



## *Corrigendum to*

# **“Quantifying the uncertainty in simulating global tropospheric composition due to the variability in global emission estimates of Biogenic Volatile Organic Compounds” published in Atmos. Chem. Phys., 13, 2857–2891, 2013**

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Some errors occurred in the Appendix Table A2 in the above-mentioned manuscript.

The products in the rows (1), (4) and (5) have been corrected. The reactant name for acetone has been changed from  $\text{CH}_3\text{COCO}_3$  to  $\text{CH}_3\text{COCH}_3$ .

**Table A2.** The photolytic reaction rates included in the modified CB05 chemical mechanism. The reaction products  $O_2$  and  $H_2O$  are not shown. The stoichiometry of each photolytic reaction is taken from Yarwood et al. (2005) except for the photolysis of  $O_2$ . The source of the absorption co-efficients and quantum yields are thus: [1] Sander et al. (2011), [2] Matsumi et al. (2002), [3] Atkinson et al. (2006). Further details: <sup>[a]</sup> absorption cross section for a C4 mono-nitrate are adopted (Roberts and Fayer, 1989), <sup>[b]</sup> the absorption cross section for  $CH_3CHO$  is adopted, <sup>[c]</sup> set equal to the photolysis frequency of  $CH_3OOH$  due a lack of laboratory measurements and <sup>[d]</sup> the absorption cross sections are set equal to an average of methyl vinyl ketone and methacrolein.

Chemical species	Products	Reference
$O_3$	$O^1D$	[1,2]
$NO_2$	$NO + O_3$	[1]
$H_2O_2$	$OH + OH$	[1]
$HNO_3$	$OH + NO_2$	[1]
$HNO_4$	$NO_2 + HO_2$	[1]
$N_2O_5$	$NO_2 + NO_3$	[1]
$CH_2O$	$CO$	[3]
$CH_2O$	$CO + HO_2 + HO_2$	[3]
$CH_3OOH$	$HCHO + OH + HO_2$	[1]
PAN	$C_2O_3 + NO_2$	[1]
$NO_3$	$NO_2 + O_3$	[1]
$NO_3$	$NO$	[1]
ORGNTR <sup>[a]</sup>	$NO_2 + 0.51XO_2 + 0.3 ALD2 + 0.9 HO_2 + 0.74 C_2O_3 + 0.74 CH_3O_2 + 1.98 RXP$	[3]
ALD2 <sup>[b]</sup>	$CH_3O_2 + CO + 2HO_2$	[3]
ROOH <sup>[c]</sup>	$OH + 0.5 XO_2 + 0.3 ALD2 + 0.9 HO_2 + 0.74 C_2O_3 + 0.74 CH_3O_2 + 1.98 RXP$	[1]
$CH_3COCHO$	$C_2O_3 + HO_2 + CO$	[1]
$O_2$	$2O(^3P)$	[1]
ISPD <sup>[d]</sup>	$0.333 CO + 0.067 ALD2 + 0.9 HCHO + 0.832 PAR + 1.033 HO_2 + 0.7 XO_2 + 0.967 C_2O_3$	[3]
$CH_3COCH_3$	$CO + 2 CH_3O_2$	[3]
$CH_3COCH_3$	$C_2O_3 + CH_3O_2$	[3]