C811CO3} \rightarrow 0.7 \text{ C811O2} + 0.3 \text{ PINIC}$	$1.00\text{E-}11 \cdot \text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint183	TrGC	$\text{C811CO3} + \text{HO}_2 \rightarrow \text{C811CO3H}$	KAPH02*rc03_ooh	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint184	TrGC	$\text{C811CO3} + \text{HO}_2 \rightarrow \text{PINIC} + \text{O}_3$	KAPH02*rc03_o3	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint185	TrGC	$\text{C811CO3} + \text{HO}_2 \rightarrow \text{C811O2} + \text{OH}$	KAPH02*rc03_oh	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint186	TrGC	$\text{C811CO3} + \text{NO} \rightarrow \text{C811O2} + \text{NO}_2$	KAPN0	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint187	TrGC	$\text{C811CO3} + \text{NO}_2 \rightarrow \text{C811PAN}$	$k_{\text{CH3CO3_NO2}}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint188	TrGC	$\text{PINIC} + \text{OH} \rightarrow \text{C811O2}$	$7.29\text{E-}12$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf1	TrGC	$\text{C89O2} + \text{HO}_2 \rightarrow \text{C89OOH}$	KR02H02*0.859	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf2	TrGC	$\text{C89O2} + \text{NO} \rightarrow \text{C89NO3}$	KR02N0*0.104	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf3	TrGNC	$\text{C89O2} + \text{NO} \rightarrow \text{C810O2} + \text{NO}_2$	KR02N0*0.896	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf4	TrGNC	$\text{C89O2} + \text{NO}_3 \rightarrow \text{C810O2} + \text{NO}_2$	KR02N03	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf5	TrGC	$\text{C89O2} \rightarrow \text{C810O2}$	$6.70\text{E-}15*\text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf6	TrGC	$\text{C89OOH} + \text{OH} \rightarrow \text{C89O2}$	$3.61\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf7	TrGJC	$\text{C89OOH} + h\nu \rightarrow \text{C810O2} + \text{OH}$	$1.14*\text{jx}(\text{ip_CH300H})+\text{jx}(\text{ip_H0CH2CH0})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf8	TrGC	$\text{C89NO3} + \text{OH} \rightarrow \text{CH}_3\text{COCH}_3 + \text{CO13C4CHO} + \text{NO}_2$	$2.56\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf9	TrGJNC	$\text{C89NO3} + h\nu \rightarrow \text{C810O2} + \text{NO}_2$	$\text{J_IC3H7N03}+\text{jx}(\text{ip_H0CH2CH0})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf10	TrGC	$\text{C810O2} + \text{HO}_2 \rightarrow \text{C810OOH}$	KR02H02*0.914	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf11	TrGNC	$\text{C810O2} + \text{NO} \rightarrow \text{C810NO3}$	KR02N0*0.104	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf12	TrGNC	$\text{C810O2} + \text{NO} \rightarrow \text{CH}_3\text{COCH}_3 + \text{C514O2} + \text{NO}_2$	KR02N0*0.896	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf13	TrGNC	$\text{C810O2} + \text{NO}_3 \rightarrow \text{CH}_3\text{COCH}_3 + \text{C514O2} + \text{NO}_2$	KR02N03	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf14	TrGC	$\text{C810O2} \rightarrow \text{CH}_3\text{COCH}_3 + \text{C514O2}$	$6.70\text{E-}15*\text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf15	TrGC	$\text{C810OOH} + \text{OH} \rightarrow \text{C810O2}$	$8.35\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410ap188bf16	TrGJC	$\text{C810OOH} + h\nu \rightarrow \text{CH}_3\text{COCH}_3 + \text{C514O2} + \text{OH}$	$1.14 * jx(ip_CH300H) + jx(ip_HOCH2CHO)$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf17	TrGNC	$\text{C810NO3} + \text{OH} \rightarrow \text{CH}_3\text{COCH}_3 + \text{CO13C4CHO} + \text{NO}_2$	4.96E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf18	TrGJC	$\text{C810NO3} + h\nu \rightarrow \text{CH}_3\text{COCH}_3 + \text{C514O2} + \text{NO}_2$	$2.84 * J_IC3H7N03 + jx(ip_HOCH2CHO)$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf19	TrGC	$\text{C514O2} + \text{HO}_2 \rightarrow \text{C514OOH}$	$KR02H02 * 0.706$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf20	TrGNC	$\text{C514O2} + \text{NO} \rightarrow \text{C514NO3}$	$KR02N0 * 0.129$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf21	TrGNC	$\text{C514O2} + \text{NO} \rightarrow \text{CO13C4CHO} + \text{HO}_2 + \text{NO}_2$	$KR02N0 * 0.871$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf22	TrGNC	$\text{C514O2} + \text{NO}_3 \rightarrow \text{CO13C4CHO} + \text{HO}_2 + \text{NO}_2$	$KR02N03$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf23	TrGC	$\text{C514O2} \rightarrow \text{CO13C4CHO} + \text{HO}_2$	$2.50E-13 * R02$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf24	TrGC	$\text{C514OOH} + \text{OH} \rightarrow \text{CO13C4CHO} + \text{OH}$	1.10E-10	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf25	TrGJC	$\text{C514OOH} + h\nu \rightarrow \text{CO13C4CHO} + \text{HO}_2 + \text{OH}$	$1.14 * jx(ip_CH300H) + jx(ip_HOCH2CHO) * 2.0$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf26	TrGC	$\text{C514NO3} + \text{OH} \rightarrow \text{CO13C4CHO} + \text{NO}_2$	4.33E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410ap188bf27	TrGJC	$\text{C514NO3} + h\nu \rightarrow \text{CO13C4CHO} + \text{HO}_2 + \text{NO}_2$	$J_IC3H7N03 + jx(ip_HOCH2CHO) * 2.0$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint189	TrGC	$\text{C811O2} \rightarrow \text{C812O2}$	$1.30E-12 * R02$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint190	TrGC	$\text{C811O2} + \text{HO}_2 \rightarrow \text{C811OOH}$	$KR02H02 * 0.859$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint191	TrGC	$\text{C811O2} + \text{NO} \rightarrow \text{C812O2} + \text{NO}_2$	$KR02N0$	Rickard and Pascoe (2009), Taraborrelli (2014)*
G410apint192	TrGC	$\text{C811CO3H} + h\nu \rightarrow \text{C811O2} + \text{OH}$	$1.14 * jx(ip_CH300H)$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint193	TrGC	$\text{C811CO3H} + \text{OH} \rightarrow \text{C811CO3}$	1.04E-11	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint194	TrGC	$\text{C811PAN} \rightarrow \text{C811CO3} + \text{NO}_2$	k_PAN_M	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint195	TrGC	$\text{C811PAN} + \text{OH} \rightarrow \text{C721CHO} + \text{CO} + \text{NO}_2$	$6.77\text{E-}12$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint196	TrGC	$\text{C812O2} \rightarrow \text{C813O2}$	$9.20\text{E-}14*\text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)*
G410apint197	TrGC	$\text{C812O2} + \text{NO} \rightarrow \text{C813O2} + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint198	TrGC	$\text{C812O2} + \text{HO}_2 \rightarrow \text{C812OOH}$	$\text{KR02H02}*0.859$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint199	TrGC	$\text{C812OOH} + h\nu \rightarrow \text{C813O2} + \text{OH}$	$1.14*\text{jx}(\text{ip_CH300H})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint200	TrGC	$\text{C812OOH} + \text{OH} \rightarrow \text{C812O2}$	$1.09\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint201	TrGC	$\text{C813O2} \rightarrow \text{CH}_3\text{COCH}_3 + \text{C512O2}$	$6.70\text{E-}15*\text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)*
G410apint202	TrGC	$\text{C813O2} + \text{NO} \rightarrow \text{CH}_3\text{COCH}_3 + \text{C512O2} + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)*
G410apint203	TrGC	$\text{C813O2} + \text{HO}_2 \rightarrow \text{C813OOH}$	$\text{KR02H02}*0.859$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint204	TrGC	$\text{C813OOH} + h\nu \rightarrow \text{CH}_3\text{COCH}_3 + \text{C512O2} + \text{OH}$	$1.14*\text{jx}(\text{ip_CH300H})+\text{jx}(\text{ip_MGLY0X})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint205	TrGC	$\text{C813OOH} + \text{OH} \rightarrow \text{C813O2}$	$1.86\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint206	TrGC	$\text{C721CHO} + \text{NO}_3 \rightarrow \text{C721CO3} + \text{HNO}_3$	$\text{KN03AL}*8.5$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint207	TrGC	$\text{C721CHO} + \text{OH} \rightarrow \text{C721CO3}$	$2.63\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint208	TrGC	$\text{C721CHO} \rightarrow \text{C721O2} + \text{CO} + \text{HO}_2$	$\text{jx}(\text{ip_H0CH2CHO})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint209	TrGC	$\text{C721CO3} + \text{HO}_2 \rightarrow \text{C721CO3H}$	KAPH02*rc03_ooh	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint210	TrGC	$\text{C721CO3} + \text{HO}_2 \rightarrow \text{C721O2} + \text{OH}$	KAPH02*rc03_oh	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint211	TrGC	$\text{C721CO3} + \text{HO}_2 \rightarrow \text{NORPINIC} + \text{O}_3$	KAPH02*rc03_o3	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint212	TrGC	$\text{C721CO3} + \text{NO} \rightarrow \text{C721O2} + \text{NO}_2$	KAPN0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint213	TrGC	$\text{C721CO3} + \text{NO}_2 \rightarrow \text{C721PAN}$	k_CH3CO3_NO2	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint214	TrGC	$\text{C721CO3} + \text{NO}_3 \rightarrow \text{C721O2} + \text{NO}_2$	KR02N03*1.74	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint215	TrGC	$\text{C721CO3} \rightarrow \text{C721O2}$	1.00E-11*R02*0.7	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint216	TrGC	$\text{C721CO3} \rightarrow \text{NORPINIC}$	1.00E-11*R02*0.3	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint217	TrGC	$\text{C721O2} + \text{HO}_2 \rightarrow \text{C721OOH}$	KR02H02*0.820	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint218	TrGC	$\text{C721O2} + \text{NO} \rightarrow \text{C722O2} + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint219	TrGC	$\text{C721O2} \rightarrow \text{C722O2}$	1.30E-12*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint220	TrGC	$\text{C721CO3H} + \text{OH} \rightarrow \text{C721CO3}$	9.65E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint221	TrGC	$\text{C721CO3H} + h\nu \rightarrow \text{C721O2} + \text{OH}$	1.14*jx(ip_CH300H)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint222	TrGC	$\text{NORPINIC} + \text{OH} \rightarrow \text{C721O2}$	6.57E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint223	TrGC	$\text{C721PAN} + \text{OH} \rightarrow \text{C721OOH} + \text{CO} + \text{NO}_2$	2.96E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint224	TrGC	$\text{C721PAN} \rightarrow \text{C721CO3} + \text{NO}_2$	k_PAN_M	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint225	TrGC	$\text{C721OOH} + \text{OH} \rightarrow \text{C721O2}$	1.27E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint226	TrGC	$\text{C721OOH} + h\nu \rightarrow \text{C722O2} + \text{OH}$	1.14*jx(ip_CH300H)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint227	TrGC	$\text{C722O2} + \text{HO}_2 \rightarrow \text{C722OOH}$	KR02H02*0.820	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint228	TrGC	$\text{C722O2} + \text{NO} \rightarrow \text{CH}_3\text{COCH}_3 + \text{C44O2} + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint229	TrGC	$\text{C722O2} \rightarrow \text{CH}_3\text{COCH}_3 + \text{C44O2}$	$6.70\text{E-15}*\text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint230	TrGC	$\text{C722OOH} + \text{OH} \rightarrow \text{C722O2}$	3.31E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint231	TrGC	$\text{C722OOH} + h\nu \rightarrow \text{CH}_3\text{COCH}_3 + \text{C44O2} + \text{OH}$	$1.14*jx(ip_CH300H)$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint232	TrGC	$\text{C44O2} + \text{HO}_2 \rightarrow \text{C44OOH}$	$\text{KR02H02}*0.625$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint233	TrGC	$\text{C44O2} + \text{NO} \rightarrow \text{HCOCH2CHO} + \text{HO}_2 + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint234	TrGC	$\text{C44O2} \rightarrow \text{HCOCH2CHO} + \text{HO}_2$	$8.80\text{E-13}*\text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint235	TrGC	$\text{C44OOH} + \text{OH} \rightarrow \text{C44O2}$	7.46E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint236	TrGC	$\text{C44OOH} + h\nu \rightarrow \text{HCOCH2CHO} + \text{HO}_2 + \text{OH}$	$1.14*jx(ip_CH300H)$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint237	TrGC	$\text{C512O2} \rightarrow \text{C513O2}$	$1.30\text{E-12}*\text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)*
G410apint238	TrGC	$\text{C512O2} + \text{HO}_2 \rightarrow \text{C512OOH}$	$\text{KR02H02}*0.706$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint239	TrGC	$\text{C512O2} + \text{NO} \rightarrow \text{C513O2} + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)*
G410apint240	TrGC	$\text{C512OOH} + h\nu \rightarrow \text{C513O2} + \text{OH}$	$1.14*jx(ip_CH300H)+jx(ip_HOCH2CHO)$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint241	TrGC	$\text{C512OOH} + \text{OH} \rightarrow \text{CO13C4CHO} + \text{OH}$	1.01E-10	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint242	TrGC	$\text{C513O2} \rightarrow \text{GLYOX} + \text{HOC2H4CO3}$	$8.80\text{E-13}*\text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)*
G410apint243	TrGC	$\text{C513O2} + \text{NO} \rightarrow \text{GLYOX} + \text{HOC2H4CO3} + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint244	TrGC	$\text{C513O2} + \text{HO}_2 \rightarrow \text{C513OOH}$	$\text{KR02H02}*0.706$	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint245	TrGC	$\text{CO13C4CHO} + \text{OH} \rightarrow \text{CHOC3COCO3}$	1.33E-10	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint246	TrGC	$\text{CO13C4CHO} + h\nu \rightarrow \text{CHOC3COO2} + \text{CO} + \text{HO}_2$	jx(ip_HOCH2CHO)*2	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint247	TrGC	$\text{CO13C4CHO} + \text{NO}_3 \rightarrow \text{CHOC3COCO3} + \text{HNO}_3$	2*KN03AL*5.5	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint248	TrGC	$\text{C513OOH} + h\nu \rightarrow \text{GLYOX} + \text{HOC2H4CO3} + \text{OH}$	$1.14*\text{jx(ip_CH300H)}+\text{jx(ip_HOCH2CHO)}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint249	TrGC	$\text{C513OOH} + \text{OH} \rightarrow \text{C513CO} + \text{OH}$	9.23E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint250	TrGC	$\text{CHOC3COCO3} \rightarrow \text{CHOC3COO2}$	1.00E-11*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint251	TrGC	$\text{CHOC3COCO3} + \text{HO}_2 \rightarrow \text{CHOC3COOOH}$	KAPH02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint252	TrGC	$\text{CHOC3COCO3} + \text{NO}_2 \rightarrow \text{CHOC3COPAN}$	k_CH3C03_N02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint253	TrGC	$\text{CHOC3COCO3} + \text{NO} \rightarrow \text{CHOC3COO2} + \text{NO}_2$	KAPN0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint254	TrGC	$\text{CHOC3COO2} \rightarrow \text{HCOCH2CO3} + \text{HCHO}$	2.00E-12*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint255	TrGC	$\text{CHOC3COO2} + \text{HO}_2 \rightarrow \text{C413COOOH}$	KR02H02*0.625	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint256	TrGC	$\text{CHOC3COO2} + \text{NO} \rightarrow \text{HCOCH2CO3} + \text{HCHO} + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint257	TrGC	$\text{C513CO} + \text{OH} \rightarrow \text{HOC2H4CO3} + \text{CO} + \text{CO}$	2.64E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint258	TrGC	$\text{C513CO} + h\nu \rightarrow \text{HOC2H4CO3} + \text{HO}_2 + \text{CO} + \text{CO}$	$\text{jx(ip_MGLYOX)}+2.15*\text{jx(ip_MGLYOX)}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint259	TrGC	$\text{C413COOOH} + \text{OH} \rightarrow \text{CHOC3COO2}$	8.33E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint260	TrGC	$\text{C413COOOH} + h\nu \rightarrow \text{HCOCH2CO3} + \text{HCHO} + \text{OH}$	$1.14*\text{jx(ip_CH300H)}+\text{jx(ip_HOCH2CHO)}$ +J_ACETOL	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint261	TrGC	$\text{CHOC3COOOH} + \text{OH} \rightarrow \text{CHOC3COCO3}$	7.55E-11	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint262	TrGC	$\text{CHOC3COOOH} + h\nu \rightarrow \text{CHOC3COO2} + \text{OH}$	$1.14 * \text{jx}(\text{ip_CH300H}) + \text{jx}(\text{ip_HOCH2CHO}) + \text{J_ACETOL}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint263	TrGC	$\text{CHOC3COPAN} \rightarrow \text{CHOC3COCO3} + \text{NO}_2$	k_PAN_M	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint264	TrGC	$\text{CHOC3COPAN} + \text{OH} \rightarrow \text{C4CODIAL} + \text{CO} + \text{NO}_2$	$7.19\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint265	TrGC	$\text{C4CODIAL} + \text{OH} \rightarrow \text{C312COCO3}$	$3.39\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint266	TrGC	$\text{C4CODIAL} + h\nu \rightarrow \text{HCOCOCH2O2} + \text{HO}_2 + \text{CO}$	$\text{jx}(\text{ip_HOCH2CHO})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint267	TrGC	$\text{C4CODIAL} + h\nu \rightarrow \text{HCOCH2CO3} + \text{HO}_2 + \text{CO}$	$\text{jx}(\text{ip_MGLY0X})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint268	TrGC	$\text{C4CODIAL} + \text{NO}_3 \rightarrow \text{C312COCO3} + \text{HNO}_3$	$2 * \text{KN03AL} * 4.0$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint269	TrGC	$\text{C312COCO3} \rightarrow \text{HCOCOCH2O2}$	$1.00\text{E-}11 * \text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint270	TrGC	$\text{C312COCO3} + \text{HO}_2 \rightarrow \text{C312COCO3H}$	$\text{KAPH02} * \text{rco3_ooh}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint271	TrGC	$\text{C312COCO3} + \text{HO}_2 \rightarrow \text{HCOCOCH2O2} + \text{OH}$	$\text{KAPH02} * (1 - \text{rco3_ooh})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint272	TrGC	$\text{C312COCO3} + \text{NO}_2 \rightarrow \text{C312COPAN}$	k_CH3C03_N02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint273	TrGC	$\text{C312COCO3} + \text{NO} \rightarrow \text{HCOCOCH2O2} + \text{NO}_2$	KAPN0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint274	TrGC	$\text{C312COCO3H} + \text{OH} \rightarrow \text{C312COCO3}$	$1.63\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint275	TrGC	$\text{C312COCO3H} + h\nu \rightarrow \text{HCOCOCH2O2} + \text{OH}$	$1.14 * \text{jx}(\text{ip_CH300H}) + \text{jx}(\text{ip_MGLY0X})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint276	TrGC	$\text{C312COPAN} \rightarrow \text{C312COCO3} + \text{NO}_2$	k_PAN_M	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint277	TrGC	$\text{C312COPAN} + \text{OH} \rightarrow \text{HCOCOCHO} + \text{CO} + \text{NO}_2$	$1.27\text{E-}11$	Rickard and Pascoe (2009), Taraborrelli (2014)*
G410apint278	TrGC	$\text{HOC2H4CO3} \rightarrow 0.7 \text{HOCH}_2\text{CH}_2\text{O}_2 + 0.3 \text{HOC2H4CO2H}$	$1.00\text{E-}11 * \text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint279	TrGC	$\text{HOC2H4CO3} + \text{NO} \rightarrow \text{HOCH}_2\text{CH}_2\text{O}_2 + \text{NO}_2$	KAPN0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint280	TrGC	$\text{HOC2H4CO3} + \text{HO}_2 \rightarrow \text{HOC2H4CO3H}$	KAPH02*rc03_ooh	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint281	TrGC	$\text{HOC2H4CO3} + \text{HO}_2 \rightarrow \text{HOCH}_2\text{CH}_2\text{O}_2 + \text{OH}$	KAPH02*rc03_oh	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint282	TrGC	$\text{HOC2H4CO3} + \text{HO}_2 \rightarrow \text{HOC2H4CO2H} + \text{O}_3$	KAPH02*rc03_o3	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint283	TrGC	$\text{HOC2H4CO3} + \text{NO}_2 \rightarrow \text{C3PAN1}$	k_CH3CO3_NO2	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint284	TrGC	$\text{HOC2H4CO2H} + \text{OH} \rightarrow \text{HOCH}_2\text{CH}_2\text{O}_2$	1.39E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint285	TrGC	$\text{HOC2H4CO3H} + \text{OH} \rightarrow \text{HOC2H4CO3}$	1.73E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint286	TrGC	$\text{HOC2H4CO3H} + h\nu \rightarrow \text{HOCH}_2\text{CH}_2\text{O}_2 + \text{OH}$	1.14*jx(ip_CH300H)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint287	TrGC	$\text{C3PAN1} \rightarrow \text{HOC2H4CO3} + \text{NO}_2$	k_PAN_M	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint288	TrGC	$\text{C3PAN1} + \text{OH} \rightarrow \text{HOCH}_2\text{CHO} + \text{CO} + \text{NO}_2$	4.51E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint289	TrGC	$\text{APINENE} + \text{NO}_3 \rightarrow \text{LNAPINABO2}$	1.2E-12*EXP(490./temp)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint290	TrGC	$\text{LNAPINABO2} \rightarrow \text{PINAL} + \text{NO}_2$	(.65*6.70E-15+.35*2.50E-13)*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint291	TrGC	$\text{LNAPINABO2} + \text{NO} \rightarrow \text{PINAL} + \text{NO}_2 + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint292	TrGC	$\text{LNAPINABO2} + \text{HO}_2 \rightarrow \text{LNAPINABOOH}$	KR02H02*0.914	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint293	TrGC	$\text{LNAPINABO2} + \text{NO}_3 \rightarrow \text{PINAL} + \text{NO}_2 + \text{NO}_2$	KR02N03	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint294	TrGC	$\text{LNAPINABOOH} + h\nu \rightarrow \text{PINAL} + \text{NO}_2 + \text{OH}$	1.14*jx(ip_CH300H)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint295	TrGC	$\text{LNAPINABOOH} + \text{OH} \rightarrow \text{C96CO3}$.65*6.87E-12+.35*1.23E-11	Rickard and Pascoe (2009), Taraborrelli (2014)*

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410bpin	TrGC	BPINENE + OH \rightarrow BPINAO2	$1.47\text{E}-11 * \text{EXP}(467/\text{TEMP}) * (0.8326 * 0.3 + 0.068) / (0.8326 + 0.068)$	Vereecken and Peeters (2012)*
G410bpint2	TrGC	BPINENE + OH \rightarrow LVROO6R1O2	$1.47\text{E}-11 * \text{EXP}(467/\text{TEMP}) * 0.8326 * 0.7 / (0.8326 + 0.068)$	Vereecken and Peeters (2012)*
G410bpint3	TrGC	BPINAO2 + HO ₂ \rightarrow BPINAOOH	KR02H02*0.914	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint4	TrGC	BPINAO2 + NO \rightarrow BPINANO3	KR02N0*0.240	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint5	TrGC	BPINAO2 + NO \rightarrow NOPINONE + HCHO + HO ₂ + NO ₂	KR02N0*0.760	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint6	TrGC	BPINAO2 \rightarrow NOPINONE + HCHO + HO ₂	$9.20\text{E}-14 * \text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint7	TrGC	BPINAOOH + OH \rightarrow BPINAO2	$1.33\text{E}-11$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint8	TrGC	BPINAOOH + h ν \rightarrow NOPINONE + HCHO + HO ₂ + OH	$1.14 * \text{jx}(\text{ip_CH300H})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint9	TrGC	BPINANO3 + OH \rightarrow NOPINONE + HCHO + NO ₂	$4.70\text{E}-12$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint10	TrGC	LVROO6R1O2 + NO \rightarrow LVROO6R3O2 + CH ₃ COCH ₃ + NO ₂	KR02N0*0.892	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint11	TrGC	LVROO6R1O2 + NO \rightarrow LVROO6R1NO3	KR02N0*0.108	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint12	TrGC	LVROO6R1O2 + HO ₂ \rightarrow LVROO6R1OOH	KR02H02*0.914	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint13	TrGC	LVROO6R1O2 \rightarrow LVROO6R3O2 + CH ₃ COCH ₃	$1.60\text{E}-13 * \text{R02}$	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint14	TrGC	LVROO6R3O2 \rightarrow LVROO6R5O2	$5.68\text{E}10 * \text{exp}(-8745/\text{TEMP})$	Vereecken and Peeters (2012), Taraborrelli (2014)*
G410bpint15	TrGC	LVROO6R3O2 + NO \rightarrow LVROO6R3O + NO ₂	KR02N0*0.890	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint16	TrGC	LVROO6R3O2 + NO \rightarrow LVROO6R3NO3	KR02N0*0.110	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint17	TrGC	LVROO6R3O2 + HO ₂ \rightarrow LVROO6R3OOH	KR02H02*0.820	Vereecken and Peeters (2012), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410bpint18	TrGC	LVROO6R3O2 \rightarrow LVROO6R3O	2.50E-13*R02	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint19	TrGC	LVROO6R3O \rightarrow LVROO6R4P + HO ₂	5.7E10*exp(-2949/TEMP)	Vereecken and Peeters (2012), Taraborrelli (2014)*
G410bpint20	TrGC	LVROO6R5O2 \rightarrow LVROO6R5P + OH	9.17E10*exp(-8706/TEMP)	Vereecken and Peeters (2012), Taraborrelli (2014)*
G410bpint21	TrGC	LVRO6R1O2 + NO \rightarrow LVRO6R3O2 + NO ₂	KR02N0*0.747	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint22	TrGC	LVRO6R1O2 + NO \rightarrow LVRO6R1NO3	KR02N0*0.253	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint23	TrGC	LVRO6R1O2 + HO ₂ \rightarrow LVRO6R1OOH	KR02H02*0.914	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint24	TrGC	LVRO6R1O2 \rightarrow LVRO6R3O2	8.80E-13*R02	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint25	TrGC	LVRO6R3O2 + NO \rightarrow LVRO6R3P + NO ₂	KR02N0*0.893	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint26	TrGC	LVRO6R3O2 + NO \rightarrow LVRO6R3NO3	KR02N0*0.107	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint27	TrGC	LVRO6R3O2 + HO ₂ \rightarrow LVRO6R3OOH	KR02H02*0.914	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint28	TrGC	LVRO6R3O2 \rightarrow LVRO6R3P	5.00E-12*R02	Vereecken and Peeters (2012), Taraborrelli (2014)
G410bpint29	TrGC	NOPINONE + OH \rightarrow NOPINDO2	1.55E-11	Lewis et al. (2005), Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint30	TrGC	NOPINDO2 + HO ₂ \rightarrow NOPINDOOH	KR02H02*0.890	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint31	TrGC	NOPINDO2 + NO \rightarrow C89CO3 + NO ₂	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint32	TrGC	NOPINDO2 \rightarrow C89CO3	2.00E-12*R02	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint33	TrGC	NOPINDOOH + OH \rightarrow NOPINDCO + OH	2.63E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint34	TrGC	NOPINDOOH + h ν \rightarrow C89CO3 + OH	1.14*jx(ip_CH300H)	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410bpint35	TrGC	$\text{NOPINDCO} + \text{OH} \rightarrow \text{C89CO3}$	3.07E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint36	TrGC	$\text{BPINENE} + \text{O}_3 \rightarrow \text{NOPINONE} + .63 \text{ CO} + .37 \text{ HOCH2OOH} + .16 \text{ OH} + .16 \text{ HO}_2$	$1.5\text{E-}17 * .051 / (1 - .027)$	Nguyen et al. (2009), Taraborrelli (2014)
G410bpint37	TrGC	$\text{BPINENE} + \text{O}_3 \rightarrow \text{NOPINOO}$	$1.5\text{E-}17 * .368 / (1 - .027)$	Nguyen et al. (2009), Taraborrelli (2014)
G410bpint38	TrGC	$\text{BPINENE} + \text{O}_3 \rightarrow \text{NOPINDO2} + \text{OH}$	$1.5\text{E-}17 * .283 / (1 - .027)$	Nguyen et al. (2009), Taraborrelli (2014)
G410bpint40	TrGC	$\text{BPINENE} + \text{O}_3 \rightarrow \text{C8BC} + \text{CO}_2$	$1.5\text{E-}17 * (.104 + .167) / (1 - .027)$	Nguyen et al. (2009), Taraborrelli (2014)*
G410bpint41	TrGC	$\text{C8BC} + \text{OH} \rightarrow \text{C8BCO2}$	3.04E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint42	TrGC	$\text{C8BCO2} + \text{HO}_2 \rightarrow \text{C8BCOOH}$	KR02H02*0.859	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint43	TrGC	$\text{C8BCO2} + \text{NO} \rightarrow \text{C8BCNO3}$	KR02N0*0.138	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint44	TrGC	$\text{C8BCO2} + \text{NO} \rightarrow \text{C89O2} + \text{NO}_2$	KR02N0*0.862	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint45	TrGC	$\text{C8BCO2} \rightarrow \text{C89O2}$	$2.50\text{E-}13 * \text{R02}$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint46	TrGC	$\text{C8BCOOH} + \text{OH} \rightarrow \text{C8BCCO} + \text{OH}$	1.62E-11	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint47	TrGC	$\text{C8BCOOH} + h\nu \rightarrow \text{C89O2} + \text{OH}$	$1.14 * \text{jx}(\text{ip_CH300H})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint48	TrGC	$\text{C8BCNO3} + \text{OH} \rightarrow \text{C8BCCO} + \text{NO}_2$	1.84E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint49	TrGC	$\text{C8BCNO3} \rightarrow \text{C89O2} + \text{NO}_2$	J_IC3H7N03	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint50	TrGC	$\text{C8BCCO} + \text{OH} \rightarrow \text{C89O2}$	3.94E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint51	TrGC	$\text{NOPINOO} \rightarrow \text{NOPINONE} + \text{H}_2\text{O}_2$	$6.00\text{E-}18 * c(\text{ind_H2O})$	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint52	TrGC	$\text{NOPINOO} + \text{CO} \rightarrow \text{NOPINONE}$	1.2E-15	Rickard and Pascoe (2009), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410bpint53	TrGC	$\text{NOPINOO} + \text{NO} \rightarrow \text{NOPINONE} + \text{NO}_2$	1.E-14	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint54	TrGC	$\text{NOPINOO} + \text{NO}_2 \rightarrow \text{NOPINONE} + \text{NO}_3$	1.E-15	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint55	TrGC	$\text{NOPINOO} + \text{SO}_2 \rightarrow \text{NOPINONE} + \text{H}_2\text{SO}_4$	7.E-14	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint56	TrGC	$\text{BPINENE} + \text{NO}_3 \rightarrow \text{LNBPINABO2}$	2.51E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint57	TrGC	$\text{LNBPINABO2} + \text{HO}_2 \rightarrow \text{LNBPINABOOH}$	KR02H02*0.914	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint58	TrGC	$\text{LNBPINABO2} + \text{NO} \rightarrow \text{NOPINONE} + \text{HCHO} + \text{NO}_2 + \text{NO}_2$	KR02N0	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint59	TrGC	$\text{LNBPINABO2} + \text{NO}_3 \rightarrow \text{NOPINONE} + \text{HCHO} + \text{NO}_2 + \text{NO}_2$	KR02N03	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint60	TrGC	$\text{LNBPINABO2} \rightarrow \text{NOPINONE} + \text{HCHO} + \text{NO}_2$	9.20E-14*R02*0.7	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint61	TrGC	$\text{LNBPINABO2} \rightarrow \text{BPINAN03}$	9.20E-14*R02*0.3	Rickard and Pascoe (2009), Taraborrelli (2014)*
G410bpint62	TrGC	$\text{LNBPINABOOH} + \text{OH} \rightarrow \text{LNBPINABO2}$	9.58E-12	Rickard and Pascoe (2009), Taraborrelli (2014)
G410bpint63	TrGC	$\text{LNBPINABOOH} + h\nu \rightarrow \text{NOPINONE} + \text{HCHO} + \text{NO}_2 + \text{OH}$	1.14*jx(ip_CH300H)	Rickard and Pascoe (2009), Taraborrelli (2014)
G410apint296	TrGC	$\text{CARENE} + \text{OH} \rightarrow \text{LAPINABO2}$	8.7E-11*(.50+.25)	Wolfe et al. (2011), Taraborrelli (2014)*
G410apint297	TrGC	$\text{CARENE} + \text{OH} \rightarrow \text{MENTHEN6ONE} + \text{HO}_2$	8.7E-11*.25*.60	Wolfe et al. (2011), Taraborrelli (2014)
G410apint298	TrGC	$\text{CARENE} + \text{OH} \rightarrow \text{LVROO6R1O2}$	8.7E-11*.25*.40	Wolfe et al. (2011), Taraborrelli (2014)
G410apint299	TrGC	$\text{CARENE} + \text{O}_3 \rightarrow \text{APINBOO}$	2.E-16*.50*.18	Wolfe et al. (2011), Taraborrelli (2014)
G410apint300	TrGC	$\text{CARENE} + \text{O}_3 \rightarrow \text{PINONIC}$	2.E-16*.50*.16	Wolfe et al. (2011), Taraborrelli (2014)
G410apint301	TrGC	$\text{CARENE} + \text{O}_3 \rightarrow \text{OH} + \text{NORPINAL} + \text{CO} + \text{HO}_2$	2.E-16*.50*.66	Wolfe et al. (2011), Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410apint302	TrGC	CARENE + O ₃ → APINAOO	2.E-16*.50*.12	Wolfe et al. (2011), Taraborrelli (2014)
G410apint303	TrGC	CARENE + O ₃ → OH + C109O2	2.E-16*.50*(.22+.66)	Wolfe et al. (2011), Taraborrelli (2014)*
G410apint304	TrGC	CARENE + NO ₃ → LNAPINABO2	9.5E-12	Wolfe et al. (2011), Taraborrelli (2014)
G410myrc	TrGC	BMYRCENE + OH → MYRCISOPO2	9.19E-12*exp(1071./temp)*0.64	Hites and Turner (2009), Orlando et al. (2000), Taraborrelli (2014)
G410myrct2	TrGC	BMYRCENE + OH → MYRCO2	9.19E-12*exp(1071./temp)*0.36	Hites and Turner (2009), Orlando et al. (2000), Taraborrelli (2014)
G410myrct3	TrGC	BMYRCENE + O ₃ → .25 CH ₃ COCH ₃ + .75 OH + .75 CH ₃ COCH ₂ O ₂ + MYRCCHO	4.7E-16	Atkinson and Arey (2003), Taraborrelli (2014)
G410myrct4	TrGNC	BMYRCENE + NO ₃ → MYRCNO3	1.1E-11	Atkinson and Arey (2003), Taraborrelli (2014)
G410myrct5	TrGC	MYRCO2 → CH ₃ COCH ₃ + HO ₂ + MYRCCHO	8.E-13*R02	Taraborrelli (2014)
G410myrct6	TrGC	MYRCO2 + HO ₂ → MYRCOOH	KR02H02	Taraborrelli (2014)
G410myrct7	TrGCN	MYRCO2 + NO → .80 CH ₃ COCH ₃ + .80 HO ₂ + .80 MYRCCHO + .80 NO ₂ + .20 MYRCNO3	KR02N0	Taraborrelli (2014)*
G410myrct8	TrGCJ	MYRCOOH + hν → CH ₃ COCH ₃ + OH + HO ₂ + MYRCCHO	1.14*jx(ip_CH300H)	Taraborrelli (2014)
G410myrct9	TrGC	MYRCOOH + OH → MYRCOOHISOPO2	1.55E-10	Baker et al. (2004), Taraborrelli (2014)
G410myrct10	TrGC	MYRCOOH + NO ₃ → CH ₃ COCH ₃ + SURCH3CHO + NISOPO2	4.7E-13	Baker et al. (2004), Taraborrelli (2014)
G410myrct11	TrGC	MYRCNO3 + OH → iC ₃ H ₇ ONO ₂ + MYRCCHOISOPO2	1.55E-10	see note
G410myrct12	TrGC	MYRCISOPO2 → MYRCISOPO	1.E-12*R02	Taraborrelli (2014)
G410myrct13	TrGC	MYRCISOPO2 + HO ₂ → MYRCISOPOOH	KR02H02	Taraborrelli (2014)
G410myrct14	TrGC	MYRCISOPO2 + NO → .8 MYRCISOPO + .8 NO ₂ + .2 MYRCISOPNO3	KR02N0	Taraborrelli (2014)
G410myrct15	TrGC	MYRCISOPO2 → MYRCHPALD + HO ₂	K16HS	Taraborrelli (2014)*
G410myrct16	TrGC	MYRCISOPO → CH ₃ COCH ₃ + SURCH3CHO + .43 MVK + .27 MACR + .7 HCHO + .29 LHC4ACCHO + HO ₂	KDEC	Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410myrct17	TrGC	MYRCISOPNO3 + OH \rightarrow CH ₃ COCH ₃ + HO ₂ + SURCH3CHO + ISOPBNO3	kadt+kads	Taraborrelli (2014)
G410myrct18	TrGC	MYRCISOPOOH + h ν \rightarrow MYRCISOPO + OH	1.14*jx(ip_CH300H)	Taraborrelli (2014)
G410myrct19	TrGC	MYRCISOPOOH + OH \rightarrow SURCH3CHO + .43 ISOPBOOH + .27 ISOPDOOH + .29 LISOPACOOH + CH ₃ COCH ₃ + HO ₂	kadt+kads	Taraborrelli (2014)
G410myrct20	TrGCJ	MYRCHPALD + h ν \rightarrow OH + ZCODC23DBCOOH + HO ₂	J_HPald	Taraborrelli (2014)
G410myrct21	TrGC	MYRCHPALD + OH \rightarrow MYRCCHOHPALD + HO ₂ + CH ₃ COCH ₃	kadt+kads	Taraborrelli (2014)
G410myrct22	TrGC	MYRCHPALD + OH \rightarrow SURCH3CO3 + SURCH3CHO + HCHO + GLYOX + HO ₂	(kadt+kads)*acho*ach2ooh	Taraborrelli (2014)
G410myrct23	TrGC	MYRCHPALD + O ₃ \rightarrow MYRCCHOHPALD + .25 CH ₃ COCH ₃ + .75 OH + .75 CH ₃ COCH ₂ O ₂	4.7E-16	Taraborrelli (2014)
G410myrct24	TrGC	MYRCCHO + OH \rightarrow SURCH3CHO + MYRCCHOISOPO2	1.55E-10	Baker et al. (2004), Taraborrelli (2014)
G410myrct25	TrGC	MYRCCHO + NO ₃ \rightarrow SURCH3CHO + NISOPO2	4.7E-13	Baker et al. (2004), Taraborrelli (2014)
G410myrct26	TrGC	MYRCCHOISOPO2 \rightarrow MYRCCHOISOPO	1.E-12*R02	Taraborrelli (2014)
G410myrct27	TrGC	MYRCCHOISOPO2 + HO ₂ \rightarrow SURCH3CHO + .43 ISOPBOOH + .27 ISOPDOOH + .29 LISOPACOOH	KR02H02	Taraborrelli (2014)
G410myrct28	TrGC	MYRCCHOISOPO2 + NO \rightarrow .8 MYRCCHOISOPO + .8 NO ₂ + .2 SURCH3CHO + .2 ISOPBNO3	KR02N0	Taraborrelli (2014)
G410myrct29	TrGC	MYRCCHOISOPO2 \rightarrow MYRCCHOHPALD + HO ₂	K16HS	Taraborrelli (2014)*
G410myrct30	TrGC	MYRCCHOISOPO \rightarrow SURCH3CHO + .43 MVK + .27 MACR + .7 HCHO + .29 LHC4ACCHO + HO ₂	KDEC	Taraborrelli (2014)
G410myrct31	TrGCJ	MYRCCHOHPALD + h ν \rightarrow SURCH3CHO + OH + ZCODC23DBCOOH + HO ₂	J_HPald	Taraborrelli (2014)
G410myrct32	TrGC	MYRCCHOHPALD + OH \rightarrow SURCH3CHO + .3 C1ODC2O2C4OOH + .2 C1OOHC2O2C4OD + .3 C1ODC3O2C4OOH + .2 C1OOHC3O2C4OD	(kadt+kads)*acho*ach2ooh	Taraborrelli (2014)
G410myrct33	TrGC	MYRCOOHISOPO2 \rightarrow MYRCOOHISOPO	1.E-12*R02	Taraborrelli (2014)
G410myrct34	TrGC	MYRCOOHISOPO2 + HO ₂ \rightarrow MYRCOOHISOPOOH	KR02H02	Taraborrelli (2014)
G410myrct35	TrGC	MYRCOOHISOPO2 + NO \rightarrow .8 MYRCOOHISOPO + .8 NO ₂ + .2 CH ₃ COCH ₃ + .2 SURCH3CHO + .2 ISOPBNO3	KR02N0	Taraborrelli (2014)
G410myrct36	TrGC	MYRCOOHISOPO2 \rightarrow MYRCOOHHPALD + HO ₂	K16HS	Taraborrelli (2014)*

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410myrct37	TrGC	MYRCOOHISOPO \rightarrow CH ₃ COCH ₃ + SURCH3CHO + .43 MVK + .27 MACR + .7 HCHO + .29 LHC4ACCHO + HO ₂ + OH	KDEC	Taraborrelli (2014)
G410myrct38	TrGCJ	MYRCOOHISOPOOH + h ν \rightarrow MYRCOOHISOPO + OH	1.14*jx(ip_CH300H)	Taraborrelli (2014)
G410myrct39	TrGCJ	MYRCOOHISOPOOH + h ν \rightarrow CH ₃ COCH ₃ + HO ₂ + SURCH3CHO + .43 ISOPBOOH + .27 ISOPDOOH + .29 LISOPACOOH + OH	1.14*jx(ip_CH300H)	Taraborrelli (2014)
G410myrct40	TrGCJ	MYRCOOHHPALD + h ν \rightarrow CH ₃ COCH ₃ + OH + SURCH3CHO + OH + ZCODC23DBCOOH + HO ₂	J_HPALD	Taraborrelli (2014)
G410myrct41	TrGC	MYRCOOHHPALD + OH \rightarrow CH ₃ COCH ₃ + OH + SURCH3CHO + .3 C1ODC2O2C4OOH + .2 C1OOHC2O2C4OD + .3 C1ODC3O2C4OOH + .2 C1OOHC3O2C4OD	(kadt+kads)*acho*ach2ooh	Taraborrelli (2014)
G410myrct42	TrGC	SURCH3CHO + OH \rightarrow SURCH3CO3 + H ₂ O	4.4E-12*EXP(365./temp)	Taraborrelli (2014)*
G410myrct43	TrGC	SURCH3CHO + h ν \rightarrow HCHO + HO ₂ + HO ₂ + CO	jx(ip_CH3CHO)	Taraborrelli (2014)
G410myrct44	TrGC	SURCH3CHO + NO ₃ \rightarrow HNO ₃ + SURCH3CO3	KN03AL	Taraborrelli (2014)
G410myrct45	TrGC	SURCH3CO3 \rightarrow HCHO + HO ₂ + CO ₂	1.00E-11*0.7*R02	Taraborrelli (2014)
G410myrct46	TrGC	SURCH3CO3 \rightarrow SURCH3CO2H	1.00E-11*0.3*R02	Taraborrelli (2014)
G410myrct47	TrGC	SURCH3CO3 + HO ₂ \rightarrow OH + HCHO + HO ₂ + CO ₂	KAPH02*0.44	Taraborrelli (2014)
G410myrct48	TrGC	SURCH3CO3 + HO ₂ \rightarrow SURCH3CO3H	KAPH02*0.41	Taraborrelli (2014)
G410myrct49	TrGC	SURCH3CO3 + HO ₂ \rightarrow SURCH3CO2H + O ₃	KAPH02*0.15	Taraborrelli (2014)
G410myrct50	TrGC	SURCH3CO3 + NO ₂ \rightarrow SURPAN	k_CH3CO3_NO2	Taraborrelli (2014)
G410myrct51	TrGC	SURCH3CO3 + NO \rightarrow NO ₂ + HCHO + HO ₂ + CO ₂	KAPNO	Taraborrelli (2014)
G410myrct52	TrGC	SURCH3CO3 + NO ₃ \rightarrow NO ₂ + HCHO + HO ₂ + CO ₂	KR02NO3*1.60	Taraborrelli (2014)
G410myrct53	TrGC	SURCH3CO2H + OH \rightarrow HCHO + HO ₂ + CO ₂	4.2E-14*exp(850./temp)	Taraborrelli (2014)*
G410myrct54	TrGC	SURCH3CO3H + h ν \rightarrow HCHO + HO ₂ + OH + CO ₂	1.14*jx(ip_CH300H)	Taraborrelli (2014)
G410myrct55	TrGC	SURCH3CO3H + OH \rightarrow SURCH3CO3 + H ₂ O	0.6*k_CH300H_OH	Taraborrelli (2014)
G410myrct56	TrGC	SURPAN \rightarrow SURCH3CO3 + NO ₂	k_PAN_M	Taraborrelli (2014)
G410myrct57	TrGC	SURPAN + OH \rightarrow HCHO + CO + NO ₂	9.50E-13*EXP(-650./temp)	Taraborrelli (2014)
G410afarn	TrGC	FARNESENE + OH \rightarrow LFARNISOPO2	2.7E-11*EXP(390./temp)	Taraborrelli (2014)*
G410afarnt2	TrGC	FARNESENE + OH \rightarrow LFARNO2	2.*1.9E-11*exp(450./temp)	Taraborrelli (2014)*
G410afarnt3	TrGC	FARNESENE + O ₃ \rightarrow FARNCHO + .25 CH ₃ COCH ₃ + .75 OH + .75 CH ₃ COCH ₂ O ₂	6.51E-15*exp(-829./temp)	Taraborrelli (2014)*
G410afarnt4	TrGC	FARNESENE + O ₃ \rightarrow MHO + MYRCCHO + .75 OH	6.51E-15*exp(-829./temp)	Taraborrelli (2014)*
G410afarnt5	TrGNC	FARNESENE + NO ₃ \rightarrow LFARNO2	2E-12+2*9.37E-12	Taraborrelli (2014)*

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410afarnt6	TrGC	LFARNO2 \rightarrow LFARNO	8.E-13*R02	Taraborrelli (2014)
G410afarnt7	TrGC	LFARNO2 + HO ₂ \rightarrow LFARNOOH	KR02H02	Taraborrelli (2014)
G410afarnt8	TrGC	LFARNO2 + NO \rightarrow .75 LFARNO + .75 NO ₂ + .25 LFARNONO2	KR02N0	Taraborrelli (2014)
G410afarnt9	TrGCJ	LFARNOOH + h ν \rightarrow LFARNO + OH	1.14*jx(ip_CH300H)	Taraborrelli (2014)
G410afarnt10	TrGC	LFARNOOH + OH \rightarrow MHOOOH + MYRCCHOISOPO2	2.7E-11*EXP(390./temp)	Taraborrelli (2014)
G410afarnt11	TrGC	LFARNONO2 + OH \rightarrow .5 MHO + .5 FARNCHO + MYRCISOPNO3	2.*1.9E-11*exp(450./temp)	Taraborrelli (2014)
G410afarnt12	TrGC	LFARNO \rightarrow .5 MHO + .5 MYRCCHO + .5 FARNCHO + .5 CH ₃ COCH ₃ + HO ₂	KDEC	Taraborrelli (2014)
G410afarnt13	TrGC	LFARNISOPO2 \rightarrow LFARNISOPO	1.E-12*R02	Taraborrelli (2014)
G410afarnt14	TrGC	LFARNISOPO2 + HO ₂ \rightarrow LFARNISOPOOH	KR02H02	Taraborrelli (2014)
G410afarnt15	TrGC	LFARNISOPO2 + NO \rightarrow .75 LFARNISOPO + .75 NO ₂ + .25 LFARNISOPNO3	KR02N0	Taraborrelli (2014)
G410afarnt16	TrGC	LFARNISOPO2 \rightarrow MYRCCHOHPALD + HO ₂ + MHO	K16HS	Taraborrelli (2014)*
G410afarnt17	TrGC	LFARNISOPO \rightarrow MHO + MYRCCHOISOPO	KDEC	Taraborrelli (2014)
G410afarnt18	TrGC	LFARNISOPNO3 + OH \rightarrow MHO + HO ₂ + ISOPBNO3	kadt+kads	Taraborrelli (2014)
G410afarnt19	TrGC	LFARNISOPOOH + h ν \rightarrow LFARNISOPO + OH	1.14*jx(ip_CH300H)	Taraborrelli (2014)
G410afarnt20	TrGC	LFARNISOPOOH + OH \rightarrow MHO + MYRCCHOISOPO2	kadt+kads	Taraborrelli (2014)
G410afarnt21	TrGC	LFARNISOPOOH + O ₃ \rightarrow MHO + MYRCCHOISOPO2 + .75 OH	2.*6.51E-15*exp(-829./temp)	Taraborrelli (2014)*
G410afarnt22	TrGC	FARNCHO + OH \rightarrow FARNCHOO2	1.9E-11*exp(450./temp)	Taraborrelli (2014)*
G410afarnt23	TrGC	FARNCHO + OH \rightarrow FARNCHOISOPO2	2.7E-11*EXP(390./temp)	Taraborrelli (2014)*
G410afarnt24	TrGC	FARNCHO + O ₃ \rightarrow OPA + MYRCCHO + .75 OH	6.51E-15*exp(-829./temp)	Taraborrelli (2014)*
G410afarnt25	TrGNC	FARNCHO + NO ₃ \rightarrow FARNCHOISOPO2	2E-12	Taraborrelli (2014)*
G410afarnt26	TrGNC	FARNCHO + NO ₃ \rightarrow FARNCHOO2	9.37E-12	Taraborrelli (2014)*
G410afarnt27	TrGC	FARNCHOO2 \rightarrow FARNCHOO	8.E-13*R02	Taraborrelli (2014)
G410afarnt28	TrGC	FARNCHOO2 + HO ₂ \rightarrow FARNCHOOOH	KR02H02	Taraborrelli (2014)
G410afarnt29	TrGC	FARNCHOO2 + NO \rightarrow .75 FARNCHOO + .75 NO ₂ + .25 FARNCHONO3	KR02N0	Taraborrelli (2014)
G410afarnt30	TrGCJ	FARNCHOOOH + h ν \rightarrow FARNCHOO + OH	1.14*jx(ip_CH300H)	Taraborrelli (2014)
G410afarnt31	TrGC	FARNCHOOOH + OH \rightarrow OPA + HO ₂ + MYRCCHOISOPO2	2.7E-11*EXP(390./temp)	Taraborrelli (2014)*
G410afarnt32	TrGC	FARNCHONO3 + OH \rightarrow OPA + MYRCISOPNO3	1.9E-11*exp(450./temp)	Taraborrelli (2014)*
G410afarnt33	TrGC	FARNCHOO \rightarrow OPA + MYRCCHO + HO ₂	KDEC	Taraborrelli (2014)

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410afarnt34	TrGC	FARNCHOISOPO2 \rightarrow FARNCHOISOPO	1.E-12*R02	Taraborrelli (2014)
G410afarnt35	TrGC	FARNCHOISOPO2 + HO ₂ \rightarrow FARNCHOISOPOOH	KR02H02	Taraborrelli (2014)
G410afarnt36	TrGC	FARNCHOISOPO2 + NO \rightarrow .75 FARNCHOISOPO + .75 NO ₂ + .25 FARNCHOISOPNO3	KR02N0	Taraborrelli (2014)
G410afarnt37	TrGC	FARNCHOISOPO2 \rightarrow OPA + MYRCCHOHPALD + HO ₂ + MHO	K16HS	Taraborrelli (2014)*
G410afarnt38	TrGC	FARNCHOISOPO \rightarrow OPA + MYRCCHOISOPO	KDEC	Taraborrelli (2014)
G410afarnt39	TrGC	FARNCHOISOPNO3 + OH \rightarrow OPA + HO ₂ + ISOPBNO3	kadt+kads	Taraborrelli (2014)
G410afarnt40	TrGC	FARNCHOISOPOOH + h ν \rightarrow FARNCHOISOPO + OH	1.14*jx(ip_CH300H)	Taraborrelli (2014)
G410afarnt41	TrGC	FARNCHOISOPOOH + OH \rightarrow OPA + MYRCCHOISOPO2	kadt+kads	Taraborrelli (2014)
G410afarnt42	TrGC	FARNCHOISOPOOH + O ₃ \rightarrow OPA + MYRCCHOISOPO2 + .75 OH	6.51E-15*exp(-829./temp)	Taraborrelli (2014)
G410afarnt43	TrGC	MHO + OH \rightarrow MHOO2	1.37E-10	Smith et al. (1996), Taraborrelli (2014)*
G410afarnt44	TrGC	MHO + O ₃ \rightarrow .25 CH ₃ COCH ₃ + .125 OPA + .25 GLYOX + .25 CH ₃ COCH ₂ O ₂ + .125 H ₂ O ₂ + .75 OH + .75 CH ₃ COCH ₂ O ₂ + .75 OPA	3.9E-16	Grosjean et al. (1996), Taraborrelli (2014)*
G410afarnt45	TrGC	MHO + NO ₃ \rightarrow MHOO2	7E-12	Smith et al. (1996), Taraborrelli (2014)
G410afarnt46	TrGC	MHOO2 \rightarrow CH ₃ COCH ₃ + OPA + HO ₂	8.E-13*R02	Taraborrelli (2014)
G410afarnt47	TrGC	MHOO2 + HO ₂ \rightarrow MHOOOH	KR02H02	Taraborrelli (2014)
G410afarnt48	TrGC	MHOO2 + NO \rightarrow .25 MHONO3 + .75 CH ₃ COCH ₃ + .75 OPA + .75 HO ₂ + .75 NO ₂	KR02N0	Taraborrelli (2014)
G410afarnt49	TrGC	MHOOOH + OH \rightarrow MHOO2	0.6*k_CH300H_OH	Taraborrelli (2014)
G410afarnt50	TrGC	MHOOOH + OH \rightarrow MHOCODOOH + HO ₂	kt*ftoh*falk*falk	Taraborrelli (2014)
G410afarnt51	TrGCJ	MHOOOH + h ν \rightarrow CH ₃ COCH ₃ + OPA + OH	1.14*jx(ip_CH300H)	Taraborrelli (2014)
G410afarnt52	TrGCJ	MHOCODOOH + h ν \rightarrow CH ₃ COCH ₃ + CH ₃ COCH ₂ O ₂ + SURCH3CO3 + OH	1.14*jx(ip_CH300H)+2.77*jx(ip_HOCH2CHO)	Taraborrelli (2014)*
G410afarnt53	TrGC	MHONO3 + OH \rightarrow MHOCODNO3 + HO ₂	kt*ftoh*falk*fch2ono2	Taraborrelli (2014)
G410afarnt54	TrGC	MHOCODNO3 + h ν \rightarrow CH ₃ COCH ₃ + NO ₂ + CH ₃ COCH ₂ O ₂ + SURCH3CO3	2.84*J_IC3H7N03	Taraborrelli (2014)
G410afarnt55	TrGC	OPA + OH \rightarrow CH ₃ COCH ₂ O ₂ + SURCH3CO3	2.E-11	Fruekilde et al. (1997), Taraborrelli (2014)*

Table 1: Gas phase reactions (... continued)

#	labels	reaction	rate coefficient	reference
G410afarnt56	TrGCJ	$\text{OPA} + h\nu \rightarrow \text{CH}_3\text{COCH}_2\text{O}_2 + \text{SURCH}_3\text{CO}_3 + \text{HO}_2$	$\text{jx(ip_CH3CHO)} + \text{jx(ip_CH3COCH3)}$	Taraborrelli (2014)

*Notes:

Rate coefficients for three-body reactions are defined via the function `k_3rd`($T, M, k_0^{300}, n, k_{\text{inf}}^{300}, m, f_c$). In the code, the temperature T is called `temp` and the concentration of “air molecules” M is called `cair`. Using the auxiliary variables $k_0(T)$, $k_{\text{inf}}(T)$, and k_{ratio} , `k_3rd` is defined as:

$$k_0(T) = k_0^{300} \times \left(\frac{300\text{K}}{T} \right)^n \quad (1)$$

$$k_{\text{inf}}(T) = k_{\text{inf}}^{300} \times \left(\frac{300\text{K}}{T} \right)^m \quad (2)$$

$$k_{\text{ratio}} = \frac{k_0(T)M}{k_{\text{inf}}(T)} \quad (3)$$

$$\text{k_3rd} = \frac{k_0(T)M}{1 + k_{\text{ratio}}} \times f_c \left(\frac{1}{1 + (\log_{10}(k_{\text{ratio}}))^2} \right) \quad (4)$$

A similar function, called `k_3rd_iupac` here, is used by Atkinson et al. (2005) for three-body reactions. It has the same function parameters as `k_3rd` and it is defined as:

$$k_0(T) = k_0^{300} \times \left(\frac{300\text{K}}{T} \right)^n \quad (5)$$

$$k_{\text{inf}}(T) = k_{\text{inf}}^{300} \times \left(\frac{300\text{K}}{T} \right)^m \quad (6)$$

$$k_{\text{ratio}} = \frac{k_0(T)M}{k_{\text{inf}}(T)} \quad (7)$$

$$N = 0.75 - 1.27 \times \log_{10}(f_c) \quad (8)$$

$$\text{k_3rd_iupac} = \frac{k_0(T)M}{1 + k_{\text{ratio}}} \times f_c \left(\frac{1}{1 + (\log_{10}(k_{\text{ratio}}/N))^2} \right) \quad (9)$$

Structure Activity Relationship (SAR) for estimation of rate constants as developed by Taraborrelli (2010).

SAR for H-abstraction:

base rate constants

$$ks = 8.42E - 13 \quad (10)$$

$$kt = 1.75E - 12 \quad (11)$$

$$kp = 1.24E - 13 \quad (12)$$

$$kcho = 5.55E - 12 \times \text{EXP}(311/T) * .95 \quad (13)$$

$$krohro = 1.6E - 13 \quad (14)$$

$$kco2h = .66 \times 4.2E - 14 \times \exp(850/T) \quad (15)$$

activation factors (parameter names starting with f)

$$fsoh = 3.44 \quad (16)$$

$$ftoh = 2.68 \quad (17)$$

$$fsooh = 7. \quad (18)$$

$$ftooh = 7. \quad (19)$$

$$fono2 = 0.04 \quad (20)$$

$$fch2ono2 = 0.2 \quad (21)$$

$$fcpan = .25 \quad (22)$$

$$fallyl = 3.6 \quad (23)$$

$$falk = 1.23 \quad (24)$$

$$fcho = 0.55 \quad (25)$$

$$fco2h = 1.67 \quad (26)$$

$$fco = 0.73 \quad (27)$$

$$fo = 8.15 \quad (28)$$

$$fpch2oh = 1.29 \quad (29)$$

$$ftch2oh = 0.53 \quad (30)$$

$$kadp = 0.45E - 11 \quad (31)$$

$$kads = 3.0E - 11 \quad (32)$$

$$kadtt = 5.5E - 11 \quad (33)$$

$$kadsecprim = 3.0E - 11 \quad (34)$$

$$kadtertprim = 5.7E - 11 \quad (35)$$

activation factors (parameter names starting with a)

$$apan = 0.56 \quad (36)$$

$$acho = 0.31 \quad (37)$$

$$acoch3 = 0.76 \quad (38)$$

$$ach2ono2 = 0.47 \quad (39)$$

$$ach2oh = 1.7 \quad (40)$$

$$ach2ooh = 0.21 \quad (41)$$

$$acoh = 2.2 \quad (42)$$

$$acooH = 2.2 \quad (43)$$

$$aco2h = 0.25 \quad (44)$$

G2110: The rate coefficient is: `k_HO2_HO2 = (1.5E-12*EXP(19./temp)+1.7E-33*EXP(1000./temp)*cair)*(1.+1.4E-21*EXP(2200./temp)*C(ind_H2O))`. The value for the first (pressure-independent) part is from Christensen et al. (2002), the water term from Kircher and Sander (1984).

G3109: The rate coefficient is: `k_NO3_NO2 = k_3rd(temp, cair, 2.E-30, 4.4, 1.4E-12, 0.7, 0.6)`.

G3110: The rate coefficient is defined as backward reaction divided by equilibrium constant.

G3203: The rate coefficient is: `k_NO2_HO2 = k_3rd(temp, cair, 1.8E-31, 3.2, 4.7E-12, 1.4, 0.6)`.

G3206: The rate coefficient is: `k_HNO3_OH = 2.4E-14 * EXP(460./temp) + 1./ (1./`

base rate constants

$(6.5\text{E-}34 * \text{EXP}(1335./\text{temp}) * \text{cair}) + 1./$
 $(2.7\text{E-}17 * \text{EXP}(2199./\text{temp}))$)

G3207: The rate coefficient is defined as backward reaction divided by equilibrium constant.

G4103a: Sander et al. (2006) recommend a zero product yield for HCHO. G4103b: Sander et al. (2006) recommend a zero product yield for HCHO.

G4107: The rate coefficient is: $k_{\text{CH3OOH_OH}} = 3.8\text{E-}12 * \text{EXP}(200./\text{temp})$.

G4109: The same temperature dependence assumed as for $\text{CH}_3\text{CHO} + \text{NO}_3$.

G4201e: The product distribution is from Rickard and Pascoe (2009), after substitution of the Criegee intermediate by its decomposition products.

G4206: The product $\text{C}_2\text{H}_5\text{OH}$, which reacts only with OH, is substituted by its degradation products $\approx 0.1 \text{HOCH}_2\text{CH}_2\text{O}_2 + 0.9 \text{CH}_3\text{CHO} + 0.9 \text{HO}_2$.

G4207: The rate constant $8.01\text{E-}12$ is for the H abstraction in alpha to the $-\text{OOH}$ group (Rickard and Pascoe, 2009) and $0.6 * k_{\text{CH3OOH_OH}}$ is for the $\text{C}_2\text{H}_5\text{O}_2$ channel. The branching ratios are calculated from the terms of the rate coefficient at 298 K.

G4211et1: personal communication C. Gross (MPICH)
G4211et2: personal communication C. Gross (MPICH)
G4211et3: personal communication C. Gross (MPICH)

G4213: The rate coefficient is: $k_{\text{CH3CO3_NO2}} = k_{\text{3rd}}(\text{temp}, \text{cair}, 9.7\text{E-}29, 5.6, 9.3\text{E-}12, 1.5, 0.6)$.

G4218: The rate coefficient is the same as for the CH_3O_2 channel in G4107 ($\text{CH}_3\text{OOH} + \text{OH}$).

G4221: The rate coefficient $\text{isk}_{\text{PAN_M}} = k_{\text{CH3CO3_NO2}}/9.\text{E-}29 * \text{EXP}(-14000./\text{temp})$, i.e. the rate coefficient is defined as backward reaction divided by equilibrium constant.

G4226ea: personal communication C. Gross (MPICH)

G4226eb: personal communication C. Gross (MPICH)

G4226ec: personal communication C. Gross (MPICH)

G4243: Orlando et al. (1998a) estimated that about 25% of the $\text{HOCH}_2\text{CH}_2\text{O}$ in this reaction is produced with sufficient excess energy that it decomposes promptly. The decomposition products are $2 \text{HCHO} + \text{HO}_2$.

G4255et2: personal communication C. Gross (MPICH)

G4255et3: personal communication C. Gross (MPICH)

G4260et2: personal communication C. Gross (MPICH)

G4260et3: personal communication C. Gross (MPICH)

G4266e: $0.4 \text{CO}_2\text{HCH}_2\text{O}_2$ approximated to $\text{HOOCH}_2\text{CO}_2\text{H}$

G4300: The product $\text{NC}_3\text{H}_7\text{O}_2$ is substituted with its degradation products $\text{C}_2\text{H}_5\text{O}_2 + \text{CO}_2 + \text{HO}_2$.

G4304: The value for the generic $\text{RO}_2 + \text{HO}_2$ reaction from Atkinson (1997) is used here.

G4306: The MCM (Rickard and Pascoe, 2009) products are $0.2 \text{IPROPOL} + 0.2 \text{CH}_3\text{COCH}_3 + 0.6 \text{IC}_3\text{H}_7\text{O}$. IPROPOL and $\text{IC}_3\text{H}_7\text{O}$ are substituted with their degradation products. We assume IPROPOL to be oxidized entirely to $\text{CH}_3\text{COCH}_3 + \text{HO}_2$ by OH. $\text{IC}_3\text{H}_7\text{O} + \text{O}_2$ produces the same products.

G4307: Analogous to G4207 for both rate coefficient and branching ratios.

G4315a: The same value as for G4107 ($\text{CH}_3\text{OOH} + \text{OH}$) is used, multiplied by the branching ratio of the CH_3O_2 channel.

G4343e: k for the major channel of $\text{ACETOL} + \text{OH}$

G4353e: $.28 \text{CH}_3\text{CHO}_2\text{CO}_2\text{H}$ yield approximated by $.14 \text{CH}_3\text{COCO}_2\text{H} + .14 \text{CH}_3\text{CHO}$

G4354e: k for $(\text{CH}_3)_2\text{CCO}$ adjusted

G4355e: k for $(\text{CH}_3)_2\text{CCO}$ adjusted

G4400: $\text{LC}_4\text{H}_9\text{O}_2$ represents $0.127 \text{NC}_4\text{H}_9\text{O}_2 + 0.873 \text{SC}_4\text{H}_9\text{O}_2$.

G4401: $\text{NC}_4\text{H}_9\text{O}$ and $\text{SC}_4\text{H}_9\text{O}$ are substituted with $2 \text{CO}_2 + \text{C}_2\text{H}_5\text{O}_2$ and $0.636 \text{MEK} + \text{HO}_2$ and $0.364 \text{CH}_3\text{CHO} + \text{C}_2\text{H}_5\text{O}_2$, respectively. The stoichiometric coefficients on the right side are weighted averages.

G4403: The alkyl nitrate yield is the weighted average yield for the two isomers forming from $\text{NC}_4\text{H}_9\text{O}_2$ and $\text{SC}_4\text{H}_9\text{O}_2$.

G4404: The product distribution is the weighted average of the single isomer hydroperoxides. It is calculated from the rate constants of single channels and the ratio of the isomers $\text{NC}_4\text{H}_9\text{O}_2$ and $\text{SC}_4\text{H}_9\text{O}_2$. The overall rate constant for this reaction is calculated as weighted average of the channels rate constants. The relative weight of the products from $\text{NC}_4\text{H}_9\text{OOH}$ and $\text{SC}_4\text{H}_9\text{OOH}$ are then 0.0887 and 0.9113. The channels producing RO_2 are given the rate coefficient $0.6 * k_{\text{CH3OOH_OH}}$ as for G4107. For $\text{NC}_4\text{H}_9\text{OOH}$ the products are $0.327 \text{NC}_4\text{H}_9\text{O}_2 + 0.673 \text{C}_3\text{H}_7\text{CHO} + 0.673 \text{OH}$. $\text{C}_3\text{H}_7\text{CHO}$ is then substituted with $2 \text{CO}_2 + \text{C}_2\text{H}_5\text{O}_2$. Hence, $0.327 \text{NC}_4\text{H}_9\text{O}_2 + 1.346 \text{CO}_2 + 0.673 \text{C}_2\text{H}_5\text{O}_2 + 0.673 \text{OH}$. For $\text{SC}_4\text{H}_9\text{OOH}$ the products are $0.219 \text{SC}_4\text{H}_9\text{O}_2 + 0.781 \text{MEK} + 0.781 \text{OH}$.

G4406e: Lumped products from Taraborrelli et al. (2009).

G4413: LMEKO_2 represents $0.459 \text{MEKAO}_2 + 0.462 \text{MEKBO}_2 + 0.079 \text{MEKCO}_2$.

G4415: Alkyl nitrate formation is neglected. The products of MEKAO and MEKCO are substituted with $\text{HCHO} + \text{CO}_2 + \text{HOCH}_2\text{CH}_2\text{O}_2$ and $\text{HCHO} + \text{CO}_2 + \text{C}_2\text{H}_5\text{O}_2$.

G4416: LMEKOOH is assumed having the composition $0.459 \text{MEKAOOH} + 0.462 \text{MEKBOOH} + 0.079 \text{MEKCOOH}$. $\text{MEKAOOH} + \text{OH}$ gives $0.89 \text{CO}_2\text{C}_3\text{CHO} +$

0.89 OH + 0.11 MEKAO2 + H₂O. CO2C3CHO is substituted with CH₃COCH₂O₂ + CO₂ and the products become 0.89 CH₃COCH₂O₂ + 0.89 CO₂ + 0.89 OH + 0.11 MEKAO2 + H₂O. MEKBOOH + OH gives 0.758 BIACET + 0.758 OH + 0.242 MEKBO2 + H₂O. MEKCOOH + OH gives 0.614 EGLYOX + 0.614 OH + 0.386 MEKCO2 + H₂O. EGLYOX is substituted with C₂H₅O₂ + 2 CO₂ and the products become 0.614 C₂H₅O₂ + 1.228 CO₂ + 0.614 OH + 0.386 MEKCO2 + H₂O.

G4417: The rate coefficient is the combination of the ones for the two isomers weighted by the relative abundances for NC4H9NO3 and SC4H9NO3, respectively. Product distribution is calculated accordingly. NC4H9NO3 + OH gives C3H7CHO + NO₂ + H₂O with C3H7CHO being substituted with 2 CO₂ + C₂H₅O₂. After substitution is obtained 2 CO₂ + C₂H₅O₂ + NO₂ + H₂O. SC4H9NO3 + OH gives MEK + NO₂ + H₂O. For the product distribution NC4H9NO3 and SC4H9NO3 account for 0.08577 and 0.91423, respectively.

G4419: The same value as for PAN is assumed.

G4420: Products are as in G4415. Only the main channels for each isomer are considered. Rate constant is the weighted average for the isomers.

G4437e: LHMVKABO2 is a lumped species of virtual composition 0.3 HMKVKA02 + 0.7 HMKVKBO2. The products are the weighted average for the permutation reactions of each single RO2 in the MCM (Rickard and Pascoe, 2009).

G4439ea: products are the weighted average for the decomposition of 0.3 HMKVKA0 + 0.7 HMKVKBO.

G4440e: as for G4439ea

G4441e: The rate coefficient and products are 30% for HMKVKA0OH and 70% for HMKVKBOOH.

G4455: CH3COCOCO3H assumed to photolyse quickly and give CH3CO3 + CO + CO2 + OH

G4462e: EZCH3CO2CHCHO 90:10 mixture of the Z- and E-isomer, CH3COCHO2CHO → .1 CH3CO3 + .1 GLYOX + .9 CH3COCHOOHCHO → .1 CH3CO3 + .1 GLYOX + .9 CH3COCOCHO → CH3CO3 + .1 GLYOX + 1.8 CO + .9 HO2; $k_{H-shift}^{1,6}$

G4463e: $k_{H-shift}^{1,6}$

G4500e: MVKO2 substituted with HCHO + ACO3 = 1.33 HCHO + .67 HCOCH2O2 + .33 CO

G4501e: coefficients from Paulot et al(2009)ACP

G4503e: k=.3E-12*.77+1.4E-12*.23

G4504e: k=1.5E-12*2.

G4505e: k=.3E-12*.46+1.E-12*.54

G4506e: k=1.5E-12*2.

G4507e: eisopao2 60percent zisopao2 40percent, MIME species

G4508e: as for G4507e

G4509e: eisopco2 42percent and zisopao2 58percent

G4510e: as for G4509e

G4511e: $k_{H-shift}^{1,6}$

G4512e: $k_{H-shift}^{1,6}$

G4521ec: k parent alkene, branching from Rickard et al(1999)

G45222: HYBIACETOOA approximated to yield BIACETOH only, CH3COCHOOHCH2OH's main reaction with OH yields BIACETOH recycling OH → substitution with BIACETOH

G4537e: O2NOCH2COCH2O2 = NO2 + 2 HCHO + CO2

G4550e: LC578OOH = .6 C57OOH + .4 C58OOH

G4550et2: LC578OOH = .6 C57OOH + .4 C58OOH

G4551e: Peroxy vinyl radical chemistry neglected

G4552e: Peroxy vinyl radical chemistry neglected

G4553e: Peroxy vinyl radical chemistry neglected

G4555e: Peroxy vinyl radical chemistry neglected

G4577et3: from MIME ZCODC23DBCOOH = 0.64 ZC1ODC23DBC4OOH + 0.36 ZC1OOHC23DBC4OD, HOOCH2COCH2O2 substituted with HCO-COCH2OOH

G4581et8: C1ODC3OOHC4OOH → C1ODC3ODC4OOH → .5 CH3CO3 + .5 CO + .5 MGLYOX + .5 HO2 + HOOCH2CO3

G4581et10: C1ODC3OOHC4OD + OH → .25 C1ODC3ODC4OD + .375 CO + .375 CH3COCOCHO + .375 CO2 + .375 HCOCOHCH3CHO → .25(MGLYOX + 2 CO + 2 HO2) + .375 CO + .375(CH3CO3 + 2 CO + HO2) + .375 CO2 + .375(MGLYOX + CO + 2 HO2)

G4582e: $k_{H-shift}^{1,6}$

G4585et3: approximation H-abstraction only at C1

G4598et3: ZC1O3HC23DBC4OD 62percent G4598et4: HCOCOHCH3CHO substituted with CH3COCHOH + CO + HO2

G4136: CH3COCH2OOCH2CO3 → CH3CO3 + CO2 + 2 HCHO

G45mbo11: approximated to yield only MBOACO

G45mbo15: approx. not as in MCM!!!

G45mbo43: NO3CH2CO3 = HCHO + CO2 + NO2 following to Paulot et al(2009)

G410apin1: H-abstraction channels neglected, branching ratios re-scaled. G410apin2: H-abstraction channels neglected, branching ratios re-scaled. G410apin3: H-abstraction channels neglected, branching ratios re-scaled.

G410apint150: endo-cyclic vinyl ROOH assumed not be formed

G410apint162: C920CO3, C109CO, C109OH neglected

G410apint191: C811NO3 neglected

G410apint196: C812OH neglected

G410apint201: C813OH neglected

G410apint202: C813NO3 neglected

G410apint237: C512OH and CO13C4CHO neglected

G410apint239: C512NO3 neglected

G410apint242: C513OH + C513CO neglected

G410apint277: C33CO is called HCOCOCHO here

G410apint295: NC101CO substituted with C96CO3

G410bpin: BPINBO2 minor and similar to BPINAO2, peroxy ring closure of BPINCO2 is 0.6/s fast and traditional chemistry is neglected BPINCO2=LVROO6R1O2. G410bpint2: BPINBO2 minor and similar to BPINAO2, peroxy ring closure of BPINCO2 is 0.6/s fast and traditional chemistry is neglected BPINCO2=LVROO6R1O2.

G410bpint14: 1,5-H-shift

G410bpint19: LVROO6R13OO formation neglected

G410bpint20: traditional reactions neglected because of complexity

G410bpint40: .167 is for the neglected lactone stemming from the dioxirane as C8BC

G410bpint61: BPINBNO3 produced but approximated by BPINANO3

G410apint296: 2- and 3-CARENE together treated as the more emitted 3-CARENE, being similar to AP-INENE and assumed to yield the same products.

G410apint303: endo-cyclic vinyl ROOH assumed not be formed

G410myrct7: 20percent alkylnitrate yield like for AP-INENE

G410myrct15: $k_{H-shift}^{1,6}$

G410myrct29: $k_{H-shift}^{1,6}$

G410myrct36: $k_{H-shift}^{1,6}$

G410myrct53: k acetic acid + OH

G410myrct42: Surrogate acetaldehyde.

G410afarn: k for addition to the isoprene unit

G410afarnt2: k for 2-methyl-2-butene

G410afarnt3: O3 + outer double bond k for 2-methyl-2-butene, OH-yield might be much lower

G410afarnt4: O3 + inner double bond, k for 2-methyl-2-butene, MYRCCHO approximation, OH-yield might be much lower

G410afarnt5: k sum of 4-methylhexa-3,5-dienal (Baker et al., 2004) and twice 2-methyl-2-butene.

G410afarnt16: $k_{H-shift}^{1,6}$

G410afarnt21: k for 2-methyl-2-butene

G410afarnt22: k for 2-methyl-2-butene

G410afarnt23: k for addition to the isoprene unit

G410afarnt24: k for 2-methyl-2-butene

G410afarnt25: k of 4-methylhexa-3,5-dienal (Baker et al., 2004).

G410afarnt26: k of 2-methyl-2-butene

G410afarnt31: k for addition to the isoprene unit

G410afarnt32: k of 2-methyl-2-butene

G410afarnt37: $k_{H-shift}^{1,6}$

G410afarnt43: 1.57E-10 for k was relative to k(trans-2-butene), with the new k=5.662E-11(IUPAC), k(MHO+OH) is 13percent lower

G410afarnt44: .25 GLYOX + .25 CH3COCH2O2 maor products of CH3OCCH2CHO2CHO ox.

G410afarnt52: CO2H3CO3 approximation

G410afarnt55: Approximation: OPACO3 is CH3COCH2O2 + SURCH3CO3.

Table 2: Photolysis reactions

#	labels	reaction	rate coefficient	reference
J1000	StTrGJ	$O_2 + h\nu \rightarrow O(^3P) + O(^3P)$	jx(ip_02)	see note
J1001a	StTrGJ	$O_3 + h\nu \rightarrow O(^1D)$	jx(ip_01D)	see note
J1001b	StTrGJ	$O_3 + h\nu \rightarrow O(^3P)$	jx(ip_03P)	see note
J2101	StTrGJ	$H_2O_2 + h\nu \rightarrow 2 OH$	JX(ip_H202)	see note
J3101	StTrGNJ	$NO_2 + h\nu \rightarrow NO + O(^3P)$	jx(ip_N02)	see note
J3103a	StTrGNJ	$NO_3 + h\nu \rightarrow NO_2 + O(^3P)$	jx(ip_N020)	see note
J3103b	StTrGNJ	$NO_3 + h\nu \rightarrow NO$	jx(ip_N002)	see note
J3104a	StTrGNJ	$N_2O_5 + h\nu \rightarrow NO_2 + NO_3$	jx(ip_N205)	see note
J3200	TrGJ	$HONO + h\nu \rightarrow NO + OH$	JX(ip_H0N0)	see note
J3201	StTrGNJ	$HNO_3 + h\nu \rightarrow NO_2 + OH$	JX(ip_HN03)	see note
J3202	StTrGNJ	$HNO_4 + h\nu \rightarrow .667 NO_2 + .667 HO_2 + .333 NO_3 + .333 OH$	JX(ip_HN04)	see note
J4100e	StTrGJ	$CH_3OOH + h\nu \rightarrow HCHO + OH + HO_2$	1.14*jx(ip_CH300H)	see note
J4101a	StTrGJ	$HCHO + h\nu \rightarrow H_2 + CO$	jx(ip_COH2)	see note
J4101b	StTrGJ	$HCHO + h\nu \rightarrow H + CO + HO_2$	jx(ip_CH0H)	see note
J4104e	StTrGJ	$HOCH_2OOH + h\nu \rightarrow OH + HO_2 + HCOOH$	1.14*jx(ip_CH300H)	see note
J4200e	TrGCJ	$C_2H_5OOH + h\nu \rightarrow CH_3CHO + HO_2 + OH$	1.14*jx(ip_CH300H)	see note
J4201	TrGCJ	$CH_3CHO + h\nu \rightarrow CH_3O_2 + HO_2 + CO$	jx(ip_CH3CHO)	see note
J4202e	TrGCJ	$CH_3C(O)OOH + h\nu \rightarrow CH_3O_2 + OH + CO_2$	1.14*jx(ip_CH3C03H)	see note
J4204e	TrGNJCJ	$PAN + h\nu \rightarrow .6 CH_3C(O)OO + .6 NO_2 + .4 CH_3O_2 + .4 CO_2 + .4 NO_3$	jx(ip_PAN)	see note
J4205ae	TrGCJ	$HOCH_2CHO + h\nu \rightarrow HCHO + 2 HO_2 + CO$	jx(ip_HOCH2CHO)*0.70	Taraborrelli (2014)
J4205be	TrGCJ	$HOCH_2CHO + h\nu \rightarrow 1.16 OH + .84 HCOCH_2O_2 + .1 HCHO + .1 CO + .06 GLYOX$	jx(ip_HOCH2CHO)*0.15	Taraborrelli (2014)
J4205ce	TrGCJ	$HOCH_2CHO + h\nu \rightarrow CH_3OH + CO$	jx(ip_HOCH2CHO)*0.15	Taraborrelli (2014)
J4206e	TrGCJ	$HOCH_2CHO + h\nu \rightarrow OH + HCHO + CO + HO_2$	1.14*jx(ip_CH300H)+jx(ip_HOCH2CHO)	Taraborrelli (2014)
J4206et2	TrGCJ	$HOCH_2CO_3H + h\nu \rightarrow HCHO + HO_2 + OH + CO_2$	1.14*jx(ip_CH300H)	Rickard and Pascoe (2009)*
J4207	TrGCJ	$PHAN + h\nu \rightarrow HOCH_2CO_3 + NO_2$	jx(ip_PAN)	see note
J4208	TrGCJ	$GLYOX + h\nu \rightarrow 2 CO + 2 HO_2$	jx(ip_GLYOX)	see note
J4209	TrGNJCJ	$HCOCO_2H + h\nu \rightarrow 2 HO_2 + CO + CO_2$	jx(ip_MGLYOX)	Rickard and Pascoe (2009)*
J4210e	TrGNJCJ	$HCOCO_3H + h\nu \rightarrow HO_2 + CO + OH + CO_2$	1.14*jx(ip_CH300H)+jx(ip_HOCH2CHO)	Rickard and Pascoe (2009)*

Table 2: Photolysis reactions (... continued)

#	labels	reaction	rate coefficient	reference
J4211e	TrGCJ	HYETHO2H + $h\nu$ → HOCH ₂ CH ₂ O + OH	1.14*jx(ip_CH300H)	Rickard and Pascoe (2009)*
J4212	TrGCJ	ETHOHNO3 + $h\nu$ → HO ₂ + 2 HCHO + NO ₂	J_IC3H7N03	see note
J4213e	TrGCJ	HOCH2CO3H + $h\nu$ → OH + HCHO + CO ₂ + OH	2*1.14*jx(ip_CH300H)	Taraborrelli (2014)
J4214e	TrGC	HOCH2CO2H + $h\nu$ → OH + HCHO + HO ₂ + CO ₂	1.14*jx(ip_CH300H)	Taraborrelli (2014)
J4215e	TrGC	CH2CO + $h\nu$ → .4 CO ₂ + .8 H + .34 CO + .34 OH + .34 HO ₂ + .16 HCHO + .16 O(³ P) + .1 HCOOH + CO	J_ketene* 0.36	Taraborrelli (2014)
J4216e	TrGC	CH3CHOHOOH + $h\nu$ → CH ₃ O ₂ + HCOOH + OH	1.14*jx(ip_CH300H)	Taraborrelli (2014)
J4217e	TrGCJ	NO3CH2CHO + $h\nu$ → HO ₂ + CO + HCHO + NO ₂	1.59*J_IC3H7N03+jx(ip_CH3C0CH3)	Taraborrelli (2014)
J4300e	TrGCJ	iC ₃ H ₇ OOH + $h\nu$ → CH ₃ COCH ₃ + HO ₂ + OH	1.14*jx(ip_CH300H)	von Kuhlmann (2001)*
J4301	TrGCJ	CH ₃ COCH ₃ + $h\nu$ → CH ₃ C(O)OO + CH ₃ O ₂	jx(ip_CH3C0CH3)	see note
J4302	TrGCJ	CH ₃ COCH ₂ OH + $h\nu$ → CH ₃ C(O)OO + HCHO + HO ₂	J_ACETOL	see note
J4303	TrGCJ	MGLYOX + $h\nu$ → CH ₃ C(O)OO + CO + HO ₂	jx(ip_MGLYOX)	see note
J4304e	TrGCJ	CH ₃ COCH ₂ O ₂ H + $h\nu$ → CH ₃ C(O)OO + HCHO + OH	1.14*jx(ip_CH300H)+J_ACETOL	Taraborrelli (2014)
J4305e	TrGCJ	HOCH2COCH2OOH + $h\nu$ → HOCH2CO + HCHO + OH	1.14*jx(ip_CH300H)+J_ACETOL	Taraborrelli (2014)
J4306	TrGNCJ	iC ₃ H ₇ ONO ₂ + $h\nu$ → CH ₃ COCH ₃ + NO ₂ + HO ₂	J_IC3H7N03	von Kuhlmann et al. (2003)*
J4307	TrGCJ	NOA + $h\nu$ → CH ₃ C(O)OO + HCHO + NO ₂	J_IC3H7N03+jx(ip_CH3C0CH3)	see note
J4309e	TrGCJ	HYPPO2H + $h\nu$ → CH ₃ CHO + HCHO + HO ₂ + OH	1.14*jx(ip_CH300H)	Taraborrelli (2014)
J4310e	TrGNCJ	PR2O2HNO3 + $h\nu$ → NOA + HO ₂ + OH	1.14*jx(ip_CH300H)	Taraborrelli (2014)
J4311e	TrGCJ	HOCH2COCHO + $h\nu$ → HOCH2CO + CO + HO ₂	jx(ip_MGLYOX)	Taraborrelli (2014)
J4312e	TrGCJ	CH3COCO2H + $h\nu$ → .5 CH ₃ CHO + .8 CO ₂ + .4 CH ₃ C(O)OO + .3 HO ₂ + .1 CH ₃ COOH + .1 OH + .2 CO	JX(IP_MGLYOX)	Taraborrelli (2014)
J4313e	TrGCJ	HCOCOCH2OOH + $h\nu$ → HCOCO3A + HCHO + OH	1.14*jx(ip_CH300H)+J_ACETOL	Taraborrelli (2014)
J4314e	TrGCJ	HCOCOCH2OOH + $h\nu$ → HOOCH2CO3 + CO + HO ₂	JX(IP_MGLYOX)	Taraborrelli (2014)
J4315e	TrGCJ	HCOCOCHO + $h\nu$ → HCOCO3A + HO ₂ + CO	2*JX(IP_MGLYOX)	Taraborrelli (2014)
J4316e	TrGC	CH3COCO3H + $h\nu$ → CH ₃ C(O)OO + OH + CO ₂	JX(IP_MGLYOX)+1.14*jx(ip_CH300H)	Taraborrelli (2014)
J4317e	TrGC	CH3CHCO + $h\nu$ → C ₂ H ₄ + CO	J_ketene*0.36*2.	Taraborrelli (2014)*
J4400e	TrGCJ	LC ₄ H ₉ OOH + $h\nu$ → OH + 0.254 CO ₂ + 0.5552 MEK + 0.5552 HO ₂ + 0.3178 CH ₃ CHO + 0.4448 C ₂ H ₅ O ₂	1.14*jx(ip_CH300H)	Rickard and Pascoe (2009)*
J4401	TrGCJ	MVK + $h\nu$ → C ₃ H ₆ + CO	jx(ip_MVK)	Taraborrelli (2014)*
J4403	TrGCJ	MEK + $h\nu$ → CH ₃ C(O)OO + C ₂ H ₅ O ₂	0.42*jx(ip_CHOH)	von Kuhlmann et al. (2003)*

Table 2: Photolysis reactions (... continued)

#	labels	reaction	rate coefficient	reference
J4404e	TrGCJ	$\text{LMEKOOH} + h\nu \rightarrow 0.538 \text{ HCHO} + 0.538 \text{ CO}_2 + 0.459 \text{ HOCH}_2\text{CH}_2\text{O}_2 + 0.079 \text{ C}_2\text{H}_5\text{O}_2 + 0.462 \text{ CH}_3\text{C(O)OO} + 0.462 \text{ CH}_3\text{CHO} + \text{OH}$	$1.14 * jx(ip_CH300H) + J_ACETOL$	Rickard and Pascoe (2009)*
J4405	TrGCJ	$\text{BIACET} + h\nu \rightarrow 2 \text{ CH}_3\text{C(O)OO}$	$2.15 * jx(ip_MGLYOX)$	see note
J4406	TrGNCJ	$\text{LC4H9NO}_3 + h\nu \rightarrow \text{NO}_2 + 0.254 \text{ CO}_2 + 0.5552 \text{ MEK} + 0.5552 \text{ HO}_2 + 0.3178 \text{ CH}_3\text{CHO} + 0.4448 \text{ C}_2\text{H}_5\text{O}_2$	$J_IC3H7N03$	see note
J4407e	TrGNCJ	$\text{MPAN} + h\nu \rightarrow .6 \text{ MACO}_3 + .6 \text{ NO}_2 + .4 \text{ MACO}_2 + .4 \text{ NO}_3$	$jx(ip_PAN)$	Taraborrelli (2014)*
J4409e	TrGCJ	$\text{CO}_2\text{H}_3\text{CO}_3\text{H} + h\nu \rightarrow \text{CH}_3\text{COCHOH} + \text{OH} + \text{CO}_2$	$1.14 * jx(ip_CH300H)$	Taraborrelli (2014)
J4410	TrGCJ	$\text{CO}_2\text{H}_3\text{CO}_3\text{H} + h\nu \rightarrow \text{CH}_3\text{C(O)OO} + \text{HO}_2 + \text{HCOCO}_3\text{H}$	J_ACETOL	Rickard and Pascoe (2009)*
J4410t2	TrGCJ	$\text{CO}_2\text{H}_3\text{CO}_2\text{H} + h\nu \rightarrow \text{CH}_3\text{C(O)OO} + \text{HCOCO}_2\text{H} + \text{HO}_2$	J_ACETOL	Taraborrelli (2014)
J4411	TrGCJ	$\text{MACR} + h\nu \rightarrow .5 \text{ MACO}_3 + .5 \text{ MACO}_2 + .5 \text{ CO} + \text{HO}_2$	$jx(ip_MACR)$	see note
J4412e	TrGCJ	$\text{MACROOH} + h\nu \rightarrow \text{MACRO} + \text{OH}$	$1.14 * jx(ip_CH300H) + 2.77 * jx(ip_HOCH_2CHO)$	see note
J4414	TrGCJ	$\text{MACROH} + h\nu \rightarrow \text{CH}_3\text{COCH}_2\text{OH} + \text{CO} + \text{HO}_2 + \text{HO}_2$	$2.77 * jx(ip_HOCH_2CHO)$	see note
J4415e	TrGCJ	$\text{MACO}_3\text{H} + h\nu \rightarrow \text{MACO}_2 + \text{OH}$	$1.14 * jx(ip_CH300H)$	Taraborrelli (2014)
J4416e	TrGCJ	$\text{LHMVKABOOH} + h\nu \rightarrow .12 \text{ CH}_3\text{COCHOH} + .88 \text{ CH}_3\text{C(O)OO} + .88 \text{ HOCH}_2\text{CHO} + .12 \text{ HCHO} + \text{OH}$	$1.14 * jx(ip_CH300H) + J_ACETOL$	Taraborrelli (2014)
J4418e	TrGCJ	$\text{CO}_2\text{H}_3\text{CHO} + h\nu \rightarrow \text{CH}_3\text{COCHOH} + \text{CO} + \text{HO}_2$	$jx(ip_HOCH_2CHO) + J_ACETOL$	Taraborrelli (2014)
J4419	TrGCJ	$\text{HO}_{12}\text{CO}_3\text{C}_4 + h\nu \rightarrow \text{CH}_3\text{C(O)OO} + \text{HOCH}_2\text{CHO} + \text{HO}_2$	J_ACETOL	Rickard and Pascoe (2009)*
J4420e	TrGCJ	$\text{BIACETOH} + h\nu \rightarrow \text{CH}_3\text{C(O)OO} + \text{HOCH}_2\text{CO}$	$2.15 * jx(ip_MGLYOX)$	Taraborrelli (2014)
J4421e	TrGC	$\text{HCOCCH}_3\text{CO} + h\nu \rightarrow .5 \text{ OH} + .25 \text{ HCOCOCH}_2\text{O}_2 + .25 \text{ CH}_3\text{C(O)OO} + .5 \text{ CH}_3\text{CHCO} + .5 \text{ CO}$	J_KETENE	Taraborrelli (2014)
J4422e	TrGC	$\text{CH}_3\text{COCHCO} + h\nu \rightarrow .0192 \text{ CH}_3\text{COCO}_2\text{H} + .1848 \text{ H}_2\text{O}_2 + .2208 \text{ MGLYOX} + .36 \text{ OH} + .36 \text{ CO} + .56 \text{ CH}_3\text{C(O)OO} + .2 \text{ CH}_3\text{CHO} + .2 \text{ CO}_2 + .2 \text{ HCHO} + .2 \text{ HO}_2$	$J_KETENE * 0.5$	Taraborrelli (2014)
J4422et2	TrGC	$\text{CH}_3\text{COCHCO} + h\nu \rightarrow \text{CH}_3\text{CHCO} + \text{CO}$	$J_KETENE * 0.5$	Taraborrelli (2014)
J4423e	TrGCJ	$\text{CH}_3\text{COCOCHO} + h\nu \rightarrow \text{CH}_3\text{C(O)OO} + \text{CO} + \text{CO} + \text{HO}_2$	$jx(ip_MGLYOX)$	Taraborrelli (2014)
J4424e	TrGCJ	$\text{CH}_3\text{COCOCHO} + h\nu \rightarrow \text{CH}_3\text{C(O)OO} + \text{HCOCO}_3\text{A}$	$2.15 * jx(ip_MGLYOX)$	Taraborrelli (2014)
J4424et2	TrGC	$\text{CH}_3\text{COCOCO}_2\text{H} + h\nu \rightarrow \text{CH}_3\text{C(O)OO} + \text{CO} + \text{CO}_2 + \text{HO}_2$	$3.15 * jx(ip_MGLYOX)$	Taraborrelli (2014)
J4502et2	TrGCJ	$\text{LISOPACOOH} + h\nu \rightarrow \text{LHC}_4\text{ACCHO} + \text{HO}_2 + \text{OH}$	$1.14 * jx(ip_CH300H)$	Rickard and Pascoe (2009)*
J4503et2	TrGNCJ	$\text{LISOPACNO}_3 + h\nu \rightarrow \text{LHC}_4\text{ACCHO} + \text{HO}_2 + \text{NO}_2$	$0.59 * J_IC3H7N03$	see note

Table 2: Photolysis reactions (... continued)

#	labels	reaction	rate coefficient	reference
J4504e	TrGCJ	ISOPBOOH + $h\nu$ → MVK + HCHO + HO ₂ + OH	1.14*jx(ip_CH300H)	Rickard and Pascoe (2009)*
J4505e	TrGNCJ	ISOPBNO ₃ + $h\nu$ → MVK + HCHO + HO ₂ + NO ₂	2.84*J_IC3H7N03	see note
J4506e	TrGCJ	ISOPDOOH + $h\nu$ → MACR + HCHO + HO ₂ + OH	1.14*jx(ip_CH300H)	Rickard and Pascoe (2009)*
J4507	TrGNCJ	ISOPDNO ₃ + $h\nu$ → MACR + HCHO + HO ₂ + NO ₂	J_IC3H7N03	see note
J4508e	TrGNCJ	NISOPOOH + $h\nu$ → NC4CHO + HO ₂ + OH	1.14*jx(ip_CH300H)	Rickard and Pascoe (2009)*
J4509	TrGNCJ	NC4CHO + $h\nu$ → NOA + 2 CO + 2 HO ₂	jx(ip_MACR)	see note
J4510e	TrGNCJ	LNISOOH + $h\nu$ → NOA + OH + .5 HOCHCHO + .5 CO + .5 HO ₂ + .5 CO ₂	1.14*jx(ip_CH300H)	Taraborrelli et al. (2009)*
J4511e	TrGCJ	LHC4ACCHO + $h\nu$ → .5 LHC4ACCO ₃ + .5 HO ₂ + .5 OH + .25 MACRO ₂ + .25 LHMVKABO ₂	jx(ip_MACR)	Taraborrelli (2014)*
J4512e	TrGCJ	LC578OOH + $h\nu$ → .25 CH ₃ COCH ₂ OH + .75 MGLYOX + .25 HOCHCHO + .75 HOCH ₂ CHO + .75 HO ₂ + OH	1.14*jx(ip_CH300H)+ 2.77*jx(ip_HOCH2CHO)	Taraborrelli (2014)
J4513e	TrGCJ	LHC4ACCO ₃ H + $h\nu$ → OH + .5 MACRO ₂ + .5 LHMVKABO ₂ + OH + CO ₂	J_HPALD	Taraborrelli (2014)*
J4514	TrGNCJ	LC5PAN1719 + $h\nu$ → .6 LHC4ACCO ₃ + .6 NO ₂ + .2 MACRO ₂ + .2 LHMVKABO ₂ + .4 CO ₂ + .4 NO ₃	jx(ip_PAN)	see note
J4515e	TrGCJ	HCOC ₅ + $h\nu$ → MACO ₂ + HOCH ₂ CO	0.5*jx(ip_MVK)	Taraborrelli (2014)
J4516e	TrGCJ	C59OOH + $h\nu$ → CH ₃ COCH ₂ OH + HOCH ₂ CO + OH	J_ACETOL+1.14*jx(ip_CH300H)	Taraborrelli (2014)
J4517e	TrGCJ	ZCODC23DBCOOH + $h\nu$ → LHC4ACCO ₃ + OH	J_HPALD	Taraborrelli (2014)
J4518e	TrGCJ	ZCO3HC23DBCOD + $h\nu$ → .62 EZCH3CO ₂ CHCHO + .38 EZCHOCCH ₃ CHO ₂ + OH + CO ₂	J_HPALD	Taraborrelli (2014)
J4519e	TrGCJ	C1OOHC2OOHC4OD + $h\nu$ → CH ₃ COCH ₂ O ₂ H + OH + 2 CO + HO ₂	2.77*JX(IP_HOCH2CHO)	Taraborrelli (2014)
J4520e	TrGCJ	C1OOHC2OOHC4OD + $h\nu$ → .5 CH ₃ COCH ₂ O ₂ H + .5 HOCHCHO + .5 CO ₂ H ₃ CHO + .5 HCHO + 1.5 OH	2.*1.14*JX(IP_CH300H)	Taraborrelli (2014)
J4523e	TrGCJ	C1ODC2OOHC4OD + $h\nu$ → MGLYOX + HOCHCHO + OH	1.14*JX(IP_CH300H)	Taraborrelli (2014)
J4524e	TrGCJ	C1ODC2OOHC4OD + $h\nu$ → CO ₂ H ₃ CHO + CO + HO ₂ + OH	2.*2.77*JX(IP_HOCH2CHO)	Taraborrelli (2014)*
J4525	TrGCJ	PEROXYRINGC2OOH + $h\nu$ → CH ₃ COCH ₂ OOCH ₂ CHO + HO ₂ + OH	1.14*JX(IP_CH300H)	Taraborrelli (2014)
J4526	TrGCJ	PEROXYRINGC2OOH + $h\nu$ → HCHO + OH + HO ₂ + CO ₂ H ₃ CHO	1.14*JX(IP_CH300H)	Taraborrelli (2014)*

Table 2: Photolysis reactions (... continued)

#	labels	reaction	rate coefficient	reference
J4525e	TrGCJ	ZCODC23DBCOD + $h\nu$ → .5 CH ₃ COCHCO + .5 HCOCCH ₃ CO + CO + HO ₂ + OH	jx(ip_N02)*0.1*0.5	Taraborrelli (2014)
J4526e	TrGCJ	CH ₃ COCH ₂ OOCH ₂ CHO + $h\nu$ → CH ₃ C(O)OO + HCHO + GLYOX + HO ₂	1.14*JX(IP_CH300H)+J_ACETOL	Taraborrelli (2014)
J4527e	TrGCJ	CH ₃ COCH ₂ OOCH ₂ CHO + $h\nu$ → CH ₃ C(O)OO + HCHO + HCHO + CO + HO ₂	jx(ip_H0CH2CHO)	Taraborrelli (2014)

*Notes:

J-values are calculated with an external module and then supplied to the MECCA chemistry.

Values that originate from the Master Chemical Mechanism (MCM) by Rickard and Pascoe (2009) are translated according in the following way:

J(11) → jx(ip_COH2)
J(12) → jx(ip_CHOH)
J(15) → jx(ip_HOCH2CHO)
J(18) → jx(ip_MACR)
J(22) → jx(ip_ACETOL)
J(23)+J(24) → jx(ip_MVK)
J(31)+J(32)+J(33) → jx(ip_GLYOX)
J(34) → jx(ip_MGLYOX)
J(41) → jx(ip_CH300H)
J(53) → J(iC₃H₇ONO₂)
J(54) → J(iC₃H₇ONO₂)
J(55) → J(iC₃H₇ONO₂)
J(56)+J(57) → jx(ip_NOA)

J4207: It is assumed that J(PHAN) is the same as J(PAN).

J4212: It is assumed that J(ETHOHNO₃) is the same as J(iC₃H₇ONO₂).

J4302: Following von Kuhlmann et al. (2003), we use J(CH₃COCH₂OH) = 0.11*jx(ip_CHOH). As an additional factor, the quantum yield of 0.65 is taken from Orlando et al. (1999a).

J4306: Following von Kuhlmann et al. (2003), we use J(iC₃H₇ONO₂) = 3.7*jx(ip_PAN).

J4307: NOA contains the chromophores of both CH₃COCH₃ and a nitrate group. It is assumed here that the J values are additive, i.e.: J(NO₃) = J(CH₃COCH₃) + J(iC₃H₇ONO₂).

J4317e: products Chong and Kistiakowsky 1964, 0.36 quantum yield at 334nm Kelley and Hase1975, factor 2 because of bigger structure and fit to C₂H₄LBA

J4401: Romero et al(2005)

J4406: It is assumed that J(LC4H₉NO₃) is the same as J(iC₃H₇ONO₂).

J4405: It is assumed that J(BIACET) is 2.15 times larger than J(MGLYOX), consistent with the photol-

ysis rate coefficients used in the MCM (Rickard and Pascoe, 2009).

J4414: It is assumed that J(MACROH) is 2.77 times larger than J(HOCH₂CHO), consistent with the photolysis rate coefficients used in the MCM (Rickard and Pascoe, 2009).

J4505: It is assumed that J(ISOPBNO₃) = 2.84 × J(iC₃H₇ONO₂), consistent with the photolysis rate coefficients used in the MCM (Rickard and Pascoe, 2009).

J4509: It is assumed that J(NC₄CHO) is the same as J(MACR).

J4511e: Peroxy vinyl radical chemistry approximated

J4513e: Peroxy vinyl radical chemistry approximated

J4514: It is assumed that J(LC5PAN1719) is the same as J(PAN).

J4524e: the loss of C₄OD is neglected

J4526: decomposition of a dialkoxy radical

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