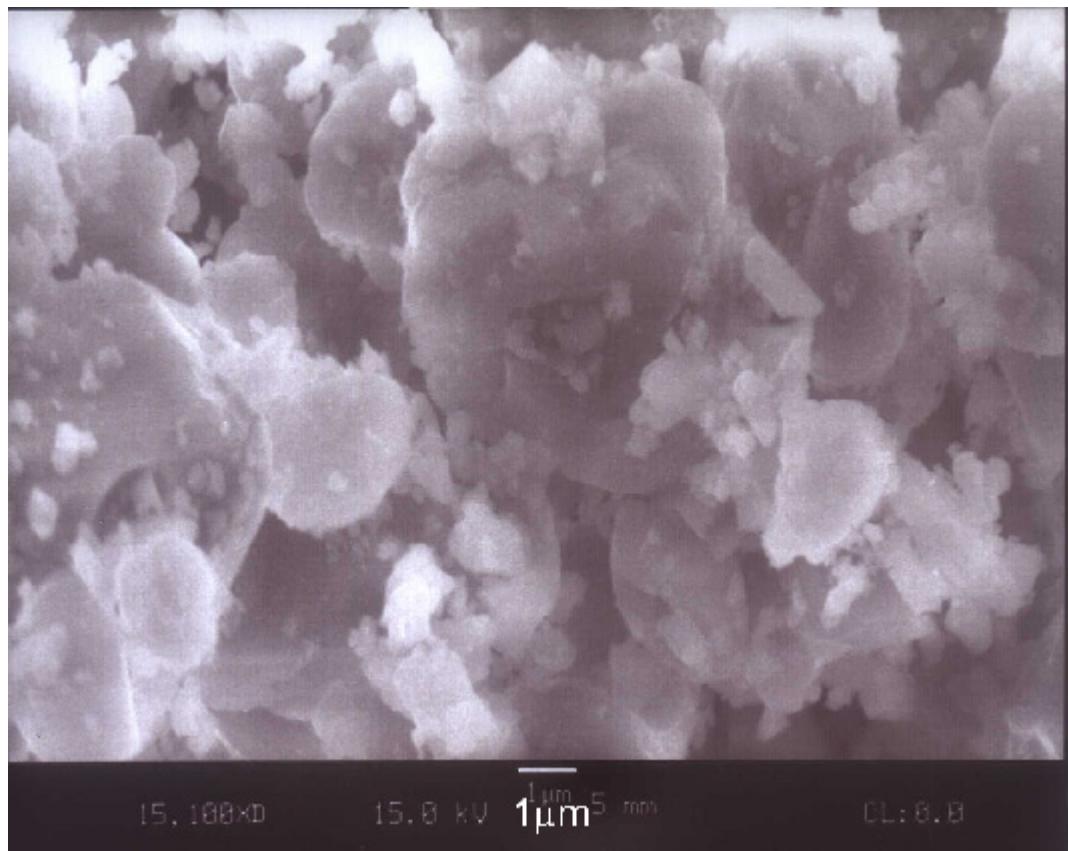
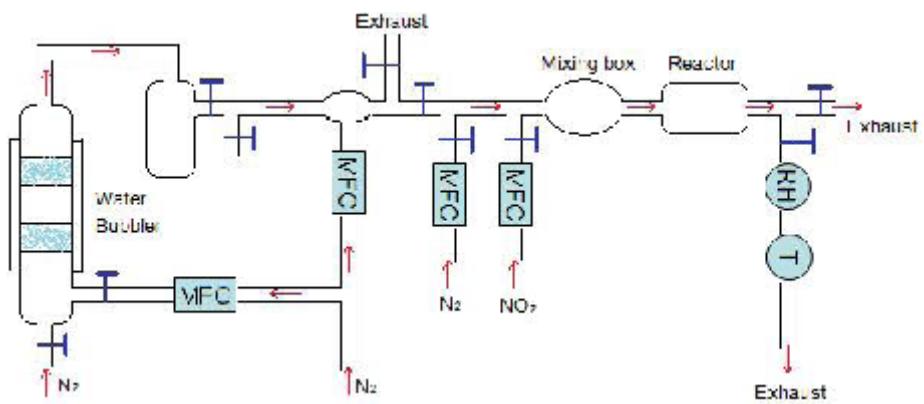


**Supplementary material for**

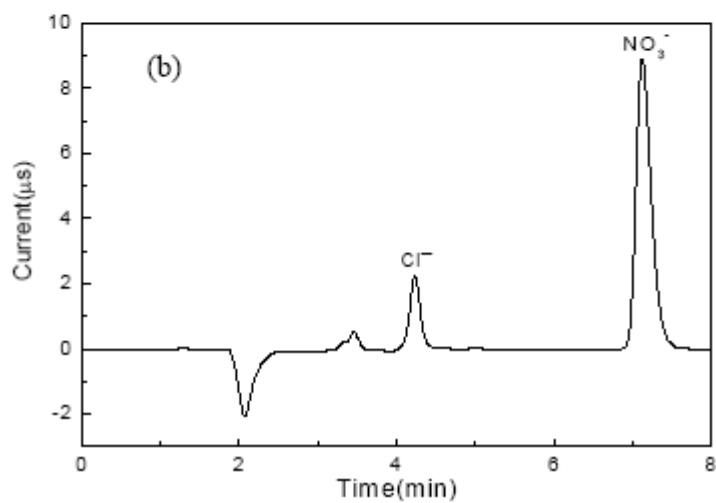
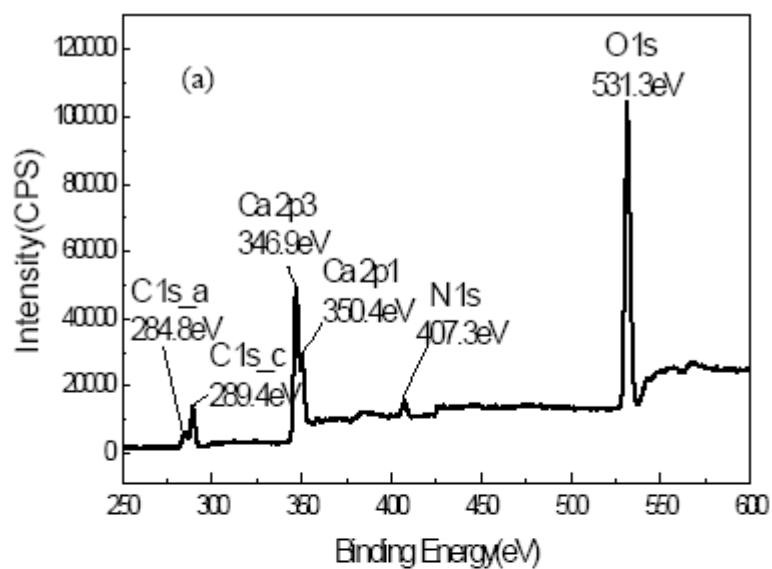
**“Kinetics and Mechanisms of Heterogeneous Reaction of NO<sub>2</sub> on CaCO<sub>3</sub> Surfaces under Dry and Wet Conditions”**



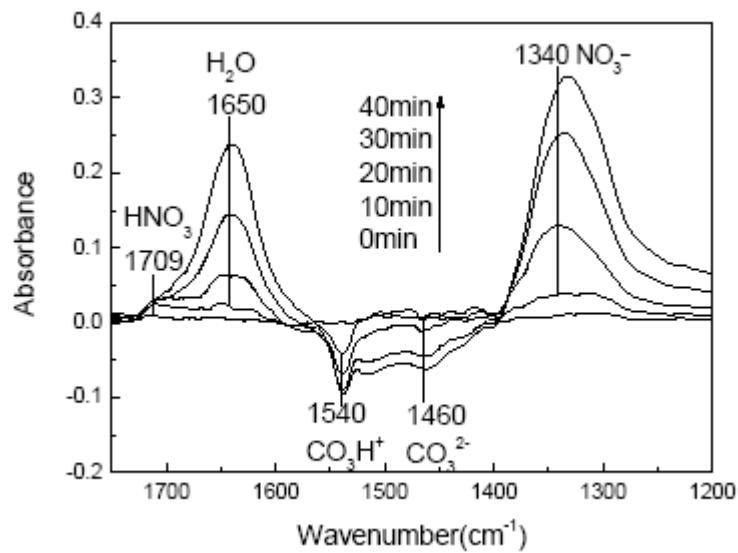
**Figure S1.** SEM picture of ground CaCO<sub>3</sub>. The scale is 1  $\mu\text{m}$ . The size of the ground CaCO<sub>3</sub> was in the range of 1-6  $\mu\text{m}$ .



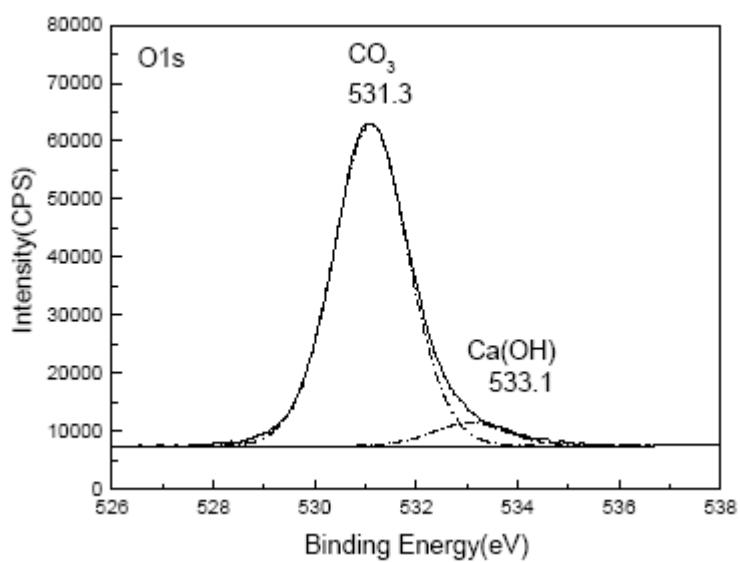
**Figure S2.** Apparatus for gas preparation.



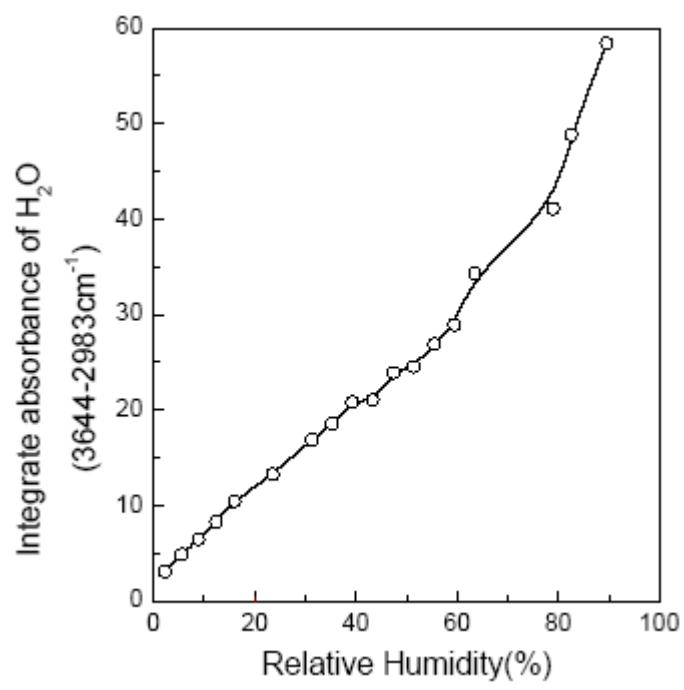
**Figure S3.** Typical XPS spectrum (a) and IC chromatogram (b) of  $\text{CaCO}_3$  particle after reacted with  $1.06 \times 10^{15}$  molecules  $\text{cm}^{-3}$   $\text{NO}_2$  for 626 min.



**Figure S4.** Typical IR spectrum of  $\text{CaCO}_3$  after reacted with  $\text{HNO}_3$  for 0 to 40 min.



**Figure S5.** XPS spectrum of O1s at the surface of CaCO<sub>3</sub> particles.



**Figure S6.** Isotherm adsorption curve of water vapor on  $\text{CaCO}_3$  particles at 296 K. IR absorbance of the stretching vibration of –OH of adsorbed water is integrated between 3644 and  $2983\text{ cm}^{-1}$ .